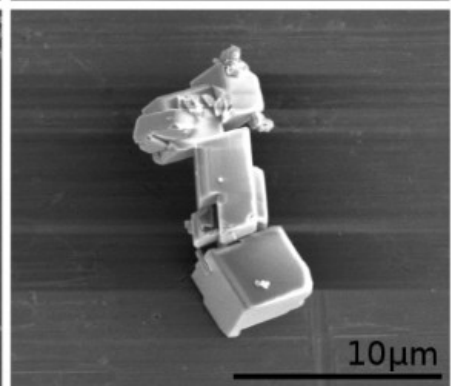
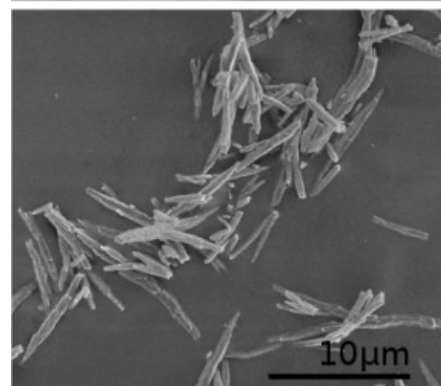
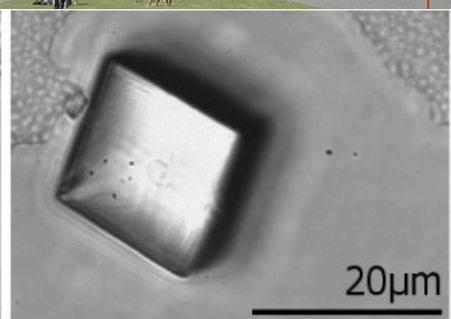
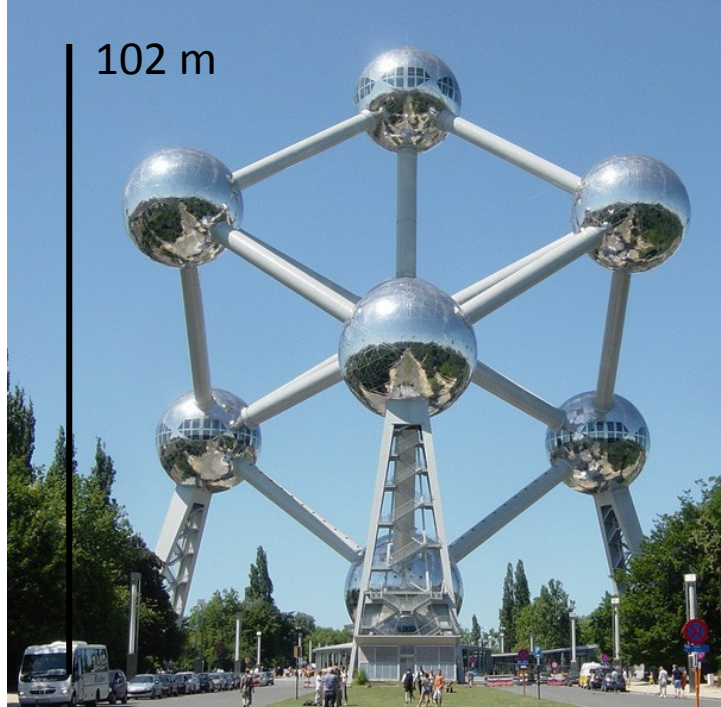


Winter School on Structural Cell Biology

Protein crystallization
Josef Houser



Crystal

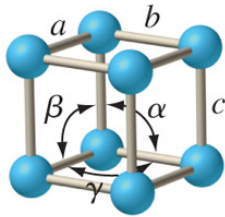


5.00 kV 50.00 μm 5.1 mm aragonite

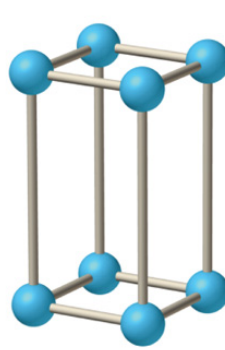
5.00 kV 50.00 μm 5.8 mm calcite

7 Crystal lattices

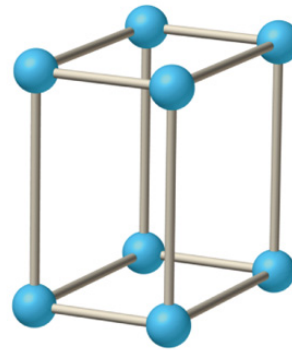
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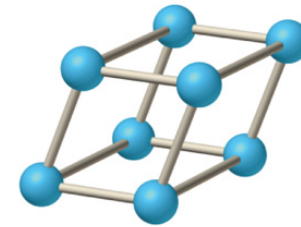
Simple cubic
 $a = b = c$
 $\alpha = \beta = \gamma = 90^\circ$



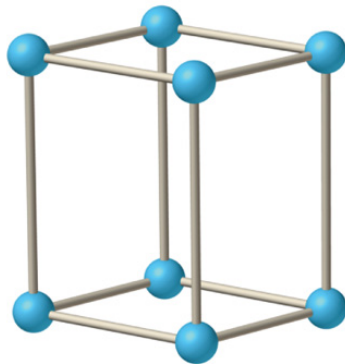
Tetragonal
 $a = b \neq c$
 $\alpha = \beta = \gamma = 90^\circ$



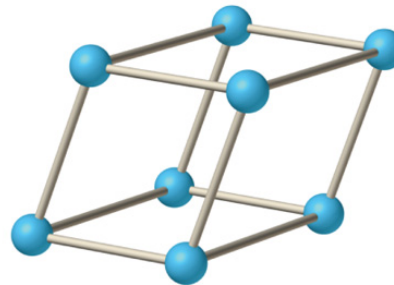
Orthorhombic
 $a \neq b \neq c$
 $\alpha = \beta = \gamma = 90^\circ$



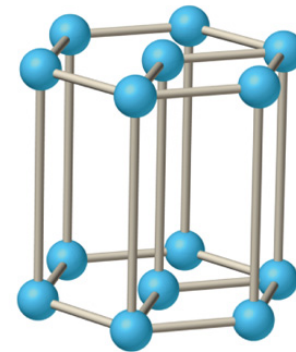
Rhombohedral
 $a = b = c$
 $\alpha = \beta = \gamma \neq 90^\circ$



Monoclinic
 $a \neq b \neq c$
 $\gamma \neq \alpha = \beta = 90^\circ$

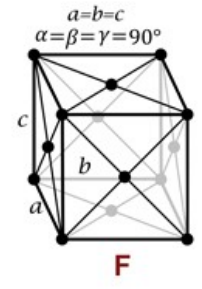
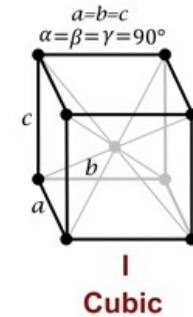
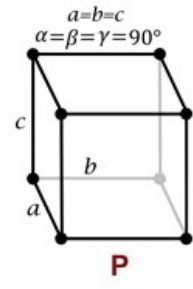
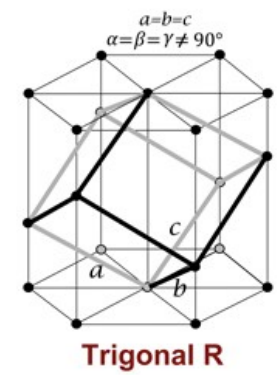
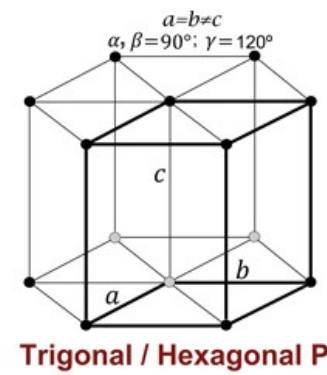
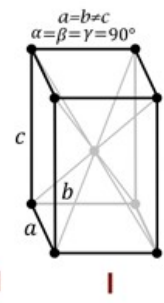
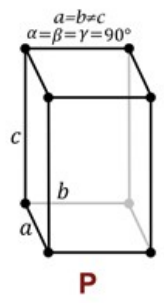
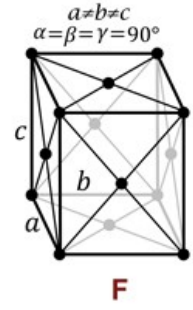
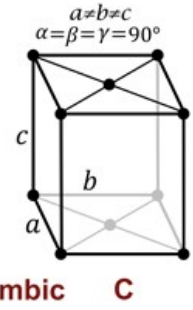
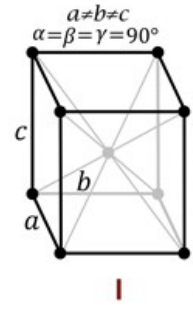
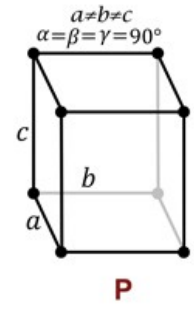
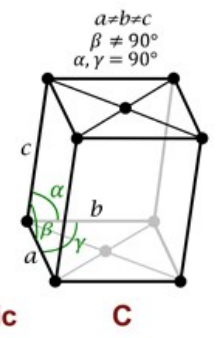
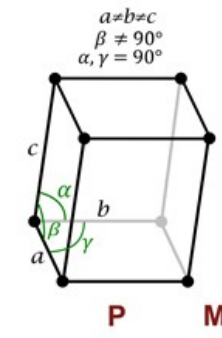
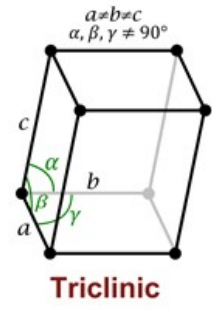


Triclinic
 $a \neq b \neq c$
 $\alpha \neq \beta \neq \gamma \neq 90^\circ$



Hexagonal
 $a = b \neq c$
 $\alpha = \beta = 90^\circ, \gamma = 120^\circ$

14 Bravais lattices

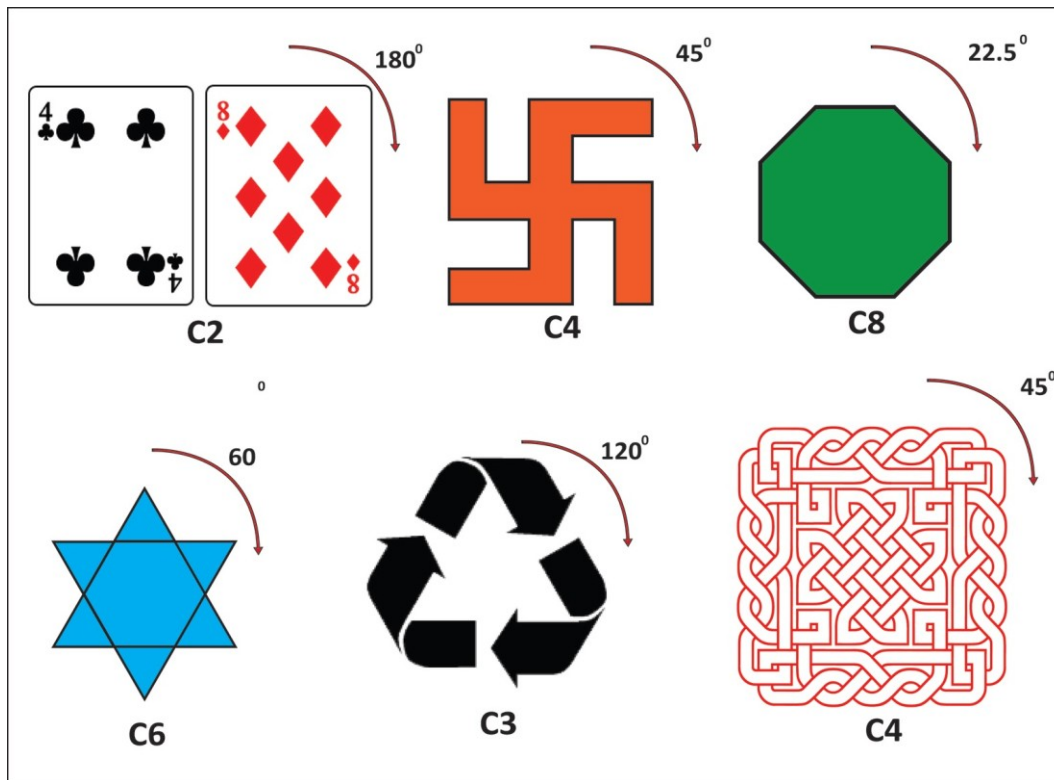


Symmetry – Space groups

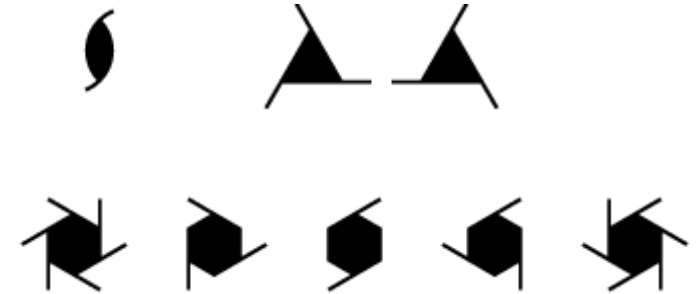
2D symmetry

3D symmetry

Proteins are chiral – no mirror symmetry



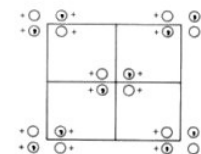
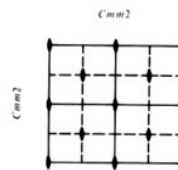
Screw axis



Combination of Bravais lattices and symmetry operations leads to **230 possible space groups** (not all for proteins).

System	Bravais Lattice	Point Group	Space Group						Fraction		
Triclinic	P	1 $\bar{1}$	P1 P $\bar{1}$						1/2		
Monoclinic	P	2 m	P2 Pm	P2 ₁ Pc	C2 Cm				1/4		
	C	2/m	P2/m	P2 ₁ /m	C2/m	P2/c	P2 ₁ /c	C2/c			
Orthorhombic	P	222	P222	P222 ₁	P2 ₁ 2 ₁ 2	P2 ₁ 2 ₁ 2 ₁	C222 ₁	C222	1/8		
	C	mm2	F222	I222	I2 ₁ 2 ₁ 2 ₁						
	F		Pmm2	Pmc2 ₁	Pcc2	Pma2	Pca2 ₁	Pnc2			
	I	mmm	Pmn2 ₁	Pba2	Pna2	Pnn2	Pmm2	Pnc2 ₁			
Tetragonal	P	4 $\bar{4}$ 4/m	P4 P $\bar{4}$ P4/m	P4 ₁ I $\bar{4}$ P4 ₂ /m	P4 ₂	P4 ₃	I4 I4 ₁		1/8		
		I	422 4mm $\bar{4}2m$	P422 P4 ₃ 22 P4mm P4 ₂ mc P $\bar{4}2m$ P $\bar{4}b2$	P42 ₁ 2 P4 ₃ 2 ₁ 2 P4bm P4 ₂ bc P $\bar{4}2c$ P $\bar{4}n2$	P4 ₁ 22 I422 P4 ₂ cm I4mm P $\bar{4}2m$ I $\bar{4}m2$	P4 ₁ 2 ₁ 2 I4 ₁ 22 P4 ₂ nm I4cm P $\bar{4}2c$ I $\bar{4}c2$	P4 ₂ 22 I4 ₁ md P4cc I4 ₁ md P4m2 I $\bar{4}2m$		P4 ₂ 2 ₁ 2 I4 ₁ cd P4nc I4 ₁ cd P $\bar{4}c2$ I $\bar{4}2d$	1/16
			4/mmm	P4/mmm P4/nmm P4 ₂ /mbc I4 ₁ /amd	P4/mcc P4/ncc P4 ₂ /mnm I4 ₁ /acd	P4/nbm P4 ₂ /mmc P4 ₂ /nmc	P4/nnc P4 ₂ /mcm P4 ₂ /ncm	P4/mbm P4 ₂ /nbc I4/mmm		P4/mnc P4 ₂ /nmm I4/mcm	
	R		3 $\bar{3}$	P3 P $\bar{3}$	P3 ₁ R $\bar{3}$	P3 ₂	R3			1/6	
		32	P312 R32	P321	P3 ₁ 12	P3 ₂ 1	P3 ₁ 12	P3 ₂ 1	1/12		
		3m $\bar{3}m$	P3m1 P $\bar{3}1m$	P31m P $\bar{3}1c$	P3c1 P $\bar{3}m1$	P31c P $\bar{3}c1$	R3m R $\bar{3}m$	R3c R $\bar{3}c$			
	Hexagonal	P same as trigonal	6 $\bar{6}$ 6/m	P6 P $\bar{6}$ P6/m	P6 ₁ P6 ₃ /m	P6 ₅	P6 ₂	P6 ₄	P6 ₃	1/12	
			622	P622	P6 ₁ 22	P6 ₅ 22	P6 ₂ 22	P6 ₄ 22	P6 ₃ 22		1/24
			6mm $\bar{6}m\bar{2}$	P6mm P $\bar{6}m\bar{2}$	P6cc P $\bar{6}c2$	P6 ₃ cm P6 ₂ m	P6 ₃ mc P6 ₂ c				
			6/mmm	P6/mmm	P6/mcc	P6 ₃ /mcm	P6 ₃ /mmc				
	Cubic	P I F	23 m3	P23 Pm3 Ia3	F23 Pn3	I23 Fm3	P2 ₁ 3 Fd3	I2 ₁ 3 Im3	Pa3	1/24	
			432	P432 P4 ₃ 2 I4 ₃ 2	P4 ₂ 32 I4 ₁ 32	F432	F4 ₃ 2	I432	P4 ₃ 2		1/48
$\bar{4}3m$ m3m			P $\bar{4}3m$ Pm3m Fd3m	F $\bar{4}3m$ Pn3n Fd3c	I $\bar{4}3m$ Pm3n Im3m	P $\bar{4}3n$ Pn3m Ia3d	F $\bar{4}3c$ Fm3m	I $\bar{4}3d$ Fm3c			

Cmm2 C_{2v}^{11} **mm2** Orthorhombic
No. 35 **Cmm2** Patterson symmetry **Cmmm**



Positions

Multiplicity,
Wyckoff letter,
Site symmetry

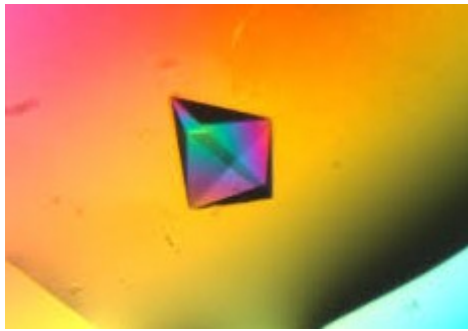
Coordinates

(0,0,0)+ $(\frac{1}{2}, \frac{1}{2}, 0)+$

8 f 1 (1) x,y,z (2) \bar{x},\bar{y},z (3) x, \bar{y},z (4) \bar{x},y,z

Example of space group information from International Crystallographic Tables

*The 11 Laue symmetries are separated by horizontal lines.



Protein

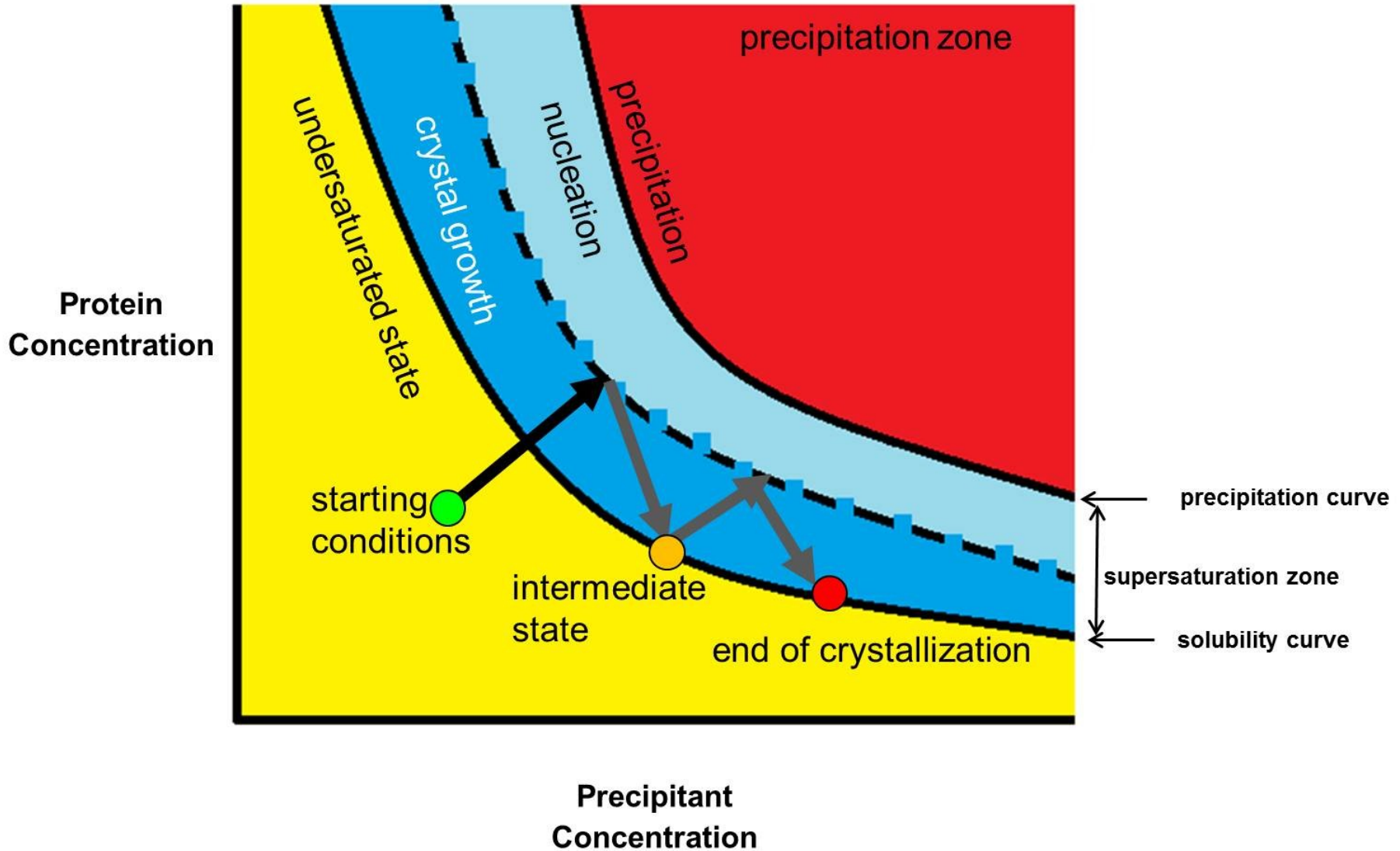
- Big molecule
- Temperature sensitive
- Hundreds-thousands of rotating bonds
- Weak interactions mostly involved
- High solvent content (30-70 %)
- Fragile crystals



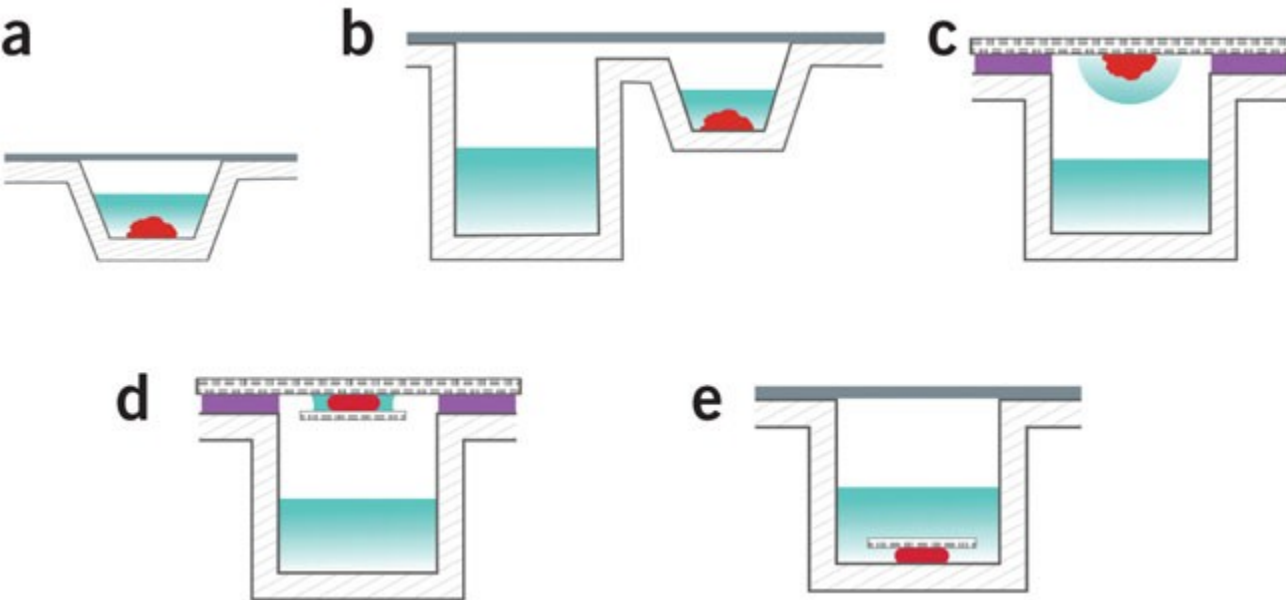
Inorganic salt (ev. organics)

- Small molecule
- Thermostable
- None/few rotating bonds
- Strong (coulombic) interactions frequent
- Low/none solvent content
- Hard crystals

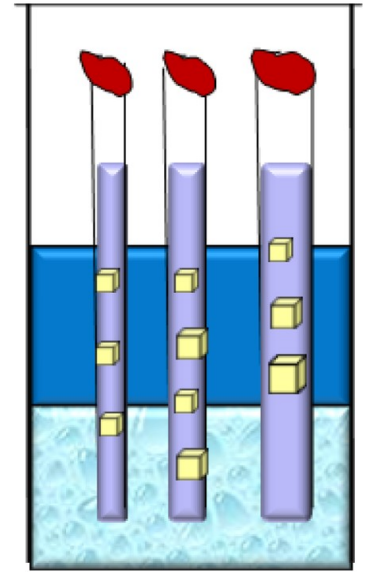
Phase diagram



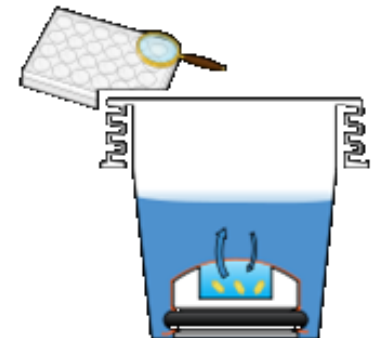
Protein crystallization techniques



Protein crystallization by vapour diffusion: **(a)** microbatch, **(b)** sitting drop and **(c,d)** hanging drop. In **d** and **e**, a sandwich is made of the mesophase (red) by placing a small glass coverslip (hatched) **(d)** below or **(e)** above the bolus. *From Nature Protocols*

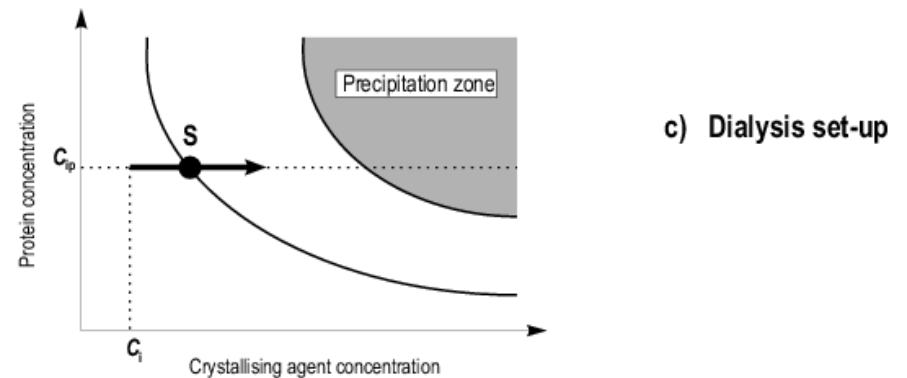
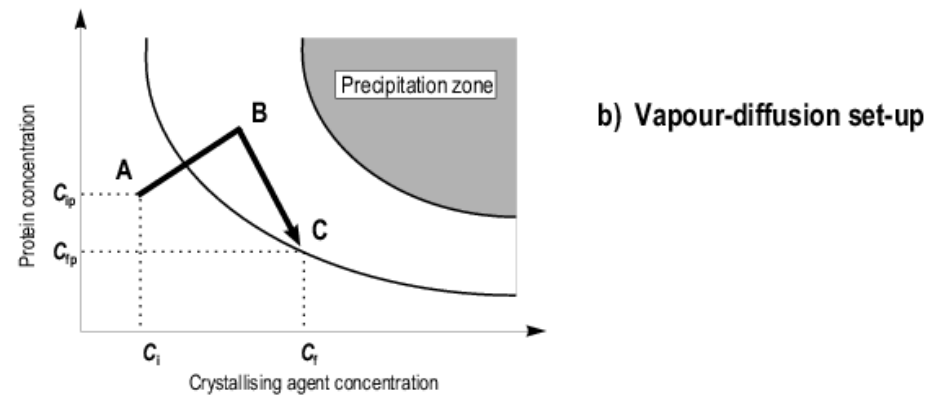
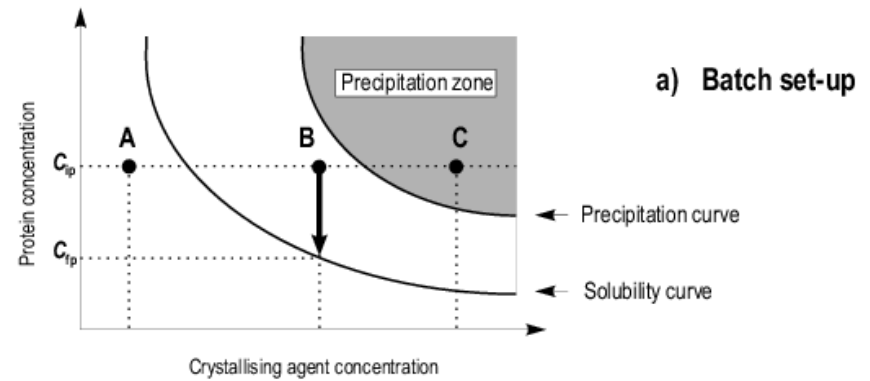


Protein crystallization by counter diffusion in capillaries.



Protein crystallization by dialysis.

Various techniques = various path in the phase diagram



Automatization vs. manual work

- ❑ High-throughput

- ❑ Low volumes (20-150 nl)

- ❑ Reproducibility

- ❑ Individual design

- ❑ Immediate visual control

- ❑ Complex sample handling



Further reading

- <http://journals.iucr.org/>
- Naomi E. Chayen: Protein Crystallization Strategies for Structural Genomics, 2007
- Terese M. Bergfors: Protein Crystallization, 2009
- Alexander McPherson: Introduction to Macromolecular Crystallography, 2011
- etc.