Metabolism of amino acids

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Intermediates of amino acid catabolism

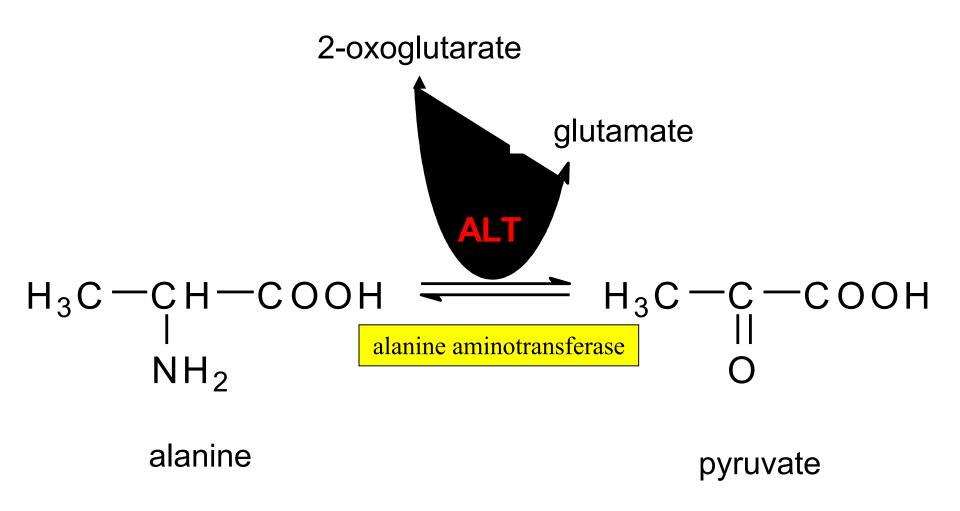
- Glucogenic (13) \rightarrow pyruvate and/or CAC intermediates
- **Ketogenic** (2) = Leu, Lys \rightarrow acetyl-CoA + acetoacetate
- Mixed (5) = Thr, Ile, Phe, Tyr, Trp

Intermediates of amino acid catabolism

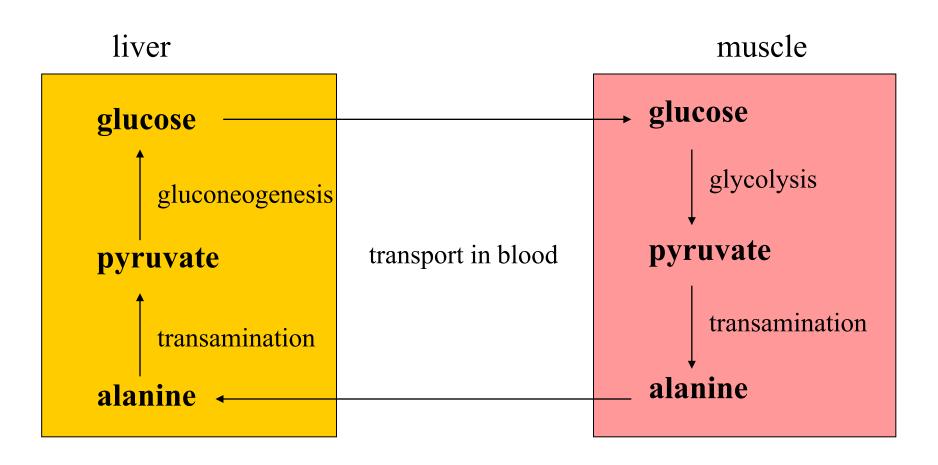
Ser, Gly, Thr, Ala, Cys, Trp pyruvate glucose lle, Leu, Lys, Thr acetoacetate acetyl-CoA Leu, Lys, Phe, Trp, Tyr Asp, Asn oxaloacetate 2-oxoglutarate Arg, Glu, Gln, His, Pro **fumarate** Phe, Tyr Asp succinyl-CoA lle, Val, Met, 7

Alanine

Transamination of alanine



Glucose-alanine cycle

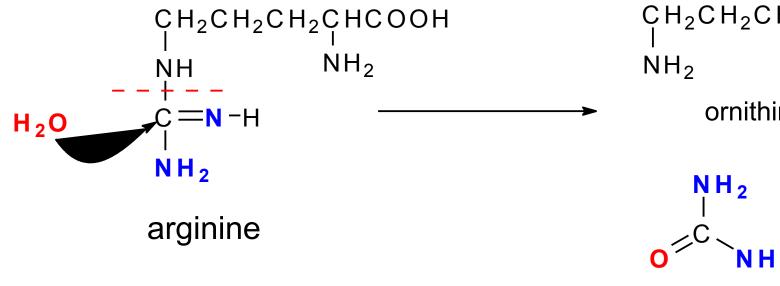


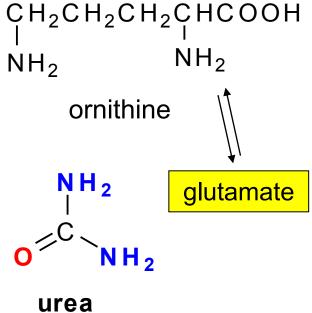
- alanine is non-toxic transport of ammonia from muscles to liver
- in the liver, alanine is the substrate for gluconeogenesis

Alanine - summary

- readily made from pyruvate (transamination)
- ALT is clinically important enzyme, mainly in liver, elevated catalytic concentration in blood serum liver diseases
- Ala is released to blood mainly from muscles, together with Gln (postresorption phase)
- semiessential AA (in metabolic stress) important substrate for gluconeogenesis

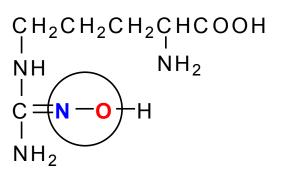
Hydrolysis of arginine → urea





NO is signal molecule from arginine

$$\begin{array}{cccc} \mathsf{CH_2CH_2CHCOOH} \\ \mathsf{NH} & \mathsf{NH_2} \\ \mathsf{C} = & \mathsf{N-H} & & & & & \\ \mathsf{O_2}, \, \mathsf{NADPH} \\ \mathsf{NH_2} & & & & & \\ \mathsf{BH_4} \end{array}$$



Exogenous NO sources

- glycerol trinitrate
- isosorbide dinitrate
- amyl nitrite
- isobutyl nitrite
- sodium nitroprusside

N = 0 • +
nitric oxide
radical

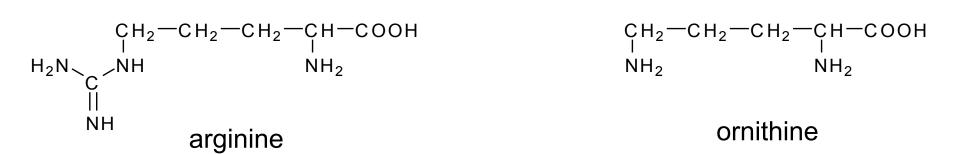
N-hydroxyarginine

$$\begin{array}{ccc} C \, H_2 C \, H_2 C \, H_2 C \, H C \, O \, O \, H \\ N \, H & N \, H_2 \\ C \, = \, O \\ I \\ N \, H_2 & citrulline \end{array}$$

Synthesis of creatine (1. part)

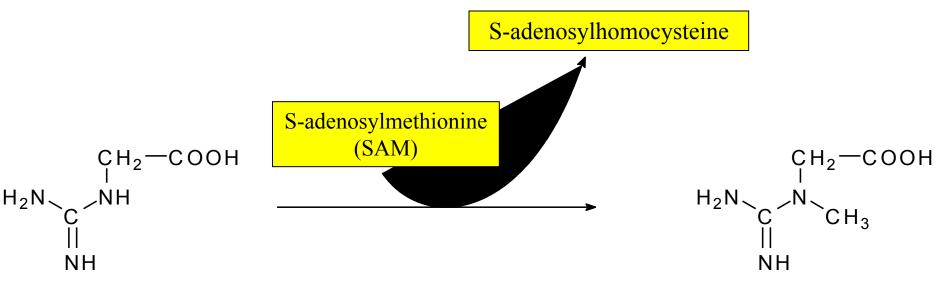
from Greek κρέας (meat)

$$\begin{array}{c} \mathsf{CH_2} - \mathsf{COOH} \\ \mathsf{NH_2} \\ \mathsf{glycine} \end{array} \qquad \begin{array}{c} \mathsf{CH_2} - \mathsf{COOH} \\ \mathsf{H_2N_{C}} \\ \mathsf{NH} \\ \mathsf{NH} \end{array}$$
 guanidinoacetate



Synthesis of creatine (2. part)

N¹-methylation of guanidinoacetate



guanidinoacetate

N-methylguanidine-N-acetate

creatine

N²-Phosforylation of creatine

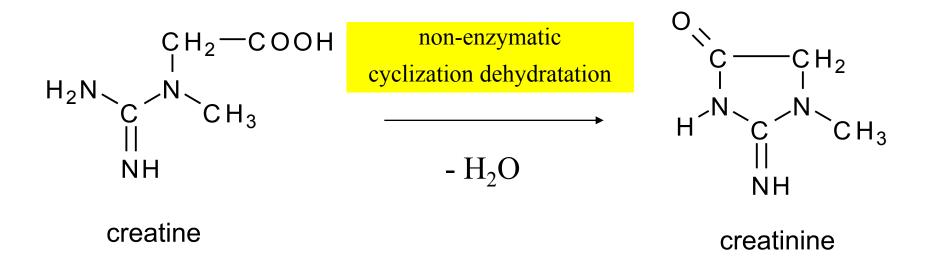
$$\begin{array}{c} CH_2-COOH \\ H_2N-C-N \\ CH_3 \\ NH \end{array}$$

$$\begin{array}{c} OH \\ CH_2-COOH \\ HO-P-N-C-N \\ OH \\ NH \end{array}$$

creatine

creatine phosphate

Creatinine is a catabolite of creatine made in <u>non-enzymatic</u> reaction



Arginine - summery

- semiessential AA (childhood)
- the most basic AA (guanidine, $pK_B = 0.5$)
- no transamination, Arg releases ornithine + urea
- $Arg + Gly + Met \rightarrow creatine$
- releases NO (vasodilator)
- OTC (over-the-counter) preparations in pharmacy

Serine

Dehydratation + deamination of serine

$$\begin{array}{c} OH & H \\ CH_2-C-COOH & \longrightarrow & H_3C-C-COOH \\ NH_2 & NH \\ \end{array}$$
 enamine imine
$$\begin{array}{c} H_2O \\ H_2O \\ \end{array}$$

$$\begin{array}{c} NH_3 + H_3C-C-COOH \\ O \\ \end{array}$$
 pyruvate

Conversion of serine to glycine

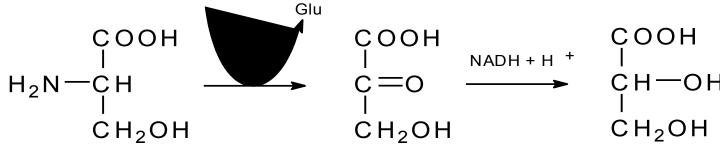
CH₂—CH—COOH + FH₄
$$\longrightarrow$$
 CH₂—COOH + HOCH₂—FH₄
OH NH₂ \longrightarrow NH₂ \longrightarrow glycine \longrightarrow H₂O + N₅N₁₀-CH₂-FH₄

cofactor:

tetrahydrofolate (FH₄)

Transamination of serine and glucose formation





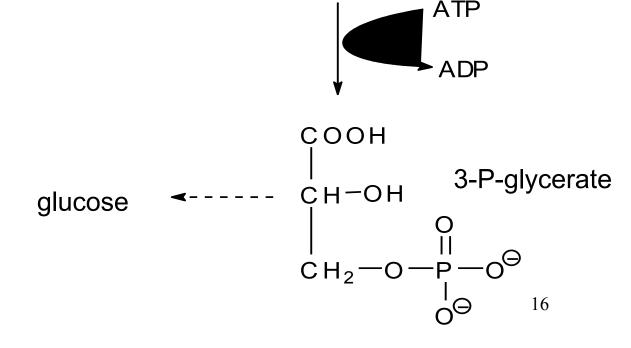
serine

reverse reaction: synthesis of serine

pathway is different - through phosphoserine

hydroxypyruvate

glycerate



Decarboxylation of serine gives ethanolamine. Methylation of ethanolamine leads to choline

Betaine is made by choline oxidation

betaine

Serine - summary

- non-essential glucogenic AA
- source of C1 fragments (attached to tetrahydrofolate)
- component of glycerophospholipids
- decarboxylation gives ethanolamine \rightarrow choline
- carbon skeleton used for selenocysteine
- serine side chain in proteins:

the site of phosphorylation

the linkage of oligosaccharides (O-glycoside bond)

nucleophilic -OH group in active site of enzyme (serine proteases)

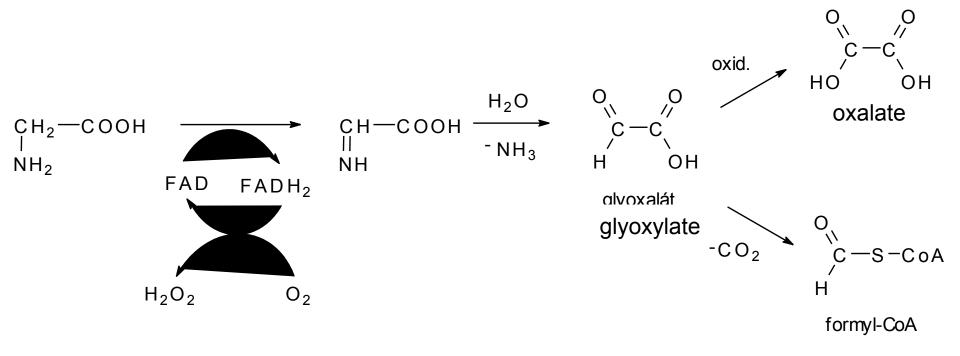
Glycine

The complete catabolism of glycine

$$CH_2$$
— $COOH + FH_4$ \longrightarrow N_5N_{10} - CH_2 - FH_4 + CO_2 + NH_3 NH_2

C1 fragment (methylene) is transferred to tetrahydrofolate

Oxidative deamination of glycine



- 60 % catabolism of glycine and ethanolamine
- 30 % catabolism of vitamin C
- 10 % food (spinach, rhubarb, mangold, tea, cocoa)

Glycine - summary

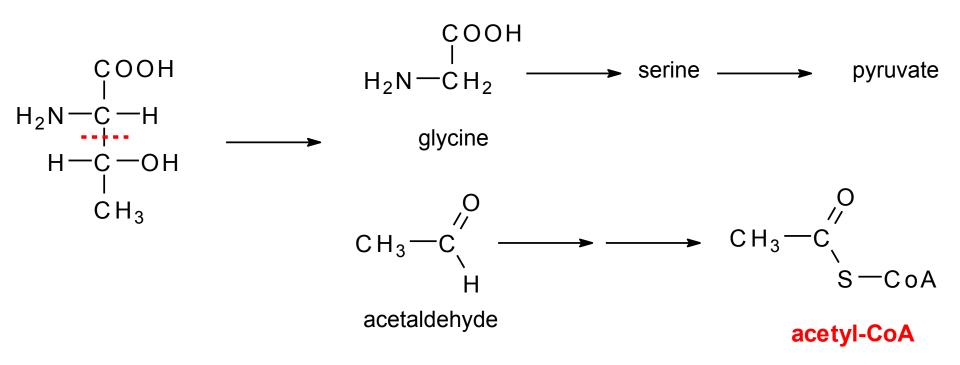
Catabolism

- complete oxidation to
 CO₂ + NH₃
- oxidative deamination to oxalate

Anabolic conversions

- donor of C1 fragment
- serine
- porphyrines
- purine bases
- creatine
- glutathione (GSH)
- conjugation agent (bile acids, xenobiotics)

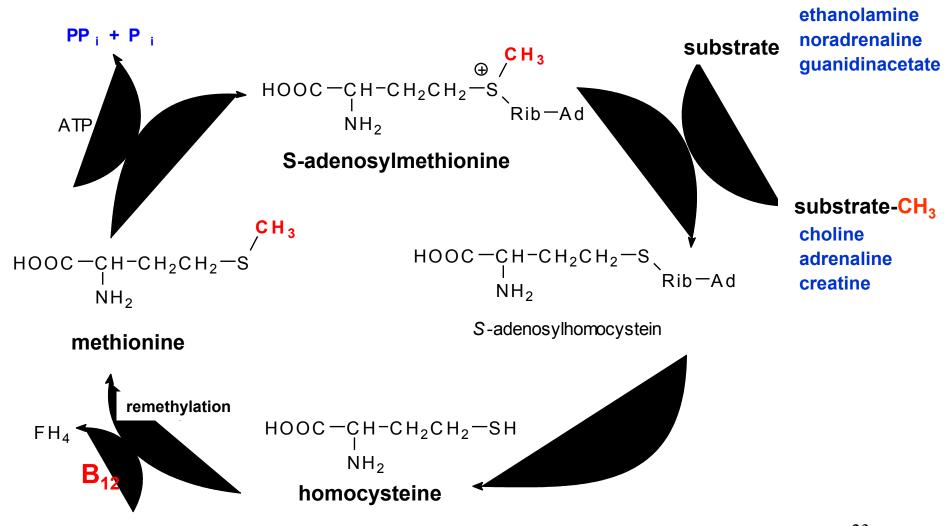
Threonine (4C) is split to glycine (2C) and acetaldehyde (2C)



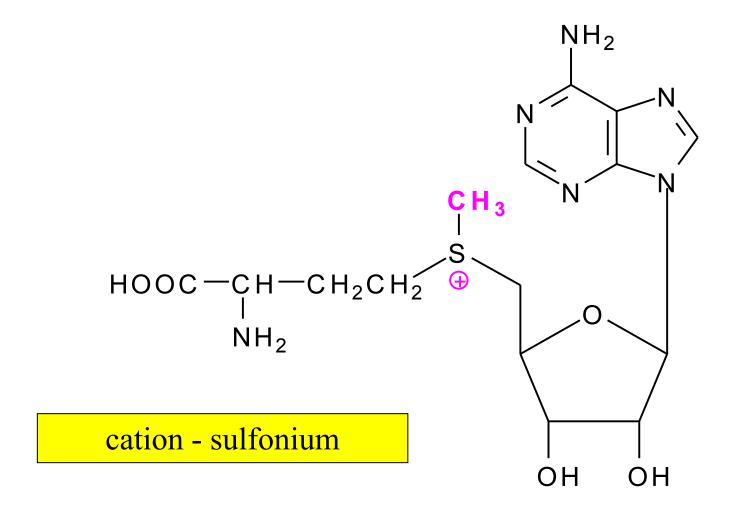
- essential AA
- two asymmetric C atoms
- the site of phosphorylation and glycosylation in proteins

CH₃—FH₄

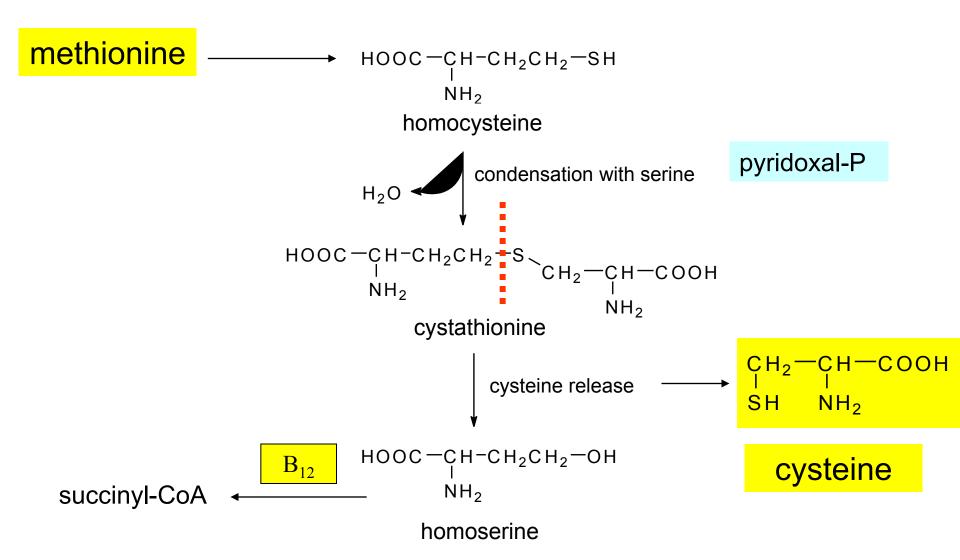
Methionine is methylation agent (homocysteine side product)



S-Adenosylmethionine (SAM) contains trivalent positively charged sulfur atom



Cysteine is made from methionine



Methionine - summery

- essential AA, rather rare in foodstufs
- S-adenosylmethionine (SAM) is methylation agent
- metabolized to cysteine \Rightarrow Cys is non-essential AA
- C-skeleton of cysteine comes from serine, sulfur atom from methionine
- final catabolite is succinyl-CoA (glucogenic)

Homocysteine is harmful

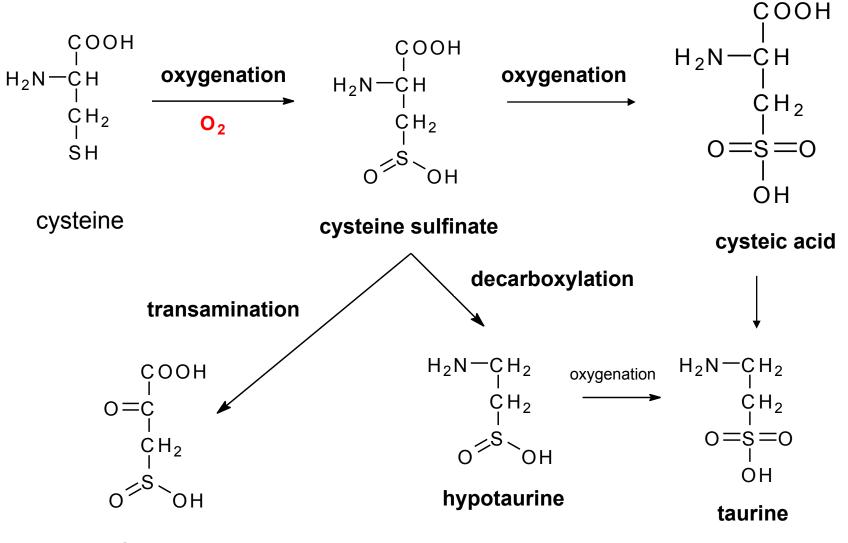
- mechanism of its action is not yet understood
- direct action on blood vessel epithelium
- decreases thrombocyte life and fibrinolysis
- supports formation of oxygen radicals damage of vessel wall
- increases LDL lipoperoxidation
- elevated blood level of homocysteine is risk factor of cardiovascular diseases

to eliminate homocysteine - three vitamins are needed:

folate, cobalamine, pyridoxin

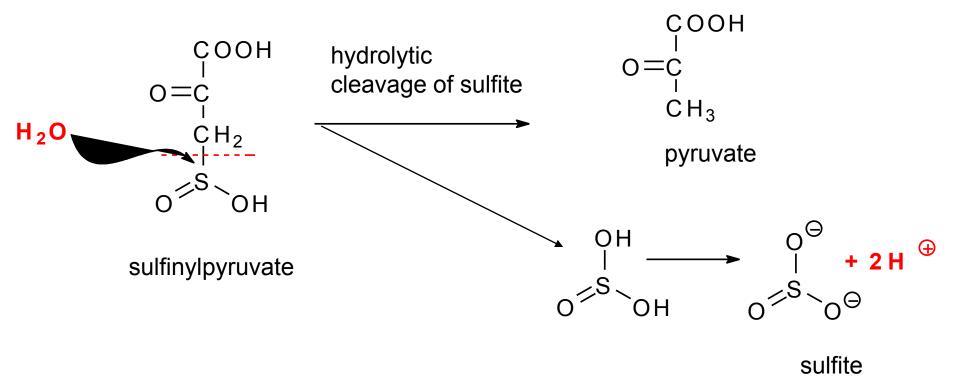
Cysteine

Cysteine catabolism: oxygenation of -SH group



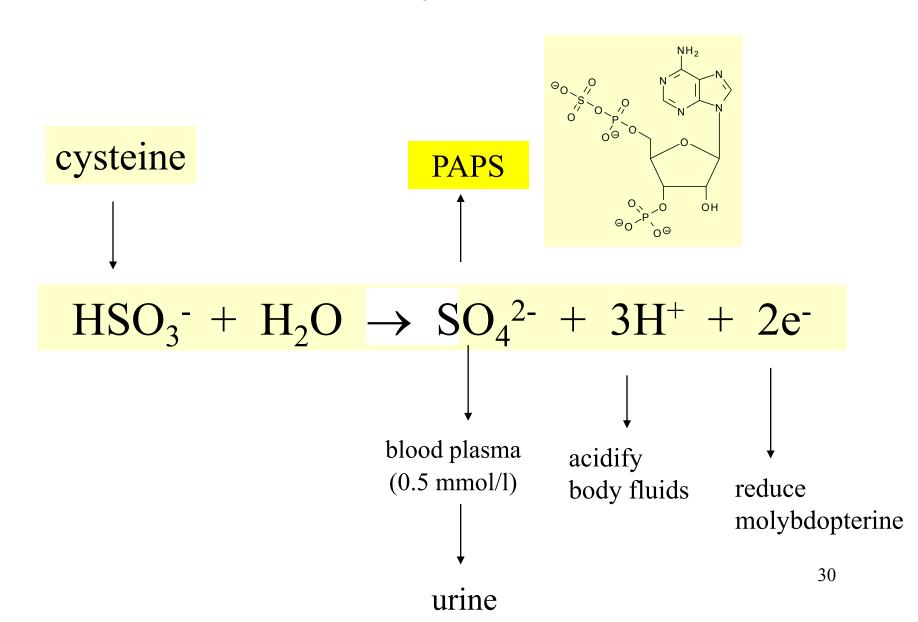
sulfinylpyruvate

The formation of sulfite



under physiol. pH – dissociation only to HSO₃⁻

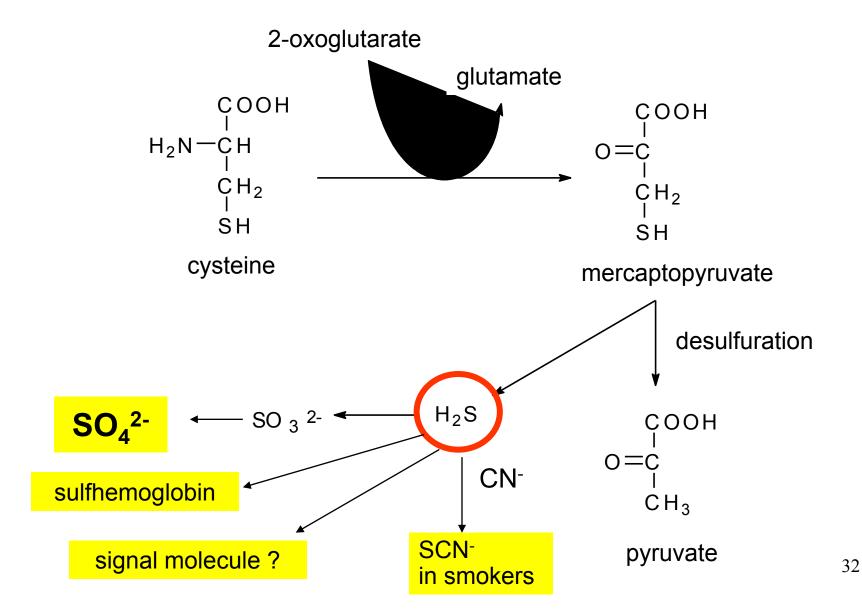
Sulfite oxidase catalyzes sulfate formation



Distinguish

Sulfite	anion SO ₃ ²⁻
Sulfide inorganic	anion S ² - (e.g. ZnS zinc sulfide)
Sulfide organic	R-S-R dialkylsulfide
Sulfate	anion SO ₄ ²⁻

Transamination of cysteine and sulfane production



Cysteine - summary

- both pathways go to pyruvate (glucogenic)
- main catabolism: sulfur oxygenation \rightarrow sulfite \rightarrow sulfate
- high protein diet leads to physiologic acidosis
- cystein provids taurine conjugation agent (e.g. bile acids)
- taurine is semiessential AA in metabolic stress
- taurine is a component of "energy drinks"
- cysteine part of glutathione (GSH) antioxidant
- decarboxylation of Cys cysteamine, in CoA-SH
- in proteins disulfide bonds (tertiary structure)
- cysteine proteases: active site contains –SH group

Six amino acids provide pyruvate

- 1. Serine dehydratation + deamination
- 2. Glycine *via* serine
- 3. Threonine *via* glycine
- 4. Alanine transamination (ALT)
- 5. Cysteine both catabolic pathways
- 6. Tryptophan *via* alanine (see later)

Aspartate

- Transamination of Asp \rightarrow oxaloacetate (CAC)
- AST (aspartate aminotransferase) clinically important enzyme
- in urea cycle, Asp donates one nitrogen into urea and releases fumarate
- decarboxylation of Asp $\rightarrow \beta$ -alanine (part of coenzyme A)
- donor of nitrogen in purine synthesis (fumarate released)
- whole structure given for pyrimidine bases synthesis
- aspartam (sweetener)
- condensation with ammonia → asparagine
 (for cell utilization, not as detoxication of ammonia)

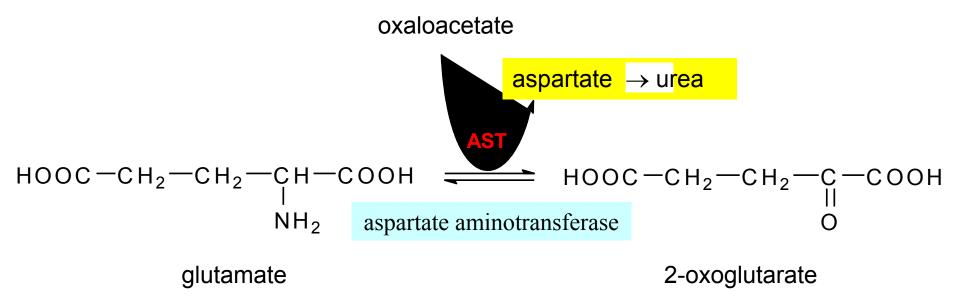
β-Alanine is made by the decarboxylation of aspartate

HOOC-CH₂-CH-COOH
$$\longrightarrow$$
 HOOC-CH₂-CH₂
NH₂
NH₂

in the structure of CoA-SH

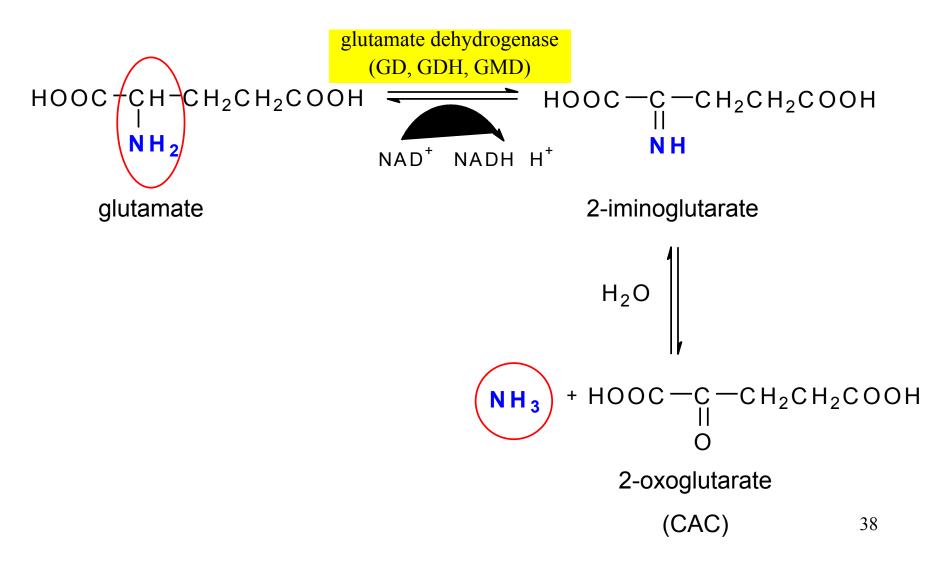
Glutamate

Glutamate with oxaloacetate afford aspartate (transamination)



AST reaction produces aspartate for urea synthesis

Dehydrogenative deamination of glutamate is the main producer of ammonia in tissues



Decarboxylation of glutamate

glutamate

GABA gama-aminobutyric acid

Glutamate - summary

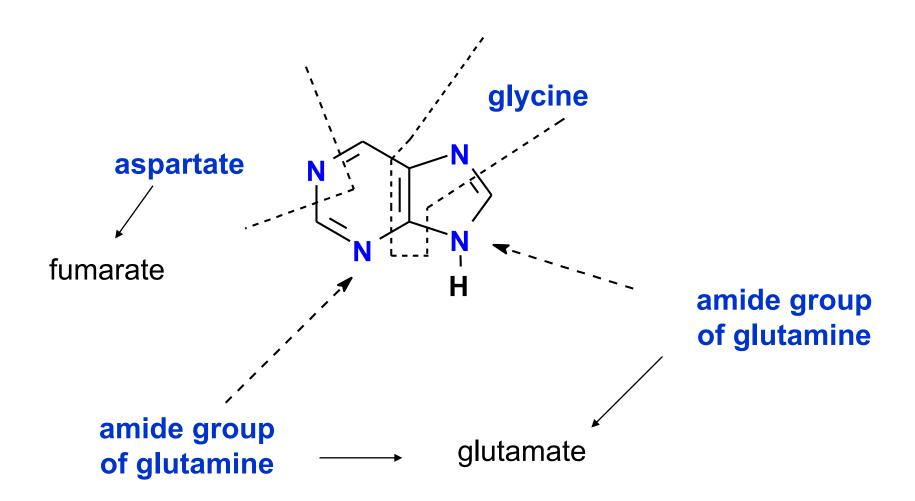
- produced in the transaminations of most AA
- glutamate dehydrogenase reaction produces most ammonia in body
- transaminations are reversible, so glutamate can be converted to 2-oxoglutarate (glucogenic)
- Glu + NH₃ \leftrightarrows Gln (ammonia detoxification)
- glutamate is readily made from glutamine, histidine, proline, ornithine
- pure monosodium glutamate (MSG, E621), flavour enhancer, can cause health problems (chinese restaurant syndrome)

Glutamine

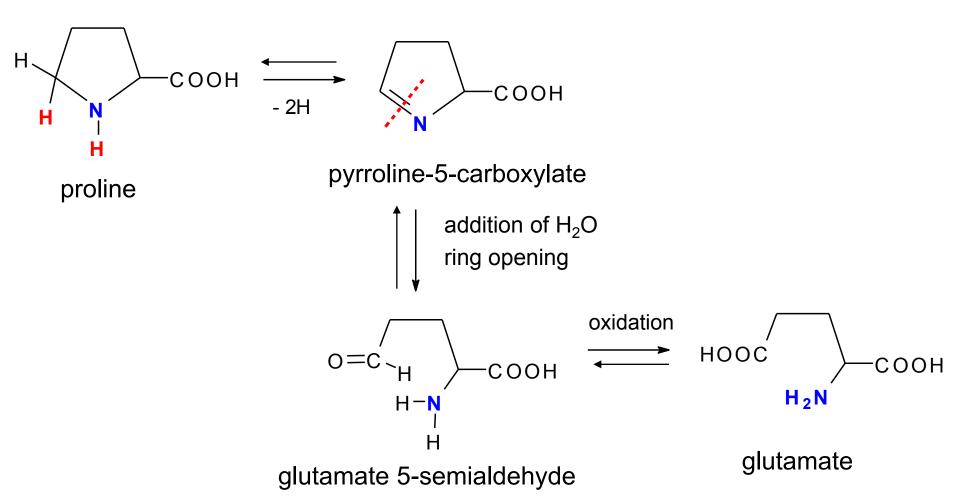
See also the previous lecture (AA-1)

- glutamine synthesis is the way of ammonia detoxification in tissues including liver
- in kidneys, glutamine releases ammonia (deamidation)
- metabolic fuel for some tissues (enterocytes, fibroblasts, lymphocytes, macrophages)
- donor of nitrogen for syntheses (glucosamine, purines)

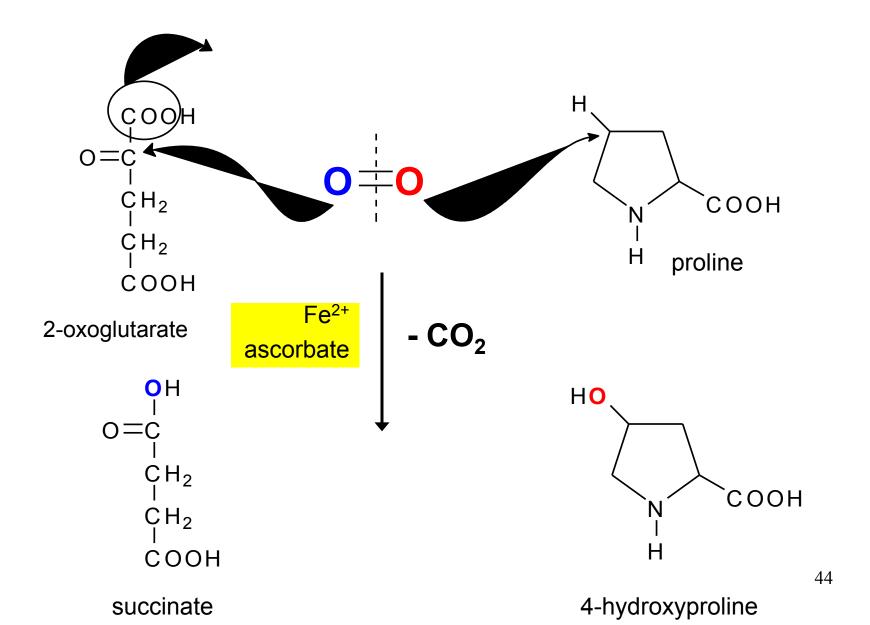
Three amino acids donate four N atoms in purine bases synthesis



Proline is converted to glutamate (and vice versa)



Hydroxylation of proline with 2-oxoglutarate as reductant

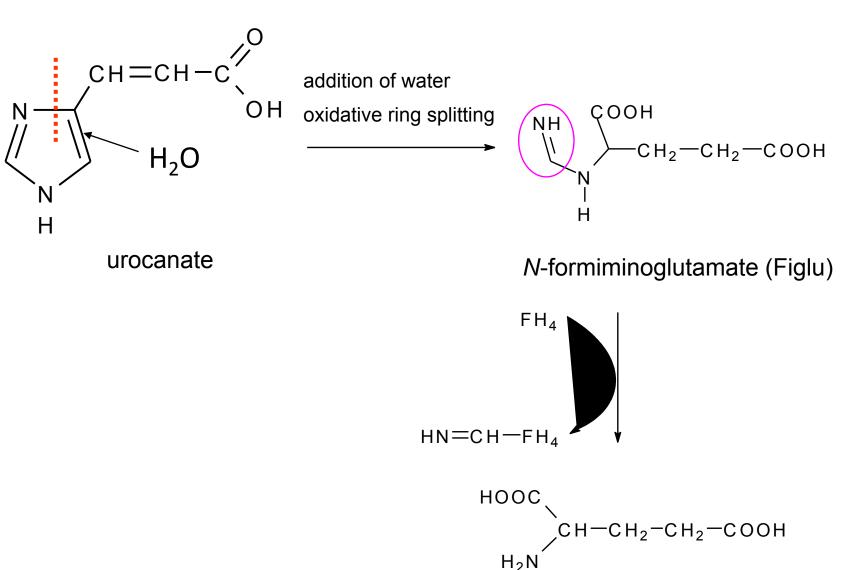


Proline - summary

- non-essential AA, can be formed from glutamate
- converted to glutamate (glucogenic)
- hydroxylation of proline in collagen is post-translation modification,
 requires ascorbate (vitamin C), Fe²⁺, and 2-oxoglutarate
 (unusual co-reductant)
- 4-hydroxyproline is catabolized to pyruvate (see Harper)

Catabolism of histidine starts with desaturation and deamination

Urocanate cleavage affords C₁ fragment



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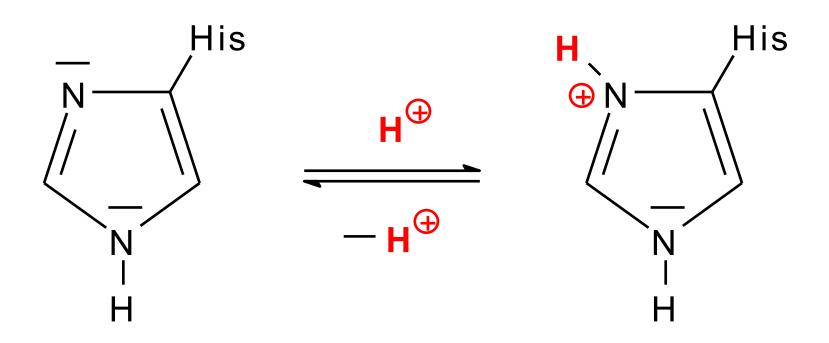
glutamate

Decarboxylation of histidine → **histamine**

$$\begin{array}{c|c} CH_2-CH-COOH \\ NH_2 \\ \hline \\ -CO_2 \\ H \end{array}$$
 histidine
$$\begin{array}{c} CH_2-CH_2-NH_2 \\ \hline \\ -CO_2 \\ \end{array}$$
 histamine

- histidine decarboxylase occurs in mast cells and basophils
- histamine stimulates HCl production in stomach
- is released in allergic reactions
- triggers inflammatory response
- antihistaminics are drugs blocking the action of histamine

Histidine is responsible for buffering actions of proteins



$$pK_{B} = 8$$

$$pK_A$$
 (His) = 6
 pK_A (His in proteins) = 6-8

Histidine - summary

- semiessential AA
- no transamination, catabolism begins with desaturation and deamination
- the source of 1C groups (formimino)
- converted to glutamate (glucogenic)
- histidine is abundant in hemoglobin buffer system
- post-translation modification: methylation of His in actine/myosine
 - → 3-methylhistidine its urine excretion is the indicator of muscle proteolysis and nutrition status

Leucine (1) - transamination + decarboxylation

$$H_3C$$

$$CH-CH_2-CH-COOH$$

$$H_3C$$

$$NH_2$$
transamination
$$H_3C$$

$$CH-CH_2-C-COOH$$

$$H_3C$$

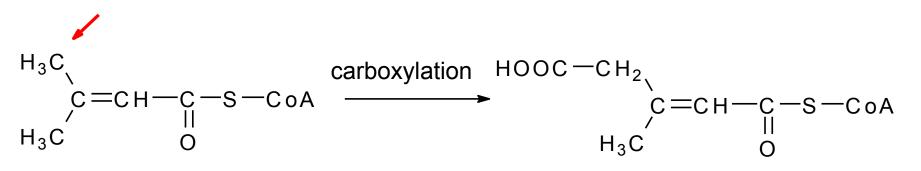
branched 2-oxoacid

$$\begin{array}{c|c} -c \circ_2 & \text{oxidative} \\ \text{decarboxylation} \\ H_3C & CH-CH_2-C-S-C \circ A \\ H_3C & O \end{array}$$

branched acyl-CoA (isovaleryl-CoA)

Leucine (2) - dehydrogenation

Leucine (3) – carboxylation at C4



acyl of dicarboxylic

branched unsaturated acid

Leucine (4) – hydratation of double bond

HOOC-CH₂

$$C = CH - C - S - CoA$$
 H_2O
 GH_2
 GH_2
 GH_3
 GH_4
 GH_4

3-hydroxy-3-methylglutaryl-CoA

(HMG-CoA)

Leucine (5) – splitting the C-C bond in HMG-CoA

acetoacetate

acetyl-CoA

Compare the final products of BCAA

Leucine	acetyl-CoA + acetoacetate	ketogenic
Isoleucine B ₁₂	acetyl-CoA + succinyl-CoA	ketogenic glucogenic
Valine B ₁₂	succinyl-CoA	glucogenic

BCAA - summery

- all BCAA are essential
- the first three reactions are the same (transamination, oxid. decarboxylation, dehydrogenation), final products are different
- leucine ketogenic, valine glucogenic, isoleucine mixed
- after meal, BCAA make about 70 % of AA in blood, because the liver does not utilize them (lack of aminotransferases)
- BCAA are most utilized in muscles and brain
- BCAA infusion are applied in severe catabolic conditions

Lysine catabolism (1)

HOOC—
$$CH$$
— $(CH_2)_4$ — NH_2 O = C — $(CH_2)_2$ — $COOH$

lysine

2-oxoglutarate

 $+H_2O$

HOOC— CH — $(CH_2)_4$ — N = C — $(CH_2)_2$ $COOH$

ketimine (Schiff base)

COOH

Lysine catabolism (2)

Lysine catabolism (3)

$$HOOC - CH - (CH_2)_3 - C \setminus H_2N - CH - (CH_2)_2COOH \\ NH_2 H$$

$$H_2N$$
-CH-(CH₂)₂COOH

allysine

glutamate

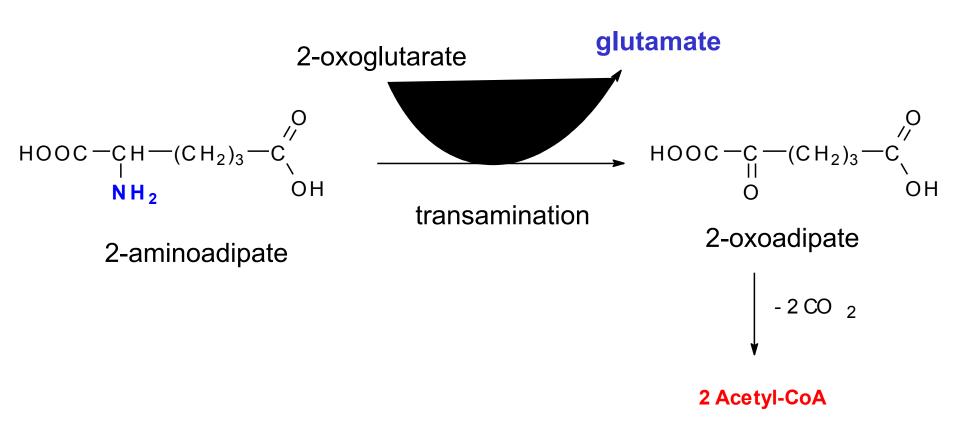
Lysine catabolism (4)

HOOC-CH-(CH₂)₃-C
$$\stackrel{O}{\longrightarrow}$$
 HOOC-CH-(CH₂)₃-C-OH NH₂ $\stackrel{O}{\longrightarrow}$ HOOC-CH-(CH₂)₃-C OH dehydrogenation $\stackrel{O}{\longrightarrow}$ HOOC-CH-(CH₂)₃-C OH $\stackrel{O}{\longrightarrow}$ OH

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2-aminoadipate

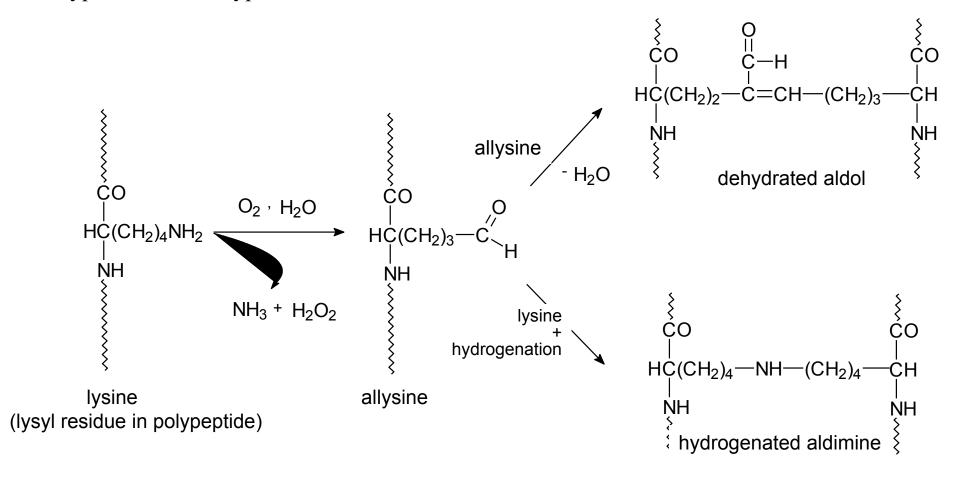
Lysine catabolism (5)



Lysine is the substrate for carnitine (the transfer of FA from cytosol to mitochondria)

Cross-links in collagen

products of reaction between the amino groups in side chains of **lysine** with the modified lysine side chains comprising the aldehyde group (the result of oxidation of lysine to **allysine**) – aldol type or aldimine type of cross-links.



Formation of fibrin clot during blood coagulation (cross-linking of fibrin)

Fibrin —
$$CH_2$$
— CH_2 — CH_2 — CH_2 — CH_2 — CH_2 — Fibrin lysine glutamine H_2 H_2 H_3 H_4 H_4 H_4 H_4 H_4 H_4 H_4 H_4 H_5 H

Lysine - summary

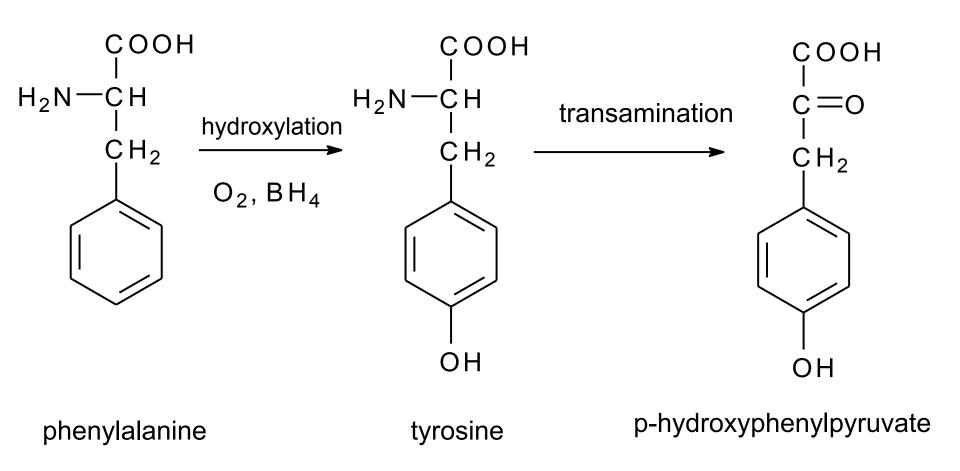
- essential AA, no transamination
- ε-amino group is removed as glutamate
- α -amino group is removed from aminoadipate by transamination
- final product acetyl-CoA (ketogenic)

Other conversions:

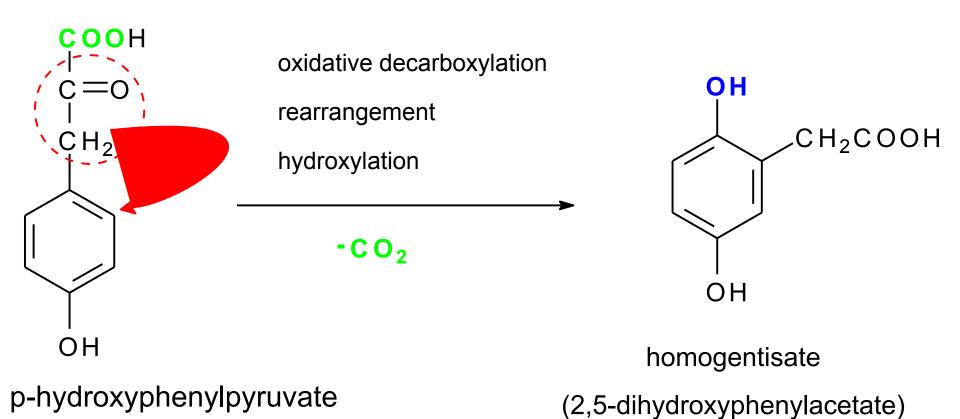
- lysine in many proteins binds ubiquitin (targeting for proteasome)
- carnitine (transport system for FA to mitochondria)
- decarboxylation \rightarrow cadaverine
- in collagen: cross bridges, hydroxylation → hydroxylysine
- in fibrin: cross linking during blood coagulation

Phenylalanine, Tyrosine

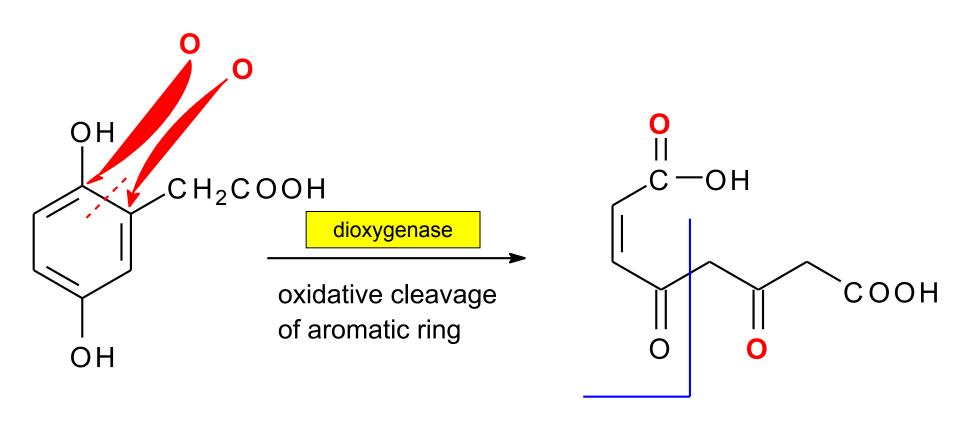
Catabolism (1)



Catabolism (2)



Catabolism (3)



maleylacetoacetate

Catabolism (4)

maleylacetoacetate

fumarylacetoacetate

Catabolism (5)

HOOC
$$H_2O$$
 HOOC $COOH$ H_3C $COOH$

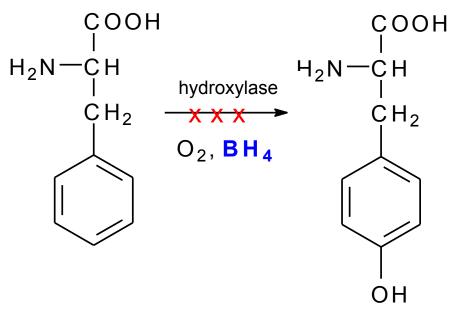
fumarylacetoacetate

fumarate

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acetoacetate

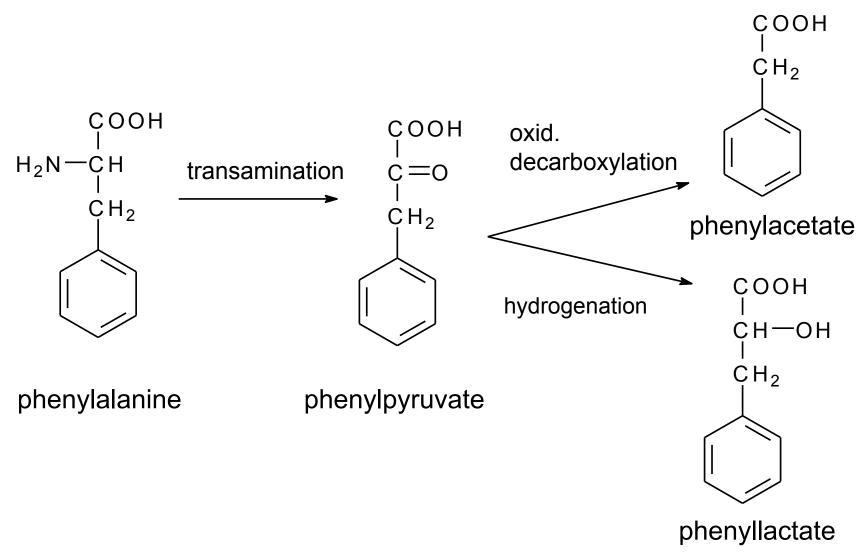
Hyperphenylalaninemia and Phenylketonuria



- deficit of hydroxylase or BH₄
- elevated blood Phe and its metabolites
- excretion of phenylpyruvateby urine

phenylalanine tyrosine

Metabolites of phenylalanine



Hyperphenylalaninemia and Phenylketonuria

- if not treated properly mental retardation and other problems
- treatment low phenylalanine diet
- products containing sweetener aspartam must be avoided
- L-aspartyl-L-phenylalanine methyl ester phenylalanine is released by hydrolysis:

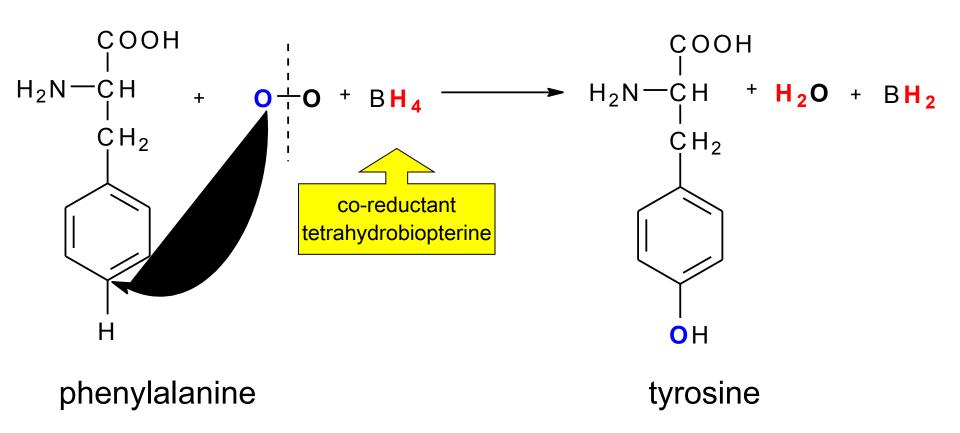
HOOC
$$H_2N$$

$$O$$

$$O$$

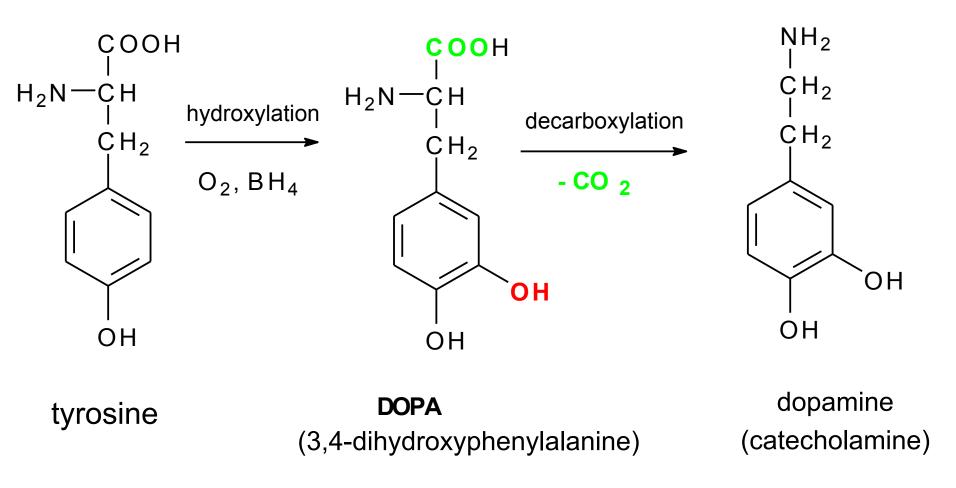
$$CH_3$$

Hydroxylation of phenylalanine gives tyrosine



Tyrosine

DOPA and dopamine from tyrosine



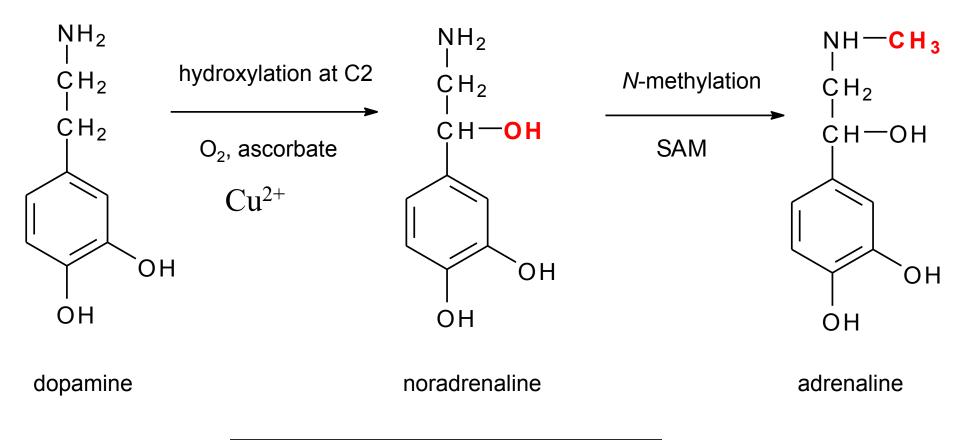
Linguistic note

- abbreviation **DOPA** comes from older English nomenclature
- oxo group and hydroxyl group were not distinguished properly:

DOPA = dioxophenylalanine

• correct chemical name is: 3-(3,4-dihydroxyphenyl)alanine

Two more catecholamines from dopamine



nor- = N-demethyl

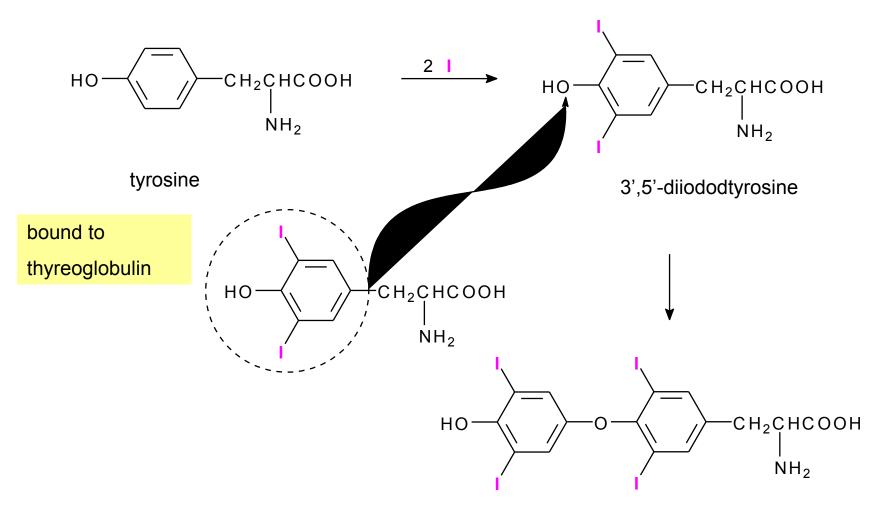
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Conversion of tyrosine to melanin, a dark pigment of skin, hair, fur

DOPA

dopaquinone

Conversion of tyrosine to thyroxine



thyroxine

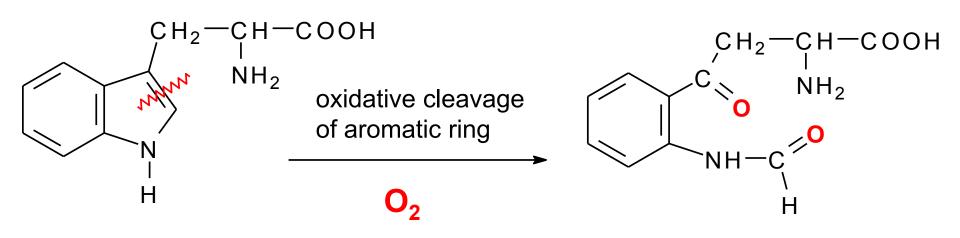
Phenylalanine, tyrosine - summary

- Phe is essential amino acid, Tyr not
- Tyr is made by Phe hydroxylation (tetrahydrobiopterine cofactor)
- catabolism is the same for both AA (mixed AA)
- provide fumarate for CAC (glucogenic)
- acetoacetate (ketone body)
- tyrosine is converted to hormones (catecholamines, thyronines)
 and dark skin pigment melanin

tryptophan

N-formylkynurenine

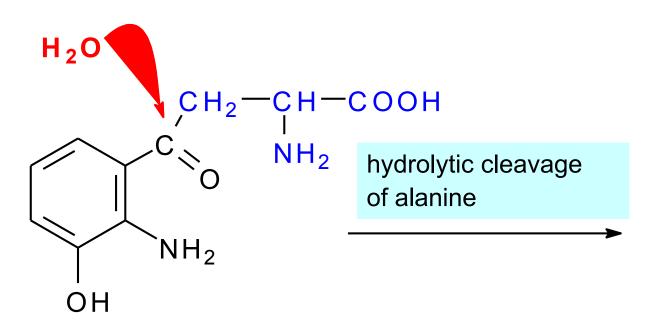
Catabolism (1)



Catabolism (2)

formamide

Catabolism (3)



$$H_3C$$
 — CH — $COOH$ NH_2 Ala

3-hydroxykynurenin

3-hydroxyanthranilate

Catabolism (4)

3-hydroxyanthranilate

Decarboxylation of tryptophan

$$\begin{array}{c} \mathsf{CH_2} - \mathsf{CH} - \mathsf{COOH} \\ \mathsf{NH_2} \\ \mathsf{NH_2} \\ \mathsf{H} \end{array} \qquad \begin{array}{c} \mathsf{CH_2} - \mathsf{CH_2} \\ \mathsf{NH_2} \\ \mathsf{NH_2} \\ \mathsf{H} \end{array}$$

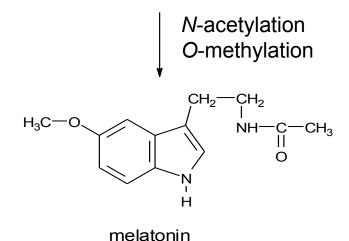
tryptophan

tryptamine

Conversion of tryptophan to melatonin

sleep-wake cycle

the hormone of darkness



Tryptophan - summary

- essential AA
- complicated catabolism
- donor of 1C fragment (formic acid formate)
- no transamination, amino group leaves as alanine (glucogenic)
- final product acetyl-CoA (ketogenic)
- source of nicotinamide and NAD⁺
- bacterial decomposition in large intestine → indole and skatole
 (3-methylindole) exhibit strong fecal odor

Five vitamins are formed in the body, only four are utilized

Vitamin	Where and how produced
Niacin	in tissues, from tryptophan
Biotin	large intestine (bacteria)
Phylloquinone	large intestine (bacteria)
Calciol	skin, from cholesterol (UV radiation)
Cobalamine	large intestine (bacteria) – not absorbed!

Seven amino acids do not undergo transamination

Amino acid	α-NH ₂ group is removed as
Arginine	ornithine
Lysine	2-aminoadipate
Methionine	homoserine
Threonine	glycine
Tryptophan	alanine
Proline	glutamate
Histidine	NH ₃ (desaturation deamination)

Biochemically relevant product AA Ala pyruvate \rightarrow glucose Arg urea, NO, creatine Ser ethanolamine \rightarrow choline \rightarrow betaine; donor of 1C fragment, selenocysteine Gly heme, creatine, GSH, conjugation reagent (e.g. glycocholate) Met donor of methyl, creatine, homocysteine, cysteine glutathione (GSH), taurine, SO₄²⁻, PAPS, cysteamine (CoA) donor of -NH₂ (urea, pyrimidines), oxaloacetate, fumarate, β -alanine (CoA) Asp Glu NH₄⁺, 2-oxoglutarate, GABA, ornithine Gln NH₄⁺, donor of -NH₂ (synthesis of glucosamine, purines) Pro glutamate, hydroxyproline His glutamate, histamine, donor of 1C fragment Lys glutamate, allysine (collagen), carnitine, cadaverine Tyr fumarate, catecholamines, thyroxine, melanins Trp nicotinamide, serotonin, melatonin, donor of 1C fragment, indole, skatole

Overview: decarboxylation of amino acids

AA	Product	Comments
Ser	ethanolamine	part of phospholipids, precursor of choline
Cys	cysteamine	part of coenzyme A (CoA-SH)
Phe	phenethylamine	structural part of stimulants (amphetamine, ephedrine etc.)
Tyr	tyramine	occurs in some foods, may cause migraine
Asp	β-alanine	part of pantothenic acid, CoA-SH, carnosine
Glu	GABA	gama-aminobutyric acid, inhibition neurotransmiter
Lys	cadaverine	product of putrefaction (decay of proteins)
Arg	agmatine	signal molecule in CNS
His	histamine	triggers allergic reactions
Trp	tryptamine	precursor of serotonine and melatonine
DOPA	dopamine	catecholamine, precursor of noradrenaline/adrenaline
Ornithine	putrescine	putrefaction product; precursor of spermidine/spermine