

C8953

NMR structural analysis - seminar

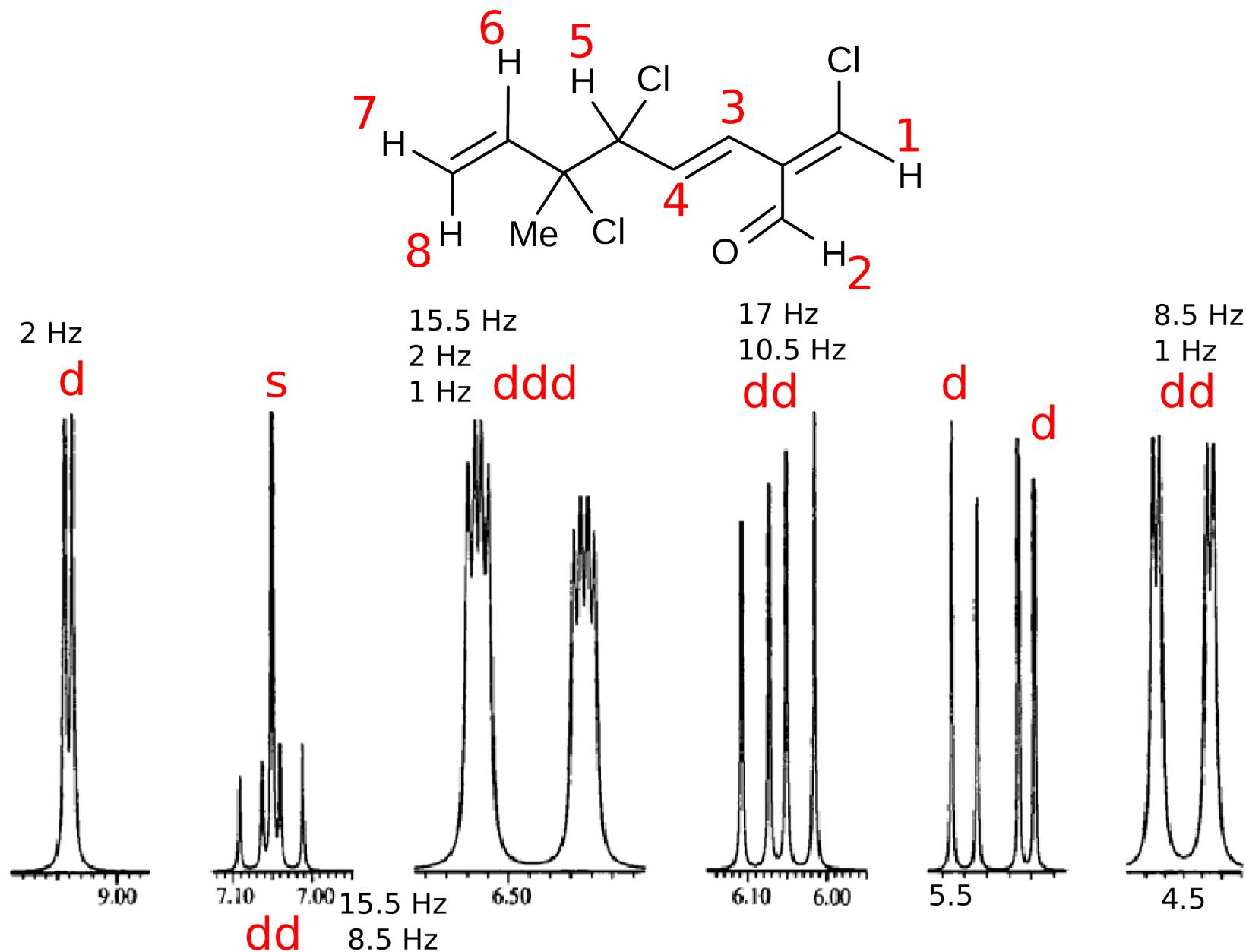
1D ^{13}C -NMR

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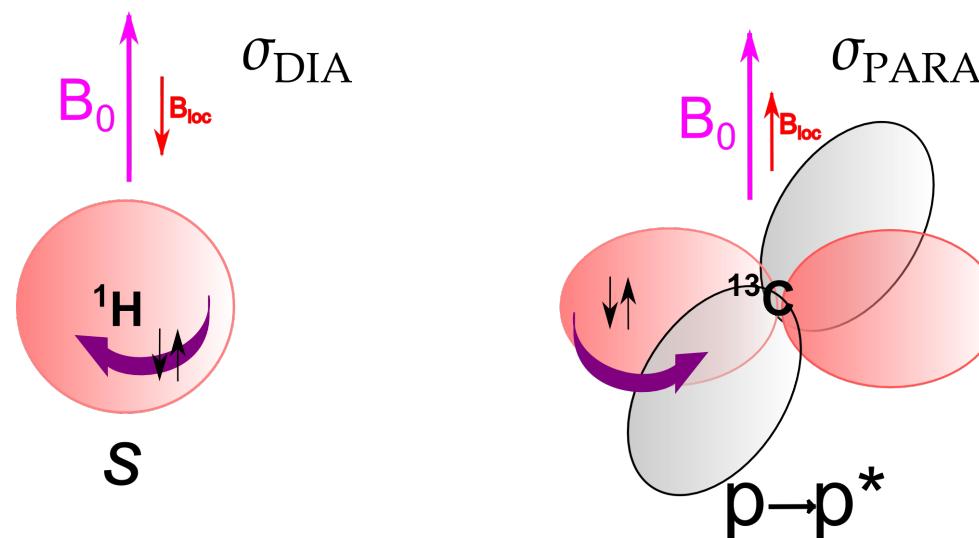
March 13, 2019

Control of 1D ^1H NMR spectrum of cartilagineal



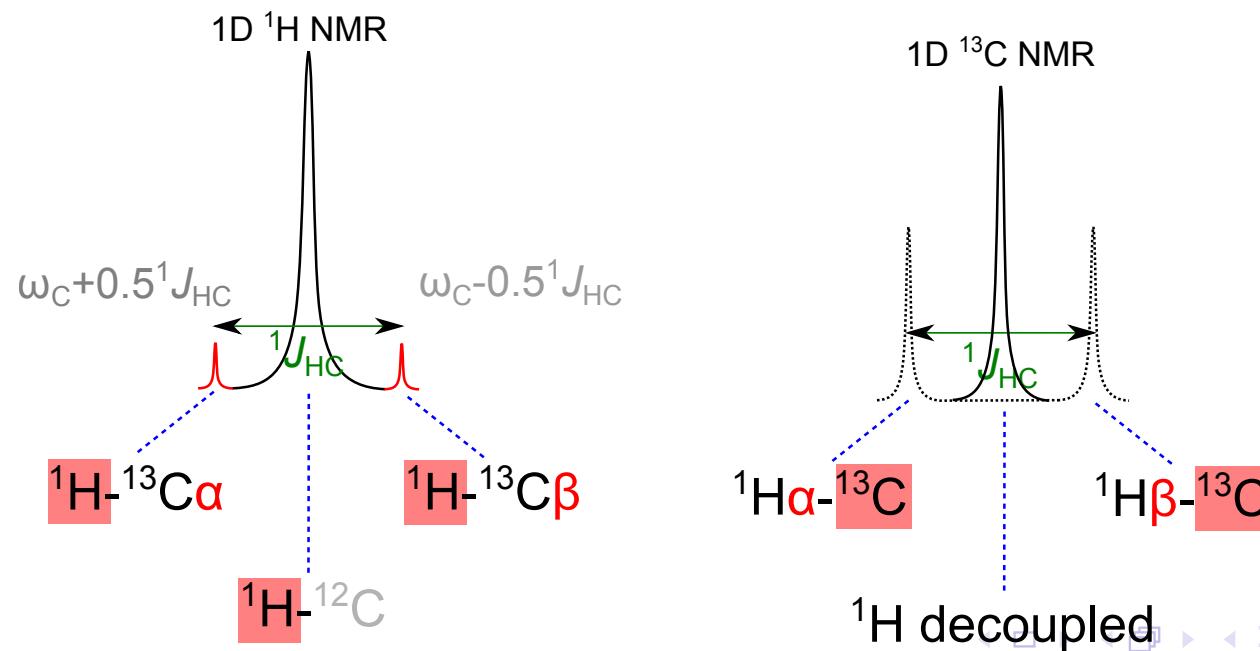
¹H vs ¹³C NMR

	¹ H	¹³ C
Spin number	¹ H: $\mathbf{s}=\frac{1}{2}$ \times ² H: s=1	¹³ C: $\mathbf{s}=\frac{1}{2}$ \times ¹² C: s=0
Abundance [%]	99.98	1.1
Gyromagnetic ratio [10 ⁷ rad.T ⁻¹ .s ⁻¹]	26.8	6.7
Chemical shift range [ppm]	0 - 15	0 - 200
Nuclear shielding	σ_{dia}	$\sigma_{\text{dia}} + \sigma_{\text{para}}$
Integration of signals	✓	✗
<i>T</i> ₁ relaxation [s]	1-20	1-40
Homonuclear <i>J</i> -interaction	✓	✗
H \leftrightarrow C <i>J</i> -interaction (\sim 100-250 Hz)	carbon satellites	(<i>n</i> + 1) splitting \times decoupling

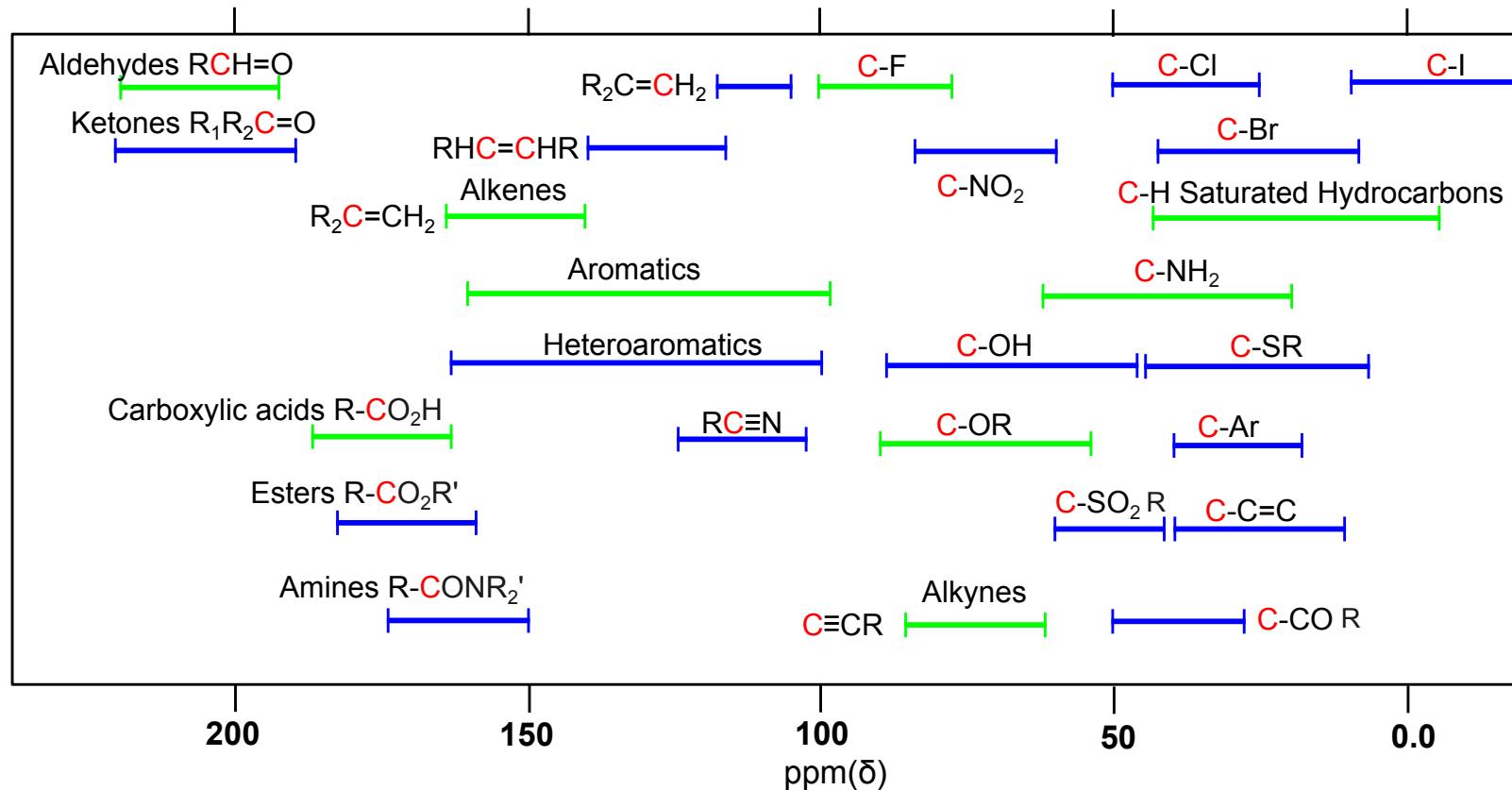


^1H vs ^{13}C NMR

	^1H	^{13}C
Spin number	$^1\text{H}: \mathbf{s}=\frac{1}{2} \times ^2\text{H}: \mathbf{s}=1$	$^{13}\text{C}: \mathbf{s}=\frac{1}{2} \times ^{12}\text{C}: \mathbf{s}=0$
Abundance [%]	99.98	1.1
Gyromagnetic ratio [$10^7 \text{ rad.T}^{-1}.\text{s}^{-1}$]	26.8	6.7
Chemical shift range [ppm]	0 - 15	0 - 200
Nuclear shielding	σ_{dia}	$\sigma_{\text{dia}} + \sigma_{\text{para}}$
Integration of signals	✓	✗
T_1 relaxation [s]	1-20	1-40
Homonuclear J -interaction	✓	✗
$\text{H} \leftrightarrow \text{C} J$ -interaction ($\sim 100\text{-}250 \text{ Hz}$)	carbon satellites	($n+1$) splitting \times decoupling



Important regions of ^{13}C chemical shifts



$^1J_{\text{CH}}$ depends on the bond order (hybridization \Leftrightarrow s-character)

- ▶ -C-H $^1J_{\text{CH}} \approx 125 \text{ Hz}$
- ▶ =C-H $^1J_{\text{CH}} \approx 160 \text{ Hz}$
- ▶ $\equiv\text{C-H}$ $^1J_{\text{CH}} \approx 250 \text{ Hz}$
- ▶ X-C-H
 - ▶ X = N, O, S, F, Cl, ... $^1J_{\text{CH}} \uparrow$
 - ▶ X = Li, Mg, ... $^1J_{\text{CH}} \downarrow$

$^2J_{\text{CH}} < 0$ or close to zero (<3 Hz)

- ▶ often not observable

in 1D ^{13}C H-C interaction suppressed by DECOUPLING
⇒ simplification of spectra (splitting removed, sensitivity)

- ▶ saturation of ^1H energy levels during decoupling enhances relatively intensity of ^{13}C signals because of heteronuclear nOe ⇒ quaternary carbons usually less intensive.

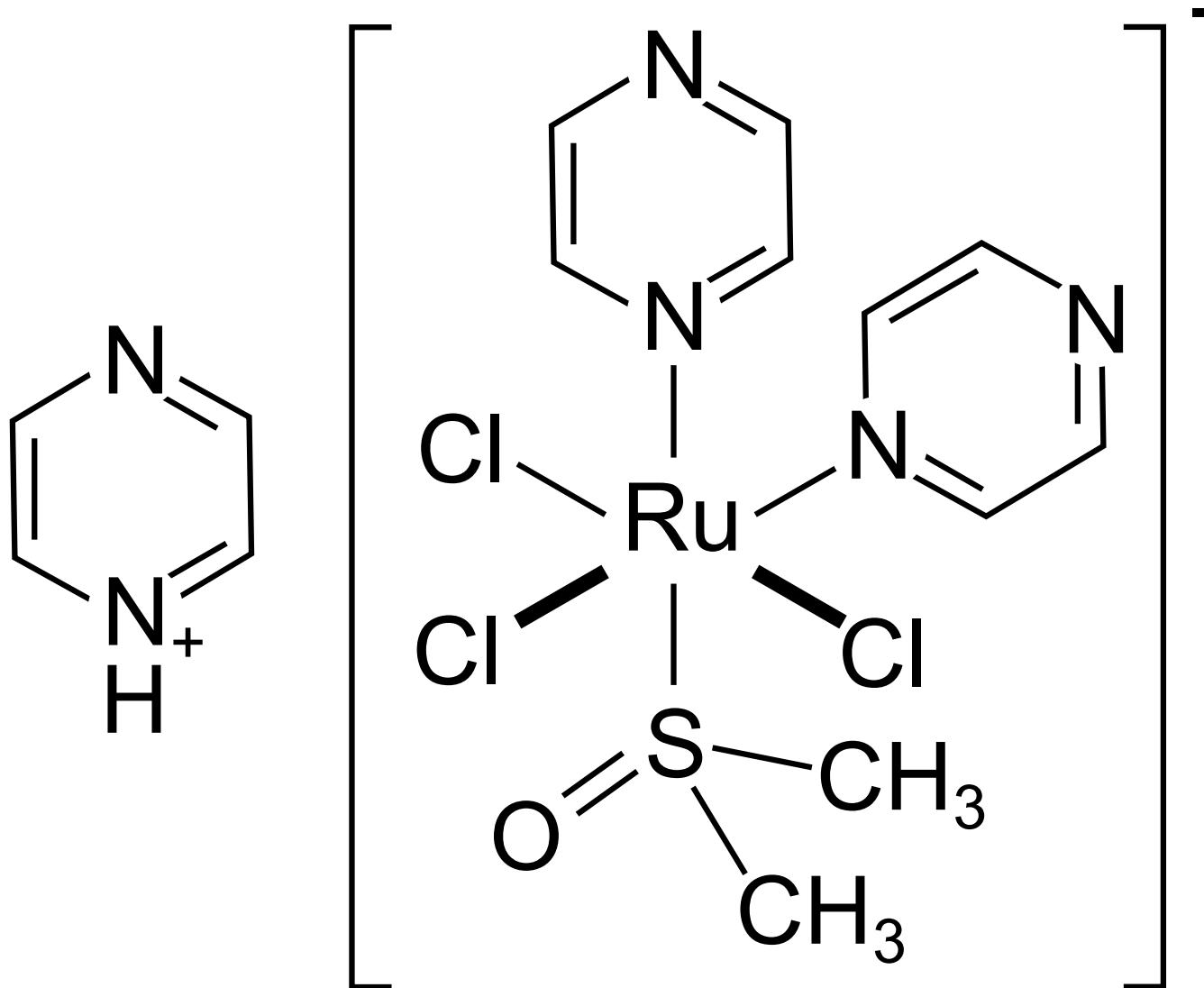
Values of chemical shift of important solvents

Abbr.	Formula	^1H	^{13}C
ACN	CH_3CN	1.9	118
Benzene	C_6H_6	7.2	128
	CHCl_3	7.2	77
DCM	CH_2Cl_2	5.3	54
DMF	$(\text{CH}_3)_2\text{NCHO}$	2.9, 8.0	32, 163
DMSO	$(\text{CH}_3)_2\text{SO}$	2.5	40
MeOH	CH_3OH	3.3, 4.8	49
Water	H_2O	4.8	-

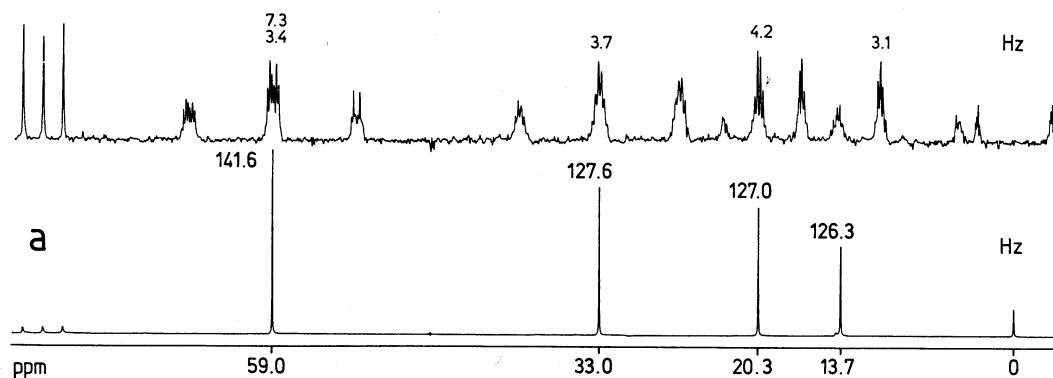
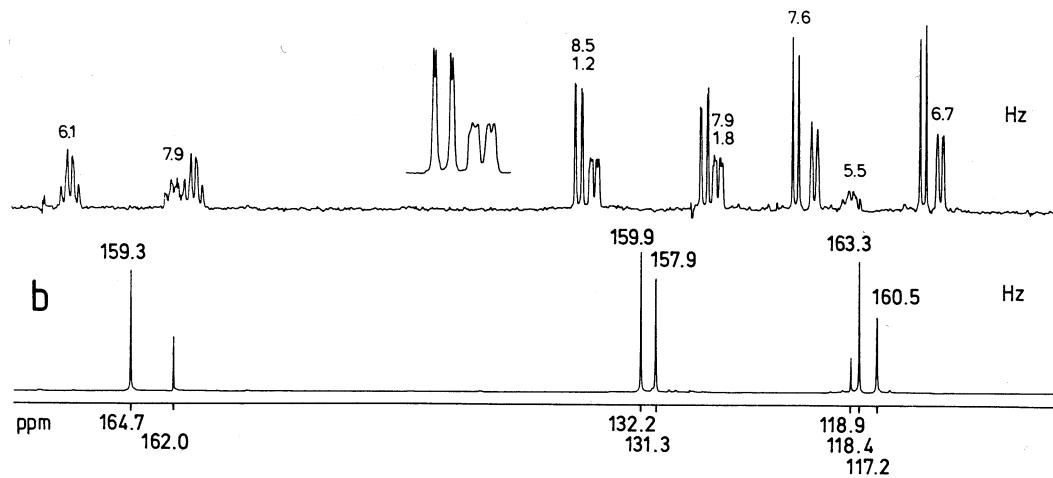
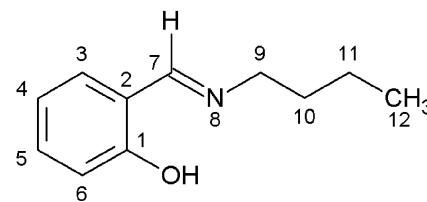
EXPLAIN effect of solvent on the position of residual ^1H water signal:

CHCl_3 - 1.6, ACN - 2.1, DMSO - 3.3, MeOH - 4.9

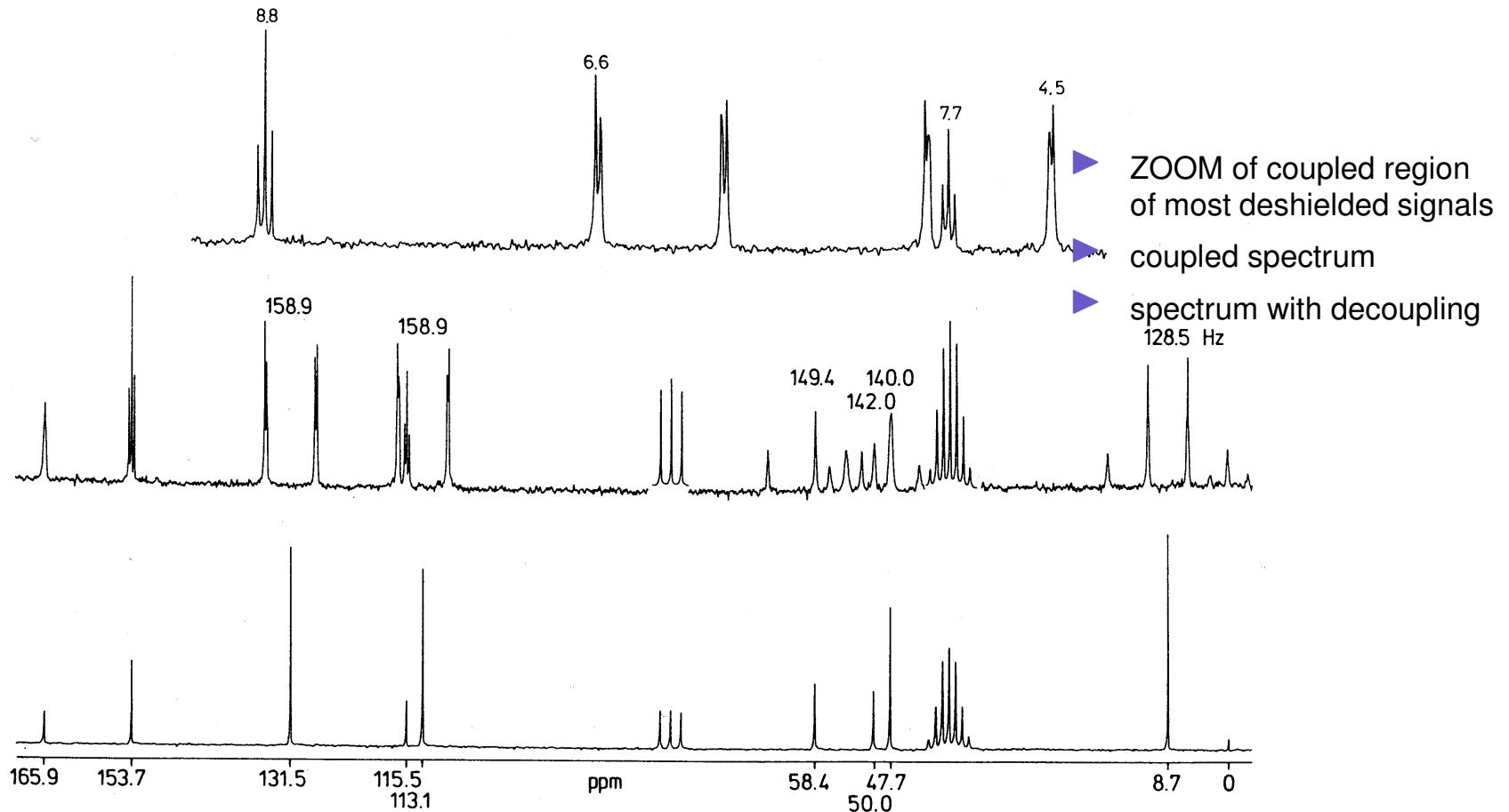
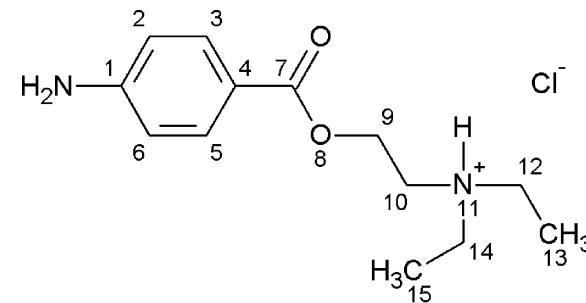
How many ^{13}C signal would you expect in the NMR spectrum?



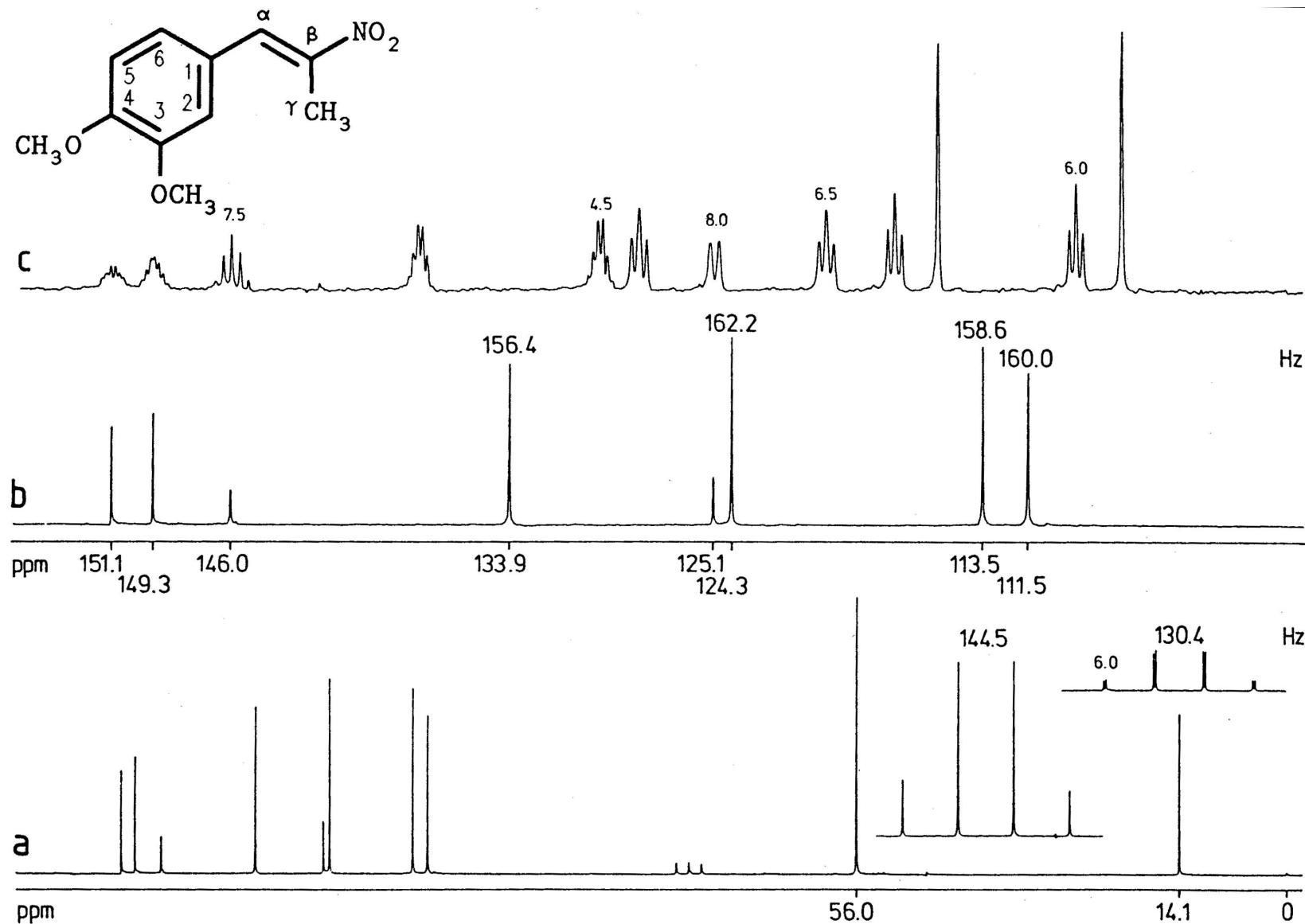
1D ^{13}C -NMR 1, bottom without CPD



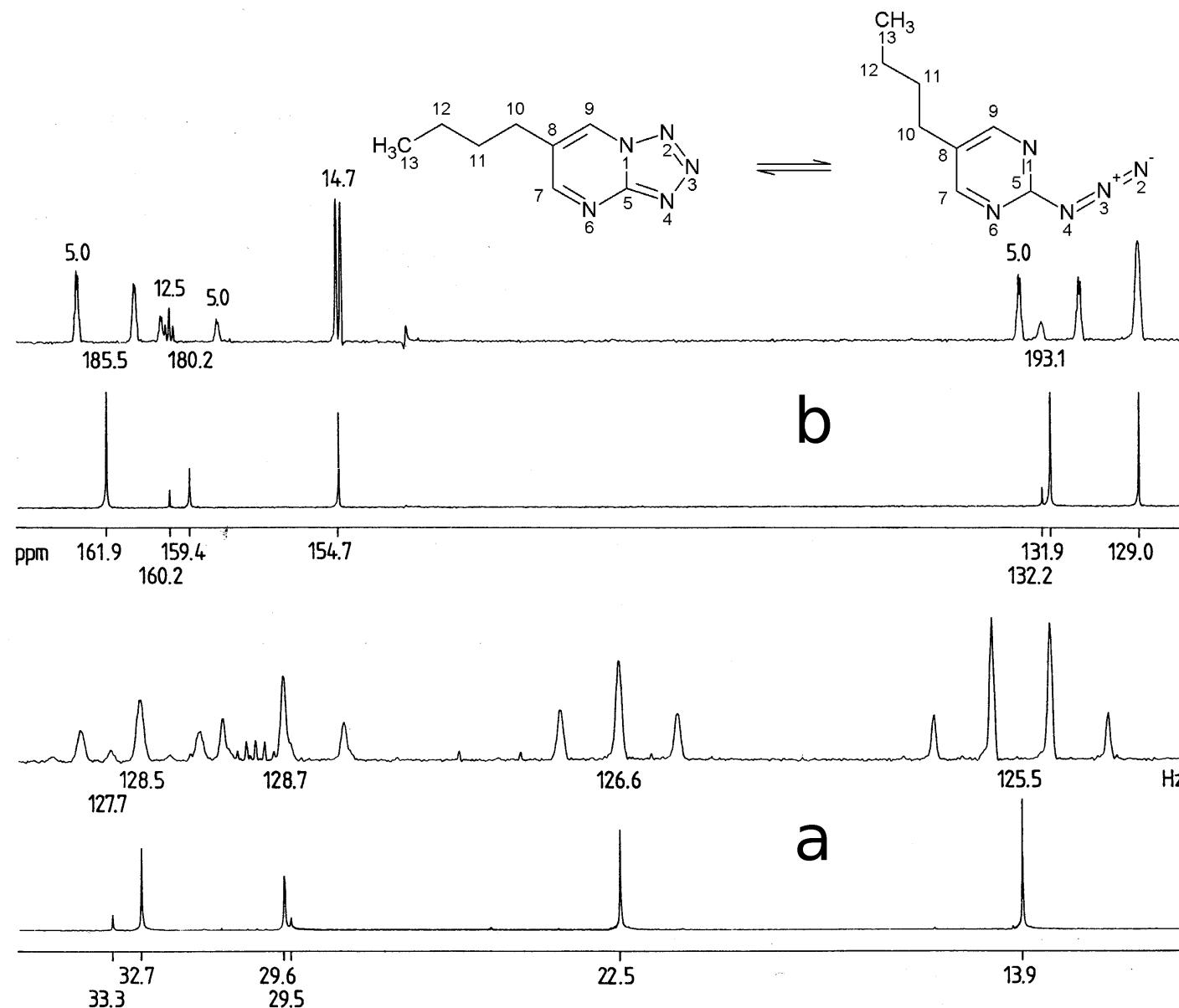
1D ^{13}C -NMR 2



1D ^{13}C -NMR 3, *b* - zoom of right region, *a* - full decoupled spectrum



1D ^{13}C -NMR 4, consider equilibrium minor-major form



Which form dominates and why?

Next topic

Vector Model + ^{13}C APT experiment