

C8953

NMR structural analysis - seminar

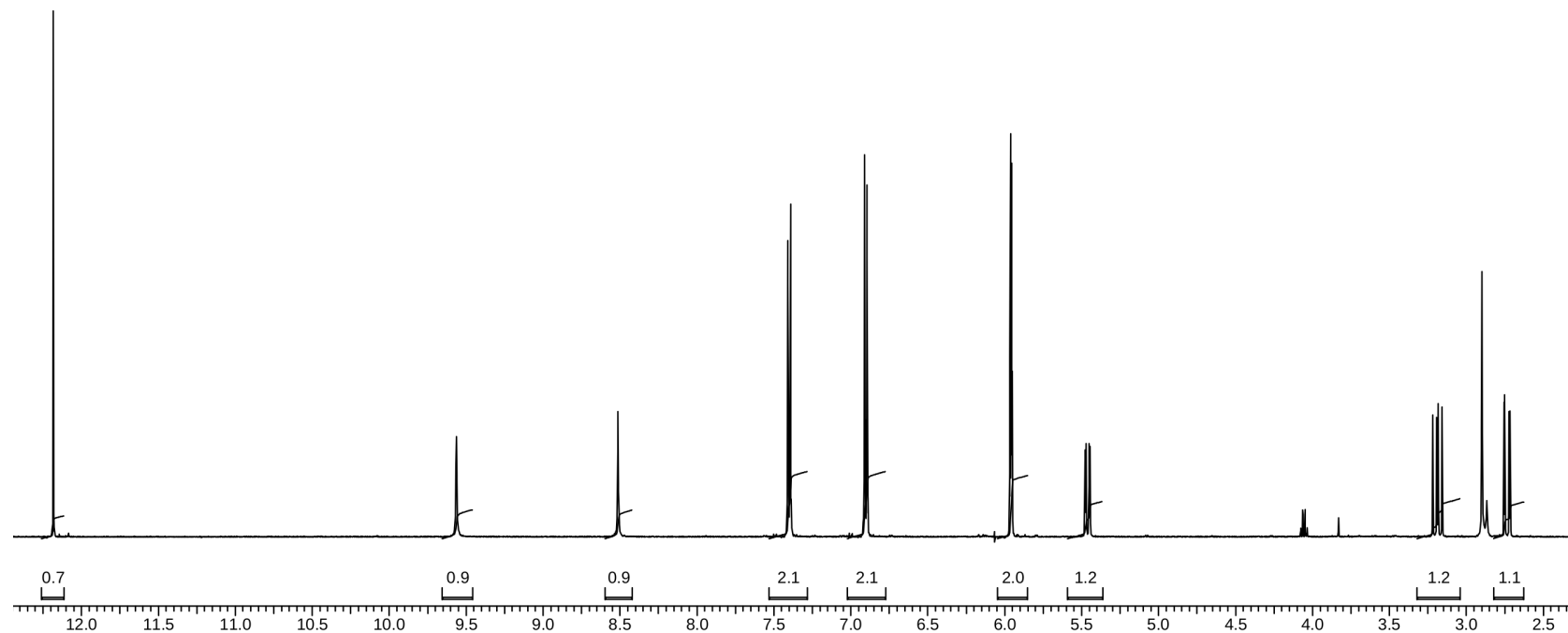
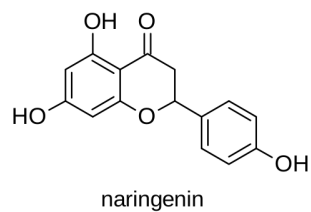
Vector model of NMR experiments + 1D spectra

Jan Novotný

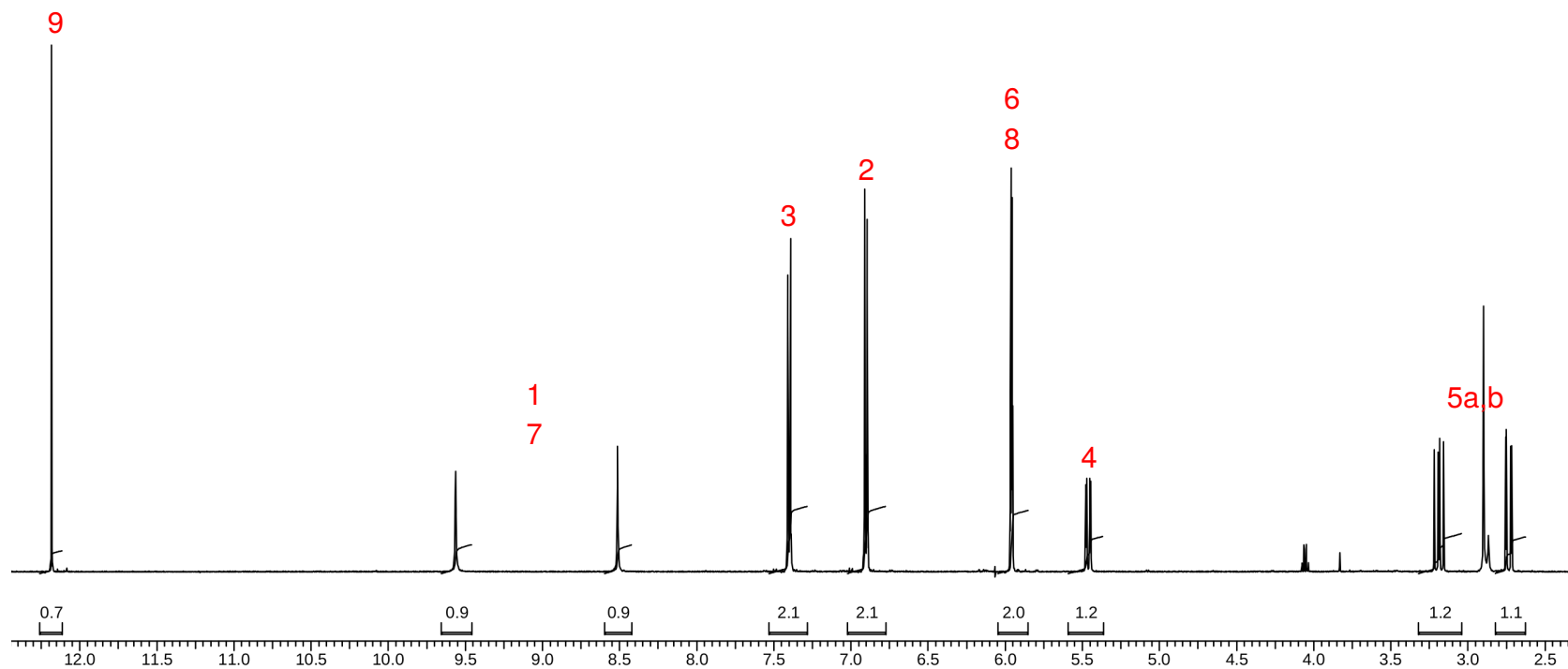
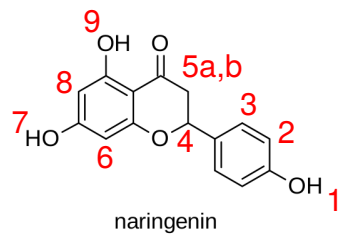
novotnyjan@mail.muni.cz

March 15, 2017

^1H NMR spectrum of naringenin in d_6 -acetone



^1H NMR spectrum of naringenin in d_6 -acetone



Basics of 1D FT spectroscopy

Draw the expected vector model of following simple pulse sequences (right-handed system, B_0 in $+z$ direction:)

▶ 90_{-x} :

▶ $90_{+y} - \Omega\tau = \pi/2$:

▶ $90_{+x} - \Omega\tau = \pi/3 - 180_{+x} - \Omega\tau = \pi/3$:

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▶ $90_{-x} : +y$

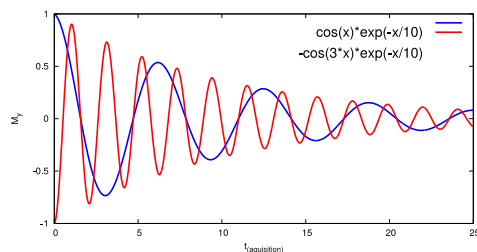
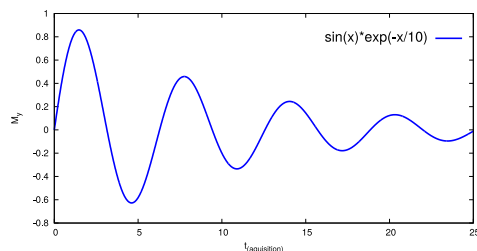
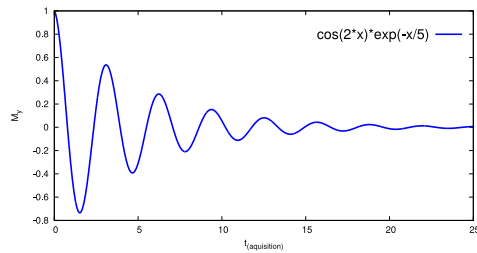
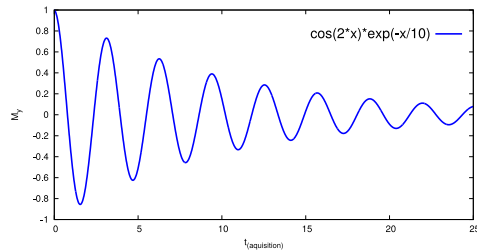
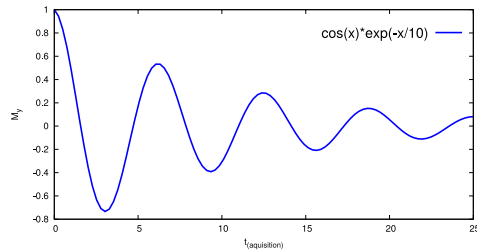
▶ $90_{+y} - \Omega\tau = \pi/2 : +x \rightarrow +y$

▶ $90_{+x} - \Omega\tau = \pi/3 - 180_{+x} - \Omega\tau = \pi/3 :$
 $-y \rightarrow -y * \cos(60) + x * \sin(60) \rightarrow +y * \cos(60) + x * \sin(60) \rightarrow +y$

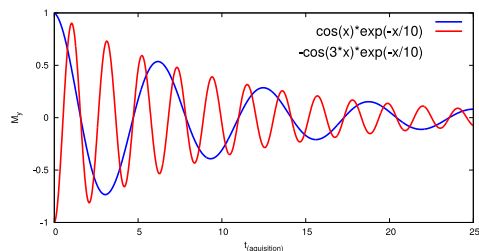
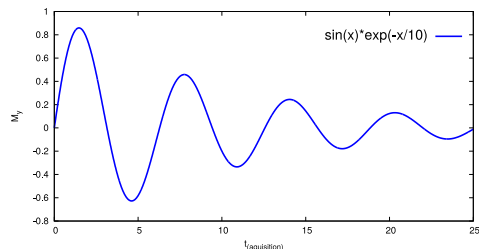
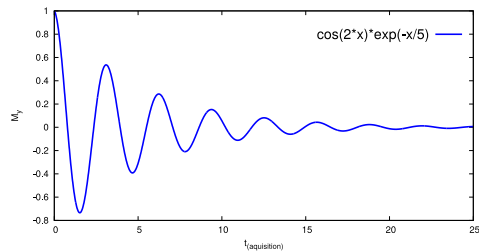
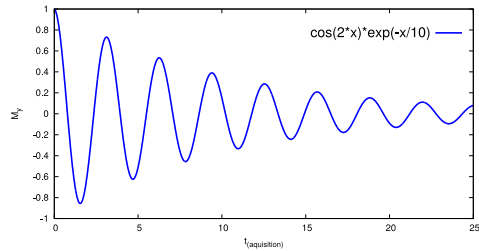
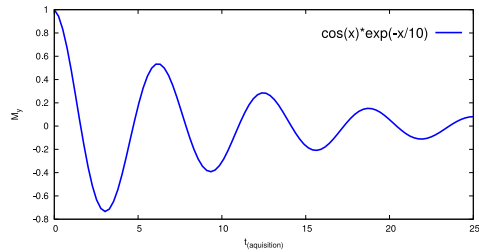
▶ $90_{+x} - \Omega\tau = \pi/3 - 180_{+y} - \Omega\tau = \pi/3 :$
 $-y \rightarrow -y * \cos(60) + x * \sin(60) \rightarrow -y * \cos(60) - x * \sin(60) \rightarrow -y$

Basics of 1D FT spectroscopy

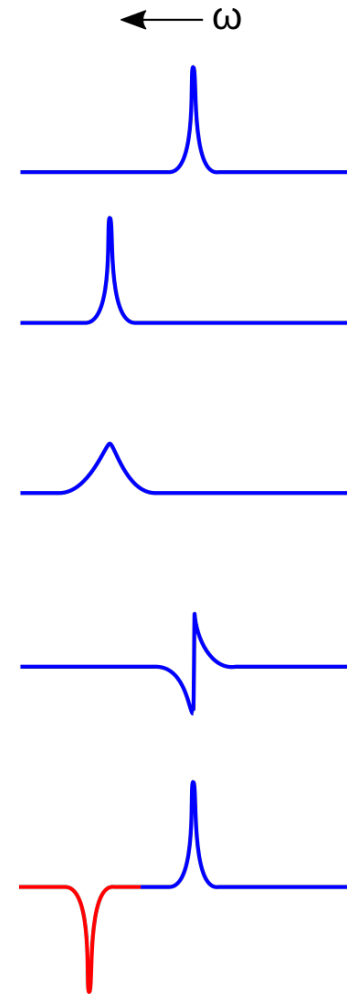
Draw FT representation of attached FID records (reciever is located in the $+y$ direction):



Basics of 1D FT spectroscopy

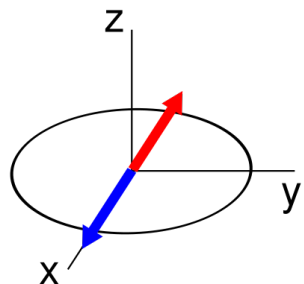
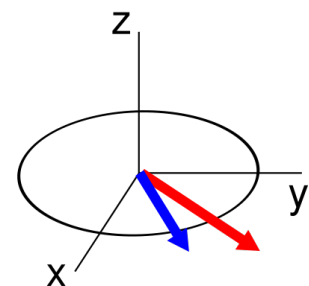
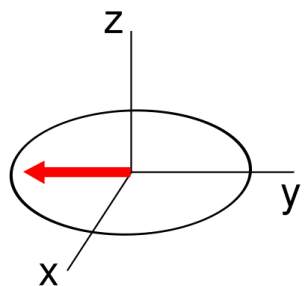
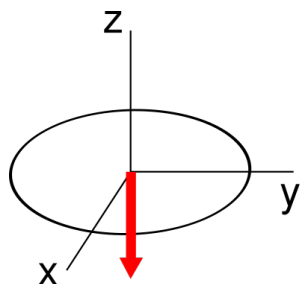


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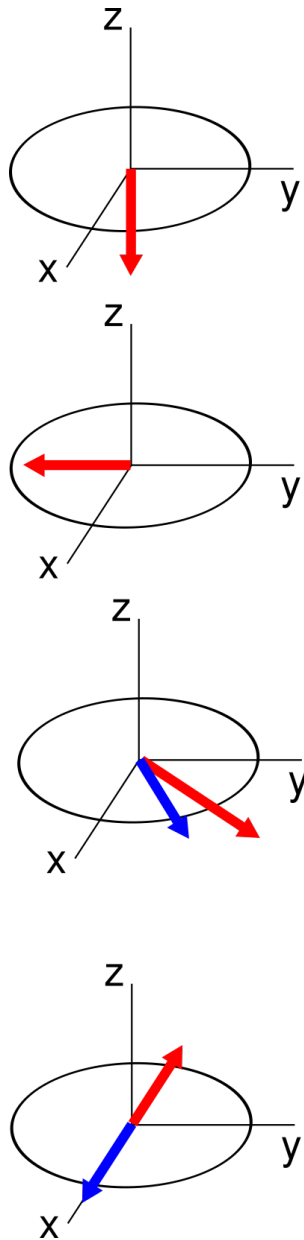


Basics of 1D FT spectroscopy

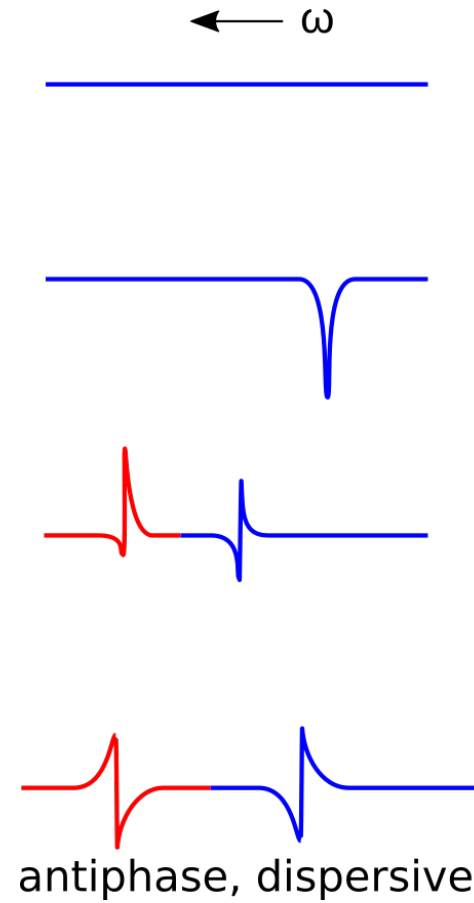
Draw FT representation of vector models
(in rotating frame, receiver is located in the
 $+y$ direction):



Basics of 1D FT spectroscopy

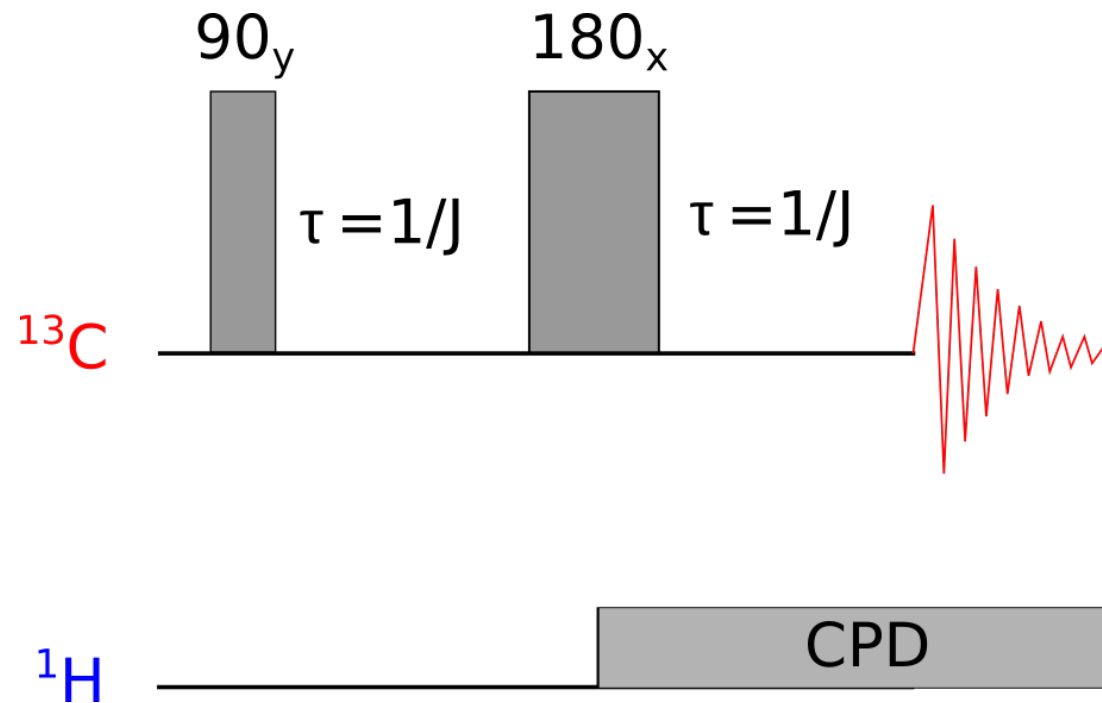


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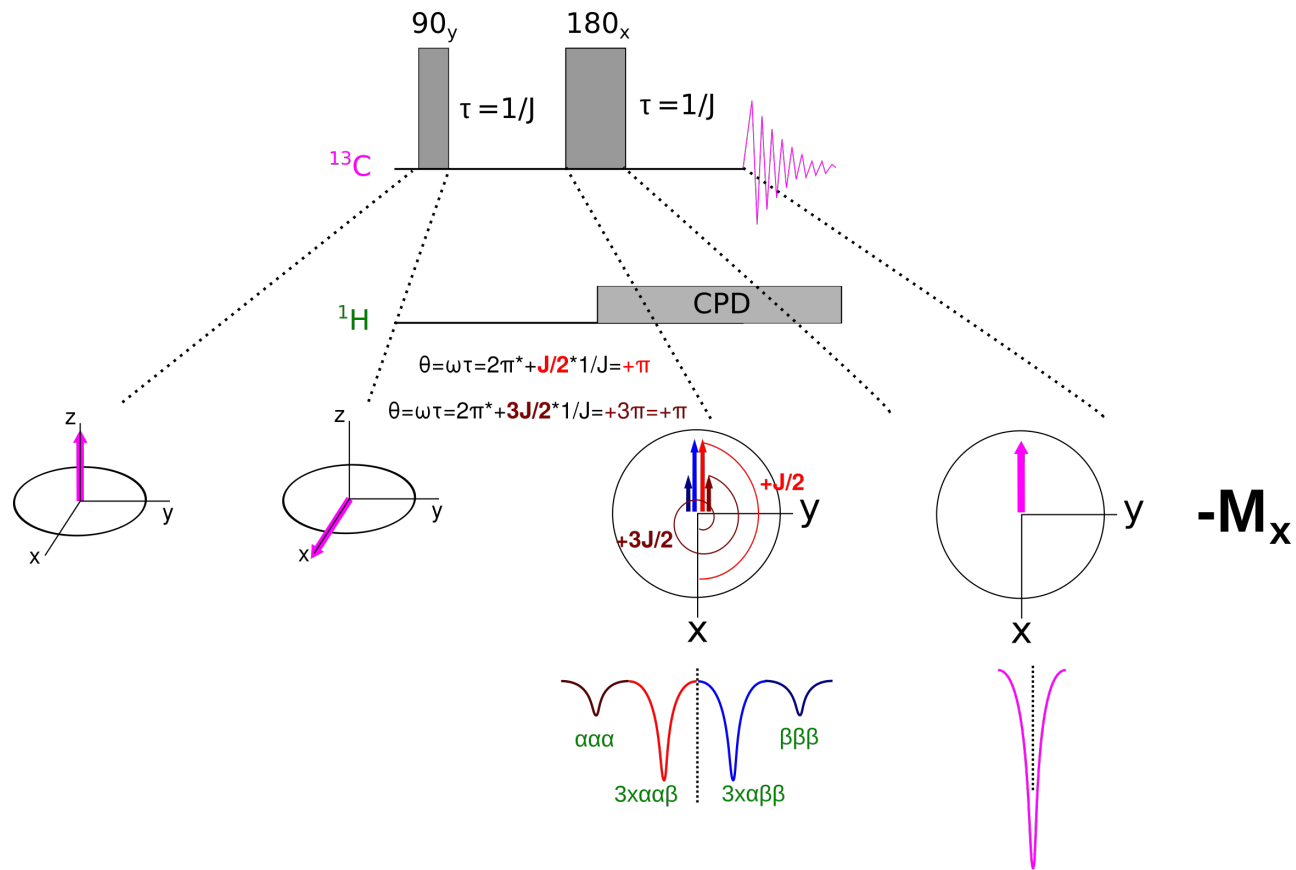
Heteronuclear spin echo of ^{13}C - $^1\text{H}_3$ group

By using vector diagrams determine the result of attached pulse sequence. First realize what is the evolution of ^{13}C signal resulting from offset? CPD=composite pulse decoupling



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APT - Attached Proton Test

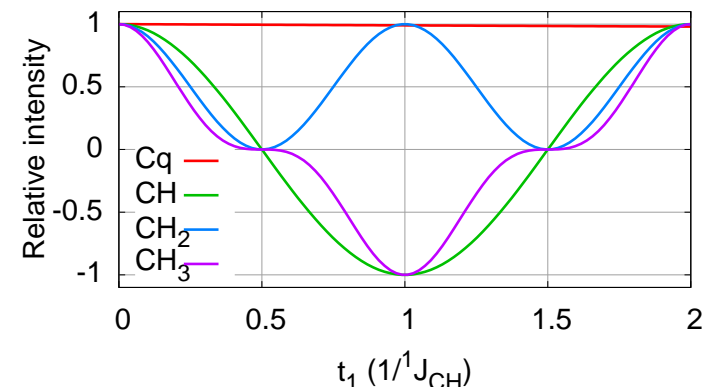
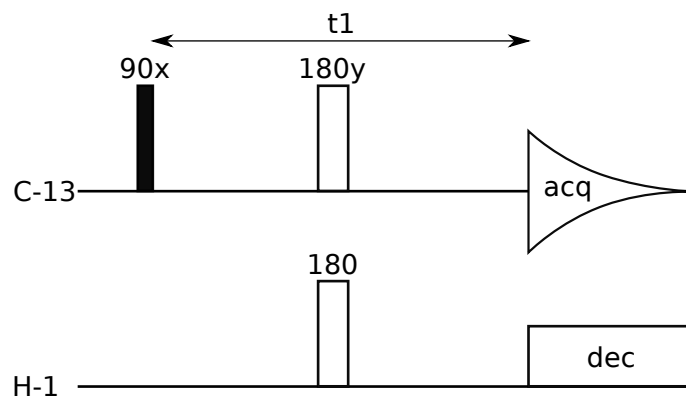
based on heteronuclear spin echo

▶ $t_1 = 1/{}^1J_{CH}$

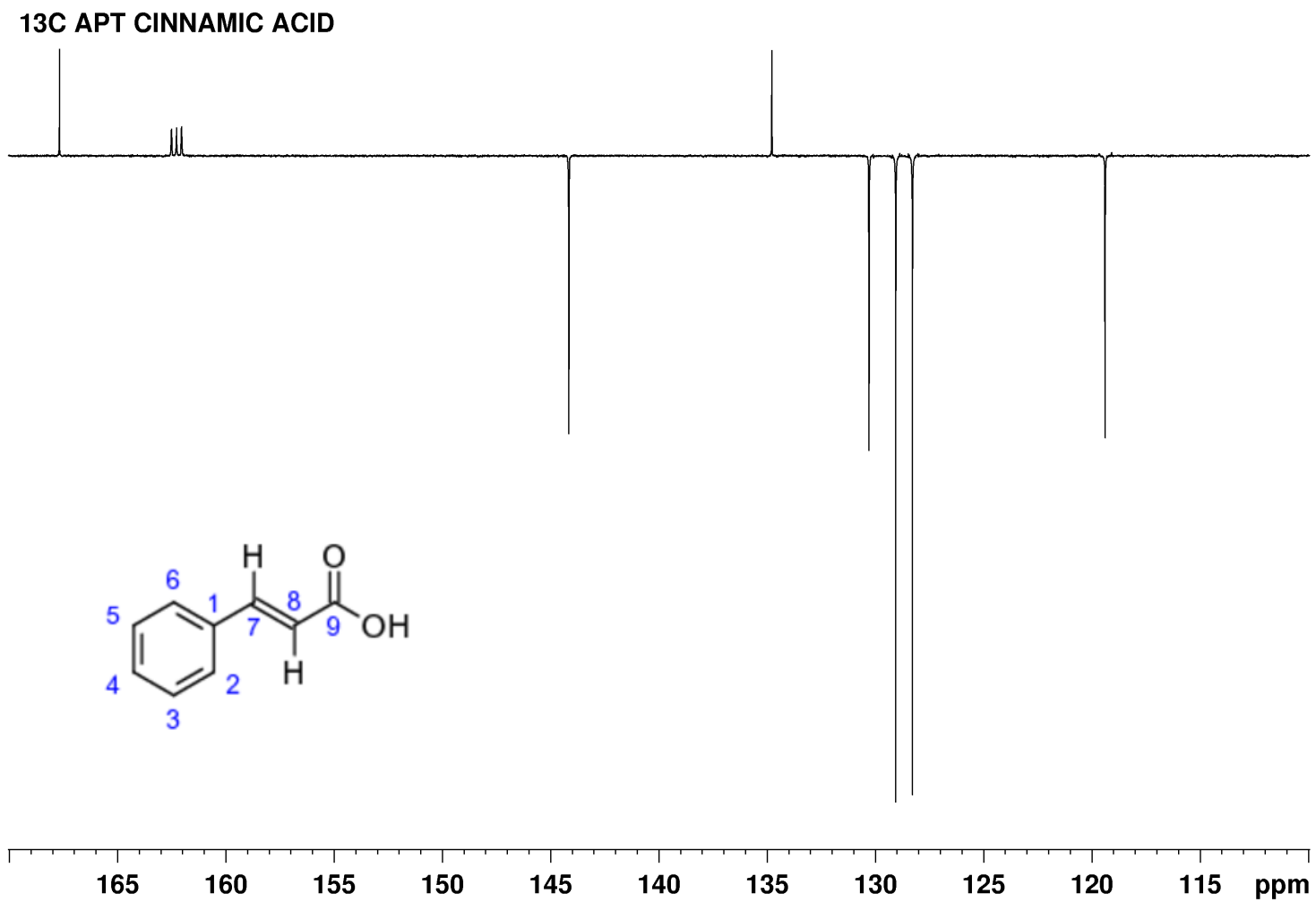
^{13}C signals are differentiated according to the number of directly bound 1H

- ▶ Cq, CH₂ positive
- ▶ CH, CH₃ negative

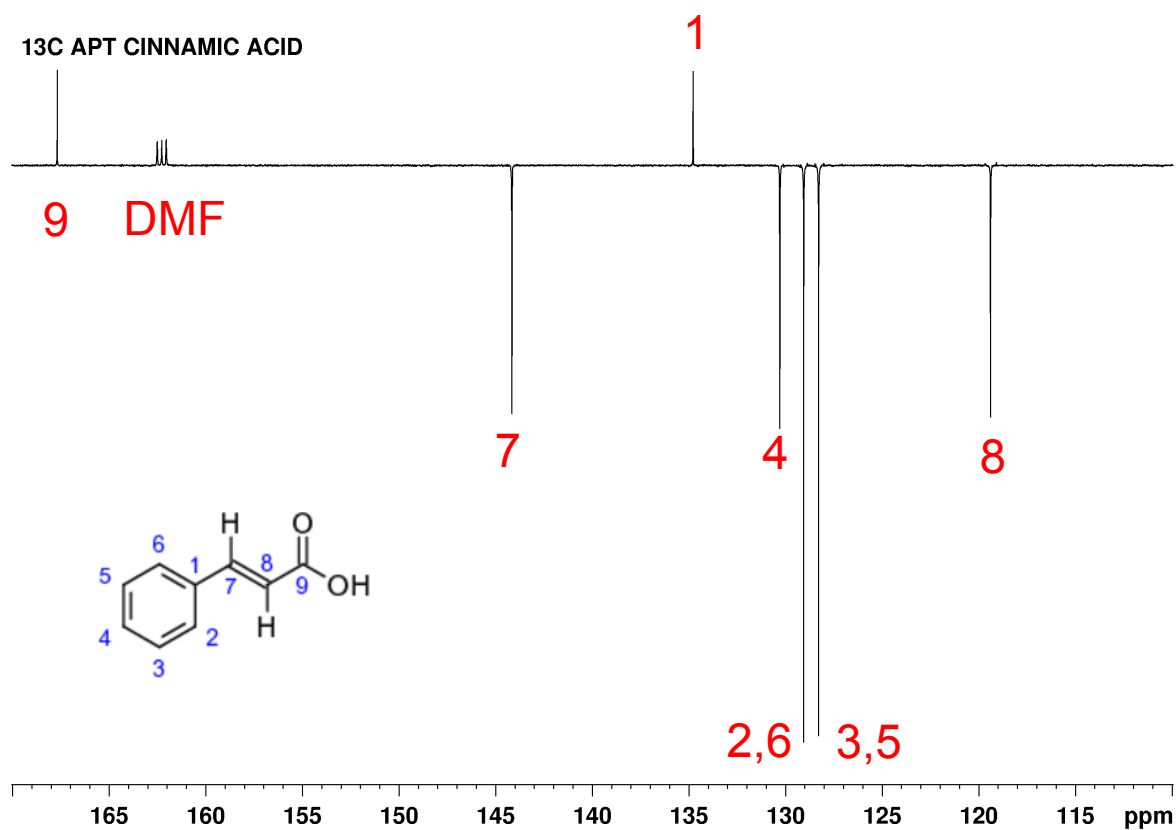
Evolution of signal governed by the value of ${}^1J_{CH} \implies$ reflected by the intensity of APT signal



^{13}C APT Cinnamic acid



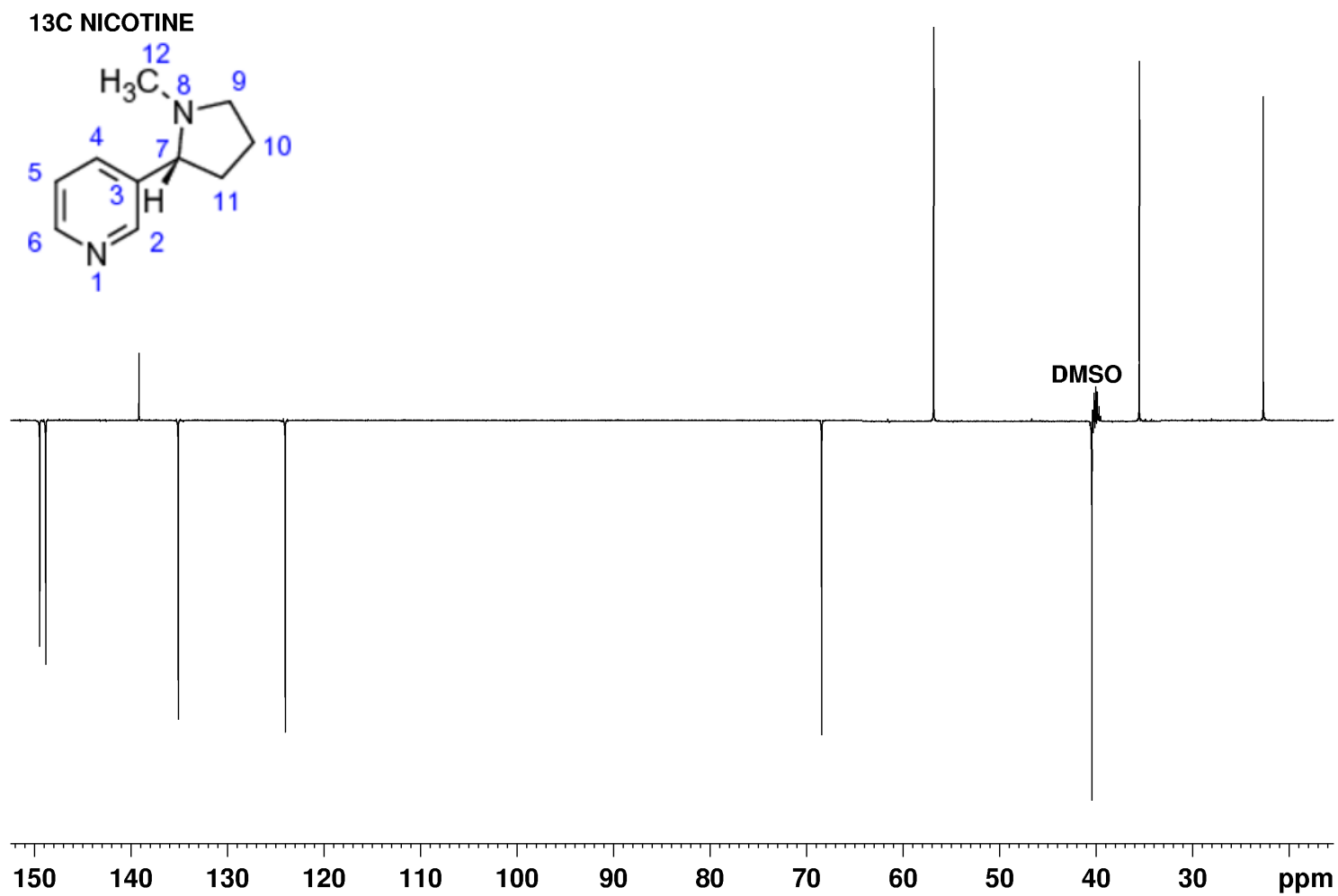
^{13}C APT Cinnamic acid



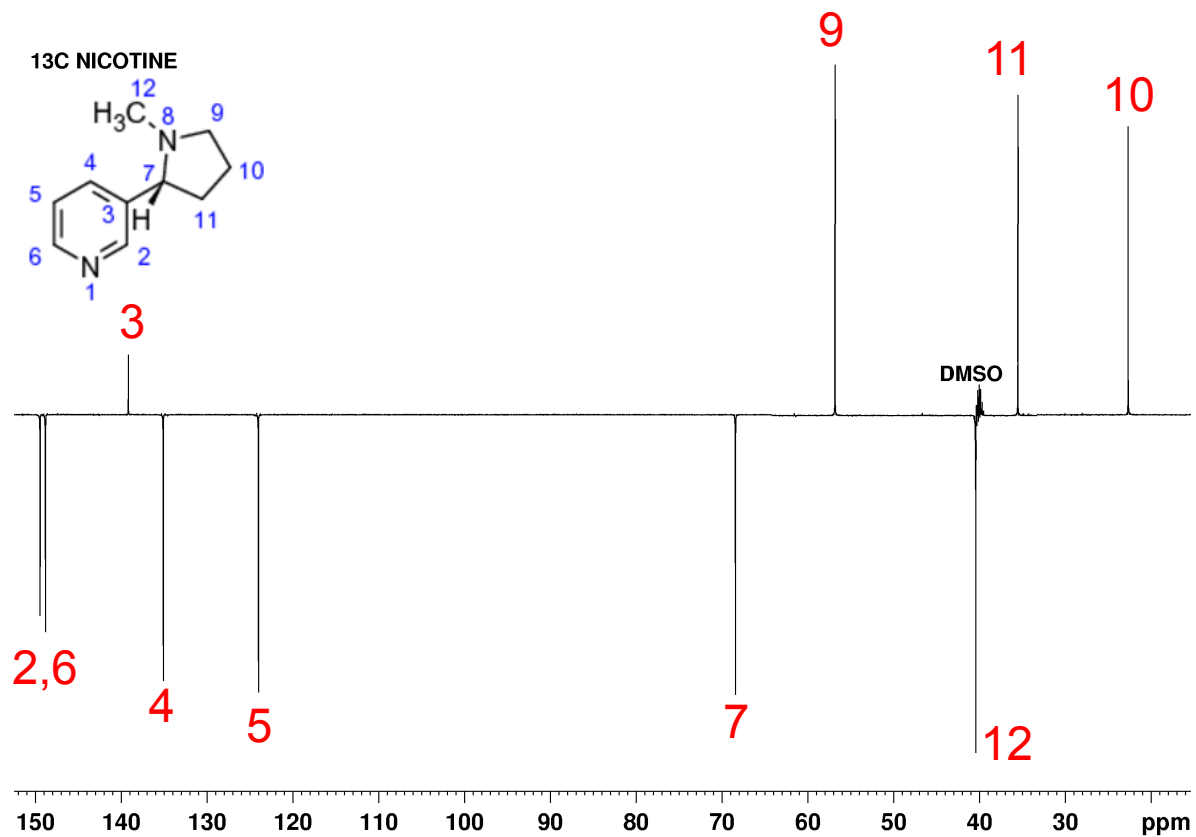
Notes:

- ▶ **C9, C1** positive quaternary
- ▶ **C7** deshielded by -M effect of carboxyl group + in neighbourhood of aromatic system
- ▶ equivalent **C2/6, C3/5** in aromatic region, para **C4** less sensitive

^{13}C APT of Nicotine



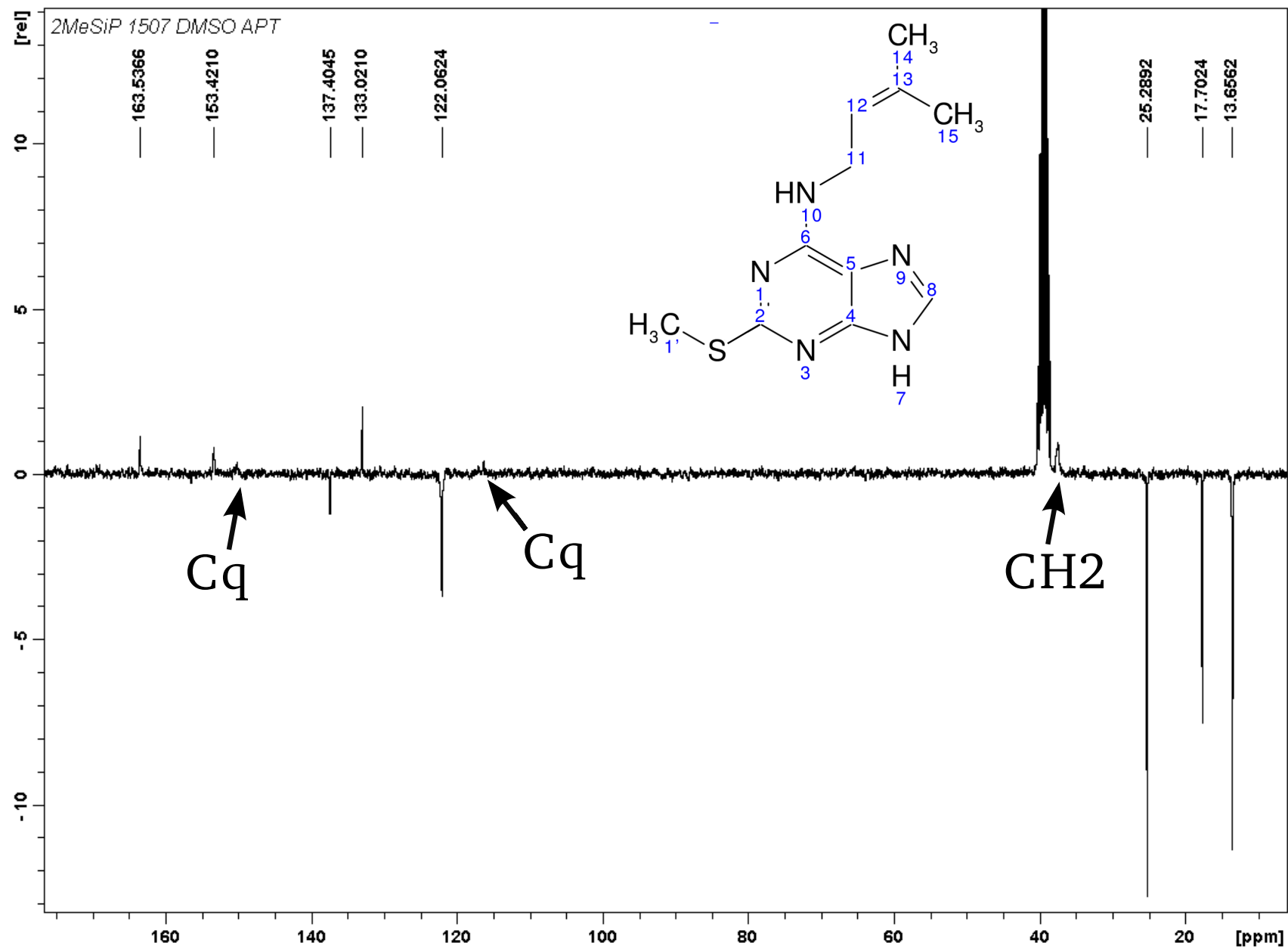
^{13}C APT of Nicotine



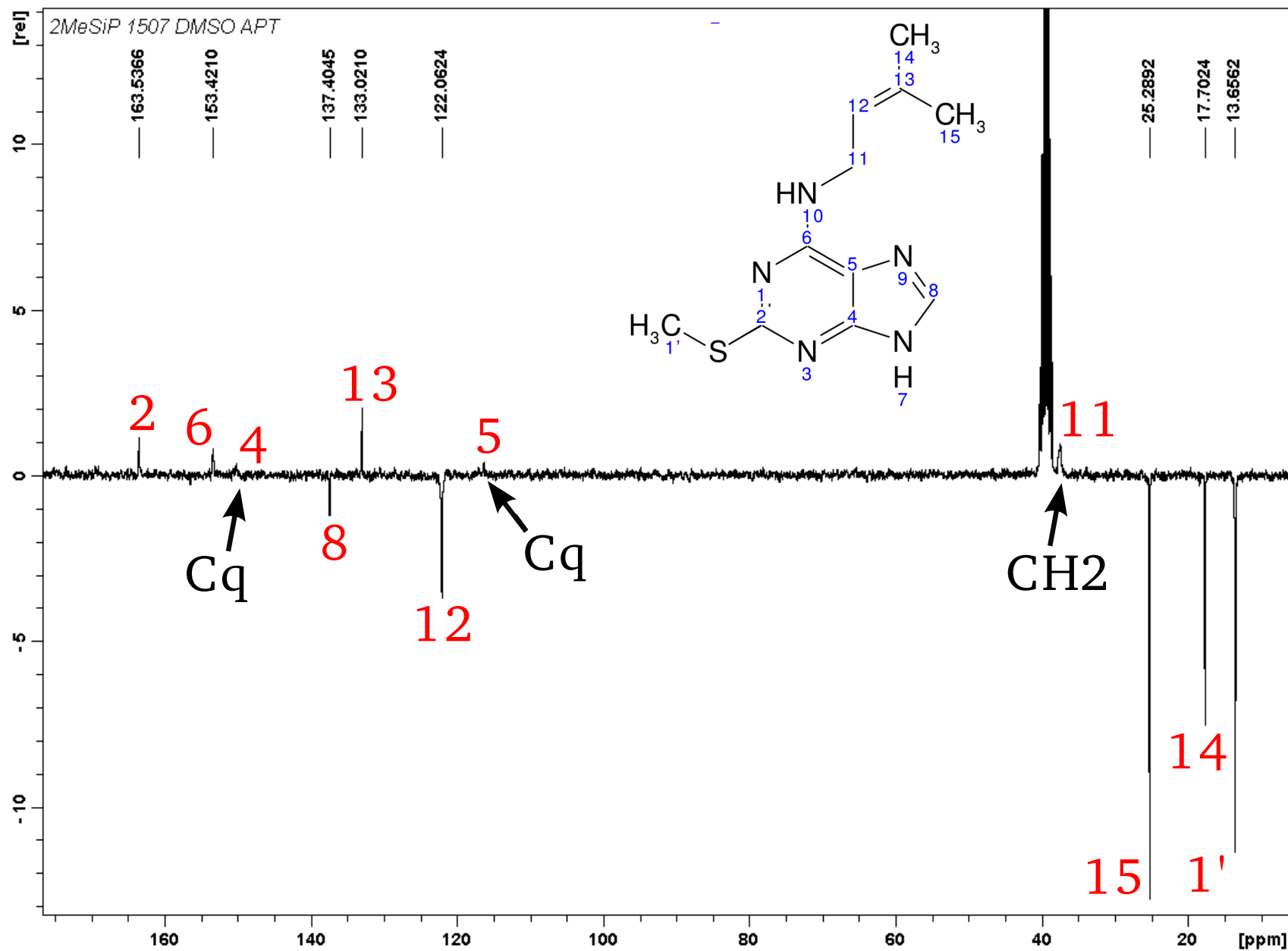
Notes:

- ▶ **C2, C6** CH negative connected to N
- ▶ **C3** quaternary, **C4** more deshielded
- ▶ **C7** tertiary carbon, in neighbourhood of aromatic system and N
- ▶ **C9** secondary, close to N; **C12** primary attached to N
- ▶ **C11** connected to tertiary carbon

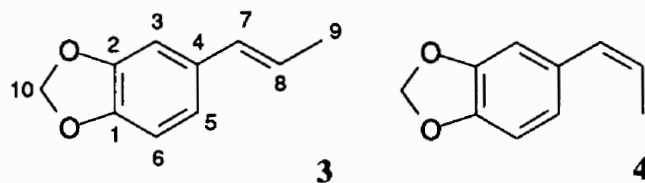
^{13}C APT 4



¹³C APT 4



^1H - ^{13}C DEPT spectrum



Which is the major product? Assign the signals as far as possible. Why does the signal at $\delta = 100.8$ exist in the spectrum 3.3.c, although its intensity should be zero?

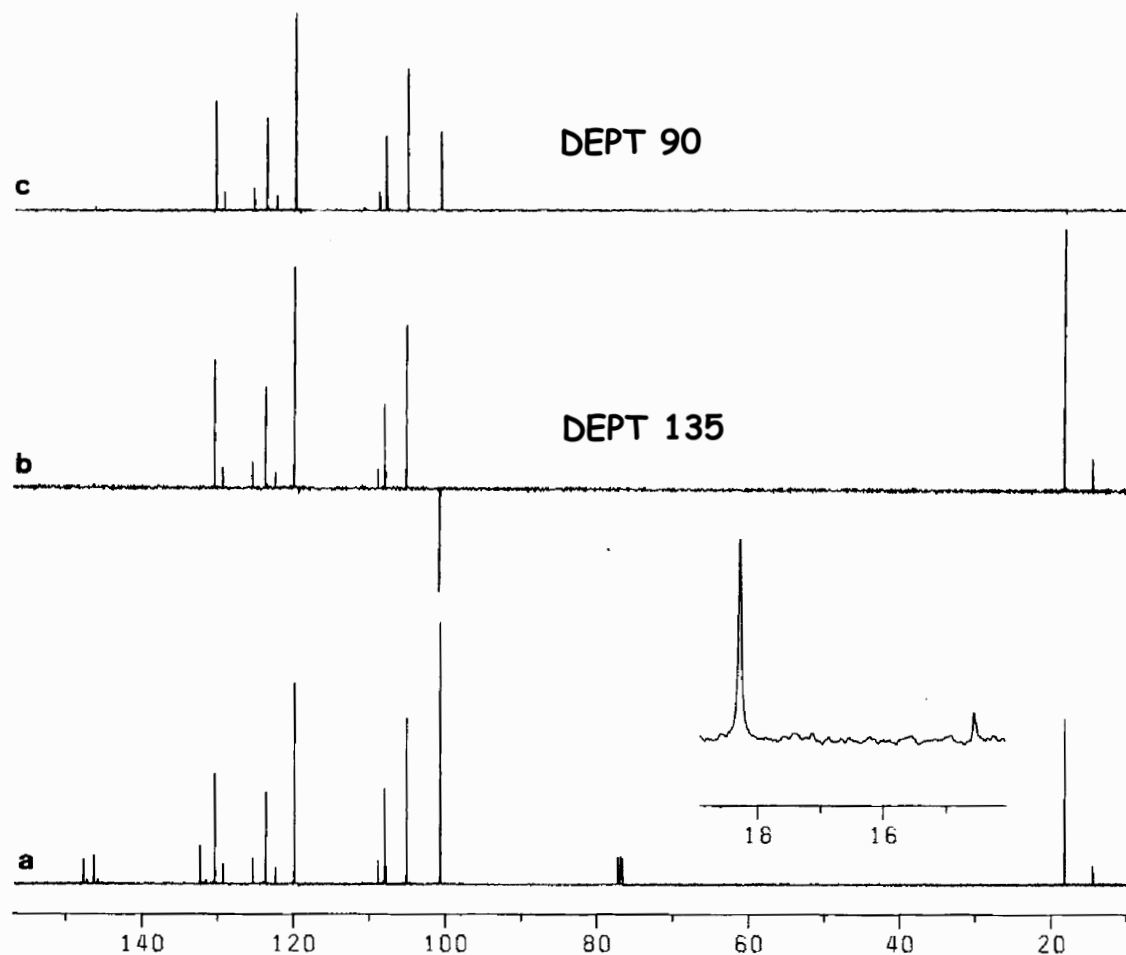
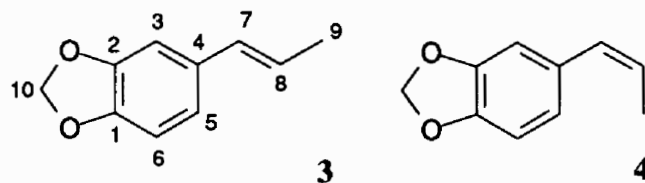


Fig. 3.3. (a) ^1H broad-band decoupled ^{13}C NMR spectrum of a mixture of **3** and **4** in CDCl_3 . Traces (b) and (c) are DEPT spectra.

^1H - ^{13}C DEPT spectrum



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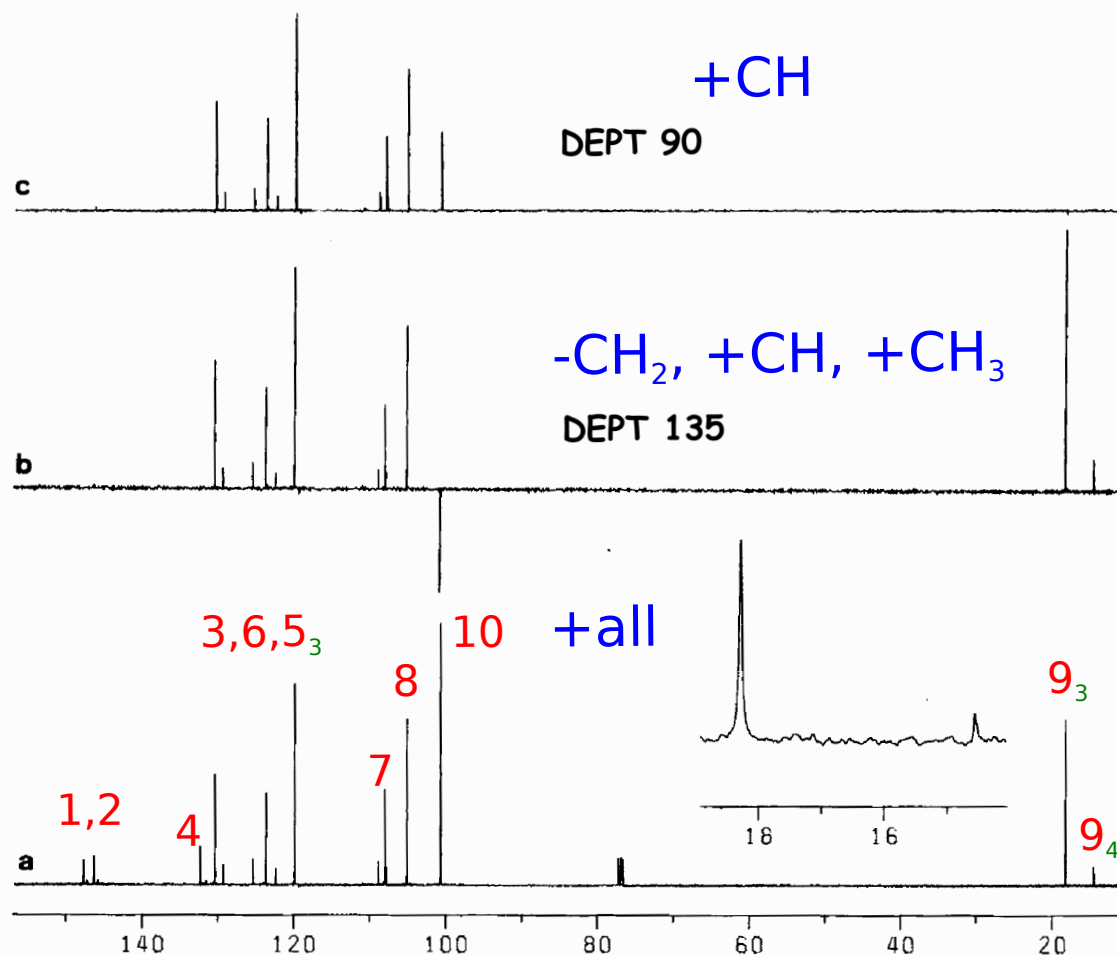


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Next topic

2D spectroscopy