

Redox homeostasis & oxidative stress

Redox homeostasis

- natural levels of oxidants (O_2) and antioxidants in each cell

Disruption of redox homeostasis

- > depletion of oxygen: metabolism disruption, acidosis in tissues, cell necrosis
- > overproduction of oxidants: depletion of antioxidants, oxidation of biological molecules (membranes, proteins, DNA ...)
-> disruption of signals (GSH), carcinogenesis, health problems, necrosis ...
= oxidative stress

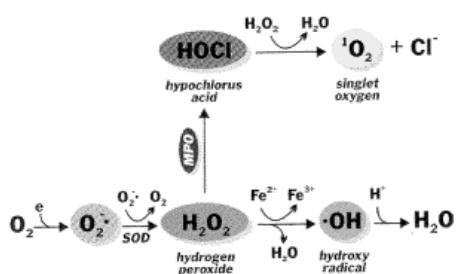
Overproduction of oxidants

Oxygen – principal molecule in living organisms
Oxygen increase or reactive derivatives -> toxicity

ROS = Reactive Oxygen Species: Sources

- production in mitochondria (byproducts)
- redox-cycling (quinones of xenobiotics)
- Fenton-reaction (metals)
- oxidations mediated via MFO (CYP)
- depletion of antioxidants (reactive molecules)

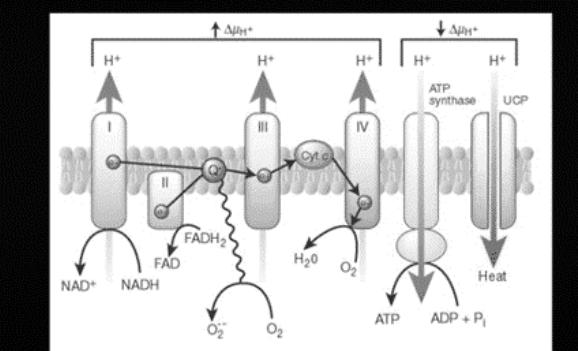
Reactive Oxygen Species (ROS)



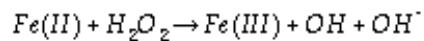
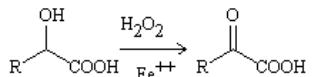
Druh	Symbol	Položas života (s) pri 37 °C
Superoxid	O_2^-	1×10^{-6}
Hydroxylový radikál	OH^-	1×10^{-9}
Alkoxylový radikál	RO^\cdot	1×10^{-6}
Peroxylový radikál	ROO^\cdot	1×10^{-2}
Singletový kyslík	O_2	1×10^{-6}
Molekulární kyslík	O_2	$> 10^2$

ROS & mitochondria

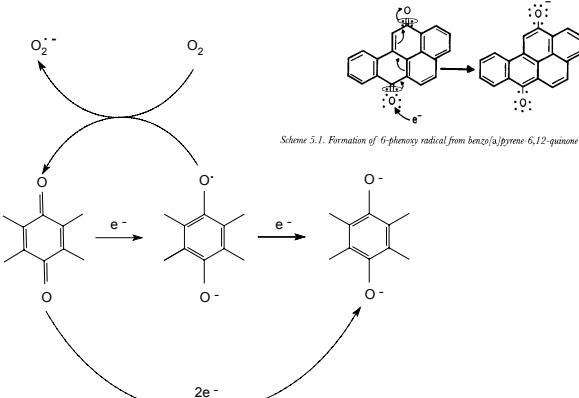
Glucose-Derived ROS: Mitochondrial Electron Transport System



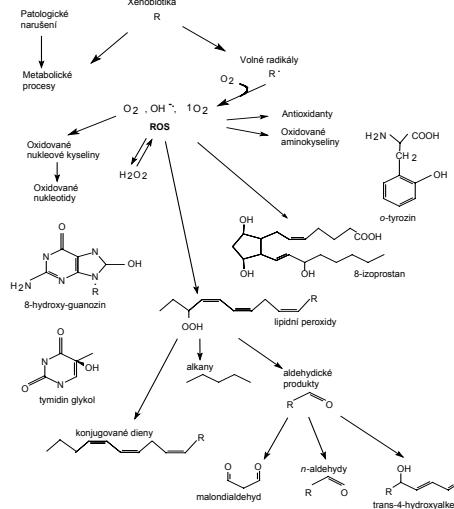
Fenton reaction



Redox-cycling and ROS formation



Toxicity of ROS



Examples of chemical-induced oxidative stress

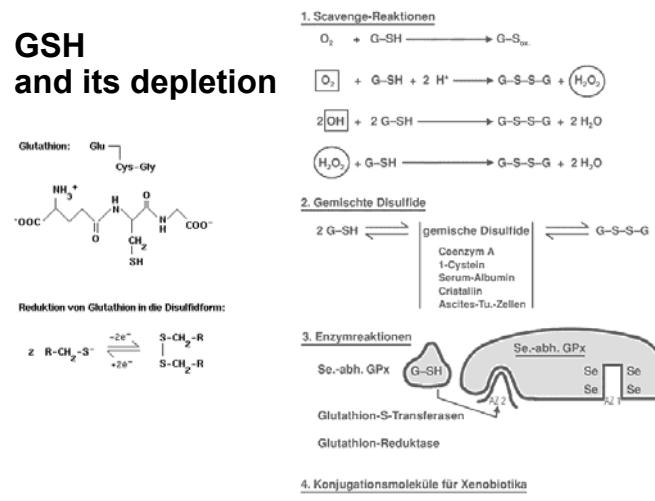
- Metals: fenton reaction $\rightarrow \text{OH}^*$

- Depletion of GSH:

reactive molecules, GST-conjugation,
metals: SH oxidation ...

- Redox-cycling chemicals: oxy-PAHs

GSH and its depletion



Biomarkers of oxidative damage

Poškození	Produkt	Stanovení	Citace
fosfolipidy	MDA	TBARS assay, HPLC, HPLC s UV-detekcí	Draper et al. 1993, Bird et al. 1983, Selim 1977
DNA	8-OH-dG	HPLC, metoda s využitím imunoafinitní izolace	Degan et al. 1991, Loft et al. 1992
proteiny	o-Tyr	spektrofotometricky, HPLC, MS	Deneshvar et al. 1997, De Zwart et al. 1998

Výsvětlivky:

MDA malondialdehyd
 8-OH-dG 8-hydroxy-2'-deoxyguanozin
 o-Tyr orthotirozin
 TBARS reaktivní látky s kyselinou thiobarbiturovou
 HPLC vysokotlaková kapalinová chromatografie
 MS hmotnostní spektrometrie

DNA damage mutagenicity and genotoxicity

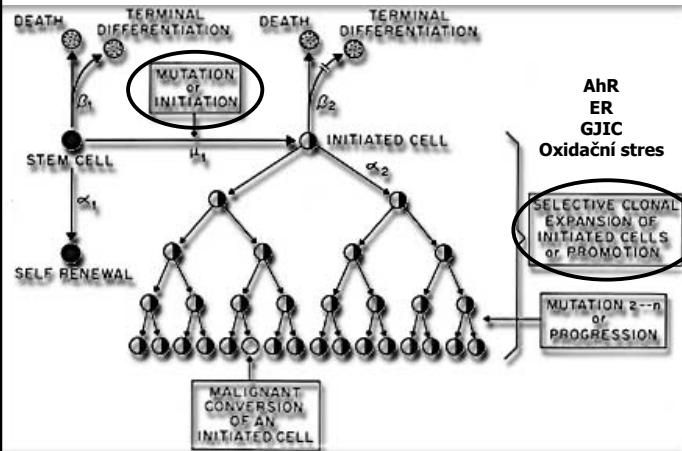
DNA:

- principal molecule for life of the cell
- structure and function carefully checked
- changes rapidly repaired
- irreversible changes -> cell death (*apoptosis*)

Mutagenesis

- changes in the sequences of deoxynucleotides
 - deletions/insertions: changes in reading frame
 - exchanges of nucleotides: changes in aminoacids
- natural mutations (billions of nucleotides/day)
 - : variability in genomes; reparations
- **chemical-induced mutagenesis**

IMPORTANT PROCESSES IN CANCEROGENESIS



Trosko and Ruch 1998, Frontiers In Bioscience 3:d208

Chemical induced DNA damage

Bases analogs

- incorporation into DNA during replication
(5-Br-Uracil: AT -> GC)

HNO₂, HSO₃, Hydroxylamine, Methoxyamine-

- deamination of bases (GC -> AT)

Alkylsulphates, N-nitroso-alkyles, cis-platinum

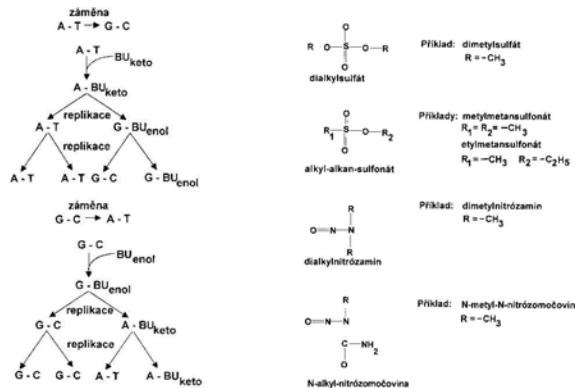
- alkylation of bases; crosslinks of dsDNA

Polycyclic aromatic hydrocarbons (PAHs) & derivatives

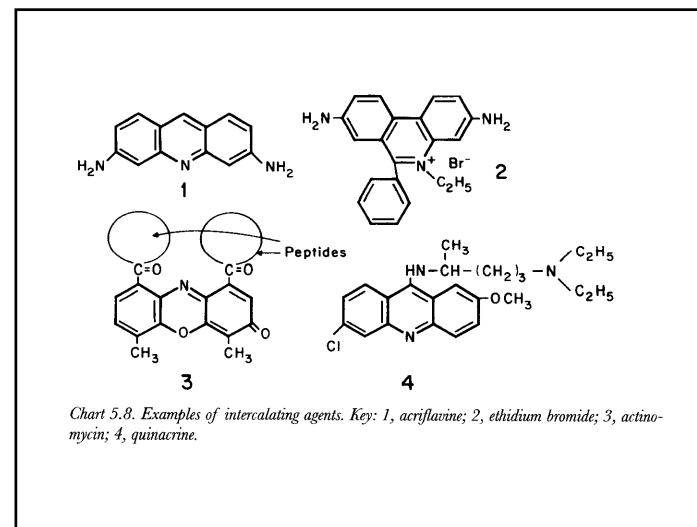
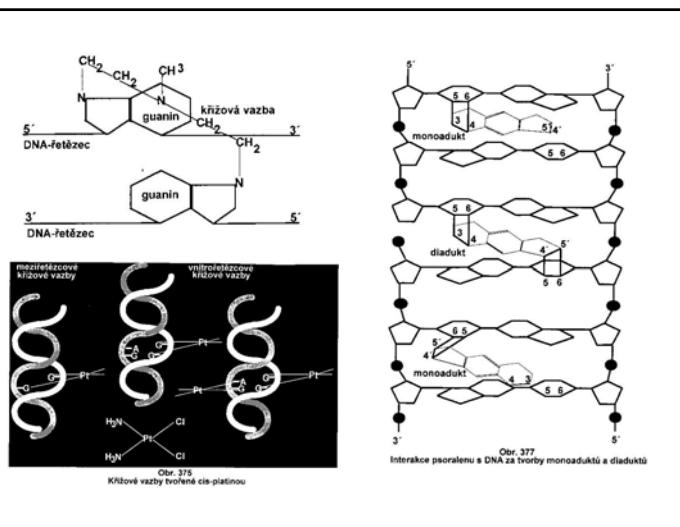
(N-acetyl-2-aminoanthracene (AAF), benzo[a]pyrene)

Mycotoxins (aflatoxins)

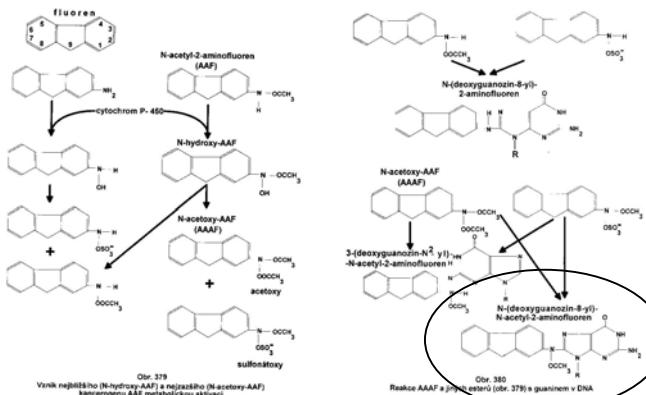
- require metabolic activation by CYPs
- adduct formation with DNA (*biomarkers*)



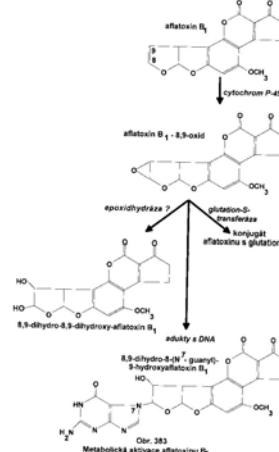
Zámeny páru bázi v DNA pod mutagenním účinkem bromuracilu



Metabolic activation of PAH and DNA-adduct formation



Metabolic activation of aflatoxin and formation of DNA-adducts



Does **chemically-induced genotoxicity** has effects in vivo ?

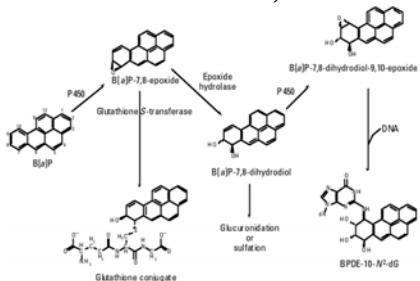
- adducts from mitochondrial DNA ?
- distance between „source of radicals“ and nuclear DNA ?
- protection mechanisms (mutation -> death)

Rubin (2002) *Oncogene* 21:7392

Thilly (2003) *Nature Genetics* 34(3):255

Mutations are not caused by chemicals

Chemicals only allow „unveil“ previously existing mutations in nuclear DNA (*non-genotoxic events cause cancer !!!*)



Physical factors & DNA damage

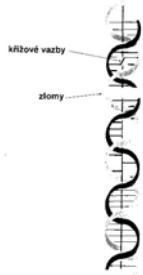
Ionizing radiation

- direct interaction with hydrogen atoms in water (and bases)
-> OH^{*} radicals; H₂O₂, O₂⁻
- oxidation of bases; dimerization ...

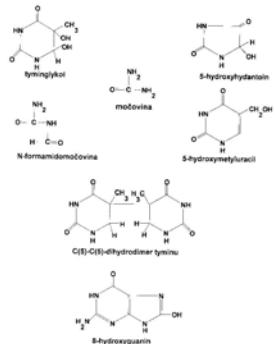
UV radiation

- interaction with aromatic cycles (bases)
- base dimerization (T=T)

Ionizing radiation effects on DNA



Obr. 384
Účinek ionizujícího záření na DNA



Obr. 385
Změny bází vyvolané ionizujícím zářením za přítomnosti kyslíku

DNA repair

Damage of DNA is carefully controlled
constitutively expressed proteins

Changes in DNA

induction of reparation enzymes ("SOS-repair")
= biomarker of DNA damage

