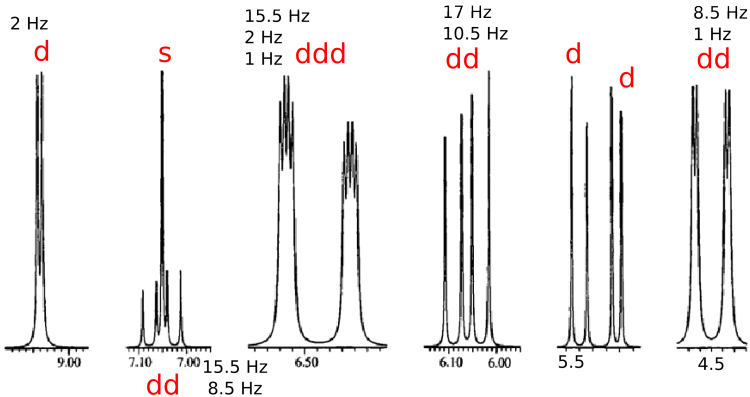
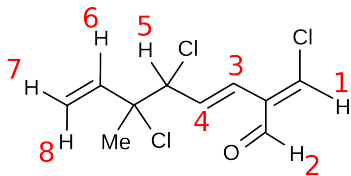


C8953
NMR structural analysis - seminar
1D ^{13}C -NMR

Kateřina Peterková, Aleř Novotný
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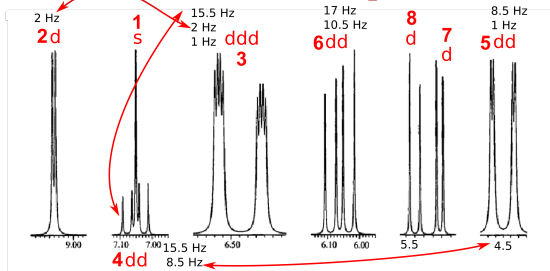
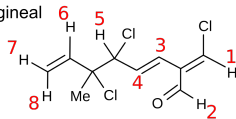
March 13, 2019

Control of 1D ^1H NMR spectrum of cartilageneal



Control of 1D ^1H NMR spectrum - cartilaginal

1D ^1H NMR - cartilaginal



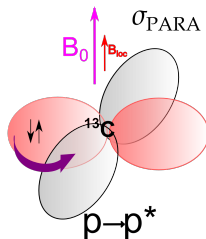
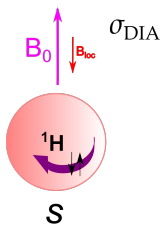
Notes:

- ▶ highest shift - **H-2** - proton of aldehydic group, splitted to doublet with $J = 2$ Hz (small value, interacting partner is relatively far away)
- ▶ the same $J = 2$ Hz belongs to doublet of doublets of doublets around 6.5 ppm, other Js: $J = 1$ Hz and $J = 15.5$ Hz - three J -constants - three partners - **H-3**
- ▶ large J -value 15.5 Hz suggests near neighbor - other signal with the same constant is doublet of doublets around 7 ppm - **H-4**
- ▶ last constant of multiplet at 6.5 ppm - $J = 1$ Hz - partner distant from H-3: either H-1 or H-5, the same J -constant belongs to doublet of doublets at 4.5 ppm - two constants, two partners which is not the case for H-1 - therefore signal at 4.5 ppm belongs to **H-5**

- ▶ just for check: both multiplets H-4 and H-5 are coupled with H-3 and with each other as well ($J = 8.5$ Hz)
- ▶ the only singlet in the spectrum is isolated **H-1**
- ▶ last unassigned doublet of doublets (6,1 ppm) must be **H-6** because it is the only proton from the trio H-6, H-7, H-8 with two unequivalent neighbors - larger coupling comes from interaction with **H-8** in *trans* position, smaller coupling comes from interaction with *cis* oriented **H-7**
- ▶ signal of the methyl group is not present in this spectrum

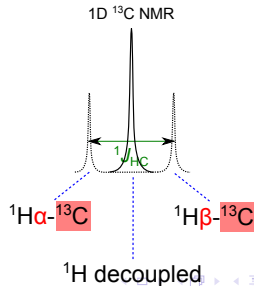
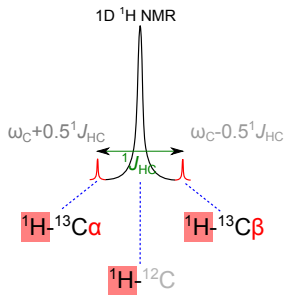
^1H vs ^{13}C NMR

	^1H	^{13}C
Spin number	$^1\text{H}: s=\frac{1}{2} \times ^2\text{H}: s=1$	$^{13}\text{C}: s=\frac{1}{2} \times ^{12}\text{C}: s=0$
Abundance [%]	99.98	1.1
Gyromagnetic ratio [$10^7 \text{ rad}\cdot\text{T}^{-1}\cdot\text{s}^{-1}$]	26.8	6.7
Chemical shift range [ppm]	0 - 15	0 - 200
Nuclear shielding	σ_{dia}	$\sigma_{\text{dia}} + \sigma_{\text{para}}$
Integration of signals	✓	✗
T_1 relaxation [s]	1-20	1-40
Homonuclear J -interaction	✓	✗
$\text{H} \leftrightarrow \text{C}$ J -interaction ($\sim 100\text{-}250 \text{ Hz}$)	carbon satellites	$(n+1)$ splitting \times decoupling

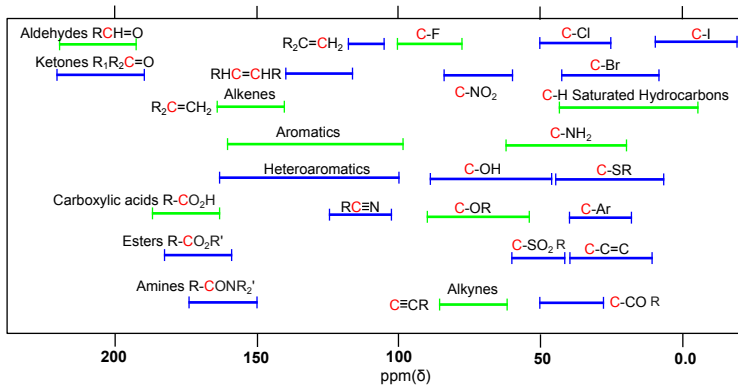


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Important regions of ^{13}C chemical shifts



$^1J_{\text{CH}}$ depends on the bond order (hybridization \Leftrightarrow s-character)

- ▶ $-\text{C}-\text{H} \quad ^1J_{\text{CH}} \approx 125 \text{ Hz}$
- ▶ $=\text{C}-\text{H} \quad ^1J_{\text{CH}} \approx 160 \text{ Hz}$
- ▶ $\equiv\text{C}-\text{H} \quad ^1J_{\text{CH}} \approx 250 \text{ Hz}$
- ▶ X-C-H
 - ▶ X = N, O, S, F, Cl, ... $^1J_{\text{CH}} \uparrow$
 - ▶ X = Li, Mg, ... $^1J_{\text{CH}} \downarrow$

$^2J_{\text{CH}} < 0$ or close to zero ($< 3 \text{ Hz}$)

- ▶ often not observable

in 1D ^{13}C H-C interaction suppressed by DECOUPLING
 \Rightarrow simplification of spectra (splitting removed, sensitivity)

- ▶ saturation of ^1H energy levels during decoupling enhances relatively intensity of ^{13}C signals because of heteronuclear nOe \Rightarrow quaternary carbons usually less intensive.

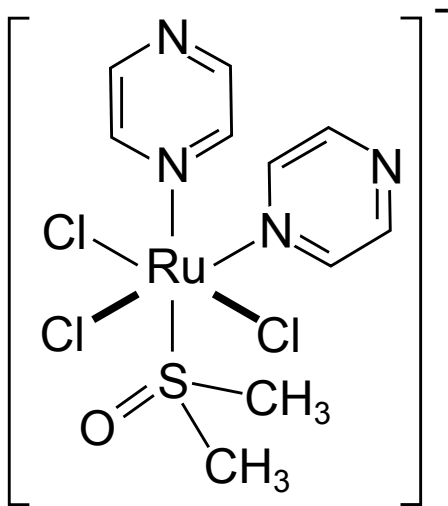
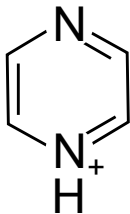
Values of chemical shift of important solvents

Abbr.	Formula	^1H	^{13}C
ACN	CH_3CN	1.9	118
Benzene	C_6H_6	7.2	128
	CHCl_3	7.2	77
DCM	CH_2Cl_2	5.3	54
DMF	$(\text{CH}_3)_2\text{NCHO}$	2.9, 8.0	32, 163
DMSO	$(\text{CH}_3)_2\text{SO}$	2.5	40
MeOH	CH_3OH	3.3, 4.8	49
Water	H_2O	4.8	-

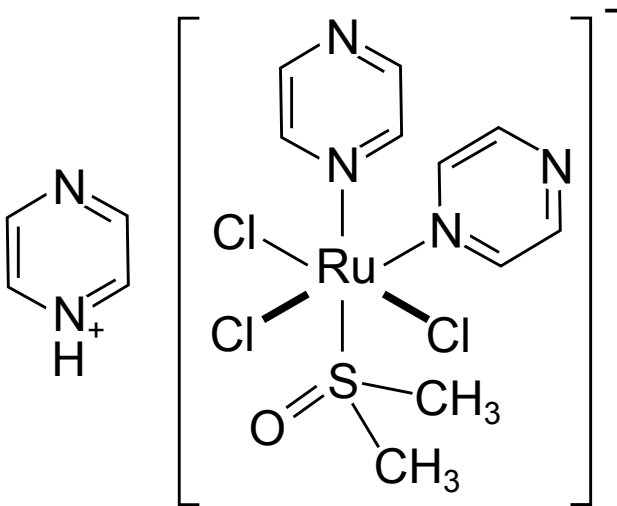
EXPLAIN effect of solvent on the position of residual ^1H water signal:

CHCl_3 - 1.6, ACN - 2.1, DMSO - 3.3, MeOH - 4.9

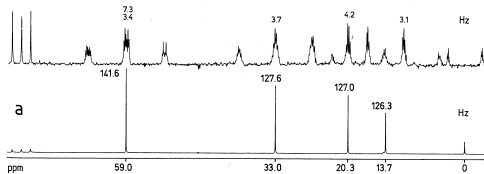
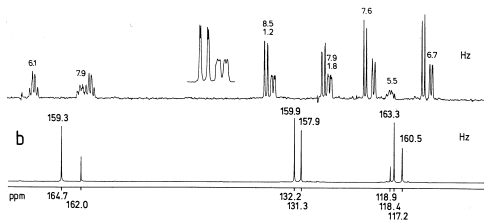
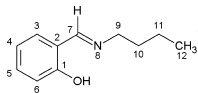
How many ^{13}C signal would you expect in the NMR spectrum?



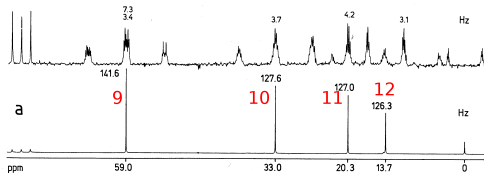
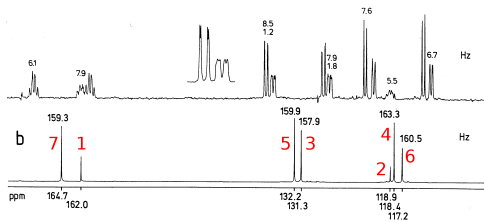
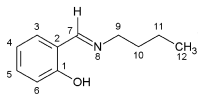
How many ^{13}C signal would you expect in the NMR spectrum? **7**



1D ^{13}C -NMR 1, bottom without CPD



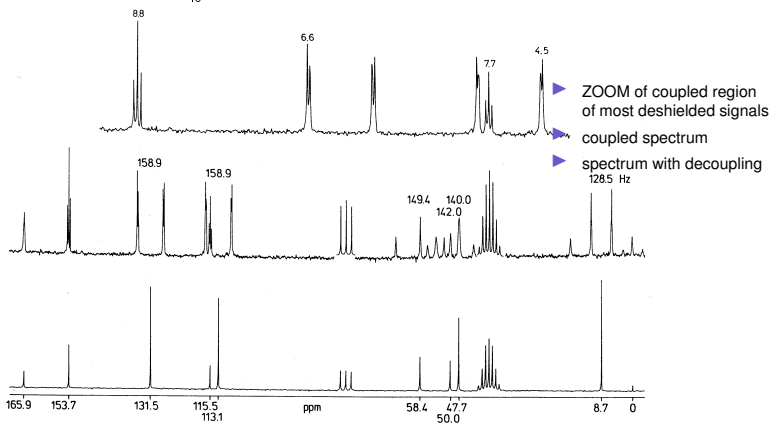
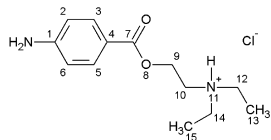
1D ^{13}C -NMR 1, bottom without CPD



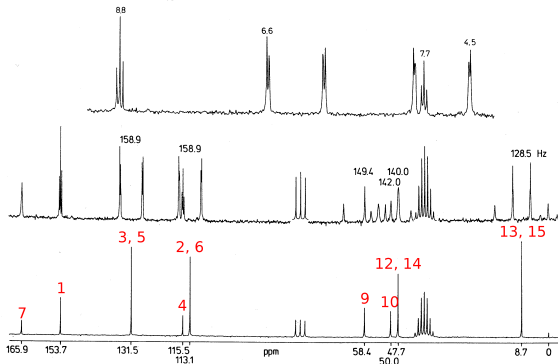
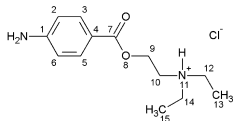
Notes:

- ▶ numbers at top of peaks refers to values J_{HC} constants
- ▶ **C1+C7** connected to electronegative groups (C1 quaternary)
- ▶ **C2** ipso aromatic, **C4+C6** shielded by M^+ of OH
- ▶ **C5+C4** NOE-enhanced in bit larger extend by close H
- ▶ **C9** → **C12**: decaying effect of N8

1D ^{13}C -NMR 2



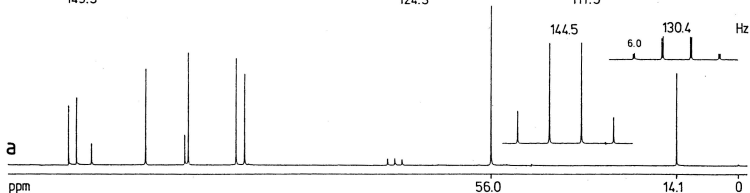
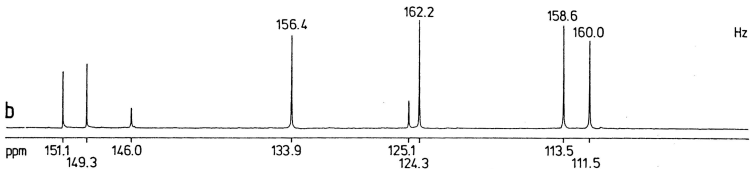
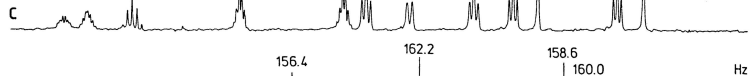
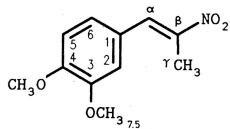
1D ^{13}C -NMR 2



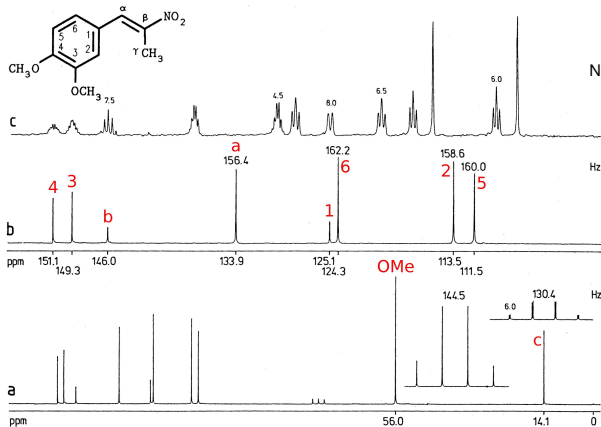
Notes:

- ▶ **C7** carbonyl, **C1** attached to N
- ▶ **C3/5** deshielded by M-CO, **C2/6** shielded by M+ of NH_2
- ▶ **C4** last quaternary aromatic signal (most isolated from H nuclei)
- ▶ **C9** effect of esteric group, ? **C10** affected by NH exchange
- ▶ **C12/C14 + C13/C15** decaying effect of N^+

1D ^{13}C -NMR 3, *b* - zoom of right region, *a* - full decoupled spectrum



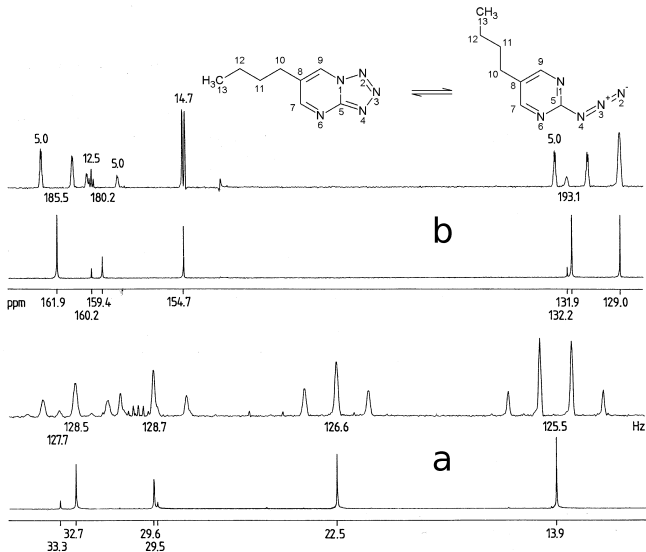
1D ^{13}C -NMR 3, *b* - zoom of right region, *a* - full decoupled spectrum



Notes:

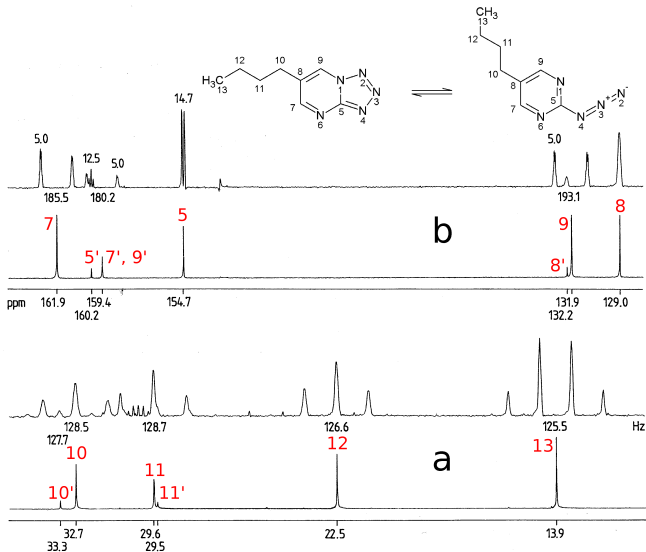
- ▶ **C3/C4** quaternary aromatic deshielded by O, **C β** quaternary coupled by CH₃ and C α H
- ▶ **C α** deshielded by NO₂
- ▶ **C1** last quaternary aromatic signal
- ▶ **C2/C6** coupled mutually and with C α , **C5** isolated (constrained) (constrained)
- ▶ quartets **OMe**, **C γ**

1D ^{13}C -NMR 4, consider equilibrium minor-major form



Which form dominates and why?

1D ^{13}C -NMR 4, consider equilibrium minor-major form



Which form dominates and why?

Next topic

Vector Model + ^{13}C APT experiment