

# Pharmaceutical colourants

## Overview of approved and used pharmaceutical colourants

A possible classification:

1. Inorganic pigments
2. Organic dyes
  - 2.1 Lipophilic dyes
    - 2.1.1 Carotenoids
    - 2.1.2 Xanthophyls
  - 2.2 Hydrophilic dyes
    - 2.2.1 Fused-rings glycosides, pseudoglycosides and their aglycones
    - 2.2.2 Phenolic dyes not based on fused rings
    - 2.2.3 Indole dyes
    - 2.2.4 Hydrophilic azo-dyes
    - 2.2.5 Hydrophilic triarylmethane dyes

## 1. Inorganic pigments

- in most insoluble or poorly soluble in both water and organic solvents
- colouring of tablet mass, capsules, surface films on coated tablets
- colouring of suspension creams and ointments

$\text{CaCO}_3$ , E 170

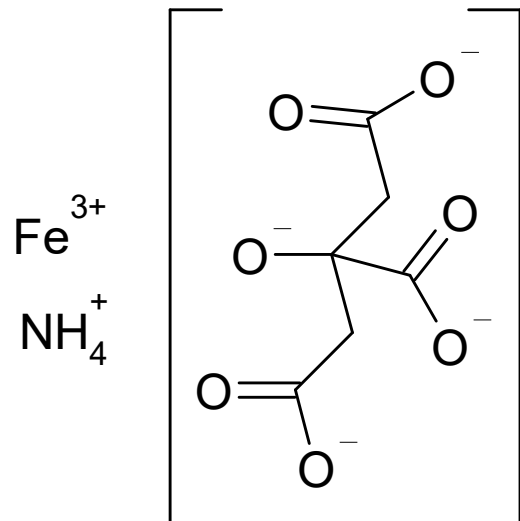
- only for surface „colouring“

$\text{TiO}_2$ , E 171, CI 77891, titane white

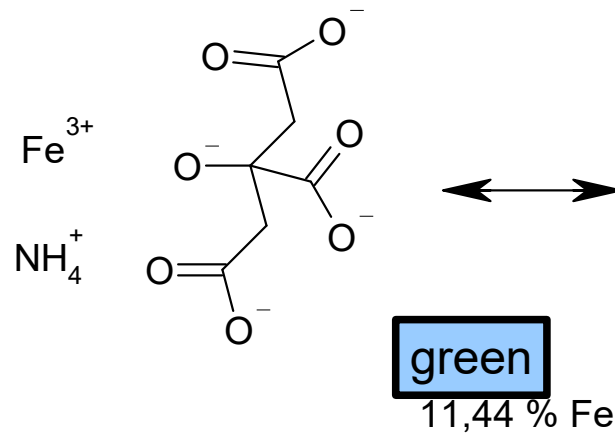
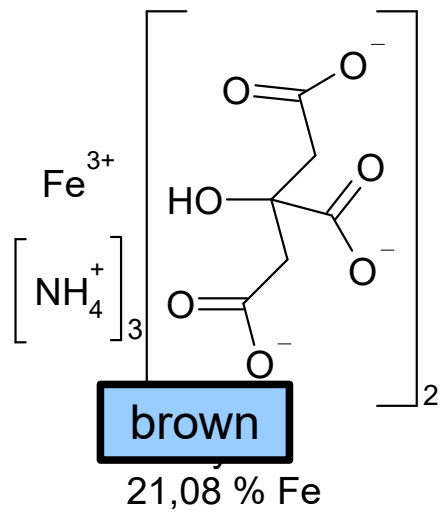
- also protecting factor against UV irradiation

Iron oxides and hydroxides E 172

Ferric-ammonium citrate



- Steinsvik S. et al.: J. Phys. Chem. Solids 58, 969 (1997)
- water soluble
- since 2002 rejected as a colourant; before 2002 approved in USA; still listed in DAC (=Deutsches Arzneimittel Codex = German Medicines Codex)



## 1. Inorganic pigments

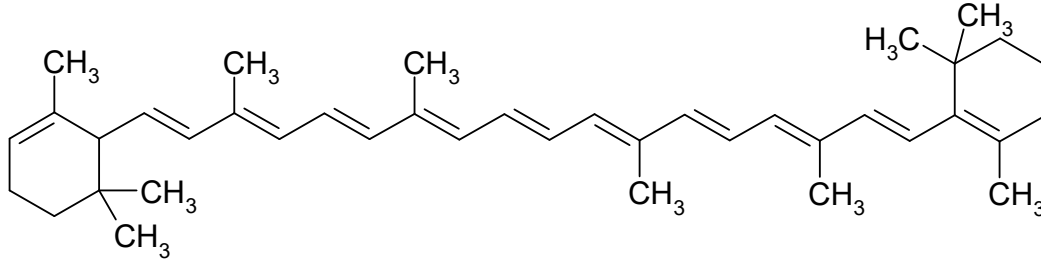
**AI E 173, CI 77000**

- finely powdered - inflammable (pyrophoric)

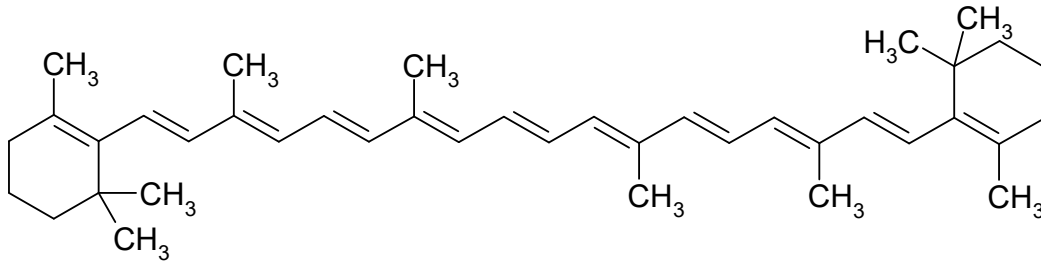
## 2.1. Lipophilic organic dyes

### Carotenoids (together E 160)

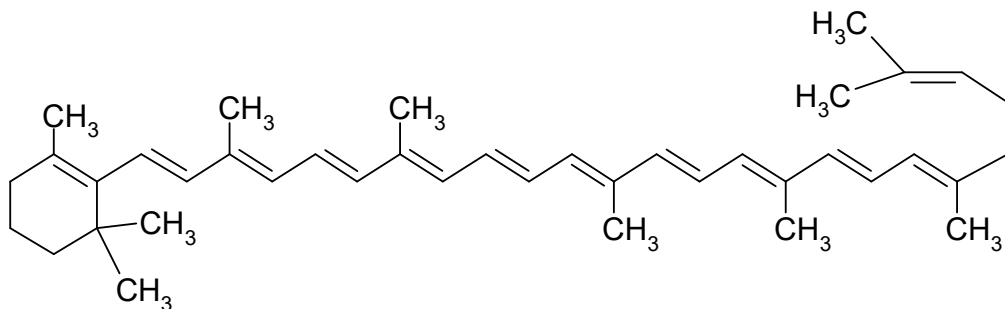
#### $\alpha$ , $\beta$ , $\gamma$ -carotene together E 160a



$\alpha$ -carotene ( $\beta,\epsilon$ -carotene)  
m.p. 187,5°C  
•log  $P_{o/w}$  = 17.49



$\beta$ -carotene  
CI 75130 (natural), CI 40800 (synthetic)  
E 160a  
m.p. 183°C  
•poorly stable on air and light  
•from light yellow to deep orange in dependence on concentration  
•0.1% for suppositories  
•log  $P_{o/w}$  = 17.63

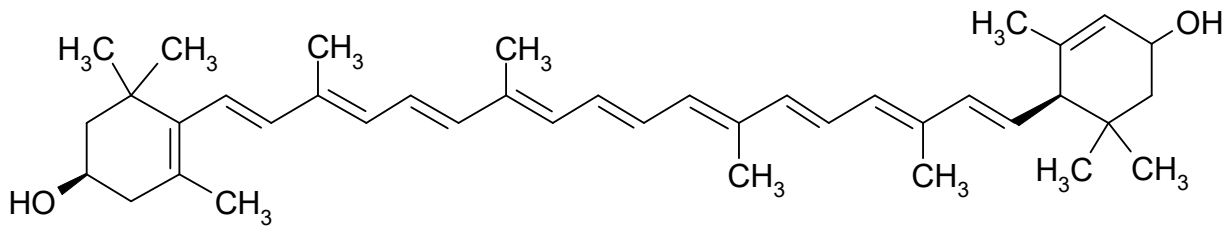


$\gamma$ -carotene ( $\beta,\theta$ -carotene)  
m.p. 154°C  
•log  $P_{o/w}$  = 17.63



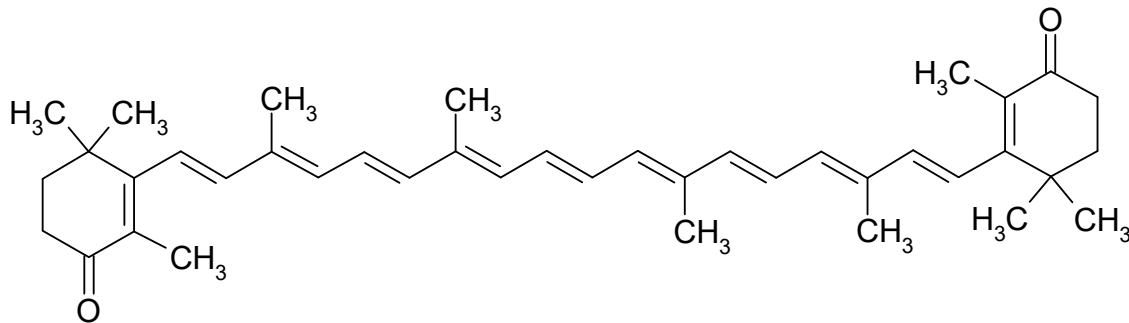
## Xanthophyls

- together E 161



### **lutein**, E 161b

- colourant of the egg yolk
- m.p. 196°C
- log P<sub>o/w</sub> =14.82



### **canthaxanthin**, CI 40850, E 161g

- orange
- log P<sub>o/w</sub> =14.1
- antioxidant, believed to be a cancer protectant



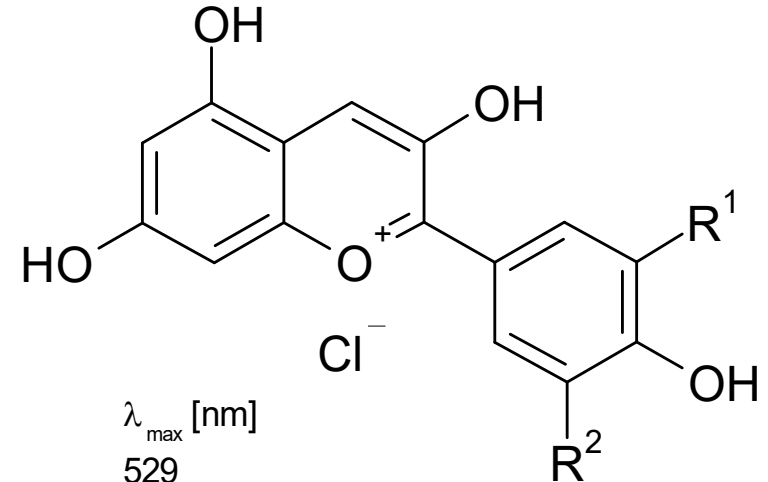
## 2.1 Hydrophilic organic dyes

### 2.1.1 Fused-rings glycosides, pseudoglycosides and their aglycones

#### Anthocyanidines

= hydroxylated and methoxylated derivatives of 2-phenylbenzopyrylium

- together E 163
- aglycones of anthocyanins



Name	R <sup>1</sup>	R <sup>2</sup>	origin	m.p. [°C]	log P <sub>o/w</sub>	colour	λ <sub>max</sub> [nm]
pelargonidine	H	H	<i>Pelargonium</i>	> 350	2.68	salmon	529
kyanidin	OH	H	<i>Cyanus</i>	> 300	2.2	red	544
delphinidine	OH	OH	<i>Delfinium</i>	> 350	2.14	blue	553
peonidine	OCH <sub>3</sub>	H	<i>Peaeonia</i>			red	543
petunidine	OCH <sub>3</sub>	OH	<i>Petunia</i>			blue-red	522
malvidine	OCH <sub>3</sub>	OCH <sub>3</sub>	<i>Malva</i>	> 300; 202-203	2.33	pink-red	553

- in plants as glycosides in most in position 3, particularly esterified with subst. cinnamic acids or acetic acid in pos. 6
- the colour of free aglycons depends on pH: blue in alkaline, red in acidic, this is not valid for glycosides
- increasing number of hydroxyles results in a more intense colour, methylation of hydroxyles shifts the shade from blue to red

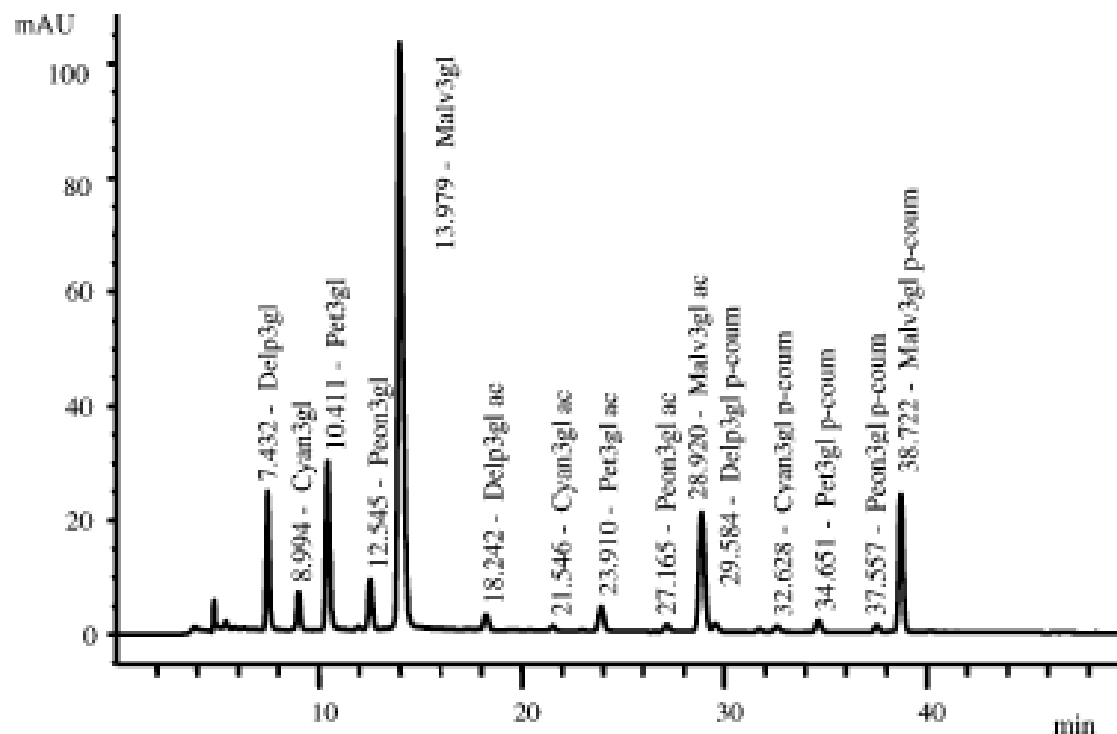
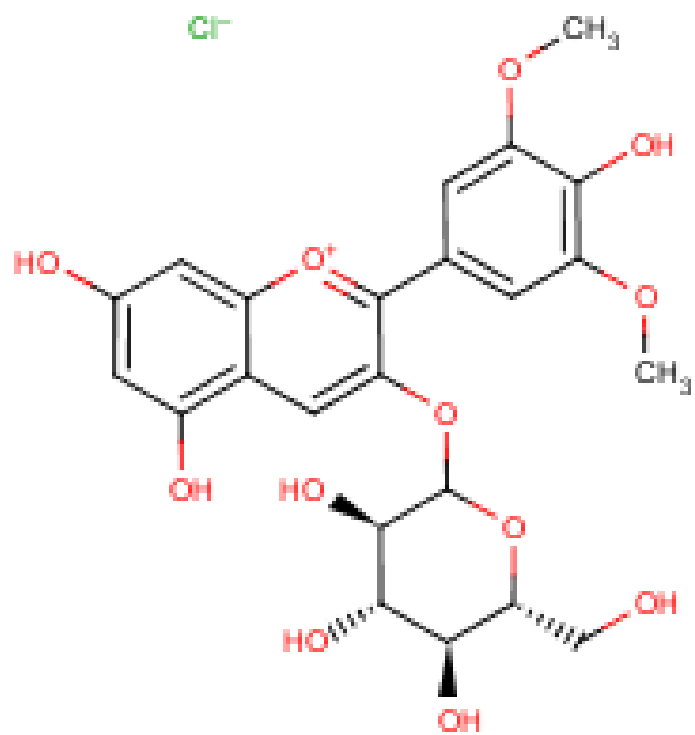
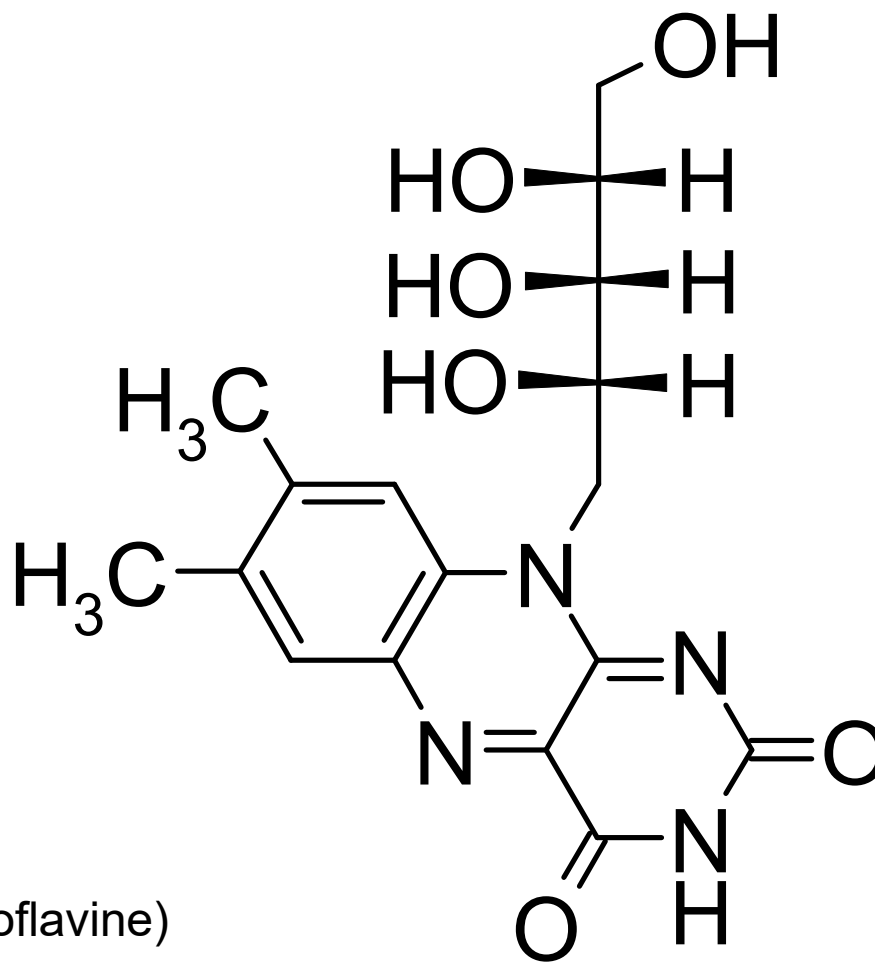


Figure 1. Representative profile of a grape extract analyzed by reversed phase HPLC at 520 nm, as described under Materials and Methods: delphinidin 3-glucoside (Delp3gl), cyanidin 3-glucoside (Cyan3gl), petunidin 3-glucoside (Pet3gl), peonidin 3-glucoside (Peon3gl), malvidin 3-glucoside (Malv3gl), and their corresponding acetic acid (ac) and *p*-coumaric acid esters (*p*-coum).



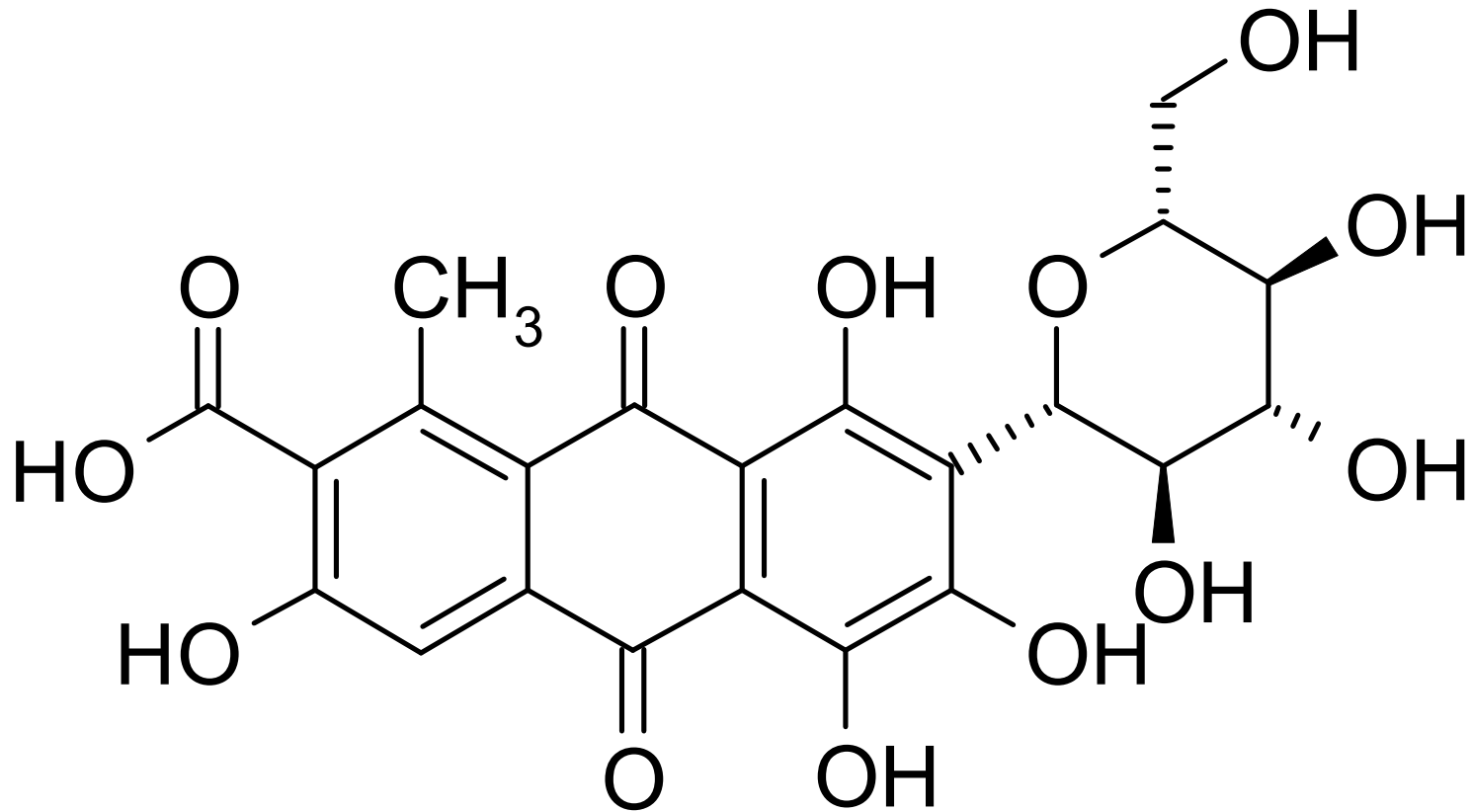
**Riboflavine** (vitamin B2, lactoflavine)

E 101

•yellow

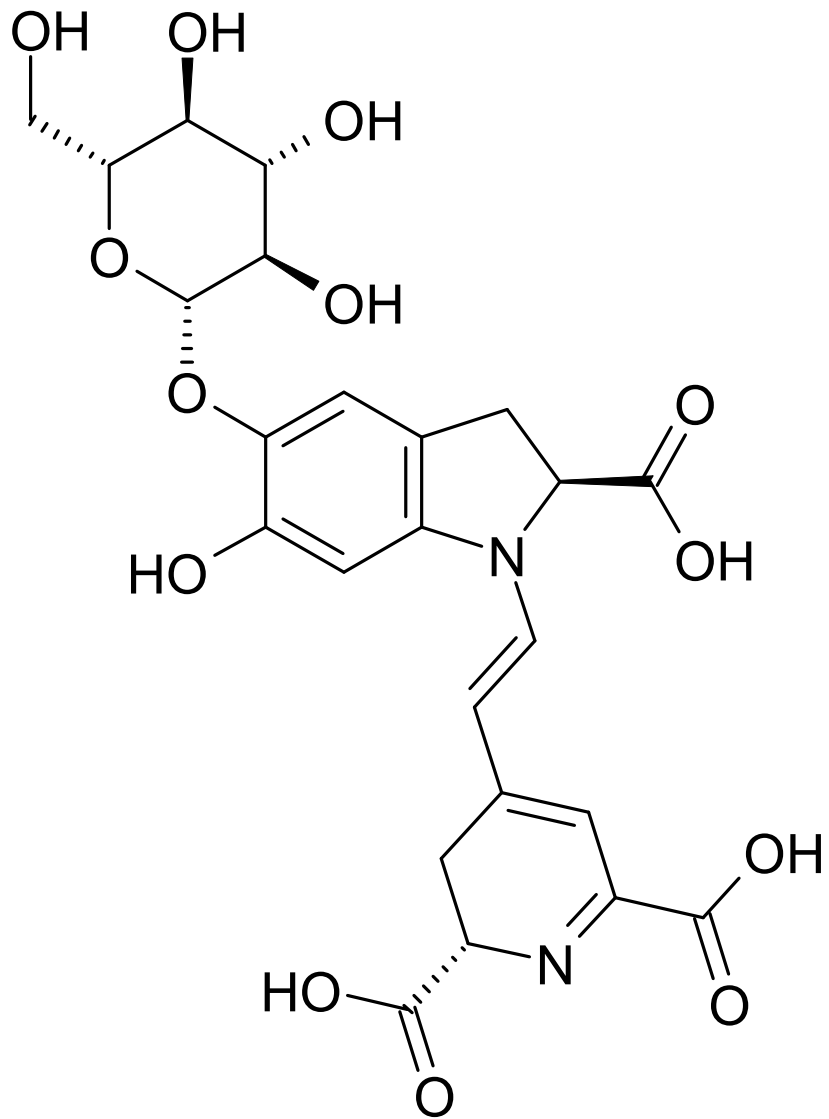
•listed in PhEur, CzP2009, BP, USP, JP

•Relative absorbance  $A_{373 \text{ nm}} / A_{267 \text{ nm}} = 0.31 - 0.33$ ,  $A_{444 \text{ nm}} / A_{267 \text{ nm}} = 0.36 - 0.39$



**Carmine** (= carminic acid, coccinellin, cochineal extract, CI 75470, E 120)

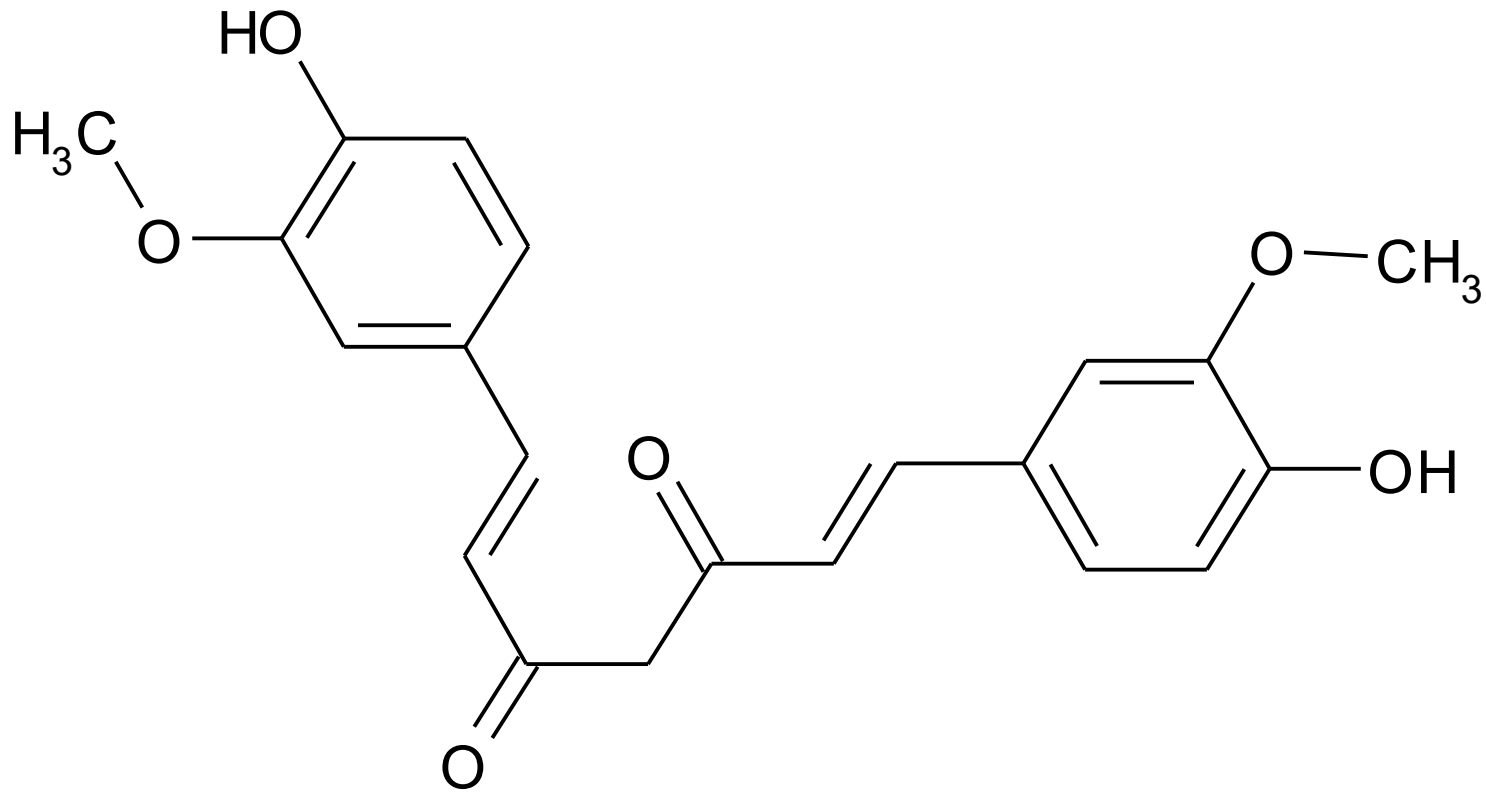
- deeply red
- fruit odour
- water solubility cca 30 g / l
- $\lambda_{\text{max}}$  (buffer pH = 3) = 490 – 493 nm



Betanine, E 162

- *Beta vulgaris* (= beet)
- isolated by water extraction at low temperature

## 2.2.2 Phenolic dyes not based on fused rings

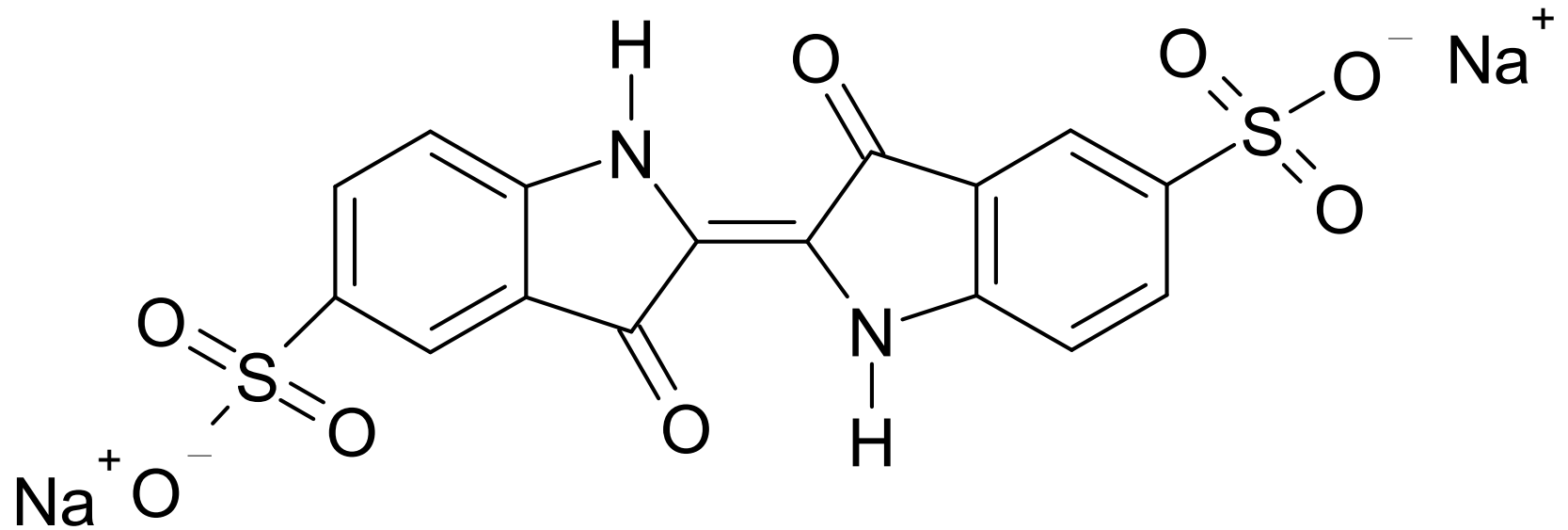


Curcumin

E 100

- yellow-orange
- approved in EU
- from turmeric root (*Curcuma longa*)

### 2.2.3 Indole dyes



indigo carmine

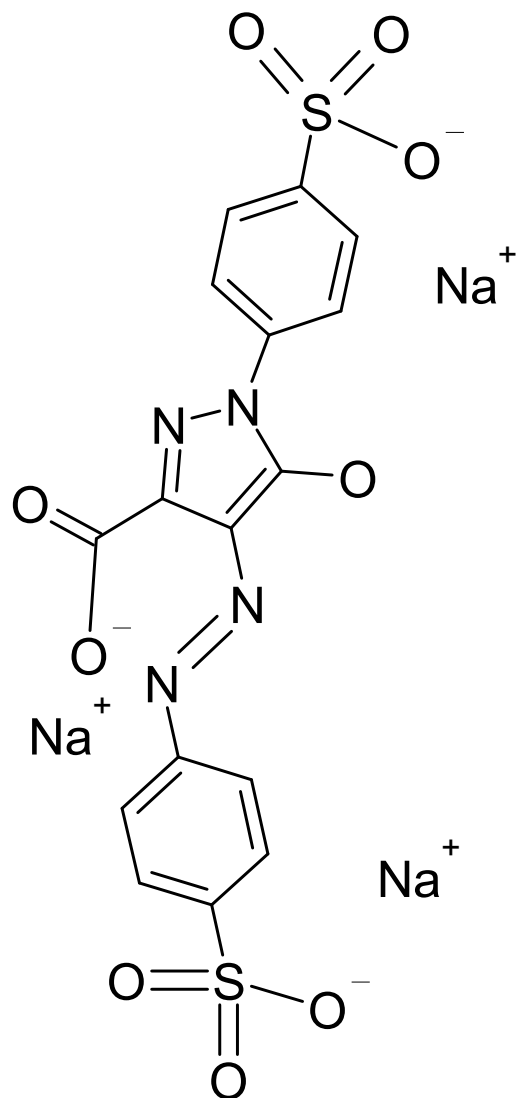
indigotindisulfonate sodium [USP]

E 132, CI 73015, FD&C blue #2

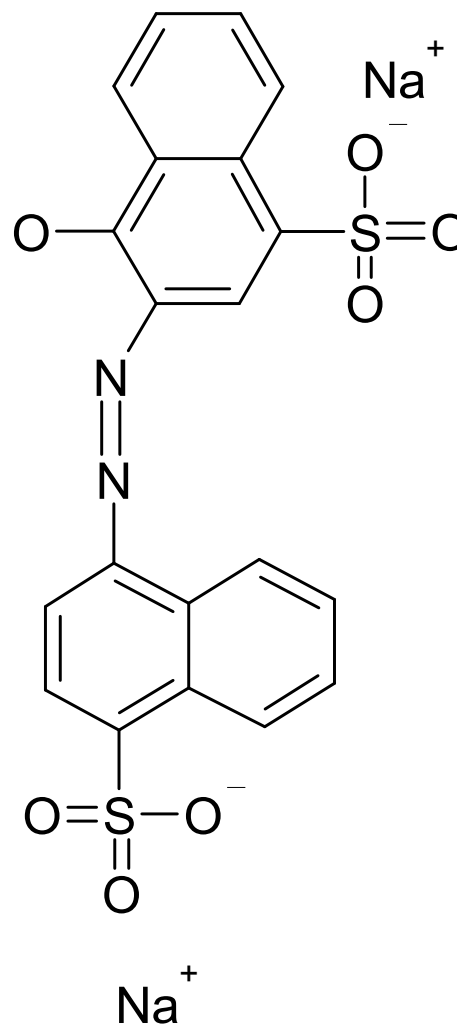
•deeply blue powder, aqueous solutions are blue or blue-purple

• $\lambda_{\text{max}} = 604 \text{ nm}$

## 2.2.4 Hydrophilic azo-dyes



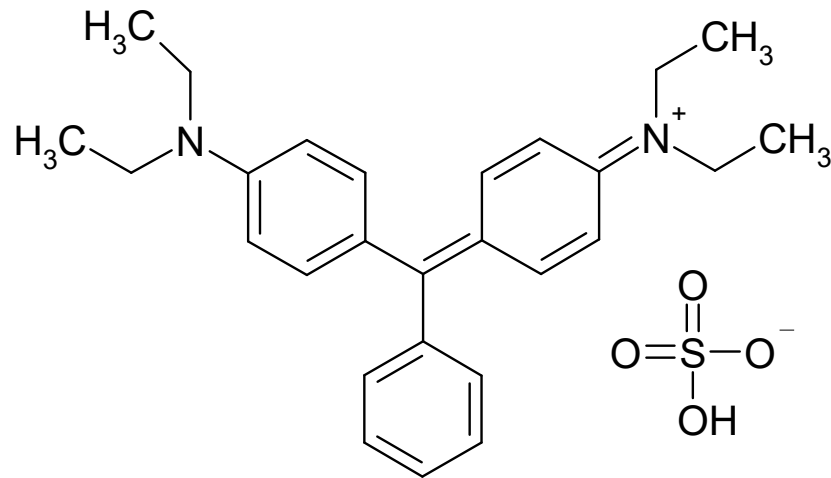
**tartrazine**, CI 19140, E 102,  
acidic yellow 23, egg yellow A  
•orange-yellow  
• $\log P_{\text{octanol/water}} = -10.17$



**carmoisine**, azorubin, E 122, CI 14720,  
C.I. Acid Red 14, chromotrop FB  
• $LD_{50} (p.o., \text{mouse}) = 8 \text{ g / kg}$



## 2.2.5 Hydrophilic triarylmethane dyes



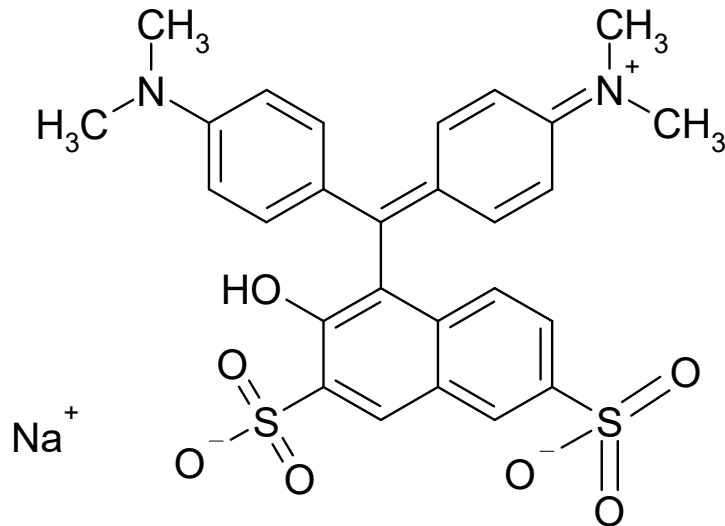
**Brilliant Green**, CI 42040, *Viride nitens*, Malachite Green G, Emerald Green

- not approved as drug and food dye, but listed in many pharmacopoeias as a therapeutic (antiseptic)

- m.p. 210°C

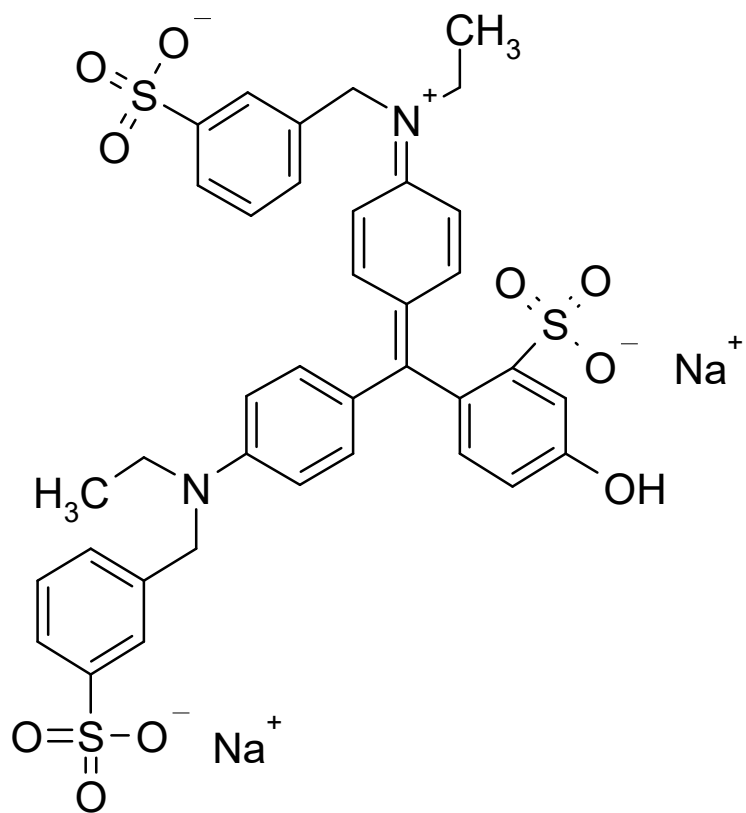
- log P<sub>o/w</sub> = 2.01

- LD<sub>50</sub> = 3 - 8 mg / kg



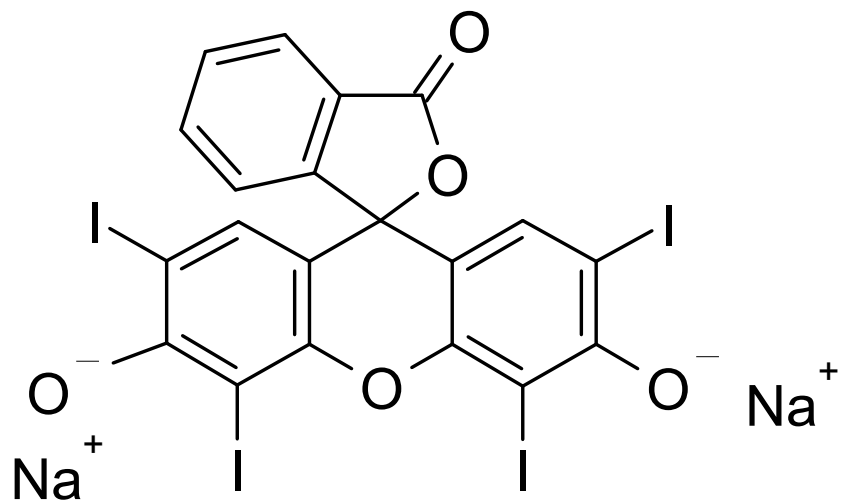
- Acid Green 50**, Green S, CI 44090, E 142, Brilliant Acid Green BS, Food Green S

- LD<sub>50</sub> = 2 g / kg



**Fast green FCF**, CI 42053, Solid Green FCF, Food Green 3

- $\lambda_{\text{max}}$  (ethanol 50 %) 622 - 626 nm
- spec. absorbance  $A_{1\text{cm}}^{1\%}$  ( $\lambda_{\text{max}}$ ; 0.003 g/l; ethanol 50%) = 1360 – 1610
- log P o/w = -5.42



**Erythrosine** (erythrosine B, CI 45430, E 127)

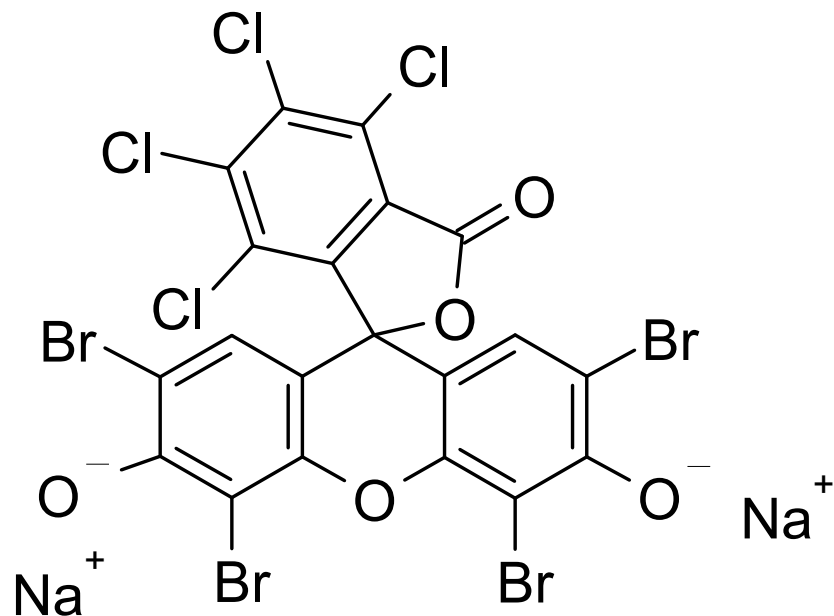
reddish-brown

$$\lambda_{\max}(\text{water}) = 524 - 527 \text{ nm}$$

specific absorbance

$$A_{1\text{cm}}^{1\%}(\lambda_{\max}; 0.005 \text{ g/l; water}) = 930 - 1170$$

$$LD_{50}(\text{p.o., rat}) = 1.84 \text{ g / kg}$$



**Phloxine B** (CI45410, D&C red #28, Eosine Blue)

deeply brown

$$\lambda_{\max}(\text{ethanol 50\%}) = 546 - 560 \text{ nm}$$

$$A_{1\text{cm}}^{1\%}(\lambda_{\max}; 0.005 \text{ g/l; ethanol 50\%}) = 930 - 1400$$

$$LD_{50}(\text{p.o., rat}) = 8.4 \text{ g / kg}$$