

Redox homeostasis & oxidative stress

Redox homeostasis

- natural levels of oxidants (O_2) and antioxidants in each cell

Disruption of redox homeostasis

- > depletion of oxygen: metabolism disruption, acidosis in tissues, cell necrosis
- > overproduction of oxidants: depletion of antioxidants, oxidation of biological molecules (membranes, proteins, DNA ...)
-> disruption of signals (GSH), carcinogenesis, health problems, necrosis ...
= oxidative stress

Overproduction of oxidants

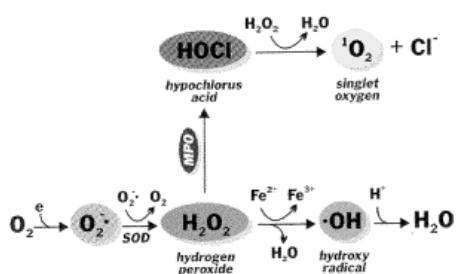
Oxygen – principal molecule in living organisms

Oxygen increase or reactive derivatives -> toxicity

ROS = Reactive Oxygen Species: Sources

- production in mitochondria (byproducts)
- redox-cycling (quinones of xenobiotics)
- Fenton-reaction (metals)
- oxidations mediated via MFO (CYP)
- depletion of antioxidants (reactive molecules)

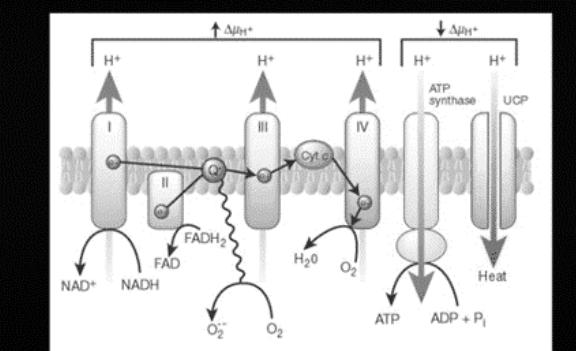
Reactive Oxygen Species (ROS)



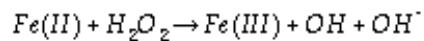
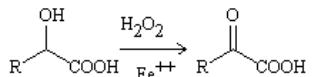
Druh	Symbol	Položas života (s) pri 37 °C
Superoxid	O_2^-	1×10^{-6}
Hydroxylový radikál	OH^\cdot	1×10^{-9}
Alkoxylový radikál	RO^\cdot	1×10^{-6}
Peroxylový radikál	ROO^\cdot	1×10^{-2}
Singletový kyslík	O_2	1×10^{-6}
Molekulární kyslík	O_2	$> 10^2$

ROS & mitochondria

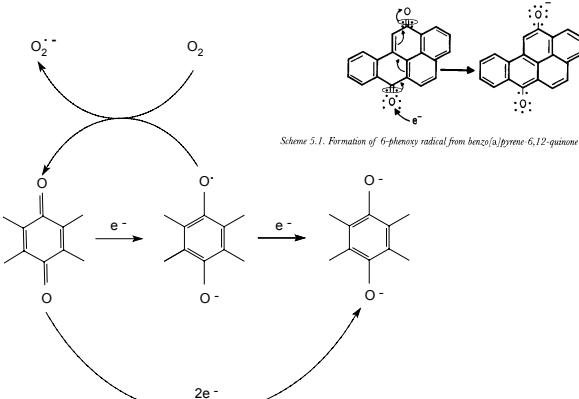
Glucose-Derived ROS: Mitochondrial Electron Transport System



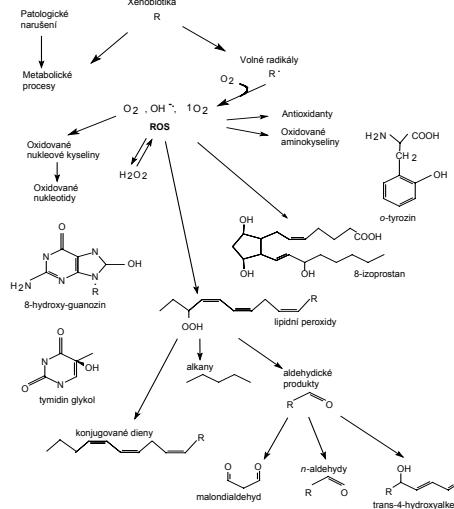
Fenton reaction



Redox-cycling and ROS formation



Toxicity of ROS



Examples of chemical-induced oxidative stress

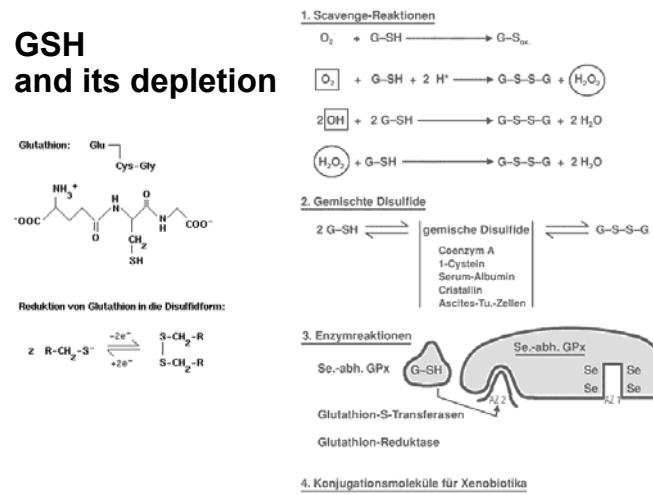
- Metals: fenton reaction $\rightarrow \text{OH}^*$

- Depletion of GSH:

reactive molecules, GST-conjugation,
metals: SH oxidation ...

- Redox-cycling chemicals: oxy-PAHs

GSH and its depletion



Biomarkers of oxidative damage

Poškození	Produkt	Stanovení	Citace
fosfolipidy	MDA	TBARS assay, HPLC, HPLC s UV-detekcí	Draper et al. 1993, Bird et al. 1983, Selim 1977
DNA	8-OH-dG	HPLC, metoda s využitím imunoafinitní izolace	Degan et al. 1991, Loft et al. 1992
proteiny	o-Tyr	spektrofotometricky, HPLC, MS	Deneshvar et al. 1997, De Zwart et al. 1998

Výsvětlivky:

MDA malondialdehyd
8-OH-dG 8-hydroxy-2'-deoxyguanozin
o-Tyr orthotirozin
TBARS reaktivní látky s kyselinou thiobarbiturovou
HPLC vysokotlaková kapalinová chromatografie
MS hmotnostní spektrometrie

DNA damage mutagenicity and genotoxicity

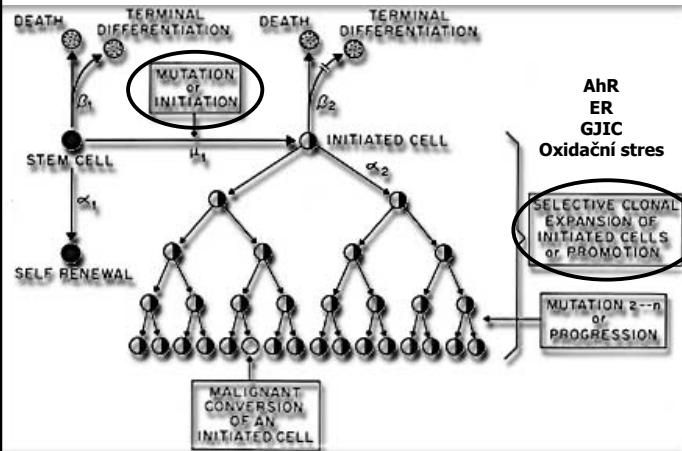
DNA:

- principal molecule for life of the cell
- structure and function carefully checked
- changes rapidly repaired
- irreversible changes -> cell death (*apoptosis*)

Mutagenesis

- changes in the sequences of deoxynucleotides
 - deletions/insertions: changes in reading frame
 - exchanges of nucleotides: changes in aminoacids
- natural mutations (billions of nucleotides/day)
 - : variability in genomes; reparations
- **chemical-induced mutagenesis**

IMPORTANT PROCESSES IN CANCEROGENESIS



Chemical induced DNA damage

Bases analogs

- incorporation into DNA during replication
(5-Br-Uracil: AT -> GC)

HNO₂, HSO₃, Hydroxylamine, Methoxyamine-

- deamination of bases (GC -> AT)

Alkylsulphates, N-nitroso-alkyles, cis-platinum

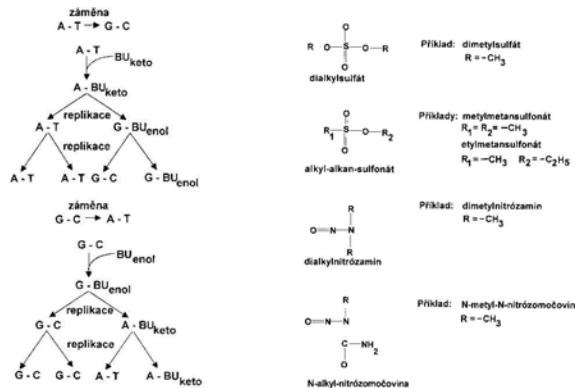
- alkylation of bases; crosslinks of dsDNA

Polycyclic aromatic hydrocarbons (PAHs) & derivatives

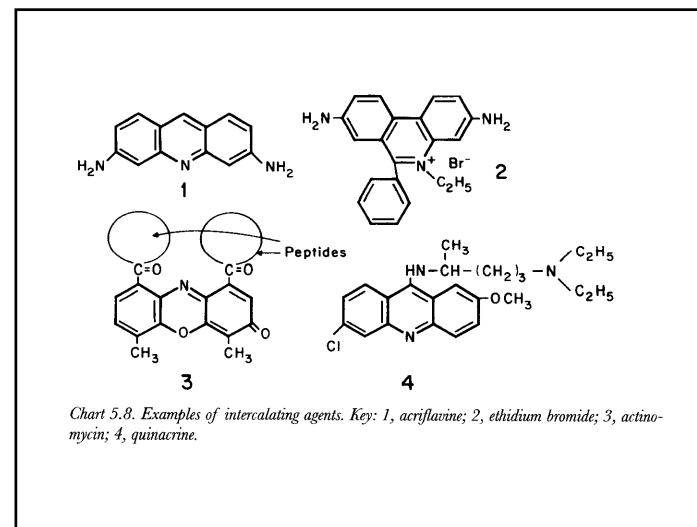
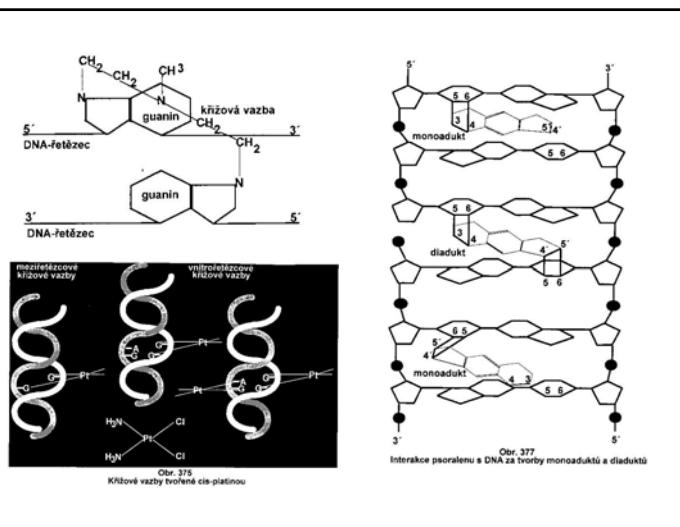
(N-acetyl-2-aminoanthracene (AAF), benzo[a]pyrene)

Mycotoxins (aflatoxins)

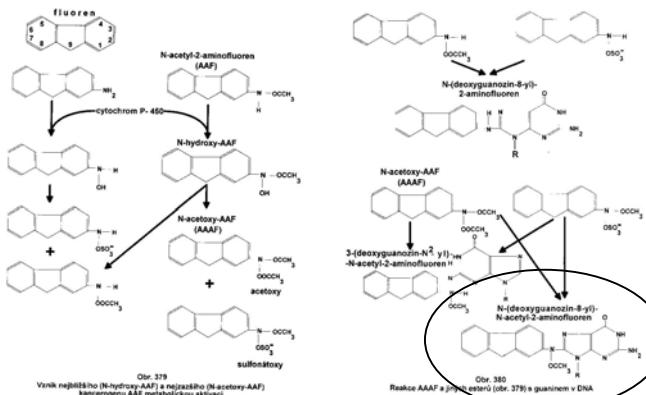
- require metabolic activation by CYPs
- adduct formation with DNA (*biomarkers*)



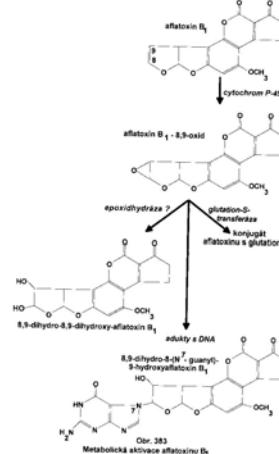
Zámeny páru bázi v DNA pod mutagenním účinkem bromuracilu



Metabolic activation of PAH and DNA-adduct formation



Metabolic activation of aflatoxin and formation of DNA-adducts



Does **chemically-induced genotoxicity** has effects in vivo ?

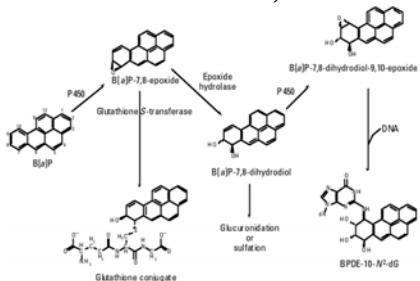
- adducts from mitochondrial DNA ?
- distance between „source of radicals“ and nuclear DNA ?
- protection mechanisms (mutation -> death)

Rubin (2002) *Oncogene* 21:7392

Thilly (2003) *Nature Genetics* 34(3):255

Mutations are not caused by chemicals

Chemicals only allow „unveil“ previously existing mutations in nuclear DNA (*non-genotoxic events cause cancer !!!*)



Physical factors & DNA damage

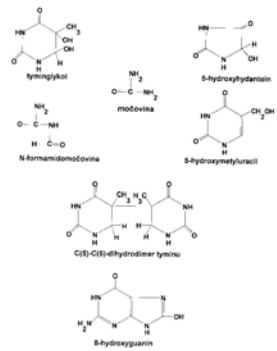
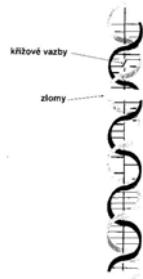
Ionizing radiation

- direct interaction with hydrogen atoms in water (and bases)
-> OH^{*} radicals; H₂O₂, O₂⁻
- oxidation of bases; dimerization ...

UV radiation

- interaction with aromatic cycles (bases)
- base dimerization (T=T)

Ionizing radiation effects on DNA



DNA repair

Damage of DNA is carefully controlled
constitutively expressed proteins

Changes in DNA

induction of reparation enzymes ("SOS-repair")
= biomarker of DNA damage

