C8953
NMR structural analysis - seminar
1D ^{13}C-NMR

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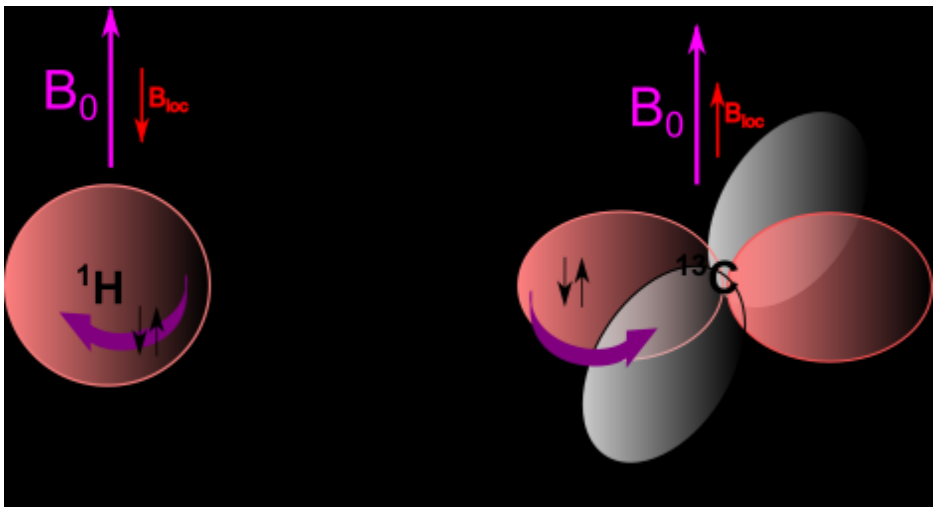
^{1}H vs ^{13}C NMR

	^{1}H	^{13}C
Spin number	**^{1}H: s=$\frac{1}{2}$** ^{2}H: s=1	**^{13}C: s=$\frac{1}{2}$** ^{12}C: s=0
Abundance [%]	99.98	1.1
Gyromagnetic ratio [10^{7} rad.T^{-1}.s^{-1}]	26.8	6.7
Chemical shift range [ppm]	0 - 15	0 - 200
Nuclear shielding	σ_{dia}	σ_{dia} + σ_{para}
Integration of signals	C	
T_1 relaxation [s]	1-20	1-40
Homonuclear *J*-interaction	C	
H↔C *J*-interaction (~ 100-250 Hz)	carbon satellites	(*n* + 1) splitting decoupling



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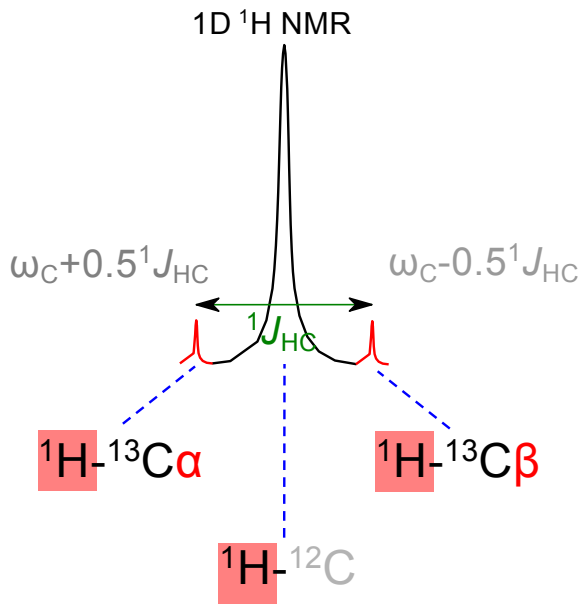
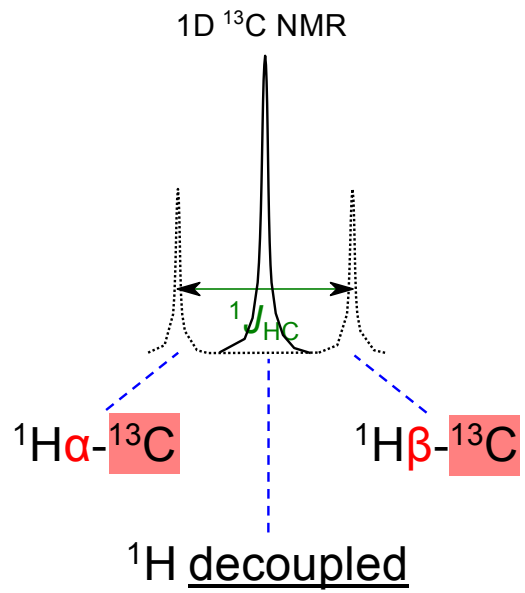
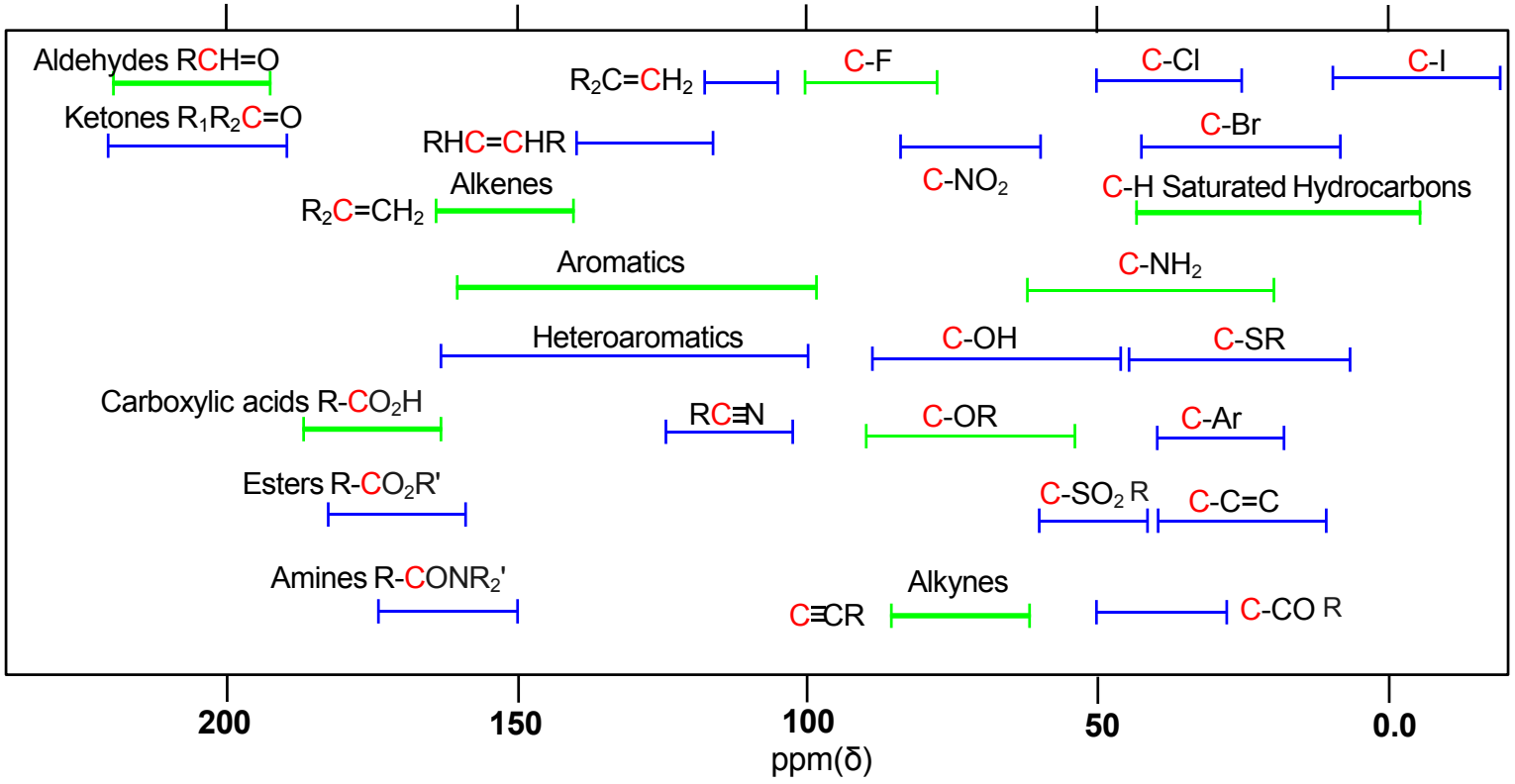




Important regions of ^{13}C chemical shifts



$^1J_{CH}$ depends on the bond order (hybridization $\Leftrightarrow$ c. haracter)

- -C-H $^1J_{CH} \approx 125\,Hz$
  =C-H $^1J_{CH} \approx 160\,Hz$
- $\equiv$C-H $^1J_{CH} \approx 250\,Hz$
- X-C-H
  - X = N, O, S, F, Cl, . . . $^1J_{CH} \Uparrow$
  - X = Li, Mg, . . . $^1J_{CH} \Downarrow$

$^2J_{CH} < 0$ or close to zero (<3 Hz)

- often not observable

in 1D ^{13}C H-C interaction suppressed by DECOUPLING $\Rightarrow$ simplification of spectra (splitting removed, sensitivity)

- saturation of ^{1}H energy levels during decoupling enhances relatively intensity of ^{13}C signals because of heteronuclear nOe $\Rightarrow$ quaternary carbons usually less intensive.

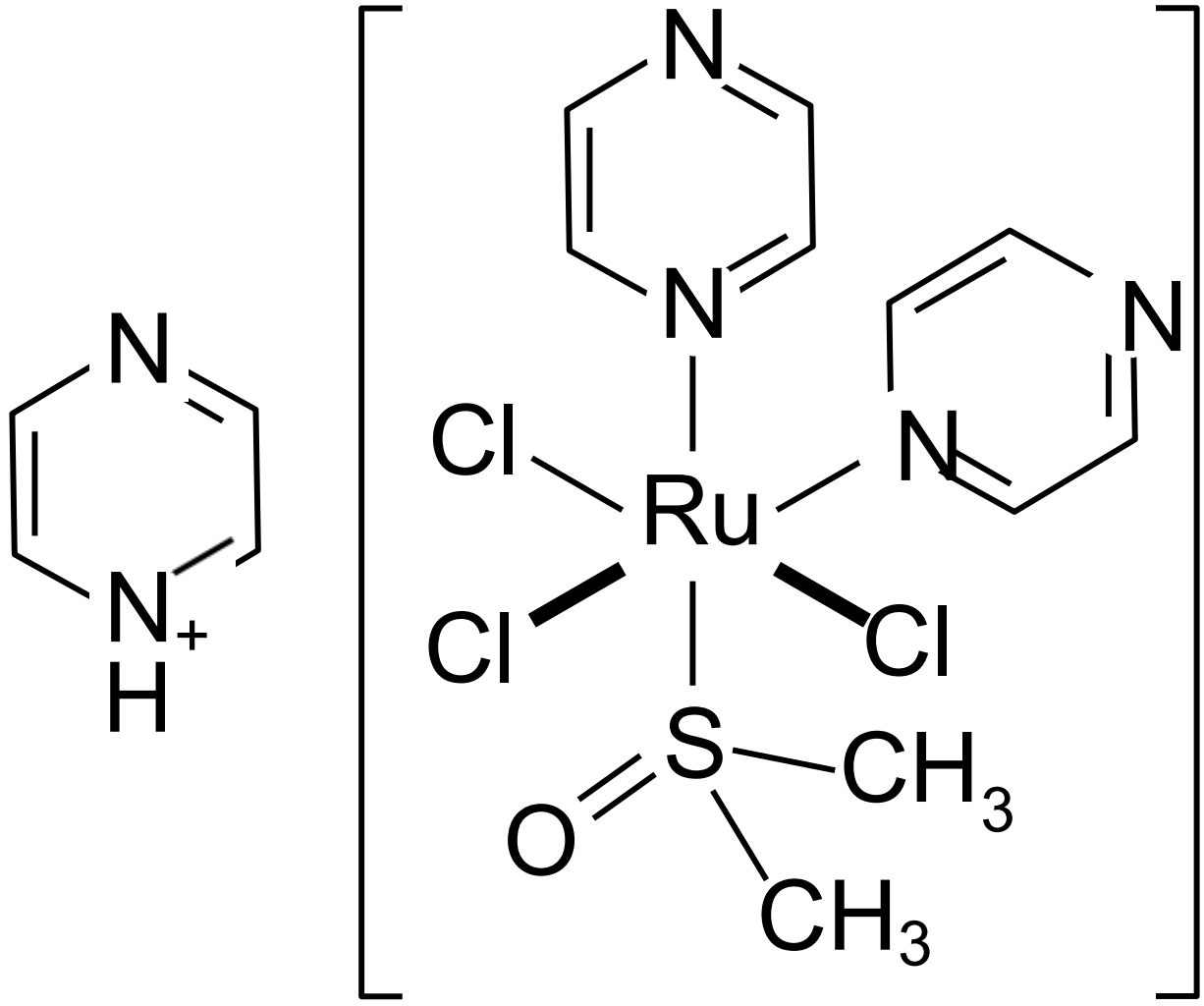
Values of chemical shift of important solvents

Abbr.	Formula	^{1}H	^{13}C
ACN	CH_3CN	1.9	118
Benzene	C_6H_6	7.2	128
	$CHCl_3$	7.2	77
DCM	CH_2Cl_2	5.3	54
DMF	$(CH_3)_2NCHO$	2.9, 8.0	32, 163
DMSO	$(CH_3)_2SO$	2.5	40
MeOH	CH_3OH	3.3, 4.8	49
Water	H_2O	**4.8**	-

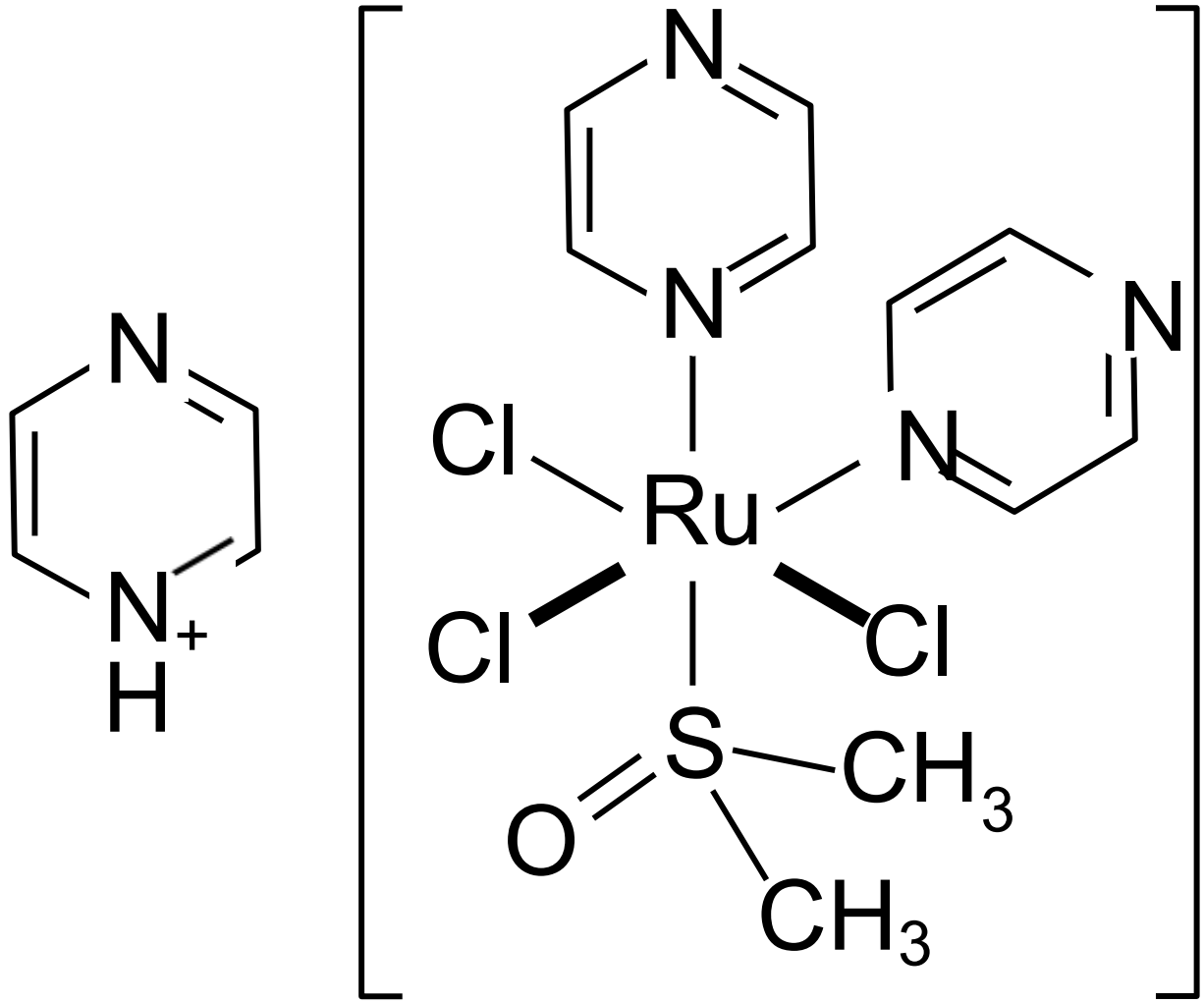
EXPLAIN effect of solvent on the position of residual ^{1}H water signal:

$CHCl_3$ - **1.6**, ACN - **2.1**, DMSO - **3.3**, MeOH - **4.9**

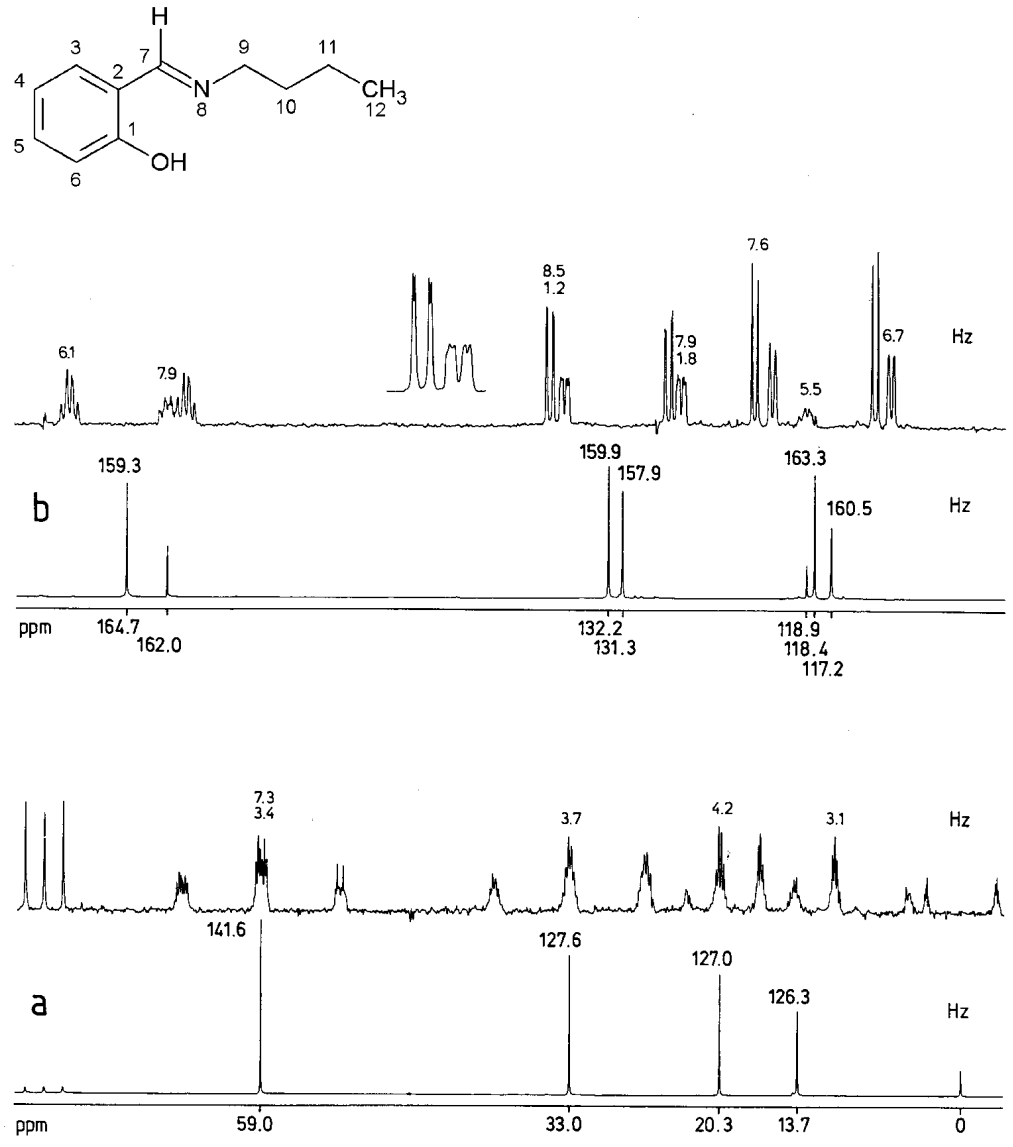
How many ^{13}C signal would you expect in the NMR spectrum?



How many ^{13}C signal would you expect in the NMR spectrum? **7**



1D ^{13}C-NMR 1, bottom without CPD



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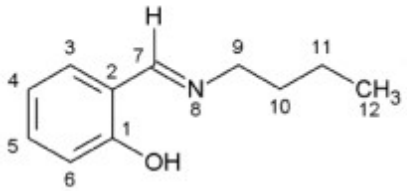
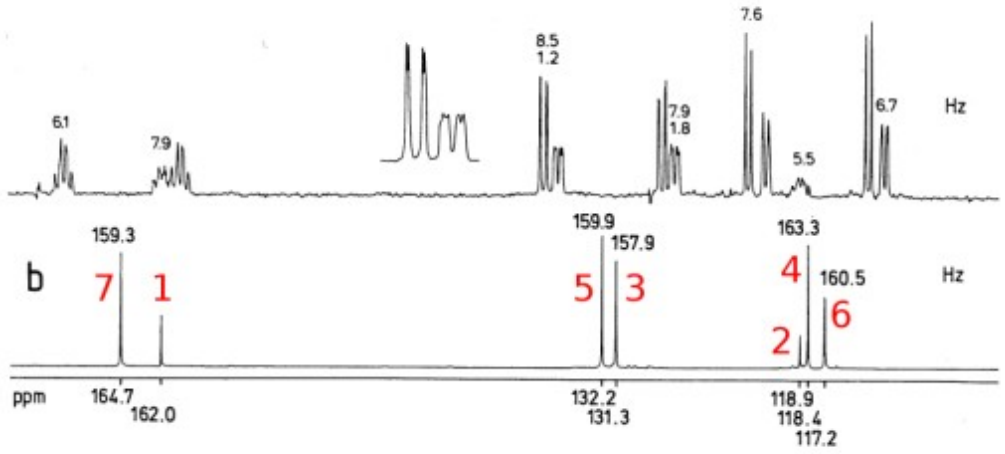
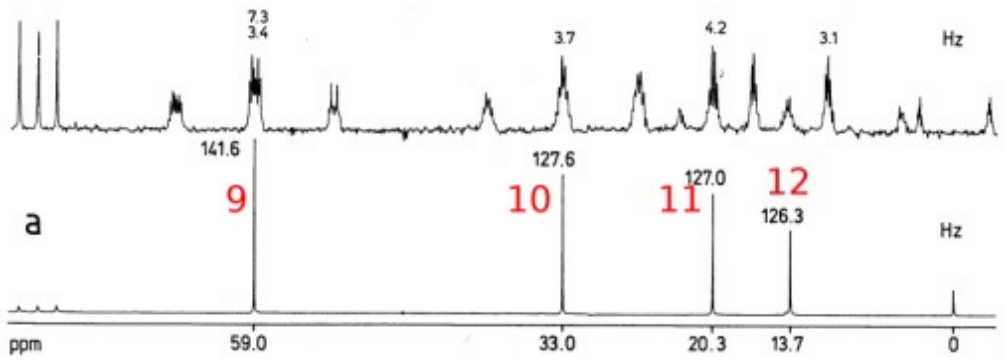


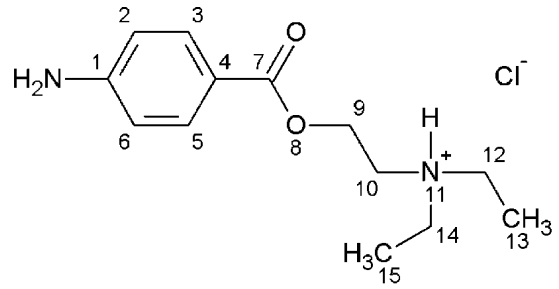


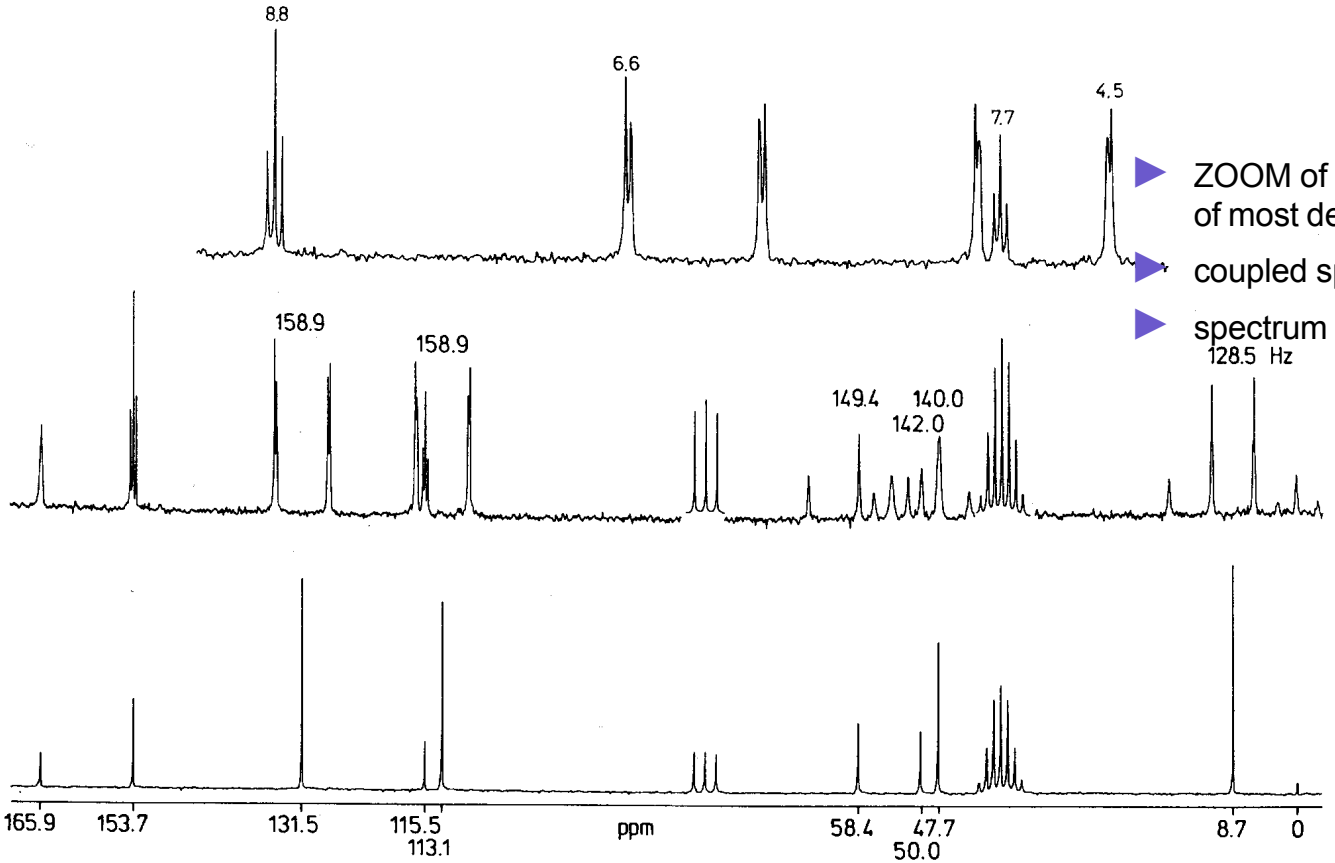


Notes:

- numbers at top of peaks refers to values J_{HC} constants
- **C1+C7** connected to electronegative groups (C1 quaternary)
- **C2** ipso aromatic, **C4+C6** shielded by M+ of OH
- **C5+C4** NOE-enhanced in bit larger extend by close H
- **C9→C12**: decaying effect of N8

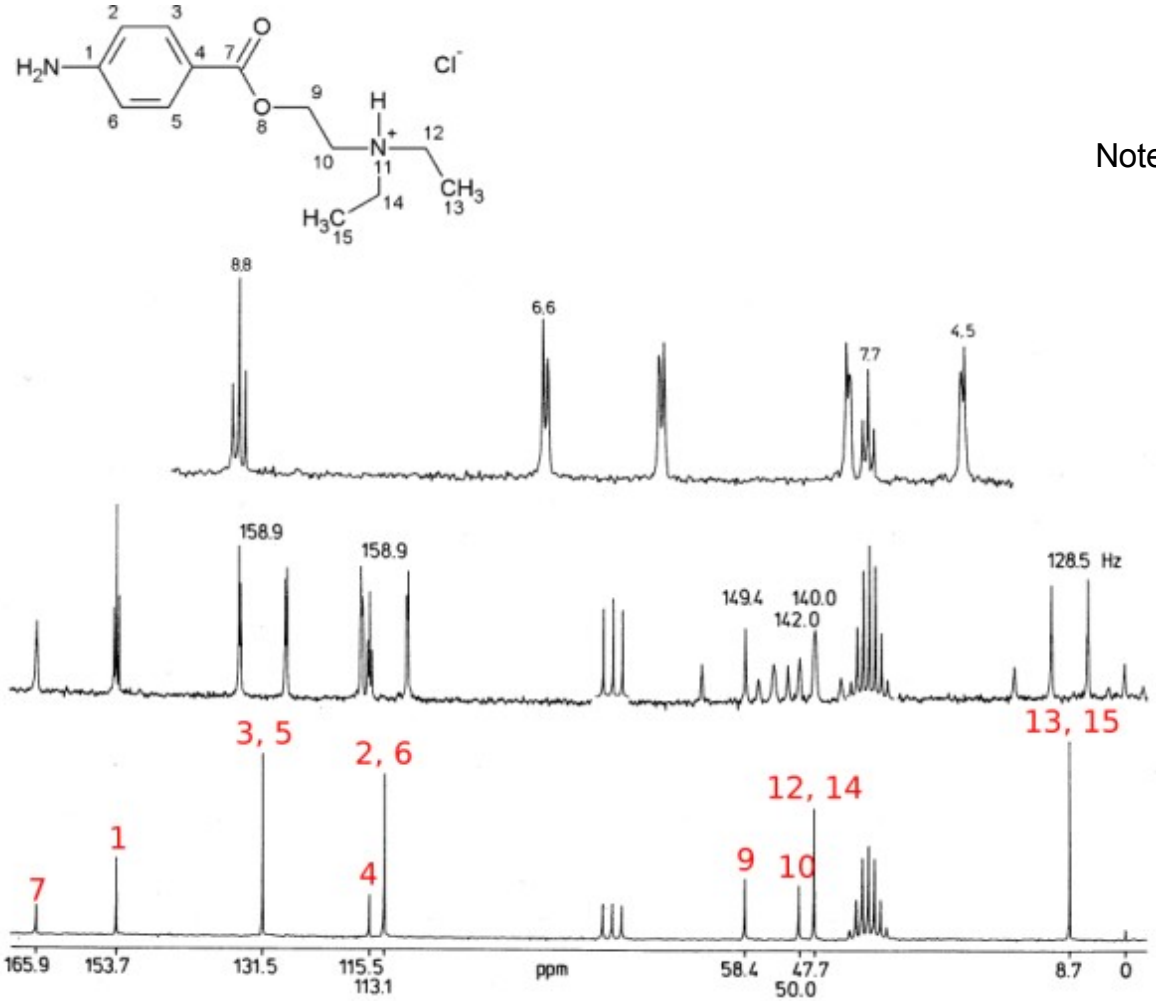
1D ^{13}C-NMR 2





- ZOOM of coupled region of most deshielded signals
- coupled spectrum
- spectrum with decoupling

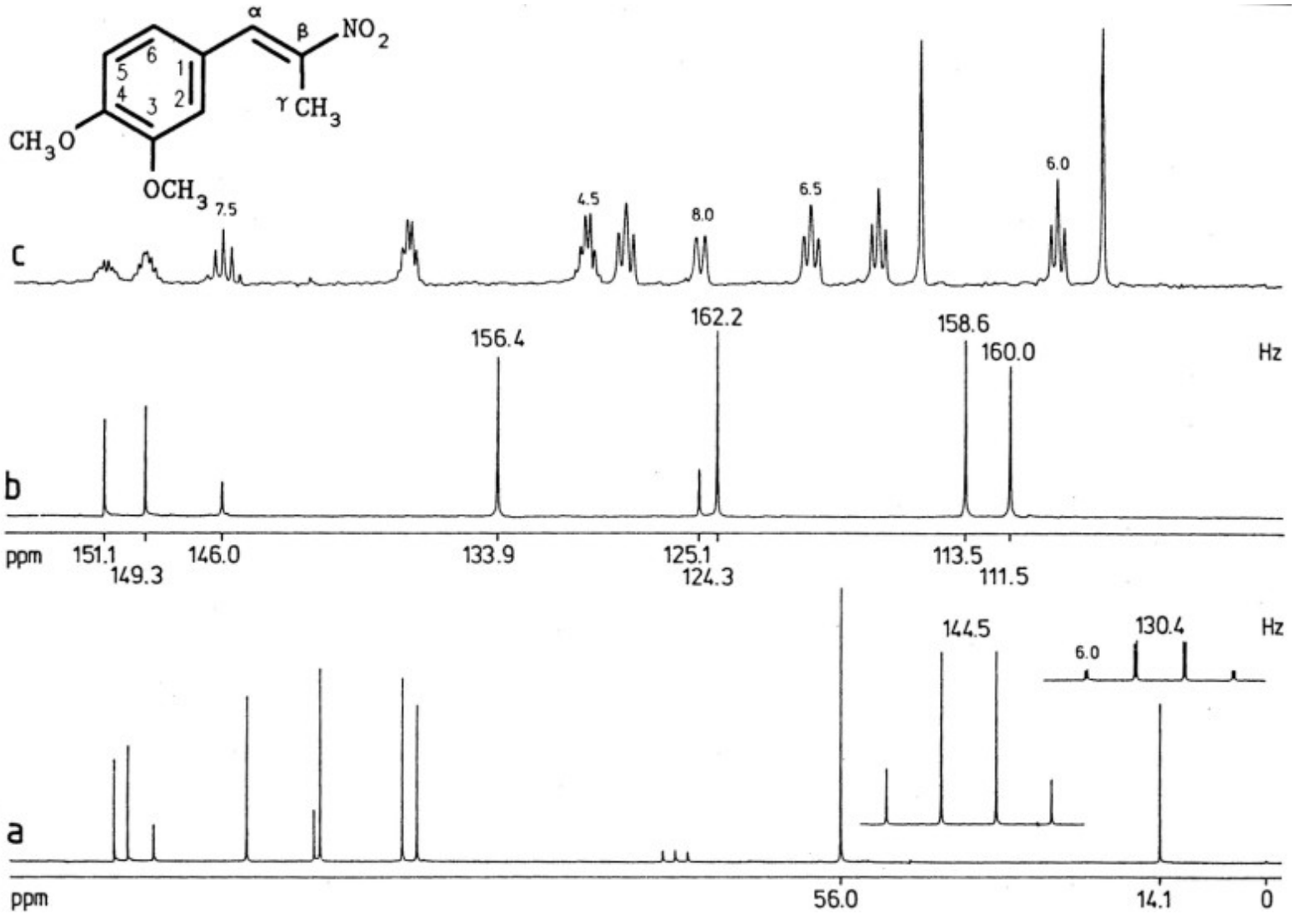
1D ^{13}C-NMR 2



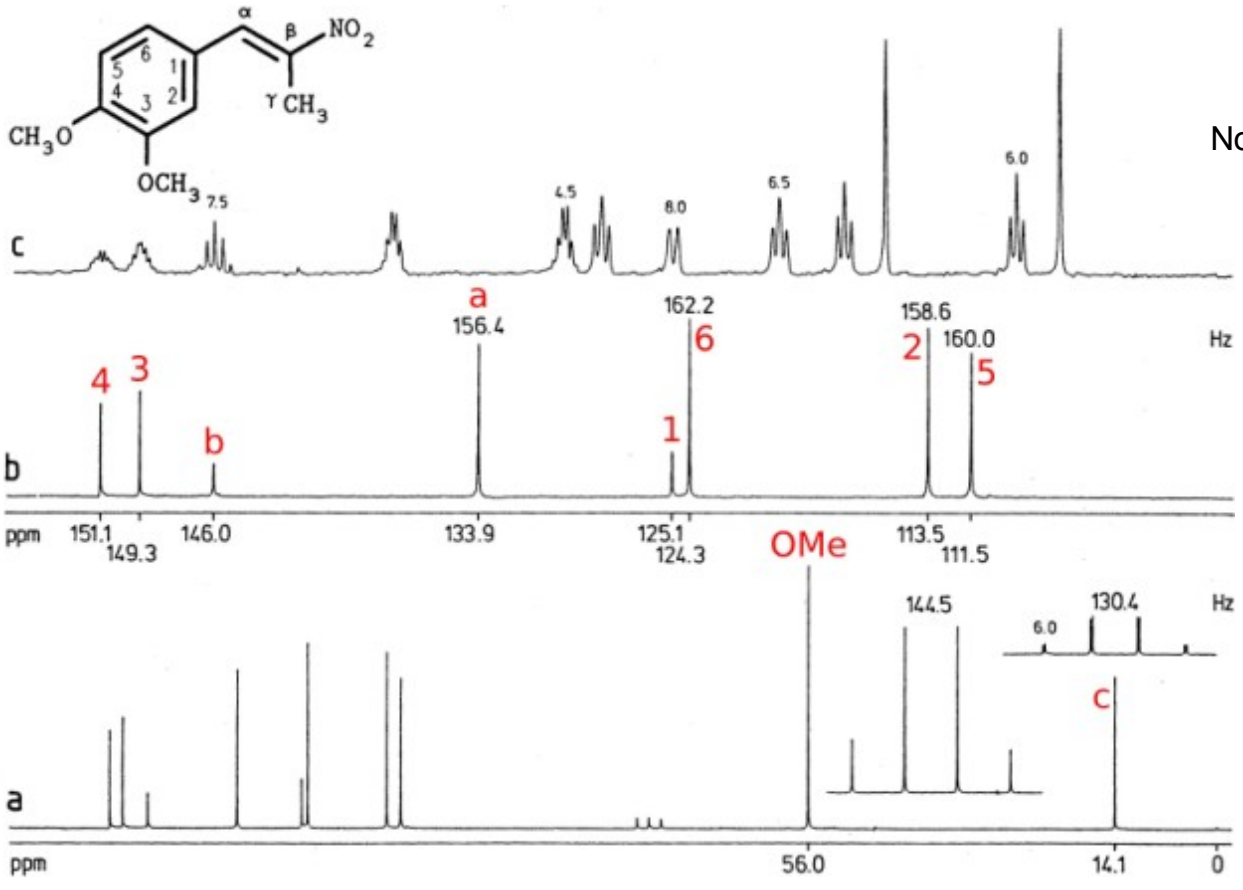
Notes:

- **C7** carbonyl, **C1** attached to N
- **C3/5** deshielded by M-CO, **C2/6** shielded by M+ of NH_2
- **C4** last quaternary aromatic signal (most isolated from H nuclei)
- **C9** effect of esteric group, ? **C10** affected by NH exchange
- **C12/C14 + C13/C15** decaying effect of N+

1D ^{13}C-NMR 3, *b* - zoom of right region, *a* - full decoupled spectrum



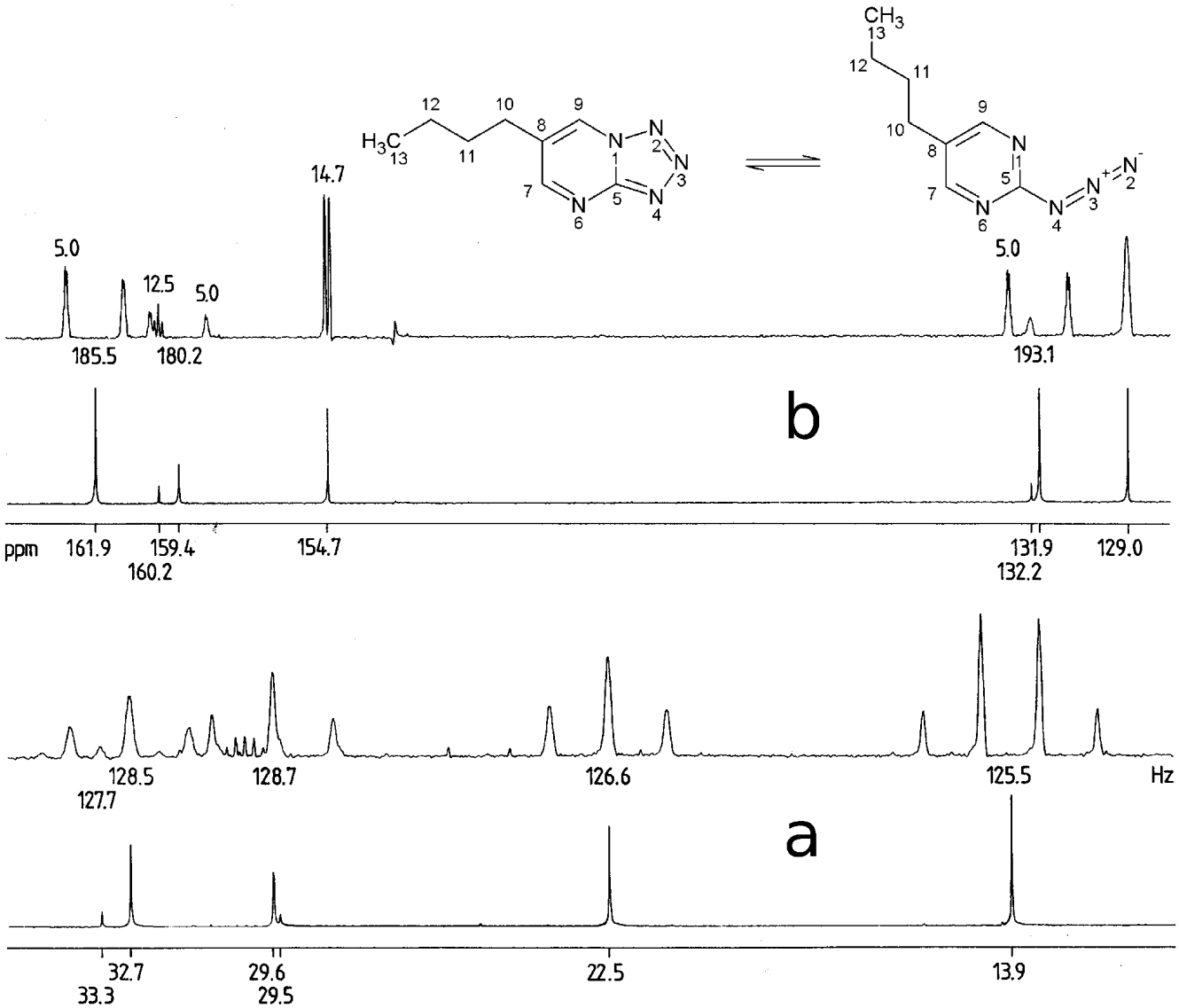
1D ^{13}C-NMR 3, *b* - zoom of right region, *a* - full decoupled spectrum



Notes:

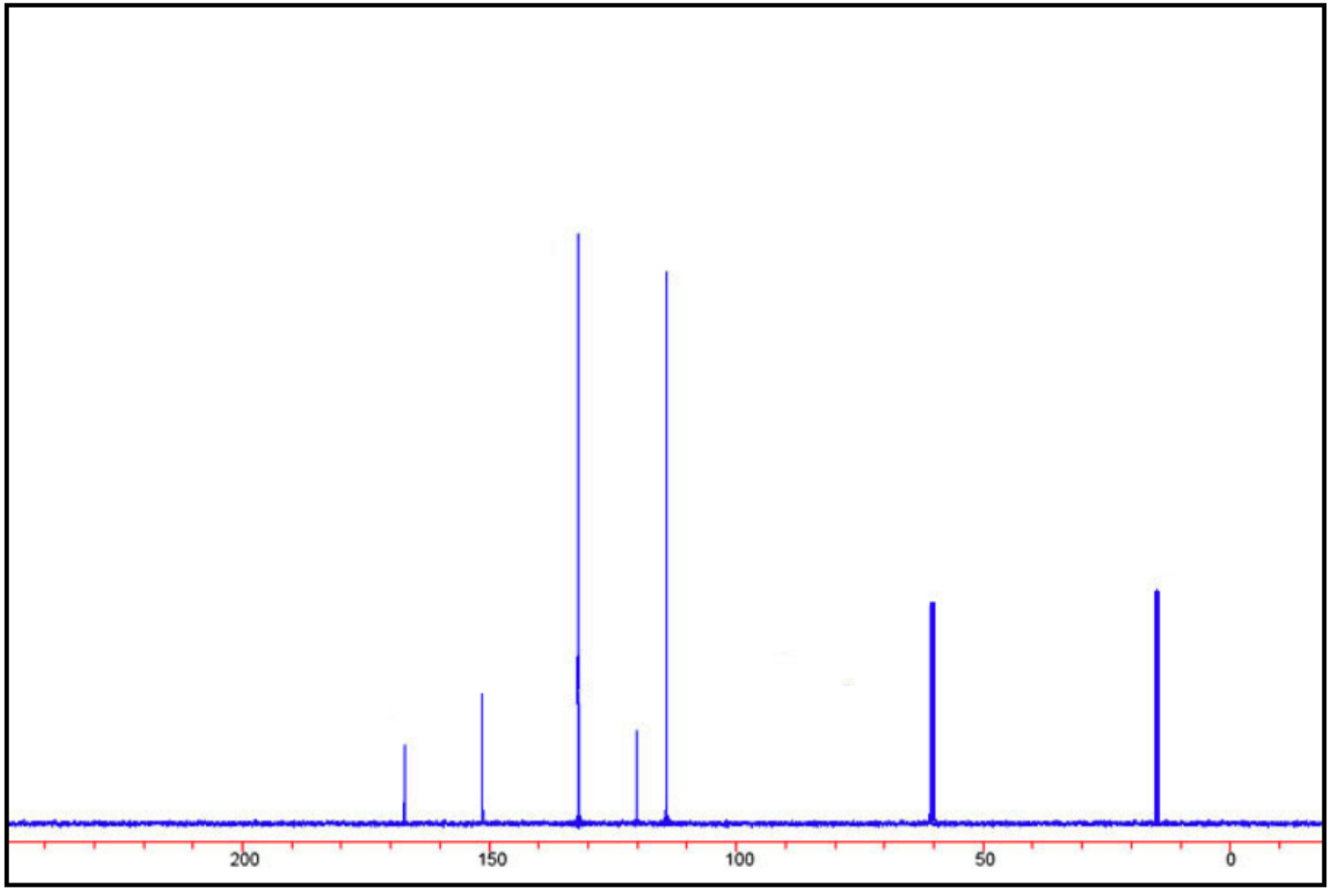
- **C3/C4** quaternary aromatic deshielded by O, **C**β quaternary coupled by CH_3 and CaH
- **C**a deshielded by NO_2
- **C1** last quaternary aromatic signal
- **C2/C6** coupled mutually and with Ca, **C5** isolated (contraintuitive)
- quartets **OMe**, **C**γ

1D ^{13}C-NMR 4, consider equilibrium minor-major form

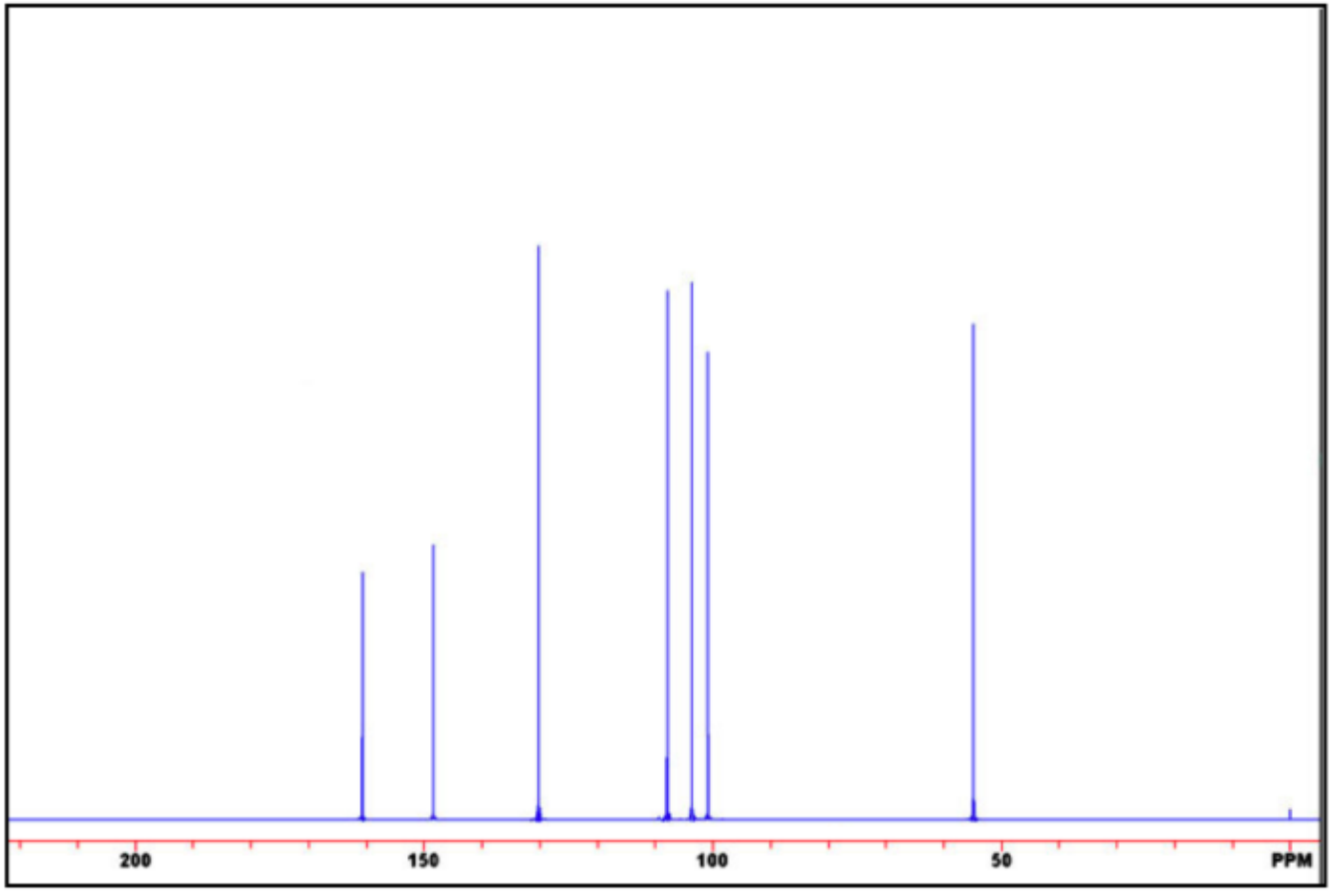


Which form dominates and why?

Draw structural formula of compound with summary formula $C_9H_{11}NO_2$ (use 1D ^{13}C NMR spectrum to find the right structure)



Draw structural formula of compound with summary formula C_7H_9NO (use 1D ^{13}C NMR spectrum to find the right structure)



Next topic

Vector Model + ^{13}C APT experiment