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(doslovná kopie)

Chemical compound/Catalogs/Inorganic compounds

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A catalog relating to the topic of [Chemical compound](#).

This page aims to list well-known [inorganic compounds](#), including [organometallic](#) compounds, to stimulate the creation of Citizen.dium articles. [Organic compounds](#), [minerals](#) and [chemical elements](#) are not included on this list. There is also an alternative listing related to this page, [inorganic compounds by element](#) (presently under construction).

This list is not necessarily complete or up to date – if you see an article that should be here but isn't (or one that shouldn't be here but is), please update the page accordingly.

Relevant links for chemical compounds are:

- The [CAS Substance Databases](#), which contains information on about 23 million compounds.
- Chemfinder [\[1\]](#) is helpful for finding information about a chemical (disable and delete [cookies!](#)).
- ChemIDplus at <http://chem.sis.nlm.nih.gov/chemidplus/> is a useful non-commercial source for chemical lookups
- <http://physchem.ox.ac.uk/MSDS/> Material Data Safety Sheets, plus other relevant links.

These (commercial) links may also provide useful information:

- Sigma Aldrich [\[2\]](#)
- Acros Organics [\[3\]](#)
- Lancaster [\[4\]](#)

Whilst most compounds are referred to by their [IUPAC systematic names](#) (following [IUPAC nomenclature](#)), "traditional" names have also been kept where they are in wide use or of significant historical interest.

See also: [inorganic compounds by element](#), [list of compounds](#), [list of organic compounds](#), [organic compound](#), [list of biomolecules](#), [list of minerals](#), [polyatomic ions](#), [list of elements by](#)

[name](#), [List of alchemical substances](#).

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A

- [Aluminium antimonide](#) – AlSb
- [Aluminium arsenide](#) – AlAs
- [Aluminium nitride](#) – AlN
- [Aluminium oxide](#) – Al₂O₃
- [Aluminium phosphide](#) – AlP
- [Aluminium chloride](#) – AlCl₃
- [Aluminium fluoride](#) – AlF₃
- [Aluminium hydroxide](#) – Al(OH)₃
- [Aluminium nitrate](#) – Al(NO₃)₃
- [Aluminium sulfate](#) – Al₂(SO₄)₃
- [Ammonia](#) – NH₃
- [Ammonium acetate](#) – CH₃COONH₄
- [Ammonium bicarbonate](#) – NH₄HCO₃
- [Ammonium cerium\(IV\) nitrate](#) – (NH₄)₂Ce(NO₃)₆
- [Ammonium chloride](#) – NH₄Cl
- [Ammonium hydroxide](#) – NH₄OH
- [Ammonium nitrate](#) – NH₄NO₃
- [Ammonium perchlorate](#)
- [Ammonium sulfate](#) – (NH₄)₂SO₄
- [Ammonium tetrathiocyanatodiamminechromate\(III\)](#) – NH₄[Cr(SCN)₄(NH₃)₂]
- [Antimony\(III\) acetate](#)
- [Antimony hydride](#) – SbH₃
- [Antimony pentachloride](#) – SbCl₅
- [Antimony pentafluoride](#) – SbF₅
- [Antimony trioxide](#) – Sb₂O₃
- [Arsine](#) – AsH₃
- [Arsenic trioxide](#) (Arsenic(III) oxide) – As₂O₃ easily confused with arsenous acid As(OH)₃

B

- [Barium carbonate](#) – BaCO₃
- [Barium chloride](#) – BaCl₂
- [Barium hydroxide](#) – Ba(OH)₂
- [Barium iodide](#) – BaI₂
- [Barium nitrate](#) – Ba(NO₃)₂
- [Barium sulfate](#) – BaSO₄
- [Barium titanate](#) – BaTiO₃
- [Beryllium hydroxide](#) – Be(OH)₂
- [Beryllium oxide](#) – BeO
- [Bismuth\(III\) oxide](#) Bi₂O₃
- [Bismuth\(III\) telluride](#) Bi₂Te₃
- [Bismuth subsalicylate](#)
- [Borane](#)

- Borax – $\text{Na}_2\text{B}_4\text{O}_7 \cdot 10\text{H}_2\text{O}$
- Boric acid – H_3BO_3
- Boron carbide – B_4C
- Boron nitride – BN
- Boron oxide – B_2O_3
- Boron trichloride – BCl_3
- Boron trifluoride – BF_3
- Bromine pentafluoride – BrF_5
- Bromine trifluoride – BrF_3
- n-Butyllithium – $\text{C}_4\text{H}_9\text{Li}$
- sec-Butyllithium
- tert-Butyllithium

C

- Cacodylic acid – $(\text{CH}_3)_2\text{AsO}_2\text{H}$
- Cadmium arsenide – Cd_3As_2
- Cadmium bromide – CdBr_2
- Cadmium chloride – CdCl_2
- Cadmium fluoride – CdF_2
- Cadmium iodide – CdI_2
- Cadmium nitrate – $\text{Cd}(\text{NO}_3)_2$
- Cadmium selenide – CdSe (of quantum dot fame)
- Cadmium sulfate – CdSO_4
- Cadmium telluride – CdTe
- Caesium bicarbonate – CsHCO_3
- Caesium carbonate – Cs_2CO_3
- Caesium chromate – Cs_2CrO_4
- Caesium chloride – CsCl
- Caesium fluoride – CsF
- Calcium carbide – CaC_2
- Calcium carbonate – CaCO_3
- Calcium chloride – CaCl_2
- Calcium fluoride – CaF_2
- Calcium hydride – CaH_2
- Calcium hydroxide – $\text{Ca}(\text{OH})_2$
- Calcium sulfate (redirect to Gypsum) – CaSO_4
- Carbon dioxide – CO_2
- Carbon monoxide – CO
- Carbonic acid – H_2CO_3
- Carbon tetrabromide – CBr_4
- Carbon tetrachloride – CCl_4
- Carbon tetraiodide – CI_4
- Carbonyl fluoride – COF_2
- Carboplatin – $\text{C}_6\text{H}_{12}\text{N}_2\text{O}_4\text{Pt}$
- Cerium(III) chloride – CeCl_3
- Cerium(IV) sulfate – $\text{Ce}(\text{SO}_4)_2$
- Chromic acid – CrO_3
- Chromium(II) acetate – $\text{Cr}_2(\text{OAc})_4$
- Chromium(III) chloride – CrCl_3

- Chromium(II) chloride – CrCl₂ (also chromous chloride)
- Chromium(III) oxide – Cr₂O₃
- Chromium(IV) oxide – CrO₂
- Chromium(VI) oxide (redirect to Chromic acid) – CrO₃
- Chromyl chloride – CrO₂Cl₂
- Cisplatin (cis-platinum(II) chloride diammine] – PtCl₂.2NH₃
- Cobalamin (redirect to Vitamin B12)
- Copper(II) acetate – Cu₂(OAc)₄(H₂O)₂
- Cobalt(II) carbonate – CoCO₃
- Cobalt(II) chloride – CoCl₂
- Cobalt(II) oxide – CoO
- Cobalt(III) oxide – Co₂O₃
- Copper(II) carbonate – CuCO₃
- Copper(I) chloride – CuCl
- Copper(II) chloride – CuCl₂
- Copper(II) nitrate – Cu(NO₃)₂
- Copper(I) oxide – Cu₂O
- Copper(II) oxide – CuO
- Copper(II) sulfate – CuSO₄
- Copper(I) sulfide – Cu₂S
- Copper(II) sulfide – CuS
- Cyanogen – (CN)₂
- Cyanogen chloride – CNCl
- Cyanuric chloride – C₃H₃N₃
- Cyclopentadienyliron dicarbonyl dimer – (C₅H₅)₂Fe₂(CO)₄ ("Fp2")

D

- Decaborane (redirect to Diborane) – B₁₀H₁₄
- Diborane – B₂H₆
- Dichlorosilane – SiH₂Cl₂
- Dimethylmercury – CH₃HgCH₃
- Dinitrogen pentoxide – N₂O₅, nitrosyl nitrate
- Disilane – Si₂H₆
- Dysprosium(III) chloride – DyCl₃

E

- Europium(III) chloride – EuCl₃

F

- Fluorosulfuric acid – FSO₂(OH)

G

- Gadolinium(III) chloride – GdCl₃
- Gadolinium(III) oxide – Gd₂O₃

- Gallium antimonide – GaSb
- Gallium arsenide – GaAs
- Gallium(III) chloride – GaCl₃
- Gallium nitride – GaN
- Gallium phosphide – GaP
- Germanium(IV) hydride – GeH₄
- Germanium(III) hydride – Ge₂H₆
- Germanium(II) fluoride – GeF₂
- Germanium(IV) fluoride – GeF₄
- Germanium(II) chloride – GeCl₂
- Germanium(IV) chloride – GeCl₄
- Germanium(II) bromide – GeBr₂
- Germanium(IV) bromide – GeBr₄
- Germanium(II) iodide – GeI₂
- Germanium(IV) iodide – GeI₄
- Germanium(II) oxide – GeO
- Germanium(IV) oxide – GeO₂
- Germanium(II) sulfide – GeS
- Germanium(IV) sulfide – GeS₂
- Germanium(II) selenide – GeSe
- Germanium(IV) selenide – GeSe₂
- Germanium telluride – GeTe
- Germanium (IV) nitride – Ge₃N₄
- Gold(I) chloride – AuCl
- Gold(III) chloride – AuCl₃
- Gold(I,III) chloride – Au₄Cl₈
- Gold(III) chloride – (AuCl₃)₂
- Gold(III) fluoride – AuF₃
- Gold(V) fluoride – AuF₅
- Gold(I) bromide – AuBr
- Gold(III) bromide – (AuBr₃)₂
- Gold(I) iodide – AuI
- Gold(III) iodide – AuI₃
- Gold(III) oxide – Au₂O₃
- Gold(I) sulfide – Au₂S
- Gold(III) sulfide – Au₂S₃
- Gold(III) selenide – AuSe
- Gold(III) selenide – Au₂Se₃
- Gold ditelluride – AuTe₂

H

- Hafnium tetrachloride – HfCl₄
- Hydrazine – N₂H₄
- Hydrazoic acid – HN₃
- Hydrobromic acid – HBr
- Hydrochloric acid – HCl
- Hydroiodic acid – HI
- Hydrogen bromide – HBr
- Hydrogen chloride – HCl

- Hydrogen fluoride – HF
- Hydrogen peroxide – H₂O₂
- Hydrogen selenide – H₂Se
- Hydrogen sulfide – H₂S
- Hydrogen telluride – H₂Te
- Hydroxylamine
- Hypochlorous acid – HClO
- Hypophosphorous acid – H₃PO₂

I

- Indium antimonide – InSb
- Indium arsenide – InAs
- Indium(I) chloride
- Indium nitride – InN
- Indium phosphide – InP
- Iodic acid – HIO₃
- Iodine heptafluoride – IF₇
- Iodine pentafluoride – IF₅
- Iodine monochloride – ICl
- Iridium(IV) chloride
- Iron(II) chloride – FeCl₂ including hydrate
- Iron(III) chloride – FeCl₃
- Iron(II) oxide – FeO
- Iron(III) nitrate
- Iron(II,III) oxide – Fe₃O₄
- Iron(III) oxide – Fe₂O₃
- Iron-sulfur cluster
- Iron(III) thiocyanate

J

K

L

- Lanthanum carbonate
- Lead azide
- Lead(IV) acetate
- Lead(II) carbonate – Pb(CO₃)
- Lead(II) chloride – PbCl₂
- Lead(II) iodide – PbI₂
- Lead(II) nitrate – Pb(NO₃)₂
- Lead(II) oxide – PbO
- Lead(IV) oxide – PbO₂
- Lead(II) phosphate – Pb₃(PO₄)₂
- Lead(II) sulfate – Pb(SO₄)₂
- Lead(II) selenide – PbSe

- [Lead styphnate](#)
- [Lead\(II\) sulfide](#) – PbS
- [Lead\(II\) telluride](#) – PbTe
- [Lead zirconate titanate](#) – Pb[Ti_xZr_yO₃ (x = 0.52; a.k.a. [Lead zirconium titanate](#))
- [Lithium aluminium hydride](#) – LiAlH₄
- [Lithium bromide](#) – LiBr
- [Lithium carbonate](#) (redirect to [Lithium salt](#)) – Li₂CO₃
- [Lithium chloride](#) – LiCl
- [Lithium citrate](#) (redirect to [Lithium salt](#)) – Li₃C₆H₅O₇
- [Lithium diisopropylamide](#)
- [Lithium hydride](#) – LiH
- [Lithium hydroxide](#) – LiOH
- [Lithium iodide](#) – LiI
- [Lithium nitrate](#) – Li₃N
- [Lithium sulfate](#) – Li₂SO₄

M

- [Magnesium carbonate](#) – MgCO₃
- [Magnesium chloride](#) – MgCl₂
- [Magnesium oxide](#) – MgO
- [Magnesium phosphate](#) – Mg₃(PO₄)₂
- [Magnesium sulfate](#) – MgSO₄
- [Manganese\(IV\) oxide](#) – MnO₂, manganese dioxide
- [Manganese\(II\) acetate](#)
- [Manganese\(II\) sulfate monohydrate](#) – MnSO₄.H₂O
- [Manganese\(II\) chloride](#) – MnCl₂
- [Manganese\(III\) chloride](#) – MnCl₃
- [Manganese\(IV\) fluoride](#) – MnF₄
- [Manganese\(II\) phosphate](#)
- [Mercury\(I\) chloride](#) – Hg₂Cl₂
- [Mercury\(II\) chloride](#) – HgCl₂
- [Mercury fulminate](#) – Hg(ONC)₂
- [Mercury\(II\) selenide](#) – HgSe
- [Mercury\(II\) sulfide](#) – HgS
- [Mercury\(II\) telluride](#) – HgTe
- [Metaphosphoric acid](#)
- [Methylcyclopentadienyl Manganese Tricarbonyl](#) – (CH₃C₅H₄)Mn(CO)₃, "MMT"
- [Methylmercury](#) – CH₃Hg⁺ - NOT a compound
- [Methylmercury hydroxide](#)
- [Molybdate orange](#)
- [Molybdenum trioxide](#) – MoO₃
- [Molybdenum disulfide](#) – MoS₂
- [Molybdenum hexacarbonyl](#) – C₆O₆Mo
- [Molybdic acid](#) – H₂MoO₄

N

- [n-Butyllithium](#) – C₄H₉Li

- Neodymium(III) chloride – NdCl₃
- Nessler's reagent
- Nickel(II) carbonate – NiCO₃
- Nickel(II) chloride – NiCl₂ and hexahydrate
- Nickel(II) hydroxide – Ni(OH)₂
- Nickelocene – Ni(C₅H₅)₂
- Nickel(II) nitrate – Ni(NO₃)₂
- Nickel(II) oxide – NiO
- Niobium pentachloride – NbCl₅
- Nitric acid – HNO₃
- Nitric oxide – NO
- Nitrogen dioxide – NO₂
- Nitrosylsulphuric acid – NOHSO₄
- Nitrous oxide – N₂O

O

- Orthophosphoric acid – H₃PO₄
- Osmium tetroxide – OsO₄, osmium(VIII) oxide
- Osmium trioxide – OsO₃, osmium(VI) oxide
- Oxybis(tributyltin)
- Oxygen difluoride – OF₂
- Ozone – O₃

P

- Palladium(II) chloride
- Palladium(II) nitrate
- Pentaborane
- Pentasulfide antimony
- Perchloric acid – HClO₄
- Perchloryl fluoride
- Persulfuric acid (Caro's acid) – H₂SO₅
- Perxenic acid – H₄XeO₆
- Petasis reagent – (C₅H₅)₂Ti(CH₃)₂
- Phenylarsine oxide
- Phenyllithium
- Phenylmercuric acetate
- Phenylphosphine
- Phosgene – COCl₂
- Phosphine – PH₃
- Phosphite – HPO₃²⁻
- Phosphomolybdic acid
- Phosphoric acid – H₃PO₄
- Phosphorous acid (Phosphoric(III) acid) – H₃PO₃
- Phosphorus oxychloride
- Phosphorus pentabromide – PBr₅
- Phosphorus pentachloride – PCl₅
- Phosphorus pentafluoride – PF₅

- Phosphorus tribromide – PBr₃
- Phosphorus trichloride – PCl₃
- Phosphorus trifluoride – PF₃
- Phosphorus triiodide – PI₃
- Phosphotungstic acid
- Platinum(IV) chloride
- Platinum(II) chloride
- Plutonium dioxide (Plutonium(IV) oxide) – PuO₂
- Potash Alum – K₂SO₄.Al₂(SO₄)₃·24H₂O
- Potassium bromide – KBr
- Potassium carbonate – K₂CO₃
- Potassium chloride – KCl
- Potassium citrate – C₆H₅K₃O₇ + H₂O
- Potassium hydrogencarbonate
- Potassium hydrogen fluoride
- Potassium hydroxide – KOH
- Potassium iodide – KI
- Potassium monopersulfate – K₂SO₄·KHSO₄·2KHSO₅
- Potassium nitrate – KNO₃
- Potassium permanganate – KMnO₄
- Potassium sulfate – K₂SO₄
- Potassium titanyl phosphate – KTiOPO₄
- Potassium vanadate – KVO₃
- Praseodymium(III) chloride – PrCl₃
- Protonated molecular hydrogen – H₃⁺
- Prussian blue (Iron(III) hexacyanoferrate(II)) – Fe₄[Fe(CN)₆]₃

Q

R

- Radium chloride – RaCl₂
- Radon difluoride – RnF₂
- Rhodium(III) chloride – RhCl₃
- Rubidium bromide – RbBr
- Rubidium chloride – RbCl
- Rubidium fluoride – RbF
- Rubidium hydroxide – RbOH
- Rubidium iodide – RbI
- Rubidium nitrate – RbNO₃
- Rubidium oxide – Rb₂O
- Rubidium telluride – Rb₂Te
- Ruthenium(VIII) oxide – RuO₄

S

- Samarium(II) iodide – SmI₂
- Samarium(III) chloride – SmCl₃
- Scandium(III) triflate – Sc(OSO₂CF₃)₃

- Scandium(III) chloride – ScCl_3 and hydrate
- Scandium(III) fluoride – ScF_3
- Scandium(III) nitrate – $\text{Sc}(\text{NO}_3)_3$
- Scandium(III) oxide – Sc_2O_3
- Sec-butyllithium –
- Selenic acid –
- Selenious acid –
- Selenium trioxide –
- Selenium dioxide – SeO_2
- Silane – SiH_4
- Silica gel –
- Silicic acid –
- Silicochloroform –
- Silicofluoric acid –
- Silicon dioxide – SiO_2
- Silver chloride – AgCl
- Silver(I) fluoride – AgF
- Silver(II) fluoride – AgF_2
- Silver iodide – AgI
- Silver nitrate – AgNO_3
- Silver sulfide –
- Soda lime –
- Sodium acetate – $\text{NaC}_2\text{H}_3\text{O}_2$
- Sodium bromide – NaBr
- Sodium bromate – NaBrO_3
- Sodium carbonate – Na_2CO_3
- Sodium chloride – NaCl
- Sodium chlorate – NaClO_3
- Sodium cyanide – NaCN
- Sodium hydride – NaH
- Sodium hydrogen carbonate (Sodium bicarbonate) – NaHCO_3
- Sodium hydrosulfide – NaSH
- Sodium hydroxide – NaOH
- Sodium iodide – NaI
- Sodium monofluorophosphate (MFP) – Na_2PFO_3
- Sodium nitrate – NaNO_3
- Sodium nitrite – NaNO_2
- Sodium percarbonate – $2\text{Na}_2\text{CO}_3 \cdot 3\text{H}_2\text{O}_2$
- Sodium phosphate; see Trisodium phosphate – Na_3PO_4
- Sodium silicate – Na_2SiO_3
- Sodium sulfate – Na_2SO_4
- Sodium sulfide – Na_2S
- Sodium sulfite – Na_2SO_3
- Sodium tellurite – Na_2TeO_3
- Stannous chloride (tin(II) chloride) – SnCl_2
- Stibine – SbH_3
- Strontium chloride – SrCl_2
- Strontium nitrate – $\text{Sr}(\text{NO}_3)_2$
- Strontium titanate – SrTiO_3
- Sulfamic acid – $\text{H}_3\text{NO}_3\text{S}$

- Sulfane – H₂S
- Sulfur dioxide – SO₂
- Sulfur monochloride
- Sulfur dichloride
- Sulfurated potash
- Sulfuric acid – H₂SO₄
- Sulfurous acid – H₂SO₃
- Sulfuryl chloride – SO₂Cl₂

T

- Tantalum carbide – TaC
- Tantalum(V) oxide – Ta₂O₅
- Telluric acid – H₆TeO₆
- Tellurium dioxide – TeO₂
- Tellurium tetrachloride – TeCl₄
- Tellurous acid – H₂TeO₃
- Terbium(III) chloride
- Tert-butyllithium
- Tetraborane(10)
- Tetrabutyltin
- Tetrachloroauric acid
- Tetraethyl lead – (CH₃CH₂)₄Pb
- Tetraethyl tin
- Tetrafluorohydrazine
- Tetramminecopper(II) sulfate – [Cu(NH₃)₄]SO₄
- Tetraphenyltin
- Tetrasulfur tetranitride – S₄N₄
- Thallium(III) sulfate
- Thallium(I) fluoride
- Thallium(III) oxide
- Thallium(I) carbonate
- Thionyl chloride – SOCl₂
- Thiophosgene
- Thiophosphoryl chloride
- Thorium dioxide – ThO₂
- Thortveitite – Scandium and Yttrium containing mineral
- Thulium(III) chloride
- Tin(II) chloride – SnCl₂
- Tin(II) fluoride – SnF₂
- Tin(IV) chloride – SnCl₄
- Titanium boride – TiB₂
- Titanium carbide – TiC
- Titanium dioxide – TiO₂, titanium(IV) oxide
- Titanium dioxide (B) – TiO₂, titanium(IV) oxide
- Titanium nitride – TiN
- Titanium(IV) bromide – TiBr₄ titanium tetrabromide
- Titanium(IV) chloride – TiCl₄ titanium tetrachloride
- Titanium(III) chloride – TiCl₃

- Titanium(II) chloride – TiCl_2
- Titanium(IV) iodide – TiI_4 titanium tetraiodide
- Titanocene dichloride – $(\text{C}_5\text{H}_5)_2\text{TiCl}_2$
- Triethylaluminium
- Trifluoromethylisocyanide
- Trifluoromethanesulfonic acid
- Tris(pentafluorophenyl)boron
- Trimethyltin chloride
- trioxidane
- Triphenylantimony (triphenylstibine)
- Tripotassium phosphate – K_3PO_4
- Trisodium phosphate – Na_3PO_4
- Tungsten carbide – WC
- Tungsten(VI) chloride – WCl_6
- Tungsten(VI) Fluoride – WF_6
- Tungstic acid – H_2WO_4
- Tungsten hexacarbonyl – $\text{W}(\text{CO})_6$

U

- Uranium hexafluoride – UF_6
- Uranyl nitrate – $\text{UO}_2(\text{NO}_3)_2$
- Uranyl zinc acetate
- Triuranium octaoxide (pitchblende or yellowcake) – U_3O_8

V

- Vanadium carbide – VC
- Vanadium oxytrichloride – VOCl_3 Vanadium(V) oxide trichloride
- Vanadium(IV) chloride – VCl_4
- Vanadium(II) chloride – VCl_2
- Vanadium(II) oxide – VO
- Vanadium(III) bromide – VBr_3
- Vanadium(III) chloride – VCl_3
- Vanadium(III) fluoride – VF_3
- Vanadium(IV) fluoride – VF_4
- Vanadium(III) oxide – V_2O_3
- Vanadium(IV) oxide – VO_2
- Vanadium(IV) sulphate – VOSO_4
- Vanadium(V) oxide – V_2O_5
- Vanadocene dichloride – $(\text{C}_5\text{H}_5)\text{VCl}_2$
- Vanadyl(acetylacetate) – $\text{VO}(\text{acac})_2$

W

- Water – H_2O

X

- Xenon difluoride – XeF_2
- Xenon hexafluoroplatinate – $\text{Xe}[\text{PtF}_6]$
- Xenon tetrafluoride – XeF_4
- Xenon tetroxide – XeO_4
- Xenic acid – H_2XeO_4

Y

- Ytterbium(III) chloride – YbCl_3
- Ytterbium(III) oxide – Yb_2O_3
- Yttrium(III) antimonide – YSb
- Yttrium(III) arsenide – YAs
- Yttrium(III) bromide – YBr_3
- Yttrium aluminium garnet
- Yttrium barium copper oxide
- Yttrium(III) fluoride – YF_3
- Yttrium iron garnet
- Yttrium(III) oxide – Y_2O_3
- Yttrium(III) sulfide – Y_2S_3

Z

- Zinc bromide – ZnBr_2
- Zinc carbonate – ZnCO_3
- Zinc chloride – ZnCl_2
- Zinc cyanide – $\text{Zn}(\text{CN})_2$
- Zinc fluoride – ZnF_2
- Zinc iodide – ZnI_2
- Zinc oxide – ZnO
- Zinc selenide – ZnSe
- Zinc sulfate – ZnSO_4
- Zinc sulfide – ZnS
- Zinc telluride – ZnTe
- Zirconium carbide – ZrC
- Zirconium(IV) chloride – ZrCl_4
- Zirconium nitride – ZrN
- Zirconium hydroxide – $\text{Zr}(\text{OH})_4$ no zirconium tetrahydroxide or hydrated zirconia?
- Zirconium(IV) oxide – ZrO_2
- Zirconium orthosilicate – ZrSiO_4

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