

C8953

NMR structural analysis seminar

Information about classes + 1D ^1H -NMR

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Information about classes

Credit:

- ▶ Max. 2 unexcused absences
- ▶ Successfully solved spectra of final exercise

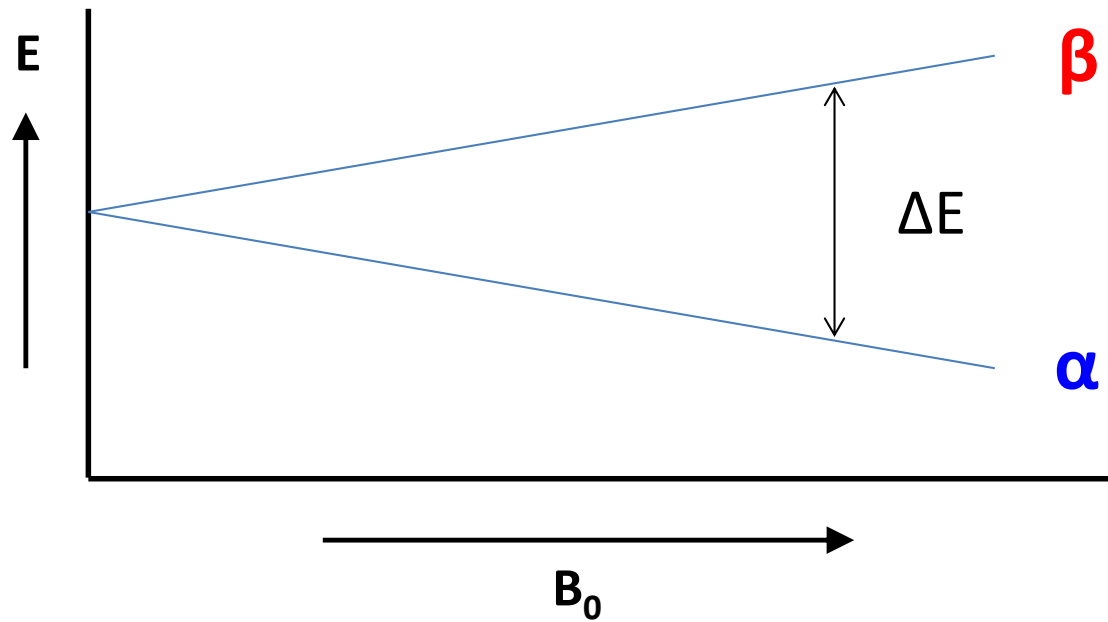
Study materials:

<https://is.muni.cz/auth/el/1431/jaro2017/C8953/um>

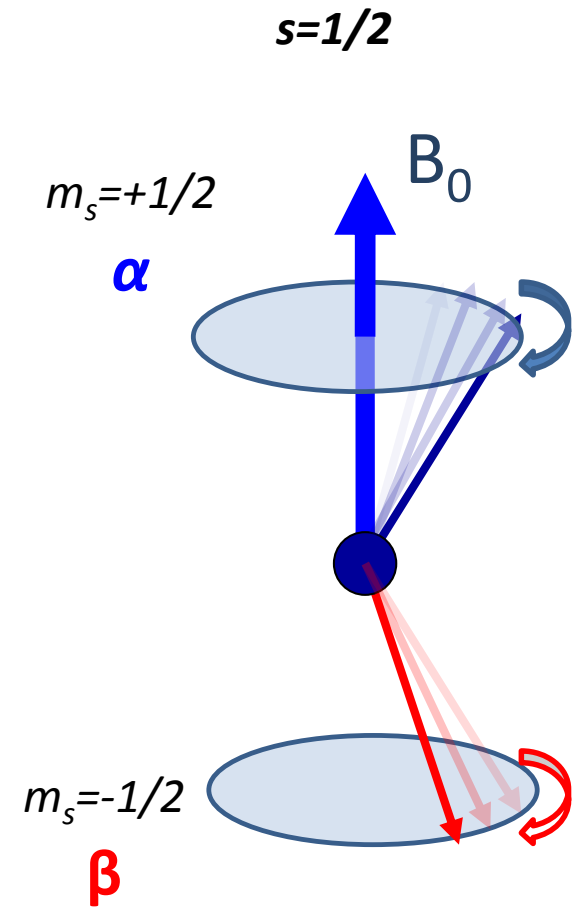
E-tests:

<https://is.muni.cz/auth/el/1431/jaro2017/C8953/odp>

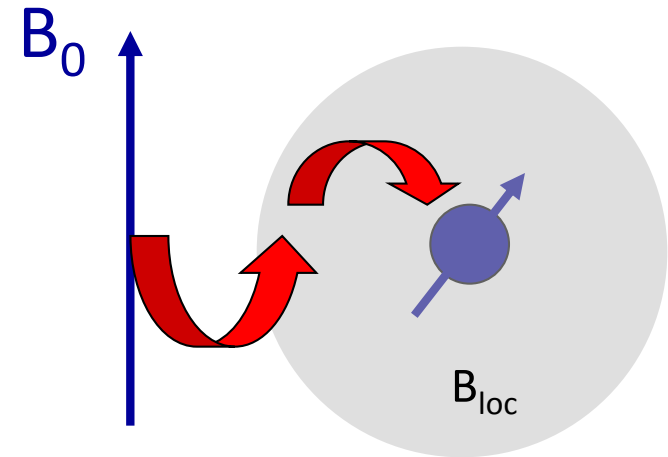
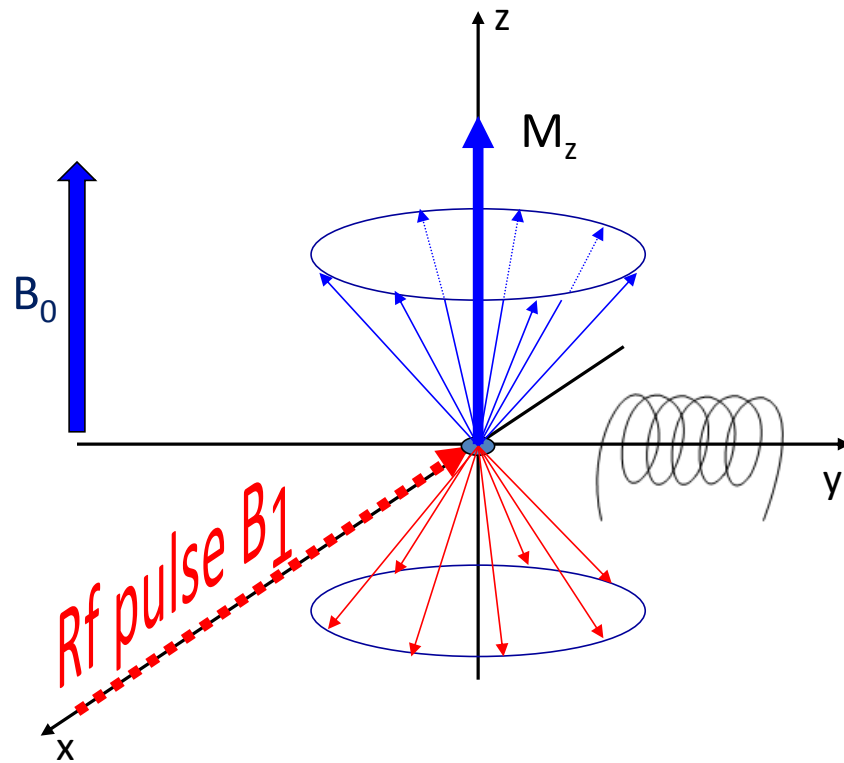
Energy levels splitting



$$N_\alpha > N_\beta$$



Behavior of nuclear spin after irradiation by RF pulse



B_0 induces local mag. field B_{loc} , which affects against B_0

↓
Nuclear shielding

Precession frequency:

Precession frequency affected by nuclear shielding:

Chemical shift:

Definition of the relative scale of the chemical shift:

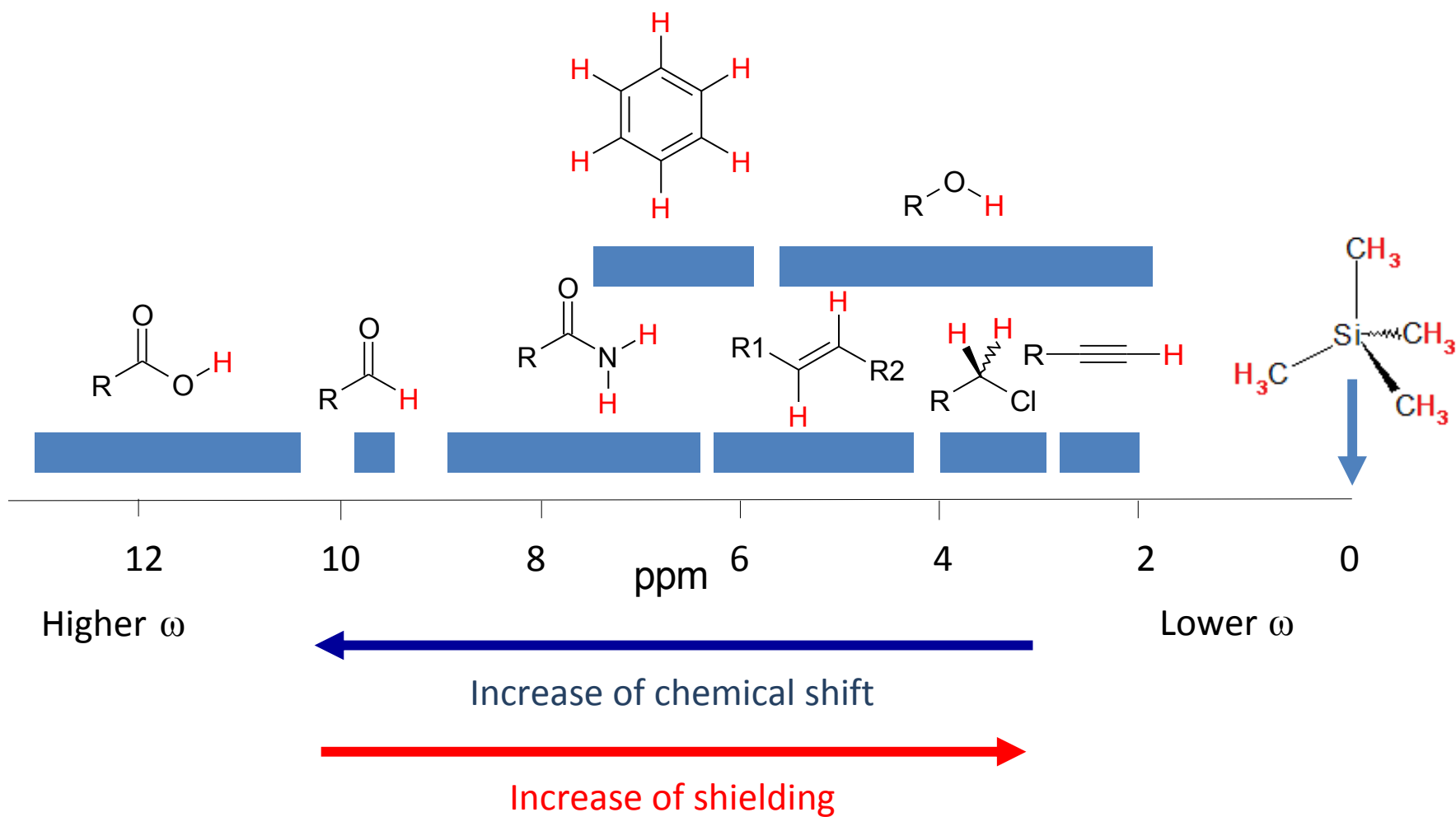
$$\omega = -\gamma B_0$$

$$\omega = -(1+\sigma)B_0$$

$$\delta = \omega - \omega_{ref}$$

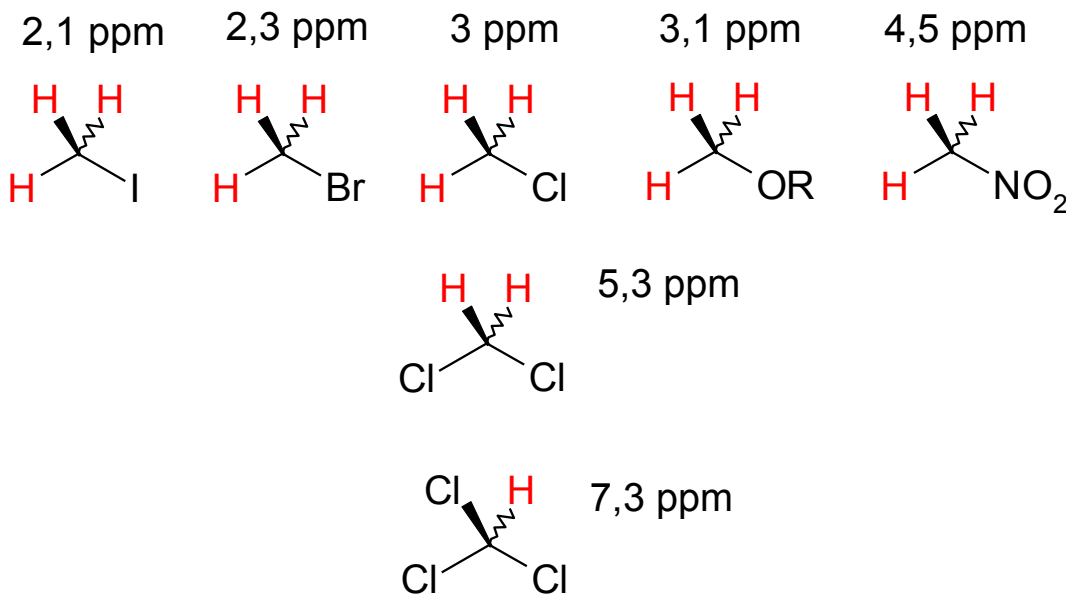
$$\delta = (\omega - \omega_{ref})/\omega_{ref} \cdot 10^6 \text{ ppm}$$

Characteristic intervals of chemical shifts values



Trends in chemical shifts

- ▶ Electronegativity, inductive and mesomeric effects of substituents
- ▶ Hybridisation
- ▶ Relative position towards the ring, double bond



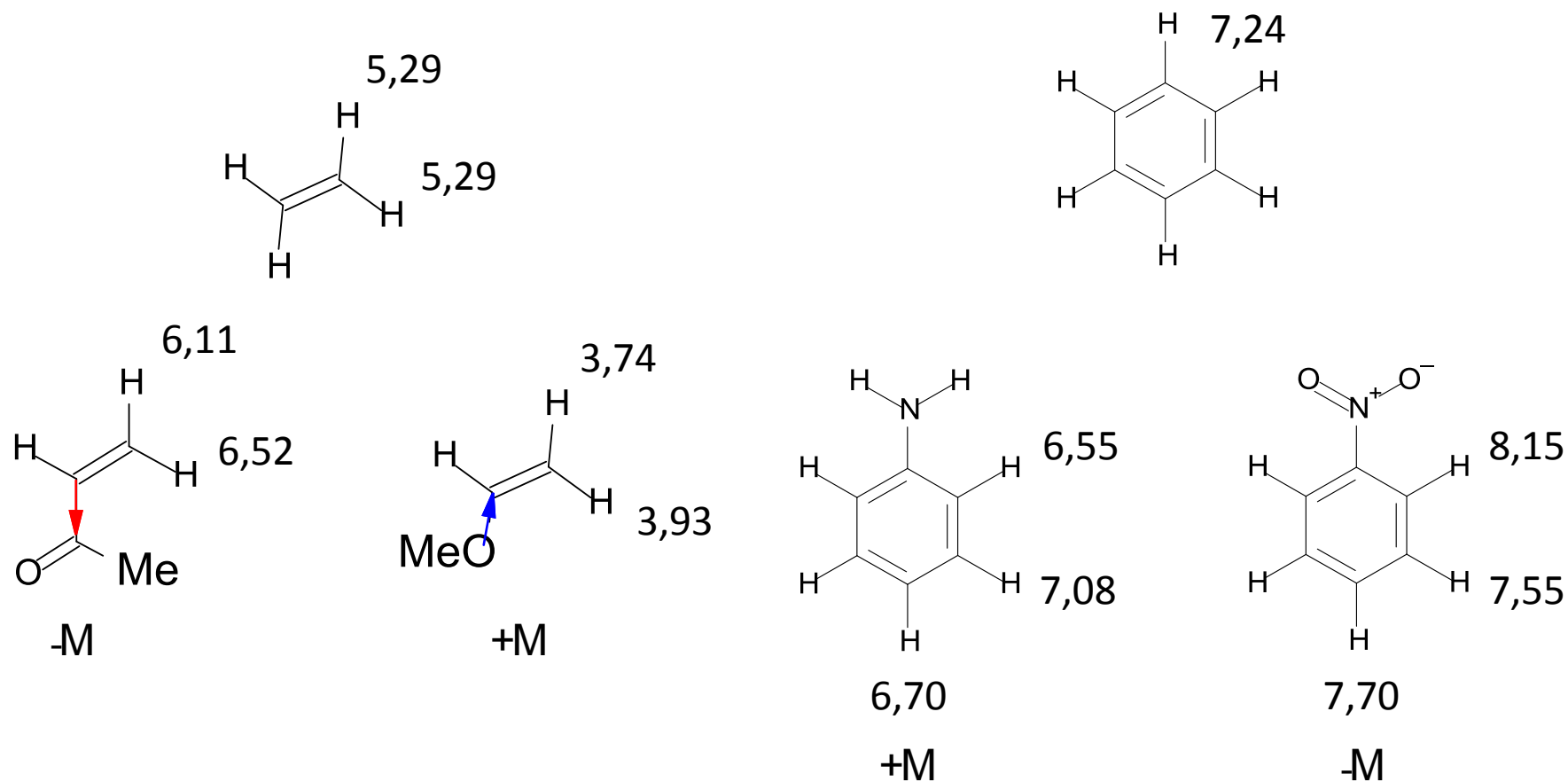
Substituents with -I effect

$=N^+R_2 > -N^+R_3 > -NO_2 > -NR_2$
 $-SO_2R > -SO_3 > -SOR > -SR$
 $-F > -OR > -NR_2 > -CR_3$
 $-F > -Cl > -Br > -I$
 $\equiv N > =NR > -NR_2$
 $-C\equiv CH > -CH=CH_2 > -CH_2-CH_3$

Substituents with +I effects

$-N-R > -O->S-$
 $-C(CH_3)_3 > -CH(CH_3)_2 > -CH_2CH_3 > -CH_3$
metals

Mesomeric effect



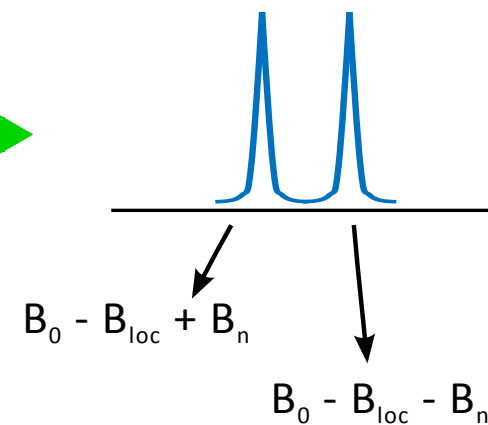
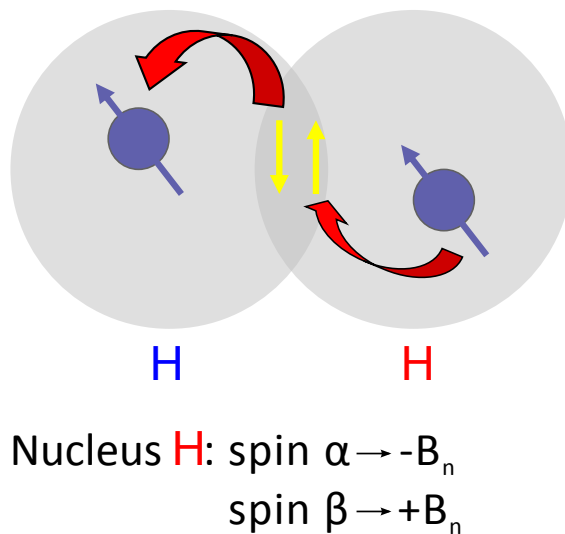
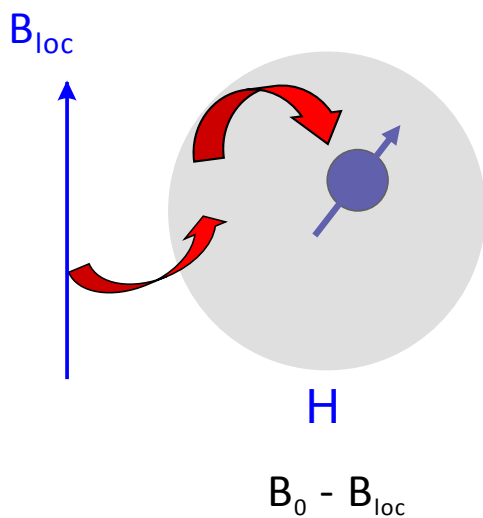
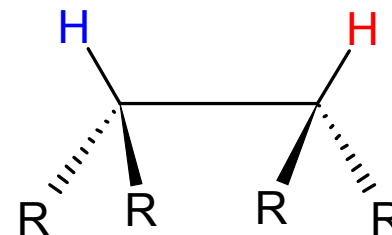
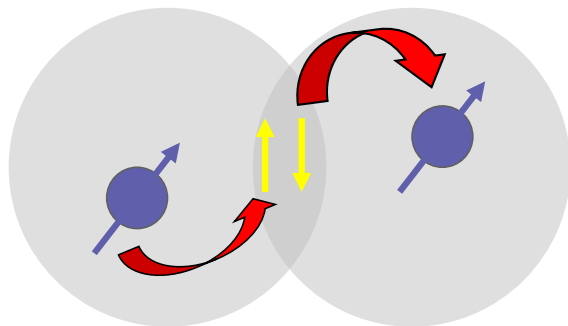
Substituents with -M effects

-F, -Cl, -Br, -I, -OH, -OR, -NH₂, -NHR, -NR₂, -SH, -SR

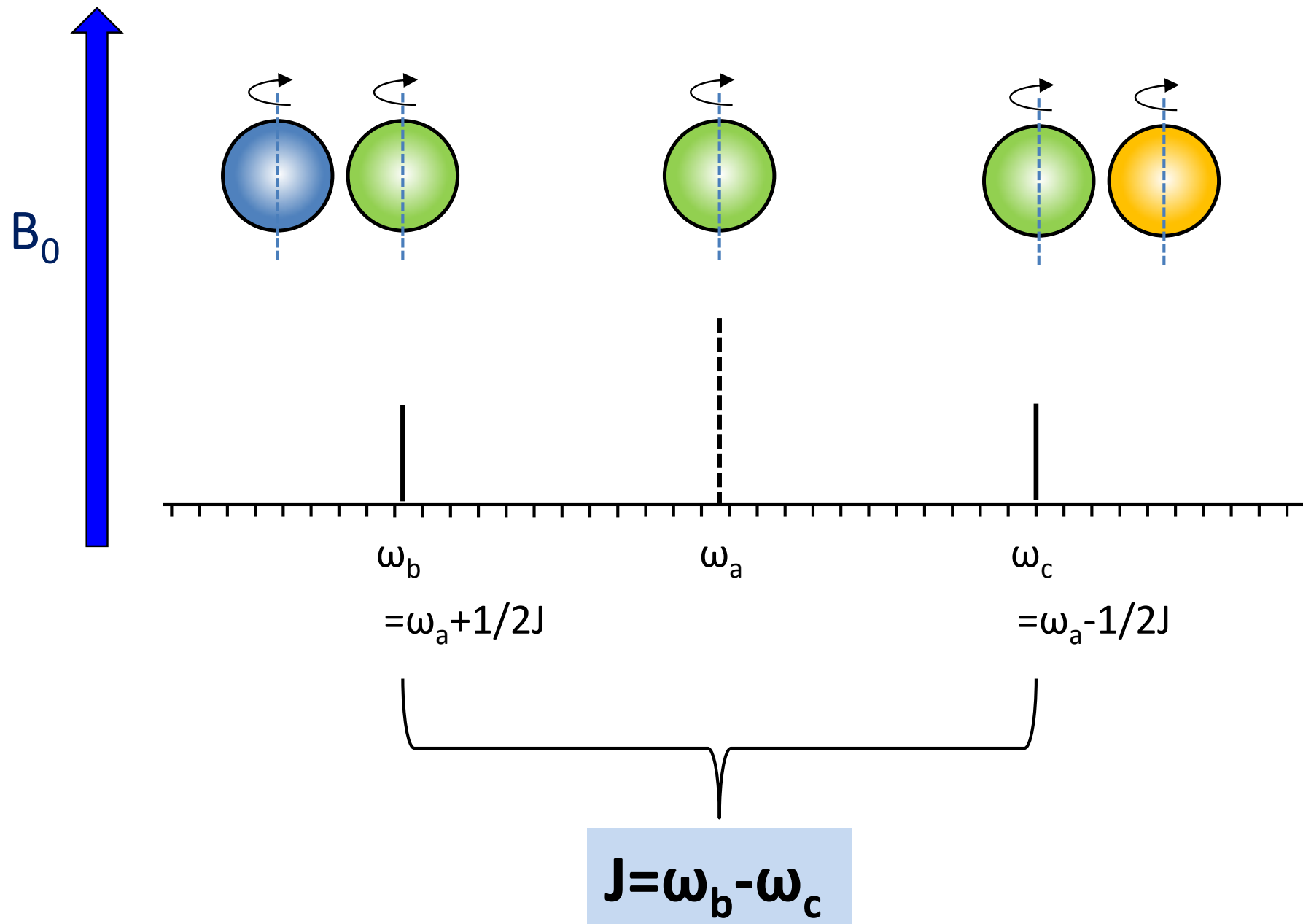
Substituents with +M effect

-CH=O, -RC=O, -C(OH)=O, -C(OR)=O, -C(NH₂)=O, -NO₂, -SO₃H, -C≡N

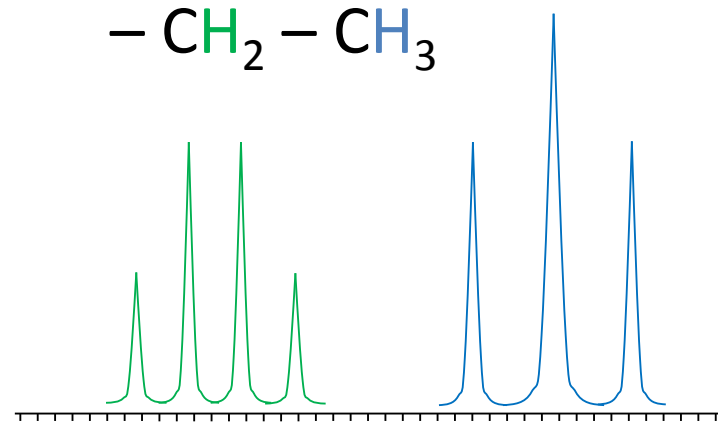
Spin-spin interaction, J -coupling



Interaction constant J



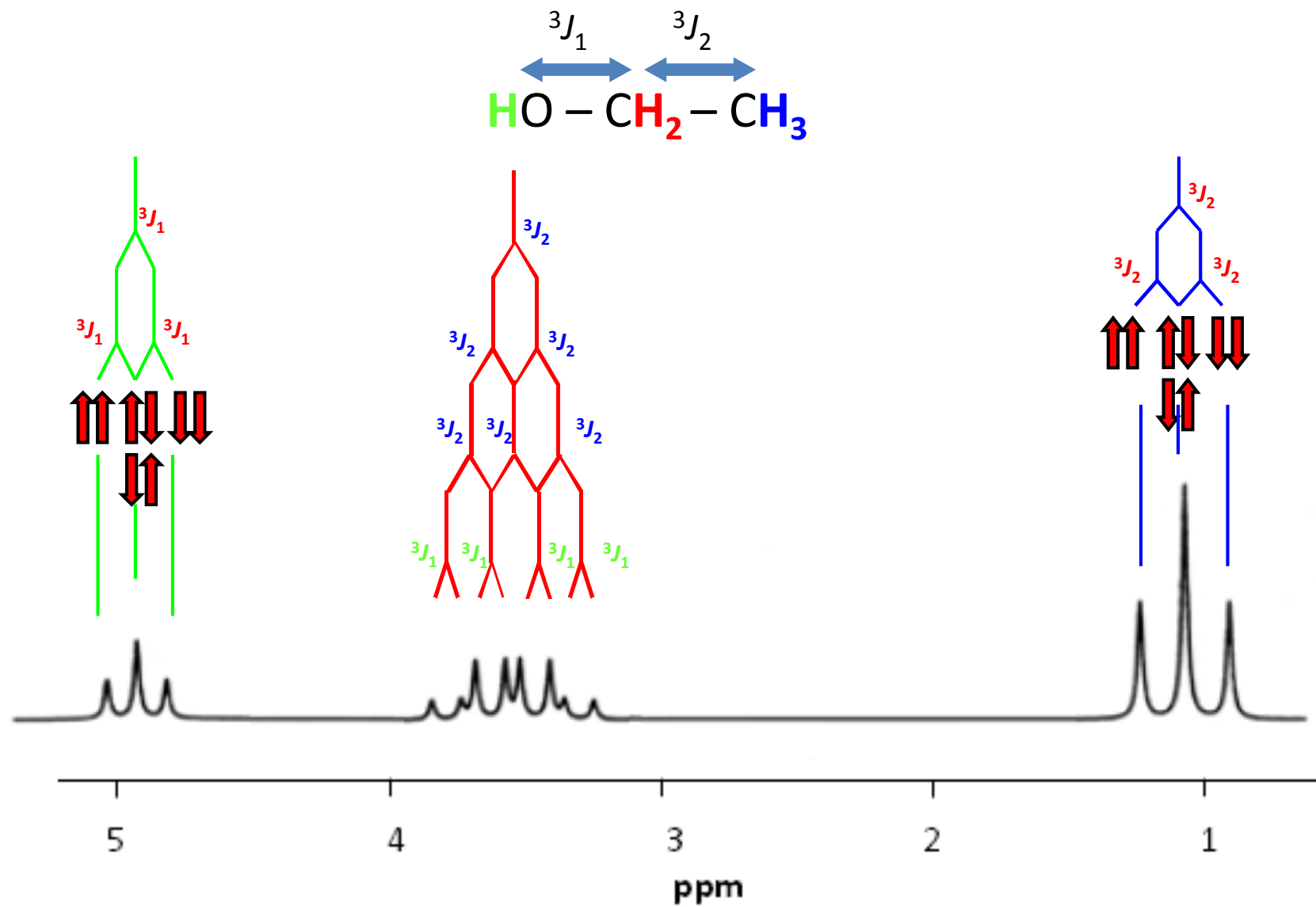
Interaction constant J



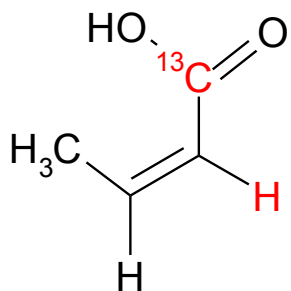
- ▶ Multiplicity of the nucleus I with the spin 1/2 is given by:
 $m = n + 1$, $n =$ number of interacting nuclei with nucleus I
- ▶ Intensity of lines in multiplet follows Pascal's triangle

			1				
			1		1		
		1		2		1	
		1	3		3	1	
	1	4		6		4	1
1	5	10		10		5	1

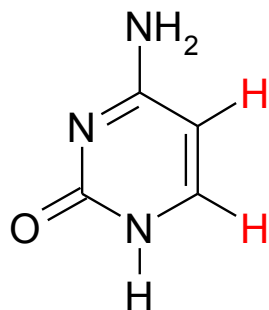
1D ^1H NMR spectrum



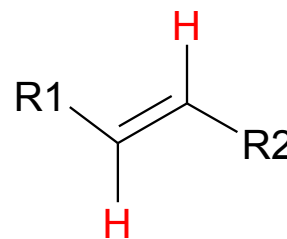
Values of J -constants - trends



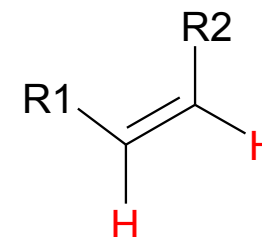
$${}^2J_{CH} = 3.1 \text{ Hz}$$



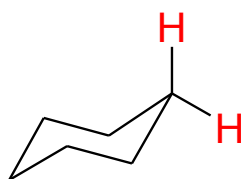
$${}^3J_{HH} = 12 \text{ Hz}$$



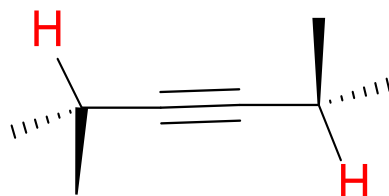
$${}^3J_{HH} = 13 - 18 \text{ Hz}$$



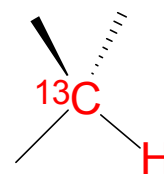
$${}^5J_{HH} = 7 - 12 \text{ Hz}$$



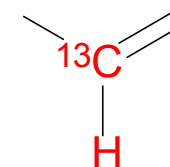
$${}^2J_{HH} = -12,5 \text{ Hz}$$



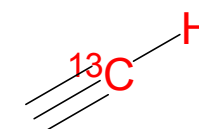
$${}^5J_{HH} = 2 - 3 \text{ Hz}$$



$${}^1J_{CH} = 125 \text{ Hz}$$

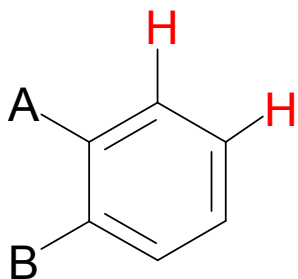


$${}^1J_{CH} = 160 \text{ Hz}$$

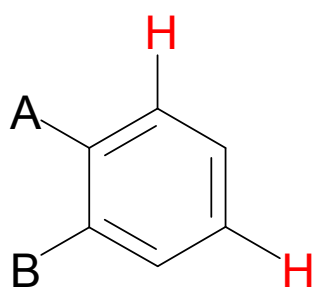


$${}^1J_{CH} = 250 \text{ Hz}$$

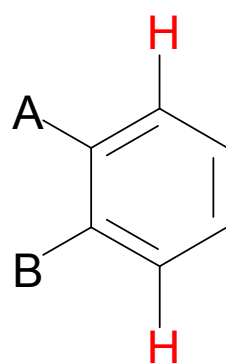
Values of J-constants - trends



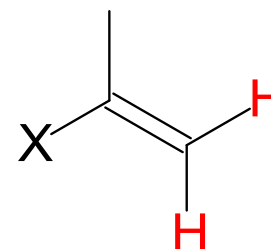
$${}^3J_{HH} = 7,5 \text{ Hz}$$



$${}^4J_{HH} = 1,5 \text{ Hz}$$



$${}^5J_{HH} = 0,7 \text{ Hz}$$



X=	Li	H	Cl	OMe	F
${}^2J_{HH}$ (Hz)	7,1	2,5	-1,4	-2,0	-3,2

1D ^1H NMR spectroscopy

- ▶ the fastest measuring, the highest sensitivity
- ▶ complicated interpretation in case of more complex systems

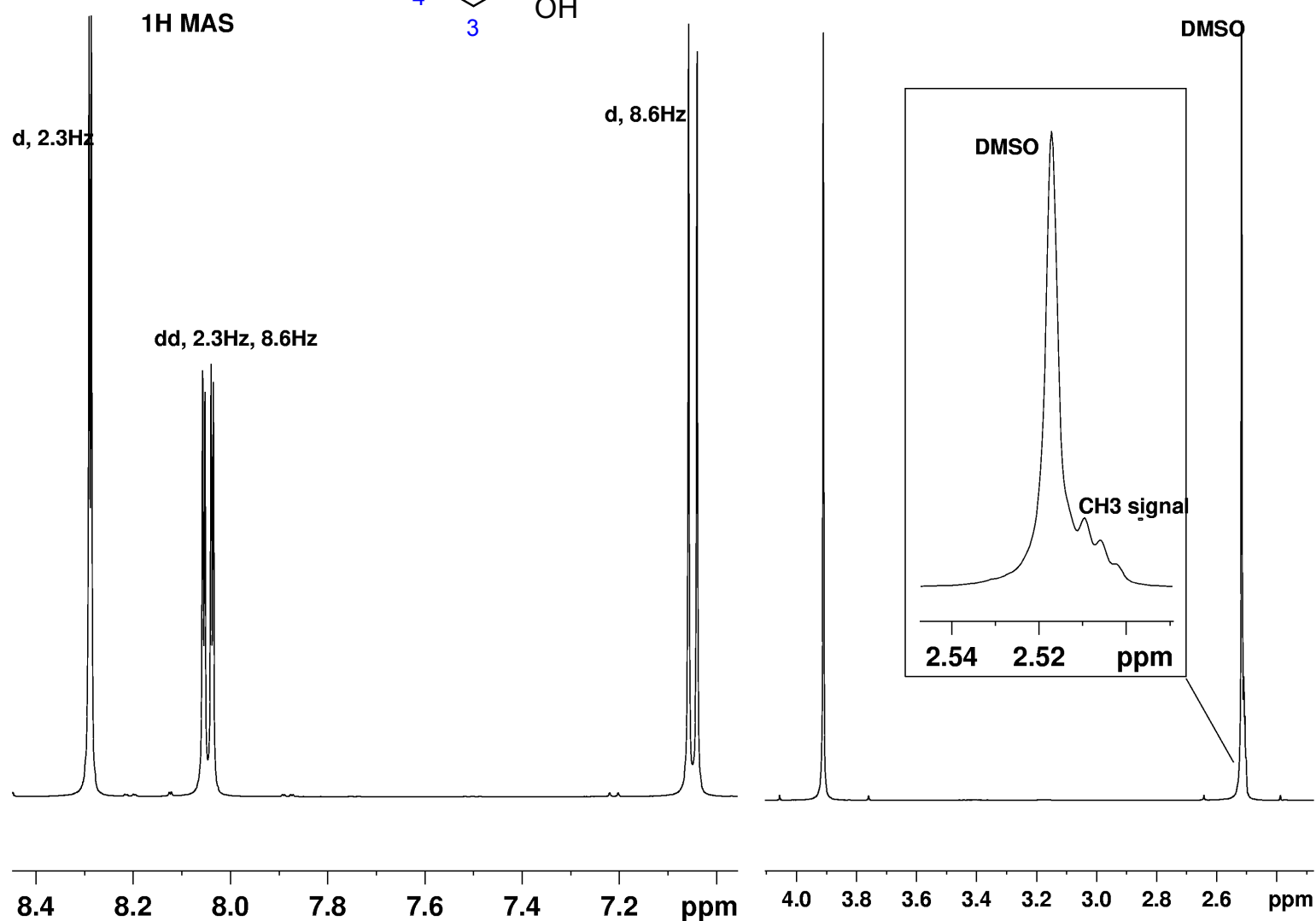
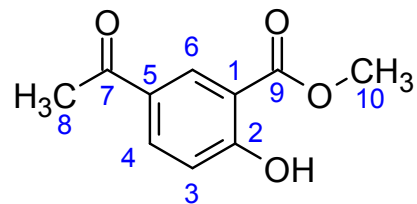
We are looking for:

- ▶ position of the signal (ppm)
- ▶ multiplicity (2J , 3J , 4J)
- ▶ intensity (integral)
- ▶ halfwidth

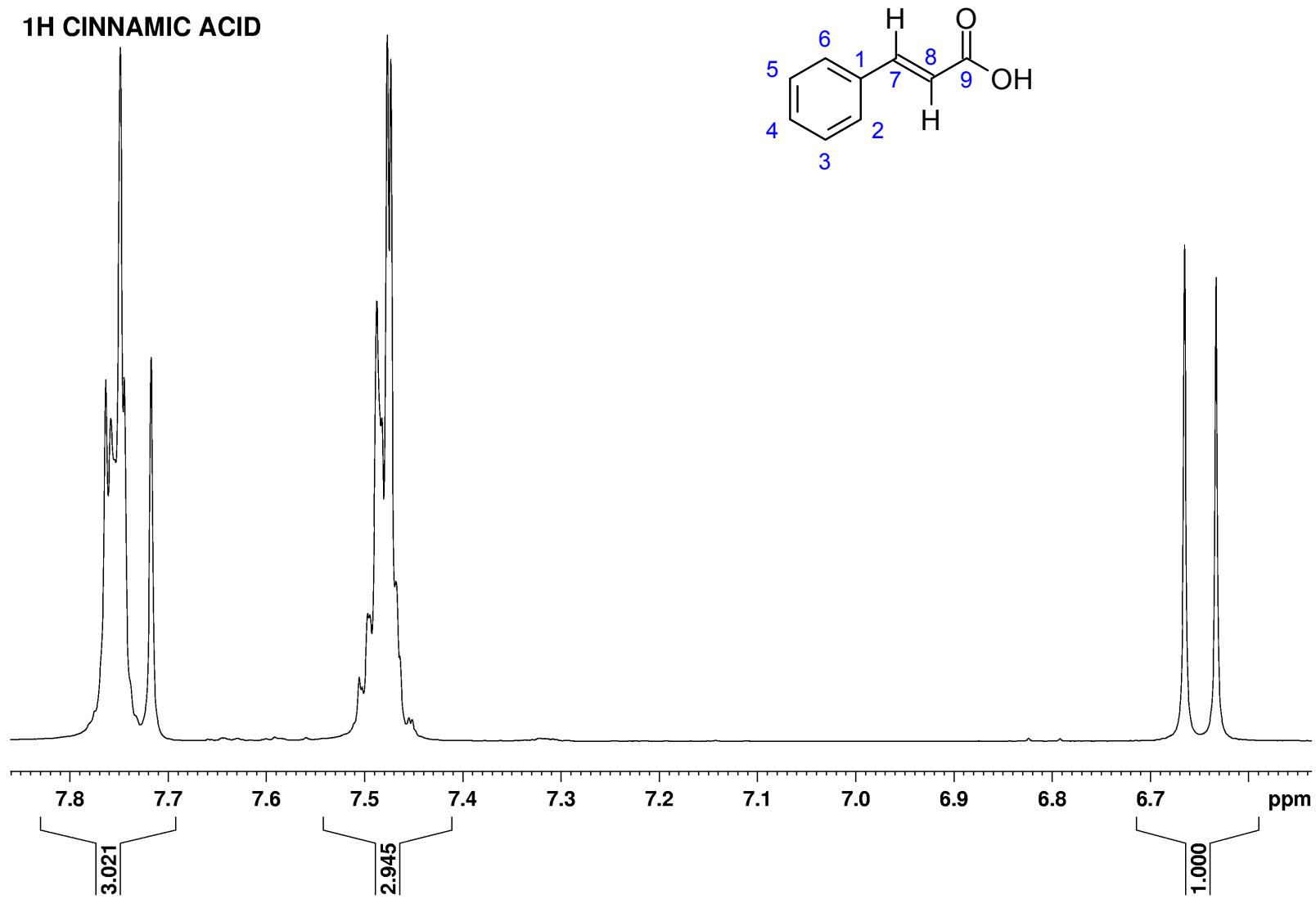
We are considering:

- ▶ chemical/magnetic equivalence
- ▶ enantiotopicity/diastereotopicity
- ▶ averaging of signals (dynamics, chemical exchange)

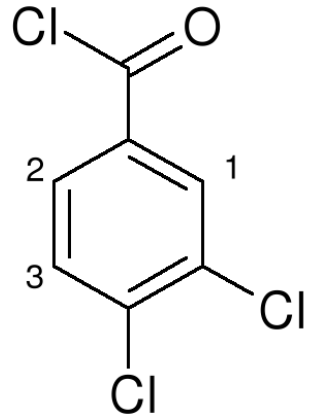
1D ^1H NMR spectrum of methyl-5-acetylsalicylate



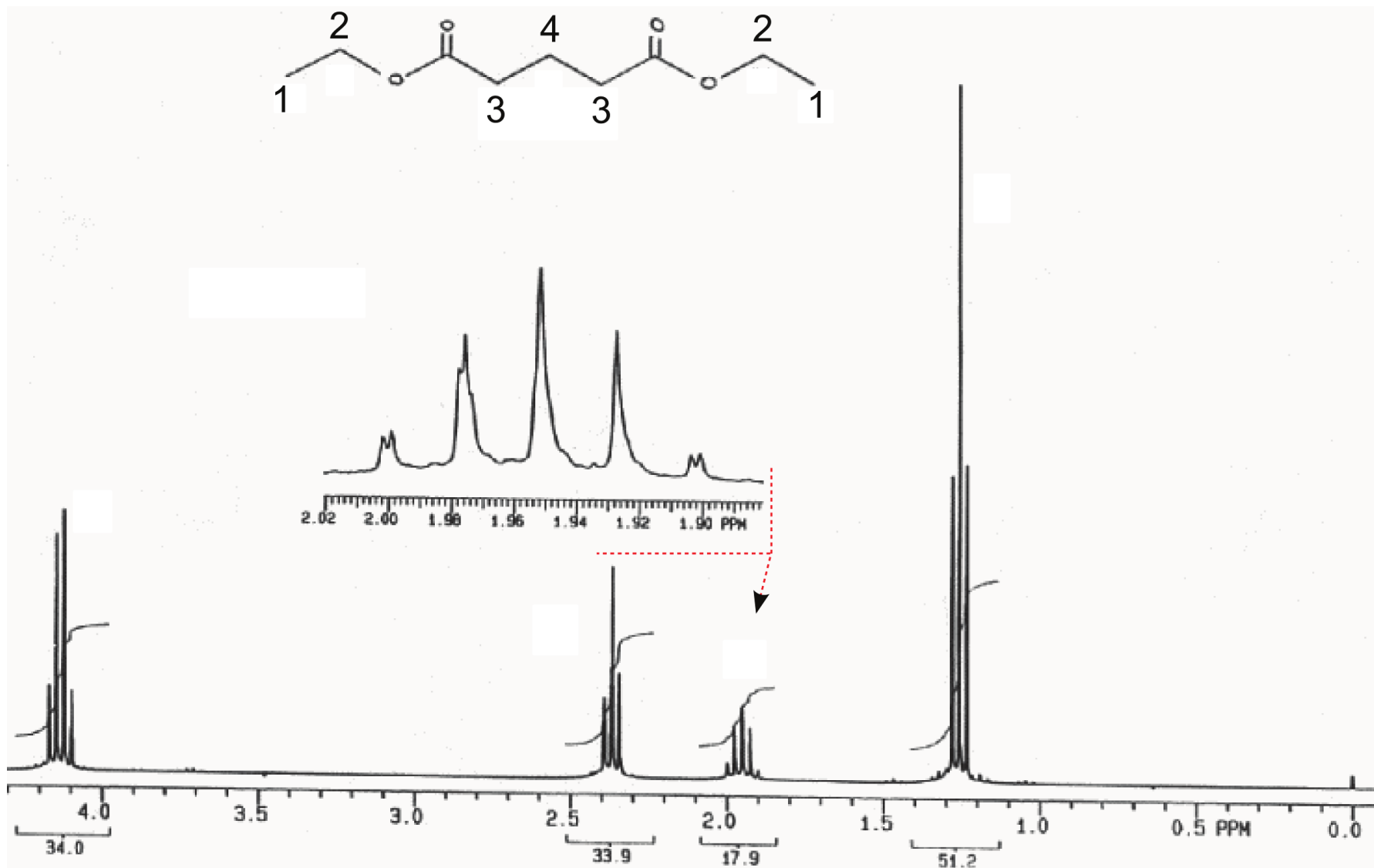
1D ^1H NMR spectrum of cinnamic acid



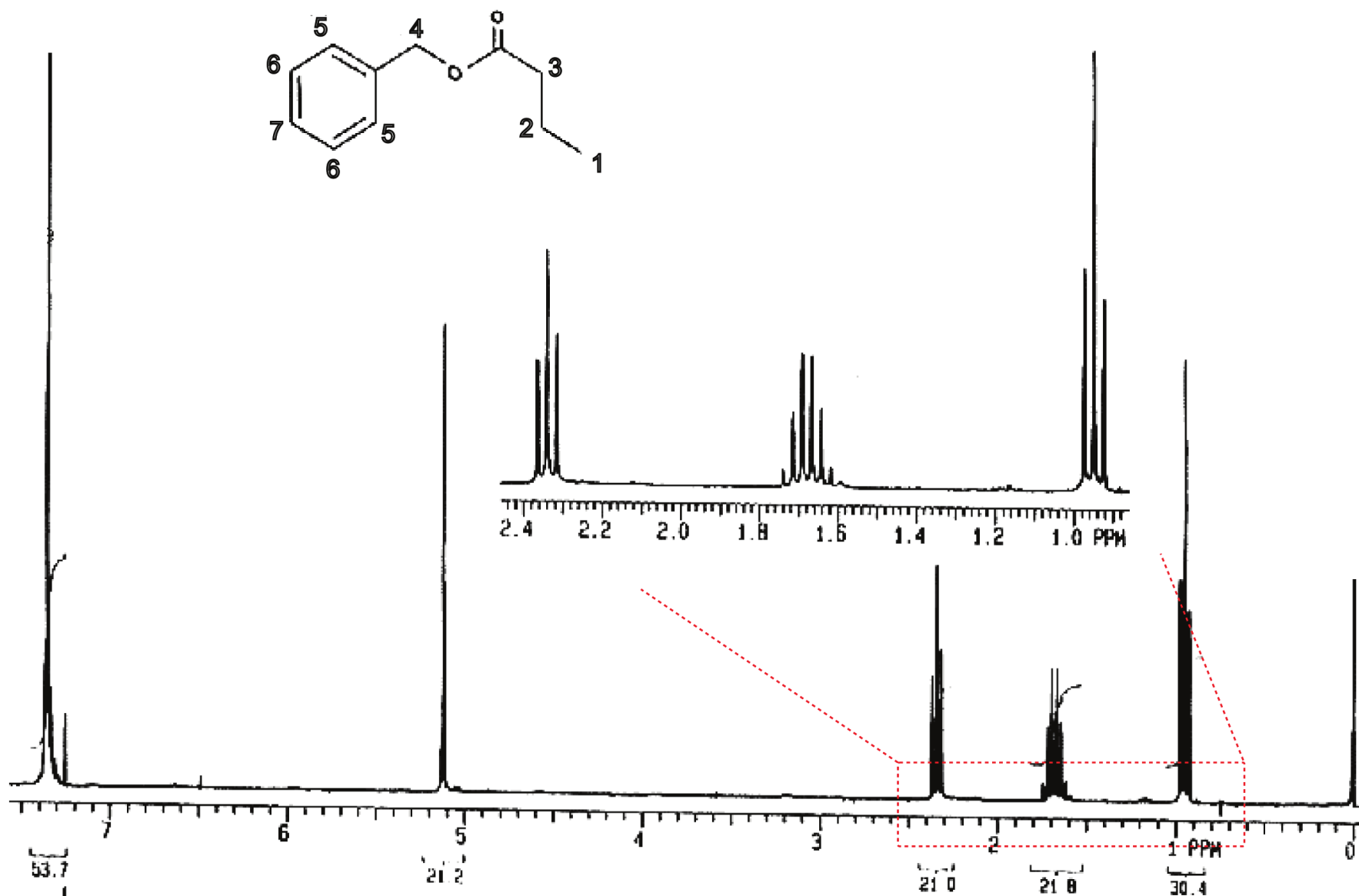
Draw approximate 1D ^1H NMR spectrum of the following compound



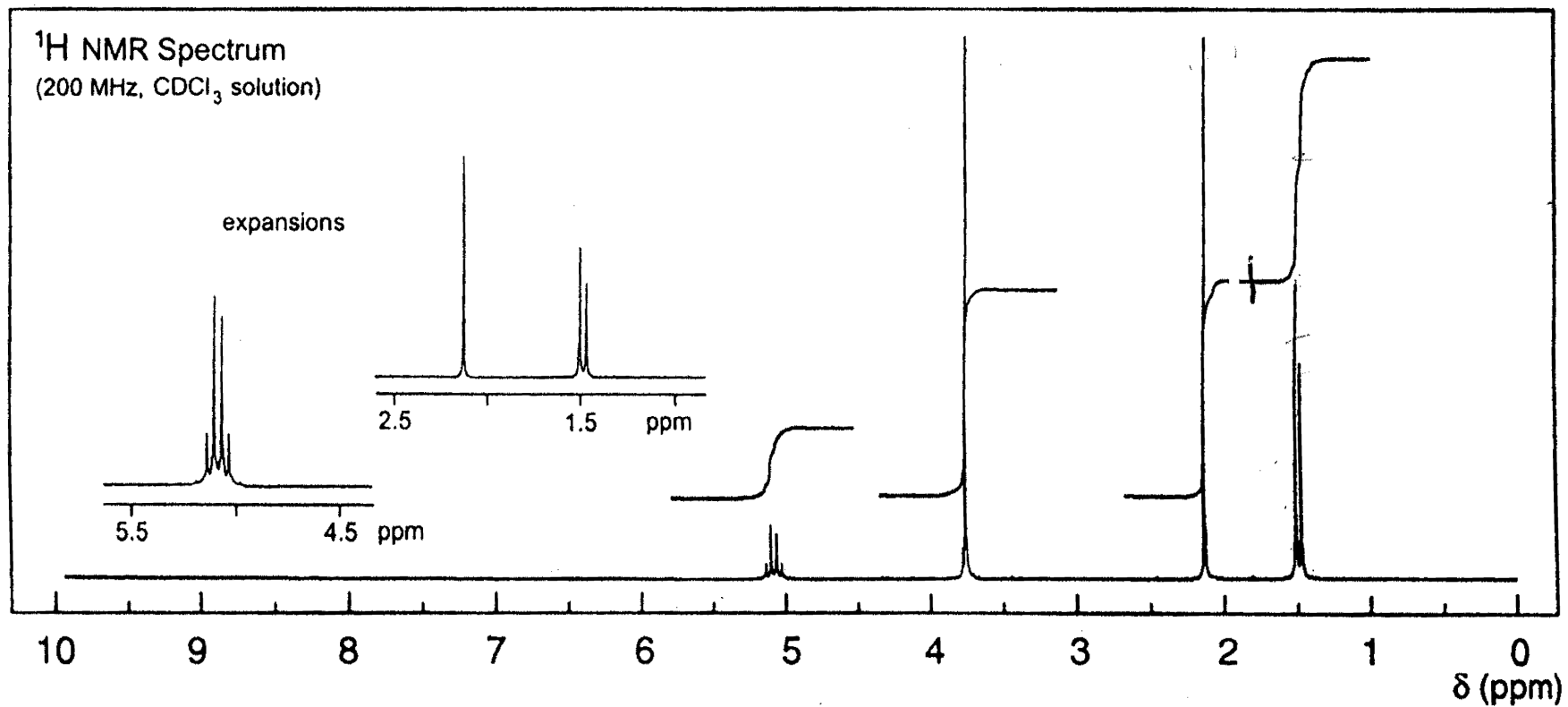
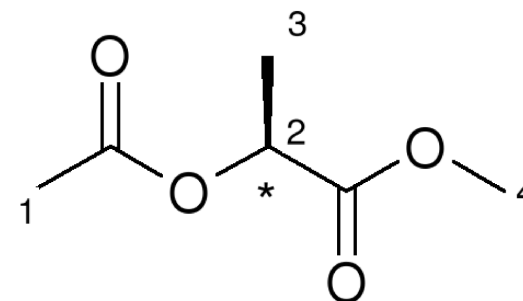
1D ^1H NMR spectrum of ethyl glutarate



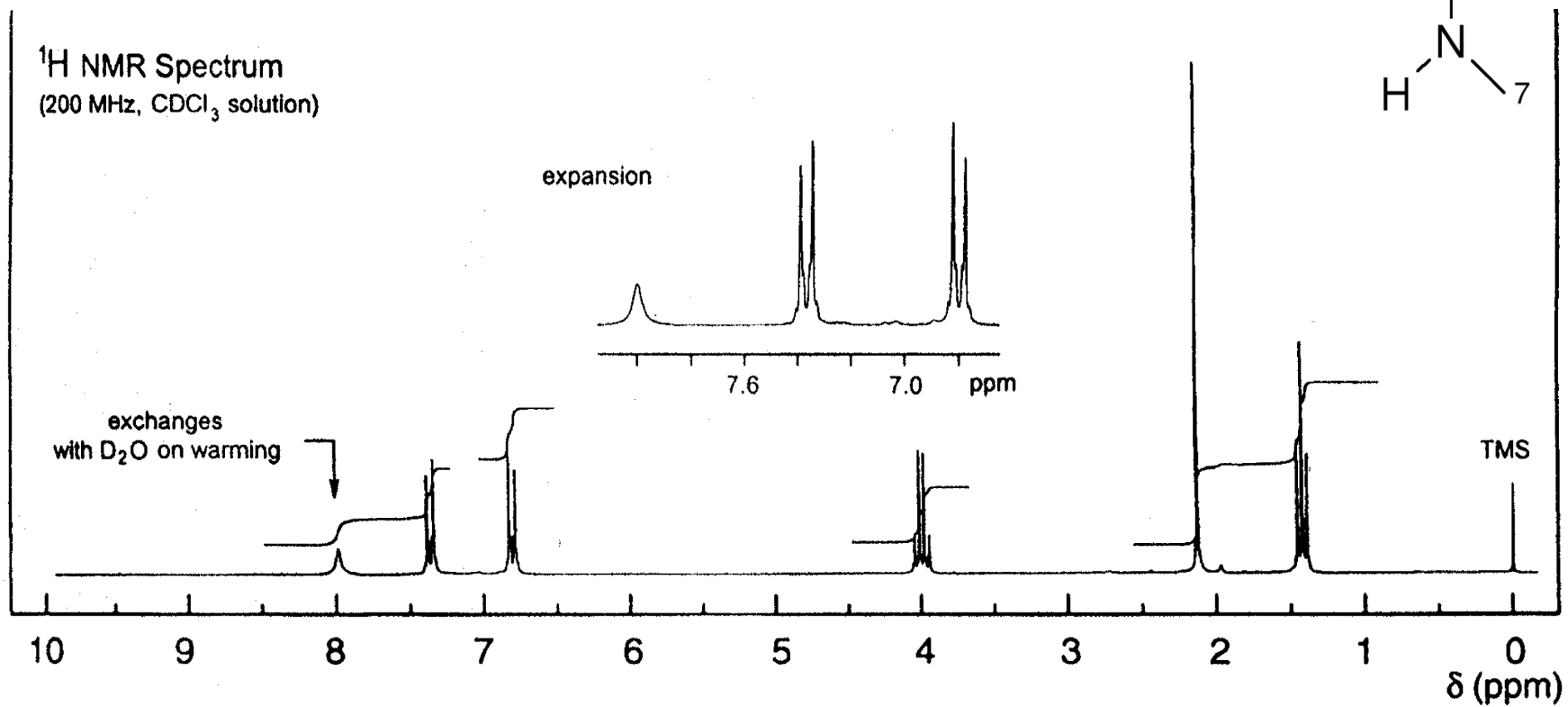
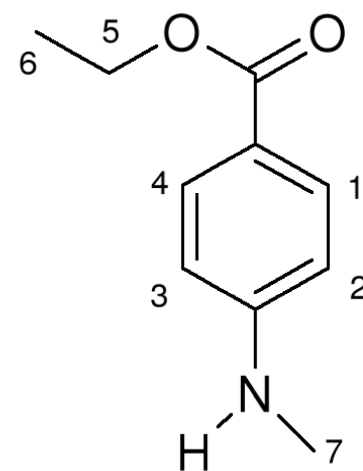
1D ^1H NMR spectrum of benzyl butyrate



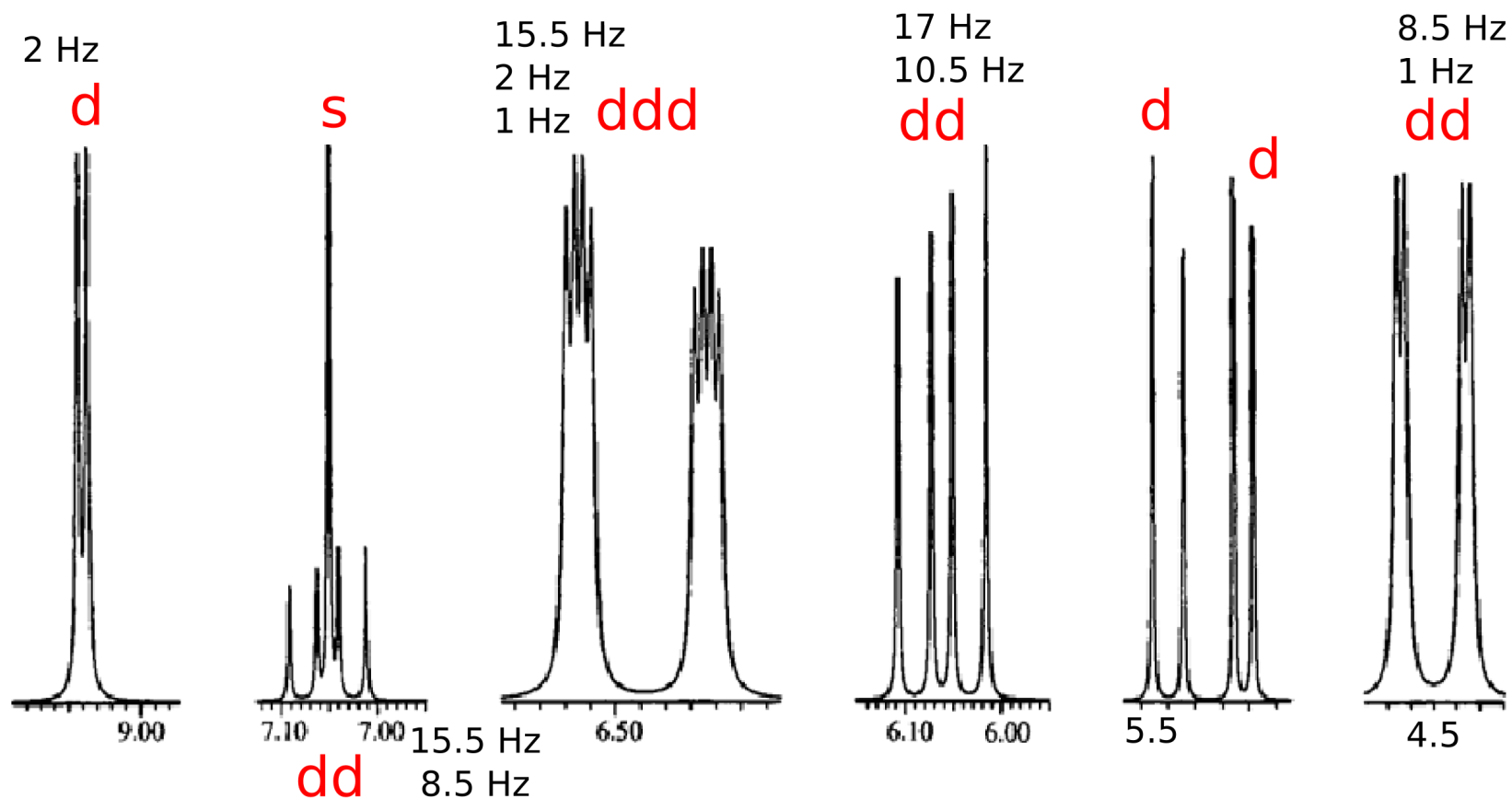
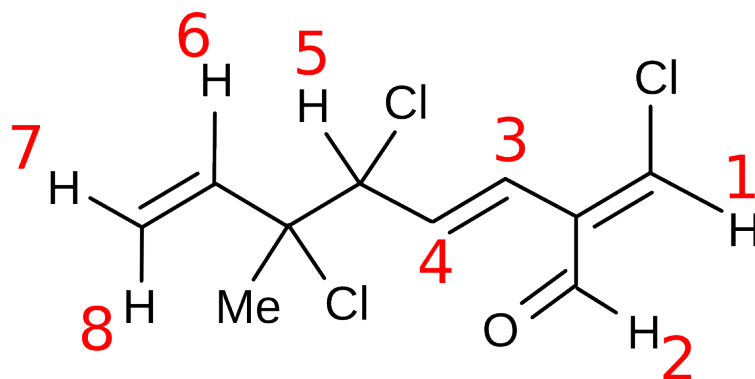
1D ^1H NMR - methyl 2-acetoxy propanoate



1D ^1H NMR - ethyl 4-(methylamino)benzoate



1D ^1H NMR spectrum of cartilagineal



Next session:

1D ^{13}C -NMR spectra