

# Organosilicon Plasma Polymers Prepared On Electrospun Polymer Nanofibers

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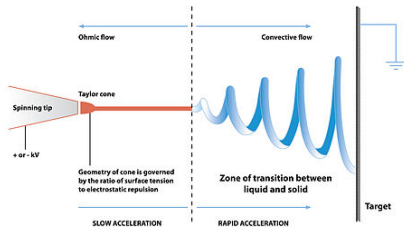


# Outline

- Electrospinning of Polymer Fibers
- PECVD of Organosilicon Films
- Plasma Sources Used
- Low Pressure CCP
- Atmospheric Pressure Plasma Jet
- Summary

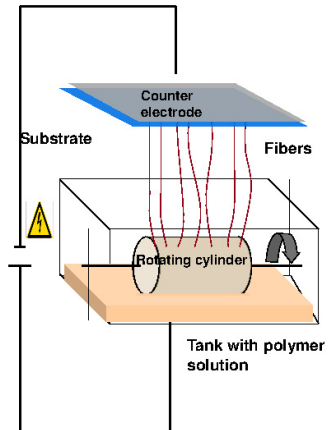
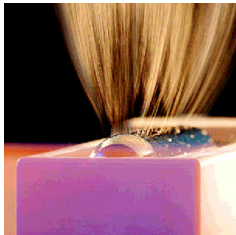
# Principle of Electrospinning

Electrospinning of polymer fibers:



- ▶ High DC voltage between electrodes
- ▶ If the electrostatic repelling force overcomes the surface tension of polymer solution or melt, the polymer formed a Taylor cone and is ejected from the surface of electrode and forms a continuous filament.

Nanospider™ technology developed by ELMARCO



# Electrospinning of PVA and PA6

PVA



PA6



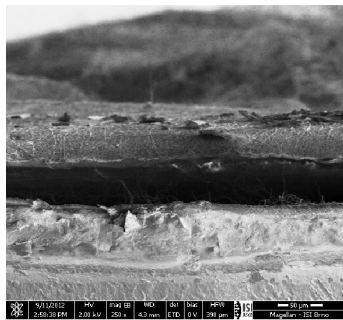
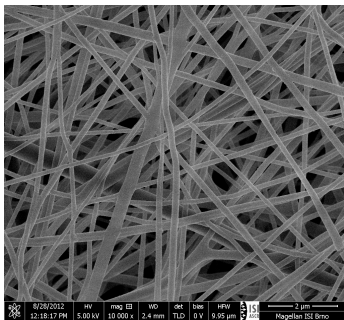
## Polyvinylalcohol (PVA)

- ▶ cylindrical electrode,
- ▶ solvent - water,
- ▶  $U = 64 - 66 \text{ kV}$  (unstable),
- ▶  $I = 0.05 \text{ A}$ ,
- ▶  $v = 12 \text{ mm/min}$ , spin 3 r/min

## Polyamide (PA6)

- ▶ wired electrode,
- ▶ solvent - acetic acid/formic acid (problems of solvent volatility),
- ▶  $U = 64 \text{ kV}$
- ▶  $I = 0.04 \text{ A}$ ,
- ▶  $v = 12 \text{ mm/min}$ , spin 2 r/min

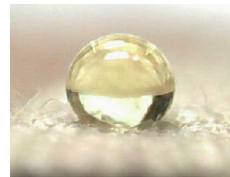
## Electrospun polymers - highly porous network of micro/nanofibers



Plasma coating of electrospun polymer micro/nanofibers can bring additional functionality for

- ▶ smart textiles
- ▶ filtration of liquids/gases
- ▶ tissue engineering
- ▶ battery separators

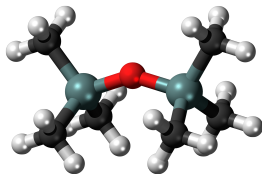
Example of functional coating on complex substrate - hydrophobic/oleophobic surfaces of plasma treated Nomex-based textiles:



M. Klíma *et al.* - cold atmospheric pressure plasma jet

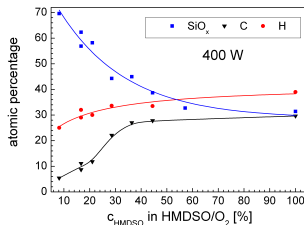
# Organosilicon Plasma Polymers and SiO<sub>x</sub> Films

Hybrid character of monomers:  
source of Si-O and C-H groups



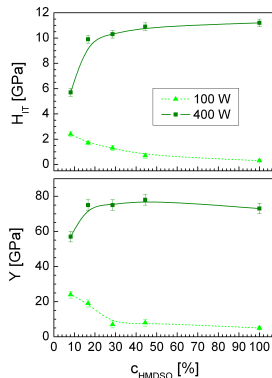
hexamethyldisiloxane (HMDSO)  
Si<sub>2</sub>OC<sub>6</sub>H<sub>12</sub>

Example for  
low pressure CCP  
in HMDSO/O<sub>2</sub>



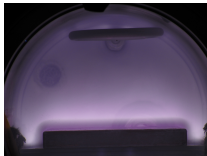
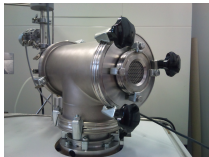
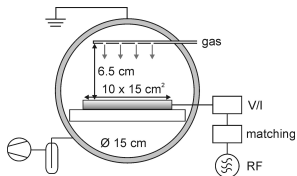
Functional coatings:

- ▶ transparent protective films on plastics
- ▶ (ultra)hydrophobic / (ultra)hydrophilic coatings
- ▶ intermediate adhesive layers on metals

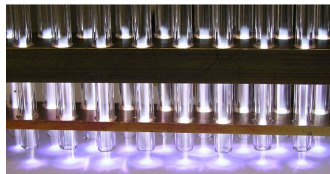
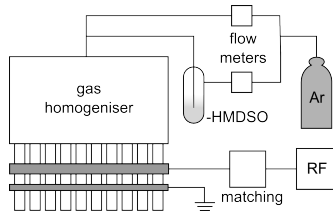


# Low and Atmospheric Pressure Discharges

- ▶ RF capacitively coupled discharge (13.56 MHz) at EMPA



- ▶ cold atmospheric pressure RF plasma multijets (13.56 MHz) at MU Brno

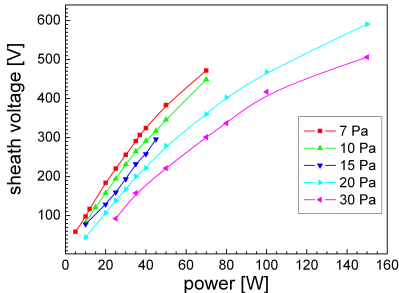


# Low Pressure CCP - Variation of Deposition Parameters

- ▶ Ar + HMDSO 1:1 mixture
- ▶ total flow rate 6 sccm
- ▶ pressure  $p$  7–30 Pa
- ▶ power  $W$  5–150 W
- ▶ dc bias-voltage at RF (substrate) electrode  $U_b$  varied with  $W$  and  $p$

Sheath voltage at substrate electrode

$$V_{sh} = 0.39V_0 + 0.73U_b$$



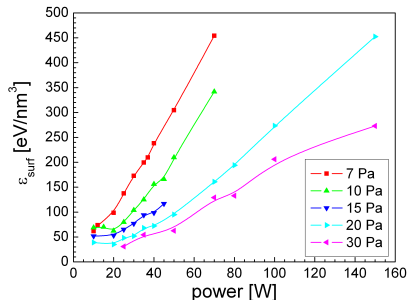
- ▶ Gas phase processes
  - energy input  $W/F|_{dep}$
- ▶ Effect of ion bombardment:

Energy dissipated per deposition rate  $R$

$$\epsilon_{surf} = \frac{\Gamma_i E_{mean}}{R} \quad (1)$$

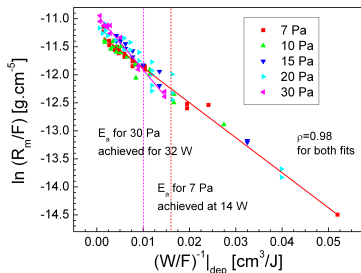
$\Gamma_i$  ion flux,  $E_{mean}$  mean ion energy

*D. Hegemann et al. Appl. Phys. Lett. 101 (2012) 211603*





# Understanding Deposition Process

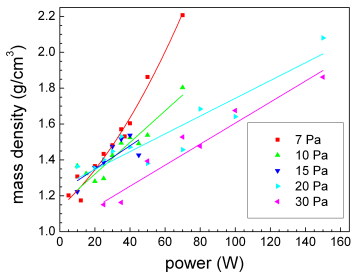
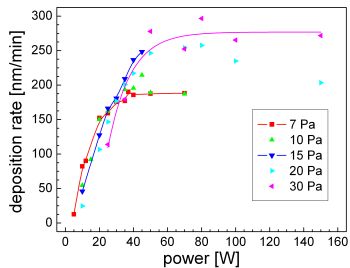


$$\frac{R_m}{F} = G \exp\left(-\frac{E_a}{W/F|_{\text{dep}}}\right) \quad (2)$$

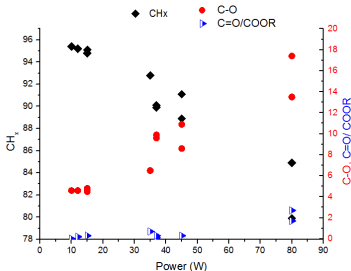
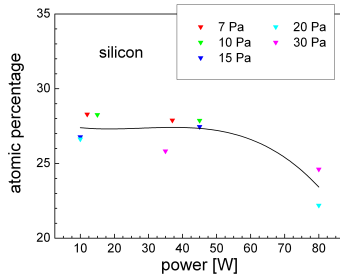
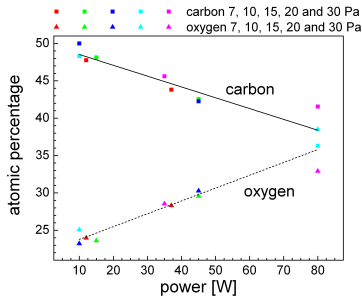
$$\frac{W}{F}|_{\text{dep}} = \frac{W}{F} \frac{d_{\text{act}}}{d_{\text{gas}}} \quad (3)$$

mass deposition rate  $R_m$ , apparent activation energy  $E_a$ , absorbed power density  $W$ , gas flow  $F$

*D. Hegemann et al. Plasma Process Polym 7 (2010) 889*

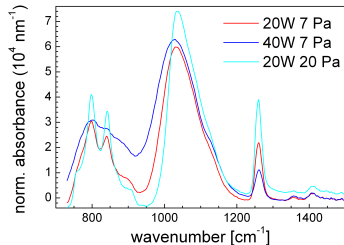
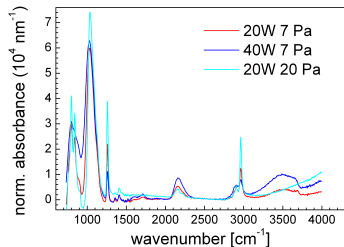


# Chemical Structure of Films by XPS



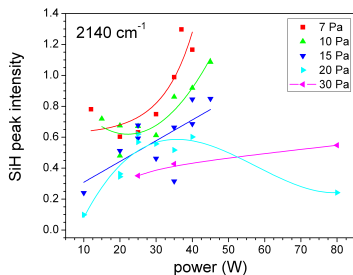
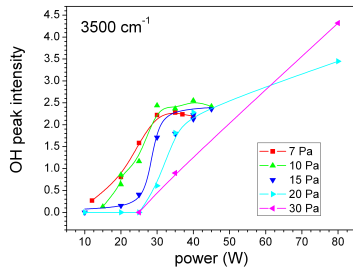
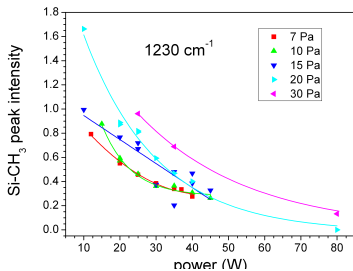
- ▶ no obvious difference for different pressures for few selected samples
- ▶ oxidation of products from Ar/HMDSO mixture with increased power, i.e. dissociation of monomer
- ▶ occurrence of C=O, COOR due to oxidation of CH groups and incorporation of the products into films

# Chemical Structure of Films by FTIR



$\sigma$ (cm <sup>-1</sup> )	Mode	Comment
<i>CH<sub>x</sub> correlated peaks</i>		
2960	$\nu_{\text{CH}_3}^{\text{a}}$	
2900	$\nu_{\text{CH}_3}^{\text{s}}$	
2925	$\nu_{\text{CH}_2}^{\text{a}}$	
2855	$\nu_{\text{CH}_2}^{\text{s}}$	
1460	$\delta_{\text{CH}_2}^{\text{a}}$	
<i>Si – CH<sub>2</sub> correlated peaks</i>		
1360	$\delta_{\text{CH}_2}$	in Si-CH <sub>2</sub> -Si
1400	$\delta_{\text{CH}_2}$	in Si-(CH <sub>2</sub> ) <sub>2</sub> -Si
<i>Si – CH<sub>3</sub> correlated peaks</i>		
1410	$\delta_{\text{CH}_3}^{\text{a}}$	in Si-Me <sub>x</sub>
1260	$\delta_{\text{CH}_3}^{\text{s}}$	in Si-Me <sub>x</sub>
845	$\rho_{\text{CH}_3}, \nu_{\text{SiC}}$	in Si-Me <sub>3</sub>
760	$\rho_{\text{CH}_3}, \nu_{\text{SiC}}$	in Si-Me <sub>3</sub>
885	$\rho_{\text{CH}_3}, \nu_{\text{SiC}}$	in Si-Me <sub>2</sub>
805	$\rho_{\text{CH}_3}, \nu_{\text{SiC}}$	in Si-Me <sub>2</sub>
775	$\rho_{\text{CH}_3}, \nu_{\text{SiC}}$	in Si-Me <sub>1</sub>
<i>Si – H correlated peaks</i>		
2140	$\nu_{\text{SiH}}$	

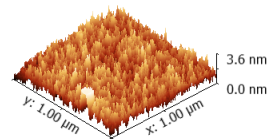
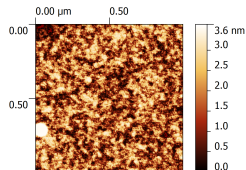
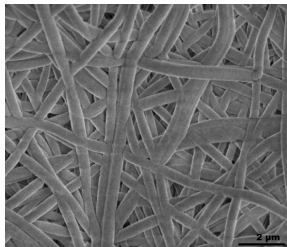
# Chemical Structure of Films by FTIR



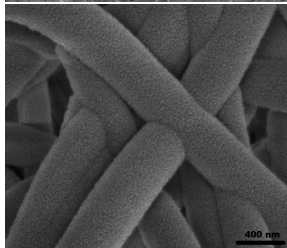
- ▶ higher retention of monomer structure (Si-CH<sub>3</sub> groups) at lower power and higher pressure
- ▶ general trends are in agreement with XPS
- ▶ more details for pressure changes (due to higher amount of samples?)

# Smooth Films by Low Pressure PECVD

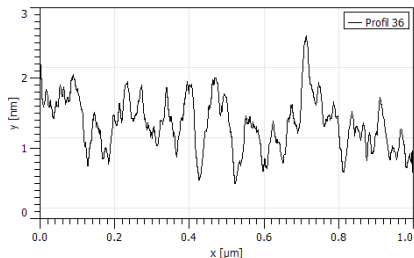
## Typical example for 20Pa and 100W



Rms roughness on Si - 0.50 nm

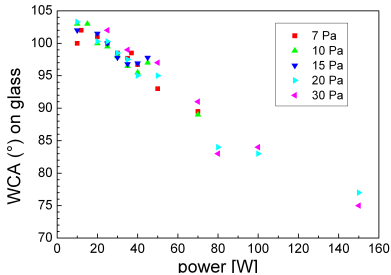


with 5 nm of sputtered Au film



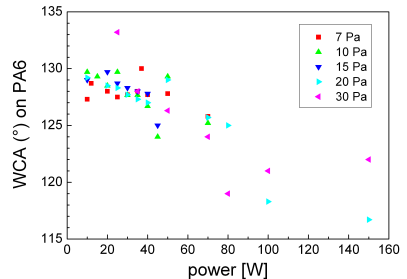
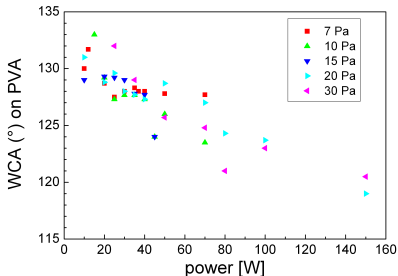
Film thickness on Si substrate 493 nm, on PVA  $140 \pm 60$  nm  
(PVA diameter  $210 \pm 40$  nm).

# Water Contact Angle



## ► hydrophobic films

- decrease of water contact angle due to increased oxidation
- nanostructure of electrospun fibers causes an increase of CA
- fibers are well protected by the films because water CA could be measured on PVA (normally dissolved in water)

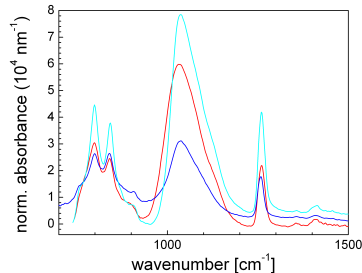
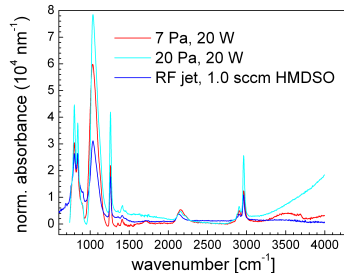


## Films Deposited by RF Jets

- ▶ atmospheric pressures,
- ▶ 20 moving nozzles,
- ▶ 10 W per nozzle,
- ▶ 3.2 slm of Ar,
- ▶ 0.8–1.2 sccm of HMDSO per nozzle (Ar bubbler),
- ▶ speed of sample movement - 60 cm/min,
- ▶ 4 or 10 passages over substrate

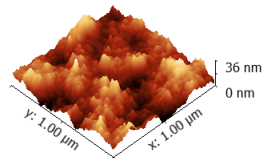
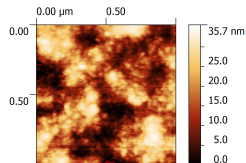
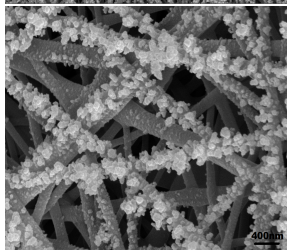
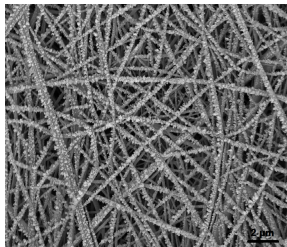
water contact angle on coated nanofibers:  
141 – 149 °

- ▶ higher value than 130° for low pressure although the film chemistry is similar

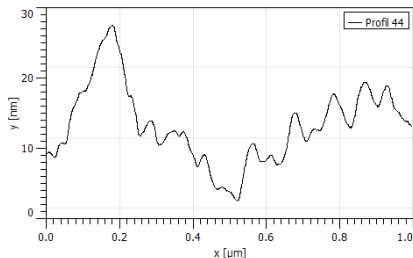


# Nanostructured Organosilicon Films

## Typical example for 0.8 sccm HMDSO and 4 passages



Rms roughness on Si - 5.15 nm



with 10 nm of sputtered Au film

Maximum film thickness on Si 280 nm.

Thickness of coated PVA nanofibers  $210 \pm 40$  nm (PVA diameter  $210 \pm 40$  nm)



# Summary

- ▶ PVA and PA6 nanofibrous textiles **can be plasma-coated in Ar/HMDSO without their damage using both**
  - ▶ low pressure CCP discharge, fibers on RF electrode (ion bombardment)
  - ▶ atmospheric pressure plasma jet (thermal load)
- ▶ **Organosilicon coatings protected PVA fibers against water** even for the nanostructured films made by RF multijets.
- ▶ **Variable surface structure** achieved due to **different discharges** used
  - ▶ low pressure coatings - very low roughness (Rms around 0.5 nm)
  - ▶ atmospheric pressure coatings - Rms of 5 nm on flat Si, nanoparticles on polymer fibers.
- ▶ **Variation of water contact angle** achieved by **surface chemistry and morphology**
  - ▶ low pressure coatings - shift from about  $102^\circ$  to  $75^\circ$  (on flat substrate) due to decrease of  $-\text{CH}_3$  groups and increase of OH
  - ▶ WCA on low pressure plasma-coated nanofibers increased to  $130\text{--}115^\circ$
  - ▶  $141\text{--}149^\circ$  was achieved for nanostructured coatings on nanofibers deposited by RF plasma multijets.

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## Our webpage:

<http://www.ceitec.eu/programs/advanced-nano-and-microtechnologies/plasma-technologies>