

Correlation between chemical structure and functional properties of organosilicon plasma polymers and SiO₂-like films

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Outline

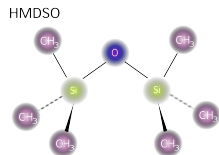
- Motivation
- Experimental
- Variety of materials
- Temperature Induced Changes
- Conclusion

Motivation

Hexamethyldisiloxane (HMDSO)

- versatile starting material for PECVD:

- ▶ source of Si-O-Si bonds (especially for HMDSO/O₂)
- ▶ source of Si-C bonds



PECVD in low pressure rf capacitively coupled discharges (CCP)

- variety of different materials can be prepared when using the mixture of HMDSO/O₂ in varying deposition conditions:

- ▶ percentage of HMDSO in HMDSO/O₂
- ▶ pressure p
- ▶ rf power P
- ▶ dc self-bias U_b (in relation with P and p)

How these materials react to annealing?

PECVD using HMDSO at atmospheric pressure: competition with low pressure process in achievement of silica-hard coatings

Will help an increased deposition temperature?

PECVD in low pressure rf CCP from HMDSO/O₂

- ▶ frequency of 13.56 MHz
- ▶ capacitive ac coupling
- ▶ asymmetric arrangement

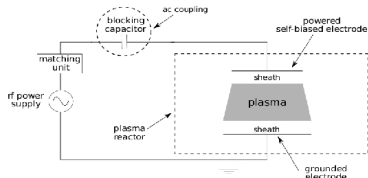
$$\xi = \frac{A_{\text{rf}}}{A_{\text{g}}} \approx 0.6$$

$$U_{\text{b}} = 0.83 V_{\text{rf}} \frac{\xi^q - 1}{\xi^q + 1}, \quad q = 1.25 - 4$$

for

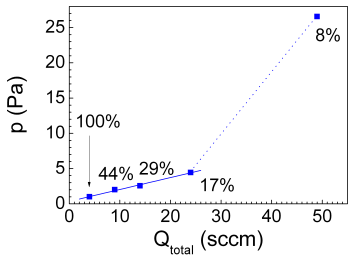
$$q = 2.5 \implies \xi = -0.55$$

- ▶ $c_{\text{hmdso}} = 5\text{--}100\%$
- ▶ $p = 1\text{--}40\text{ Pa}$
- ▶ $P = 100\text{--}450\text{ W}$
- ▶ $U_{\text{b}} = \text{from } -20 \text{ to } -335\text{ V}$

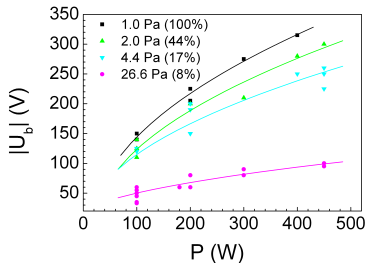
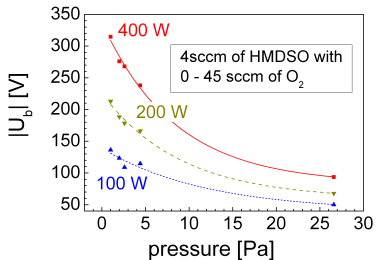


Relation between power, self-bias and pressure

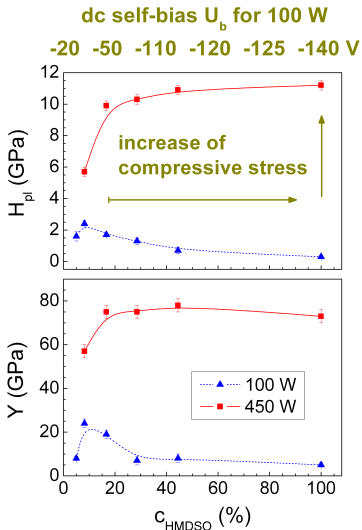
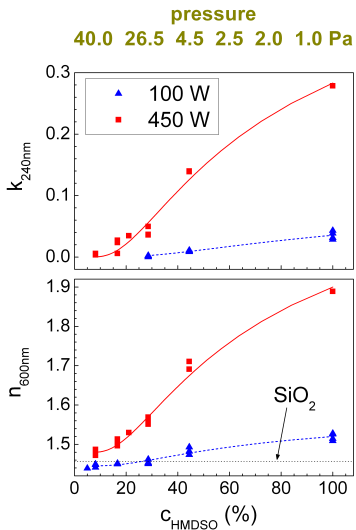
variation of HMDSO % in the mixture \iff variation of Q_{O_2} at the fixed $Q_{HMDSO} = 4$ sccm



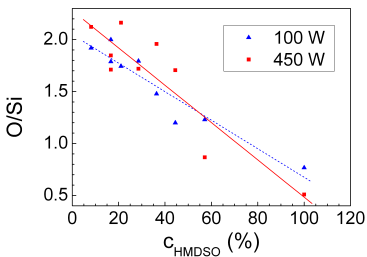
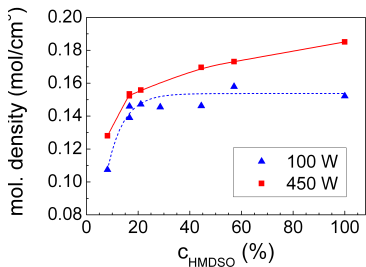
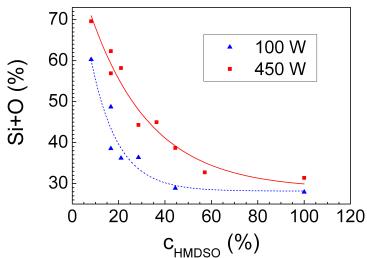
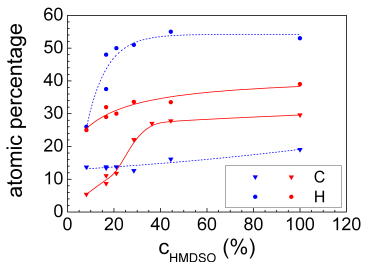
decrease of HMDSO % \implies increase of pressure
(especially for 5–8 %)
 \implies decrease of ion energy flux



Mechanical and optical properties



Composition and density

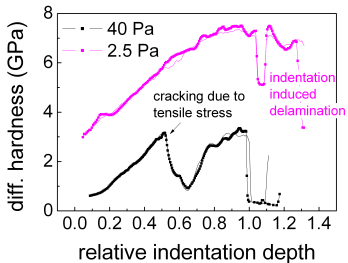


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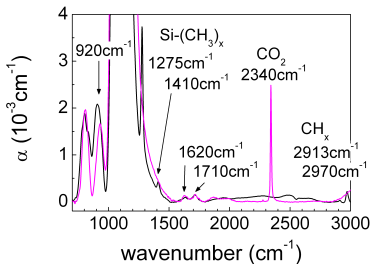
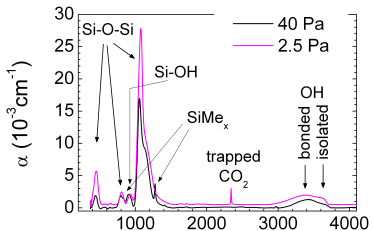
5% HMDSO, effect of pressure

p (Pa)	Q_{HMDSO} (sccm)	Q_{O_2} (sccm)	P (W)	U_b (V)
40	4	80	100	-20
2.5	0.7	14	100	-120



p (Pa)	H_{IT} (GPa)	E_{IT} (GPa)
40	1.6 ± 0.3	8 ± 1
2.5	6.9 ± 0.5	63 ± 4

Si (%)	O (%)	C (%)	H (%)
25	43	10	22
23	56	1	20

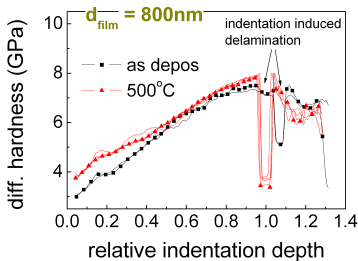
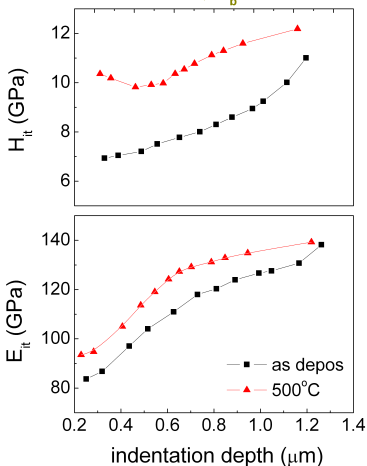


Temperature Induced Changes

- ▶ Annealing induced changes in SiO_2 -like films deposited in low pressure (2.5 Pa) CCP from 5 % HMDSO/ O_2
- ▶ Improvement of mechanical properties of SiO_2 -like films deposited in atmospheric pressure dielectric barrier discharge (DBD) by slight increase of deposition temperature
- ▶ Improvement of mechanical properties of $\text{SiO}_x\text{C}_y\text{H}_z$ film deposited in CCP from 8 % HMDSO/ O_2 at 450 W by annealing
- ▶ Annealing experiments for the films deposited in CCP from HMDSO-rich mixtures (17, 44 and 100 % of HMDSO)

5% HMDSO, 2.5 Pa / SiO₂-like film in CCP

5% HMDSO/O₂, 2.5 Pa,
P = 100W, U_b = -120V



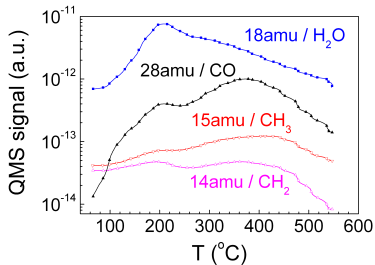
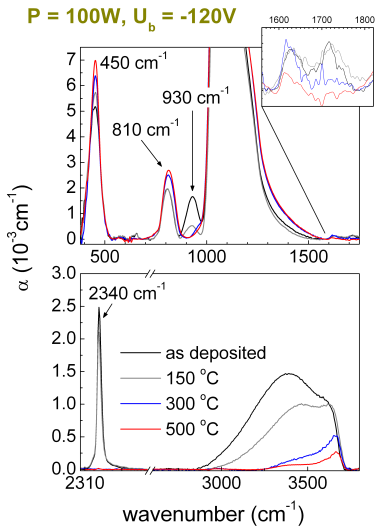
Originally:

- ▶ compressively stressed film with good fracture toughness

Annealing:

- ▶ slight stress relaxation due to annealing
- ▶ decrease of hardness induced by stress relaxation compensated by creation of new Si-O-Si bonds instead of Si-OH

5% HMDSO, 2.5 Pa / SiO₂-like film in CCP



As deposited film:

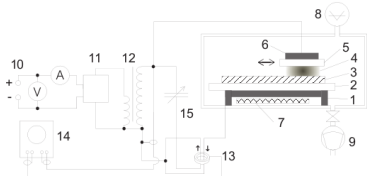
- ▶ Si 23%, O 56%, C 1%, H 20%

SiO₂-like films in atmospheric pressure DBD

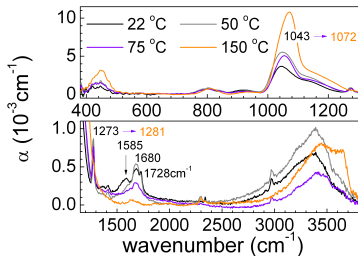
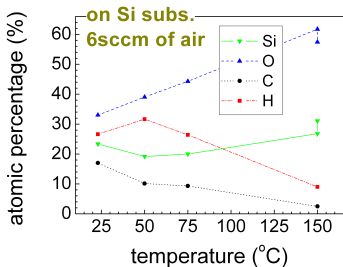
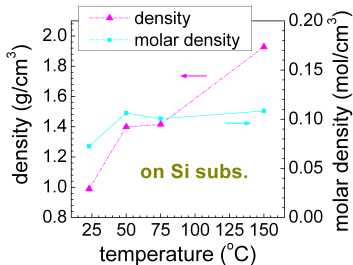
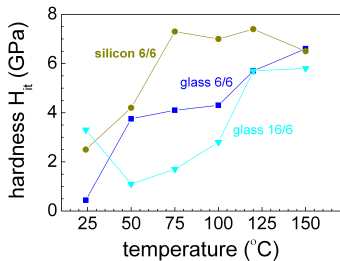


- ▶ Townsend-like (homogeneous) discharge at 6 kHz
- ▶ max. power 10 W/cm³
- ▶ discharge gap of 0.5 mm
- ▶ upper electrode covered by Simax (1.5 mm thick)
- ▶ bottom covered by glass substrate or glass plate (1mm thick) with Si substrate
- ▶ **substrate temperature 23–150 °C**

- ▶ HMDSO / synthetic air / nitrogen
- ▶ 6 sccm of N₂
- ▶ 6 slm of synthetic air
- ▶ 6 or 16 sccm of air through liquid HMDSO
 - ▶ 6 sccm ⇒ 70 ppm of HMDSO, 200 ppm of O₂ in N₂
 - ▶ 16 sccm ⇒ 173 ppm of HMDSO, 532 ppm of O₂ in N₂



SiO₂-like films in atmospheric pressure DBD



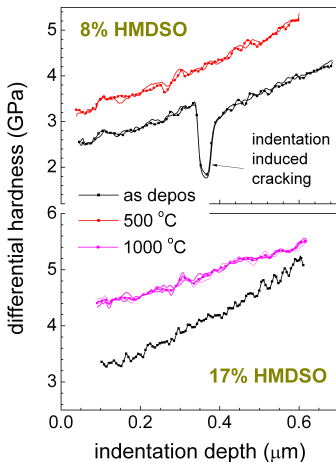
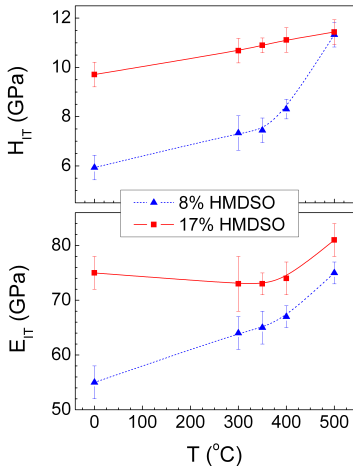
CCP 450 W, 8 and 17% HMDSO

Critical depth for indentation induced crack initiation in μm :

8% HMDSO: $p = 26.5\text{Pa}$, $U_b = -100\text{V}$

17% HMDSO: $p = 4.5\text{Pa}$, $U_b = -250\text{V}$

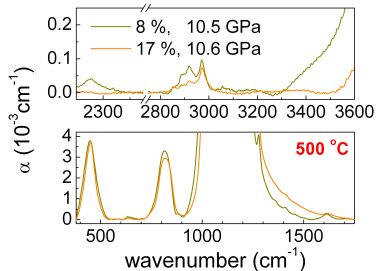
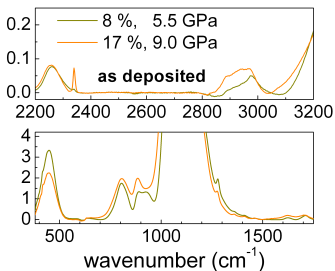
% / T ($^{\circ}\text{C}$)	-	300	350	400	500
8	0.34	0.42	0.44	0.45	> 1
17	> 1	> 1	> 1	> 1	0.46



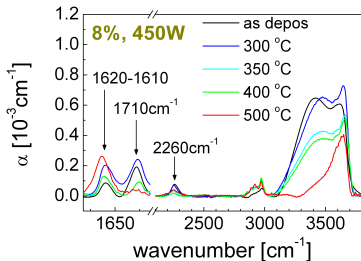
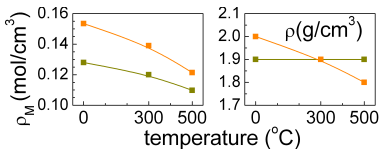
CCP 450 W, 8 and 17 % HMDSO

as deposited:

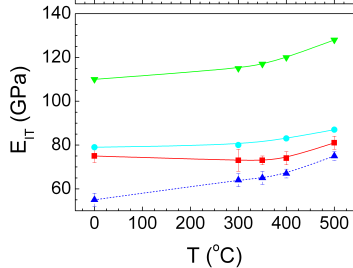
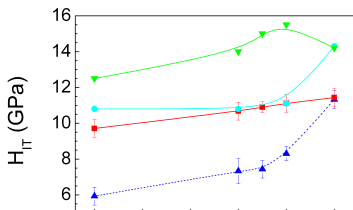
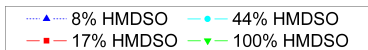
%	8%	17%
Si	22	20
O	47	37
C	5	11
H	25	32



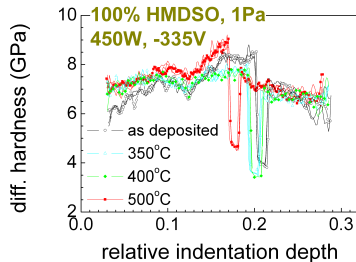
% HMDSO	T (°C)	O/Si	H/Si	C/Si
8	-	2.12	1.12	0.24
8	300	1.94	0.81	0.29
8	500	1.93	0.55	0.21
17	-	1.85	1.60	0.56
17	300	1.79	1.37	0.57
17	500	1.99	1.01	0.57



CCP 450 W, 44 and 100 % HMDSO



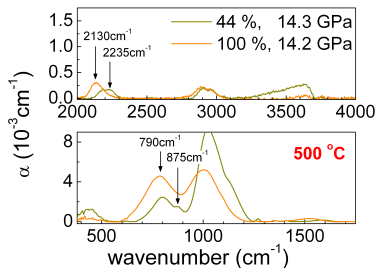
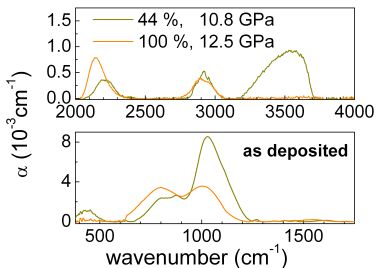
$c_{\text{hm dso}}$ (%)	8	17	44	100
p (Pa)	26.5	4.5	2	1
U_b (V)	-100	-250	-300	-335



- ▶ 100 and 44 % films are compressively stressed but the hardness does not decrease with annealing temperature
- ▶ fracture toughness is not as good as for 8 and 17% films and does not improve with annealing

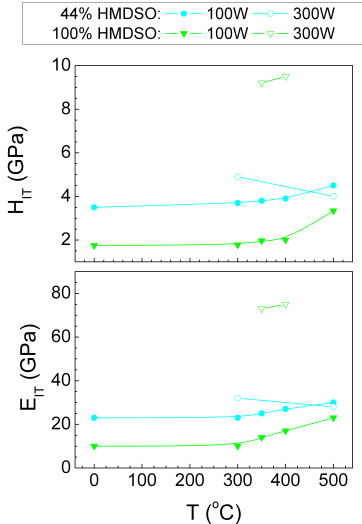
CCP 450 W, 44 and 100 % HMDSO

c_{hmdso}	T ($^{\circ}\text{C}$)	Si (%)	O (%)	C (%)	H (%)	O/Si	C/Si	H/Si
44 %	–	15	18	28	39	1.2	1.9	2.6
44 %	500	14	20	31	35	1.4	2.2	2.5
100 %	–	17	9	32	42	0.5	1.9	2.5
100 %	500	20	7	39	34	0.4	2.0	1.7



c_{hmdso}	T ($^{\circ}\text{C}$)	ρ (g/cm^3)	ρ_{M} (mol/cm^3)
44 %	–	1.8	0.17
44 %	500	1.8	0.16
100 %	–	1.8	0.17
100 %	500	1.8	0.15

CCP 100 and 300 W, 44 and 100 % HMDSO

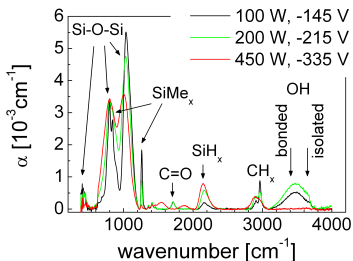


- ▶ Films deposited at 300 W were not stable \Rightarrow peeled off when not annealed

Self-bias U_b for 44 and 100 % HMDSO/O₂ (2 and 1 Pa) at different rf powers:

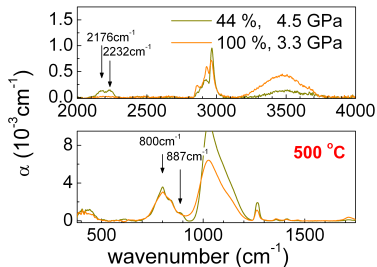
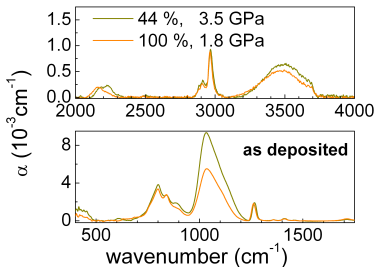
P (W) / c_{hmdso} (%)	44	100
100	-120	-145
300	-230	-270
450	-300	-335

- ▶ 100 % HMDSO film is soft polymeric material which hardness can be slightly increased by annealing and significantly improved by increased rf power.



CCP 100 W, 44 and 100 % HMDSO

c_{hmdso}	T ($^{\circ}\text{C}$)	Si (%)	O (%)	C (%)	H (%)	O/Si	C/Si	H/Si
44 %	–	15	20	17	48	1.3	1.1	3.2
44 %	500	15	25	20	40	1.7	1.3	2.7
100 %	–	12	18	21	49	1.5	1.8	4.1
100 %	500	12	19	22	47	1.6	1.8	3.9



c_{hmdso}	T ($^{\circ}\text{C}$)	ρ (g/cm^3)	ρ_{M} (mol/cm^3)
44 %	–	1.4	0.14
44 %	500	1.4	0.12
100 %	–	1.2	0.13
100 %	500	1.3	0.14

Conclusion

- ▶ Understanding of the changes in film mechanical properties (because of different deposition parameters or annealing) requires complete study of the film composition, chemical bonds and film density. Thermal desorption spectroscopy is advantageous.
- ▶ Molar density increased with increased deposition temperature of APTD films and decreased with annealing temperature of CCP films.
- ▶ Significant hydrocarbon desorption observed only for increased deposition temperature of APTD-SiO_xC_yH_z films.
- ▶ For CCP films, carbon desorption detected only for annealing of SiO₂-like film (5% HMDSO, 2.5 Pa) - rather decrease of CO₂ than CH_x. Desorption of -OH and -H more important.
- ▶ Annealing of compressively stressed CCP films led to their stabilization due to stress relaxation.
- ▶ Annealing of hard compressively stressed films (either SiO_x:H or SiO_xC_yH_z) did not decrease but increase the hardness - stress relaxation is compensated by cross-linking of material, i.e. replacing the end-groups (Si-OH, Si-H, C=O) by new strong bonds.