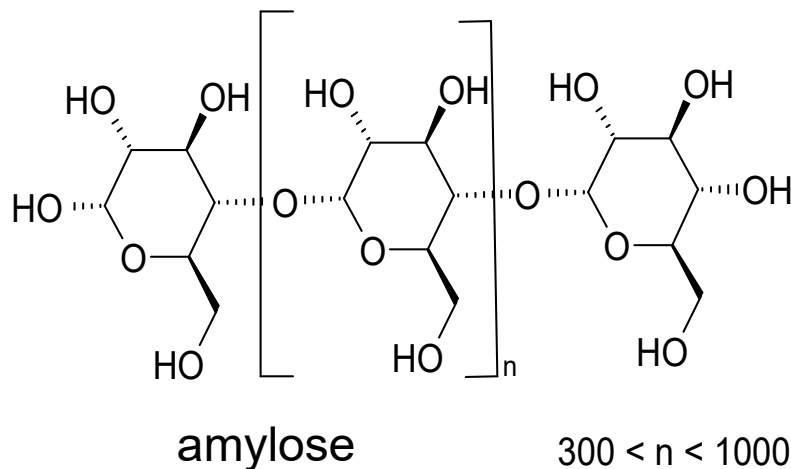


# Polysaccharides

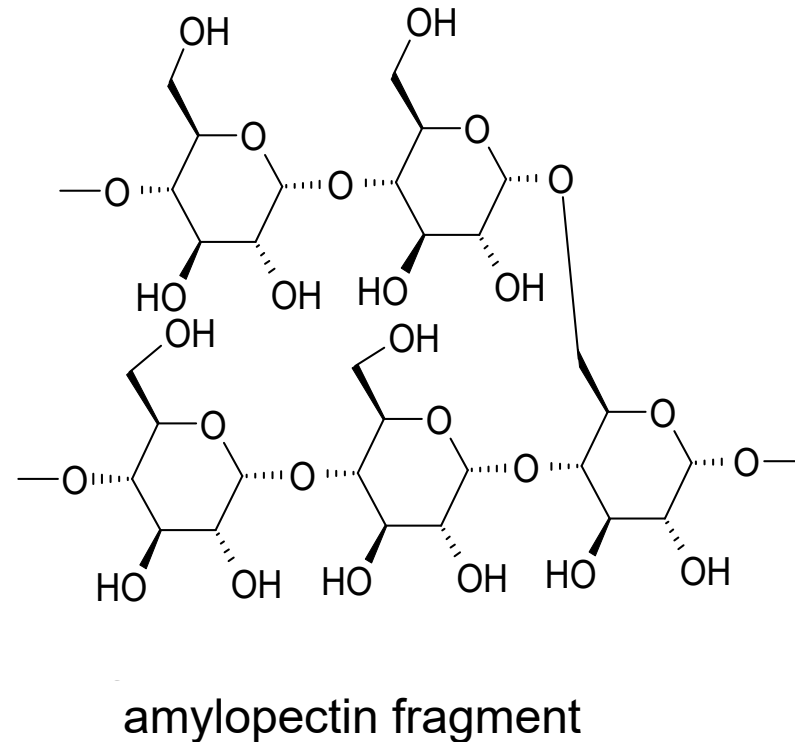
## Compounds supporting disintegration of tablets

### starch

- mixture of amylose and amylopectine [9005-25-8]
- PhEur: *Maydis amyllum*, *Oryzae amyllum*, *Solani amyllum*, *Tritici amyllum*
- USPNF: Corn starch, Potato starch, Tapioca starch, Wheat starch



(1→4)-α-D-glucopyranan

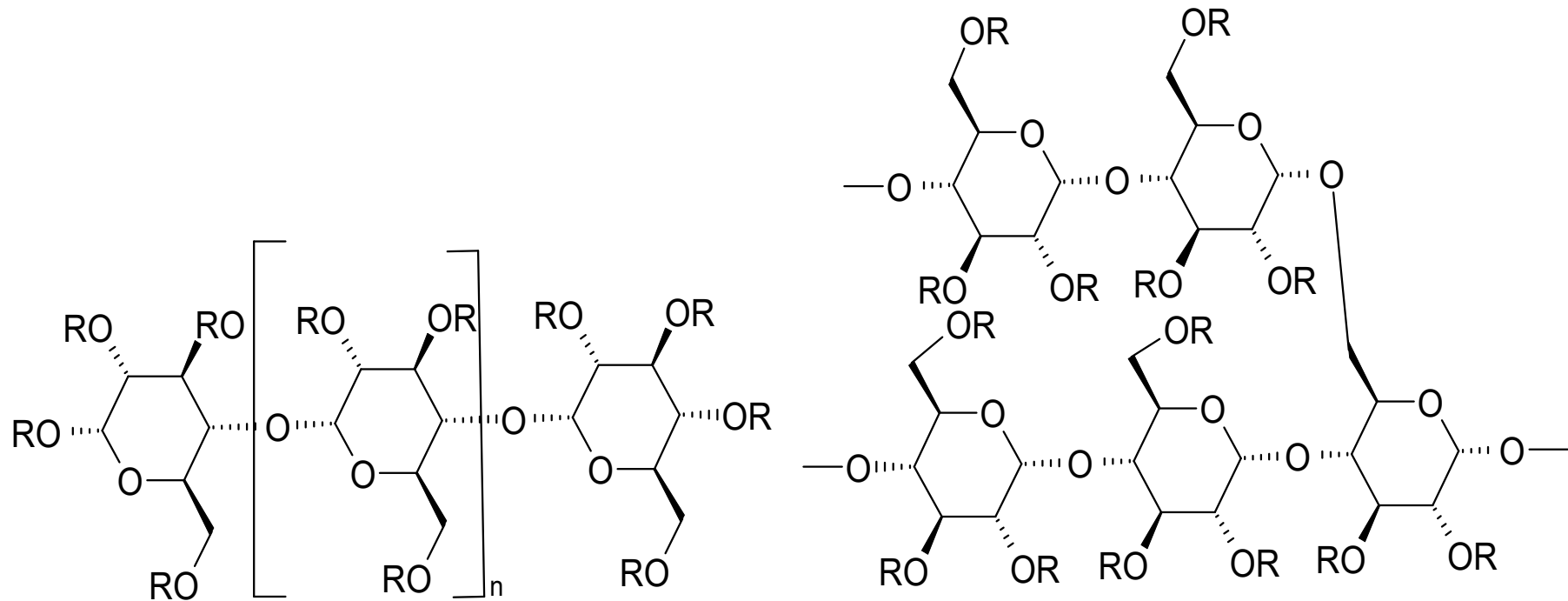


- trituration diluent, tablet granulate binder, tablet disintegrant

# Polysaccharides

Compounds supporting tablet disintegration and stabilizers of aggregate properties of coarsely dispersed systems

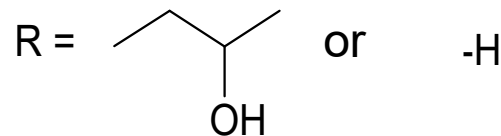
hydroxypropylstarch [113894-92-1, 9049-76-7], E1440



hydroxypropylamylose

$300 < n < 1000$

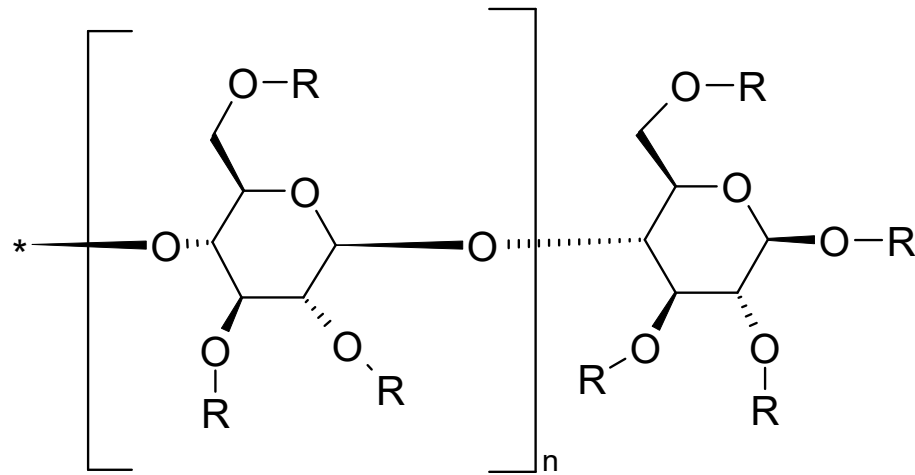
hydroxypropylamylopectine fragment



- a compound stabilizing suspensions and emulsions by increasing viscosity; capsule binder; disintegrant

# Constitutive excipients and stabilizers of aggregate properties of coarsely dispersed systems

## Celullulose and its ethers



**R = H cellulose** = (1→4)-β-D-glucopyranan; microcrystalline -  $n \approx 220$  – binder and tbl. disintegrant (Avicel<sup>®</sup>); amorphous  $n \approx 500$

**R = CH<sub>3</sub> (27 – 32 %) or H methylcellulose**  $50 \leq n \leq 1000$ , viscosity enhancer – emulsion and suspension stabilizer, tbl. binder and disintegrant (Methocel<sup>®</sup>, E461)

**R = CH<sub>2</sub>CH<sub>3</sub> ethylcellulose** compound for hydrophobic coating of tbl., binder and filler of tbl., viscosity enhancer (Aquacoat<sup>®</sup>, E462)

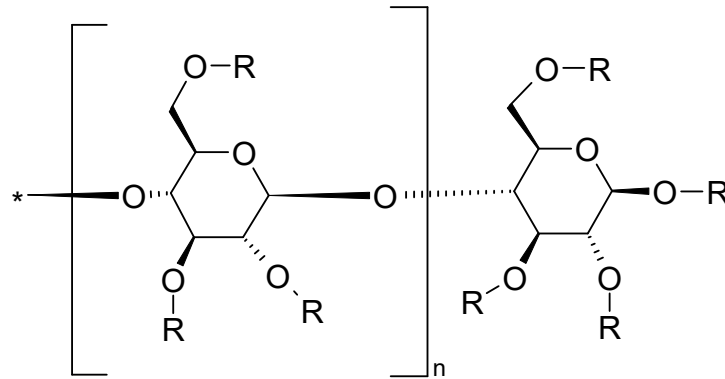
**R = (CH<sub>2</sub>CH<sub>2</sub>O)<sub>m</sub> H or H hydroxyethylcellulose** compound for coating of tbl., binder and filler of tbl., viscosity enhancer (Cellosize HEC<sup>®</sup>, Tylose PHA<sup>®</sup>)

**R = (CH<sub>2</sub>CH<sub>2</sub>O)<sub>m</sub> H or CH<sub>3</sub> or H hydroxyethylmethylcellulose** compound for coating of tbl., binder and filler of tbl., viscosity enhancer, suspension stabilizer (Tylopur MH<sup>®</sup>, Tylose MB<sup>®</sup>)

**R = (CH<sub>2</sub>CH(CH<sub>3</sub>)O)<sub>m</sub> H or H hydroxypropylcellulose**, hypolose: compound for coating of tbl., binder and filler of tbl., viscosity enhancer, suspension stabilizer (Klucel<sup>®</sup>, Nisso HPC<sup>®</sup>)

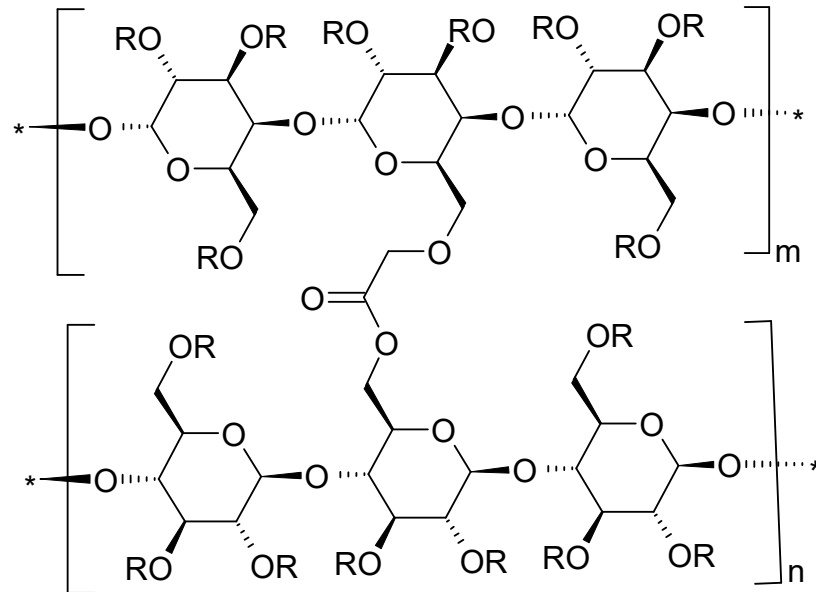
**R = (CH<sub>2</sub>CH(CH<sub>3</sub>)O)<sub>m</sub> H or CH<sub>3</sub> or H hypromellose** = hydroxypropylmethylcellulose – coating and film-forming substance, release rate controlling polymer, viscosity enhancer, suspension stabilizer (Benecel MHPC<sup>®</sup>, E464)

## Cellulose ethers continued



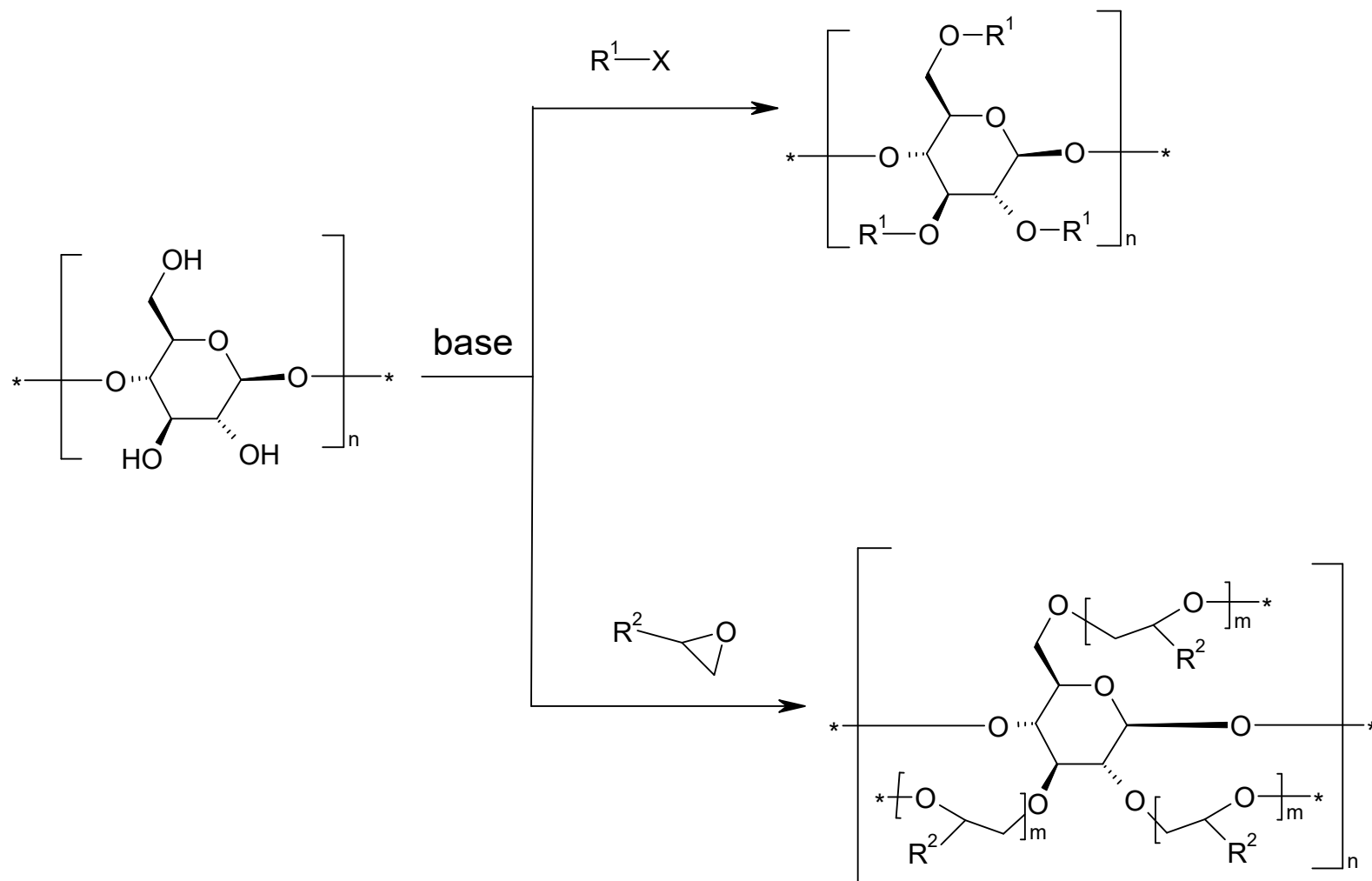
**carboxymethylcellulose**  $R = -CH_2COOH$  or  $-H$ , carmellose, most often  $Na^+$  or  $Ca^{2+}$  salt:

*Carmellosum natricum* and *calcicum PhEur* - binder and disintegrant of tbl. and cps., viscosity enhancer, suspension stabilizer, coating substance (E466, Tylose CB<sup>®</sup> -  $Na^+$  salt, Nymcel ZSC<sup>®</sup> -  $Ca^{2+}$  salt)



**crosscarmellose** – crosslinking with carboxymethyl fragments linked by ester and ether bonds

# Synthesis of saccharide ethers

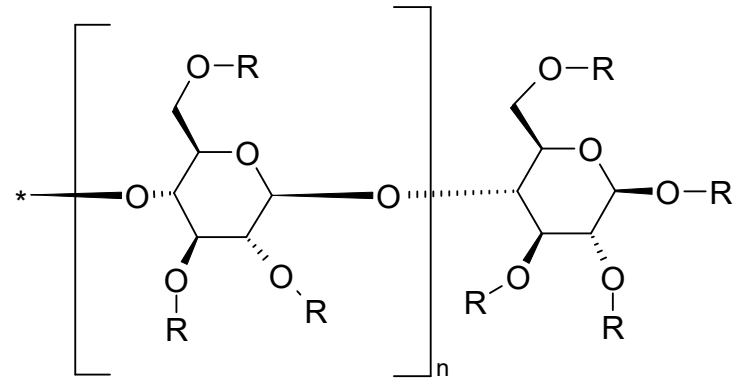


$X = -\text{Cl}, -\text{Br}, -\text{I}, -\text{OSO}_2\text{OR}^1, -\text{O}-\text{SO}_2\text{-C}_6\text{H}_4\text{-CH}_3$

$R^1 = -\text{CH}_3, -\text{CH}_2\text{CH}_3, -\text{CH}_2\text{COO}^-\text{Na}^+$

$R^2 = -\text{H}, -\text{CH}_3, 0 \leq m < \infty$

## Cellulose esters



**R = -NO<sub>2</sub> or -H** cellulose nitrate 4% solution in Et<sub>2</sub>O = *Collodium* = *Collodion* = *Pyroxylin*;

film-forming a gel-forming substance, wound sealant (*Sol. Novikov*)

**R = -OCCH<sub>3</sub> or -H** cellulose acetate = *Cellulosi acetas PhEur* compound for coating of tbl. and prolongation of drug release from them, diluent for tbl. and cps.

**R = -OCCH<sub>3</sub> or**  **or H** celacefate = cellulose acetate phthalate,

*Cellacefatum PhEur*: acetate groups

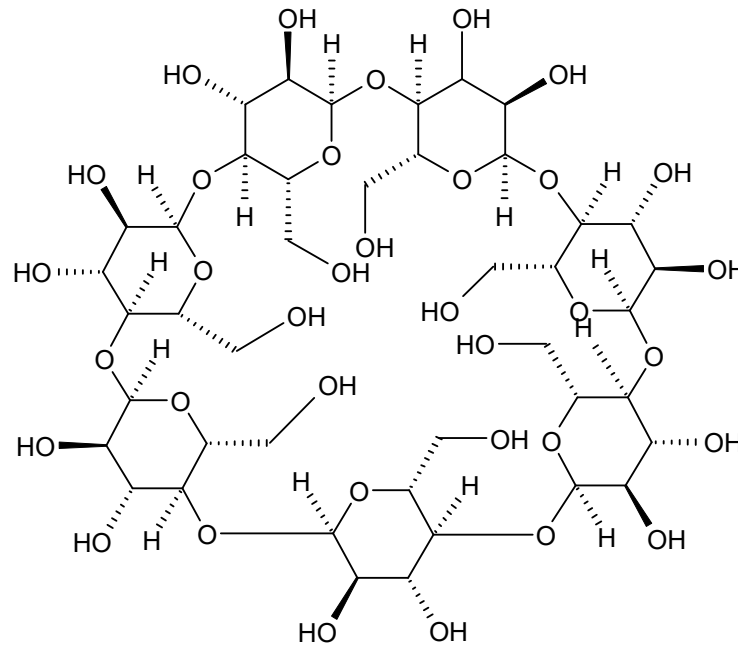
30.0 – 36.0 % hydrogenphthalate groups, 21.5 – 26 % film-forming compound for coating of tbl., matrix binder

**R = -OCCH<sub>3</sub> or -OC(CH<sub>2</sub>)<sub>2</sub>CH<sub>3</sub> or -H** celaburate = cellulose acetate butyrate,

*Cellaburatum PhEur*: 2.0 – 30.0 % acetyls, 16.0 – 53.0 % butyryls; hydrophobic contact lens material (*Cabucocon A*, [9004-36-8])

## Cyclic oligosaccharides Cyclodextrins and their derivatives

- cyclic glucose oligomers with 5 – 8 units: cyclo- $\alpha$ -(1 $\rightarrow$ 4)-D-oligoglucopyranosides



betadex

**alfadex** =  $\alpha$ -cyclodextrin = cycklo- $\alpha$ -(1 $\rightarrow$ 4)-D-hexaglukopyranoside

- parenteral DP, complexes only molecules of low  $M_r$

**betadex** =  $\beta$ -cyclodextrin = cyclo- $\alpha$ -(1 $\rightarrow$ 4)-D-heptaglukopyranoside

- Betadexum* PhEur (the only one listed in the pharmacopoeia); solubilizer – complexing agent, masking of unpleasant taste, reducing of local irritation (mucosa, eye), enhancer of permeation of a drug through mucosa etc.

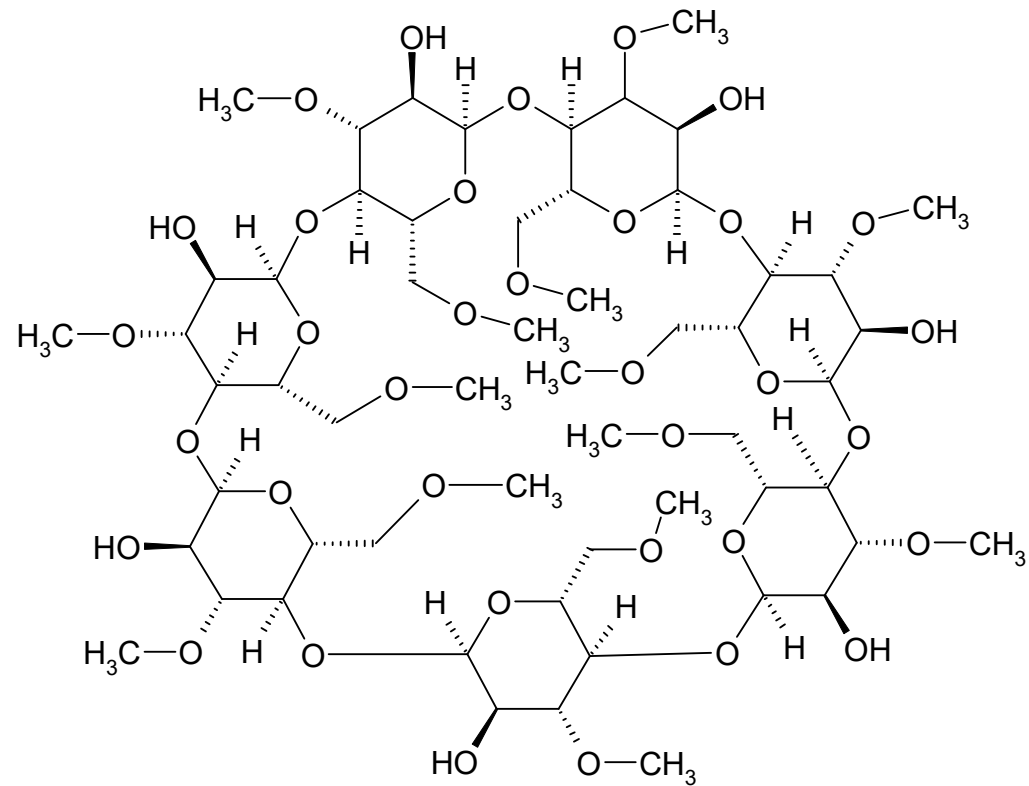
- nephrotoxic - not applicable for parenteral products

**gamadex** =  $\gamma$ -cyclodextrin = cycklo- $\alpha$ -(1 $\rightarrow$ 4)-D-octaglukopyranoside

## $\beta$ -cyclodextrin ethers

### **Dimethyl- $\beta$ -cyclodextrin**

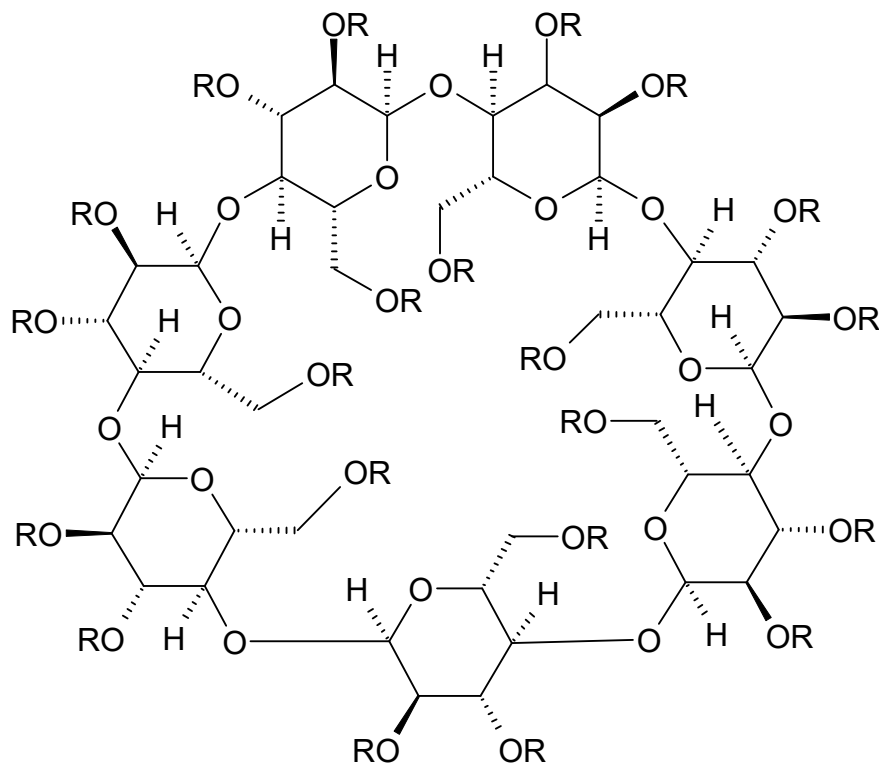
- selectively methylated in positions 2 and 6 of each anhydroglucose unit



- similar usage as betadex



## $\beta$ -cyclodextrin ethers continued



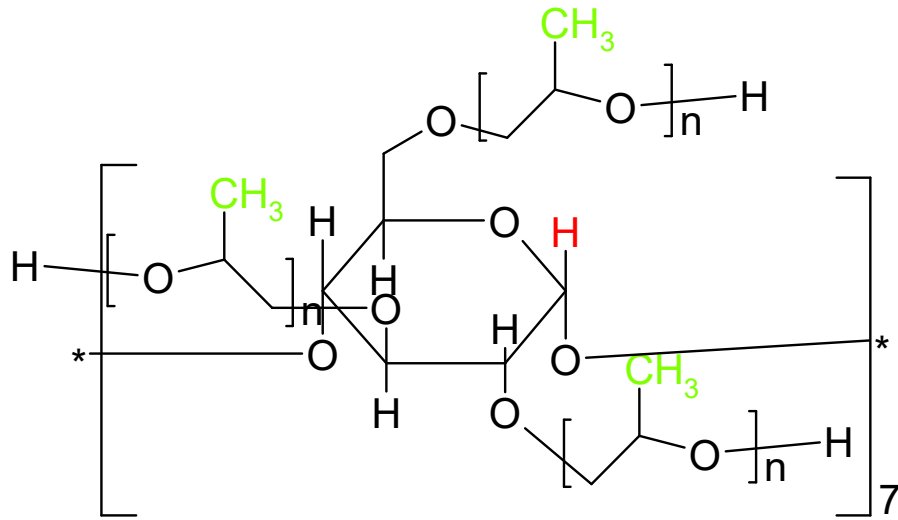
R =  $-(\text{CH}_2\text{CH}_2\text{O})_n\text{H}$  **2-hydroxyethyl- $\beta$ -cyclodextrin**

• R =  $-(\text{CH}_2\text{CH}(\text{CH}_3)\text{O})_n\text{H}$  **2-hydroxypropyl- $\beta$ -cyclodextrin**

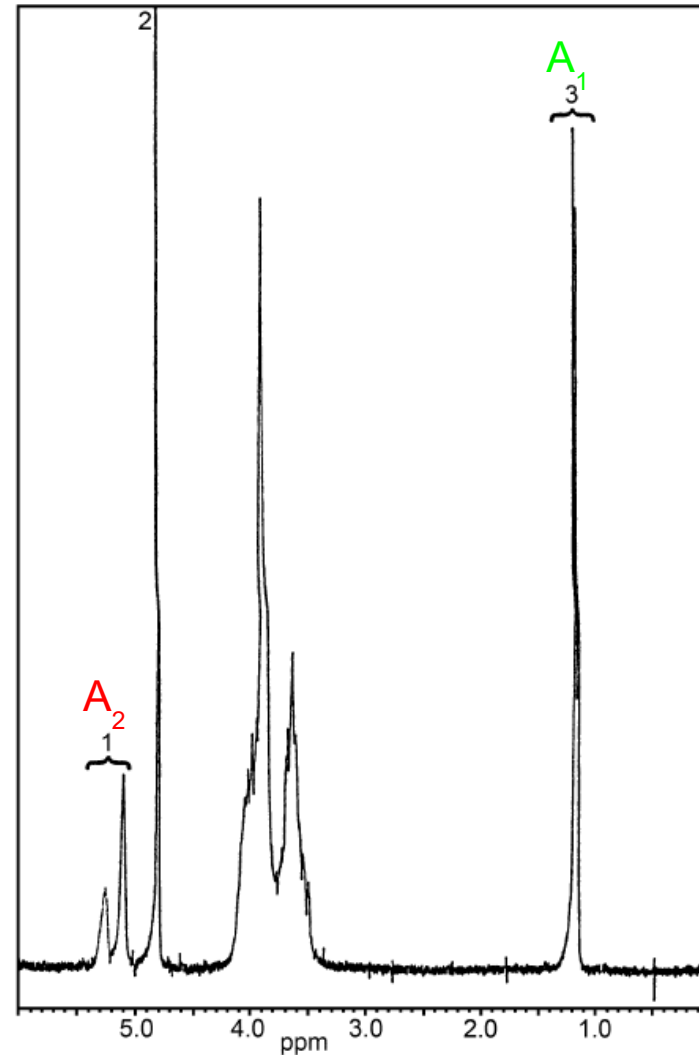
• more suitable for parent. applications than betadex: not nephrotoxic

• *Hydroxypropylbetadexum PhEur*: molar substitution MS; 0.40  
 $\leq \text{MS} \leq 1.50$  is a number of 2-hydroxypropyl units per 1  
 (anhydro)glucose unit, determined by  $^1\text{H-NMR}$

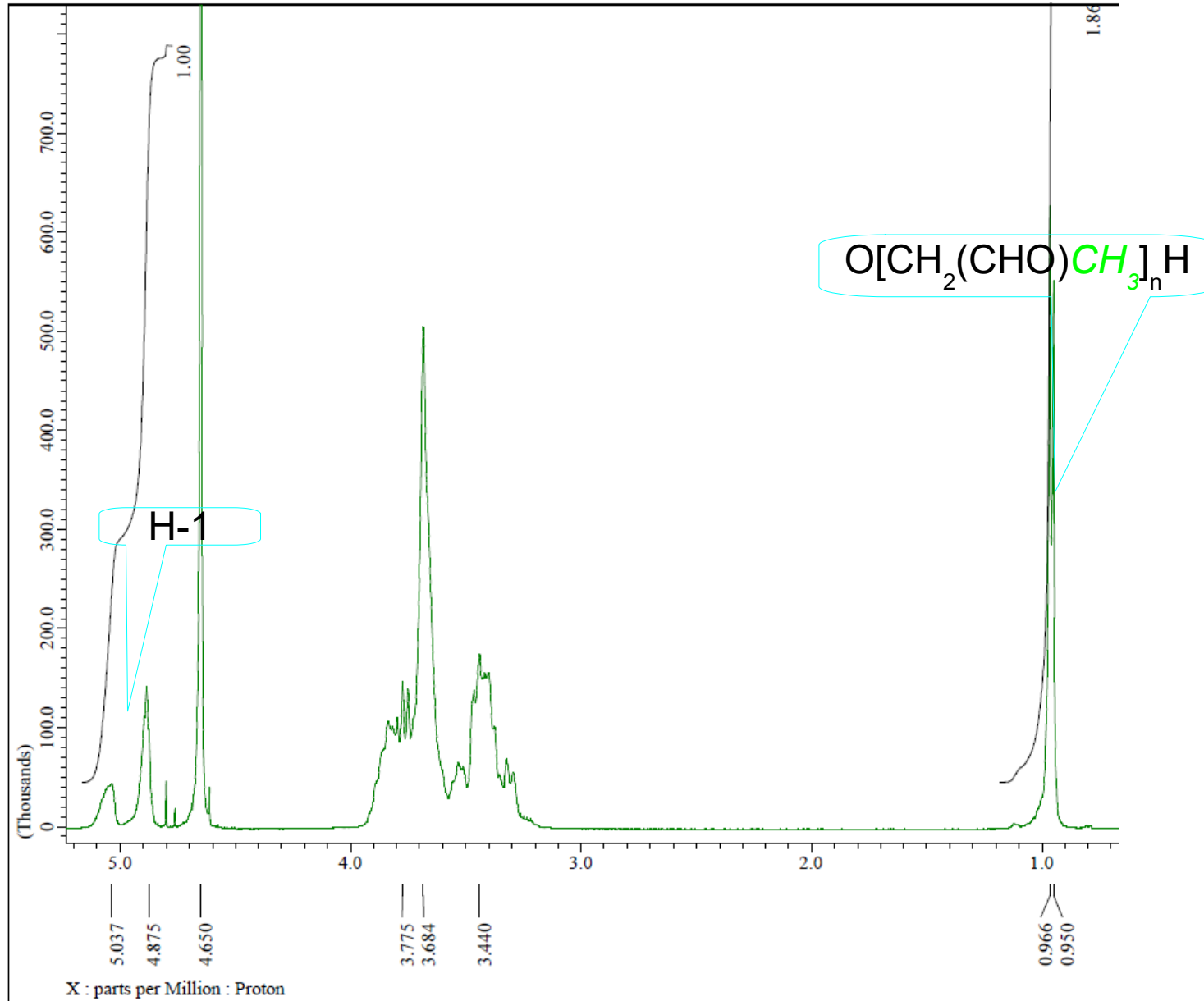
$$MS = \frac{A_1}{3A_2}$$



- in D<sub>2</sub>O
- MS is calculated from the ratio between the signal from the 3 protons of the methyl group that is part of the hydroxypropyl group and the signal from the proton attached to the C<sub>1</sub> carbon (glycosidic proton) of the anhydroglucose units.

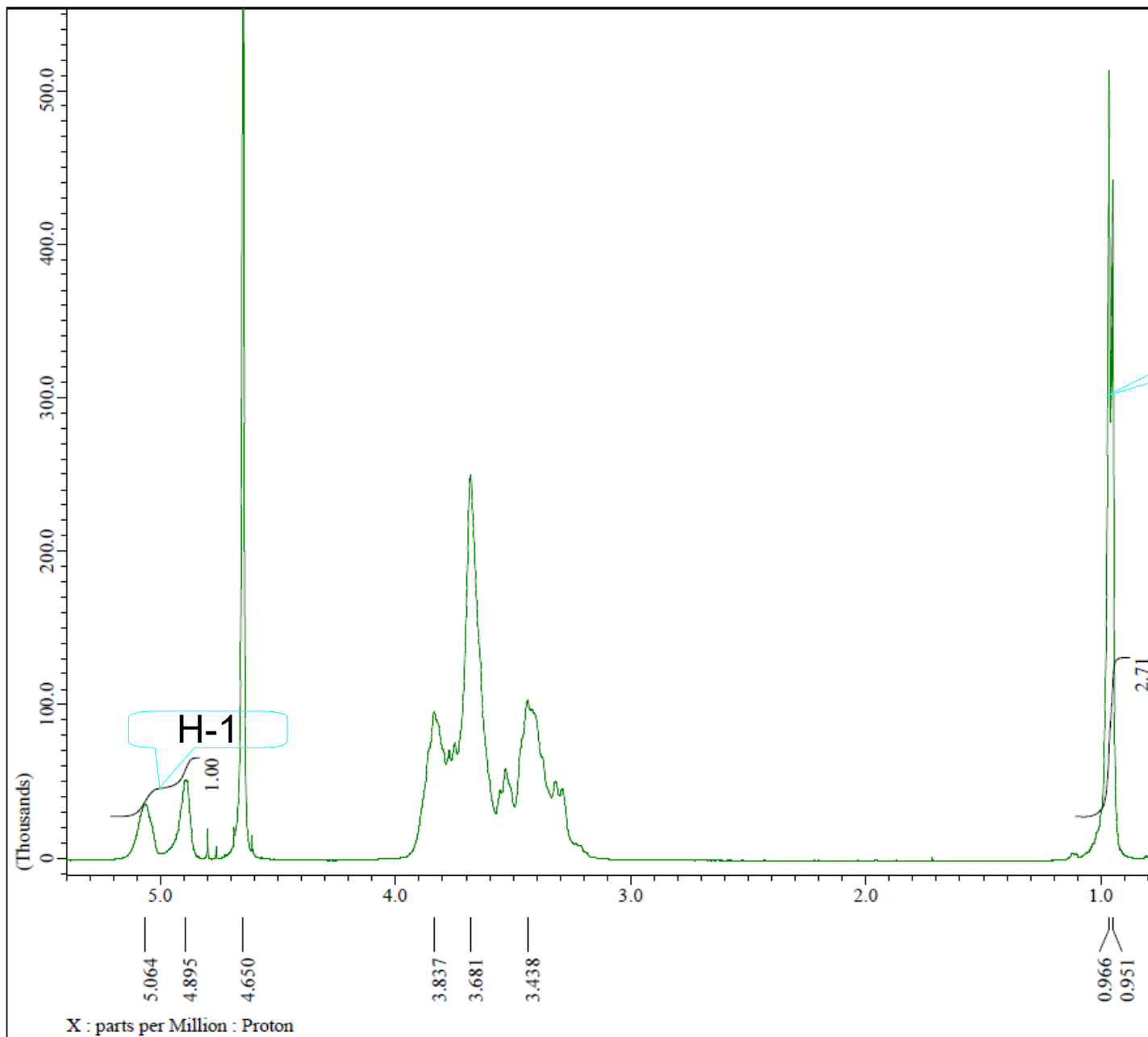


A practical example of molar substitution of hydroxypropylbetadex  
Kleptose HPB, batch No. E0266, EZC400R 400 MHz, D<sub>2</sub>O



MS = 1.83/3 = **0.62** corresponds with the pharmacopoeia

A practical example of molar substitution of hydroxypropylbetadex  
Kleptose HP, batch No. E0010, EZC400R 400 MHz, D<sub>2</sub>O



Substituted oligo- and polymers of glucose of ether type in PhEur and characterization of their molar substitution

Compound	Expression of molar substitution
Hydroxypropylbetadex	from ratio of areas in $^1\text{H-NMR}$ spectrum
Carmelose (carboxymethylcellulose)	-
Carmelose sodium salt	by means of content of Na: 6.5 – 10.8 %
Carmelose calcium salt	-
Low substituted carmelose sodium salt	by means of content of Na: 2,0 – 4,0 %
Methylcellulose	content of methoxy groups 26.0 – 33.0 % - determination as $\text{CH}_3\text{I}$ by means of GC
Ethylcellulose	content of ethoxy groups 44.0 – 51.0 % - determination as $\text{CH}_3\text{CH}_2\text{I}$ by means of GC
Hyprolose (hydroxypropylcellulose)	-
Hypromelose (hydroxypropylmethylcellulose)	content of both methoxy- and 2-hydroxypropoxy groups defined for each type of substitution, determination as $\text{CH}_3\text{I}$ and $\text{ICH}_2\text{CH}(\text{I})\text{CH}_3$ by means of GC
Carboxymethylstarch A, B, C	by means of content of Na: A 2.8 – 4.2 %; B 2.0 – 3.4 %; C 2.8 – 5.0 %
Hydroxypropylstarch	0.7 – 5.0 % hydroxypropyl groups; partial hydrolysis by $\text{DCI} / \text{D}_2\text{O}$ directly in NMR tube, measurement of $^1\text{H-NMR}$ , comparison of areas of dublet of methyl of 2-hydroxypropyl group and internal standard 3- trimethylsilylpropan-1-sulfonic acid

Substituted oligo- and polymers of glucose of **ester** type in PhEur and characterization of their molar substitution

Compound	Expression of molar substitution
Cellulose acetate	Content of acetyl groups 29.0 – 44.8%: splitting off $\text{CH}_3\text{CO}$ groups with NaOH, titration of unreacted NaOH with $\text{H}_2\text{SO}_4$
Celaburate (cellulose acetate butyrate)	2.0 – 30.0 % acetyl groups and 16.0 – 53.0 % butyryl groups; determination of acetic and butanoic acids by HPLC after hydrolysis
Celacefate (cellulose acetate phthalate)	30.0 – 36.0 % phthaloyl groups and 21.5 – 26.0 % acetate groups; determination of hydrogenphthalate groups by a direct titration with NaOH; acetate groups: splitting off $\text{CH}_3\text{CO}$ groups with NaOH, titration of unreacted NaOH with $\text{H}_2\text{SO}_4$