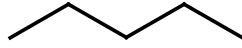
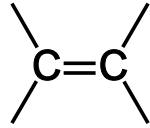


Funkční skupiny



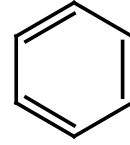
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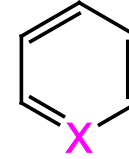
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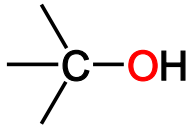
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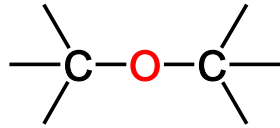
aromatická
sloučenina



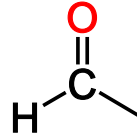
heteroaromatická
sloučenina



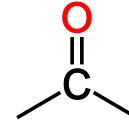
alkohol (1°,2°,3°)



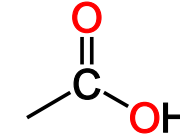
ether



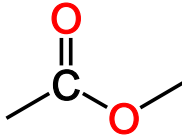
aldehyd



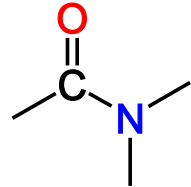
keton



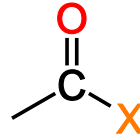
karboxylová
kyselina



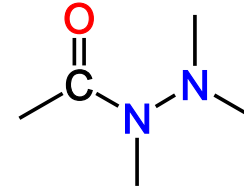
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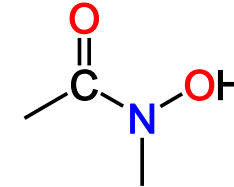
amid (1°,2°,3°)



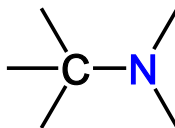
acylhalogenid



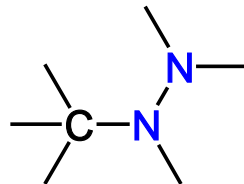
hydrazid



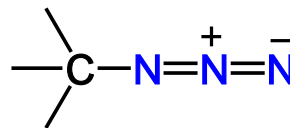
hydroxamová
kyselina



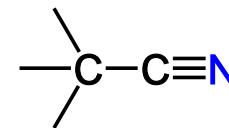
amin (1°,2°,3°)



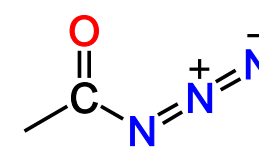
hydrazin



azid



nitril



acylazid

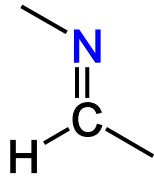
2. Polární kovalentní vazby; kyseliny a báze

1

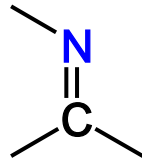
Organická chemie



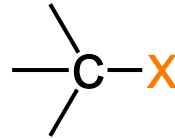
Funkční skupiny



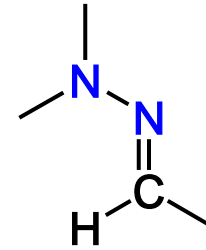
aldimin



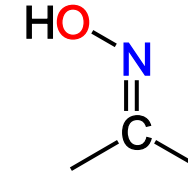
ketimin



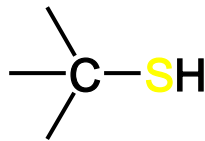
alkylhalogenid (1°,2°,3°)



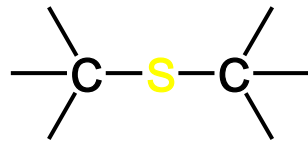
hydrazon



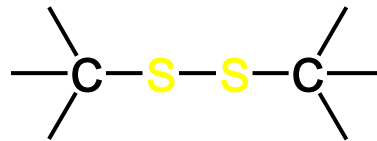
oxim



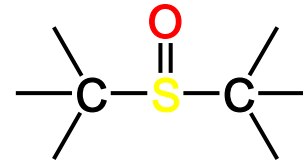
thiol (merkaptan)



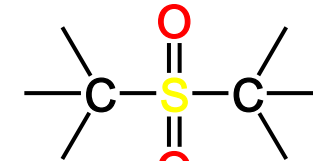
sulfid



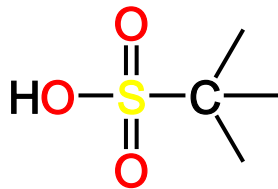
disulfid



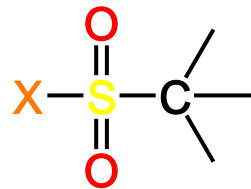
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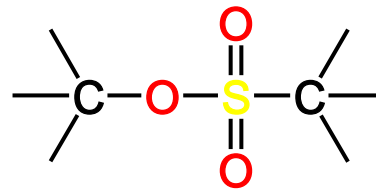
sulfon



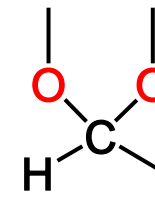
sulfonová
kyselina



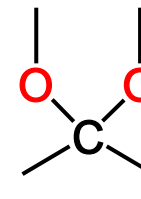
sulfonylhalogenid



sulfonát ester



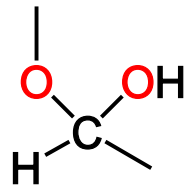
acetal



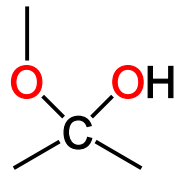
ketal



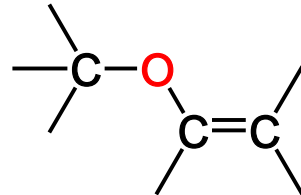
Funkční skupiny



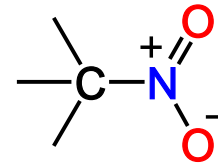
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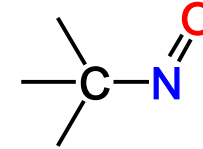
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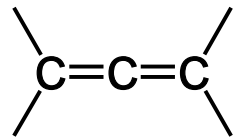
enol ether



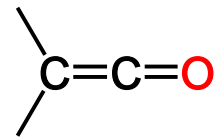
nitrosloučenina



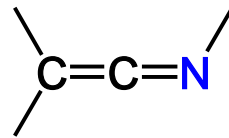
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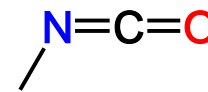
alen



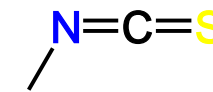
keten



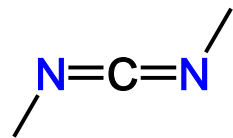
ketenimin



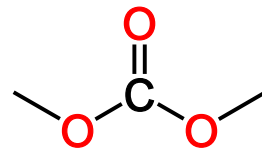
isokyanát



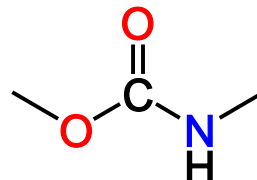
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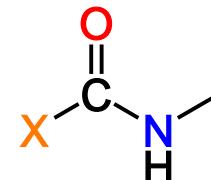
karbodiimid



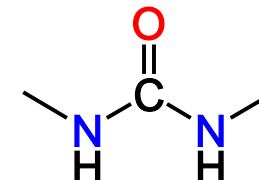
karbonát



karbamát



karbamoylhalogenid



močovina

