

Phytochemistry

**Introduction to Phytochemistry, Historical
Background**

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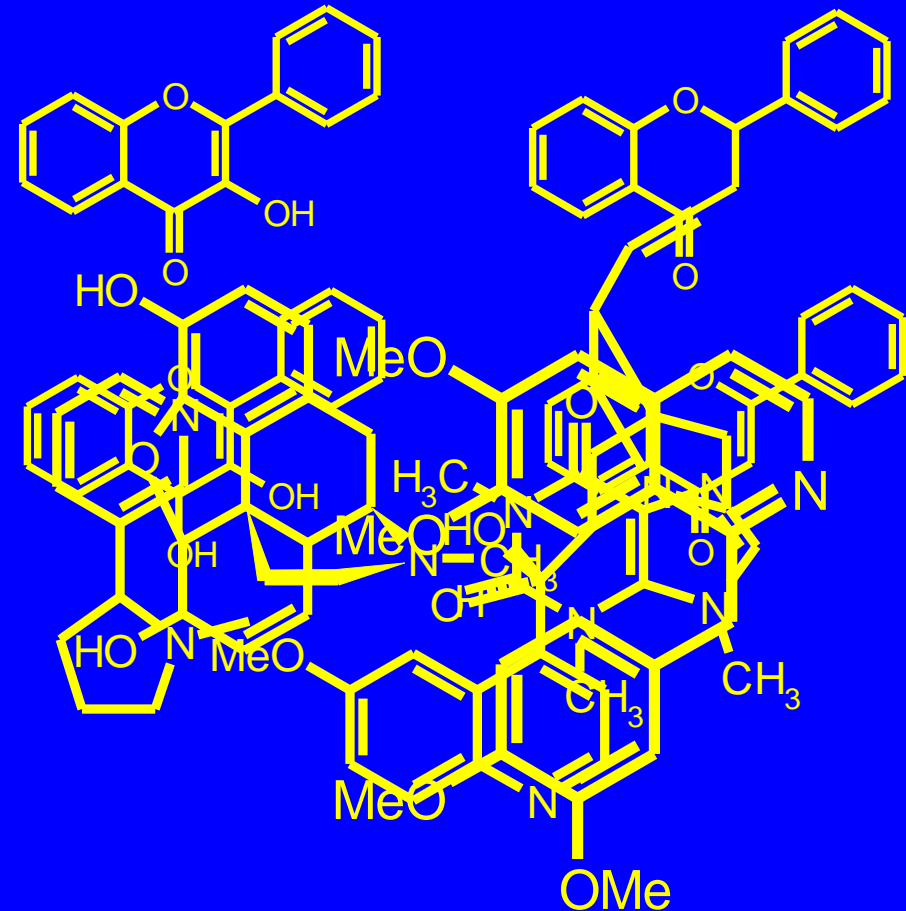
Faculty of Pharmacy

Masaryk University

- Greek origin of the word „*phytochemistry*“
- All chemical aspects concerning the plant kingdom and life of plants
 - Plant nutrition and primary metabolism of plants (photosynthesis etc.) except for plant physiology and plant biochemistry
 - Chemical studies of organic carbon-containing compounds created by plants (especially secondary metabolites)
 - For example: comparative phytochemistry – comparing the differences of natural compounds synthesis in different plant species and botanical families

Founders

- 1664 Boyle (England) - flavonoids
- 1806 Sertürner (Deutschland) - morphine
- 1817-1820 Pelletier and Caventou (France) - quinine
- 1848 Merck (Deutschland) - papaverine
- 1829 Posselt a Reimann - nicotine and 1819 Runge - caffeine (Deutschland)



- Till the fifties, a slow progress
 - Compounds isolated and identified only in crystal form, identification based on chemical degradation and biogenetic principals
- 1957, 5000 natural compounds including alkaloids
- After WW2, a great progress in chromatographic and spectral applications (separation and identification)
- Present time – more than 600 000 natural compounds

Primary and Secondary Plant Metabolism

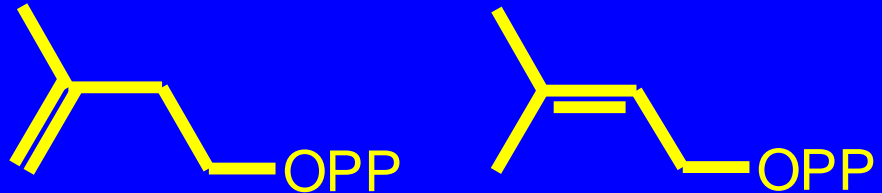
- *Primary metabolites*: ubiquitous, vital compounds
 - Simple sugars
 - Amino acids
 - Purines and pyrimidines of nucleic acids
 - Chlorophyll
 - Fatty acids and low-molecular weight carboxylic acids

- *Secondary metabolites*: specific conditions necessary for their production.
- Production in different plant organs, transportation and chemical modifications.
- Generally not necessary for life.
- Nomenclature by different systems:
 - Trivial
 - Semi trivial
 - Chemical
 - According to the biosynthetic origin

- Main classes of secondary metabolites:
 - Terpenoids
 - Alkaloids and other nitrogen-containing compounds
 - Phenolic compounds

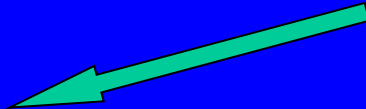
Terpenoids

- Biosynthetic origin is isopentyl-diphosphate and dimethylallyl-diphosphate



- Lipophilic properties
- Unsaturated hydrocarbons, different oxidation pattern.
- Individual subdivisions according to the number of C5 units (isoprene units)
- Approx. 20 000 compounds

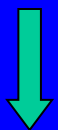
Dimethylallyl-diphosphate + Isopentenyl-diphosphate



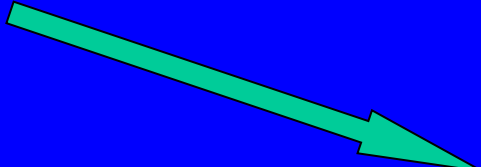
Geranyl-diphosphate → Monoterpenes → Iridoids



Farnesyl-diphosphate → Sesquiterpenes → Sesquiterpenic lactones



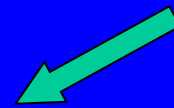
Geranyl-geranyl-diphosphate



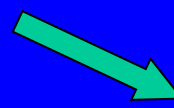
Triterpenoids (C30)



Diterpenoids



Limonoides
Cardenolides
Quassinoides
Cucurbitacines



Saponins
Phytosterols

<i>Group of Terpenoids</i>	<i>Characteristics</i>
Monoterpenes	Volatile compounds, essential oils
Iridoids	Usually lactones, bitter taste, often glycosides
Sesquiterpenes	Essential oils, higher boiling point
Sesquiterpenic lactones	Asteraceae family, bitter taste
Diterpenoids	Resins and plant growth hormones
Triterpenoidal and steroidal saponins	Foam producing compounds, hemolytic activity
Cardenolides and bufadienolides	Cardioactive glycosides, toxins
Phytosterols	Building blocks of membranes
Cucurbitacins	Bitter taste, typical for Cucurbitaceae
Carotenoids	Plant pigments

Alkaloids and other Nitrogen-Containing Metabolites

- The most known nitrogen-containing compounds
- Nitrogen part of structure (usually in a cyclic system)
- Basic
- Limited distribution in species or families – chemotaxonomic limitations
- More than 10 000 compounds

Peptides

Non-protein amino acids

Amino acids

Amines

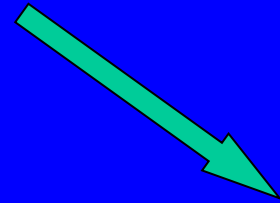
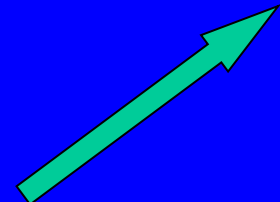
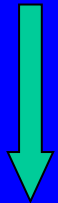
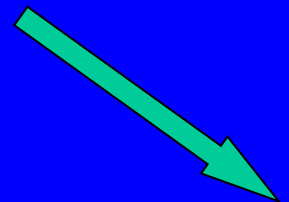
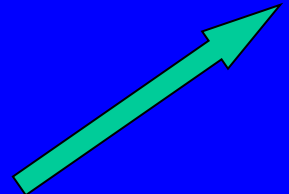
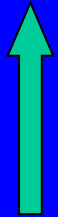
Cyanogenic glycosides

(protein building blocks)

Aldoximes

Alkaloids

Glucosinolates

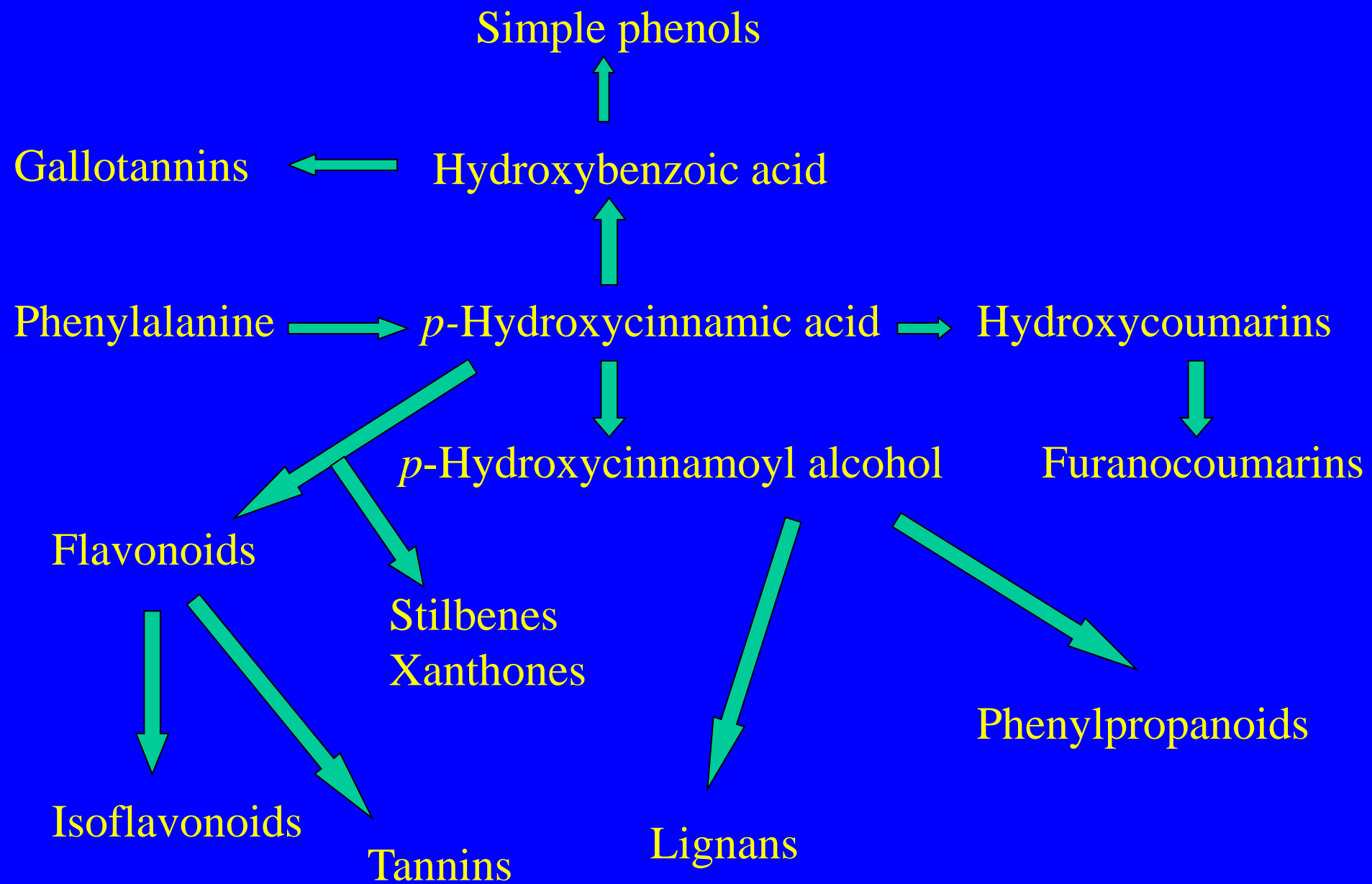


<i>Alkaloids</i>	
Amaryllidaceae	Limited to Amaryllidaceae, 275 different structures
Betalain	Yellow and red pigments
Diterpenoid alkaloids	Very poisonous, mainly Ranunculaceae
Indole	Many structures, mainly Apocynaceae and Loganiaceae
Isoquinoline	Widely distributed, greatest group, many subdivisions
Lycopodium	150 structures, clubmosses, mosses
Monoterpene	Relative to iridoids, different heterocycles
Sesquiterpene	Mainly orchids
Peptide	Mainly Rhamnaceae
Pyrrolidine and piperidine	Polyhydroxyalkaloids, inhibitors of enzymes, pharmacologically active
Pyrrolizidine	Senecio, Asteraceae
Quinoline	Rutaceae
Steroidal	Apocynaceae, Buxaceae, Liliaceae, Solanaceae
Tropane	Solanaceae

<i>Non-protein amino acids</i>	Often present in seeds, especially Leguminosae
<i>Amines</i>	Mainly volatile compounds, sporadic (scattered) occurrence, in traces
<i>Cyanogenic glycosid</i>	Widely distributed in Rosaceae, low concentrations
<i>Glucosinolates</i>	Brassicaceae
<i>Purines a pyrimidines</i>	Nucleic acids, methylxanthines

Phenolic compounds

- Aromatic compounds with one or more hydroxyl groups
- Possible and common in glycosidic form
- Biosynthesis mainly from phenylalanine via shikimic acid pathway



<i>Subclass of phenolic compounds</i>	<i>Characteristics</i>
Anthocyanins	Red and blue pigments
Anthochlors	Yellow pigments - chalcones and aurones
Benzofurans	Distributed through some higher plants and lichens
Chromones	Small group
Coumarins	700 compounds, widely distributed
Flavonoids	May compounds, widely distributed
Isoflavonoids	Mainly Leguminosae, rare as glycosides, estrogen-like effect
Lignans	Wood and bark
Phenols a phenolic acids	Universal for plant
Phenylpropanoids	Many structures, form glycosides, as aglycones part of essential oils
Quinonoids	Benzoquinones, naphthoquinones
Tannins	Tannins hydrolyzable, condensed
Xanthones	Pigments, mainly Gentianaceae

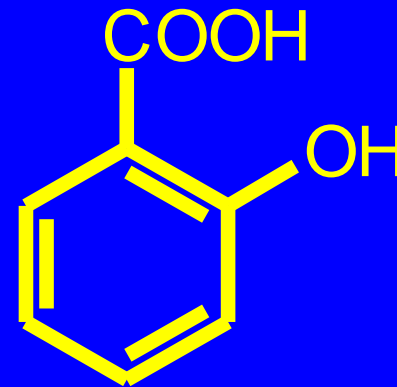
Functions of Secondary Metabolites

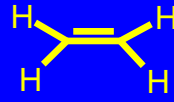
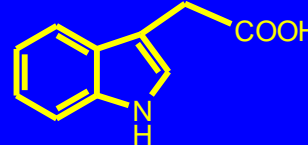
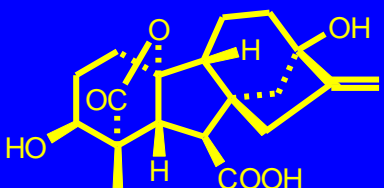

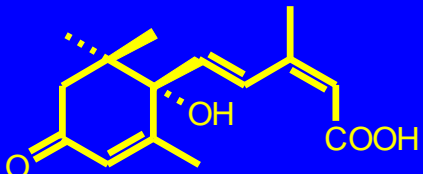
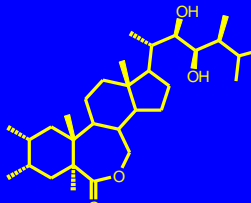
- Difficult to establish the true function
- Multifunctional
- Occurrence both as secondary and primary metabolites

- Common the occurrence of a group of very similar compounds, difficult to establish function of a selected metabolite
- Advantage of the presence of a group of similar compounds before the presence of a single metabolite

Functions of Secondary Metabolites

- Multifunction
 - **Salicylic acid:** signal stress molecule, metabolic signal for pollen production, protection against herbivores, allelopathic effect



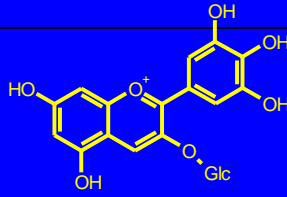
Ethylene		Simple structure, hormone of ripening, seed germination, inflorescence development, stress response
Indol-3-acetic acid		30 different conjugates, „auxine“, tropism, elongation, induction of cambial cells division
Gibberellines		102 of structures, elongation and growth of plant organs, stratification and vernalization
Cytokinines		14 structures, cell division, cell differentiation, inhibition of abscission
Abscisic acid		Seed ripening, inhibition of seed germination, durability to lack of water
Brassinosteroids		More than 40 structures, stem elongation, increase of ethylene production, wood growth, stamina growth
Polyamines	$H_2N(CH_2)_3NH(CH_2)_4NH(CH_2)_3NH_2$	4 structures, interaction with RNA, effect on DNA and RNA replication, growth and development of cells, ripening, flowering

Plant Pigments and Odors

- Mainly as attractants for pollinators (bats, birds, bees, butterflies and others)
- Different pollinators – different attractants
 - bees – blue and yellow (flavonoids and anthocyanins), soft odors
 - bats – night pollinators – fruit odors, sulphuric odors (for example methylsulphides)
- Compounds always in combinations (anthocyanins + flavonoids in blue, carotenoids + flavonoids in yellow)

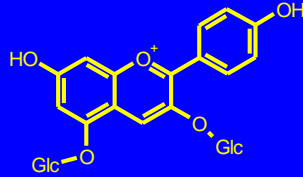
- Extra large panel of compounds as plant odor components
- Monoterpenes, sesquiterpenes, benzoates, derivatives of anthranilic acid, simple sulphuric compounds, simple esters, amines
- Different panel of compounds in leaves, flowers and fruits
 - Different purposes of presence – color and odor should repel
 - Sometimes opposite – attraction because of seed spreading

Delphinidin-3,5-diglucoside



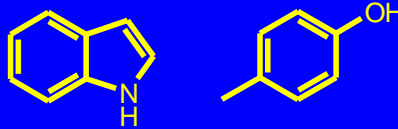
Blue color, attractive for humble bees

Pelargonidin-3,5-diglucoside

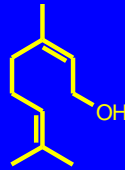


Scarlet color

Indol and p-cresol

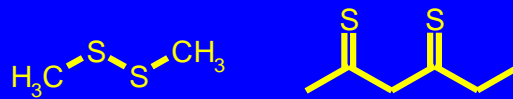


Geraniol



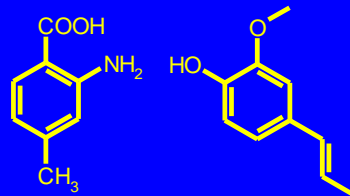
Attractive for male bee

Dimethyldisulphid, 2,4 dithiopentan



Attractive for bats

Methylantranilate and isoeugenol



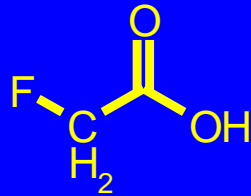
Synergise to attract butterfly pollinators

Protection Against Herbivores

Toxic for:

- Species - salicin
 - All species - ouabain
 - For plant alone - cyanogenic glycosides
-
- Concentration increases after stimulus - elicitation
 - Appreciated insecticides (rotenone, pyrethroids)
 - Differences in activities of similar structures (monoterpenoids more effective when oxygenated, pyrrolizidine alkaloids when hydroxylated and in the form of esters)

Fluoracetic acid



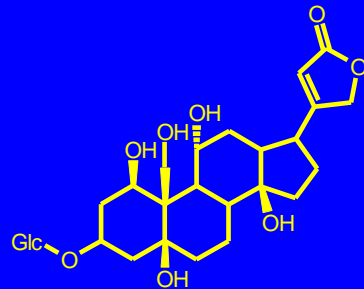
Dichapetalum cymosum, interaction with Krebs cyclus, respiratory poison for mammals

Miserotoxin



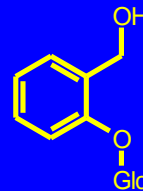
3-nitro-1-propanol-β-D-glucoside, *Astragalus atropubescens*, toxic for cattle, mitochondrial poison

Oubain



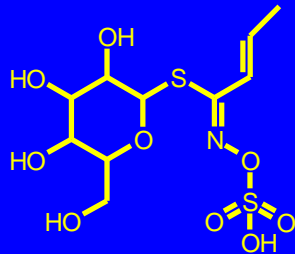
Acokanthera oubaio, cardio poison for mammals

Salicin



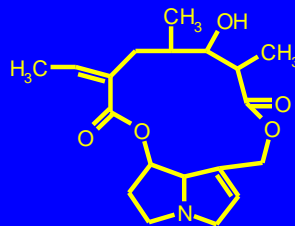
Salix spp.

Sinigrin



Brassica, toxic especially for cattle

Senecionin



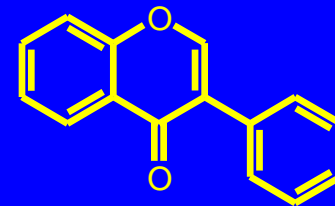
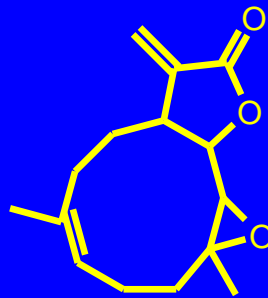
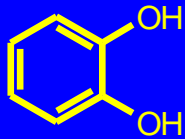
Senecio, hepatotoxic, mainly for cattle

Antifungal effect

- Different types of barriers: defensive proteins, lignification, membranes and wax-compound layers, low molecular compounds
- 1. constitutive
- 2. synthesis *de novo* as a response to infections – so called phytoalexins (more than 300 compounds identified)
- Attacking fungi produced toxins – toxicity possible for the plant and also for the consumer

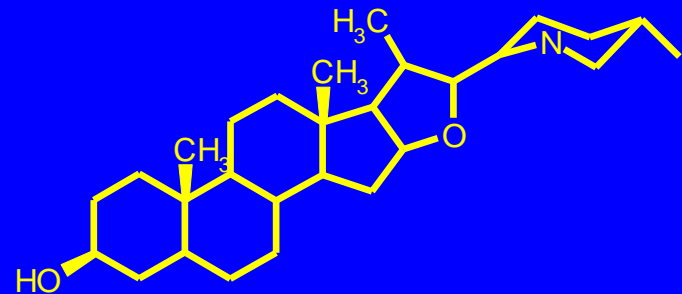
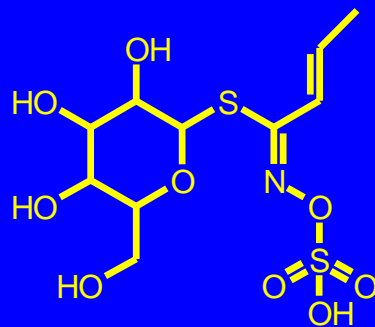
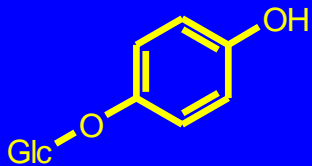
- Antifungal effect
 - Compounds located on the surface

» catechol, parthenolide, luteone



- Vacuolar located compounds

» arbutin, sinigrin, tomatine, avenacin, avenacoside



- Compounds with activity of animal hormones
 - steroids:
 - » ecdysones
 - terpenoids
 - estrogen-like compounds
 - » formononetin
- Plant compounds entering the symbiotic relationships
 - Lichens and root parasites
 - Chemical changes after infliction of microbes and parasites

Methods of Isolation and Analysis of Plant Secondary Metabolites

- Reasons for isolation and separation of compounds:
 - Initial extract shows promising biological activity
 - Chemotaxonomic studies
 - Isolation of commercially interesting compounds, for example for breeding
- Necessity clearly select targets
- To find an idea about the chemical character of the target compounds
 - Polarity, presence of acidic and basic functionalities, chemical relationship of single compounds

- To find an idea about the physical character of the target compounds
 - Melting point, boiling point, UV absorption and other spectral characteristics
- Make a thorough literature research.

‘It is not necessary to develop a bicycle again.’
- 1. Cultivation of medicinal plants, collection/harvest, drying, storage of drugs

The term „drug“ means unmodified or modified conserved or fresh plants or plant parts, animals or their parts or metabolic products of plants, animals and microorganisms.

- 2. Quality control
- 3. Isolation and separation of physiologically active compounds
- 4. Identification of physiologically active compounds

Preparative Separation

- Search in the available literature – make a literature research on the topic
- Optimization of methods
- Look for all known properties of target compounds.
- First make a trial with small amount of sample.
- Use sensitive methods.
- Partial material, do not work on the whole amount at once.
- Make a taxonomical search.

Analytical Separation

- Search in the available literature – make a literature research on the topic
- Choose optimal method, modify or confirm the functionality/ reliability.
- If necessary pre-purify the sample.

Phytochemical Literature

- Chemical abstracts, Biological abstracts
- Beilstein abstracts
- Phytochemical Analysis - phytochemical methods
- Journal of Natural Products, Phytochemistry, Planta Medica
- Journal of Chromatography, Journal of Liquid Chromatography, Chromatographia.

Cultivation of Medicinal plants, Collection/Harvest, Drying, Storage of Drugs

- Small consumer/individual person – collection of plants in wild nature versus mass production for business



- Active principles are usually present unevenly in all parts of plant.
- Proper term of harvest.
 - Concentration of content compounds changes in the life of plant, seasonal variations (concentration could change in seasons, weeks, hours).

- Menthol in spearmint only in older plants, young plant contain precursors only.
- Camphor collected from 40 year old trees.
- variability in the season:
 - » Subterranean parts are collected in autumn.
 - » Seeds and fruits from beginning till full maturity.
- Daily variability of content compounds depends on external factors:
 - » Changes of light and dark
 - » Temperature
 - » Humidity
- Collection of wild plants is the oldest way of harvest. Drug collected in this way does not standard content of active compounds.

- Standard cultural cultivation of medicinal plants is better for standard content of active compounds.
- Usage in pharmaceutical industry.
- Main conditions for successful propagation:
 - Choice of location.
 - Choice of cultivar.
 - Time of harvest.
 - Usage of breed cultivars with high resistance against diseases and noxious animals.
 - Resistance to frost and drought.
 - Economical aspects.

- **Phytotherapeutic usage:**
 - **Fresh plants:**
 - » To get compounds which could decompose by drying and storage.
 - » By-pass of energy-consuming drying
 - » Homeopathy
 - **Conserved plants:**
 - Rapid processing.
 - Stopping of undesired changes: treatment of alcohol vapors on harvested material, lyophilization- freeze drying.
 - In some cases – active compound synthesis arises after collection – post mortal synthesis - cardioactive glycosides.

- **Drying:**
 - Fastest and most economic way of conservation.
 - Bad drying can destroy material
 - Removal of water – we stop the invasion of microbes and fungi, stopping of unwanted metabolic procedures which could change compounds.
- Good drying procedure does not affect the level of active compounds.

- Well defined drying temperature:
- Usual drying temperatures
 - Leaves, herbs, flowers 30-40 °C.
 - Roots and bark 30- 65 °C.
 - Volatiles containing plants less than 35 °C.
 - Use of short-termed higher heating to stop enzymatic procedures and decomposition of unwanted compounds.
 - Sometimes opposite – wanted procedures, by enzymatic process could be created active content compounds – fermentation.
 - Fermentation can change the „design“ of drug, taste, color etc. (black tea).

- Time of drying:
 - Herbs, leaves, flowers several days.
 - Roots several weeks.
- Way of drying:
 - Directly on sun light.
 - Natural heat.
 - Well air conditioned room.
 - Special drying cabinets with controlled temperature.

- Regular Conditions of Storage:
 - Bad conditions – excessive humidity causing the invasion of molds (*Penicillium*, *Rhizopus* and *Eurotium*).
 - Important to maintain low air humidity – max. 65 %.
 - Direct sun light is bad.
 - Fluctuations in temperature.
 - between 5-15 °C.

Extraction Techniques

- Spontaneous isolation without extraction.
- Usually necessary at least one step of sample extraction and one step of sample purification.
- Not only separation from high molecular compounds, but also removal of low molecular compounds which can disturb the final steps of separation and purification.
- *Matrix effect.*