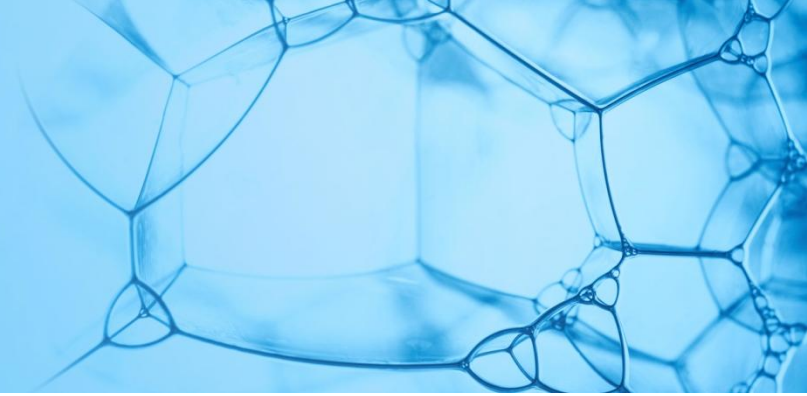


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PROTEIN ENGINEERING

10. EXAMPLE OF UTILIZING PROTEIN ENGINEERING TO ENHANCE ENZYME STABILITY

Loschmidt Laboratories
Department of Experimental Biology
Masaryk University, Brno

Outline

- ❑ Process design criteria
- ❑ Engineering enzyme stability and resistance to an organic cosolvent by modification of residues in the access tunnel
 - motivation
 - aims
 - results
 - conclusions
- ❑ Method of protein stabilization – patent application

Process design criteria

- higher activity at process conditions
- increased process stability**
- increased thermostability** to run at higher temperatures
- stability to organic solvents**
- absence of substrate and/or product inhibition
- increased selectivity (enantio-, regio-, chemo-)
- accept new substrate
- catalyse new reactions

Motivation



- ❑ **organic cosolvents** can have a **positive effect on catalysis**
 - improving substrate solubility
 - alteration substrate specificity and enantioselectivity
 - suppression of water-induced side reactions
- ❑ **higher concentration of organic co-solvents** usually cause **protein denaturation**

Aims

- ❑ to **identify mutations influencing stability** of haloalkane dehalogenases in organic cosolvent
- ❑ to **construct haloalkane dehalogenase with improved stability** in buffer containing DMSO

- ❑ **error-prone PCR** (epPCR) by Taq polymerase, MnCl_2
- ❑ **screening** by pH indicator assay in MTPs with 42 - 52% DMSO
phenol red: red \longleftrightarrow yellow (pH lower than 6.6)
- ❑ **purification** by affinity chromatography
- ❑ **thermodynamic stability and structural characterization**
by circular dichroism, fluorescence spectroscopy, differential scanning calorimetry and X-ray crystallography
- ❑ **functional characterization and kinetic stability**
by activity assay (Iwasaki method) and steady-state kinetics

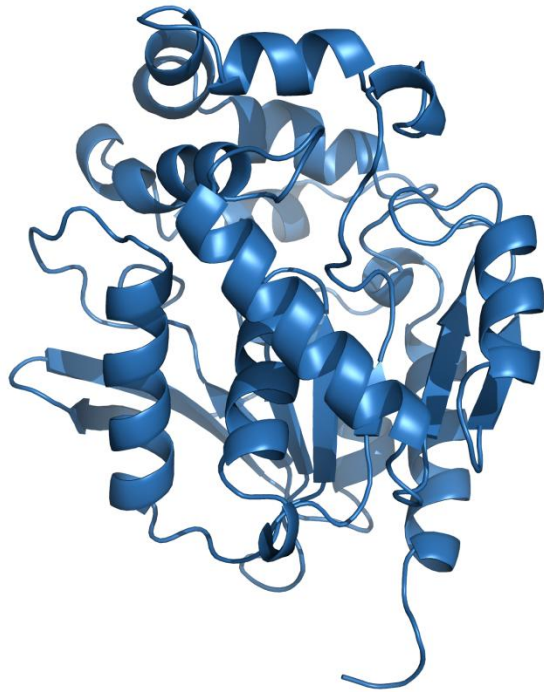
□ thermodynamic x kinetic stability

Definitions of various stability parameters.

Measure	Symbol	Type of stability	Definition
Free energy of unfolding	ΔG_u	Thermodynamic	Change in Gibbs free energy going from the folded to unfolded state
Melting temperature	T_m	Thermodynamic	The temperature at which half of the protein is in the unfolded state
Unfolding equilibrium constant	K_u	Thermodynamic	The concentration of unfolded species divided by the concentration of folded species
Half-concentration	$C_{1/2}$	Thermodynamic	The concentration of denaturant needed to unfold half of the protein (chemical equivalent of T_m)
Observed deactivation rate constant	$k_{d,obs}$	Kinetic	Overall rate constant for going from native to deactivation species
Half-life	$\tau_{1/2}$	Kinetic	Time required for residual activity to be reduced to half
Temperature of half-inactivation	T_{50}	Kinetic	Temperature of incubation to reduce residual activity by half during a defined time period
Optimum temperature	T_{opt}	Kinetic	Temperature leading to highest activity
Total turnover number	TTN	Kinetic	Moles of product produced over the lifetime of the catalyst

- ❑ **site-directed mutagenesis** by QuikChange
- ❑ **gene synthesis**
- ❑ **saturation mutagenesis** by inverse PCR using a synthetic oligonucleotide with one degenerated NNK codon
- ❑ **molecular basis of resistance to organic cosolvent** by molecular dynamics simulations in 40% DMSO

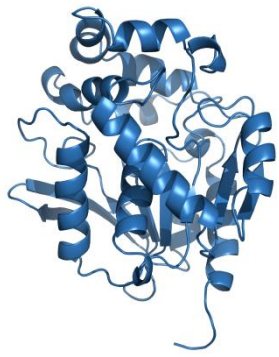
Studied HLD



DhaA

from *Rhodococcus rhodochrous*

Directed evolution



DhaA

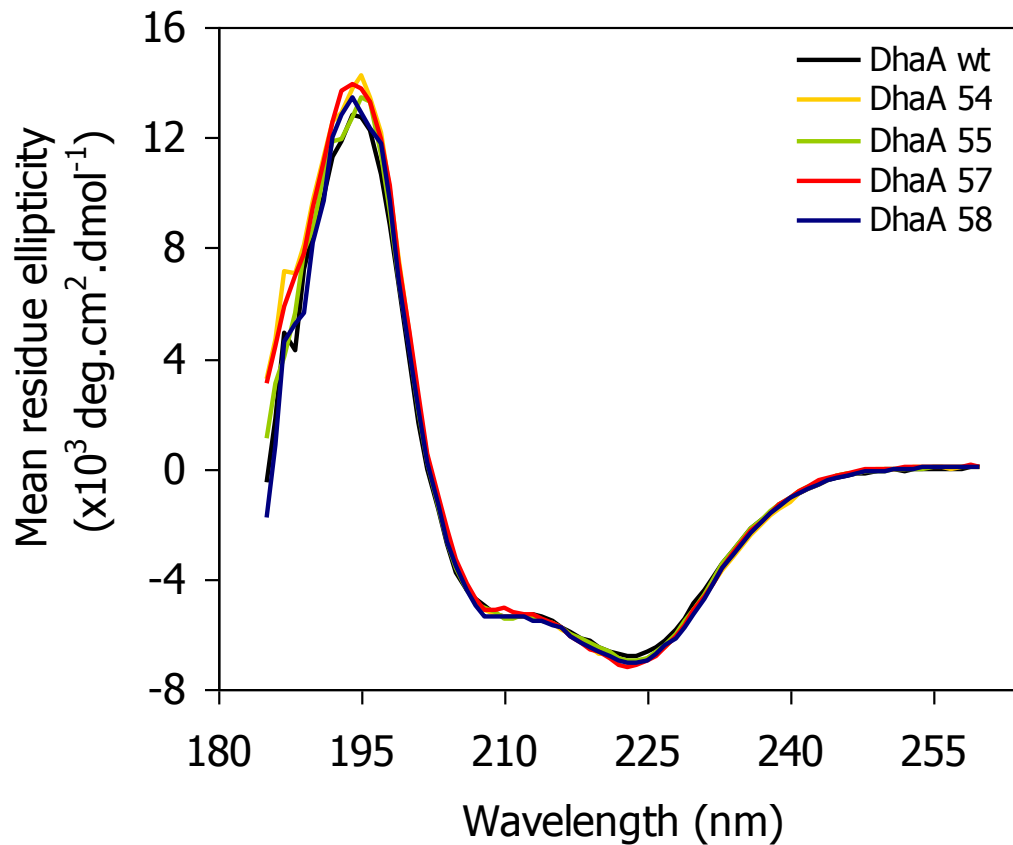
epPCR
→



7 000 colonies

4 positive hits

Mutant resistant to DMSO



$T_m = 50.4 \text{ C}$

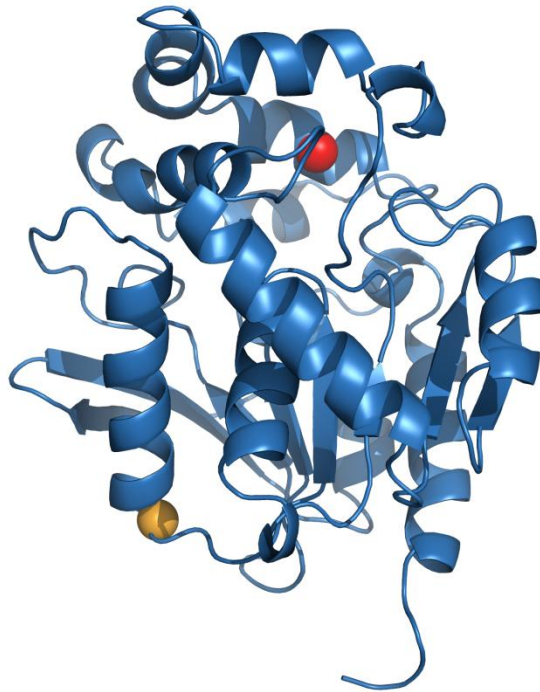
$T_m = 50.6 \text{ }^\circ\text{C}$

$T_m = 52.6 \text{ }^\circ\text{C}$

$T_m = 55.9 \text{ }^\circ\text{C}$

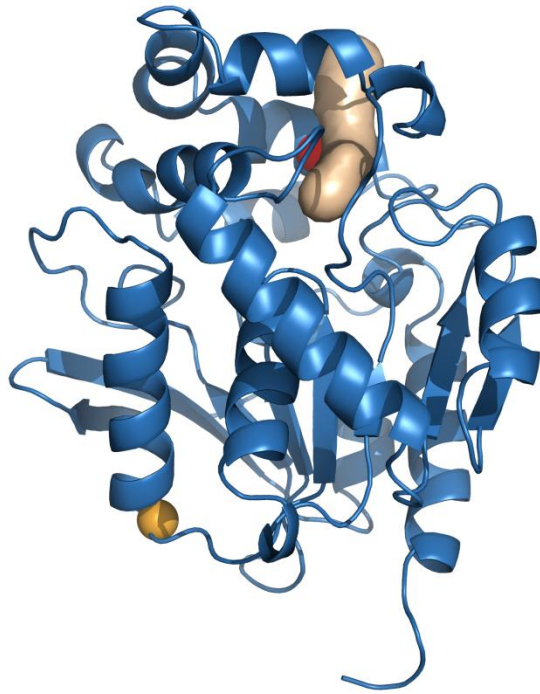
$T_m = 52.5 \text{ }^\circ\text{C}$

Mutant resistant to DMSO



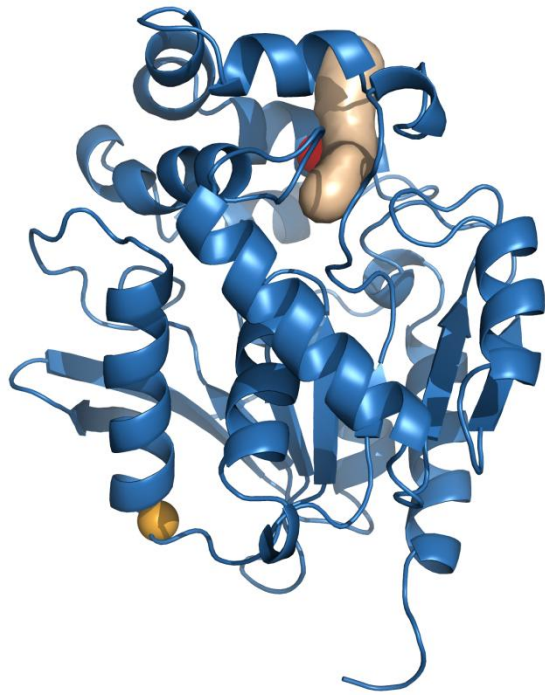
DhaA 57

Mutant resistant to DMSO



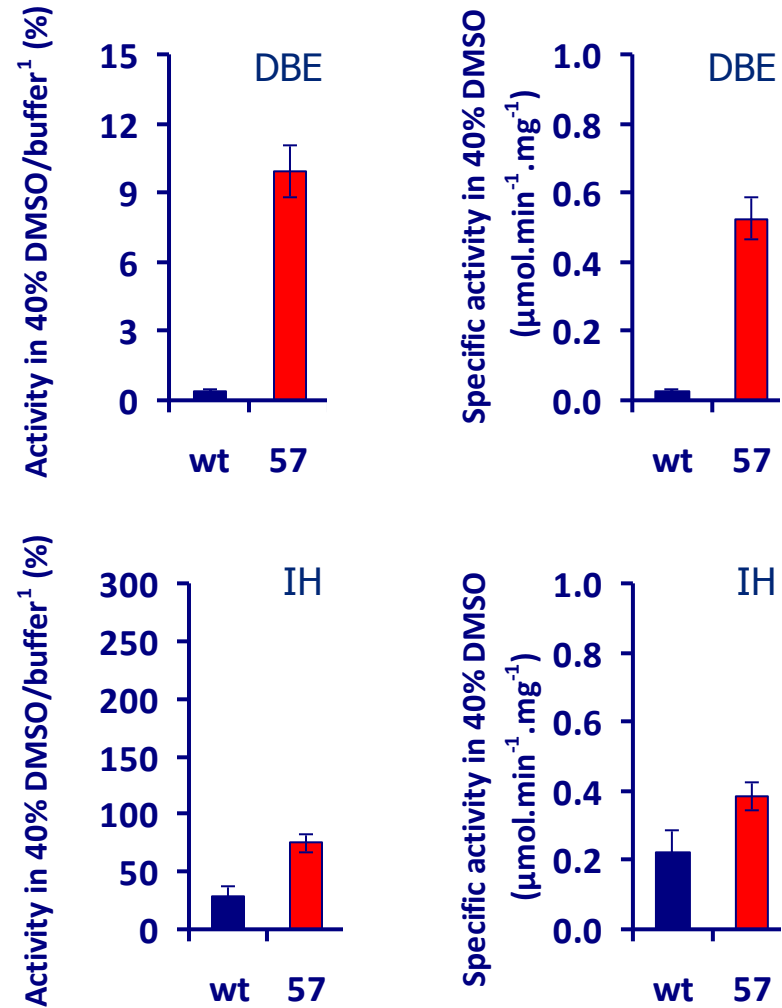
DhaA 57

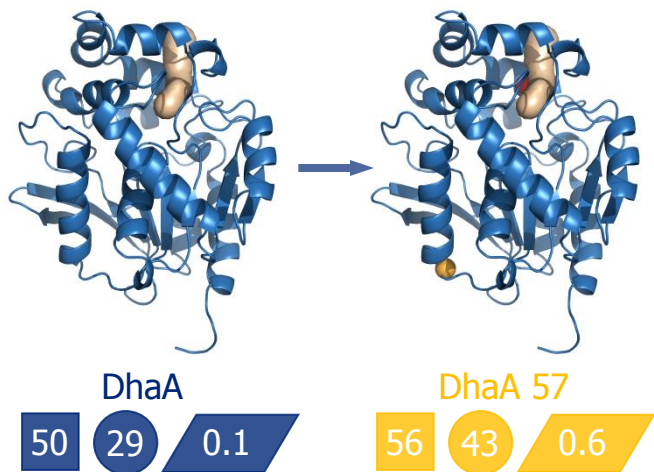
Mutant resistant to DMSO



DhaA 57

DBE – 1,2 dibromoethane; IH – iodohexane

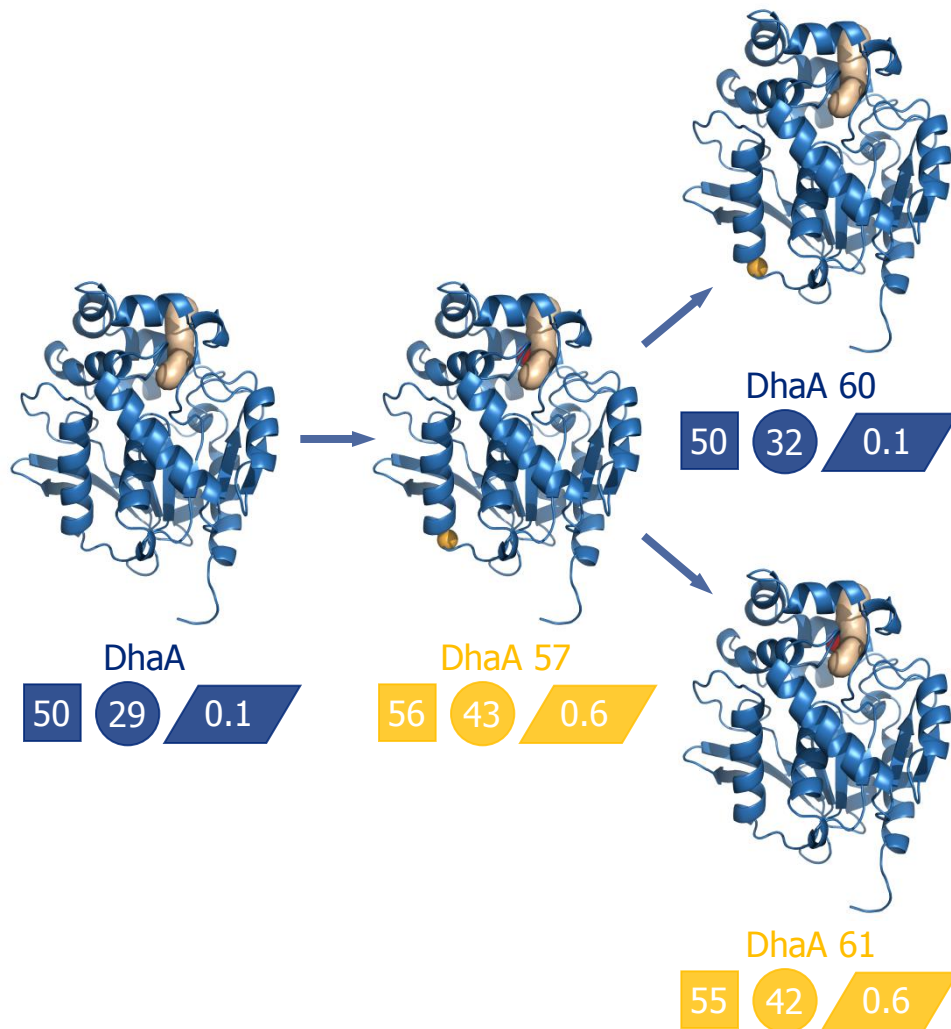




■ melting temperature in buffer (°C)

● half-concentration of DMSO (%)

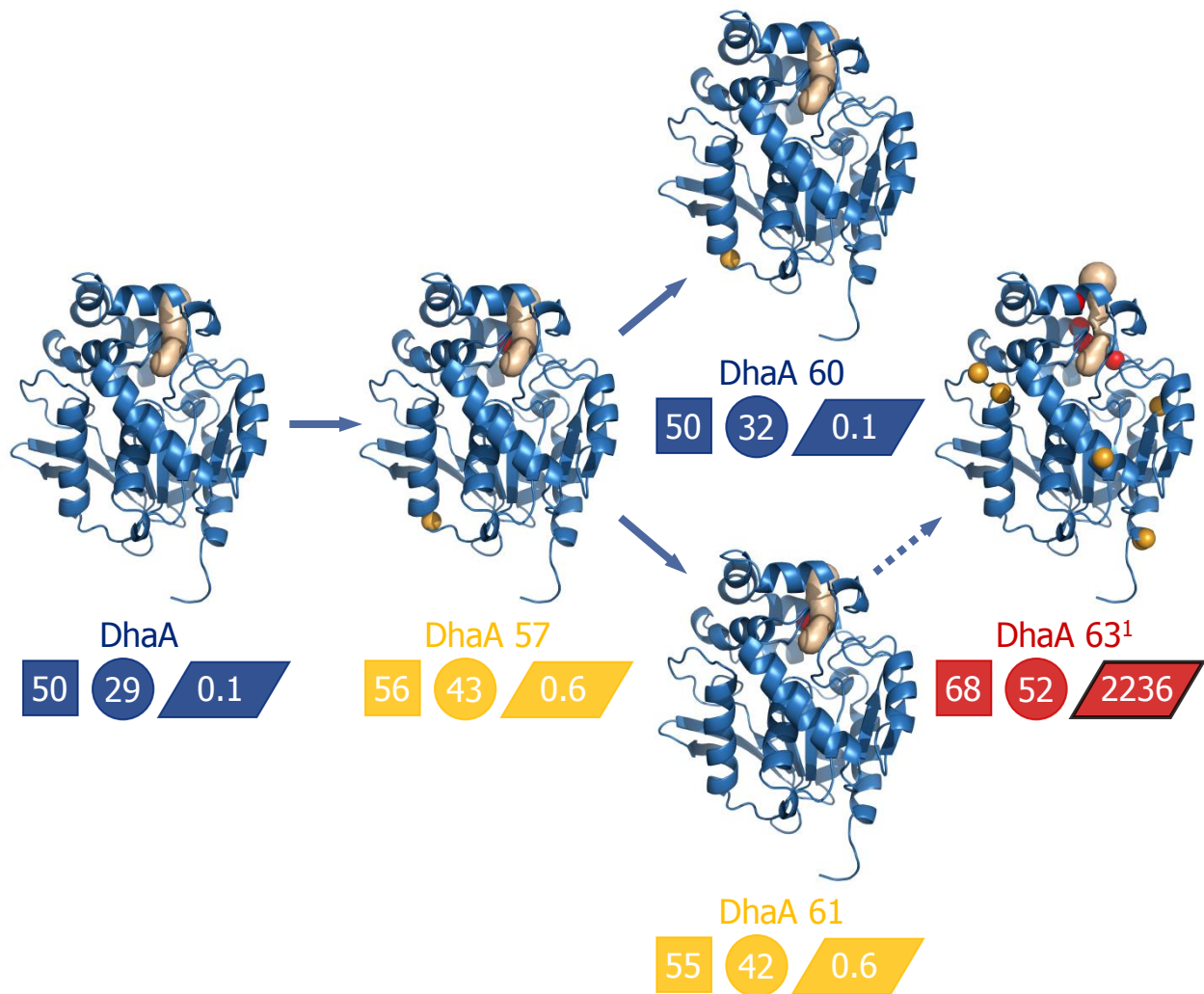
▤ half-life in 40% DMSO at 37 °C (h)



■ melting temperature in buffer (°C)

● half-concentration of DMSO (%)

▤ half-life in 40% DMSO at 37 °C (h)



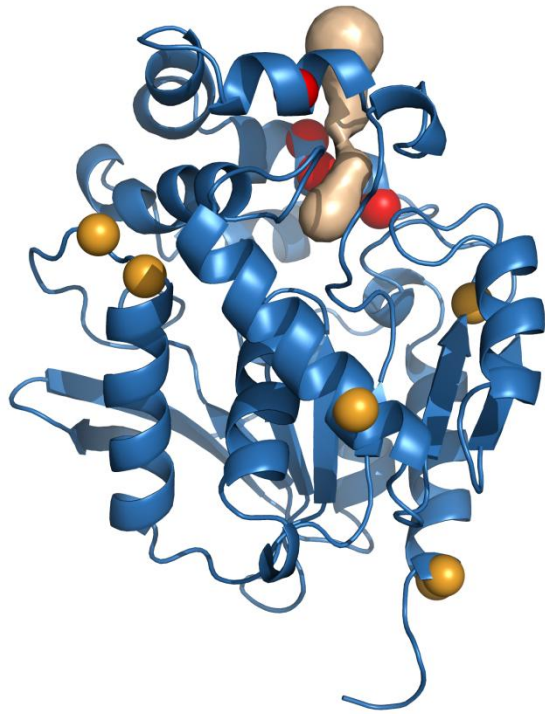
■ melting temperature in buffer (°C)

● half-concentration of DMSO (%)

▤ half-life in 40% DMSO at 37 °C (h)

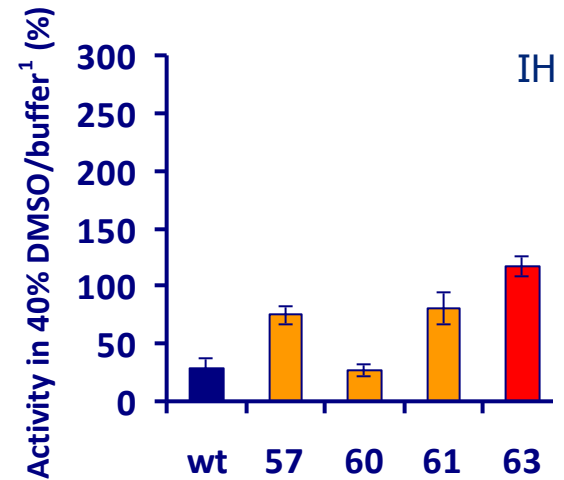
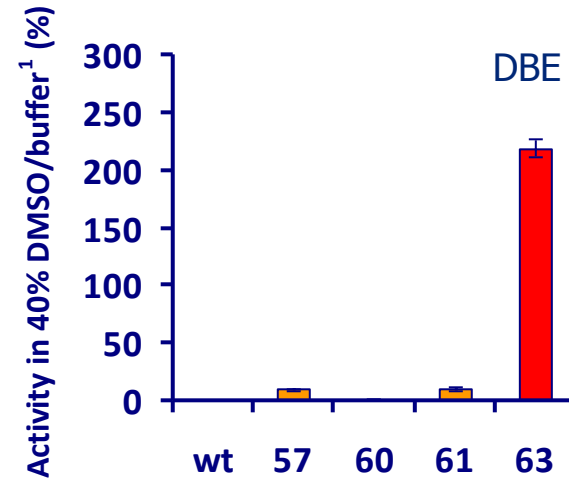
¹Gray, K.A. et al.: *Adv. Synth. Catal.* 343, 607-617 (2001)

Mutant resistant to DMSO

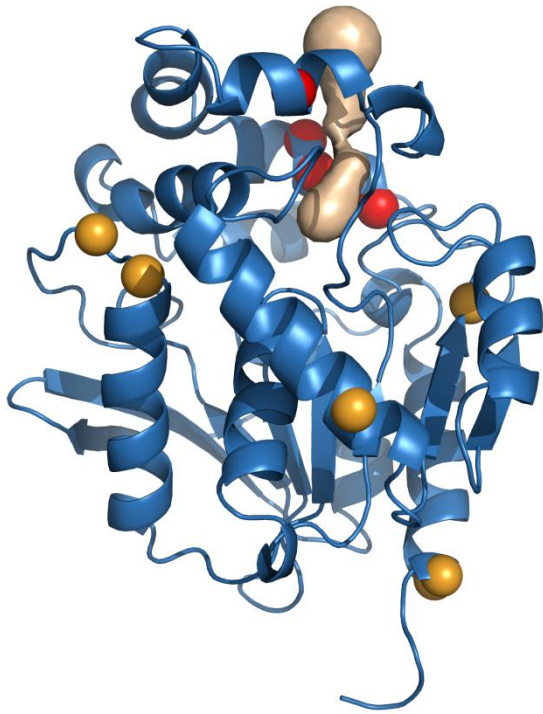


DhaA 63

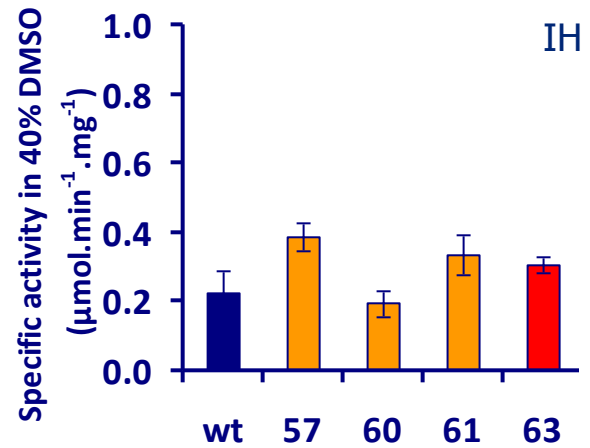
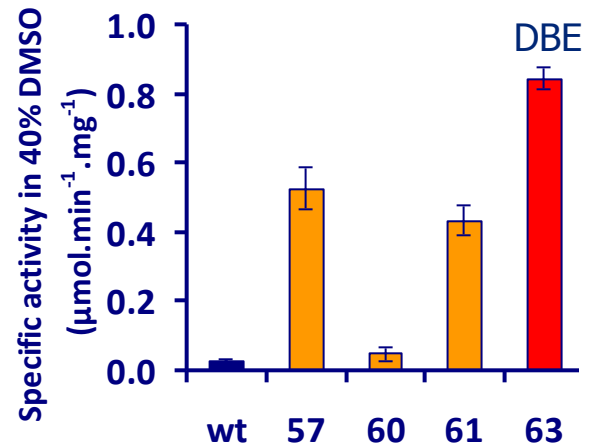
DBE – 1,2 dibromoethane; IH – iodohexane



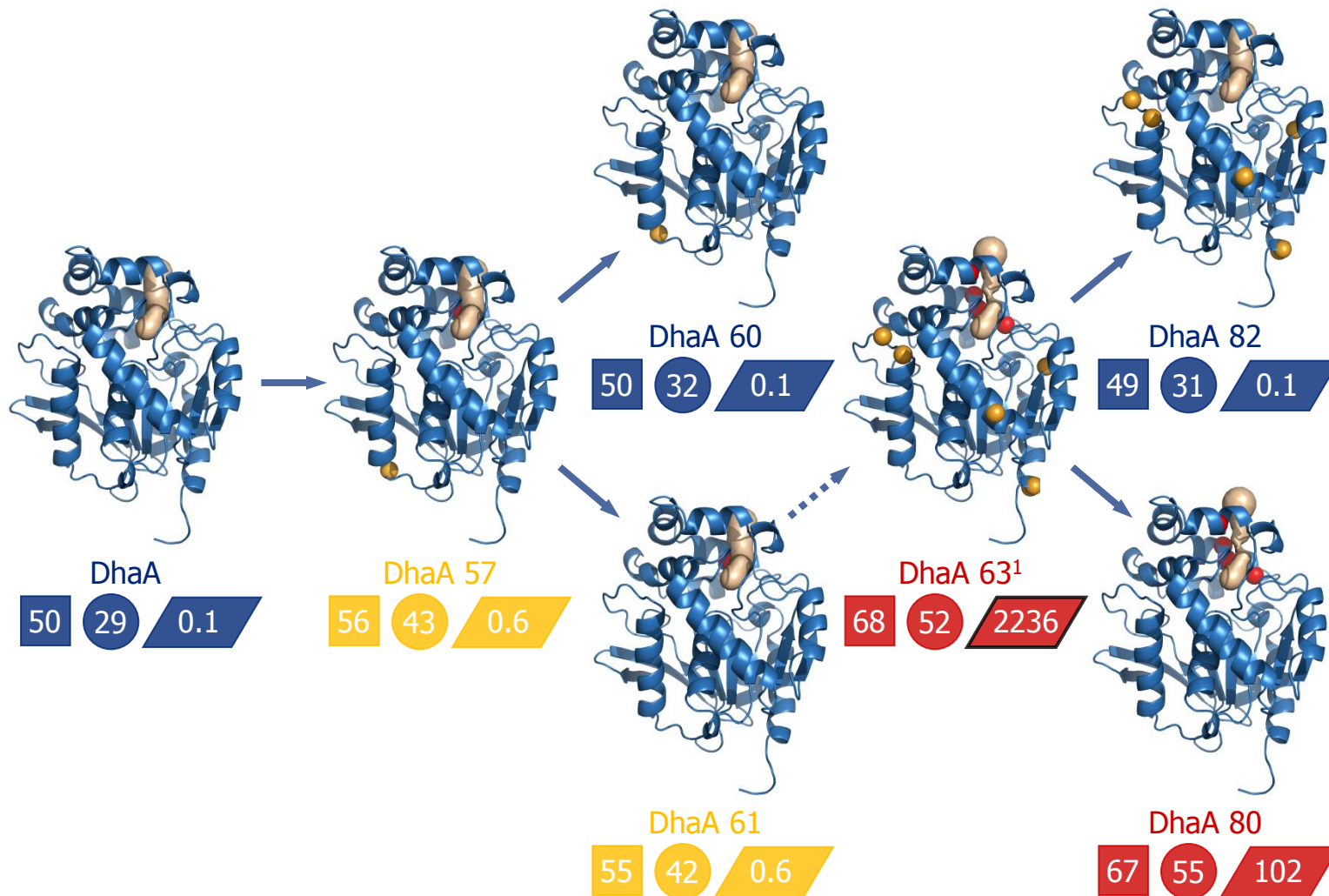
Mutant resistant to DMSO



DhaA 63



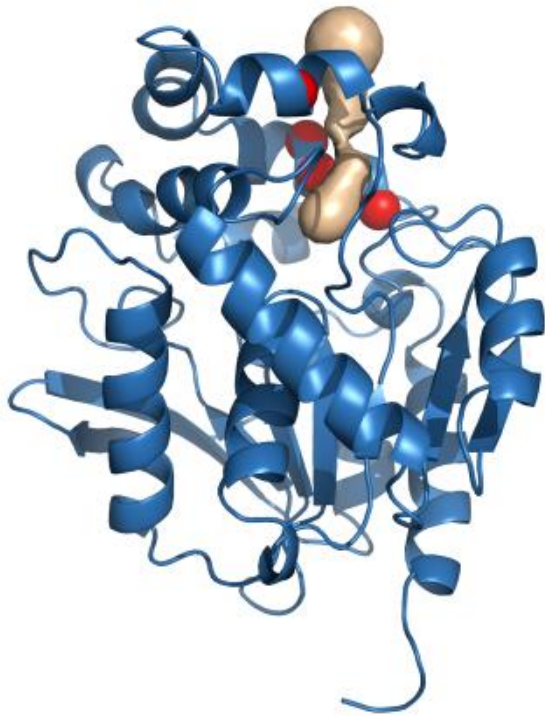
DBE – 1,2 dibromoethane; IH – iodohexane



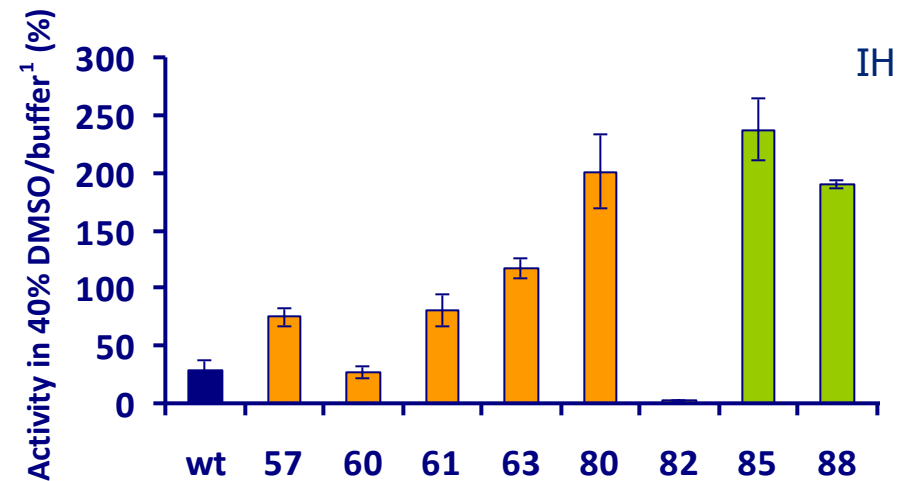
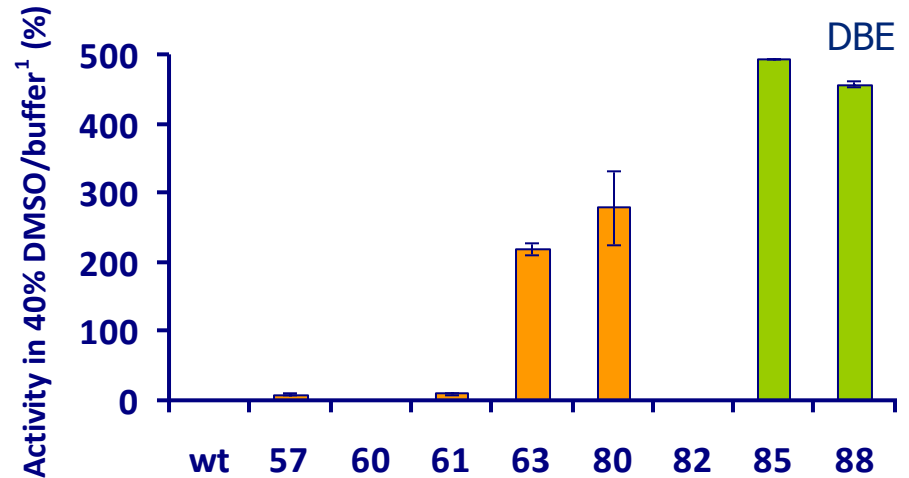
- melting temperature in buffer (°C)
- half-concentration of DMSO (%)
- ▱ half-life in 40% DMSO at 37 °C (h)

¹Gray, K.A. et al.: *Adv. Synth. Catal.* 343, 607-617 (2001)

Mutant resistant to DMSO

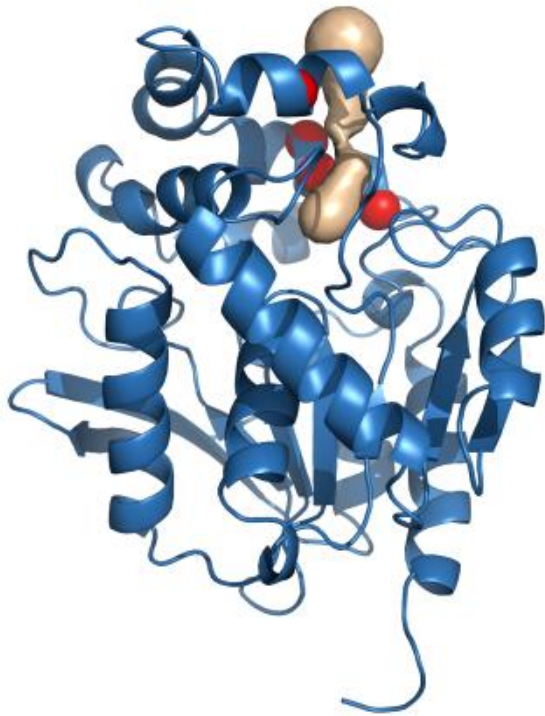


DhaA 85, DhaA 88

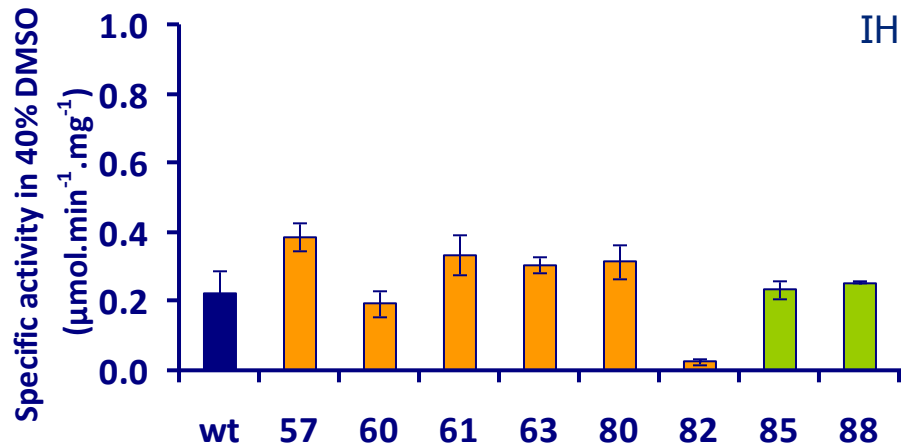
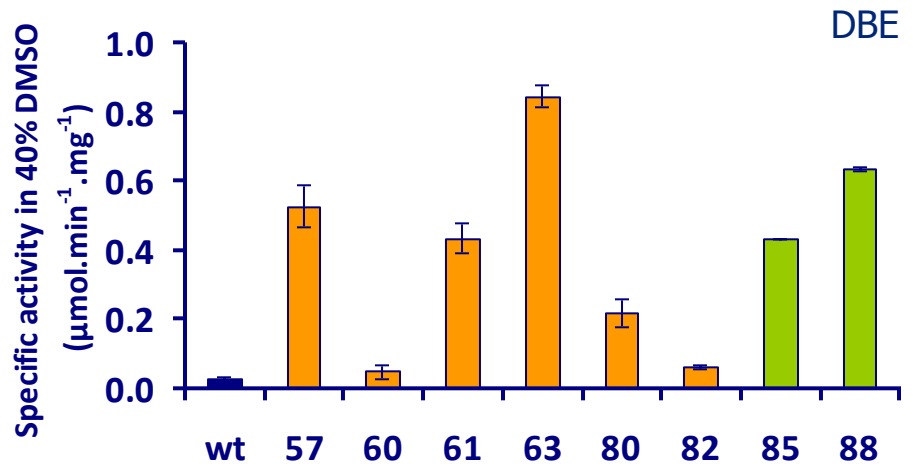


DBE – 1,2 dibromoethane; IH – iodohexane

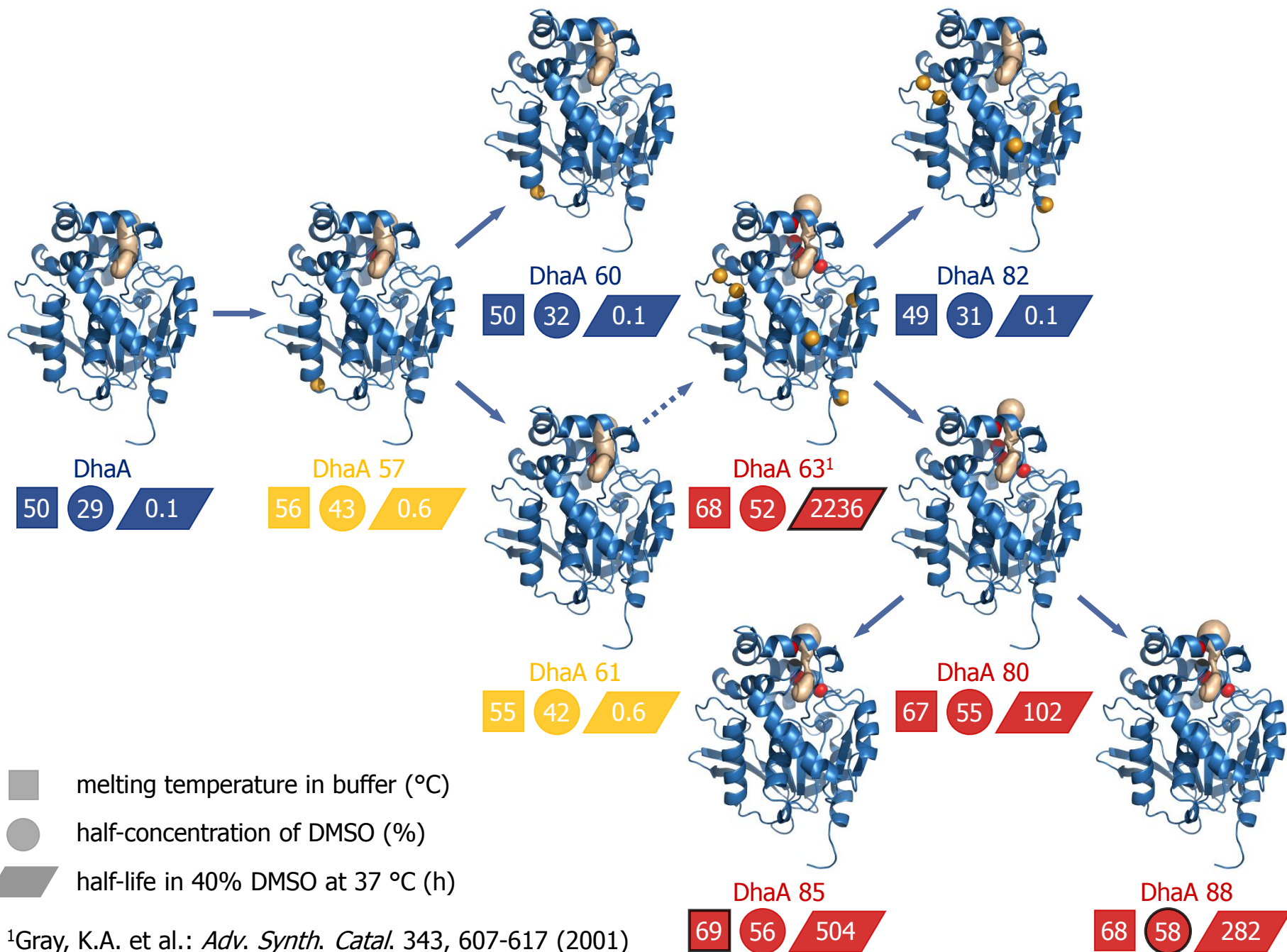
Mutant resistant to DMSO



DhaA 85, DhaA 88



DBE – 1,2 dibromoethane; IH – iodohexane



¹Gray, K.A. et al.: *Adv. Synth. Catal.* 343, 607-617 (2001)

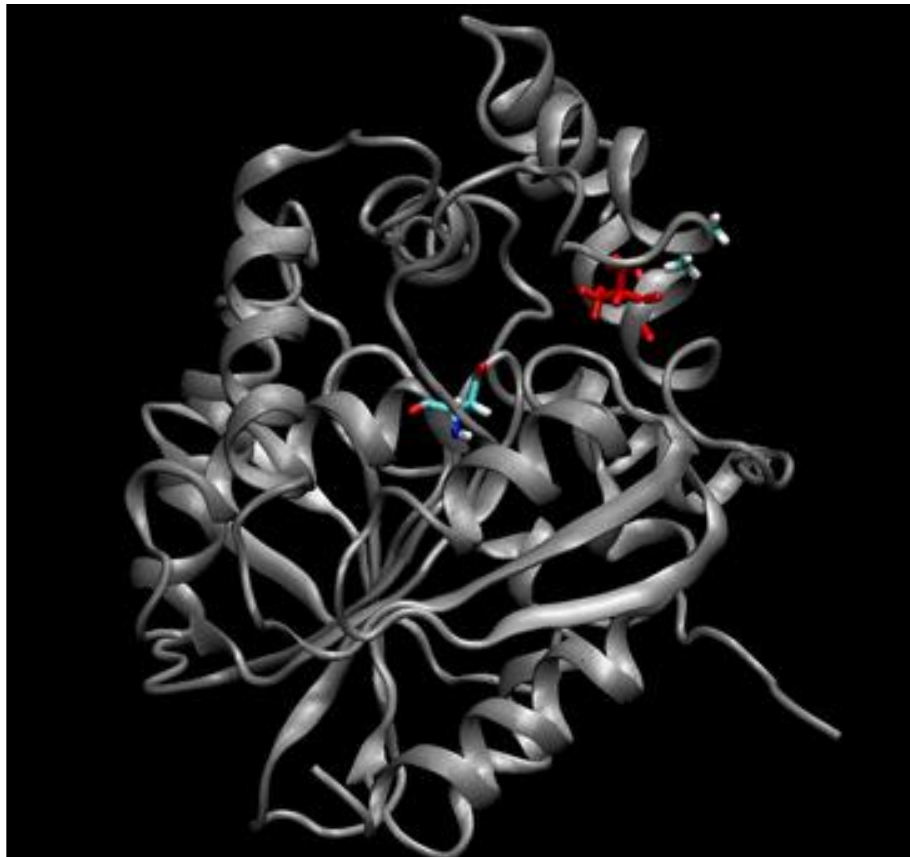
Mutant resistant to DMSO

DhaA wt

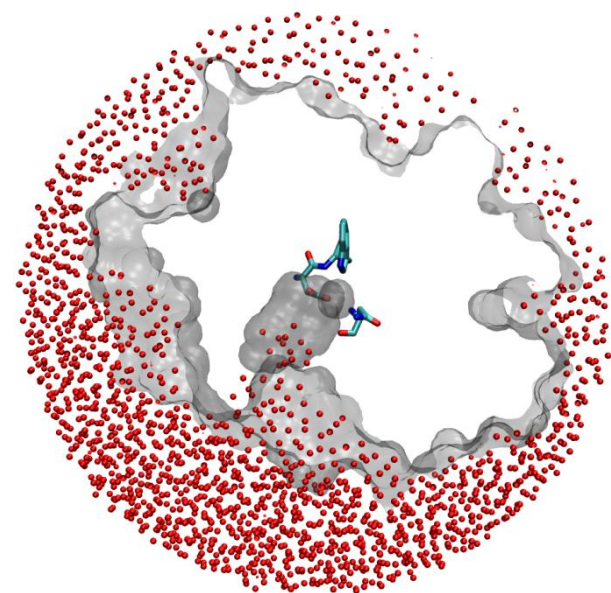


Mutant resistant to DMSO

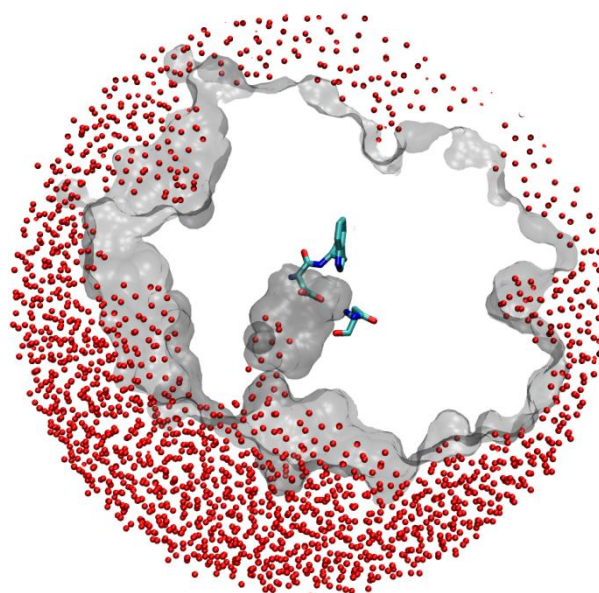
DhaA 57



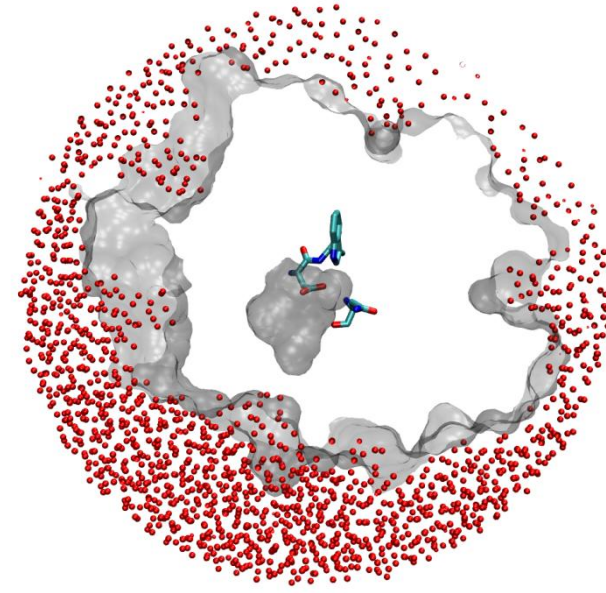
Mutant resistant to DMSO



DhaA wt



DhaA 57



DhaA 80

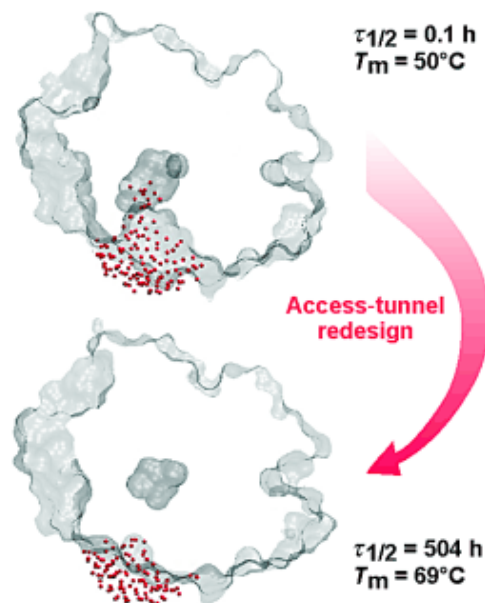
Conclusion

- ❑ **resistance** towards organic cosolvents **correlates with thermostability**
- ❑ mutations lining access tunnel modulate **occupancy** of active site by solvent and can **stabilize** protein
- ❑ robust catalysts were developed:
4 point mutations, $T_m \uparrow$ **19 C**, $T_{1/2}$ (40% DMSO) **min** → **days**
- ❑ **engineering of access tunnels represents novel strategy for engineering of robust catalysts**

Protein Stability

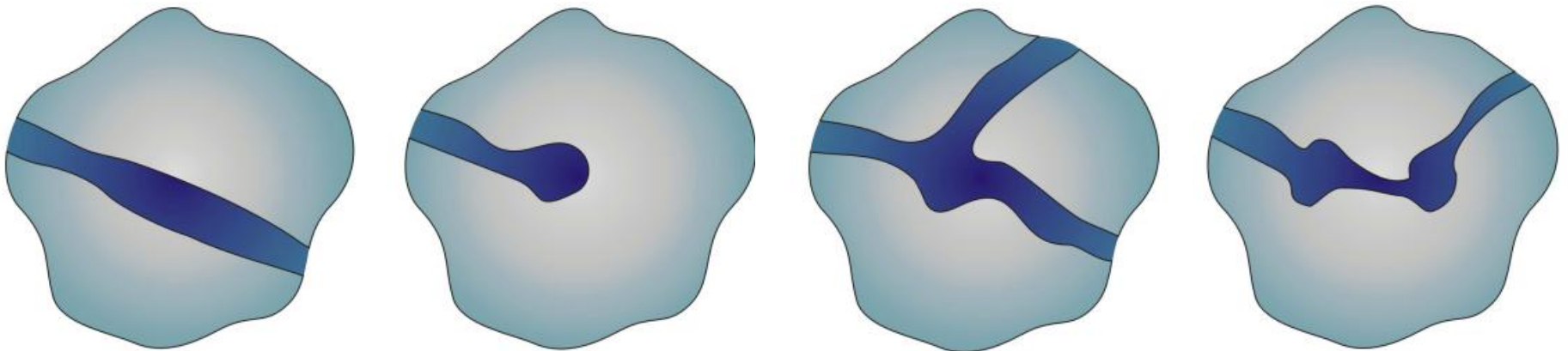
Engineering Enzyme Stability and Resistance to an Organic Cosolvent by Modification of Residues in the Access Tunnel**

*Tana Koudelakova, Radka Chaloupkova, Jan Brezovsky, Zbynek Prokop, Eva Sebestova, Martin Hesseler, Morteza Khabiri, Maryia Plevaka, Daryna Kulik, Ivana Kuta Smatanova, Pavlina Rezacova, Rudiger Ettrich, Uwe T. Bornscheuer, and Jiri Damborsky**



Method of protein stabilization

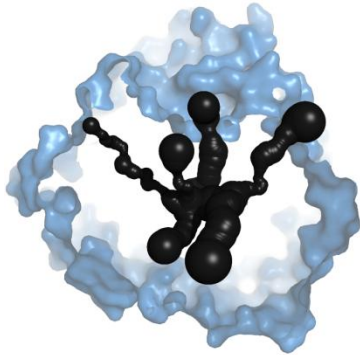
- ❑ method for modification of the access routes in order to achieve better stability of protein towards temperature and solvents
- ❑ definition of the access routes: channel x tunnel



Method of protein stabilization

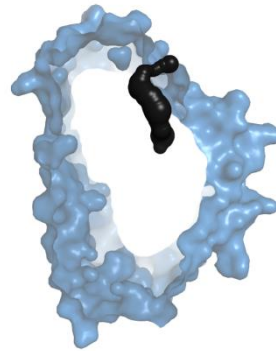
- ❑ method for modification of the access routes in order to achieve better stability of protein towards temperature and solvents
- ❑ definition of the access routes: channel x tunnel
- ❑ general concept, tunnels found in all enzyme classes

1. OXIDOREDUKTASES



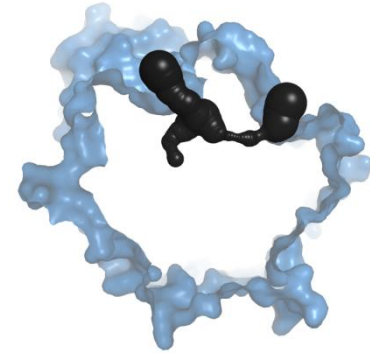
Cytochrome CYP3A4
EC 1.1.3.6

2. TRANSFERASES



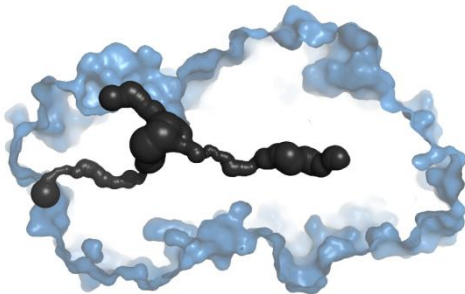
Chalcone synthase
EC 2.3.1.74

3. HYDROLASES



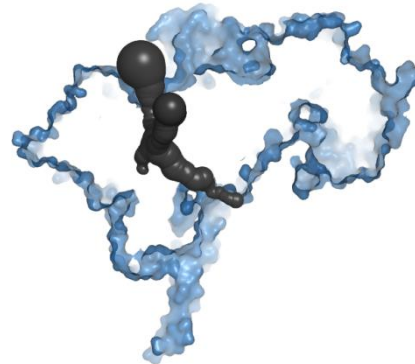
Acetylcholinesterase
EC 3.1.1.7

4. LYASES



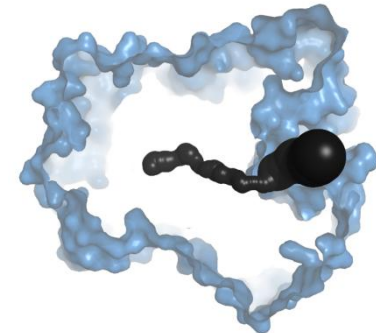
Tryptophan synthase
EC 4.2.1.20

5. ISOMERASES



Methylmalonyl-CoA mutase
EC 5.4.99.2

6. LIGASES



Asparagine synthetase
EC 6.3.1.1

Method of protein stabilization

- ❑ procedure of protein stabilization
 - identification the amino acids lining access routes based on knowledge of structure (CAVER, HotSpot Wizard)
 - modification of selected amino acids „hot spots“ (site-directed mutagenesis, random mutagenesis)
 - analysis of constructed variants/libraries, assessment of the result of modification
- ❑ rational focused mutagenesis based on detailed knowledge of structure and function
 - creation of small focused “smart” libraries
 - increase likelihood of beneficially modifying property

Method of protein stabilization

- ❑ modification of shape and physico-chemical properties of tunnels
 - selective discrimination between the molecules of a substrate/product and undesired solvent molecules inside the access routes
 - strengthening of hydrophobic interactions within the tunnel
 - thermostability enhancement
- ❑ high thermostability and resistance against organic cosolvents = required process design criteria
- ❑ invention describing the method of stabilization patented

Damborsky, J., Prokop, Z., Koudelakova, T., Stepankova, V., Chaloupkova, R., Chovancova, E., Gora, A., Brezovsky, J., 2011: Method of thermostabilization of a protein and/or stabilization towards organic solvents. Patent PV 2011-680.

Method of protein stabilization

- identification of tunnels – CAVER¹



The banner features the CAVER logo on the left, which consists of a stylized protein structure with an orange circle. To the right of the logo, the text reads "software tool for protein analysis and visualisation". On the far right, there is a 3D molecular model of a protein structure. Below this, a dark blue navigation bar contains four icons and their corresponding text: a lightbulb icon for "concept" (learn more about caver viewer), a download arrow icon for "download" (download latest build and plugins), a hand cursor icon for "online" (launch caver viewer using java web start), and a question mark icon for "find help" (fix troubles, report bugs, request features).

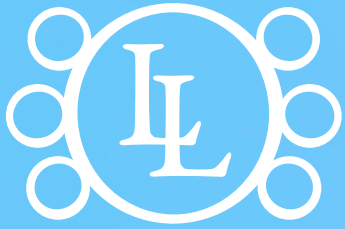
www.caver.cz

¹Chovancova E. *et al.*, 2012, PLoS Comp. Biol. 8: e1002708

Helpful references

- ❑ Koudelakova, T. et al. (2013) Engineering enzyme stability and resistance to an organic cosolvent by modification of residues in the access tunnel, *Angew. Chem. Int. Ed.* 52: 1959-1963
- ❑ Gray, K.A. (2001) Rapid evolution of reversible denaturation and elevated melting in a microbial haloalkane dehalogenase, *Adv. Synth. Catal.* 343: 607-617
- ❑ Chovancova, E. (2012) CAVER 3.0: A Tool for Analysis of Transport Pathways in Dynamic Protein Structures, *PLoS Comput. Biol.* 8: e1002708

❑ **QUESTIONS?**



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11. EXAMPLE OF UTILIZING PROTEIN ENGINEERING TO ENHANCE ENZYME ENANTIOSELECTIVITY

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