

Restorative dentistry – aesthetics I.

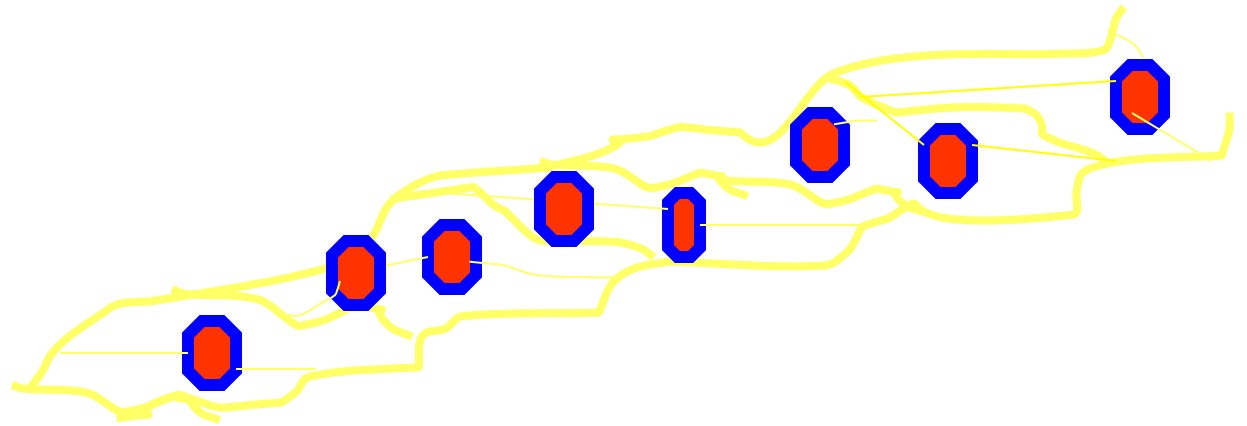
Contemporary trends

- Minimally invasive approach
- Adhesive materials and techniques
 - direct composite restorations
 - indirect composite and ceramic restorations
(luted adhesively)

Composites

Composites in dentistry

Chemically bonded mixture of organic matrix and inorganic fillers



matrix – transfers mechanical loading on inorganic fillers, protects the filler against moisture

filler - support of the material, carries the loading

coupling agents- enable the homogenous distribution of the filler in matrix

Composition - matrix

- Bis GMA – Bowen's monomer
- (2,2-bis[4-(2hydroxy-3-metakryloyloxypropoxy)fenyl]propan)
- Bis DMA
- UDMA
- TEGMA /triethylenglykoldimethacrylate
- EGMA ethylenglykoldimethacrylate
- eBis –GMA
- HDMA hexandioldimethacrylate

Composition - matrix

- Acid modified resins (compomers)
- Polysiloxa chains with polymerizable groups (ormocers)
- Silorans (ring opening monomers)

Filler

- Milled quartz
- Aluminium silicate glass
- Silicon dioxide
- Prepolymer
- Complexes of microfiller (agglomerates)

Macrofiller

- Particles μm or tenths of μm

Good mechanical resistance , abrasion resistance, bad polishability.

Microfiller

- Silicium dioxide (pyrogenous)
- Particles hundreths μm

Less amount of filler. Lower mechanical resistance, good polishability.

Microfiller in complex particles

- Prepolymer
- Agglomerates
- Higher amount of filler, good mechanical resistance, good polishability

Nanoparticles

- Particles 10 nm and less
- Special technology, size, shape and binding to monomer

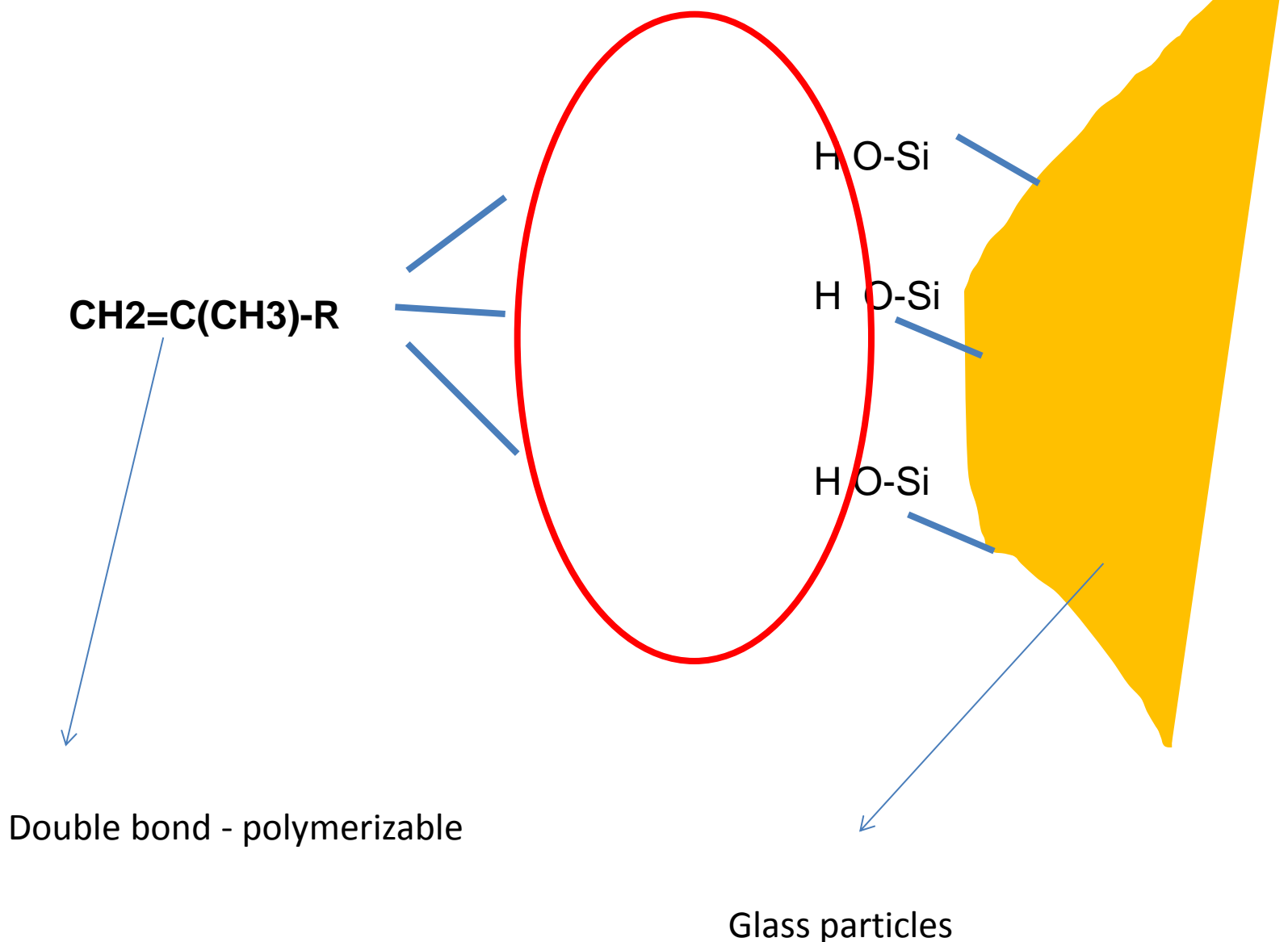
Hybrid filler

- Macro particles + microparticles
- Macro particles + microparticles+ prepolymer
- macroparticles + microparticles + prepolymer + nanoparticles

Coupling agent

G -methacryloxypropyltrimetoxysilan
(A 174)

Binding of the coupling agents to glass particles



Activator and initiator
Pigments
Fluorescents
Absorbers of light
Inhibitors

Polymerization

- Selfcuring composites

Dibenzoylperoxide – tertiary amin

Low colour stability

Light curing composites

- Initiator and sometimes also activator
 - Camphorchinon CQ
 - Phenylpropandion PPP
 - Trimethylbenzoylphosphinoxid TPO

Camphorquinon - CQ

Yellow colour

Activator: etyl-4-(N,N'-dimetylamino)benzoát (4EDMAB), N,N'-dimetylaminoethylmetakrylát (DMAEMA)

Light shades of composites: combination of CQ and other initiators.

Spectrum of absorption

Fotoiniciator	Spectrum o absorption (nm)	Maximum (nm)
CQ	440 - 500	470
PPD	380 – 430	400
TPO	350 - 410	380

Bonding



Adhesion

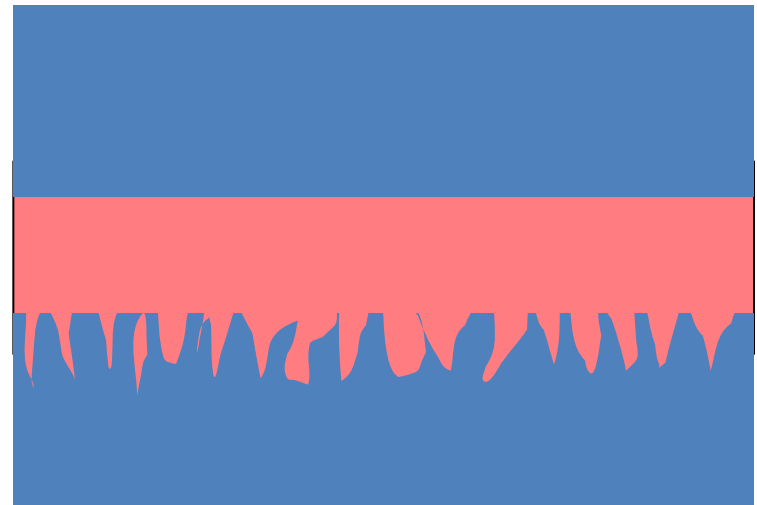
➤ **Mechanic**

➤ **Specific**

Adhesion

Mechanic

Irregularities of the surface



Adhesion

➤ **Specific**

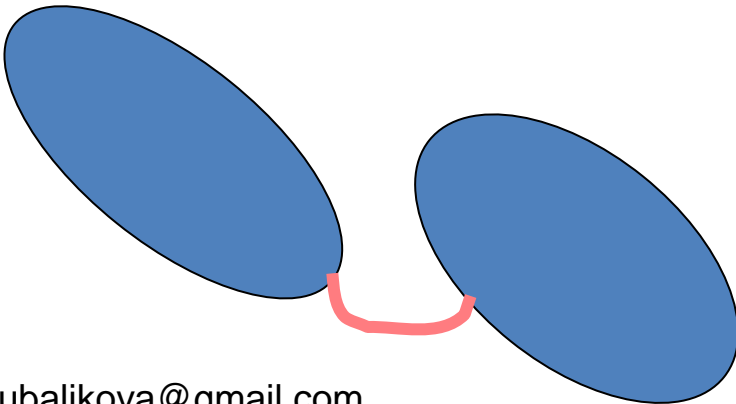
Physical

Chemical

Adhesion

➤ Specific

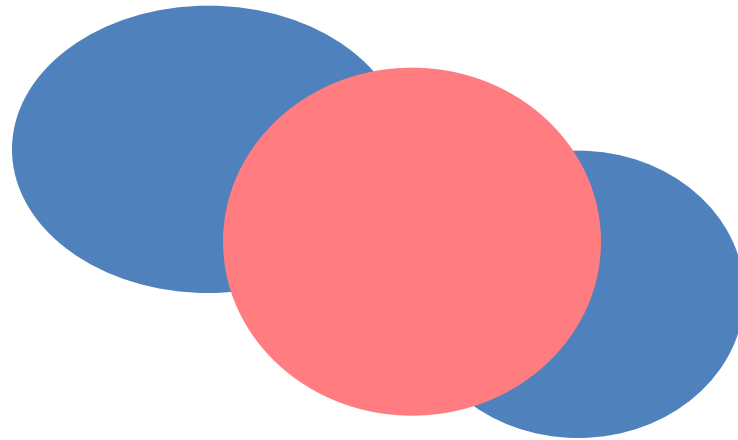
Physical – intermolecular forces - Van der Waals, hydrogenium bridges



Adhesion

➤ **Specific**

Chemical



Adhesion

- **Sandblasting**
- **Electrolytic**
- **Silanization**
- **Plazma coating**
- **Silanization**

Adhesive preparation of surfaces

- Creates irregularities
- Increases surface energy

Adhesion of dental materials

Composites - micromechanical

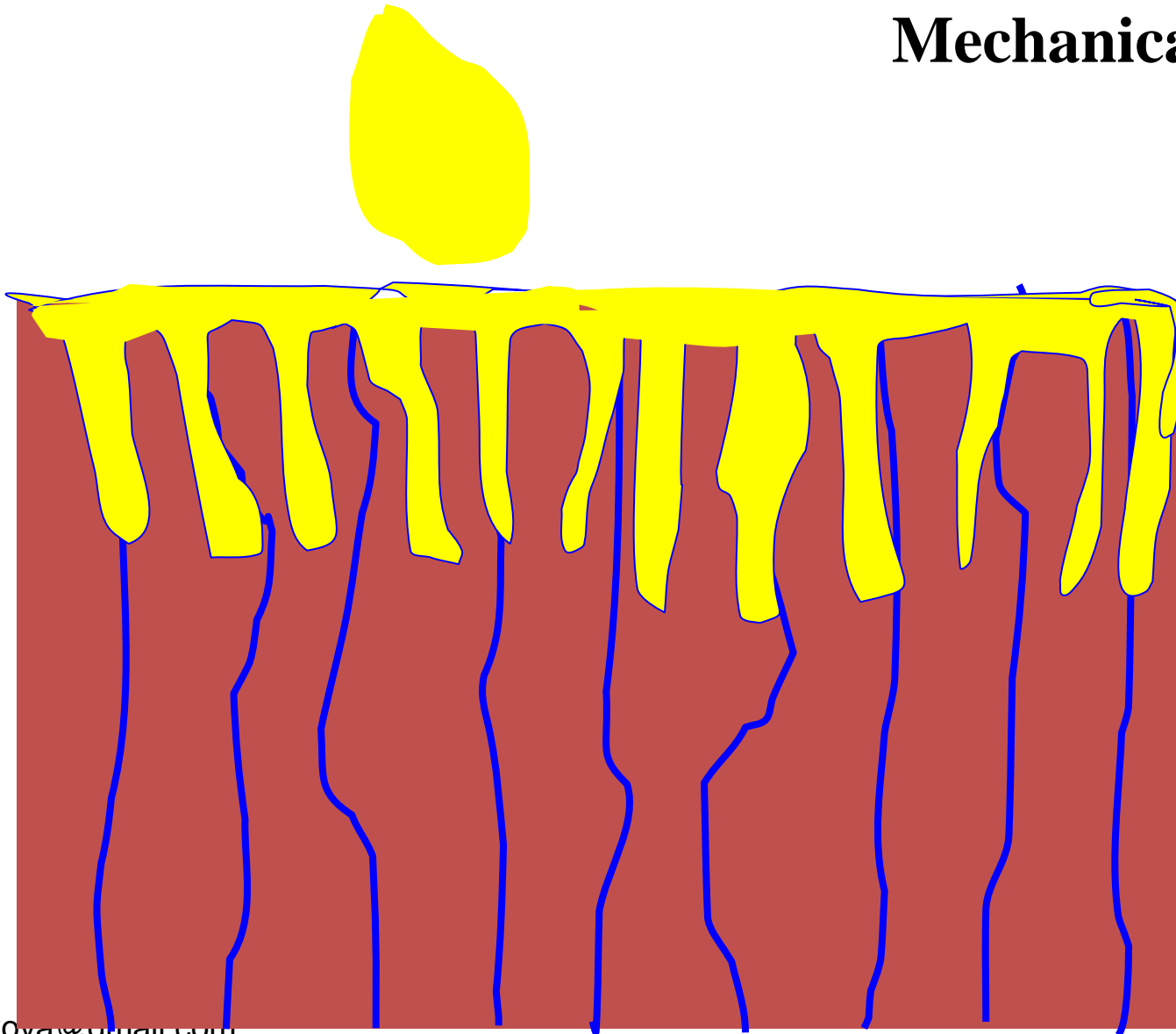
Adhesives – micromechanical, specific

Glassionomers - specific

Composits

Connection to enamel

Mechanical



Adhesion to dentin

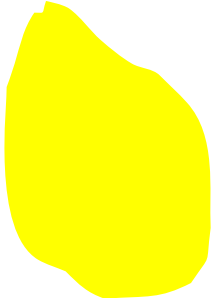
Dentin:

- o More water and organic substances in comp to enamel
- o Low surface energy
- o Tubular liquid
- o Connection with pulp chamber
- o Smear layer

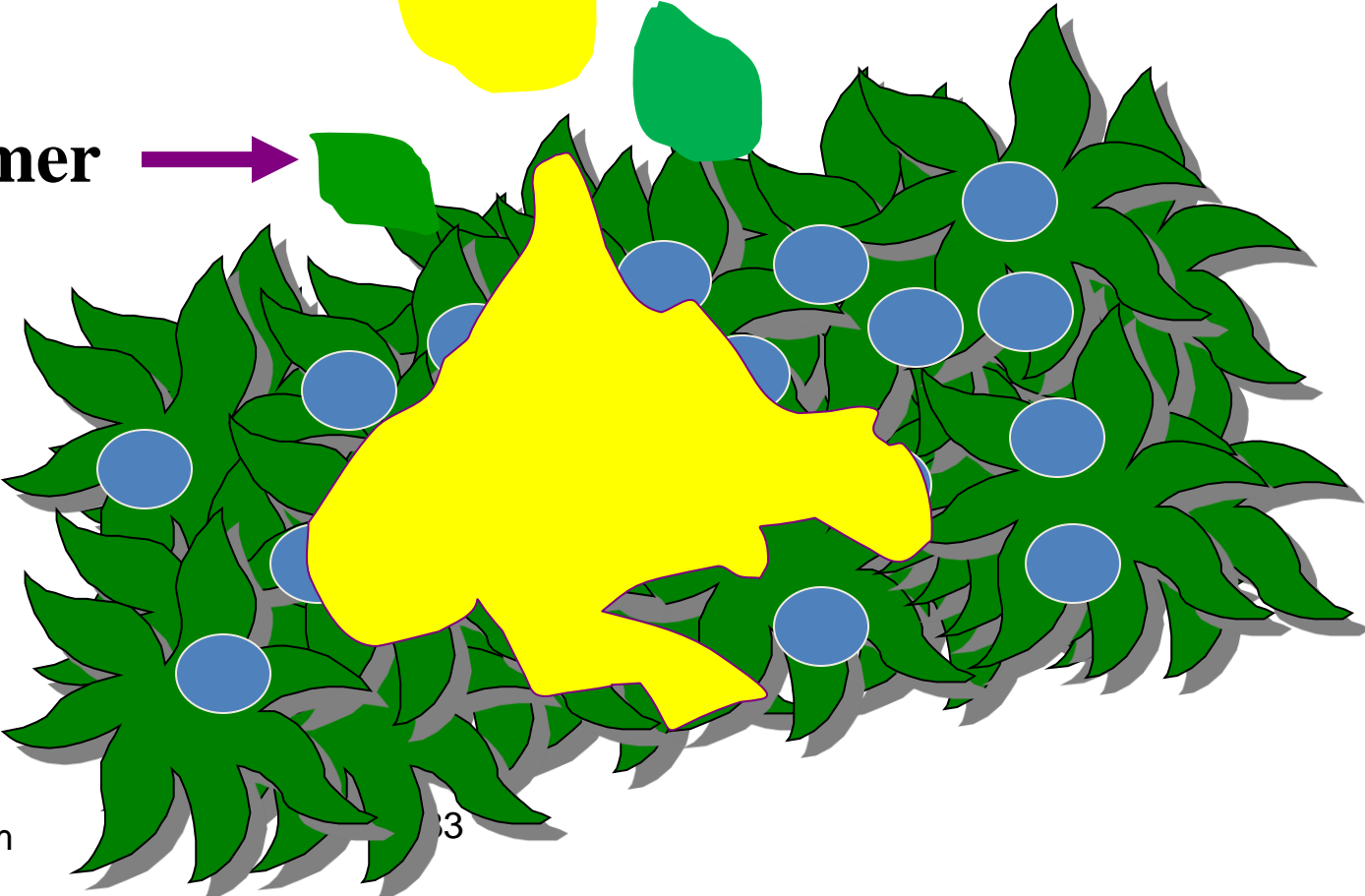
Connection to dentin

Mechanical

bond



primer



BONDING AGENTS



Generations

1st Generation: (1956)

- Glycerophosphoric acid
- DMA Resin
- Resin to tooth
- No longer used (poor clinical results: 1-3 MPa)

2nd Generation: (1970's)

- Unfilled Resin
- Bis-GMA or HEMA
- Ionic bond to calcium
- No longer used (weak bond strength, microleakage)

Generations

3rd Generation: (1980's)

- Etch + Hydrophilic Primer + Unfilled Resin
- Partial removal and/or Modification of smear layer
- Resin did not penetrate through smear layer

4th Generation: (1982)

- Total Etch (Phosphoric Acid) + Primer + Adhesive
- Complete removal of smear layer
- “Wet bonding” (risk of being too wet or dry)
- Formation of hybrid layer and resin tags
- Good clinical results

Generations

5th Generation: (1990's)

- Total Etch + Adhesive
- Hydrophilic monomers
- Formation of hybrid layer and resin tags

6th Generation: (late 1990's – 2000)

- Self-etching primer + Hydrophobic adhesive
- Partial removal of smear layer
- Chemical instability of primer

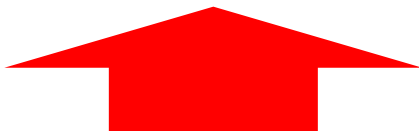
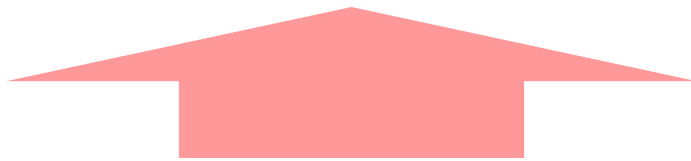
7th Generation: (2000's)

- One bottle
- Partial removal of smear layer
- Chemical instability

Adhesive materials



- Connection without gap
- Less invasive preparation
- Less risk of secondary caries
- Higher resistance
- No problem with mercury



Amalgam



Adhesive systems contain resin monomers

- 4-META
- HEMA
- TEGMA
- PENTA P
- 5-NMSA
- Bis-GMA

Adhesive systems contain resin monomers

- Hydrophobic monomers - bond

Works in enamel

Does not work in dentin without primer

- Amphiphilic monomers – hydrophobic + hydrophilic part

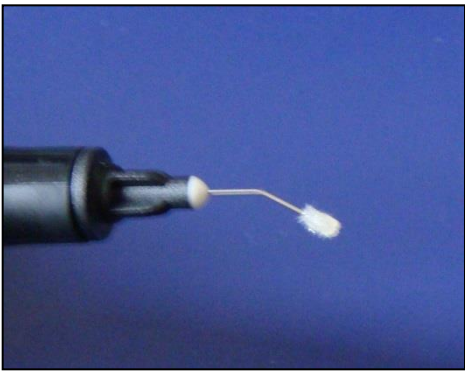
- primer

Primer is necessary for dentin

If applied on enamel – residual of water can be removed

Dissolving agents

- Aceton
- Alcohol
- Water
- Watwer/alcohol



Adhesives – classification acc to clinical steps

3- ERA

Acid etching	Washing	Priming	Bonding
Acid etching	Washing	Priming a bonding	
Selfetching priming			Bonding
Selfetching bonding			

2- ERA

2- SEA

1- SEA

Adhesives

- Acid etching technique
- Selfetching adhesive systems

Adhesives

- Acid etching technique

Etching

Washing

Priming Bonding

Adhesives

- Selfetching adhesive systems

Priming

Bonding

Less bonding strength in comparison to acid etching technique

Smaer layer infiltrated with adhesive systém

Dentin tag

Surface of dentin

Smear layer

Lateral tubul
Dentin tubul

Very mild acidic SEA Mild acidic sea Middle acidic SEA Strong acidic SEA

pH primeru

$\geq 2,5$

≈ 2

$\approx 1,5$

< 1

$\approx 100-300 \text{ nm}$

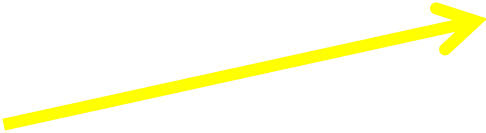
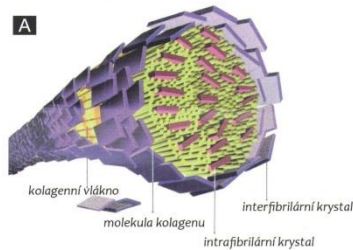
$\approx 1 \mu\text{m}$

$1-2 \mu\text{m}$

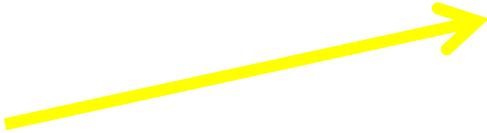
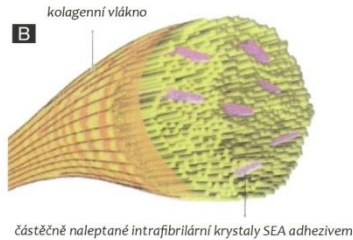
$3-5 \mu\text{m}$

Thickness of hybrid layer

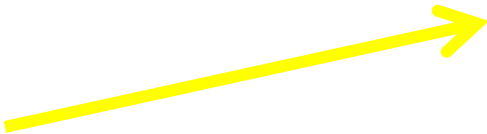
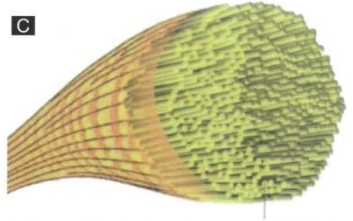
Dudek M. Adhezivní spoj
a adhezivní systémy I. LKS 11/2013



Collagen fibers with mineral salts
(peri and intra)



Collagen fibers with
intrafibrillar crystals



Collagen fibers without any crystals

(risk of hydrolysis and degradation with metalloproteinasis)

Dudek M. Adhezivní spoj
a adhezivní systémy I. LKS 11/2013

Adhesives

- Active and passive bonding

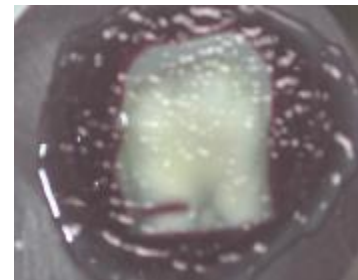
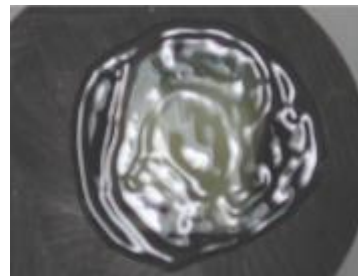
Active – rubbing with microbrush (SEA, dentin)

Passive – without any rubbing (ERA, enamel)

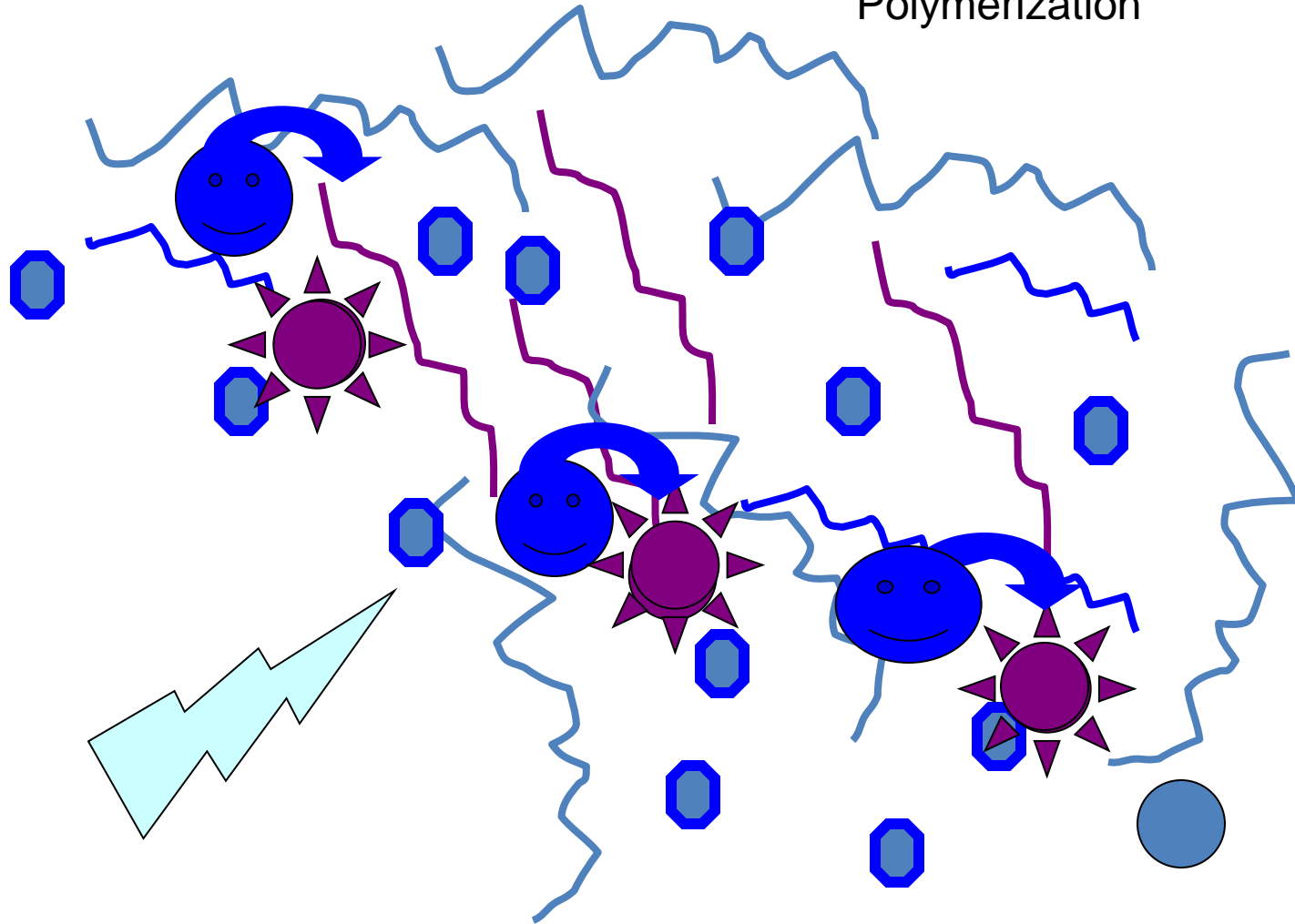
Blow up

Too thin layer does not allow complete conversion (polymerization) due to oxygen

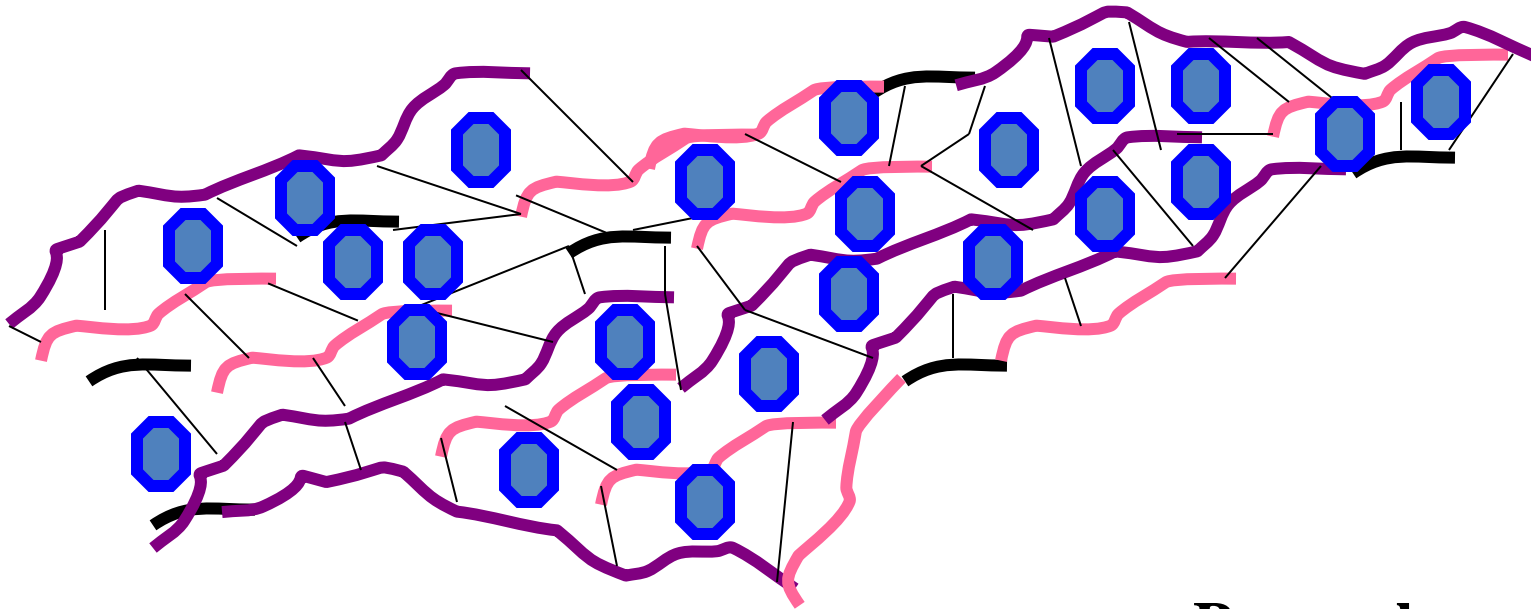
Too thick layer can contain still dissolving agent.



Polymerization



Polymer network

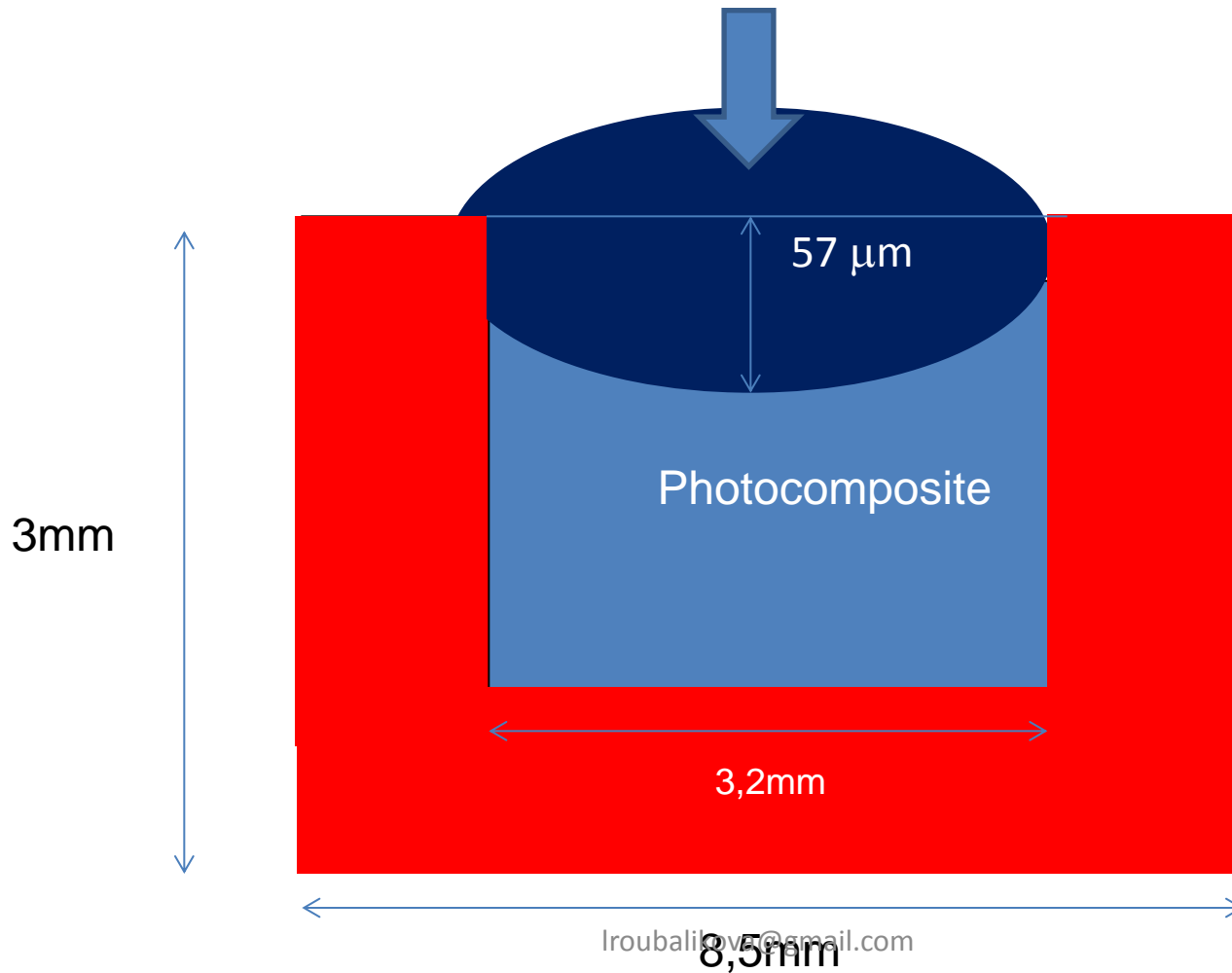


Pre -gel
Gel
Post -gel

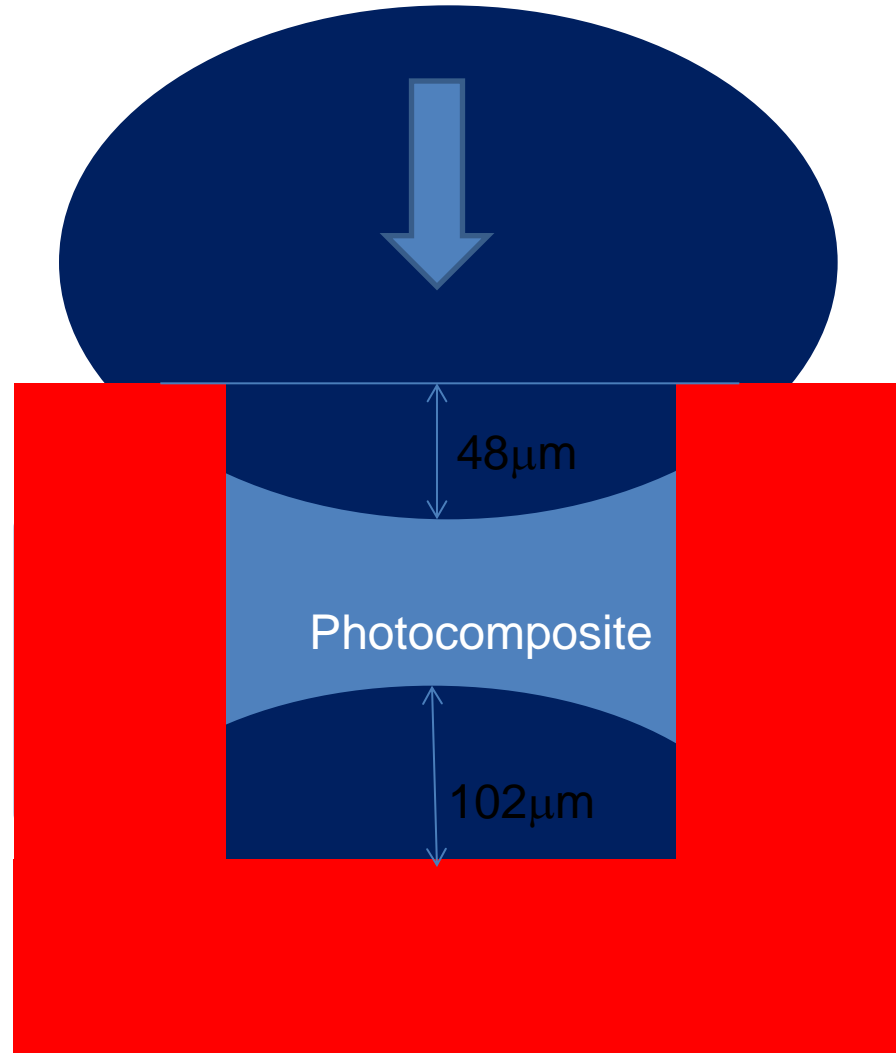
Byung, Suh

Principles of adhesion dentistry

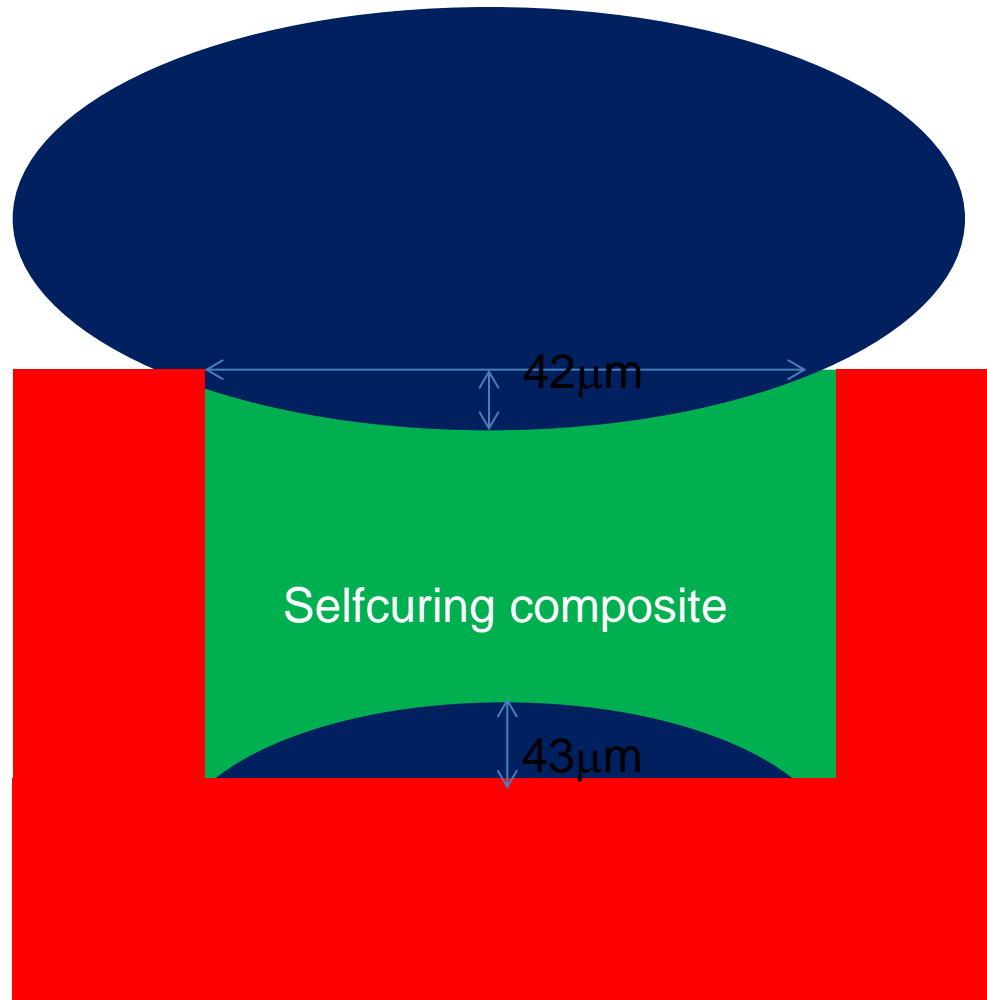
2013



Polymerization strain and stress



Polymerization strain and stress



Pregel phase

- The material is still soft, forces of polymerization shrinkage can release, deformation of the surface

Gel point

- Material became hard

Postgel phase

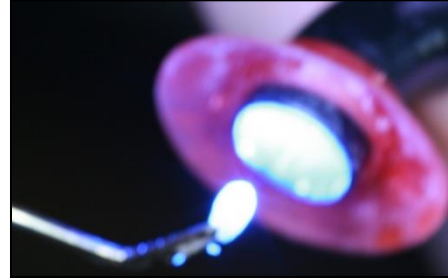
- Material is not soft, polymerization continues, due to polymerization shrinkage the polymerization stress occurs – forces on bonded surfaces can cause gaps or cracks.

- Selfcuring composites – longer pregel phase
- Photocomposites – Gel point comes earlier

Forces of polymerization shrinkage depend on

- Composite material (content of filler)
- Geometry of the cavity (C-factor)
- Placement of the composite
- Mode of polymerization

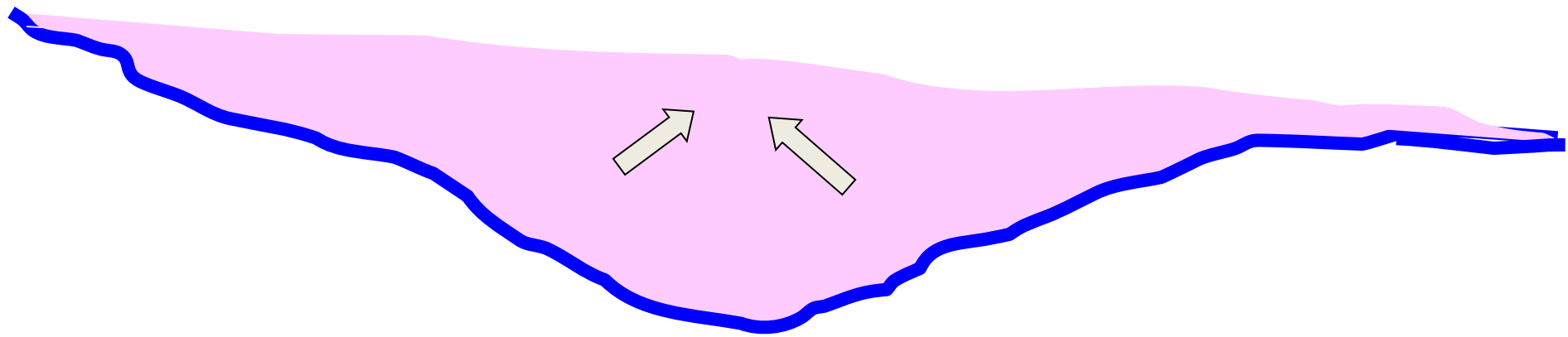
Light curing



Too short illumination, low output energy causes lower conversion of material, the risk of fracture is higher, material changes the colour.

C – factor (Configuration factor)

Surface of adhesion/free surface of the filling



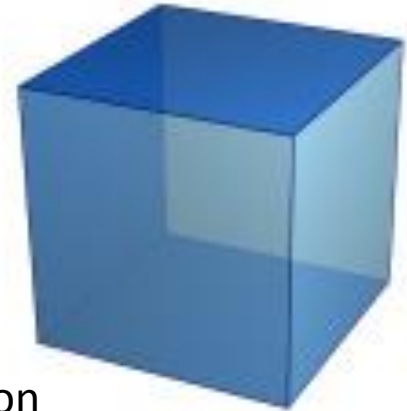
1/1 and less is optimal



5



2



1

Big free surface – lower polymerization stress,
The forces can easier release through free surface – its deformation

Forces of polymerization shrinkage
depend on

- Composite material (content of filler)

Higher content of filler - lower
shrinkage, higher polymerization stress.

Mode of polymerization

- Longer pre gel phase – releasing of polymerization forces

Continual polymerization

Min. 500 mW/cm² 40 s



2 step polymerization

10 s cca 140 mW/cm²  *750 mW/cm² 30 s*



Soft start

*Continuos increasing to 750 mW/cm²
during 10 s and polymerization 30s*

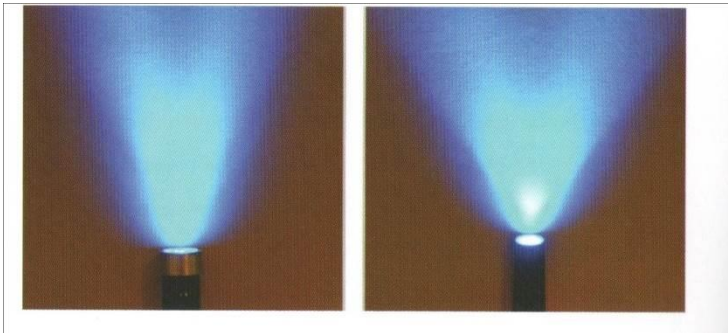


step polymerization with interruption

100 – 300 mW/cm² 3-5 s, přerušení na 3 min, pak polymerovat 750 mW/cm² po 30 s



Period of pre gel phase



Dentists polymerize mostly from the distance 4 mm – 10 mm. Diffusion of light. Soft start programm is not necessary.

Polymerization units

Quartz tungsten
halogen (QTH)

- Light emitting diode (LED)
- Plasma – arc (PAC)

- Energy and spectrum

Polymerization units

Quartz tungsten
halogen (QTH)

600 -800 mW/cm²

- Light emitting
diode (LED)

1000 -1800 mW/cm²
modré

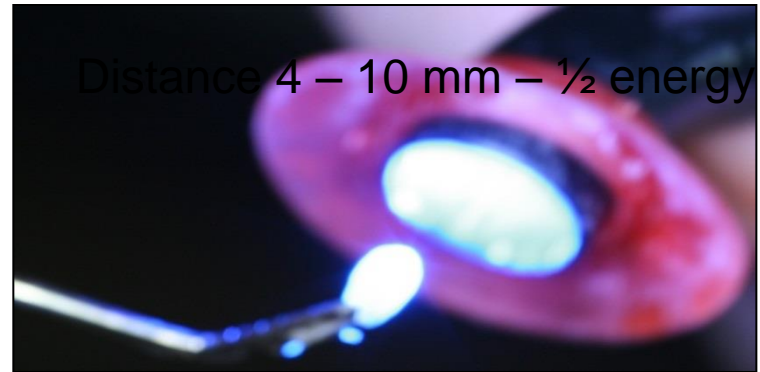
50 – 100 mW/cm²
fialové

- Plasma – arc (PAC)

1500 - 2000mW/cm²

Time of polymerization

- Recommended output power 12000 – 16000 mJ/cm²

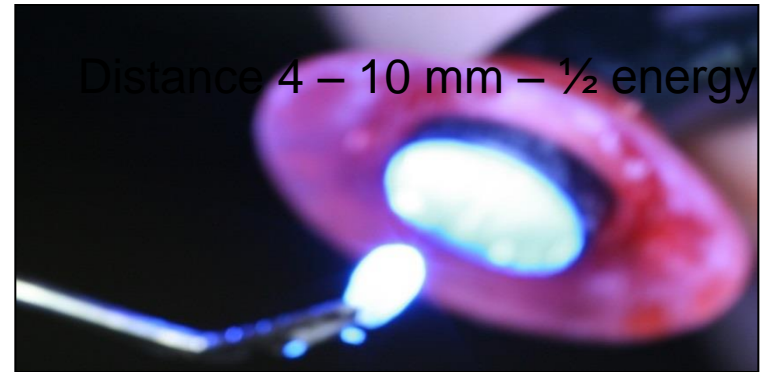


$$\frac{12\,000 \text{ mWs/cm}^2}{\text{Intensity mW/cm}^2} =$$

Time of polymerization in s

Time of polymerization

- Recommended output power 12000 – 16000 mJ/cm²



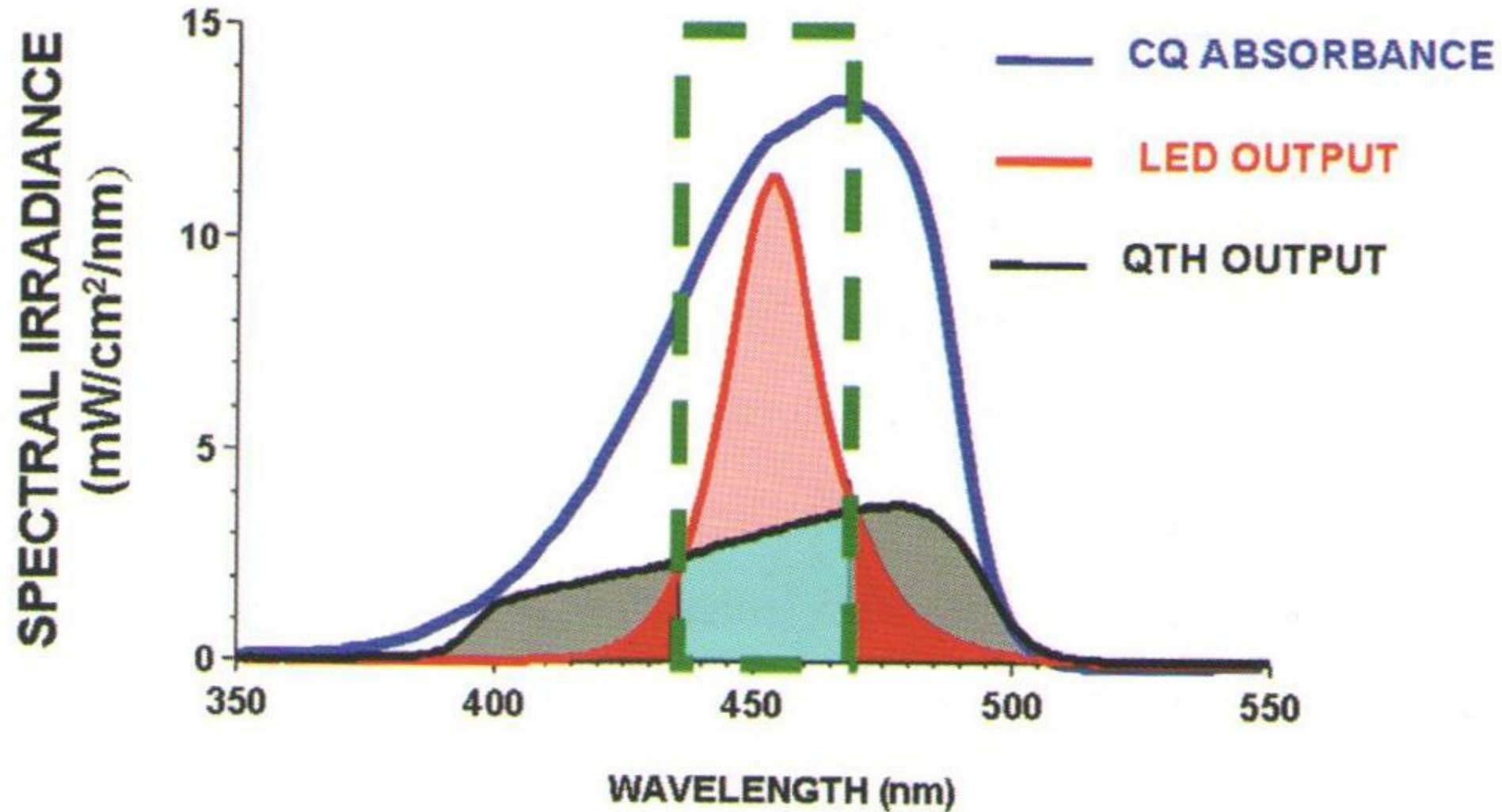
$$\frac{12\,000 \text{ mWs/cm}^2}{\text{Intensity mW/cm}^2} =$$

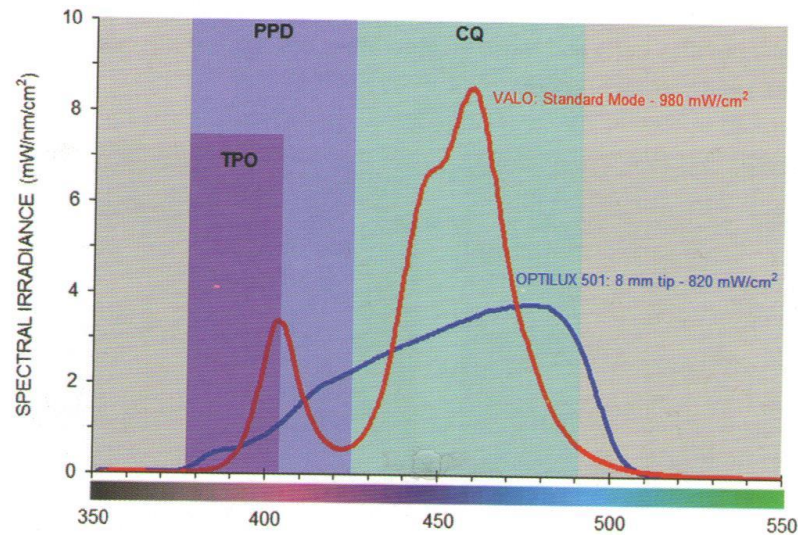
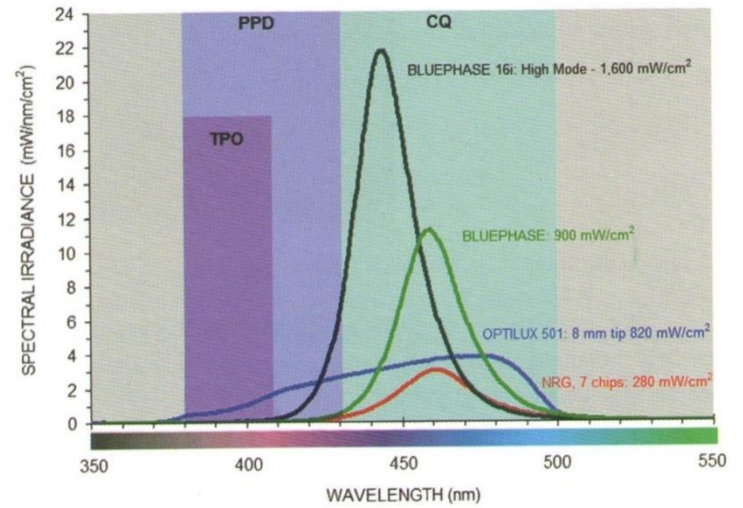
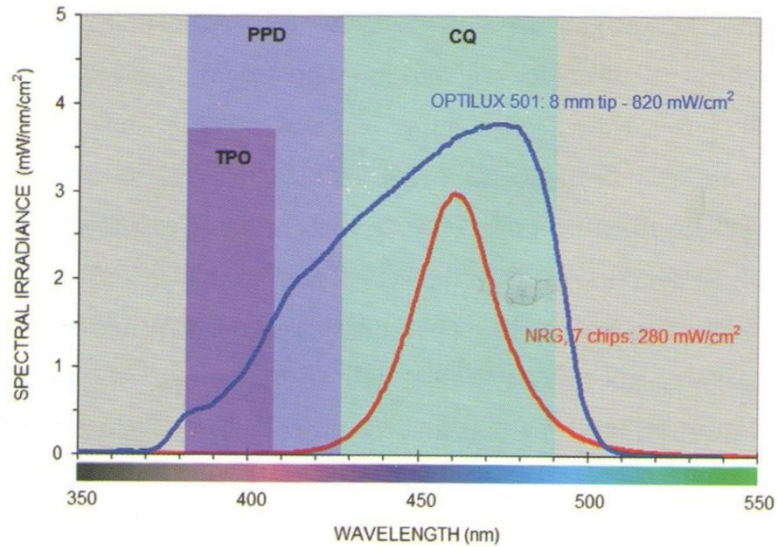
Time of polymerization in s

Photoinitiators – spectrum of absorption

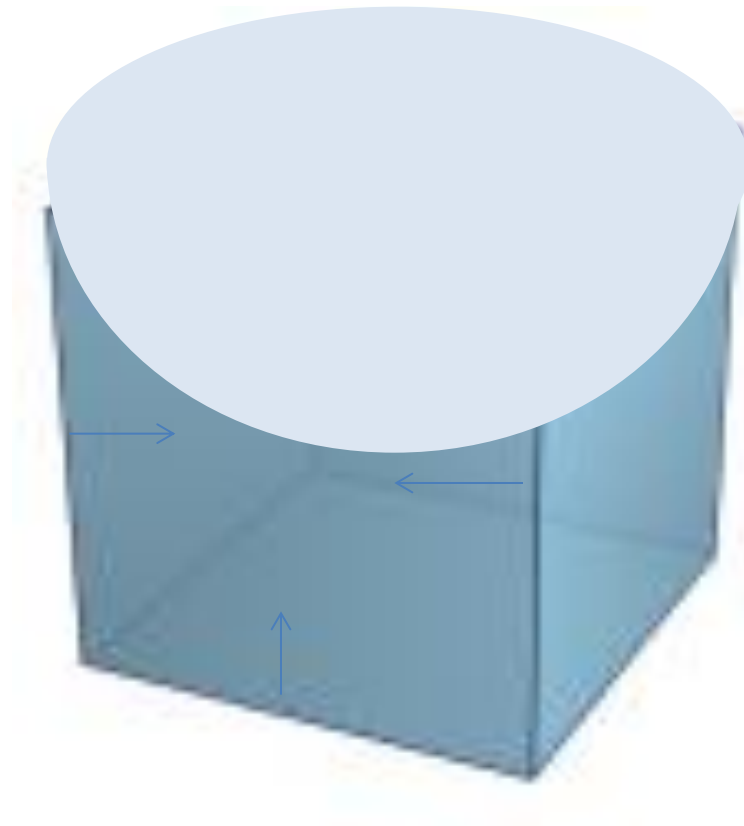
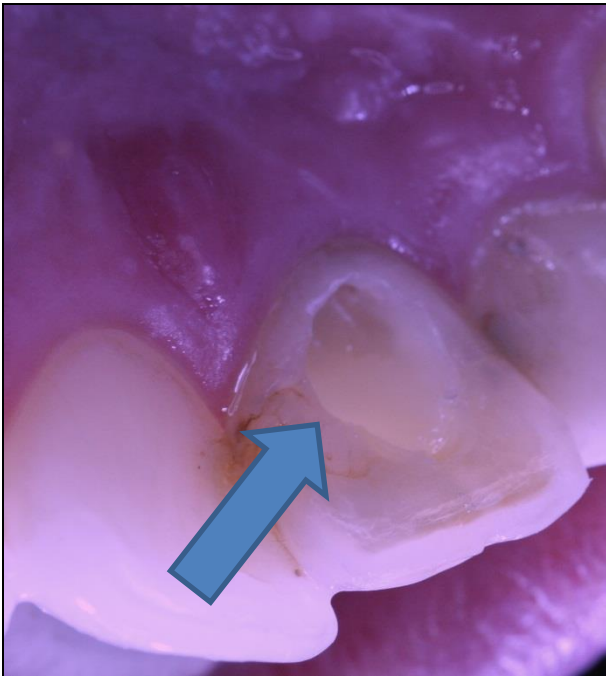
Photoinitiator	Spectrum of absorption (nm)	Maximum (nm)
CQ	440 - 500	470
PPD	380 – 430	400
TPO	350 - 410	380

ABSORPTIVE REGION THAN FROM QTH LIGHT





High C- faktor



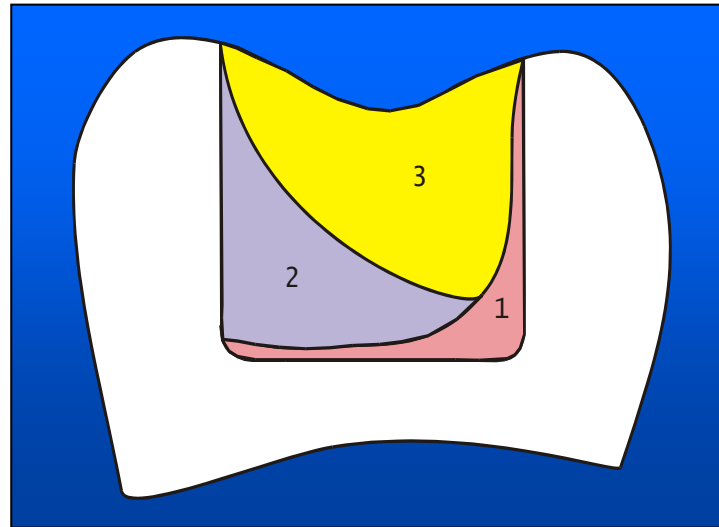
GAP in dentin, cracks in enamel



Working with layers

- Incremental technique

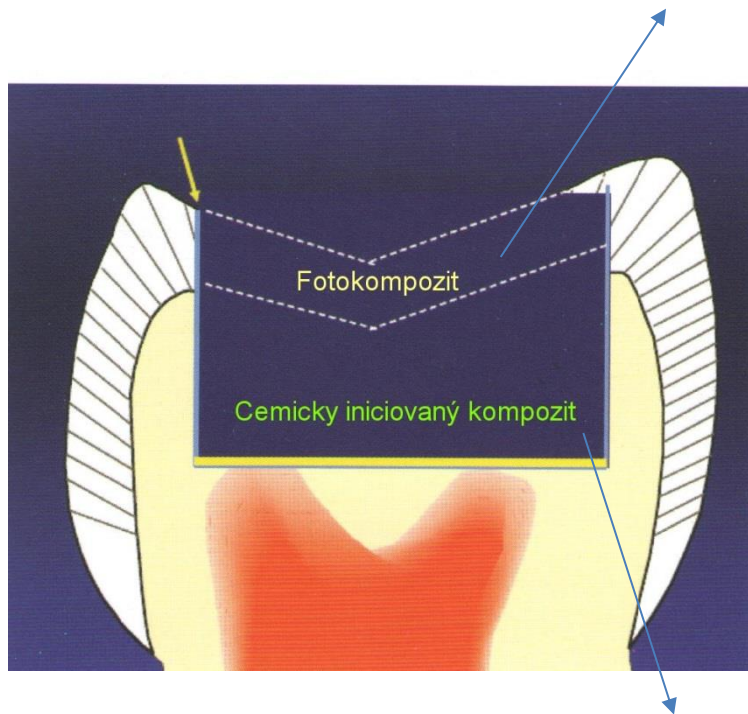
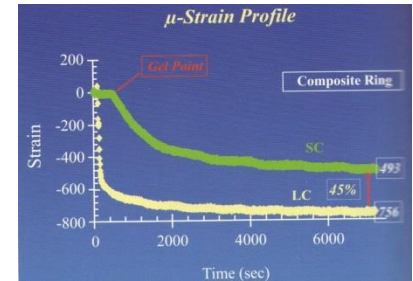
Flowable +thin layers



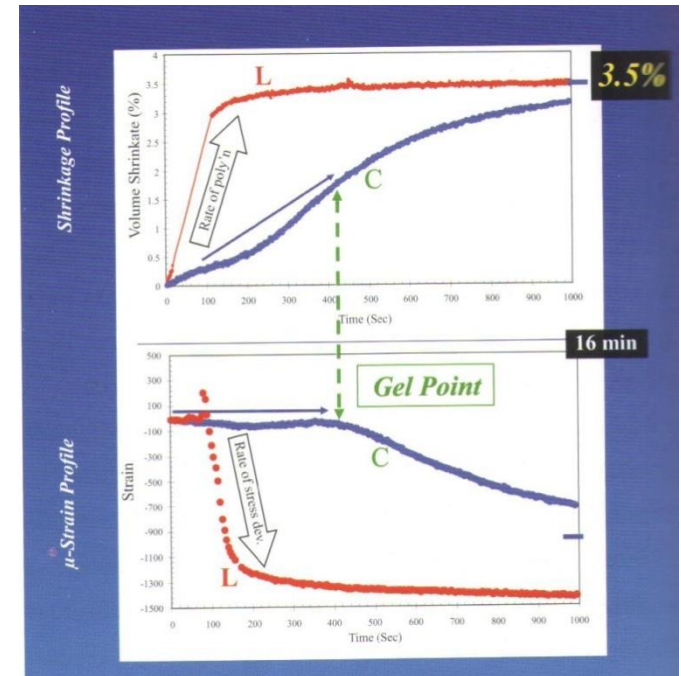
Free surface – as big as possible in each layer

COMBINATION OF MATERIALS

Light curing composite



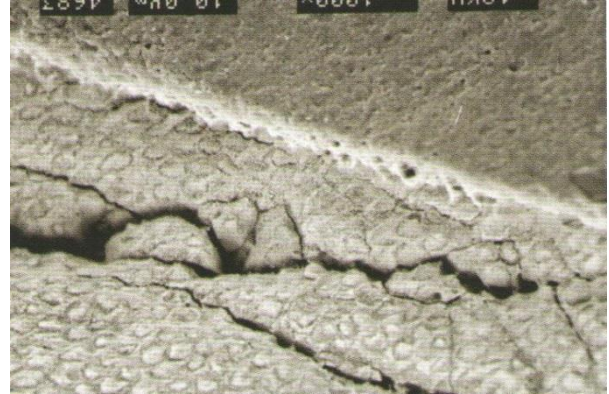
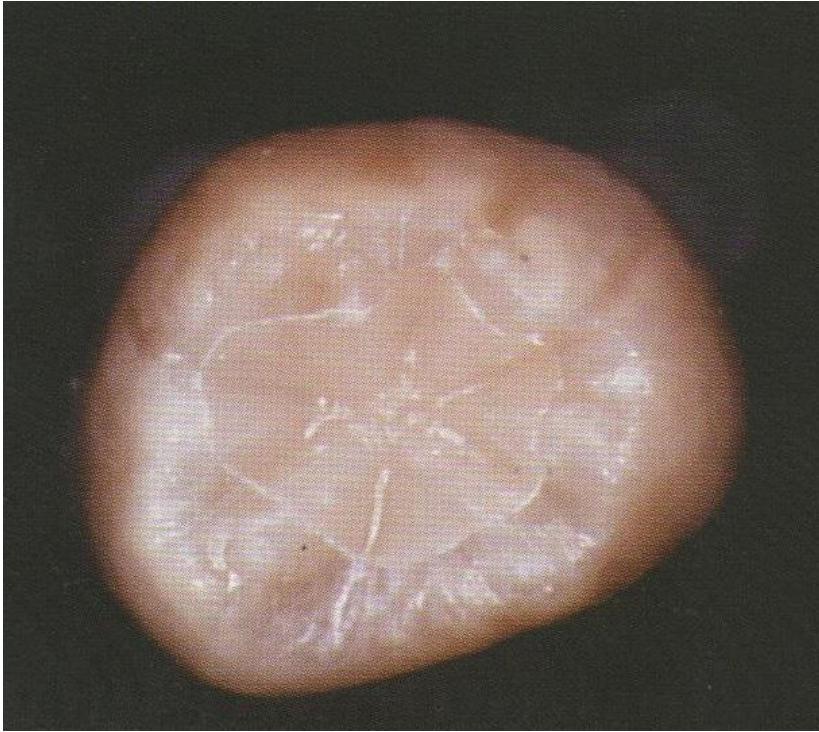
Selfcuring composite



Glassionomer and composite

Sandwich technique
GIC replaces lost dentin
Composite replaces lost
enamel

Sealing of the filling



FLOWABLES

Flowables indications

Filling of minicavities, pit and fissure sealing, tunnel

- Reparations
- Splinting
- Marginal adaptation
- Treatment of infractions
- Block out of undercuts

Flowables

- Less amount of filler
 - Higher polymerization shrinkage
 - Lower modulus of elasticity
 - Lower polymerization stress
1. *generation nízký obsah plniva, malá mechanická odolnost*
 2. *generation: nanoparticles –higher amount of filler*

Flowables indications

Filling of minicavities, pit and fissure sealing, tunnel

- Reparations
- Splinting
- Marginal adaptation
- Treatment of infractions
- Block out of undercuts

Bulk fill

Application in bulks, deep polymerization (4 – 5mm)

Heterogenous group

1. Flowables – SDR Flow (Dentsply), Venus Bulk Fill (Heraeus Kulzer), X-tra fill (VOCO), Filtek Bulk Fill (3M ESPE).
2. Packable composites (Tetric EvoCeram Bulk Fill (Ivoclar – Vivadent) a QuiXfill (Dentsply).
3. Sonic Fill (KaVo) – sonic activated composite

Sonic Fill



Sonic activation – materials become Flowable.

Bulk fill

- More translucent
- More photoinitiators
- Combination of materials (flowable bulk fill + hybrid composite)
- Packable composite + flowable on the bottom
- Sonic fill – combination with other materials is not necessary but useful.

Bulk fill

- The problem of polymerization stress is not solved completely.
- Thinner layers than 4 mm recommended.