

The metabolic functions of the liver

Catabolism of haemoglobin, bilirubin

Metabolism of iron

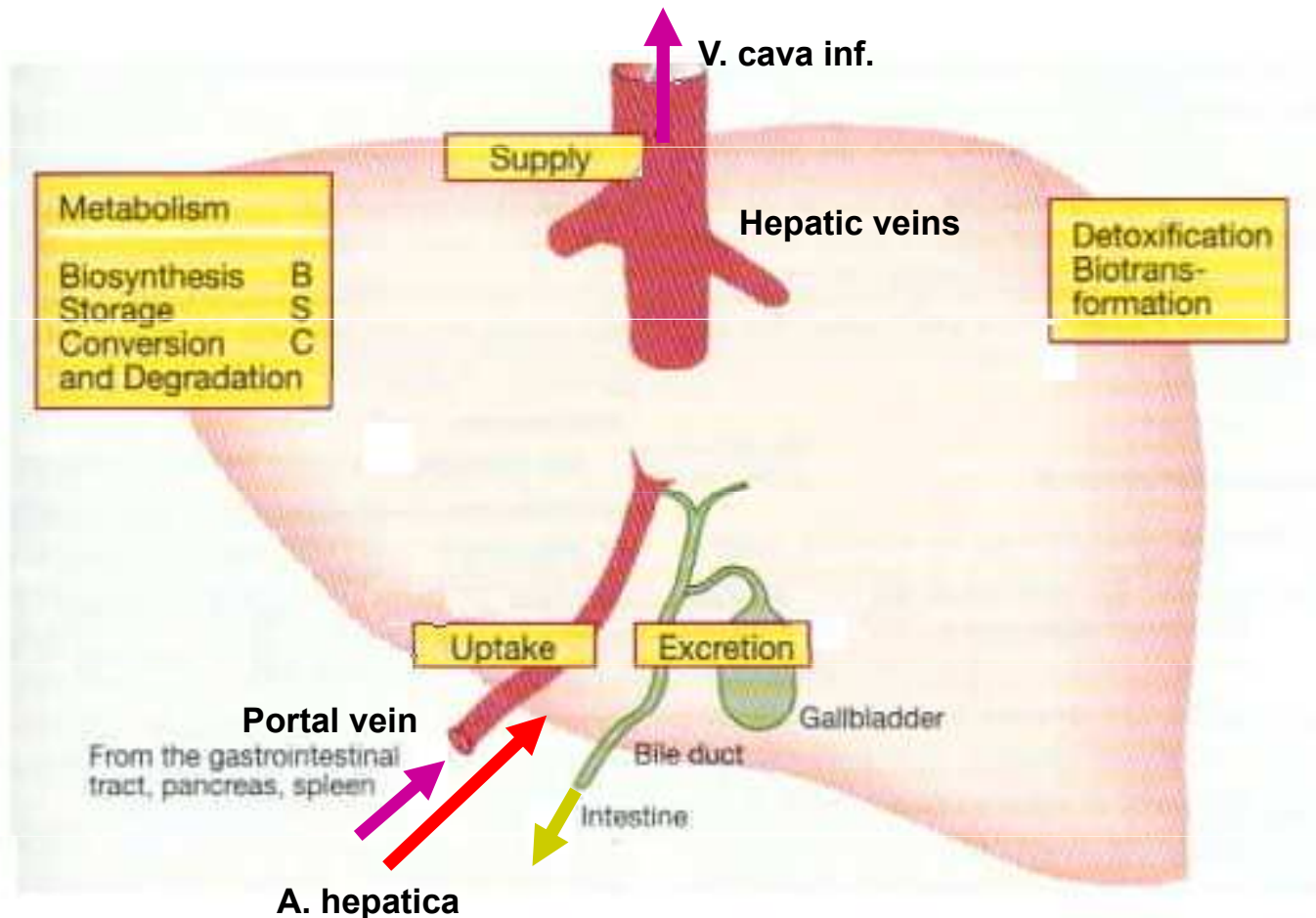
Biochemistry II

Lecture 4

2008 (J.S.)

The major metabolic functions of the liver:

- uptake of most nutrients from the gastrointestinal tract,
- intensive intermediary metabolism, conversion of nutrients,
- controlled supply of essential compounds (glucose, VLD lipoproteins, ketone bodies, plasma proteins, etc.),
- ureosynthesis, biotransformation of xenobiotics (detoxification),
- excretion (cholesterol, bilirubin, hydrophobic compounds, some metals).



The hepatocytes (the hepatic parenchymal cells) have an immensely broad range of synthetic and catabolic functions with a substantial reserve metabolic capacity.

Many of them are the **specialized metabolic functions of the liver:**

Metabolism of saccharides

- Primary regulation of the blood glucose concentration. E.g. in the postprandial state, there is an uptake of about 60 % of glucose supplied in portal blood and stored as glycogen, or in hypoglycaemia, glycogenolysis and gluconeogenesis is initiated.
- The liver cells meet their energy requirements preferentially from fatty acids, not from glycolysis. Glucose (also as glycogen store) is altruistically spared for extrahepatic tissues.

Metabolism of lipids

- Completion and secretion of VLDL and HDL.
- Ketogenesis produces ketone bodies, precious nutrients. They cannot be utilized in the liver, but they are supplied to other tissues.
- Secretion of cholesterol and bile acids into the bile represents the major way of cholesterol elimination from the body.
- Dehydrogenation of cholesterol to 7-dehydrocholesterol and 25-hydroxylation of calciols play an essential role in calcium homeostasis.

Metabolism of nitrogenous compounds

- Deamination of amino acids that are in excess of requirements.
- Intensive proteosynthesis of major plasma proteins and blood-clotting factors.
- Uptake of ammonium, ureosynthesis.
- Bilirubin capturing, conjugation, and excretion.

Biotransformation of xenobiotics

- Detoxification of drugs, toxins, excretion of some metals.

Transformation of hormones

- Inactivation of steroid hormones – hydrogenation, conjugation.
- Inactivation of insulin, about 50 % insulin inactivated in its only passage through the liver (GSH:insulin transhydrogenase splits the disulfide bonds, then proteolysis of the two chains).
- Inactivation of catecholamines and iodothyronines, conjugation of the products.

Vitamins

- Hydroxylation of calciols to calcidiols, splitting of β -carotene to retinol.
- The liver represent a store of retinol esters and cobalamin (B₁₂).

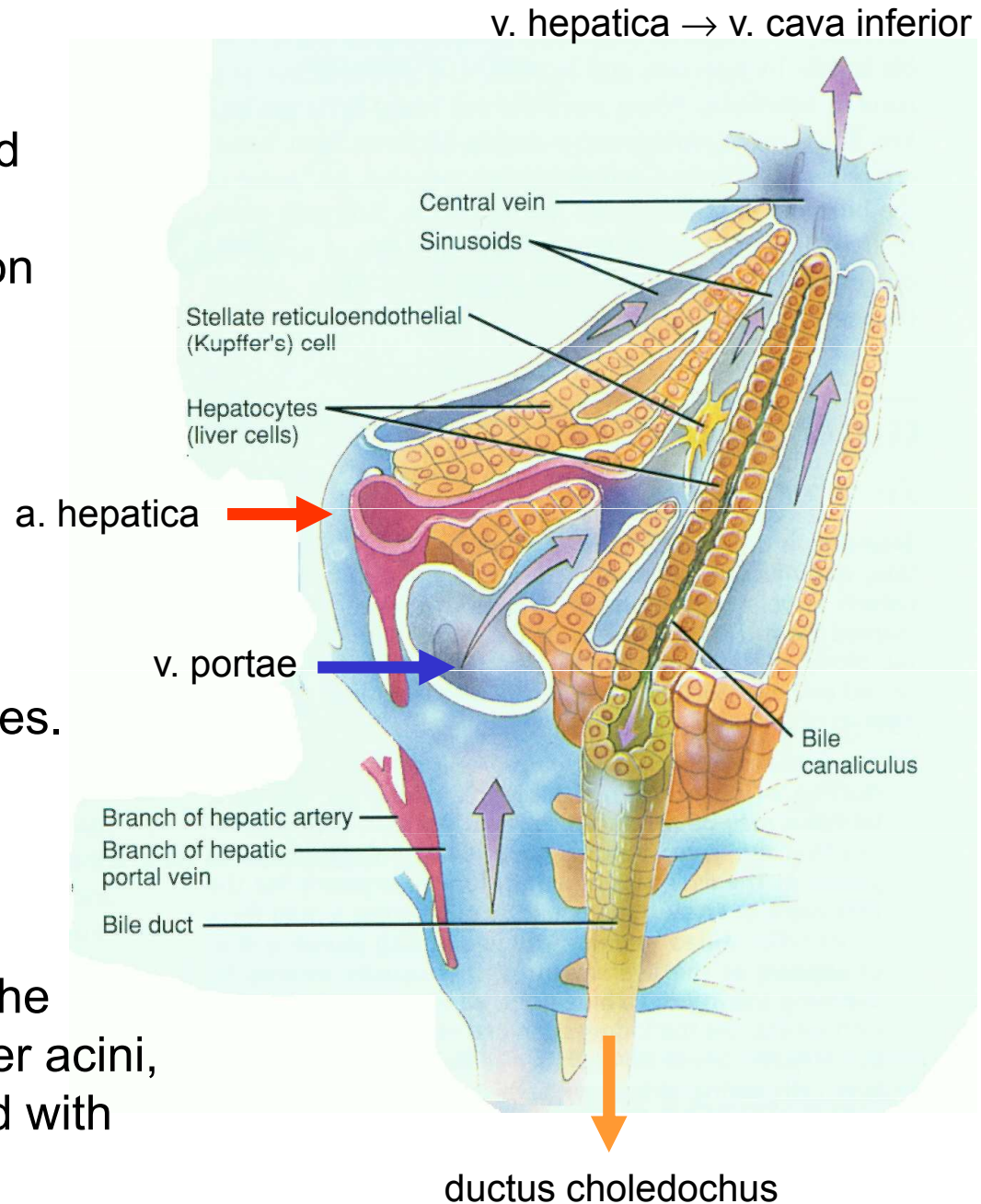
Iron and copper metabolism

- Synthesis of transferrin, coeruloplasmin, ferritin stores, excretion of copper.

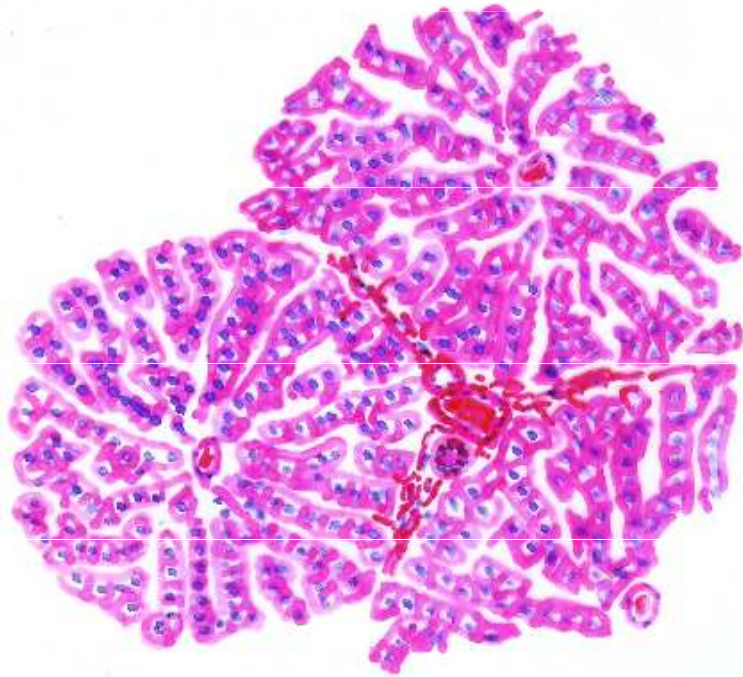
The liver receives venous blood from the intestine. Thus all the products of digestion, in addition to ingested drugs and other xenobiotics, perfuse the liver before entering the systemic circulation.

The mixed portal and arterial blood flows through sinusoids between columns of hepatocytes.

Hepatocytes are differentiated in their functions according to the decreasing pO_2 . In a simple liver acini, there are three zones equipped with different enzymes.



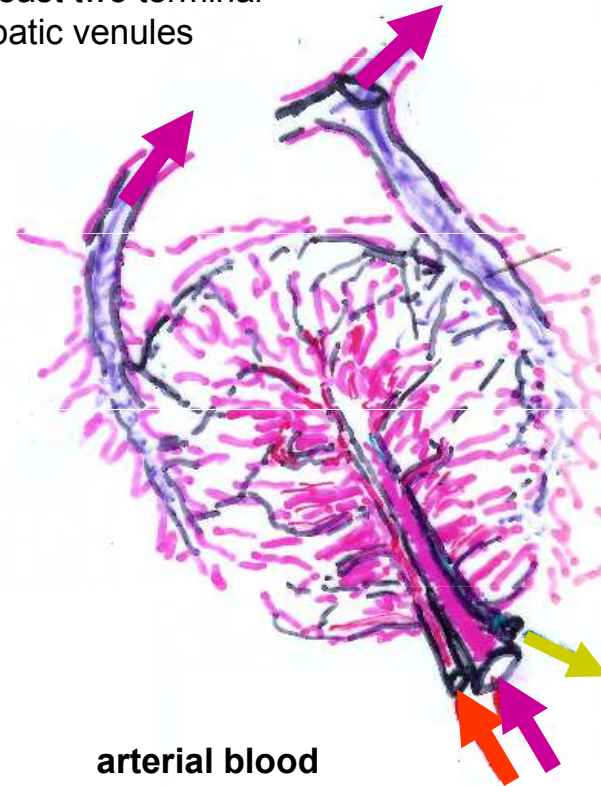
Hexagonal hepatic lobules
round terminal hepatic venules
are not functional units



simple liver ACINUS – a functional unit

efferent vessels

at least two terminal
hepatic venules



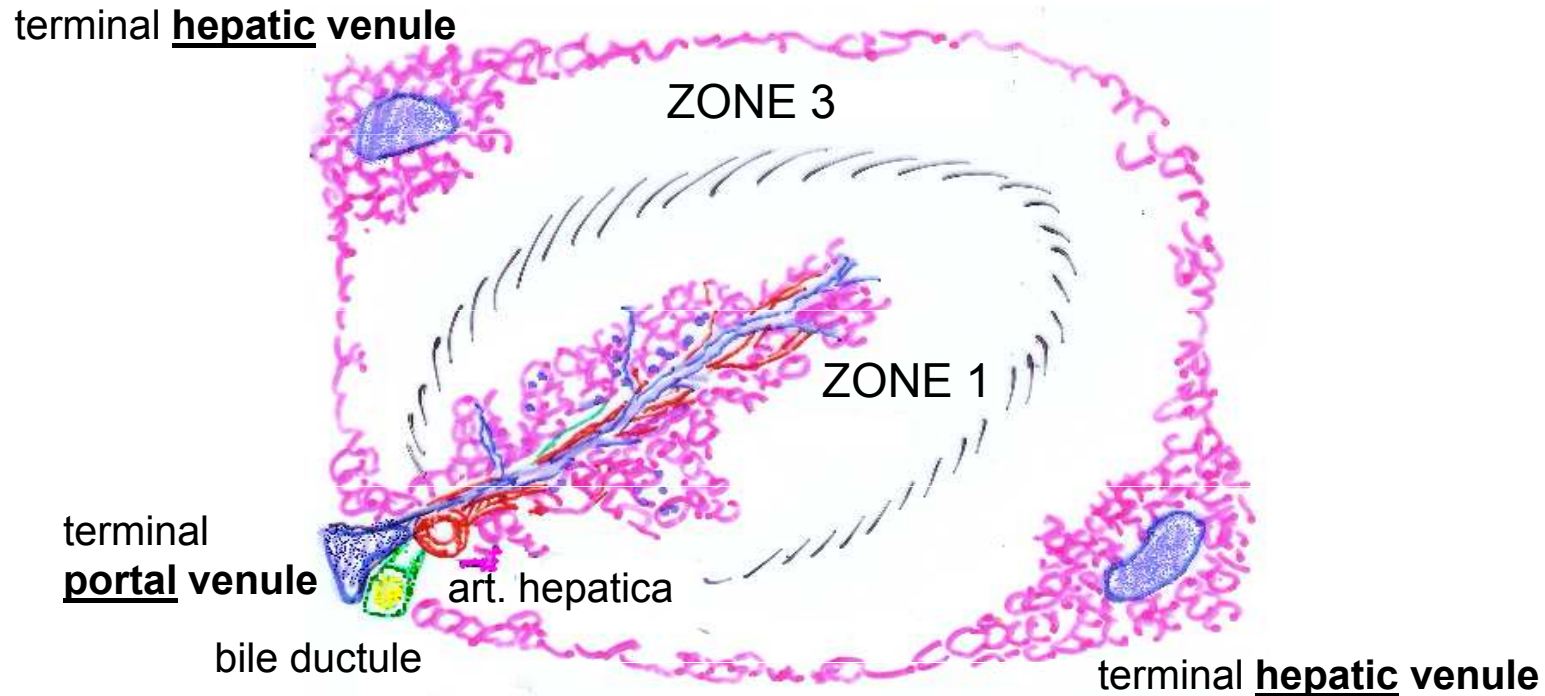
arterial blood
terminal branches
aa. hepaticae

bile ductules

portal (venous) blood
from intestine, pancreas,
and spleen

portal field with afferent vessels

Metabolic areas in the acini



Zone 1 – periportal area

high pO_2

cytogenesis, mitosis

numerous mitochondria

glycogenesis and glycogenolysis

proteosynthesis

ureosynthesis

Zone 3 – microcirculatory periphery

low pO_2

high activity of ER (cyt P450, detoxification)

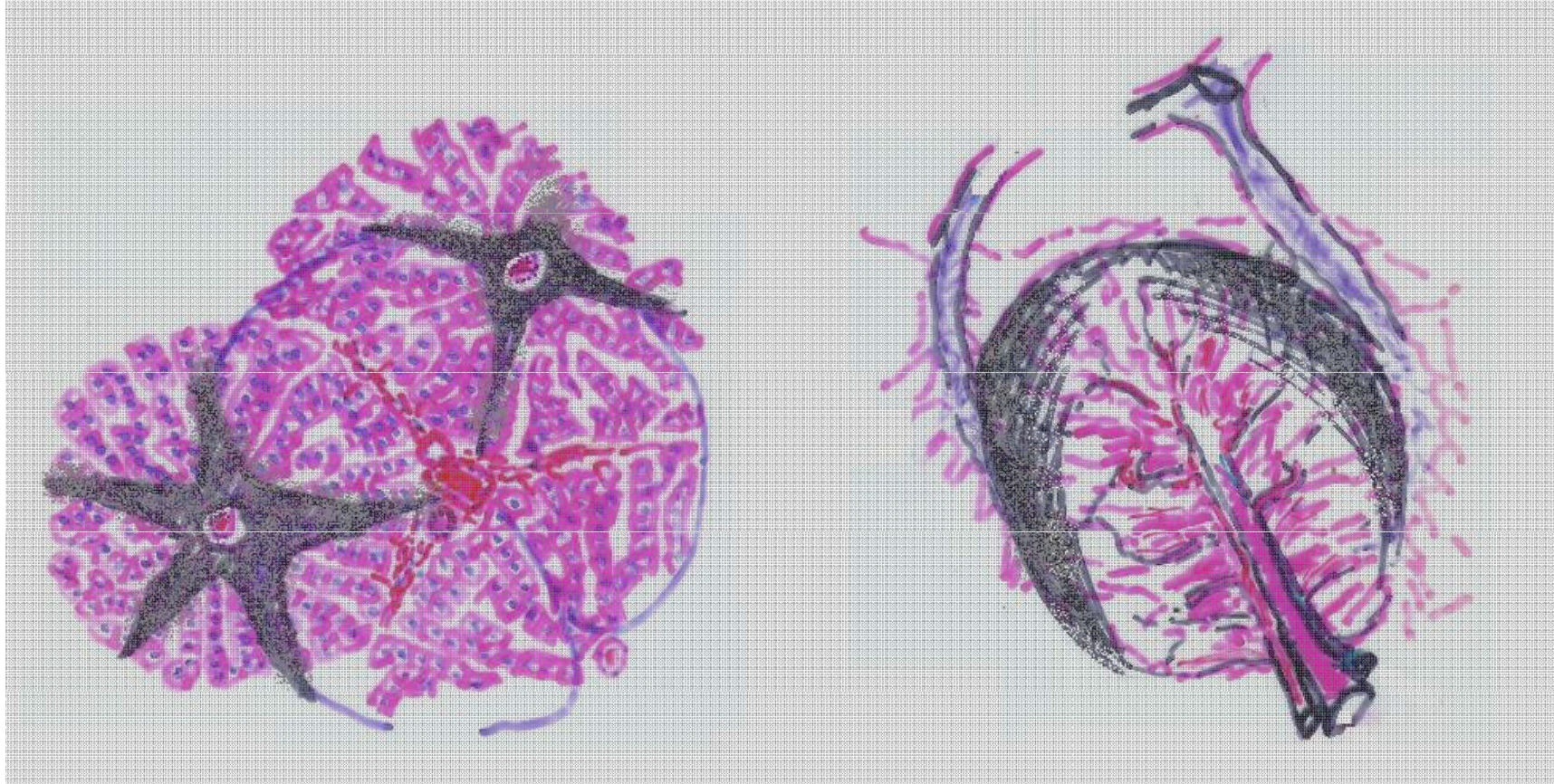
pentose phosphate pathway

hydrolytic enzymes

glycogen stores, fat and pigment stores

glutamine synthesis

Liver of a patient who died in hepatic coma:.



Seastar-shaped necrotic lesion around the terminal hepatic venule. This shape is produced by necrosis creeping along zones 3 of the simple acini, intercalating between them and reaching portal spaces.

Liver – production of bile

Composition of bile	Mass concentration / g/l	
	Hepatic bile	Gall-bladder bile
Inorganic salts	8.4	6.5
Bile acids	7 – 14	32 – 115
Cholesterol	0.8 – 2.1	3.1 – 16.2
Bilirubin glucosiduronates	0.3 – 0.6	1.4
Phospholipids	2.6 – 9.2	5.9
Proteins	1.4 – 2.7	4.5
pH	7.1 – 7.3	6.9 – 7.7

Functions

The **bile acids** emulsify lipids and fat-soluble vitamins in the intestine. High concentrations of bile acids **and phospholipids** stabilize micellar dispersion of cholesterol in the bile (crystallization of cholesterol → cholesterol gall-stones).

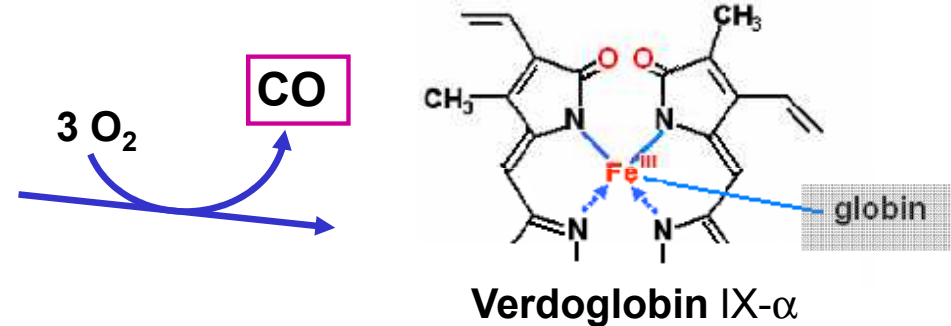
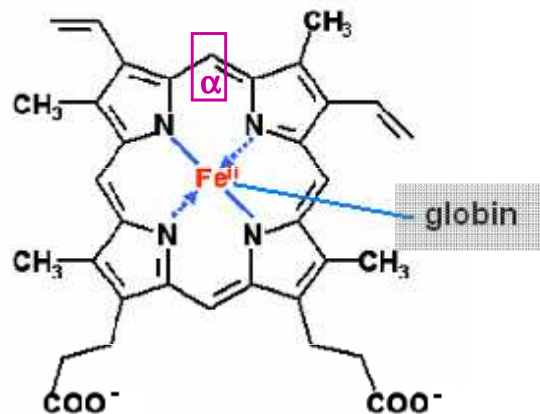
Excretion of cholesterol and bile acids is the major way of removing cholesterol from the body. Bile also removes hydrophobic metabolites, drugs, toxins and metals (e.g. copper, zinc, mercury).

Neutralization of the acid chyme in conjunction with HCO_3^- from pancreatic secretion.

Degradation of haemoglobin to bilirubin– bile pigments

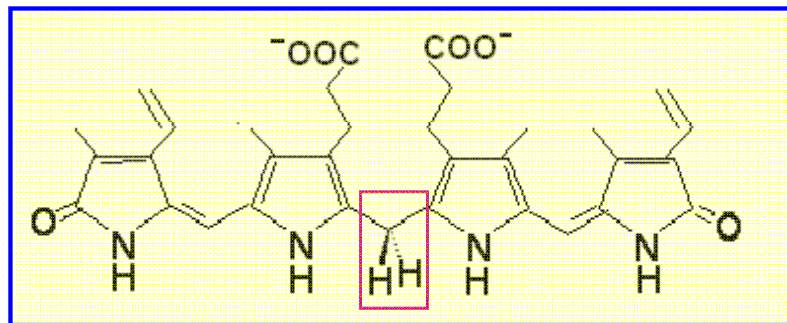
Erythrocytes are taken up by the **reticuloendothelial cells** (cells of the spleen, bone marrow, and Kupffer cells in the liver) by phagocytosis.

Haemoglobin

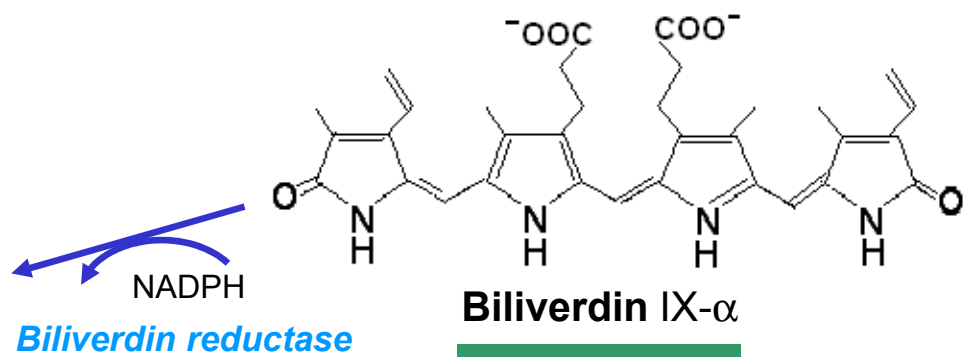


Haem oxygenase
(*NADPH:cyt P450 oxidoreductase*)

Fe^{3+}
globin



Bilirubin IX-α

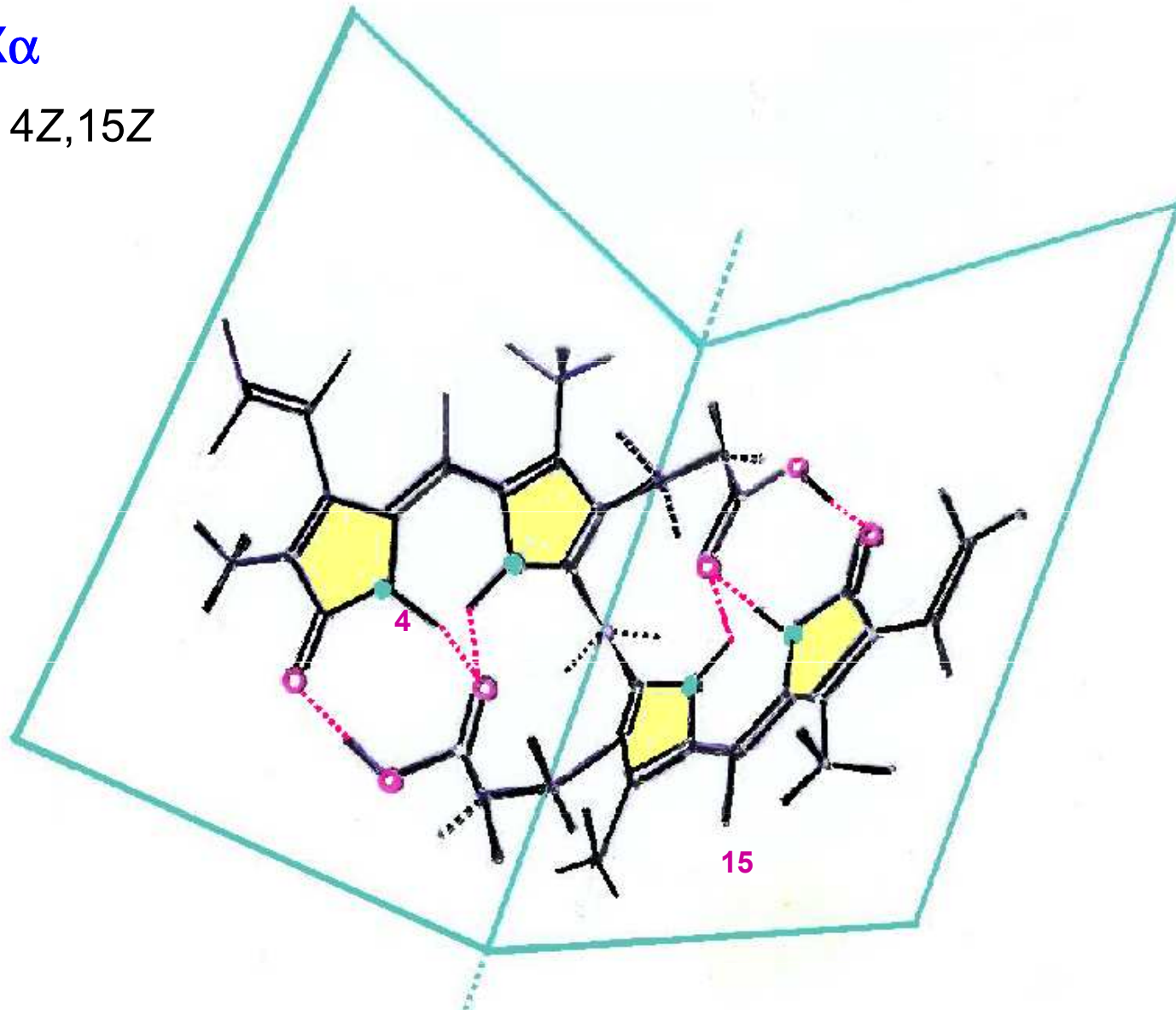


Biliverdin reductase

In blood plasma, **hydrophobic bilirubin molecules** (called **unconjugated bilirubin**) are transported in the form of **complexes bilirubin-albumin**.

Bilirubin IX α

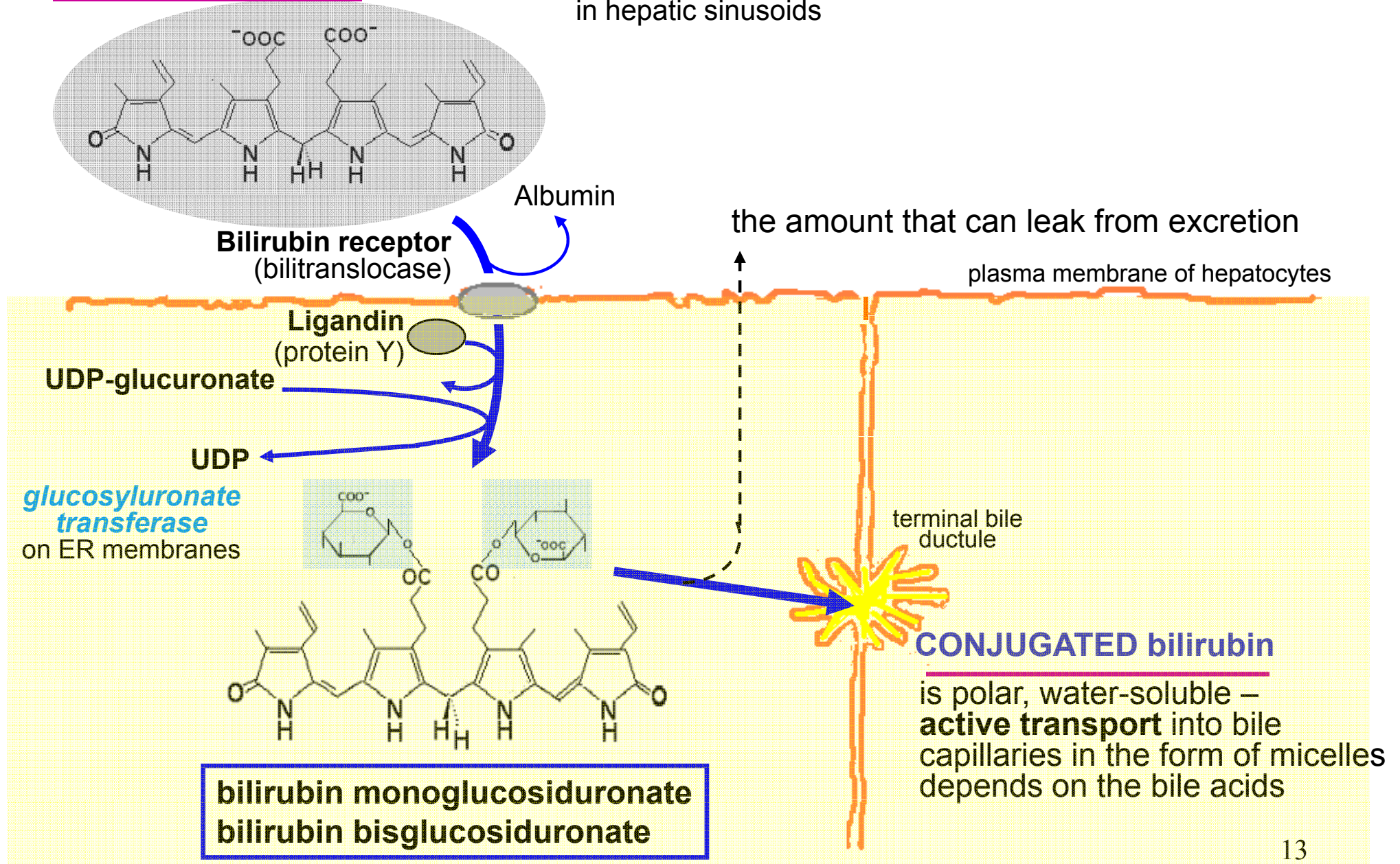
configuration 4Z,15Z



Polarity of the two carboxyl groups of unconjugated bilirubin is masked by formation of hydrogen bonds between the carboxyl groups and the electronegative atoms within the opposite halves of bilirubin molecules.

The hepatic uptake, conjugation, and excretion of bilirubin

UNCONJUGATED bilirubin (bilirubin-albumin complex) in hepatic sinusoids



The formation of urobilinoids by the intestinal microflora

Conjugated bilirubin

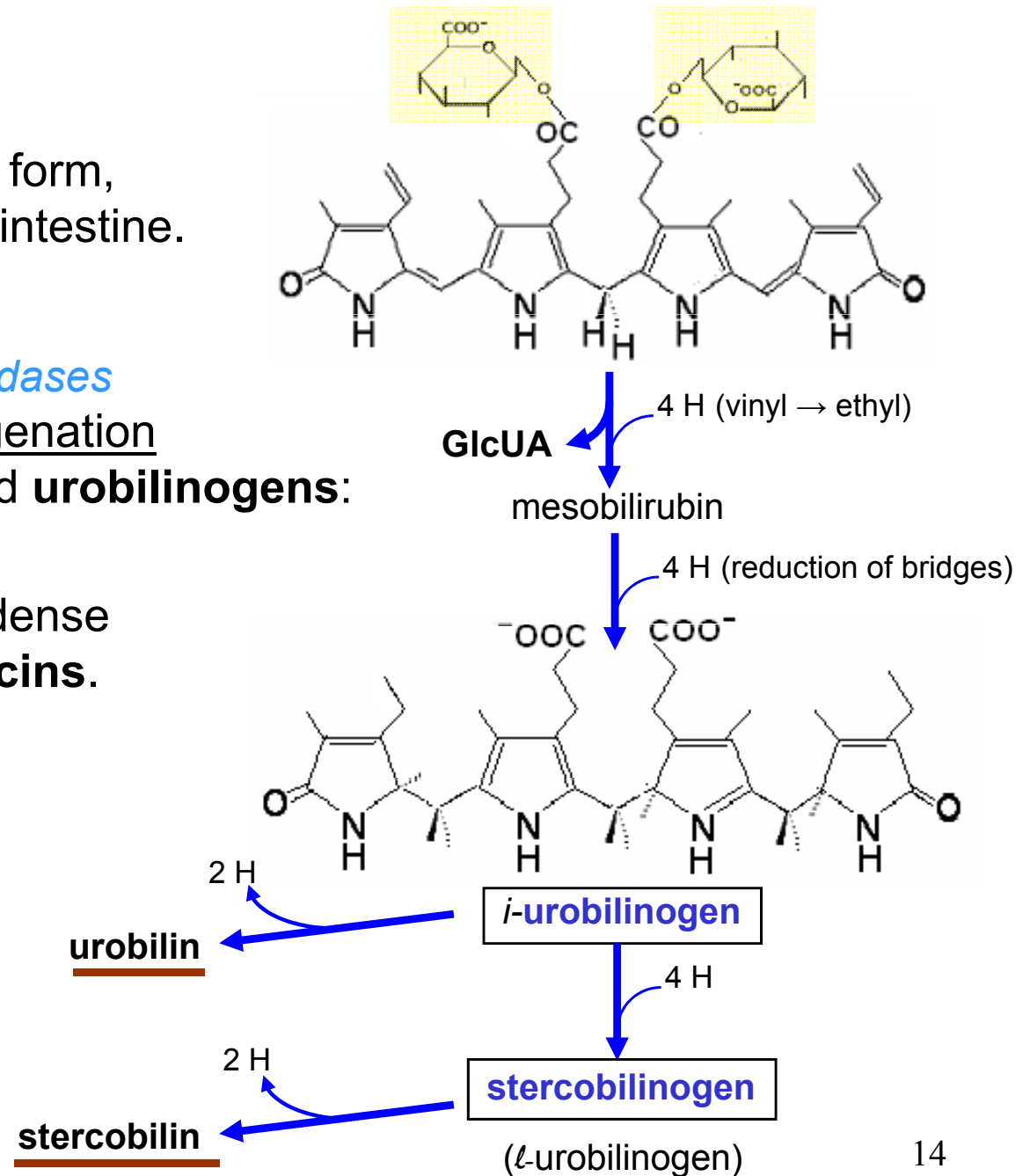
is secreted into the bile. As far as bilirubin remains in the conjugated form, it cannot be absorbed in the small intestine.

In the large intestine, bacterial *reductases* and *β -glucuronidases* catalyze deconjugation and hydrogenation of free bilirubin to mesobilirubin and **urobilinogens**:

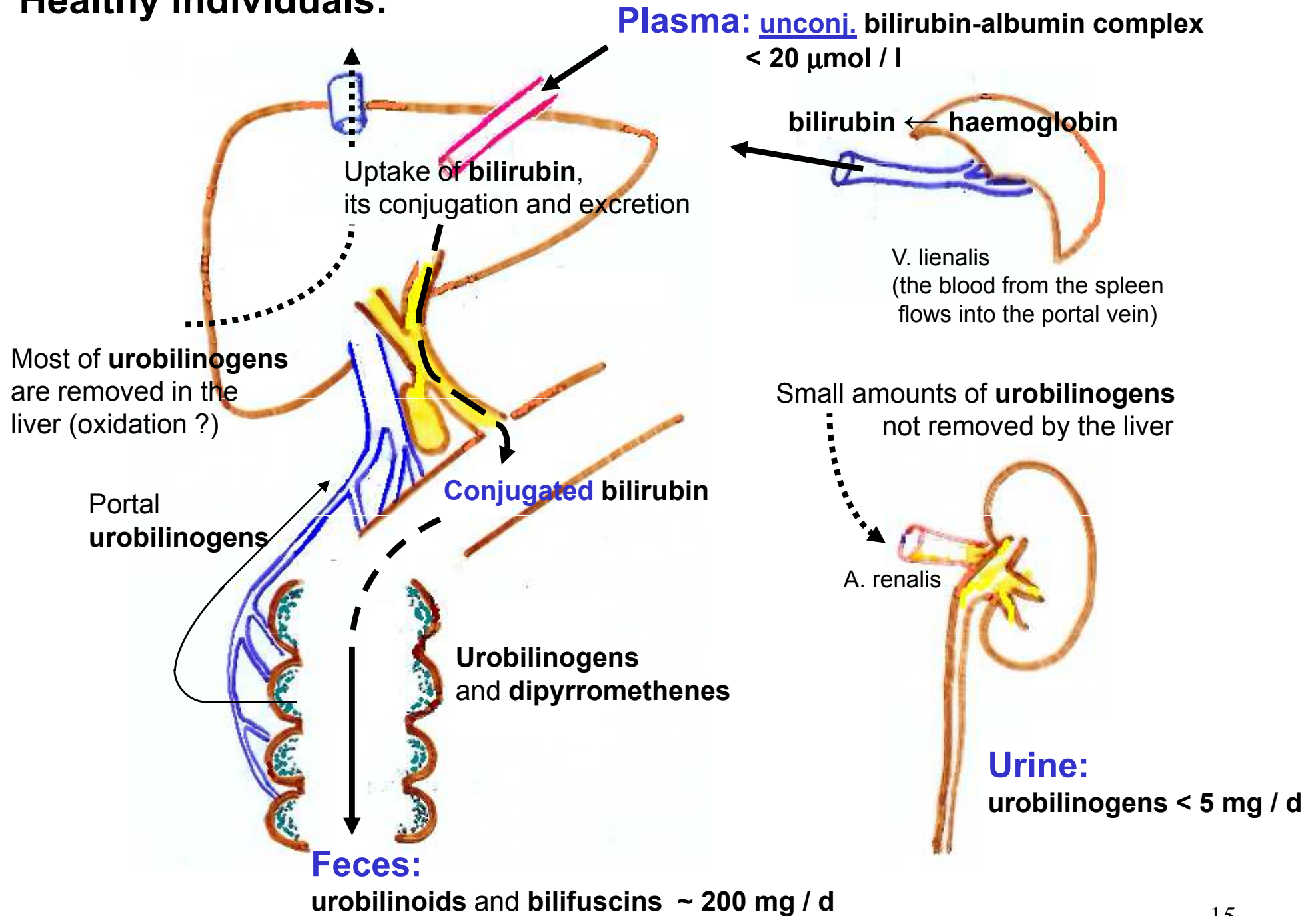
A part of urobilinogens is split to **dipyrromethenes**, which can condense to give intensively coloured **bilifuscins**.

Urobilinogens are partly

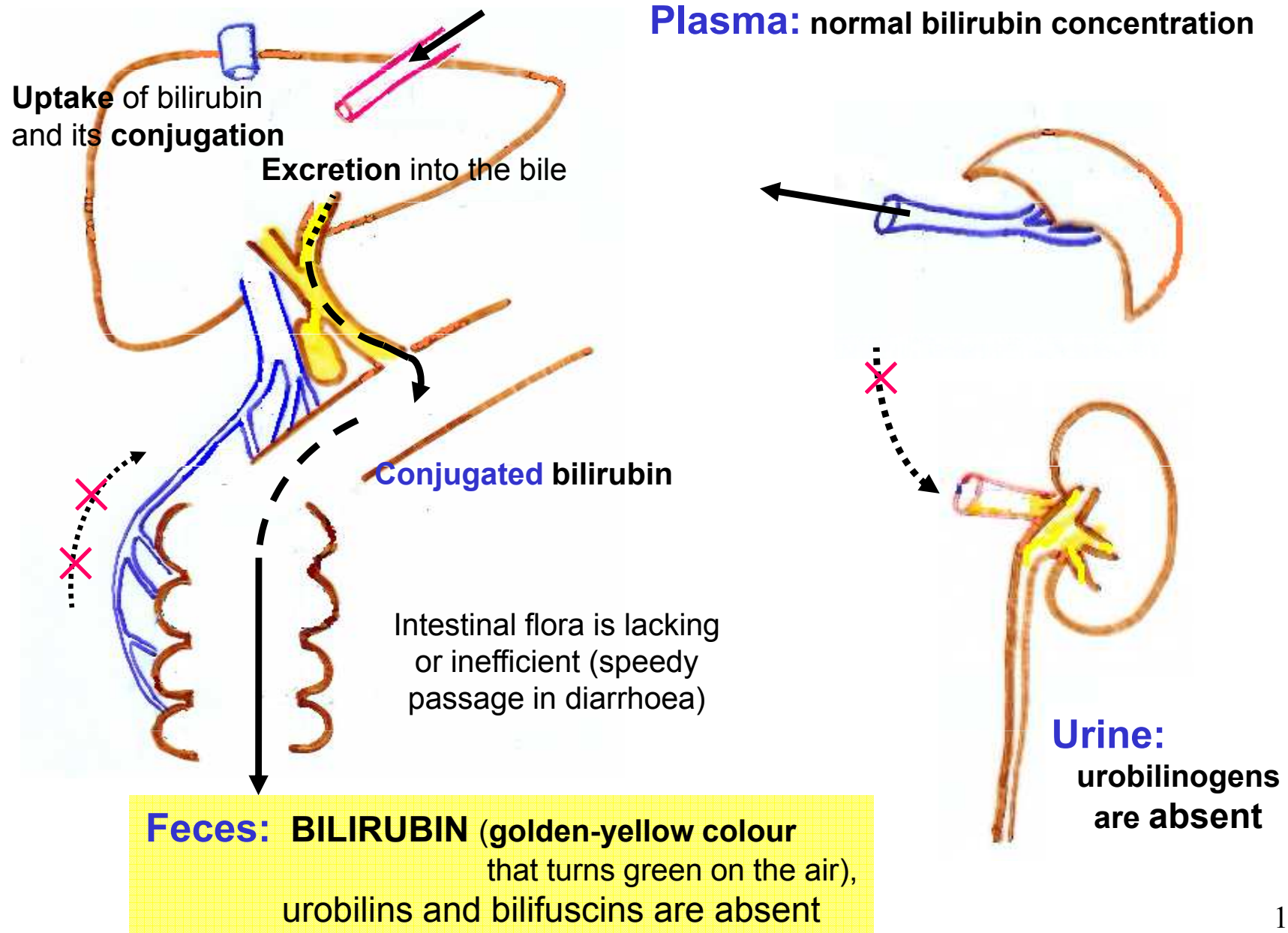
- absorbed (mostly removed by the liver), a small part appears in the urine,
- partly excreted in the feces; on the air, they are oxidized to dark brown faecal **urobilins**.



Healthy individuals:



In the **absence of intestinal microflora** (before colonization in newborns or during treatment with broad-spectrum antibiotics):



Major types of hyperbilirubinaemias

Hyperbilirubinaemia – serum bilirubin $> 20 - 22 \mu\text{mol} / \text{l}$

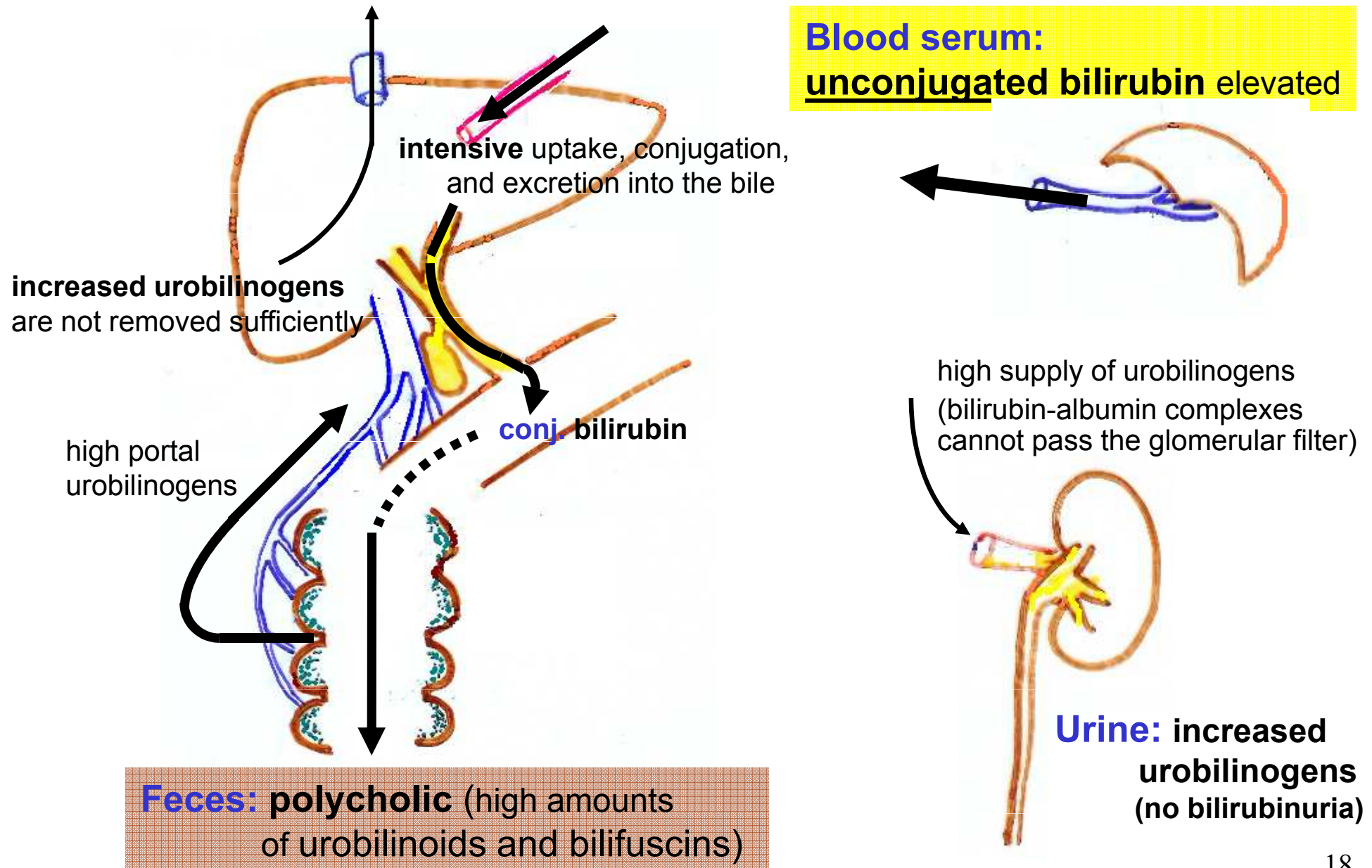
Icterus (jaundice) – yellowish colouring of scleras and skin,
serum bilirubin usually more than $30 - 35 \mu\text{mol} / \text{l}$

The causes of hyperbilirubinaemia are conventionally classified as

- **prehepatic (haemolytic)** – increased production of bilirubin,
- **hepatocellular** due to inflammatory disease (infectious hepatitis), hepatotoxic compounds (e.g. ethanol, acetaminophen), or autoimmune disease; chronic hepatitis can result in liver cirrhosis – fibrosis of hepatic lobules,
- **posthepatic (obstructive)** – insufficient drainage of intrahepatic or extrahepatic bile ducts (cholestasis).

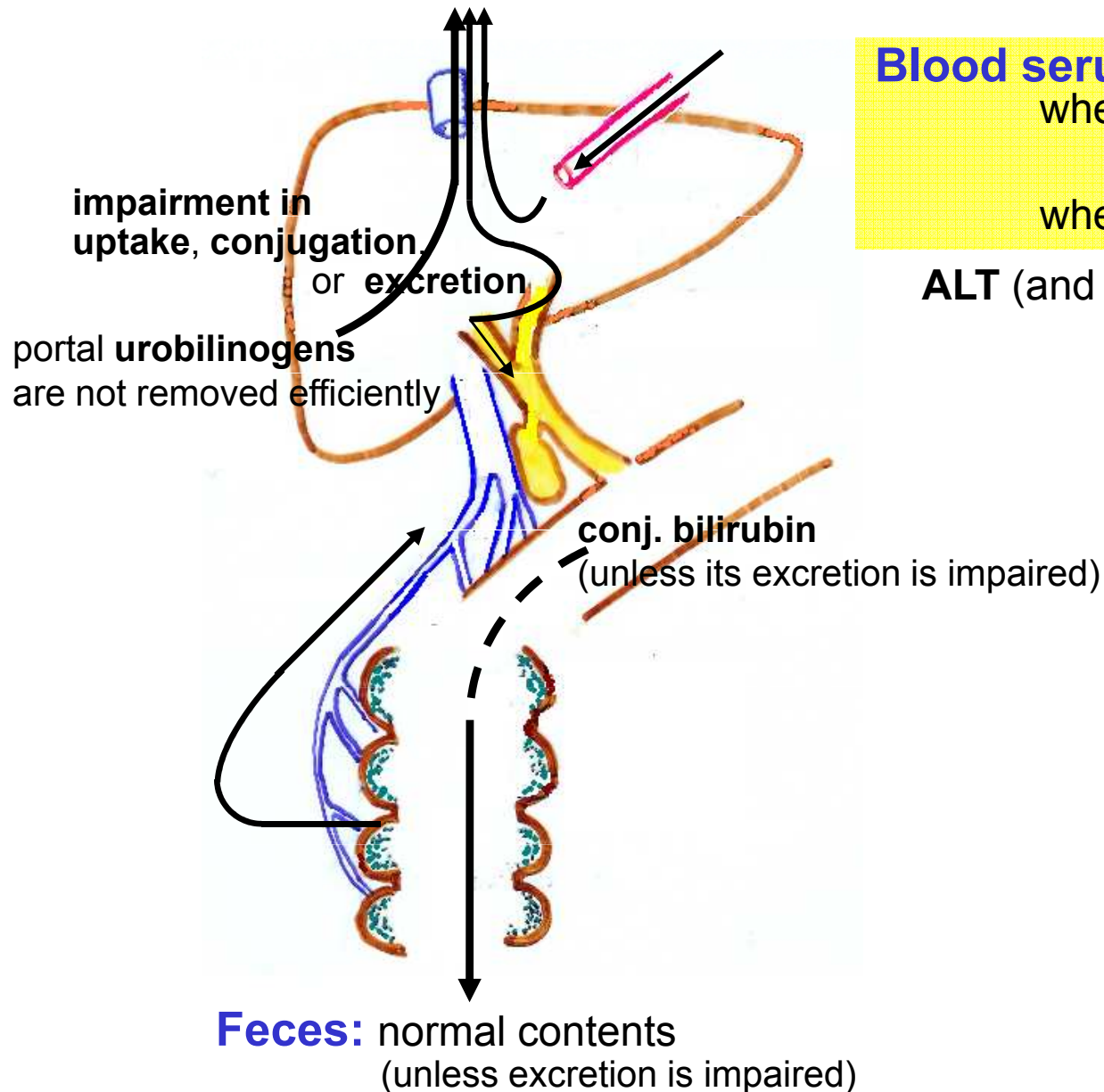
Prehepatic (haemolytic) hyperbilirubinaemia

– excessive erythrocyte breakdown



Hepatocellular hyperbilirubinaemia

The results of biochemical test depend on whether an impairment of hepatic **uptake**, **conjugation**, or **excretion** of bilirubin predominates.



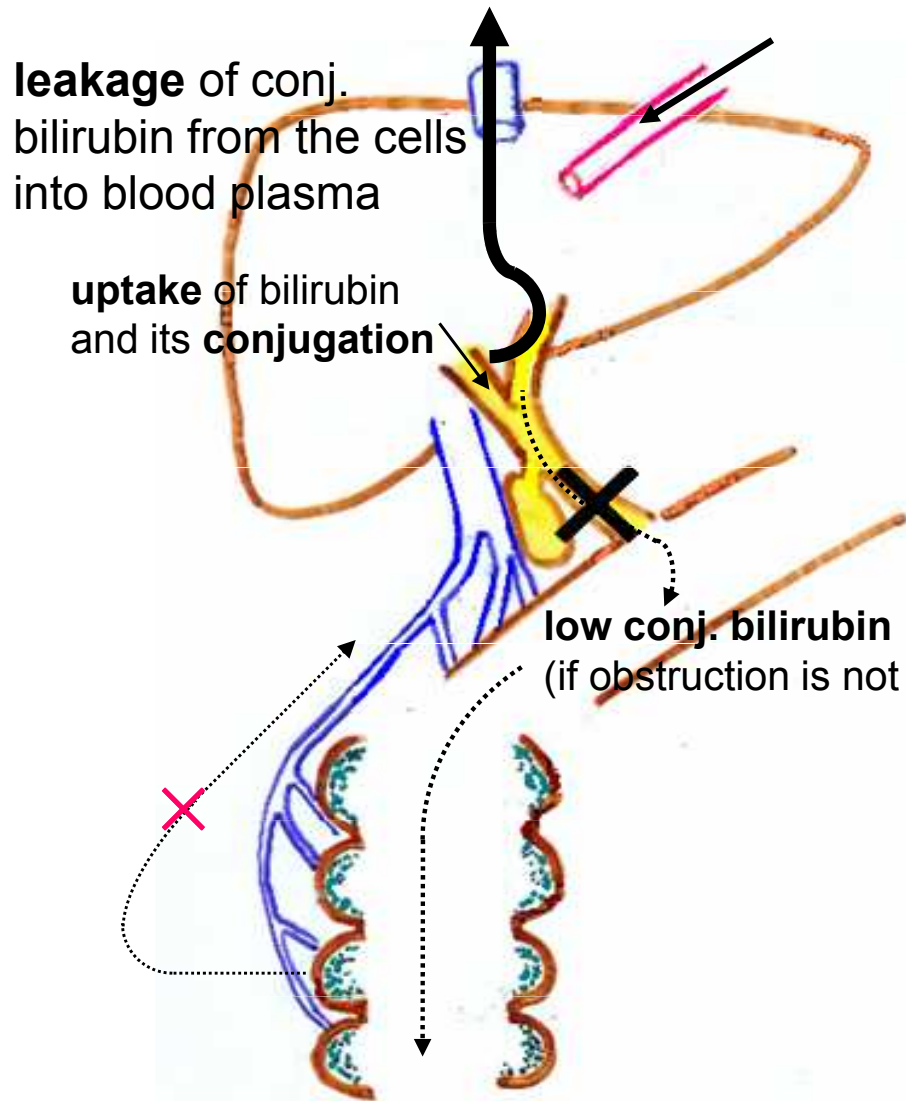
Blood serum: unconj. bilirubin is elevated, when its uptake or conjugation is impaired
conj. bilirubin is elevated, when its excretion or drainage is impaired

ALT (and AST) catalytic concentrations increased

urobilinogens and conjugated bilirubin pass into the urine (not unconj. bilirubin-albumin complexes)

Urine: increased urobilinogens (unless bilirubin excretion is impaired)
bilirubinuria (when plasma conj. bilirubin increases)

Obstructive (posthepatic) hyperbilirubinaemia



leakage of conj. bilirubin from the cells into blood plasma

uptake of bilirubin and its conjugation

low conj. bilirubin (if obstruction is not complete)

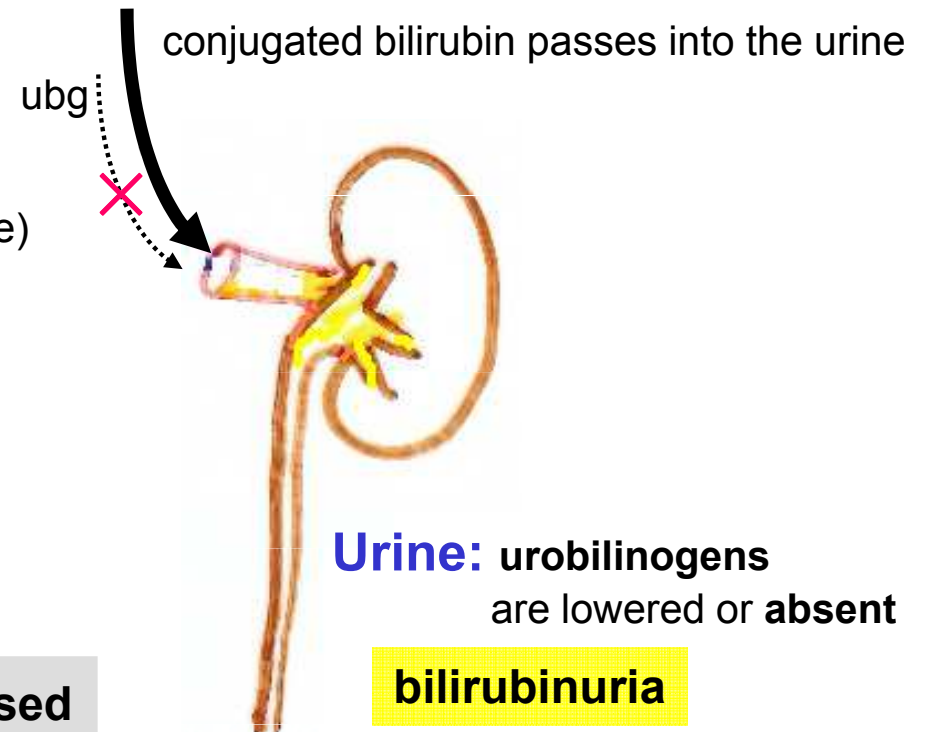
Feces: urobilinoids and bilifuscins decreased or absent (grey, acholic feces)

Blood serum:

conjugated bilirubin elevated

bile acids concentration increased

catalytic concentration of **ALP** increased



Urine: urobilinogens are lowered or absent

bilirubinuria

Summary:

Type	Bilirubin			Urobilinogens	
	Blood serum	Urine	(derivatives) Feces	Blood	Urine
PREHEPATIC (or haemolytic)	increased (unconjugated)	absent	increased	increased	increased
HEPATOCELLULAR	increased (unconj./conj.)	present (unconj.)	normal to decreased	increased or decreased	increased or decreased
OBSTRUCTIVE (posthepatic)	increased (conjugated)	present (unconj.)	decreased or absent	decreased or absent	decreased or absent

Laboratory tests

for detecting an impairment of liver functions ("liver tests")

- **Plasma markers of hepatocyte membrane integrity**

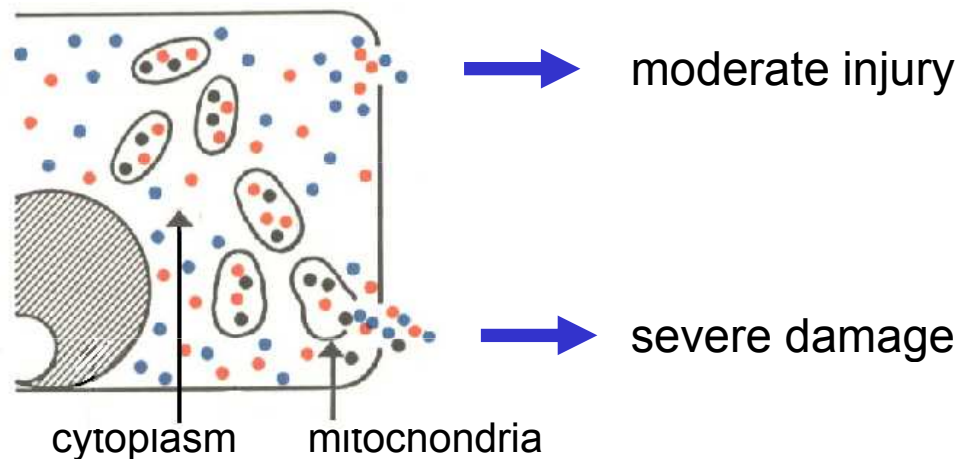
Catalytic concentrations of intracellular enzymes in blood serum increase:

An assay for alanine aminotransferase (ALT) activity is the most sensitive one.

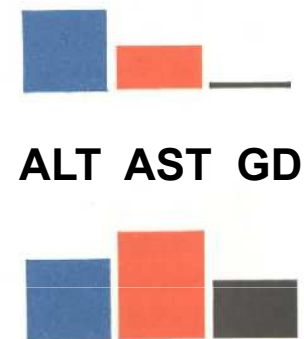
In severe impairments, the activities of

aspartate aminotransferase (AST) and

glutamate dehydrogenase (GD) also increase.



Increase of catalytic concentrations:



- **Tests for decrease in liver proteosynthesis**

Serum concentration of albumin (biological half-time about 20 days), transthyretin (prealbumin, biological half-time 2 days) and transferrin, blood coagulation factors (prothrombin time increases), activity of serum non-specific **choline esterase (ChE)**.

- **Tests for the excretory function and cholestasis**

Serum bilirubin concentration

Serum catalytic concentration of alkaline phosphatase (ALP) and γ -glutamyl transpeptidase (γ GT)

Test for urobilinogens and bilirubin in urine

Estimation of the excretion rate of **bromosulphophthalein (BSP test)** is applied to convalescents after acute liver diseases.

- **Tests of major metabolic functions** are not very decisive:

Saccharide metabolism low glucose tolerance (in oGT test)

Lipid metabolism increase in VLDL (triacylglycerols) and LDL (cholesterol)

Protein catabolism decreased urea, ammonium increase
(in the final stage of liver failure, hepatic coma)

- **Special tests to specific disorders:** serological tests to viral hepatitis, serum α -foetoprotein (liver carcinoma), porphyrins in porphyrias, etc.

Metabolism of iron

The body contains 4 – 4.5 g Fe:

In the form of haemoglobin 2.5 – 3.0 g Fe,
tissue ferritin stores up to 1.0 g Fe in men (0.3 – 0.5 g in women),
myoglobin and other haemoproteins 0.3 g Fe,
circulating transferrin 3 – 4 mg Fe.

The **daily supply** of iron in mixed diet is about **10 – 20 mg**.
From that amount, not more than **only 1 – 2 mg are absorbed**.
Iron metabolism is regulated by control of uptake, which have to
replace the daily loss in iron and prevent an uptake of excess iron.

A healthy adult individual loses on average 1 – 2 mg Fe per day in desquamated cells (intestinal mucosa, epidermis) or blood (small bleeding, so that women are more at risk because of net iron loss in menstruation and pregnancy).

There is no natural mechanism for eliminating excess iron from the body.

10 – 20 mg Fe



8 – 19 mg Fe



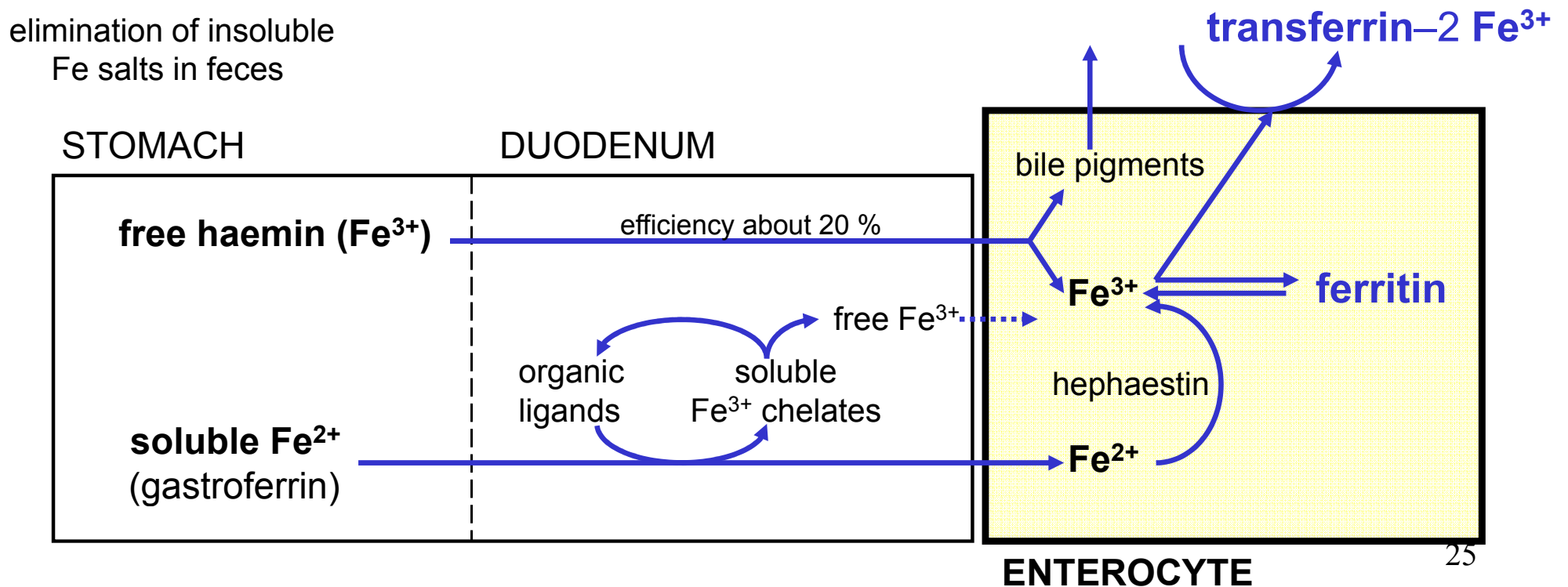
elimination of insoluble Fe salts in feces

Absorption of iron in duodenum and jejunum

Phosphates, oxalate, and phytate (*myo*-inositol hexakis(dihydrogen phosphate), present in vegetable food) form insoluble Fe^{3+} complexes and disable absorption.

Fe^{2+} is absorbed much easier than Fe^{3+} . Reductants such as **ascorbate** or **fructose** promote absorption, as well as Cu^{2+} .

Gastroferrin, a component of gastric secretion, is a glycoprotein that binds Fe^{2+} maintaining it soluble and prevents its oxidation to Fe^{3+} , from which insoluble iron salts are formed.



Transferrin (Trf)

is a plasma glycoprotein (a major component of β_1 -globulin fraction), M_r 79 600.

Plasma (serum) **transferrin concentration 2.5 – 4 g / l** (30 – 50 $\mu\text{mol} / \text{l}$)

Transferrin molecules have two binding sites for Fe ions,
total iron binding capacity (TIBC) for Fe ions is higher than **60 $\mu\text{mol} / \text{l}$** .

Serum Fe^{3+} (i.e. transferrin- Fe^{3+}) concentration is about **10 – 20 $\mu\text{mol} / \text{l}$** ,
14 – 26 $\mu\text{mol} / \text{l}$ in men,
11 – 22 $\mu\text{mol} / \text{l}$ in women.

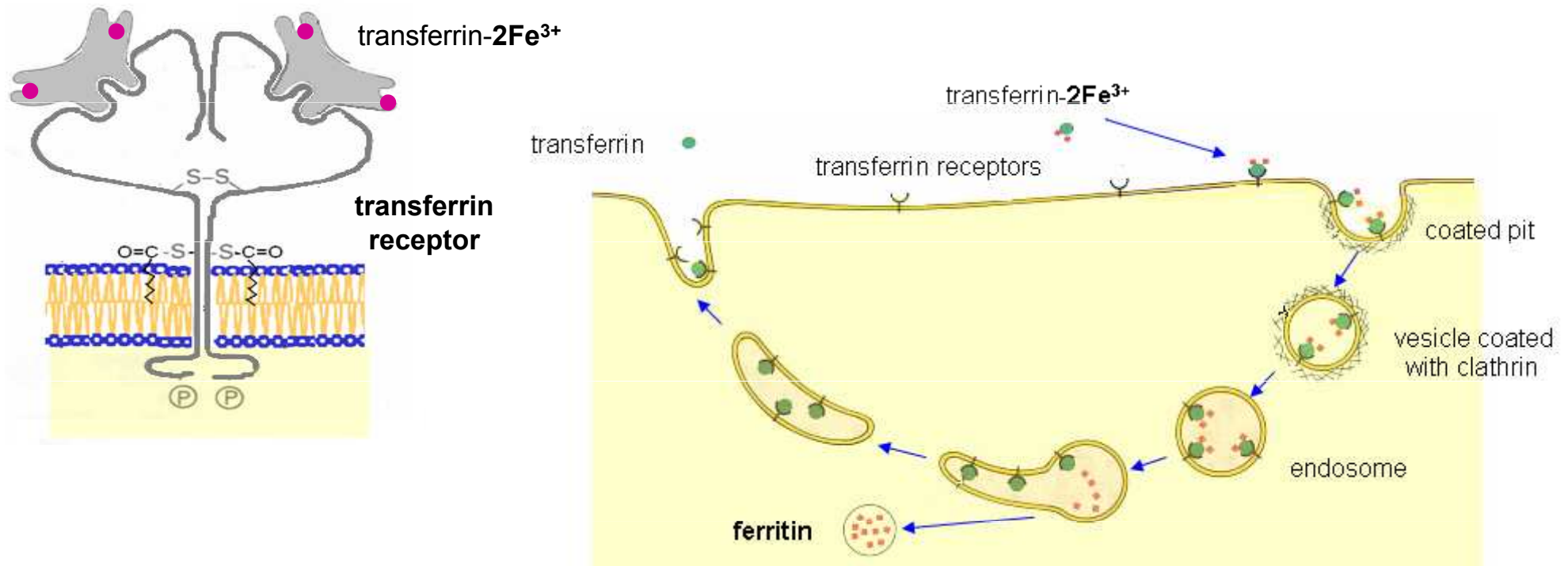
Circadian rhythm exists, the morning concentrations are higher by 10 - 30 % than those at night..

Saturation of transferrin with Fe^{3+} equals usually **about 1/3**.

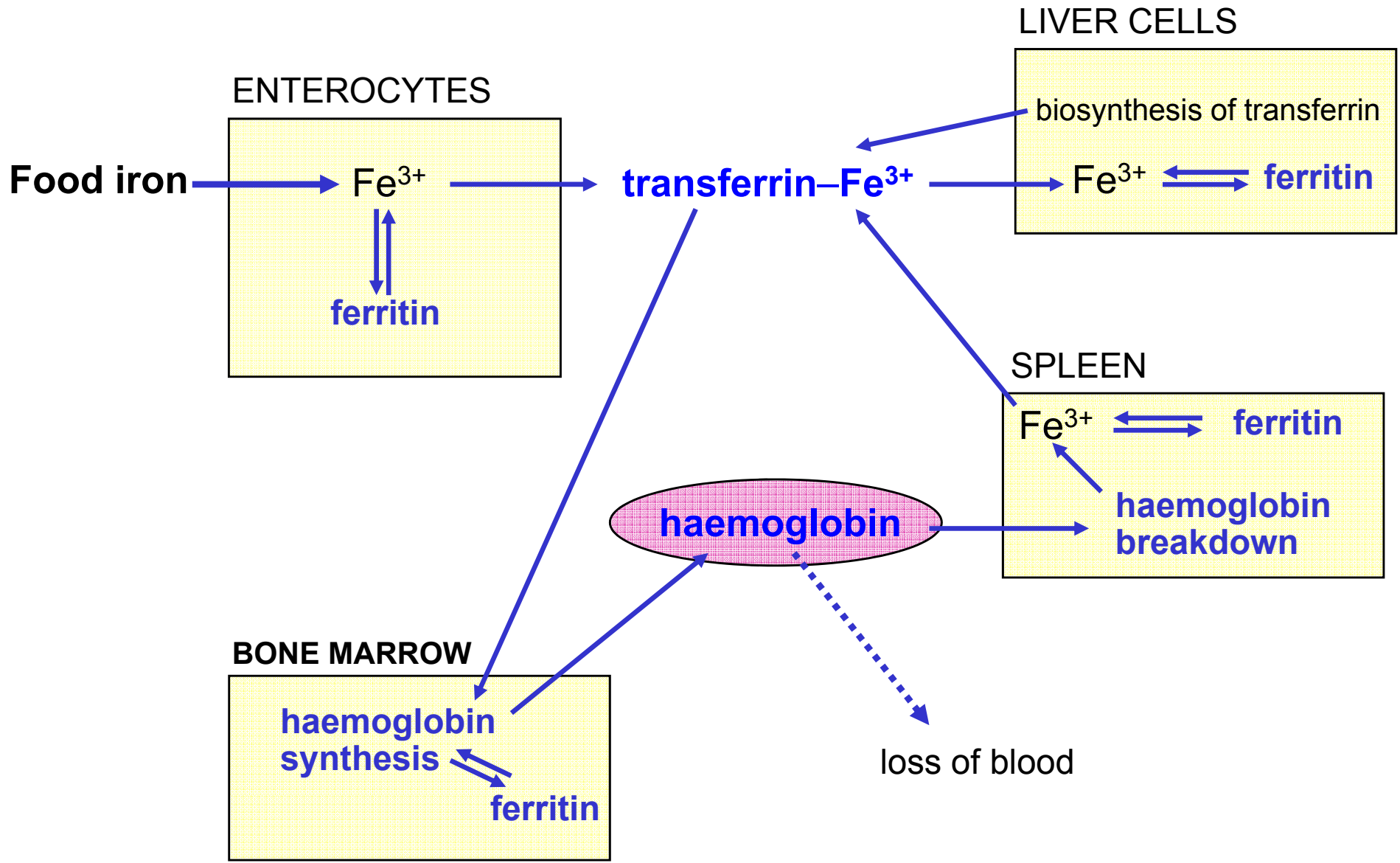
Because the biosynthesis of transferrin is stimulated during iron deficiency (and plasma iron concentration decreases), the decrease in saturation of transferrin is observed.

Iron is taken up by the cells

through a **specific receptor-mediated endocytosis**.



Some receptors are released from the plasmatic membranes. Increase in serum concentration of those **soluble transferrin receptors** is the earliest marker of iron deficiency.



Ferritin

Ferritin occurs in most tissues (especially in the liver, spleen, bone-marrow, and enterocytes).

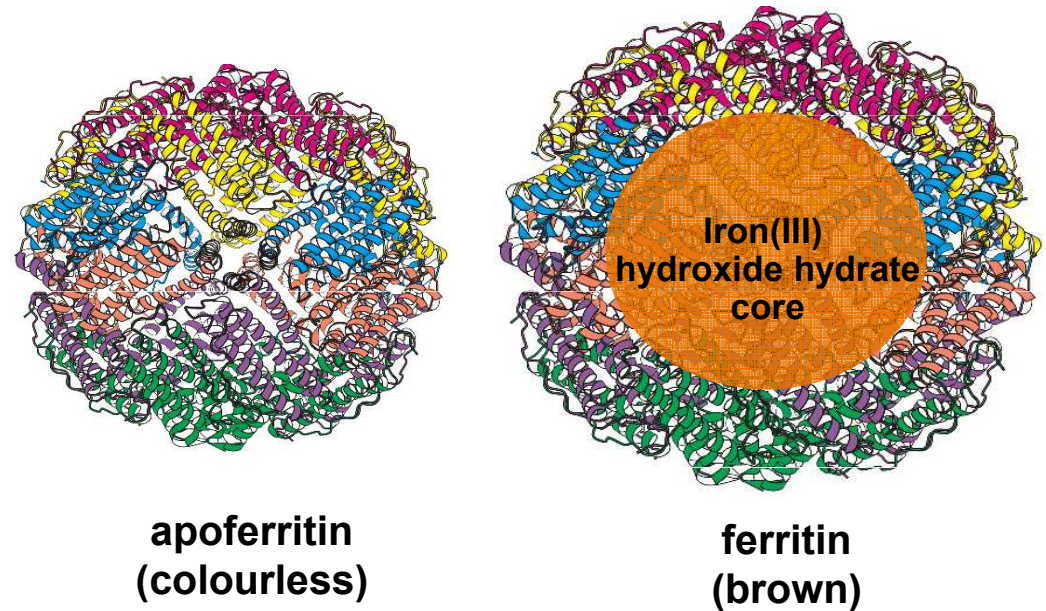
The protein apoferritin is a ball-shaped homopolymer of 24 subunits that surrounds the core of hydrated iron(III) hydroxide.

One molecule can bind few **thousands of Fe^{3+} ions**, which make up to 23 % of the weight of ferritin.

Minute amounts of ferritin are released into the blood plasma from the extinct cells. **Plasma ferritin concentration** 25 – 300 $\mu\text{g/l}$ is proportional to the ferritin stored in the cells, unless the liver is impaired (increased ferritin release from the hepatocytes).

If the loading of ferritin is excessive, ferritin aggregate into its degraded form, **haemosiderin**, in which the mass fraction of Fe^{3+} can reach 35 %.

Ferritin was discovered by V. Laufberger, professor at Masaryk university, Brno, in 1934.



Hepcidin

is a polypeptide ($M_r \sim 2000$, 25 amino acid residues, from which 8 are Cys), discovered as the liver-expressed antimicrobial peptide, LEAP-1, in 2000.

It is produced by the liver (to some extent in myocard and pancreas, too) as a **hormone that limits the accessibility of iron** and also exhibits certain antimicrobial and antifungal activity.

The **biosynthesis of hepcidin is stimulated in iron overload and in inflammations** (hepcidin belongs to acute phase proteins type 2), and is suppressed during iron deficiency .

Notice the fact that the same two factors stimulating hepcidin synthesis inhibit the biosynthesis of transferrin.

Effects of hepcidin: It – reduces Fe^{2+} absorption in the duodenum,
– prevents the release of recyclable Fe from macrophages,
– inhibits Fe transport across the placenta,
– diminishes the accessibility of Fe for invading pathogens.

Hepcidin is filtered in renal glomeruli and not reabsorbed in the renal tubules. So the amount of hepcidin excreted into the urine corresponds with the amount synthesized in the body. There is a positive correlation between this amount of hepcidin and the concentration of ferritin in blood plasma.