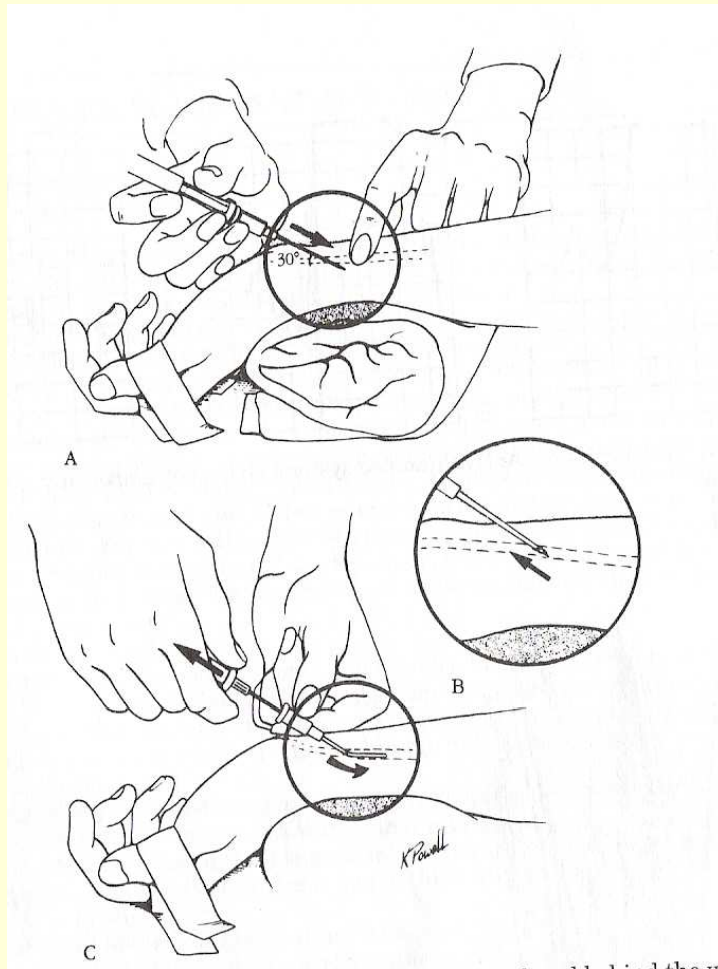


# Therapeutic and diagnostic interventions in critically ill patients

ARK

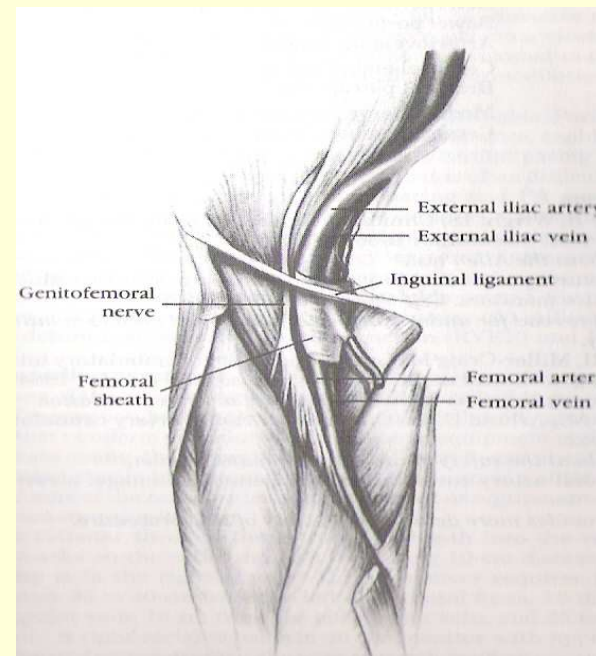


Arterial line placement : a. radialis, femoralis

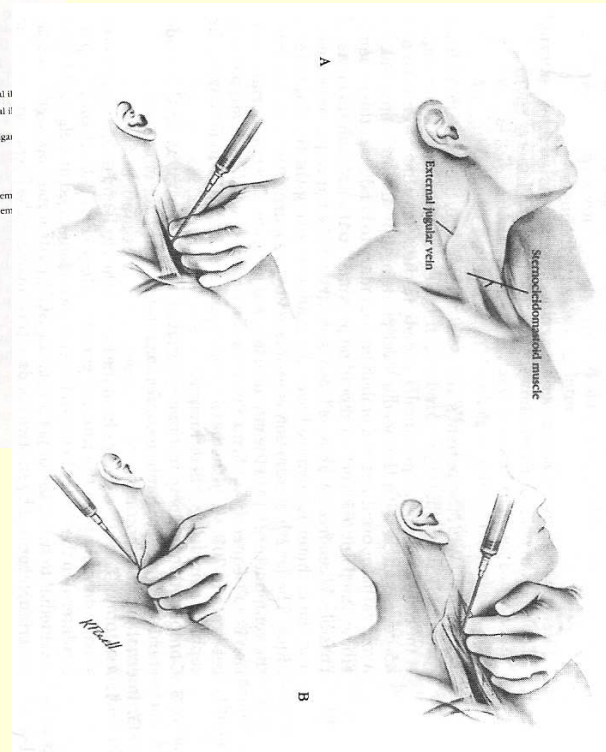
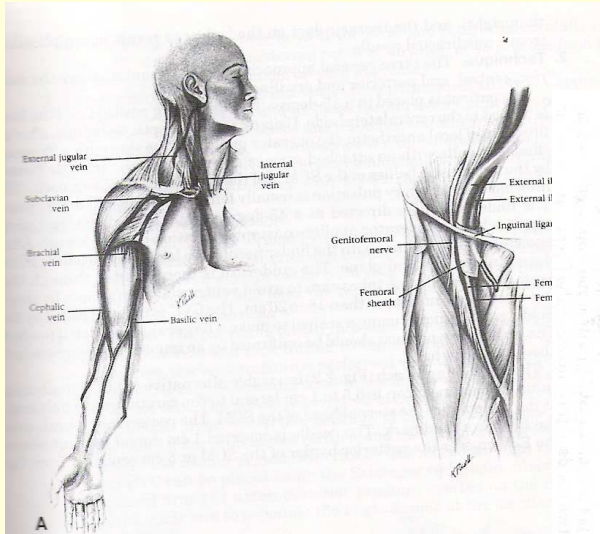
indication : invasive blood measurement, blood sampling for biochemical evaluation and blood gas analysis

Allen test

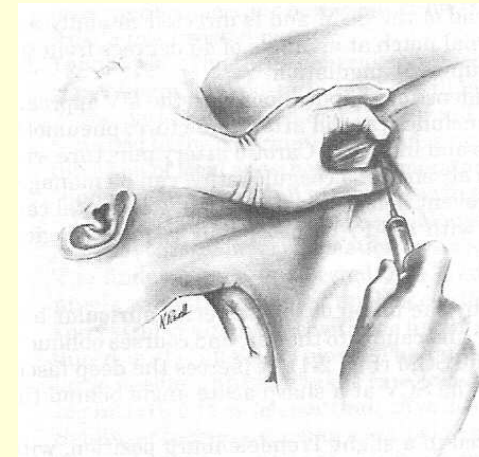
complication: infection, bleeding, ischemia



# Central venous catheterization



- v. subcl : good fixation, less frequent infection
- v. jugularis: easy access in Swan Ganz cannulation
- v femoralis: high flow, more infection



DISORDER	HCO <sub>3</sub> <sup>-</sup> (mmol/l) (mEq/L)	pCO <sub>2</sub> (mmHg)	SBE mmol/l (mEq/L)
<b>Metabolic acidosis</b>	<22 mmol/l (<22 mEq/L)	= (1.5 x HCO <sub>3</sub> <sup>-</sup> ) + 8 = 40 + SBE	<-3
<b>Metabolic alkalosis</b>	>26	= (0.7 x HCO <sub>3</sub> <sup>-</sup> ) + 21 = 40 + (0.6 x SBE)	>+3
<b>Acute respiratory acidosis</b>	= [(pCO <sub>2</sub> - 40) / 10] + 24	>45	= 0
<b>Chronic respiratory acidosis</b>	= [(pCO <sub>2</sub> - 40) / 3] + 24	>45	= 0.4 x (pCO <sub>2</sub> - 40)
<b>Acute respiratory alkalosis</b>	= [(40 - pCO <sub>2</sub> ) / 5] + 24	<35	= 0
<b>Chronic respiratory alkalosis</b>	= [(40 - pCO <sub>2</sub> ) / 2] + 24	<35	= 0.4 x (pCO <sub>2</sub> - 40)

!chyba(pCO<sub>2</sub>-40) u  
respir alkaloz

příklad:

1. chceme vědět jestli má pacient čistou MAC, pH 7.2, HCO<sub>3</sub> 14 mmol/l, paCO<sub>2</sub> 3,9 kPa  
jestli má pacient MAC a 14 HCO<sub>3</sub> pak podle vzorce (nahore):

pCO<sub>2</sub> = (1.5\*14)+8=29mmHg=3,9kPa, což odpovídá skutečnému pCO<sub>2</sub>, pacient má čistou MAC

2. chceme vědět jestli má pacient čistou MAC, pH 7.1, HCO<sub>3</sub> 14 mmol/l, paCO<sub>2</sub> 5,3 kPa  
jestli má pacient MAC a 14 HCO<sub>3</sub> pak podle vzorce (nahore):

pCO<sub>2</sub> = (1.5\*14)+8=29mmHg=3,9kPa, ale pacient ma 5,3 kPA, měl by mít nižší pCO<sub>2</sub> (kompenzace), to znamená že má zároveň respirační acidosu

atd...

# metabolická acidosa, MAC

## INCREASED ANION GAP

### Endogenous acids

Renal failure

Ketoses - diabetic, alcohol, starvation

Lactic acid

Unknown anions: liver failure, sepsis

Non-ketotic hyperosmolar hyperglycaemia

### Toxic ingestions

Ethylene glycol

Salicylate

Paraldehyde

Methanol

Toluene

Iron\*

## NORMAL ANION GAP

**Renal tubular acidosis** *Urine SID ( $Na^+ + K^+ - Cl^-$ ) > 0*

**Non-renal** *Urine SID ( $Na^+ + K^+ - Cl^-$ ) < 0*

anion gap  $AG = Na^+ + K^+ - Cl^- - HCO_3^-$

norma AG is 8-12 mmol/l

korigovaný normalní AG = 0.2 (albumin g/l) + 1.5 (phosphate mmol/l).

# MAL

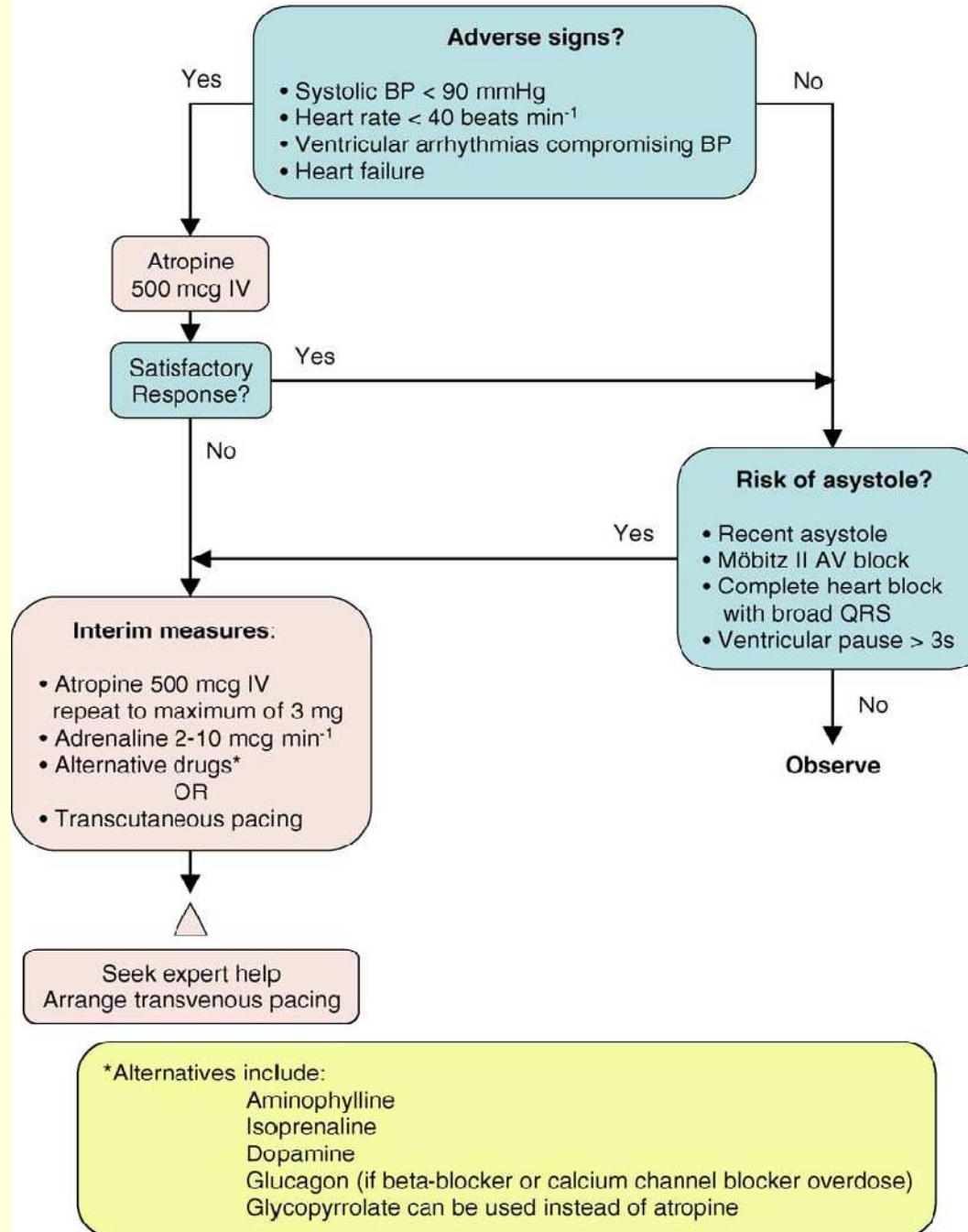
- Severe depletion in free water inducing a parallel increase in  $\text{Na}^+$  and  $\text{Cl}^-$ . Since the concentration of  $\text{Na}^+ > \text{Cl}^-$ , the difference between them increases.
- $\text{Cl}^-$  is lost from the GI tract or urine (diuretic use or abuse) in excess of  $\text{Na}^+$ .
- $\text{Na}^+$  is administered in excess of  $\text{Cl}^-$ .
- There is a severe deficiency of intracellular cations such as magnesium or potassium. This decreases intracellular  $\text{Cl}^-$  and secondarily total body  $\text{Cl}^-$  is reduced.

- Diuretic use (or abuse) is perhaps the most common aetiology of metabolic alkalosis.
- Gastrointestinal losses of  $\text{Cl}^-$  include vomiting, gastric drainage, and rarely, chloride wasting diarrhoea (villous adenoma).
- Administration of non-chloride sodium salts can occur with massive blood transfusions (sodium citrate), parenteral nutrition (sodium acetate), plasma volume expanders (acetate or citrate), Ringer's solution (sodium lactate) or overzealous use of sodium bicarbonate.
- For several hours (or longer) following recovery from hypercarbia, metabolic alkalosis (chloride-responsive) will persist so long as the respiratory failure was chronic enough to induce renal compensation ( $\text{Cl}^-$  excretion).
- Mineralocorticoid excess: primary hyperaldosteronism (Conn's syndrome), secondary hyperaldosteronism, Cushing's syndrome, Liddle's syndrome, Bartter's syndrome, exogenous corticoids, and excessive liquorice intake.

# Bradycardia Algorithm

(includes rates inappropriately slow for haemodynamic state)

If appropriate, give oxygen, cannulate a vein, and record a 12-lead ECG



## Tachycardia Algorithm (with pulse)

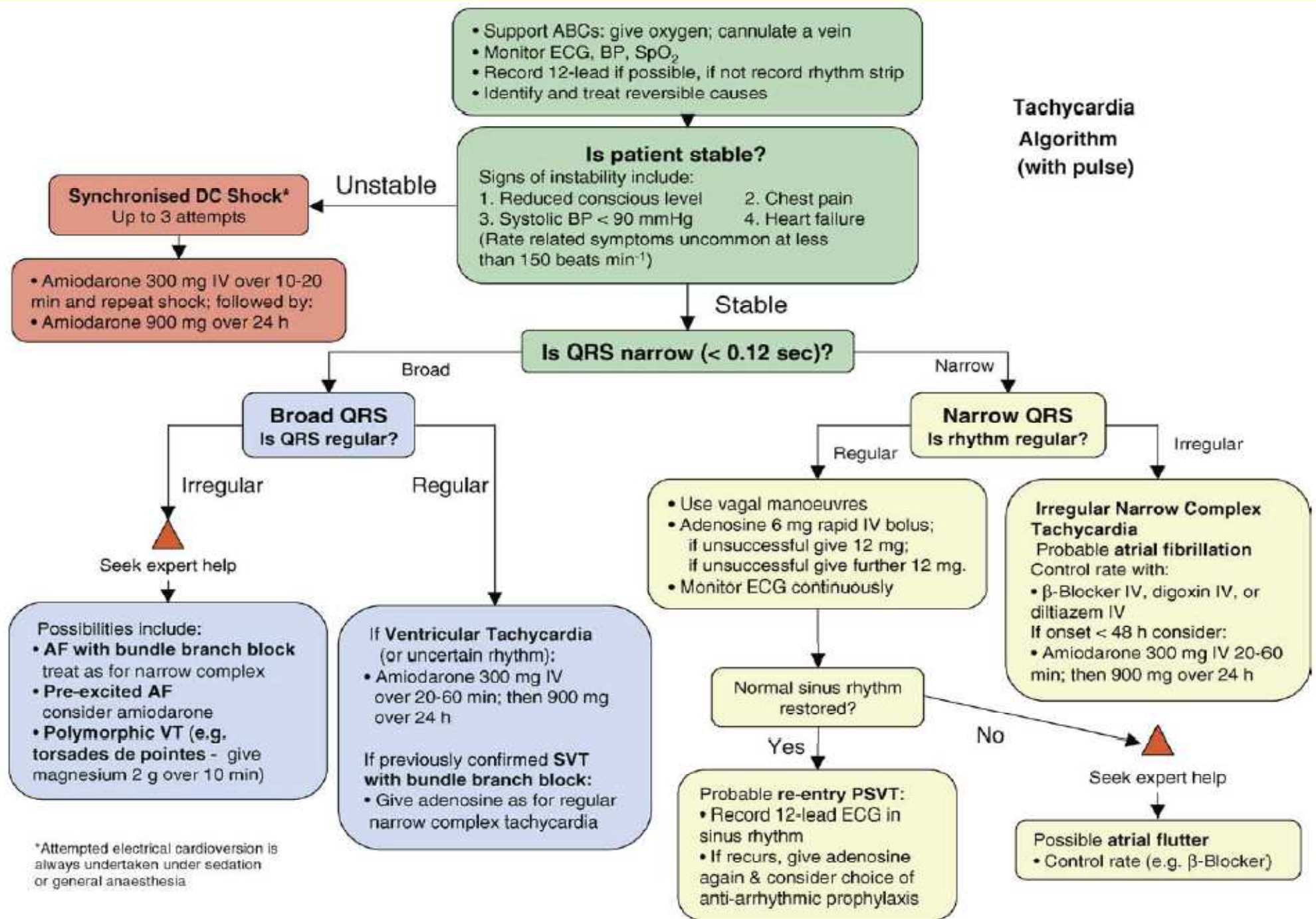


Figure 4.12 Tachycardia algorithm.



# Cardial pacing

## Indikation

### **Symptomatic** bradyarrhythmias

- AV blockade III degree
- bifascicular blockade, Mobitz II in myocardial infarction (sinus bradycardia)
- AV dissociation
- -torsade points

## VVI

rate, mA, mV

technic:

- transvenous : better , but time consuming
- transcutaneous (external stimulation)- method of choice



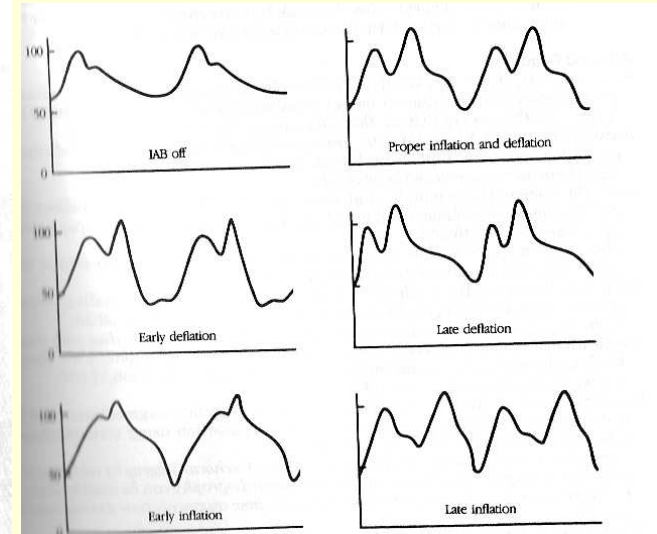
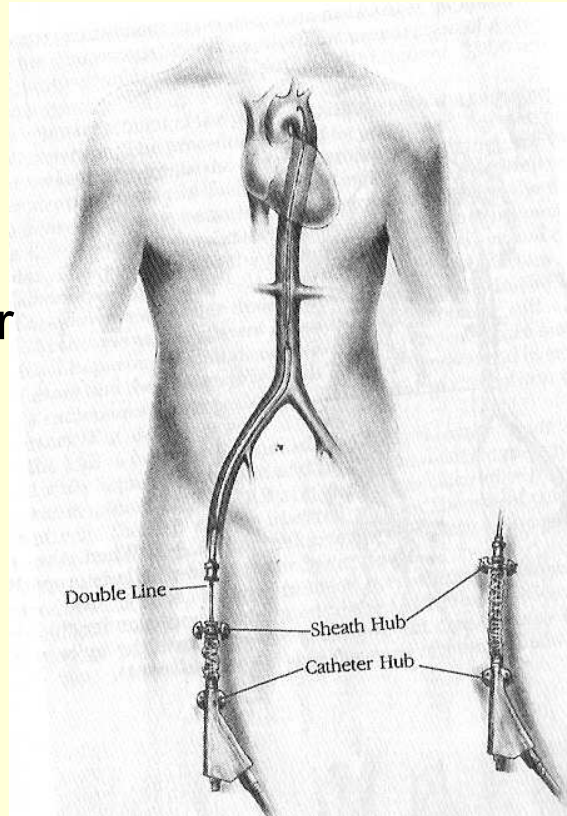
# Intra Aortic Balloon Pump (IABP) Counterpulsation

indication: cardiogenic shock  
in myocardial infarction

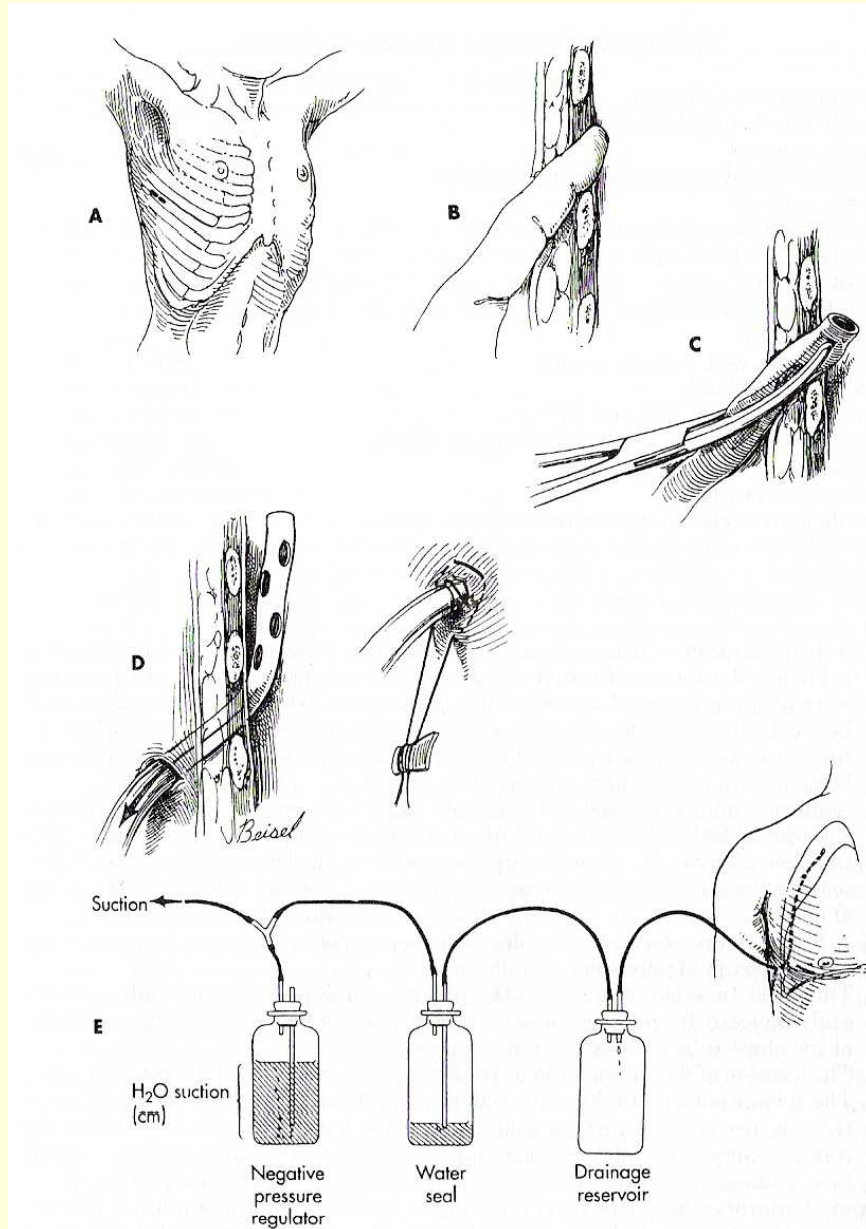
Baloon inflation in diastole,  
improves coronary perfusion

deflation during systole:  
lowering of systemic vascular  
resistance

ECG triggering



# Chest tube



Indication:

pneumothorax

haemothorax

empyema, pleural effusion

How to distinguish between  
transudate and exsudate  
(empyema) ?

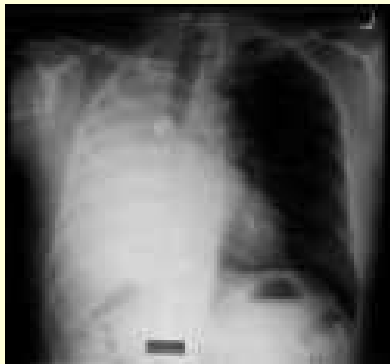
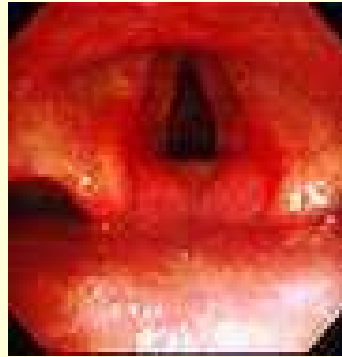
Transudate : Pleur/Plasma ratio

proteins < 0.5

LD < 0.6

Cholesterol < 0,3.

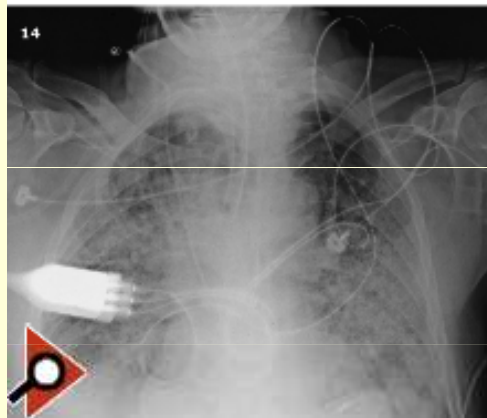
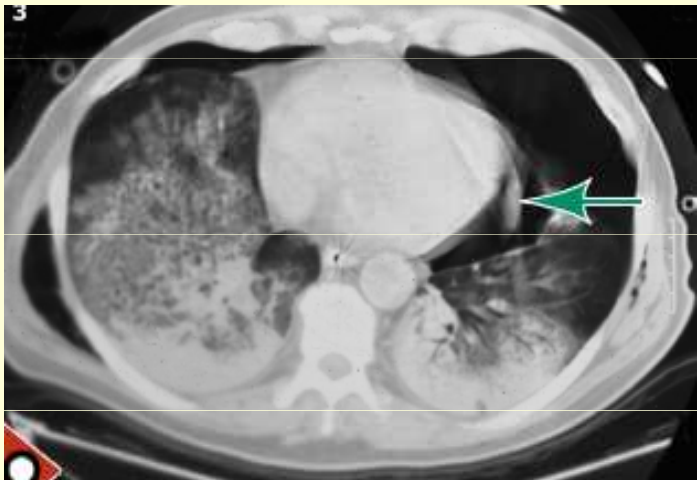
# Bronchoscopy



Indication: atelectasis, smoke injury  
ventilator pneumonia, hemoptysis,  
foreign bodies

BAL: bronchoalveolar lavage 150-400  
ml of fluid, quantitative culture on BAL  
fluid

# Imaging



# percutaneous tracheostomy

