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E0321

Sustainable Development

– The biggest challenge today?



Michal Bittner, Ph.D.

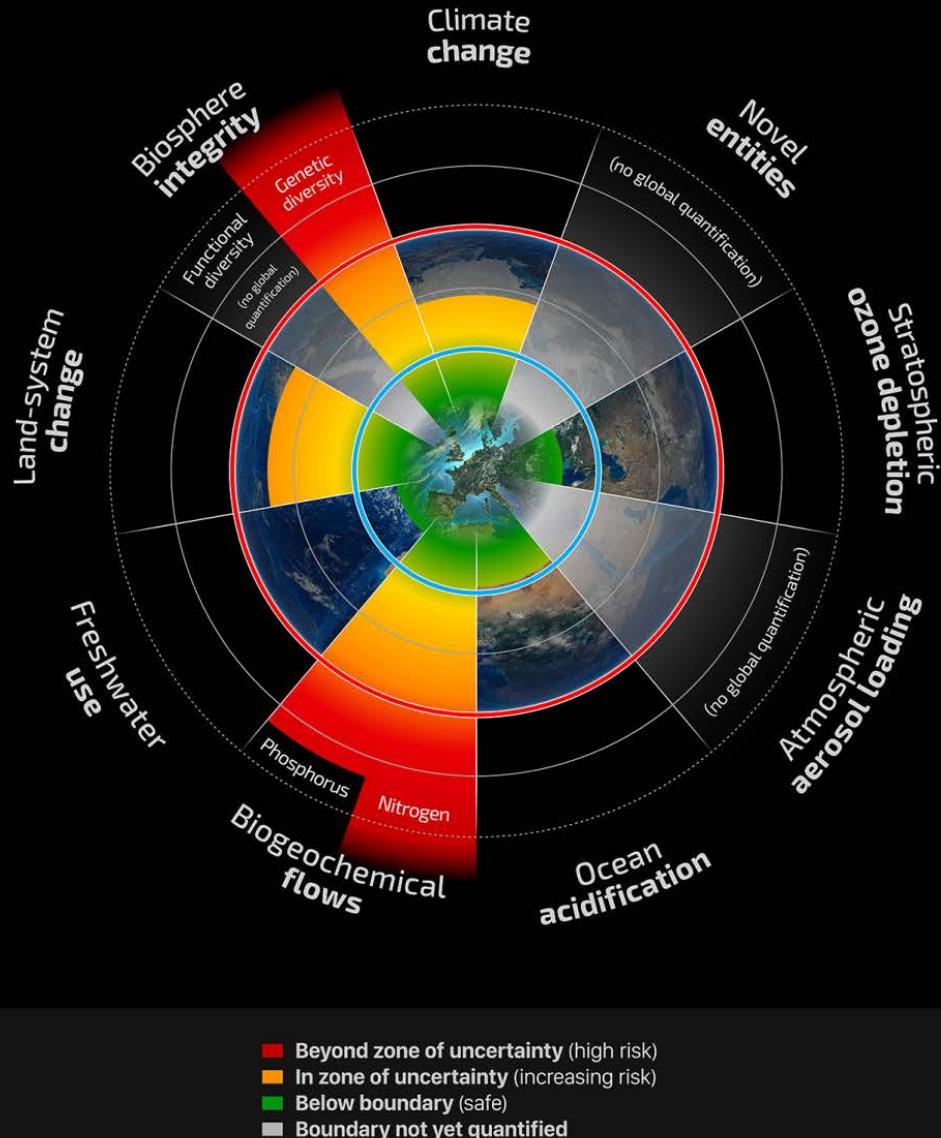
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Brno, Česko

Environmental Science Dpt.
Baylor University
Waco, Texas, USA



Planetary Boundaries

A safe operating space for humanity





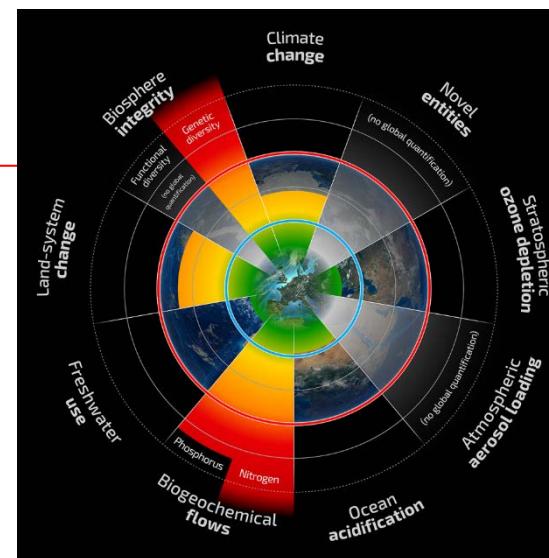
III. Stratospheric Ozone Depletion

Earth-system process	Control variable(s)	Planetary boundary (zone of uncertainty)	Current value of control variable
Stratospheric ozone depletion (R2009: same)	Stratospheric O ₃ concentration, DU	<5% reduction from pre-industrial level of 290 DU (5%–10%), assessed by latitude	Only transgressed over Antarctica in Austral spring (~200 DU)

Boundary: Average conc. of stratospheric O₃ no lower than 276 Dobson units

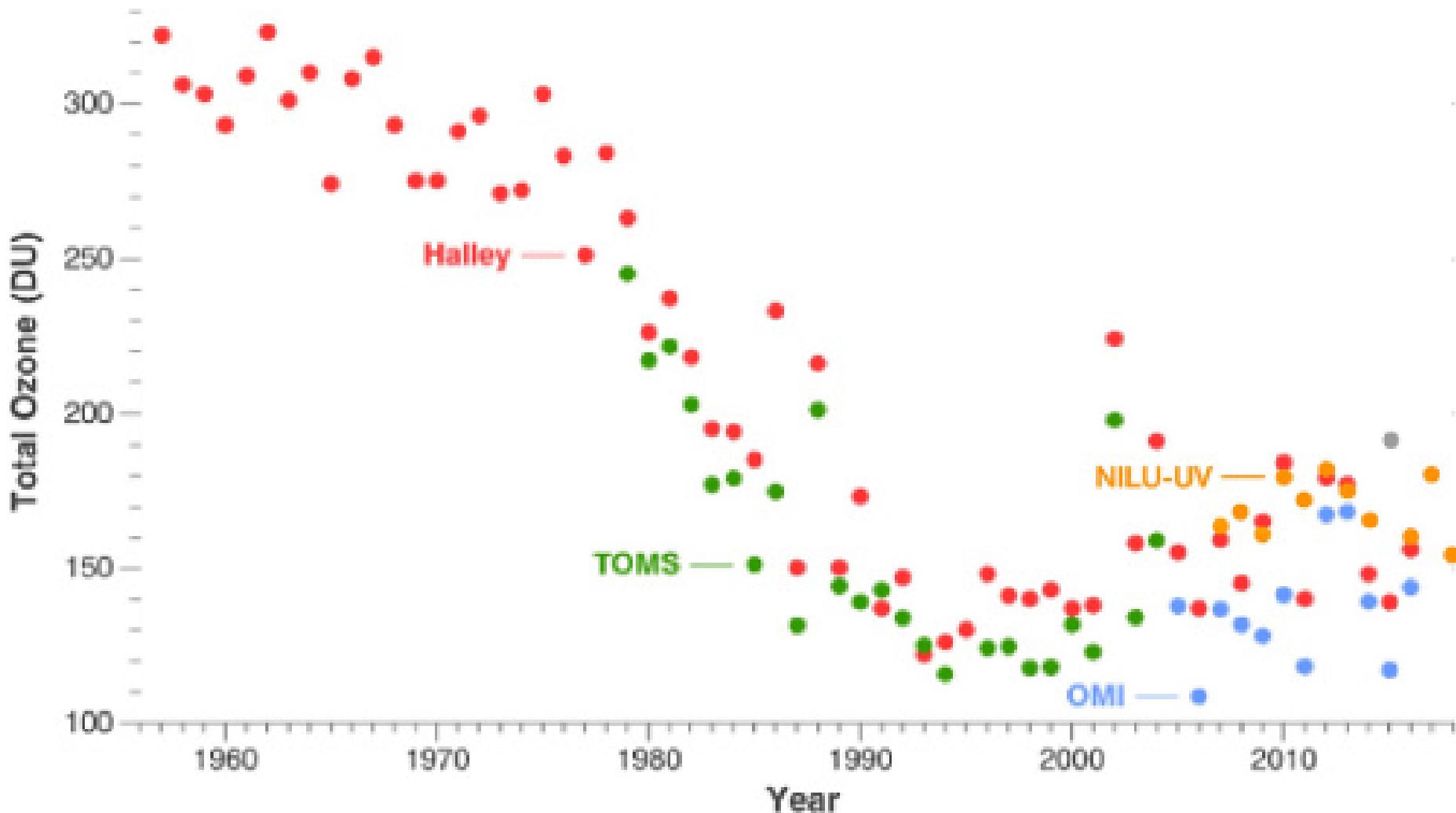
Current level: 283 Dobson units

Diagnosis: Safe, and improving





Stratospheric ozone above Antarctica



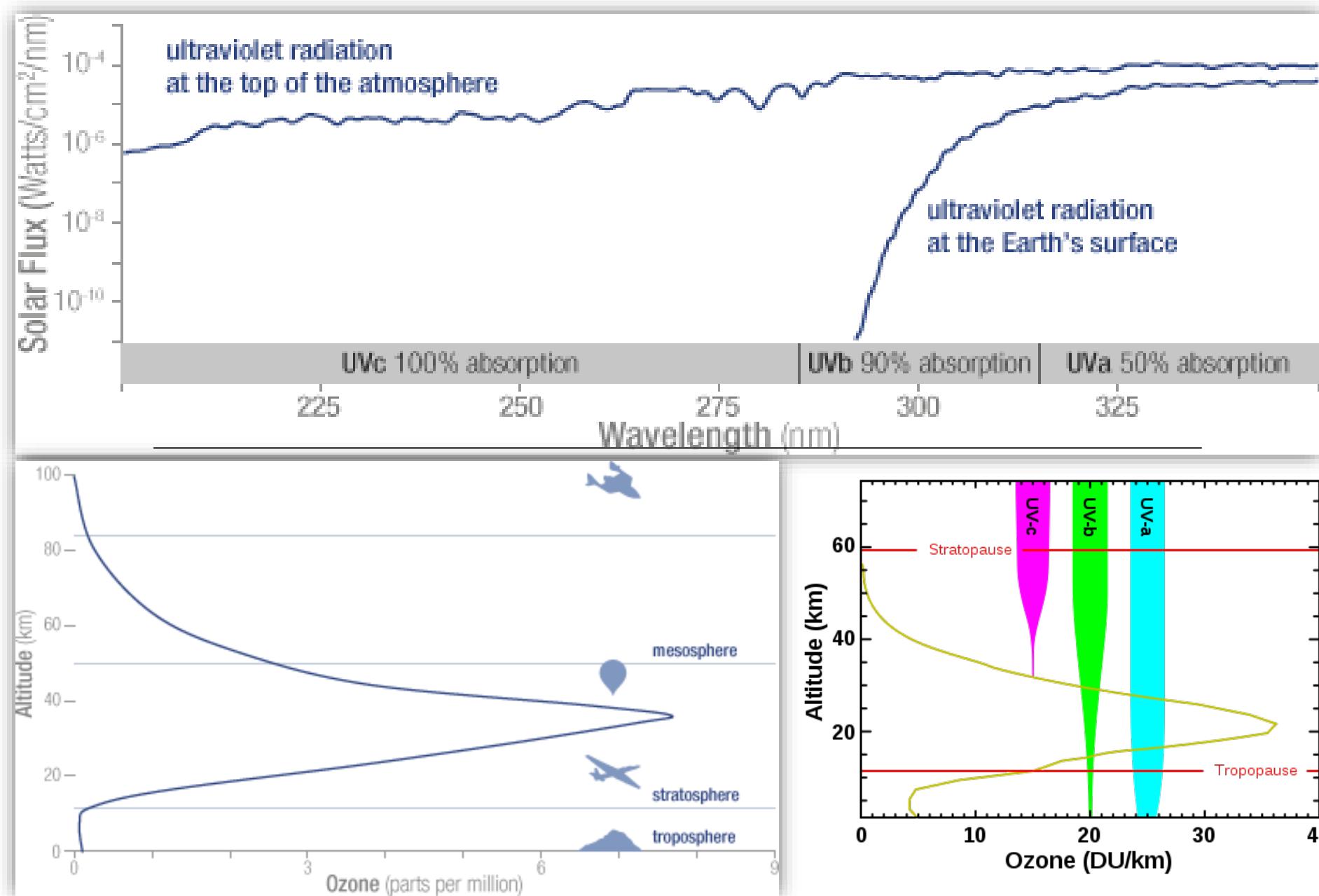


Stratospheric O₃ depletion - is it only the one environmental/health issue linked to O₃?



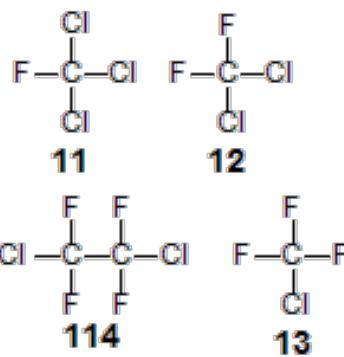
Significance of the Ozone Depletion

O₃ – protection of biosphere against harmful UVB radiation



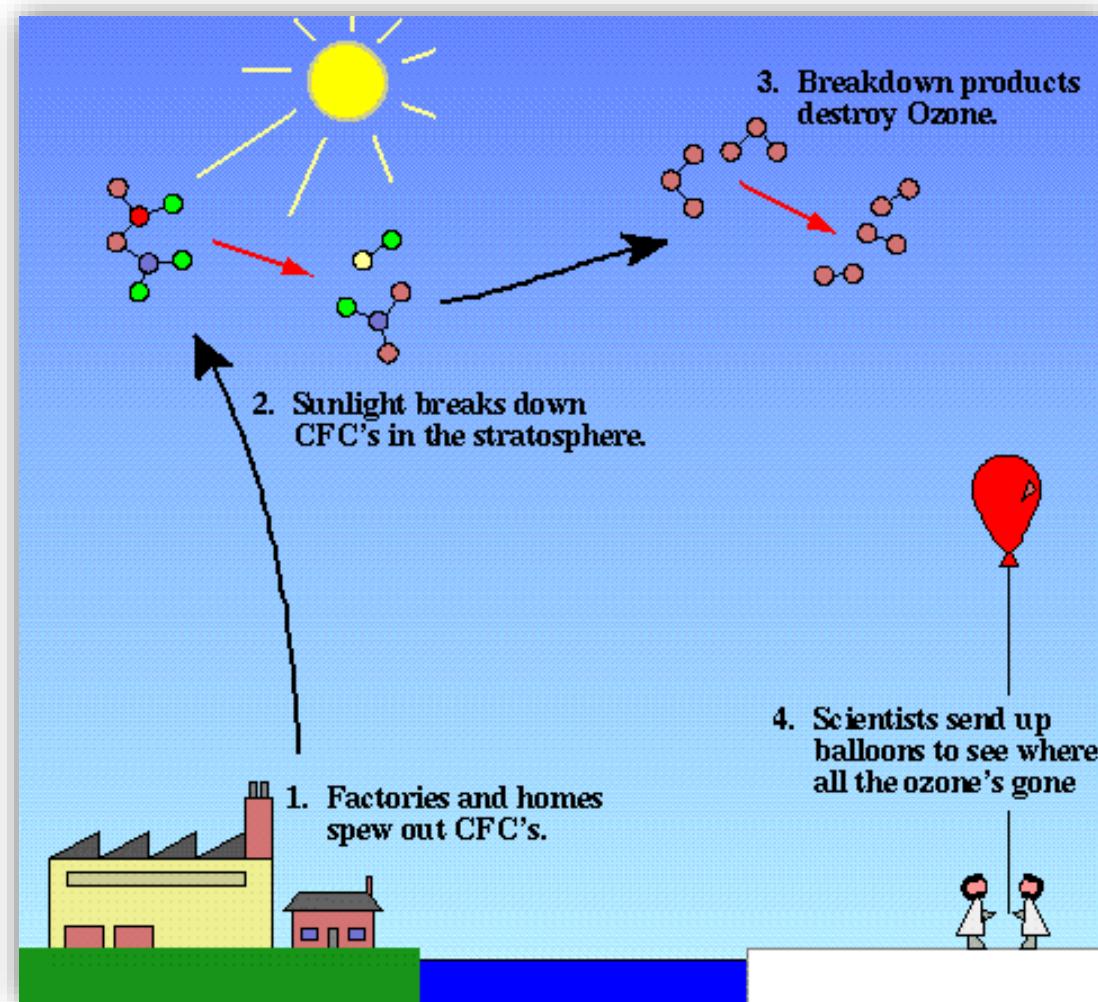
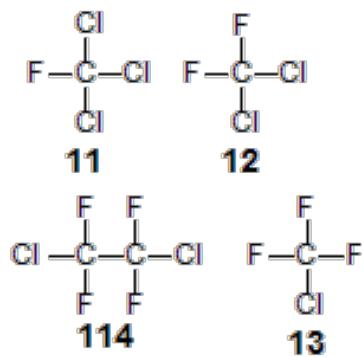
Degradation of O₃ layer

- Cl· radicals from Chloro-fluoro-carbons (CFC, Freon)
- Br· radicals from Bromo-fluoro-carbons (BFC, Halon)



Degradation of O₃ layer

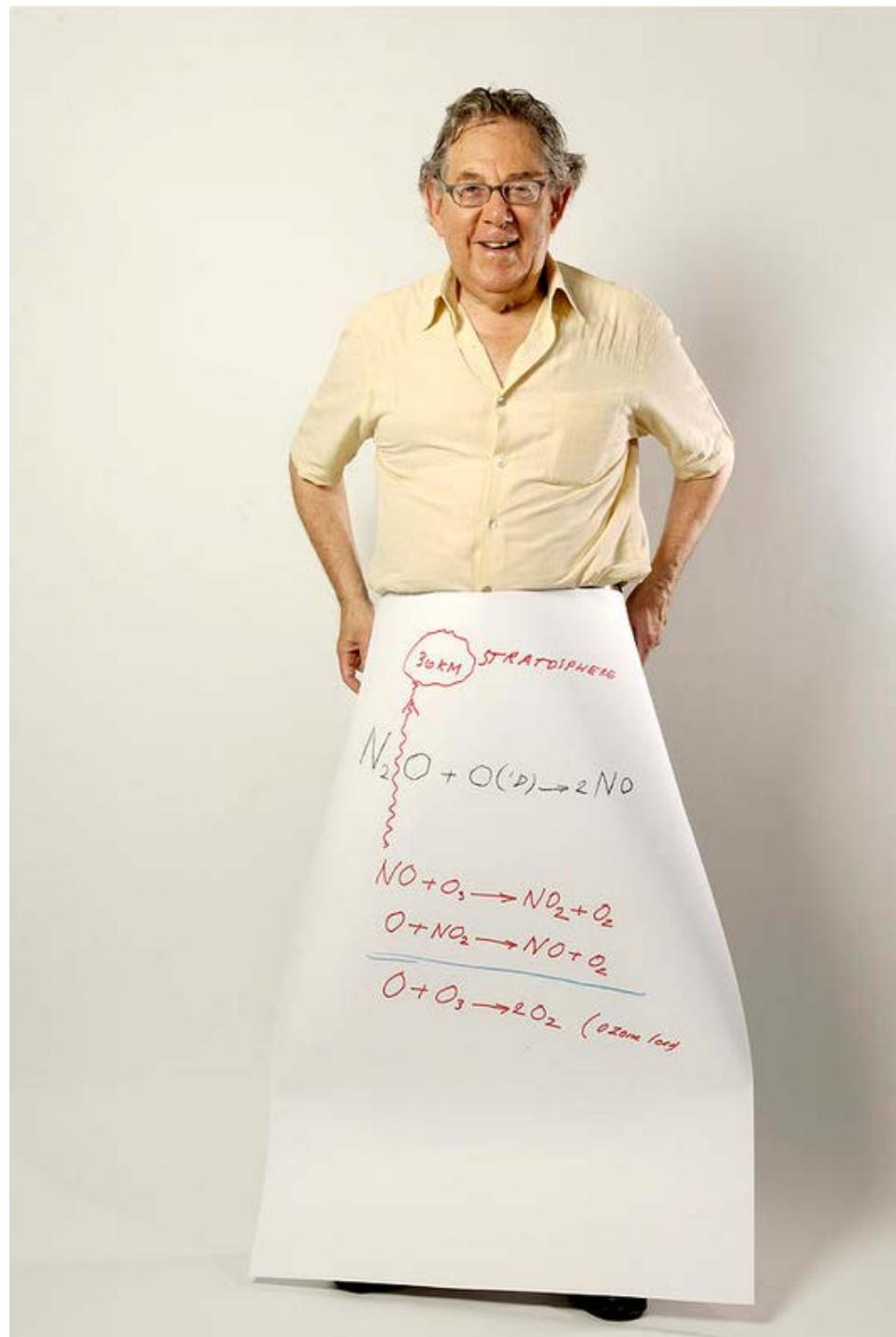
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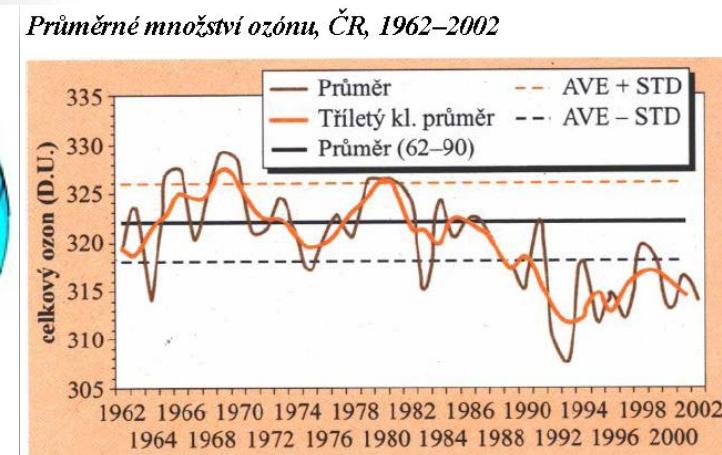
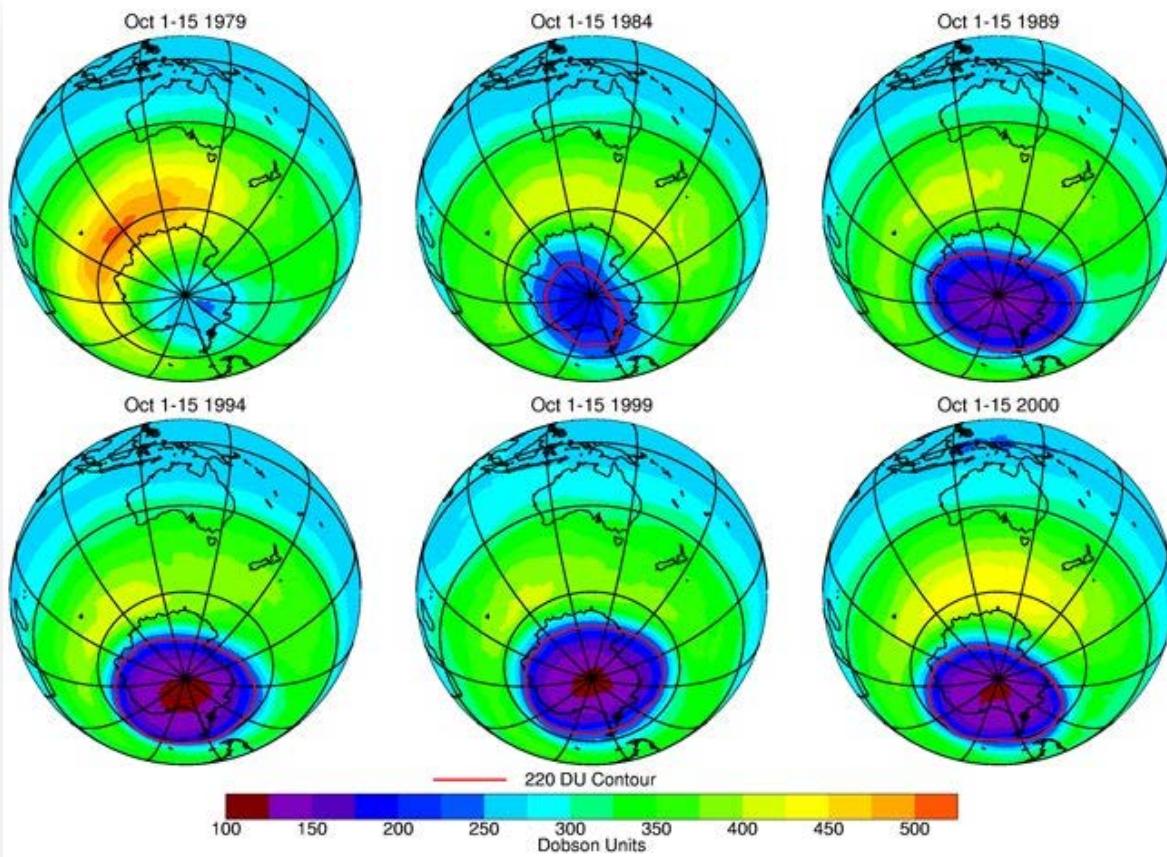
Degradation of O₃ layer

- N₂O from fertilizers



Ozone hole

- ozone depletion primarily over the South pole area
- however, significant O₃ depletion observed everywhere



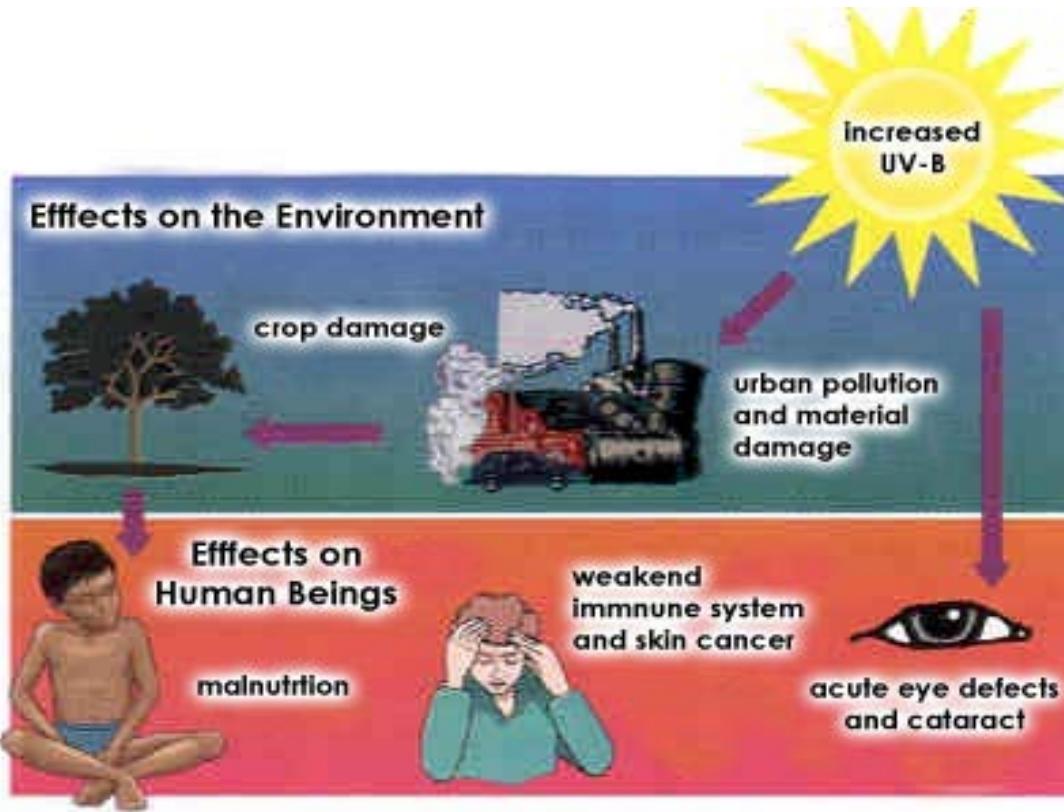
Consequences of O₃ depletion



What are the consequences of the stratospheric ozone depletion?

Less O₃ = more cancer

1% ↓ conc. O₃ ≈ 2% ↑ intensity UVB ≈ 4% ↑ skin cancer hazard



- majority of melanoms are on sunlit parts of the skin
- greatest incidence in Australia





Impact of increased UVB irradiation on crop

Possible changes in plant characteristics	Consequences	Selected sensitive crops
<ul style="list-style-type: none">■ Reduced photosynthesis■ Reduced water-use efficiency■ Enhanced drought stress sensitivity■ Reduced leaf area■ Reduced leaf conductance■ Modified flowering (either inhibited or stimulated)■ Reduced dry matter production	<p>Enhanced plant fragility</p> <p>Growth limitation</p> <p>Yield reduction</p>	<p>Rice</p> <p>Oats</p> <p>Sorghum</p> <p>Soybeans</p> <p>Beans</p>

NB: Summary conclusions from artificial exposure studies.

Source: modified from Krupa and Kickert (1989) by Runeckles and Krupa (1994) in: Fakhri Bazzaz, Wim Sombroek, *Global Climate Change and Agricultural Production*, FAO, Rome, 1996.

Ozone hole

- solution



What can we do with a stratospheric ozone depletion?

Top

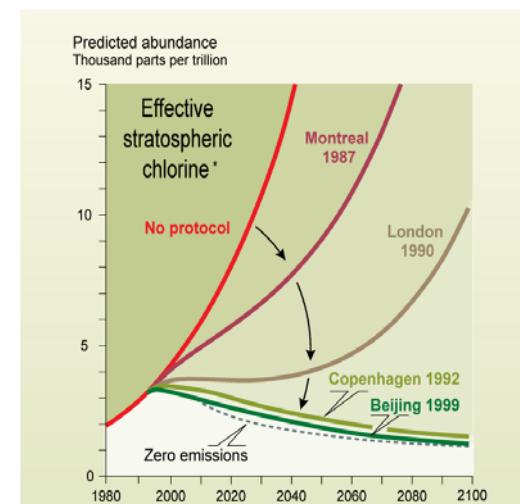




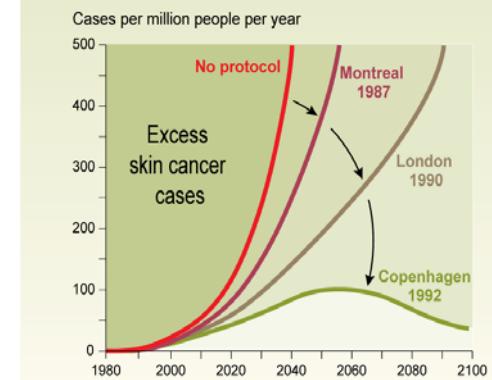
Effect of accepted solutions

1985 – Vienna Convention for the Protection of the Ozone Layer
1987 – Montreal protocol + amendments

THE EFFECTS OF THE MONTREAL PROTOCOL AMENDMENTS AND THEIR PHASE-OUT SCHEDULES



* Chlorine and bromine are the molecules responsible for ozone depletion.
"Effective chlorine" is a way to measure the destructive potential of all ODS gases emitted in the stratosphere.



Source: Twenty Questions and Answers about the Ozone Layer: 2006 Update,
Lead Author: D.W. Fahey, Panel Review Meeting for the 2006 ozone assessment.



Solutions and consequences

1985 – Vienna Convention for the Protection of the Ozone Layer
1987 – Montreal protocol + amendments

The Nobel Prize in Chemistry 1995



Paul J.
Crutzen

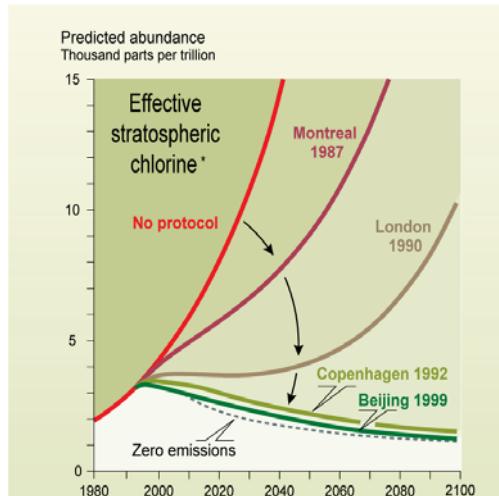


Mario J.
Molina

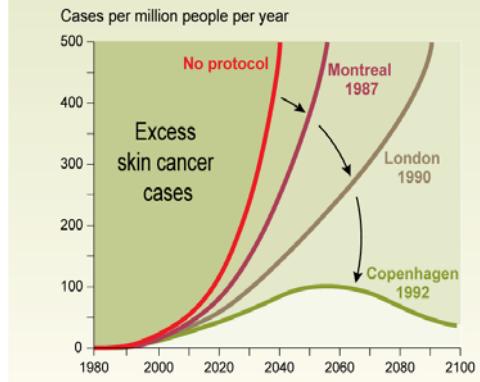


F. Sherwood
Rowland

THE EFFECTS OF THE MONTREAL PROTOCOL AMENDMENTS AND THEIR PHASE-OUT SCHEDULES



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*„for their work in atmospheric chemistry,
particularly concerning the formation and
decomposition of ozone.“*



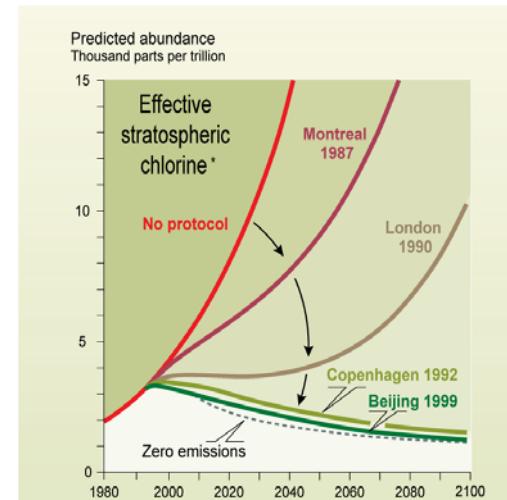
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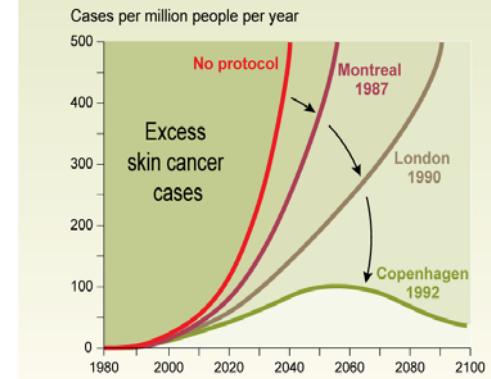
THE EFFECTS OF THE MONTREAL PROTOCOL AMENDMENTS AND THEIR PHASE-OUT SCHEDULES

Costs of CFC abandonment

- 1988-2000 – product. decreased for 90%
- overall costs of abandonment - 40 bil. \$
- no job losses
- 1/3 simply not necessary
- new HFC in cars increased car price for 50-150 \$ (prognosed 1000-1500 \$)
- CH_3Br for soil **sterilisation** replaced e.g. with crop rotation
- CH_3Br for stores **fumigation** replaced with CO_2

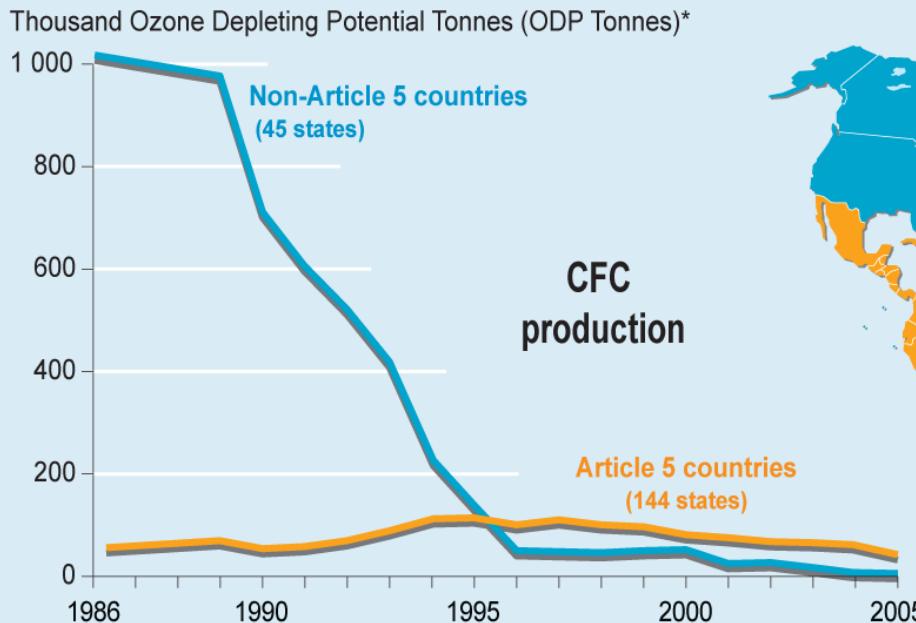


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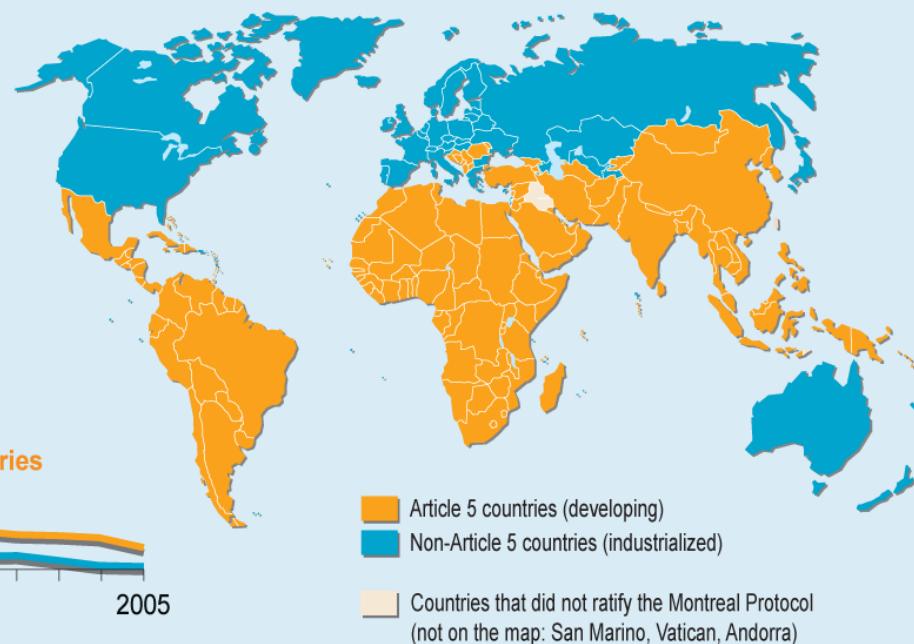
A common but differentiated responsibility



CFC
production

Article 5 countries
(144 states)

* Tonnes multiplied by the ozone depleting potential of the considered gas.



Source: United Nations Environment Programme Ozone Secretariat

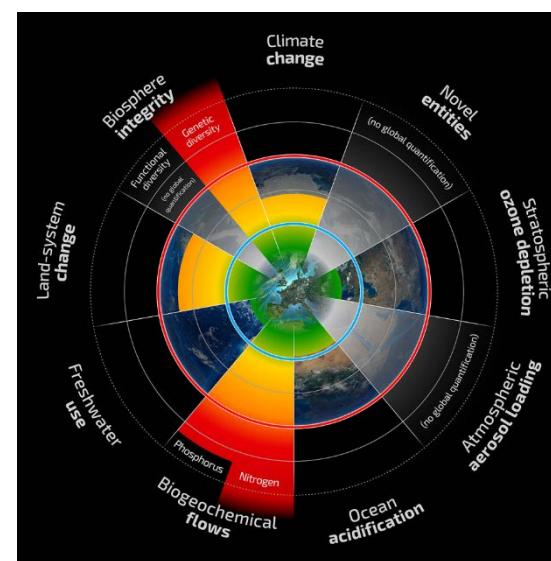
Lesson from successful solution of global issue

- cooperation of all the following stakeholders:
 - scientific discoveries and monitoring – **problem notification**
 - UNEP – **international coordinator of legal measures**
 - environmental activists – **pressure to solve the issue**
 - responsible consumers – **purchasing according to env. info**
 - technical experts - **developing env.-friendly alternatives**
 - flexible and responsible industry



IV. Ocean acidification

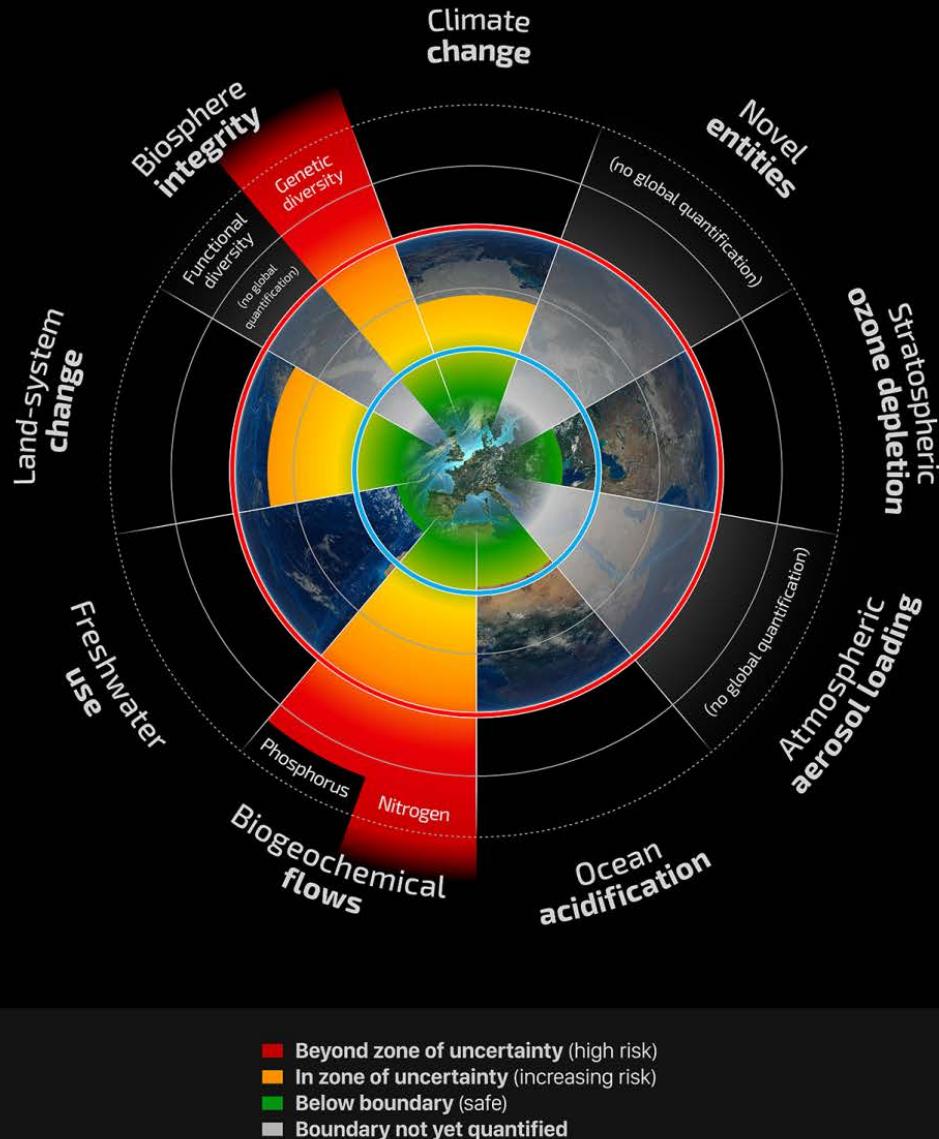
Earth-system process	Control variable(s)	Planetary boundary (zone of uncertainty)	Current value of control variable
Ocean acidification (R2009: same)	Carbonate ion concentration, average global surface ocean saturation state with respect to aragonite (Ω_{arag})	$\geq 80\%$ of the pre-industrial aragonite saturation state of mean surface ocean, including natural diel and seasonal variability ($\geq 80\% - \geq 70\%$)	$\sim 84\%$ of the pre-industrial aragonite saturation state





Planetary Boundaries

A safe operating space for humanity





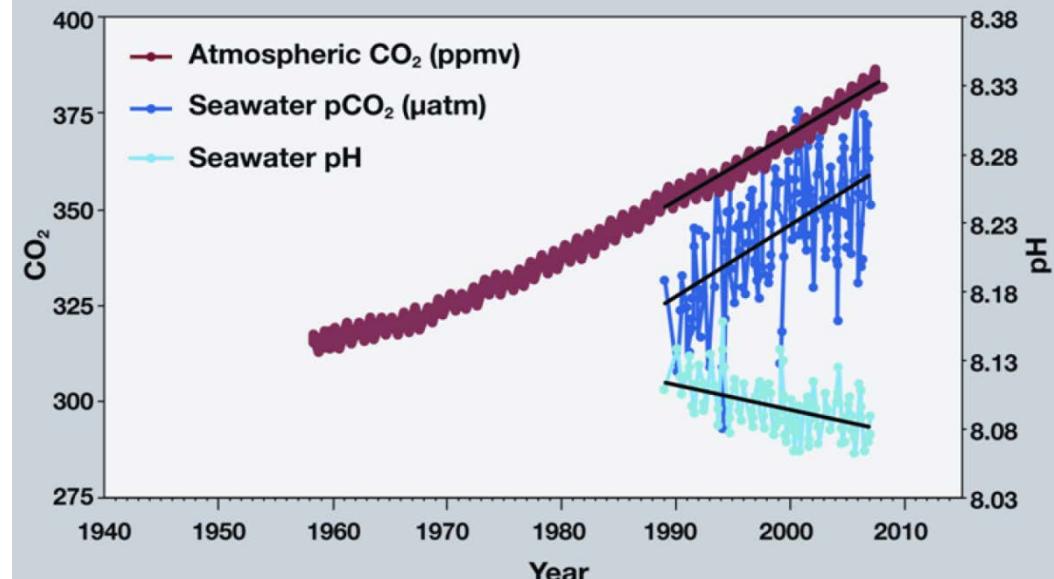
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Ocean acidification

- what is the cause?



OCEAN ACIDIFICATION

HOW WILL CHANGES IN OCEAN CHEMISTRY AFFECT MARINE LIFE?

CO₂ absorbed from the atmosphere

$\text{CO}_2 + \text{H}_2\text{O} + \text{CO}_3^{2-} \rightarrow 2 \text{HCO}_3^-$

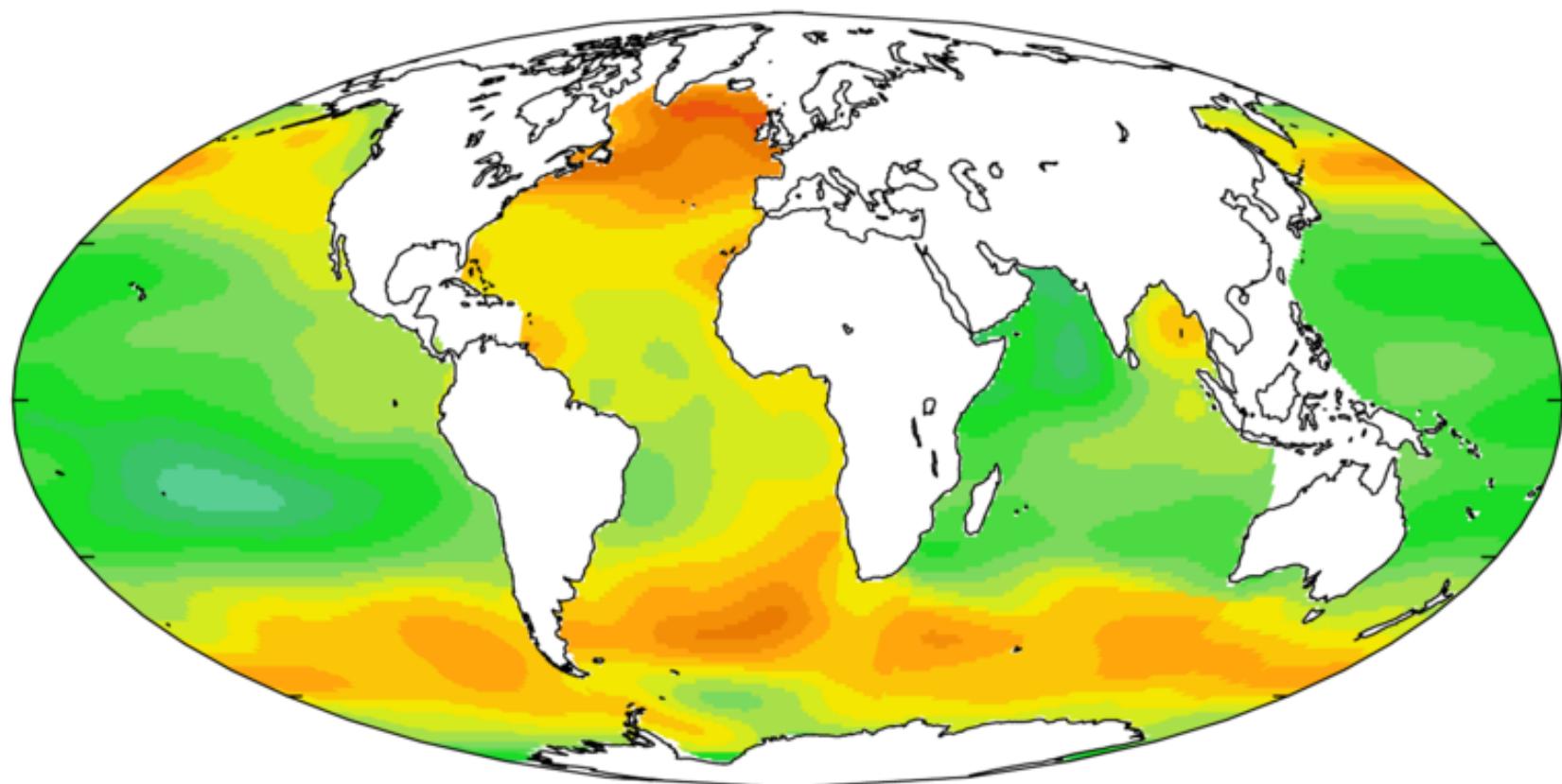
carbon dioxide water carbonate ion 2 bicarbonate ions

consumption of carbonate ions impedes calcification

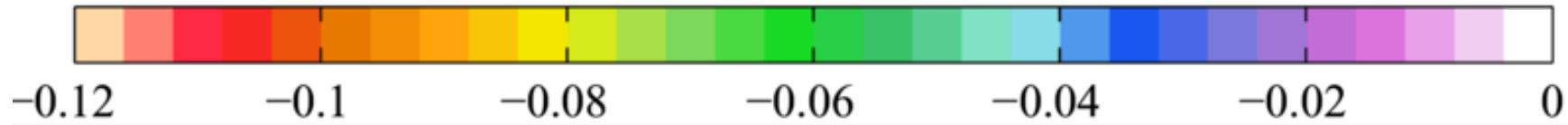
This section provides a detailed look at the chemical process of ocean acidification. It shows CO₂ being absorbed from the atmosphere into the ocean. The chemical reaction is: $\text{CO}_2 + \text{H}_2\text{O} + \text{CO}_3^{2-} \rightarrow 2 \text{HCO}_3^-$. The diagram illustrates the conversion of carbon dioxide, water, and carbonate ions into two bicarbonate ions. Below the reaction, it states that the consumption of carbonate ions impedes calcification. The right side of the slide features two images: a microscopic view of marine life and a scanning electron micrograph (SEM) of a shell fragment, which appears textured and eroded, representing the physical effects of acidification on marine organisms.



Change in pH of oceans 1700-2000



Δ sea-surface pH [-]





„Natural laboratory“

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Natural lab shows sea's acid path

By Richard Black

Environment correspondent, BBC News website



Scientists study conditions at the bottom of the Mediterranean Sea

Natural carbon dioxide vents on the sea floor are showing scientists how carbon emissions will affect marine life.

Dissolved CO₂ makes water more acidic, and around the vents, researchers saw a fall in species numbers, and snails with their

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26 March 2014 Last updated at 23:03 GMT

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How climate change will acidify the oceans

By Roger Harrabin

BBC environment analyst, Normanby Island



Off the remote eastern tip of Papua New Guinea a natural phenomenon offers an alarming glimpse into the future of the oceans, as increasing concentrations of CO₂ in the atmosphere make sea water more acidic.

Streams of volcanic CO₂ bubbles emerge from deep under the seabed here, like a giant jacuzzi.

As the bubbles of carbon dioxide dissolve into the water, carbonic acid is

In today's Magazine

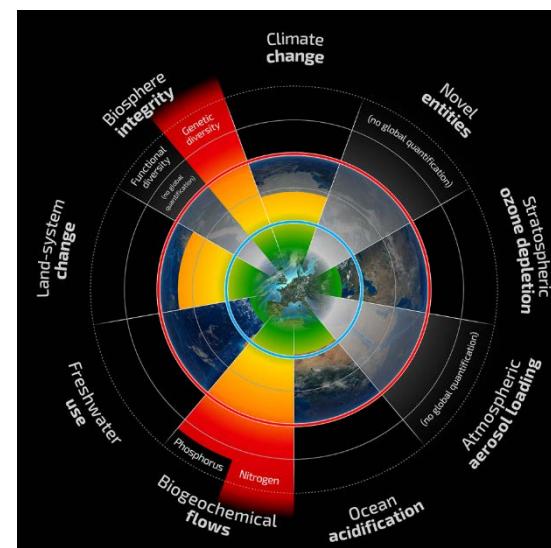
One lonely man and his hoard of Nazi art

Malaysia plane: 10 questions that are still unresolved



V and VI. Biogeochemical flows of P and N

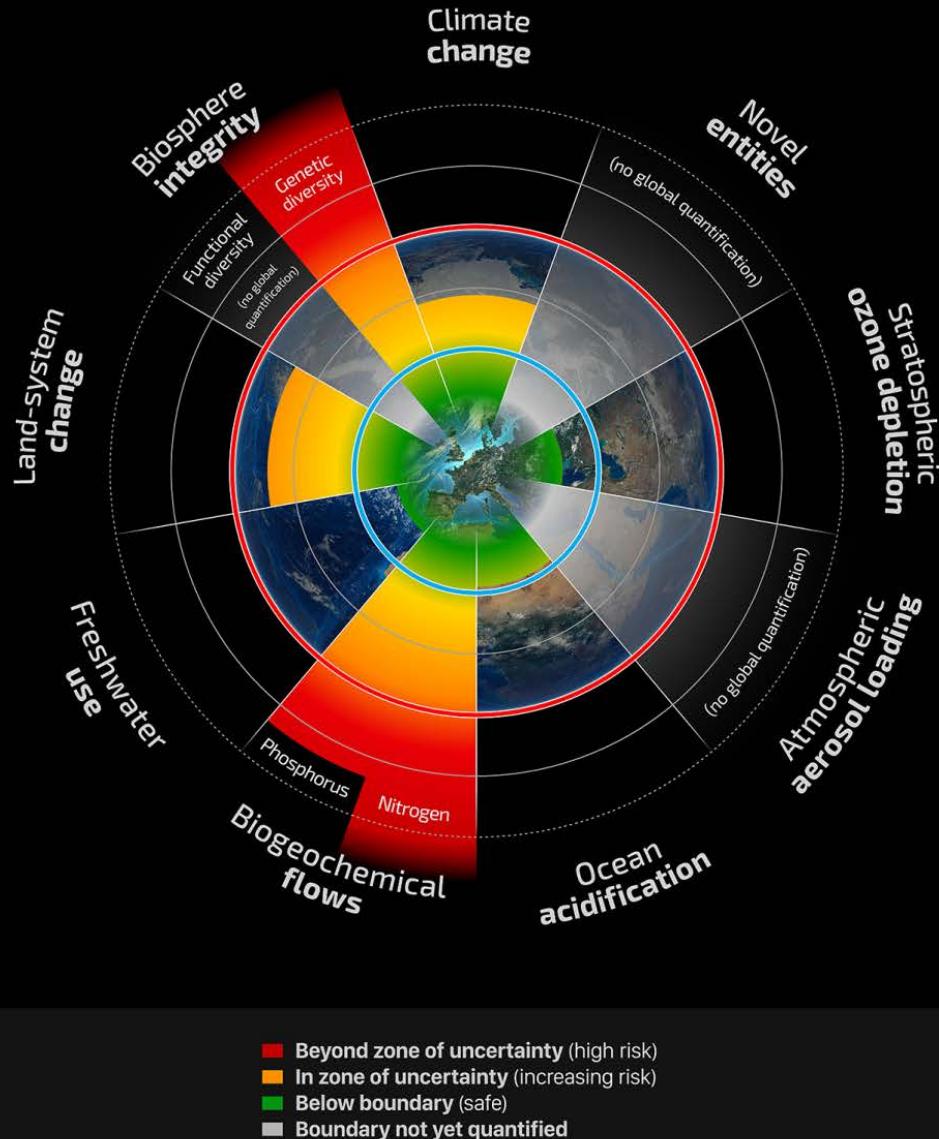
Earth-system process	Control variable(s)	Planetary boundary (zone of uncertainty)	Current value of control variable
Biogeochemical flows: (P and N cycles) (R2009:	<i>P Global</i> : P flow from freshwater systems into the ocean	11 Tg P yr ⁻¹ (11–100 Tg P yr ⁻¹)	~22 Tg P yr ⁻¹
Biogeochemical flows: (interference with P and N cycles))	<i>P Regional</i> : P flow from fertilizers to erodible soils	6.2 Tg yr ⁻¹ mined and applied to erodible (agricultural) soils (6.2-11.2 Tg yr ⁻¹). Boundary is a global average but regional distribution is critical for impacts.	~14 Tg P yr ⁻¹
	<i>N Global</i> : Industrial and intentional biological fixation of N	62 Tg N yr ⁻¹ (62–82 Tg N yr ⁻¹). Boundary acts as a global 'valve' limiting introduction of new reactive N to Earth System, but regional distribution of fertilizer N is critical for impacts.	~150 Tg N yr ⁻¹



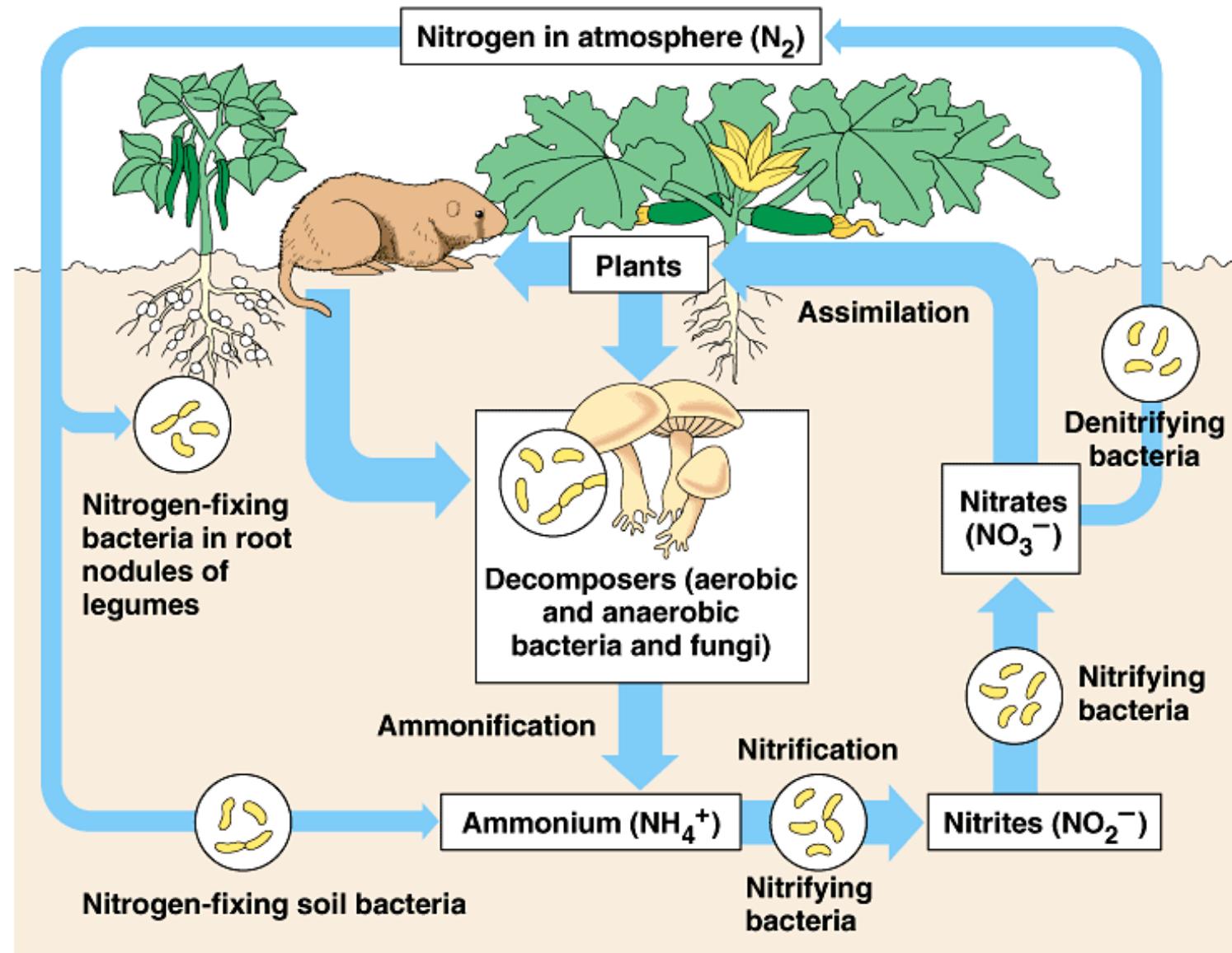


Planetary Boundaries

A safe operating space for humanity



N - Nitrogen – natural geochemical cycle

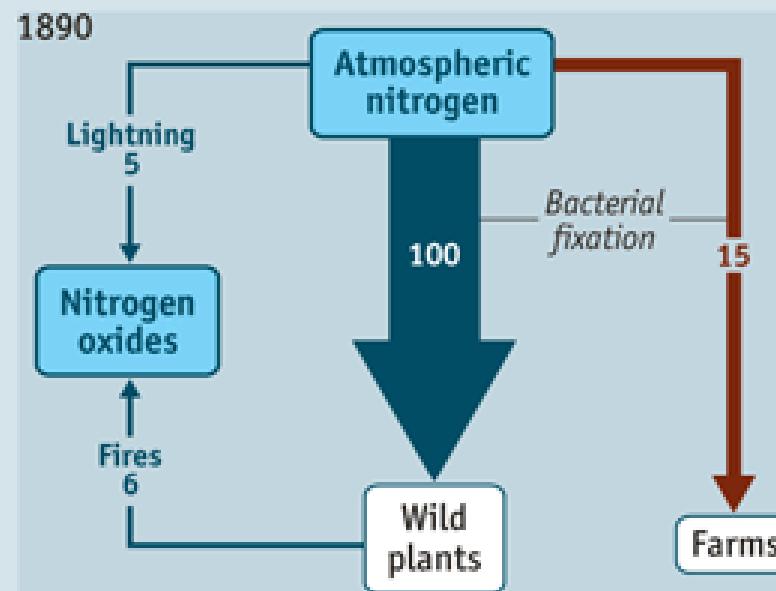




N - Nitrogen

Unbalancing the cycle

Nitrogen flows, megatonnes



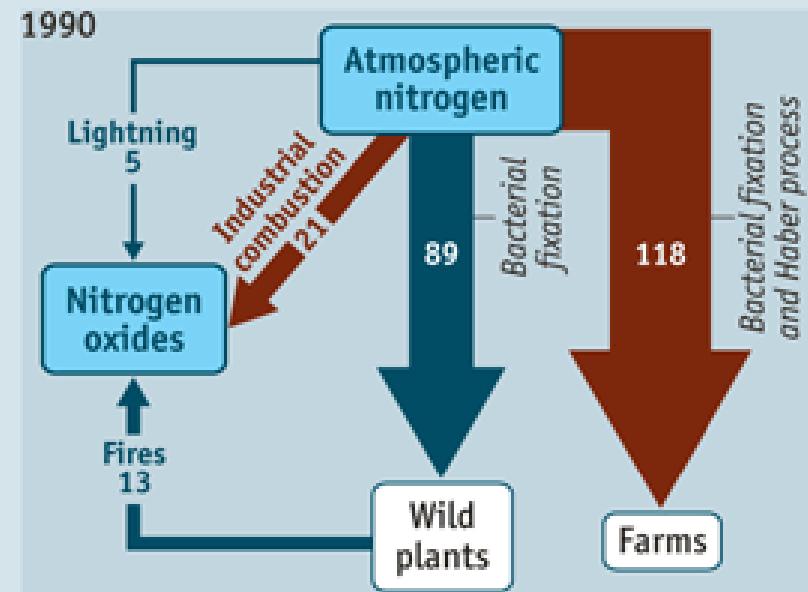
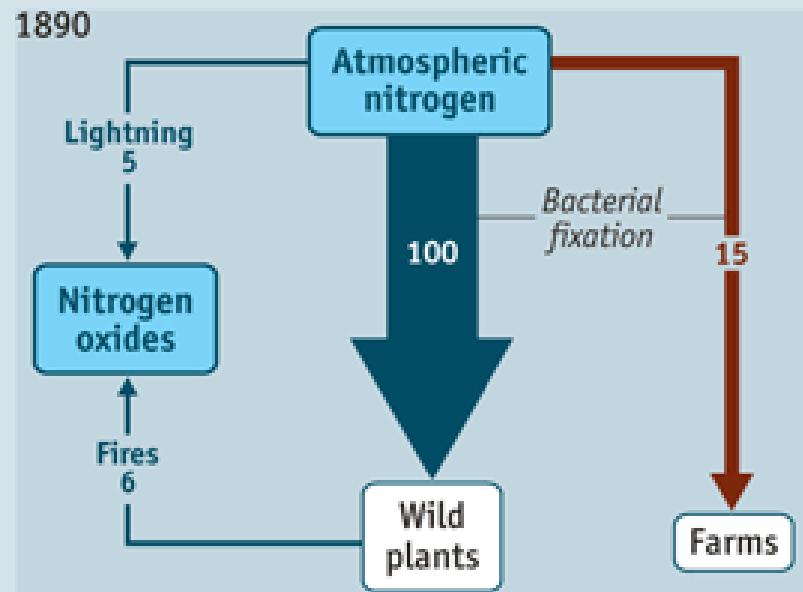
Source: Galloway and Cowling, *Ambio*

N - Nitrogen

- today, human activity changes more N_2 to reactive forms of N than all terrestrial processes together
- Haber-Bosch 80 Mt_N/yr, leguminosis 40 Mt_N/yr, fossil fuels combustion 20 Mt_N/yr, biomass combustion 10 Mt_N/yr

Unbalancing the cycle

Nitrogen flows, megatonnes





N - Nitrogen

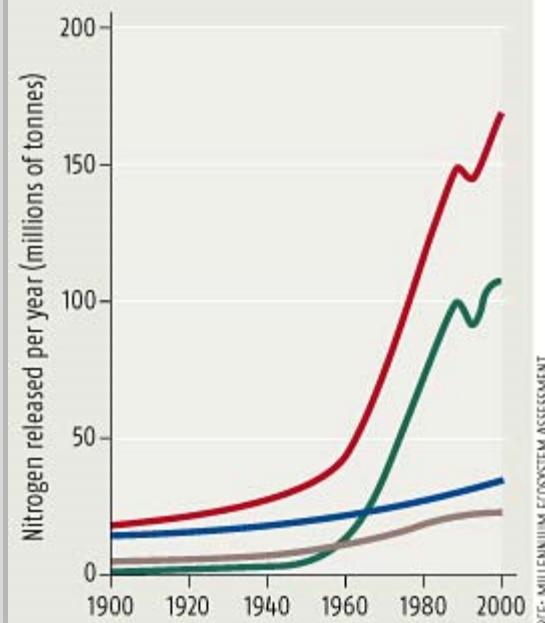
- major reason of N_2 fixation ?



NITROGEN POLLUTION

The amount of reactive nitrogen released into the environment is increasing

- Total human input
- Fertiliser and industrial uses
- Nitrogen fixation in agri-ecosystems
- Fossil fuels





N - Nitrogen

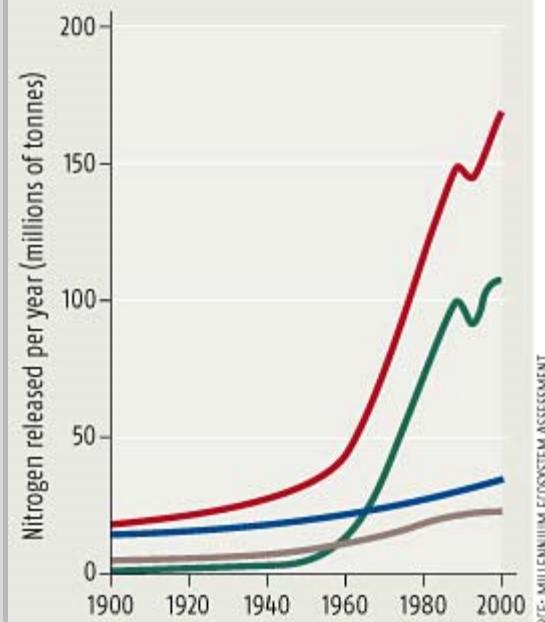
- major reason of N_2 fixation ?
- N-fertilizers
- significant part ends in water – eutrophication and nitrates issue
- significant part ends in atmosphere N_2O is GHG and O_3 decomp.
- overall decrease of resilience of planetary systems thanks to high input of reactive nitrogen molecules



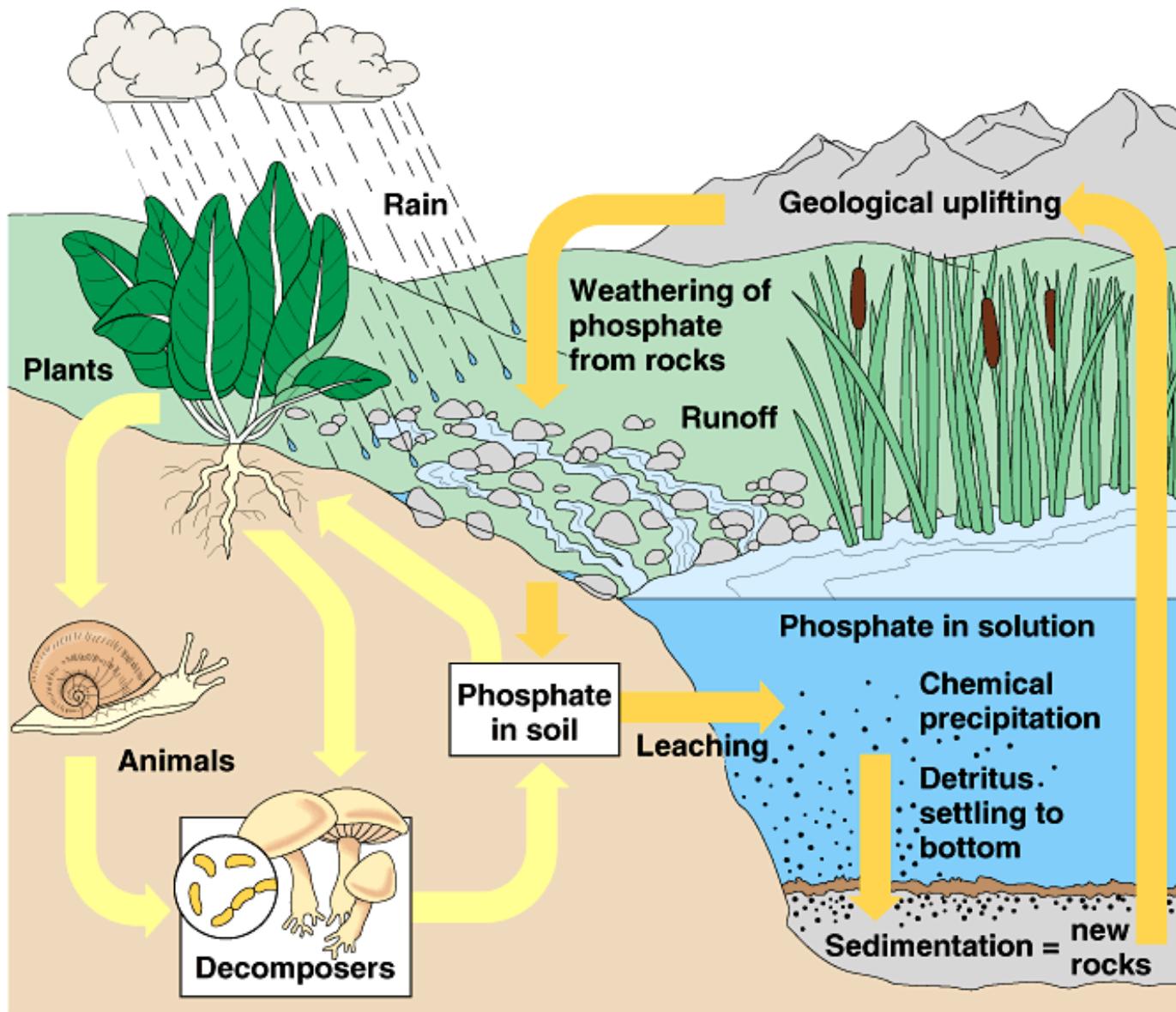
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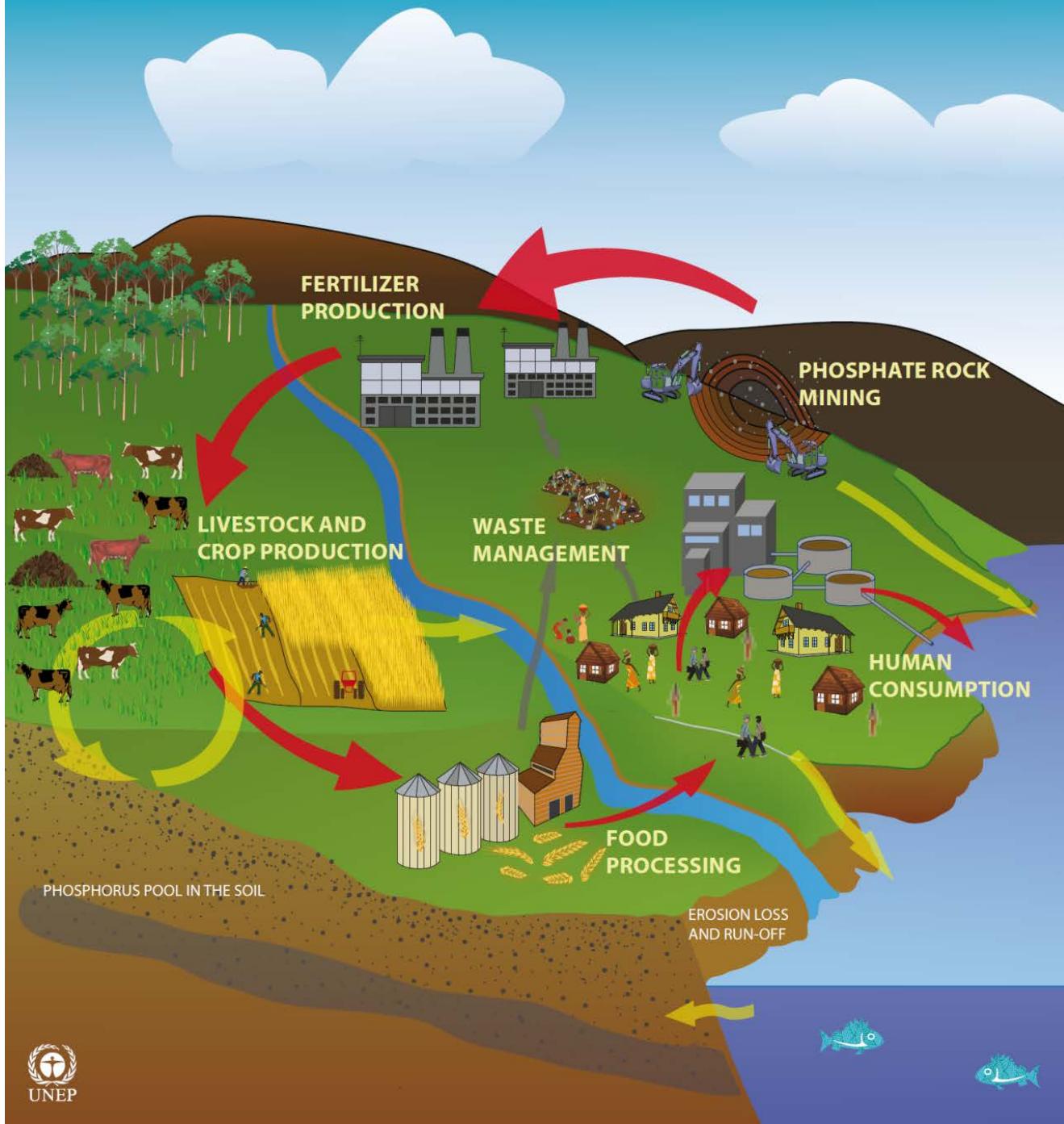
- Total human input
- Fertiliser and industrial uses
- Nitrogen fixation in agri-ecosystems
- Fossil fuels



P – phosphorus – natural geochemical cycle

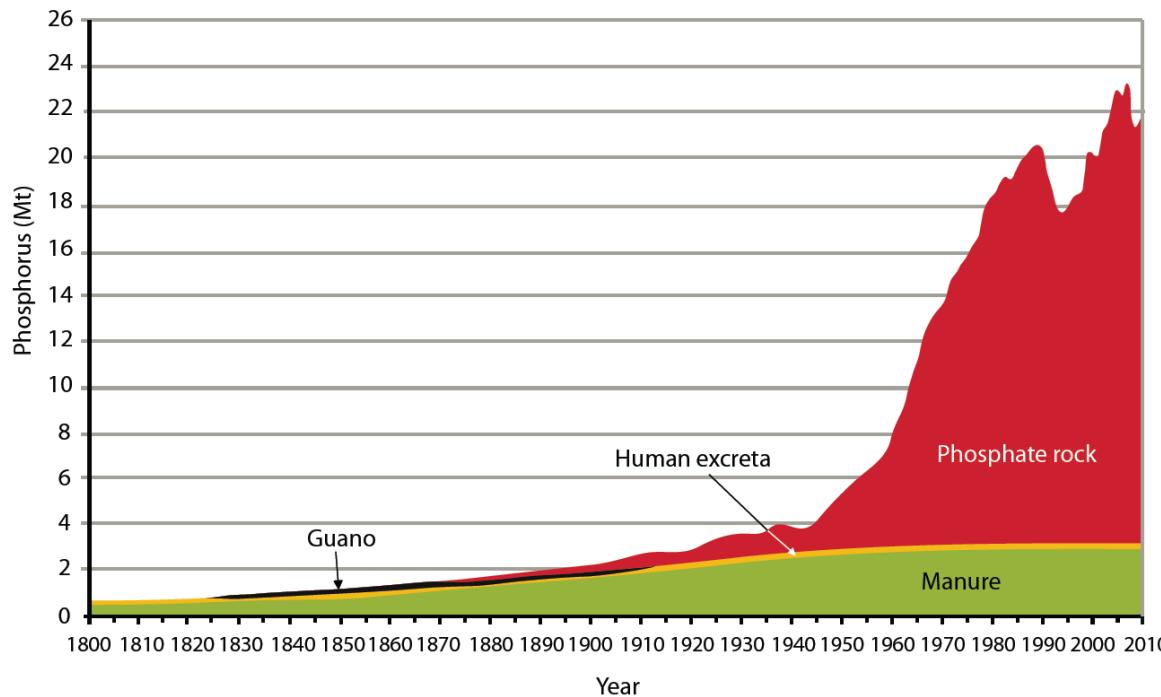


Human intervention to P cycle



P - phosphorus

- primary source - **weathering or apatite mining**
- anthropogenic flow to oceans - 8-9x higher amount
- from 20 Mt_N/yr industr. P – half ends in oceans
- higher risk of **anoxic events**



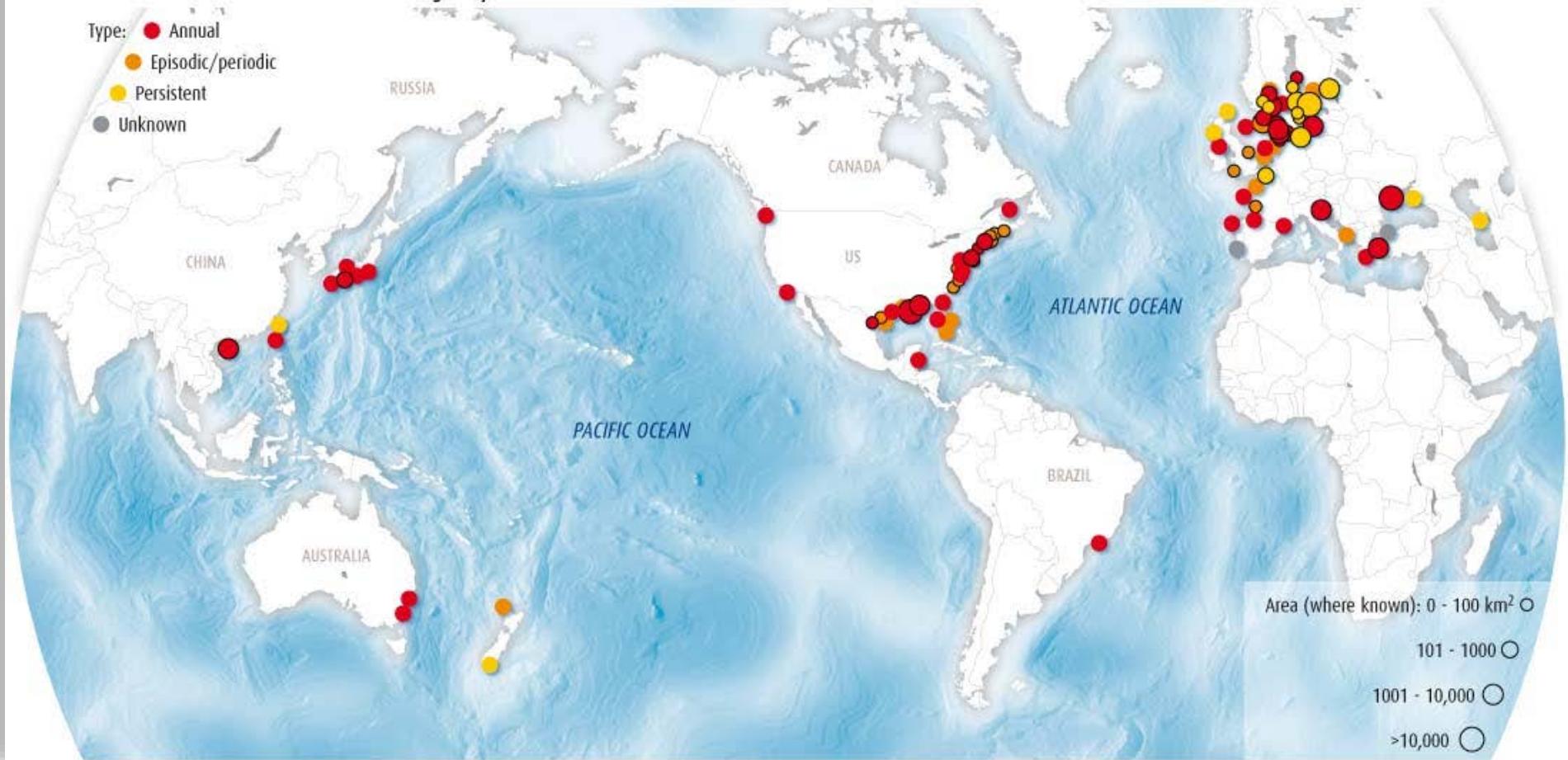


P + N = anoxic zones in oceans

200 AND COUNTING

The number of dead zones around the world is doubling every decade

- Type:
- Annual
 - Episodic/periodic
 - Persistent
 - Unknown





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Deadly Trio in the oceans

My New Scientist

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The oceans are heating, acidifying and choking

› 19:58 04 October 2013 by [Fred Pearce](#)
› For similar stories, visit the [Climate Change](#) Topic Guide

We know the oceans are warming. We know they are acidifying. And now, to cap it all, it turns out they are suffocating, too. A new health check on the state of the oceans warns that they will have lost as much as 7 per cent of their oxygen by the end of the century.

The cascade of chemical and biological changes now under way could see coral reefs irreversibly destroyed in 50 to 100 years, with marine ecosystems increasingly taken over by [jellyfish](#) and toxic algal blooms.

The [review](#) is a repeat of a study two years ago by the [International Programme on the State of the Ocean](#) (IPSO), a coalition of scientists. It concludes that things have become worse since the first study.

"The health of the oceans is spiralling downwards far more rapidly than we had thought, exposing organisms to intolerable and unpredictable evolutionary pressure," says [Alex Rogers](#) at the University of Oxford, the scientific director of IPSO.

Deadly trio

Rogers describes a "deadly trio" of linked global threats. The first is global warming: surface sea water has been [warming](#) almost as fast as the atmosphere. The second is [acidification](#) – a result of the water absorbing ever more CO₂ from the atmosphere. The third is [deoxygenation](#).



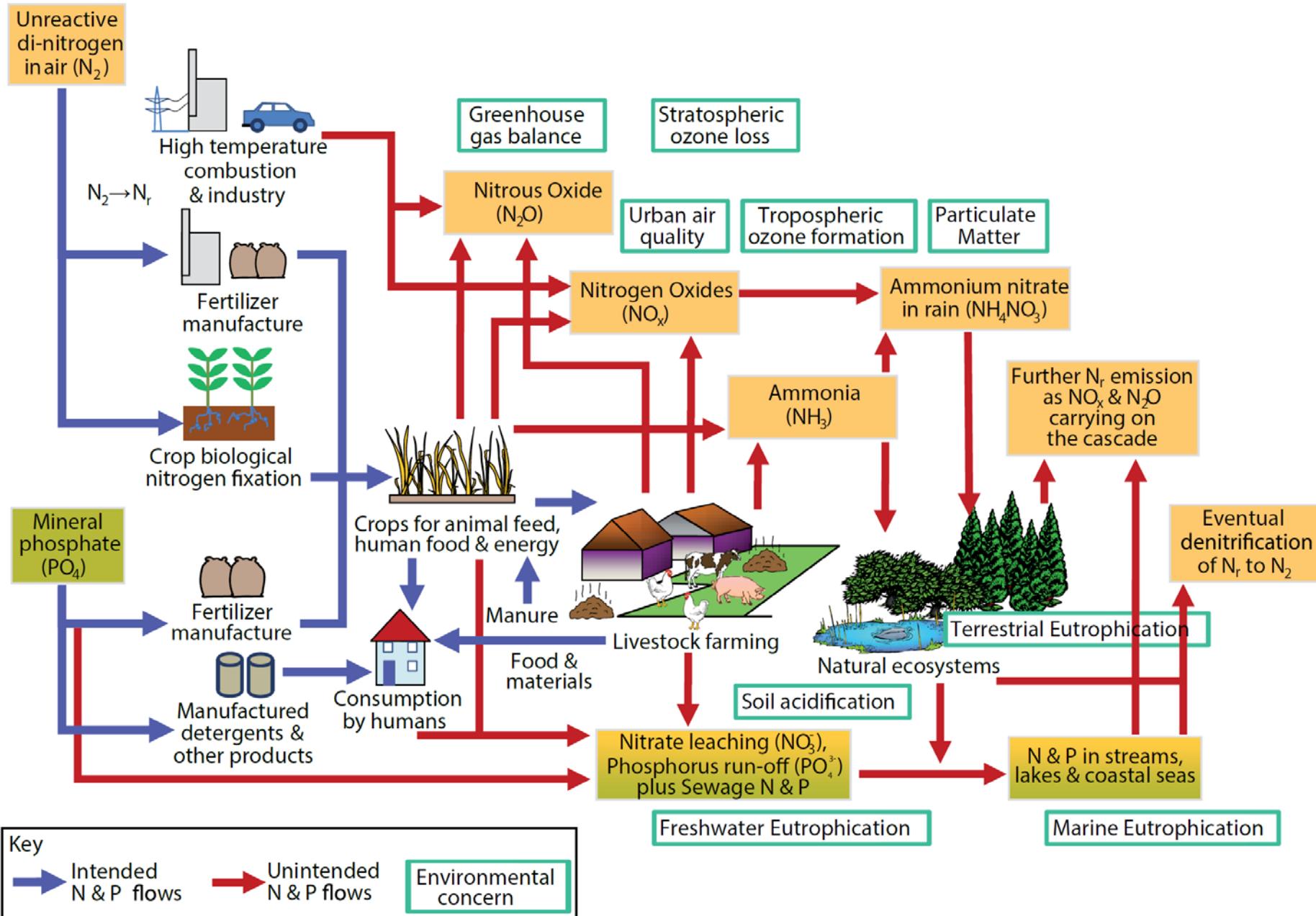
Getting harder to breathe underwater (*Image: Incredible Features/Barcroft Media*)

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[Hyundai i40 2013, 1.7 CRDI](#)



Simplified view of the nitrogen and phosphate cascade





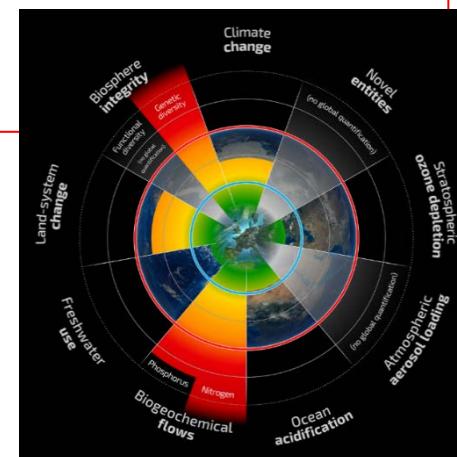
VII. Global freshwater consumption

Earth-system process	Control variable(s)	Planetary boundary (zone of uncertainty)	Current value of control variable
Freshwater use (R2009: Global freshwater use)	<p><i>Global:</i> Maximum amount of consumptive blue water use ($\text{km}^3 \text{yr}^{-1}$)</p> <p><i>Basin:</i> Blue water withdrawal as % of mean monthly river flow</p>	<p><i>Global:</i> $4000 \text{ km}^3 \text{ yr}^{-1}$ ($4000\text{--}6000 \text{ km}^3 \text{ yr}^{-1}$)</p> <p><i>Basin:</i> Maximum monthly withdrawal as a percentage of mean monthly river flow. For low-flow months: 25% (25–55%); for intermediate-flow months: 30% (30–60%); for high-flow months: 55% (55–85%)</p>	$\sim 2600 \text{ km}^3 \text{ yr}^{-1}$

Boundary: No more than 4000 km^3 of fresh water consumed per year

Current level: 2600 km^3 per year

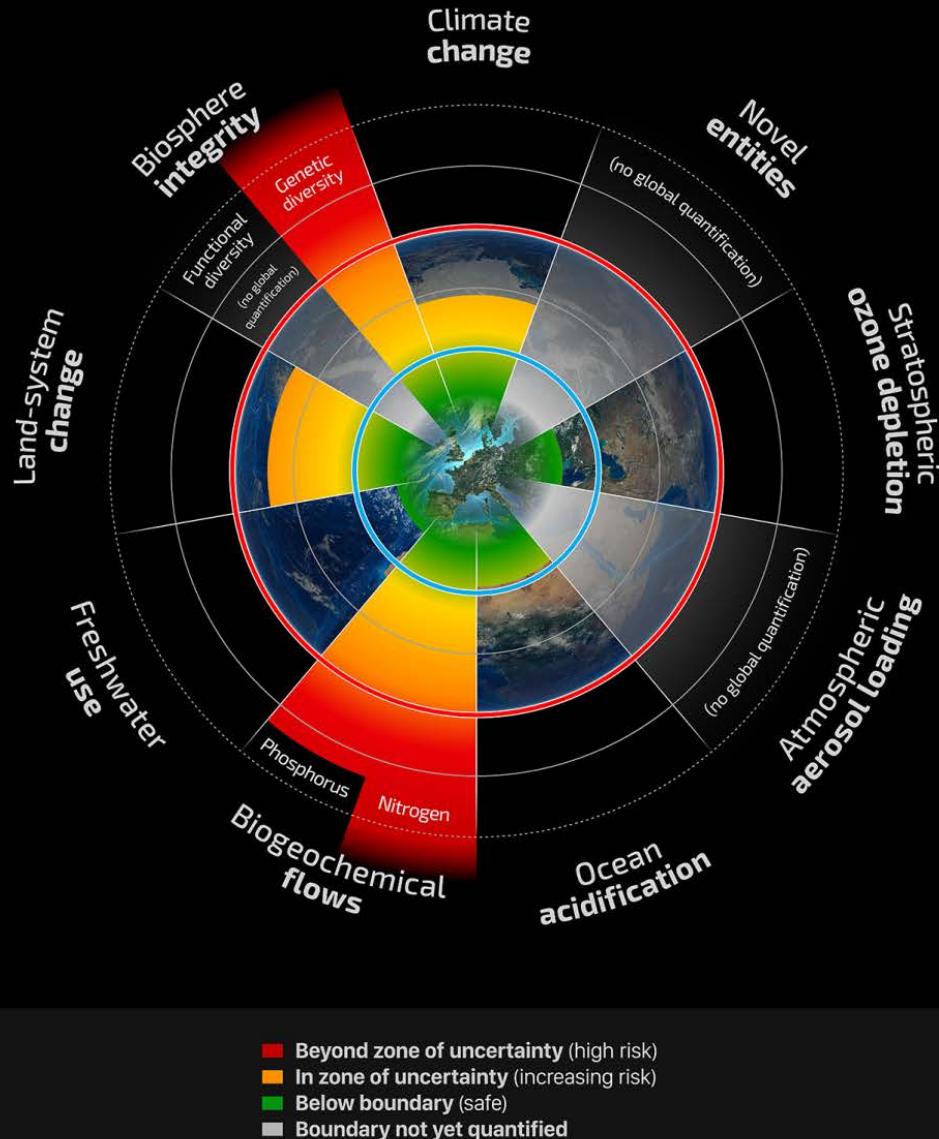
Diagnosis: Boundary will be approached by mid-century





Planetary Boundaries

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FW issue

- People are a dominant force changing **flow of water in rivers**
- cca 25 % water does not reach the ocean
- consequences for the biodiversity, nutrition, aquatic and terestrial ecosystems



8 Mighty Rivers Run Dry From Overuse

Main About the Freshwater Initiative Restoring Rivers Reducing Water Use News Videos

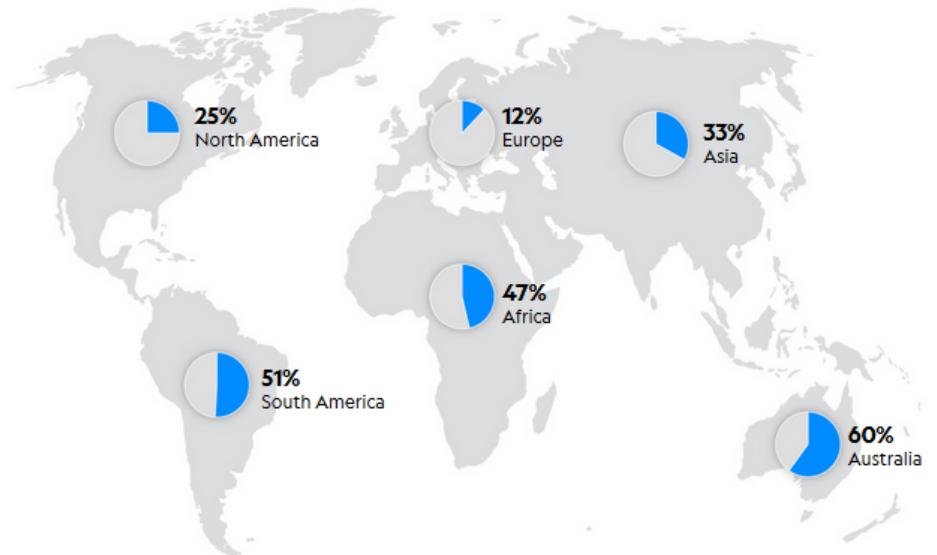




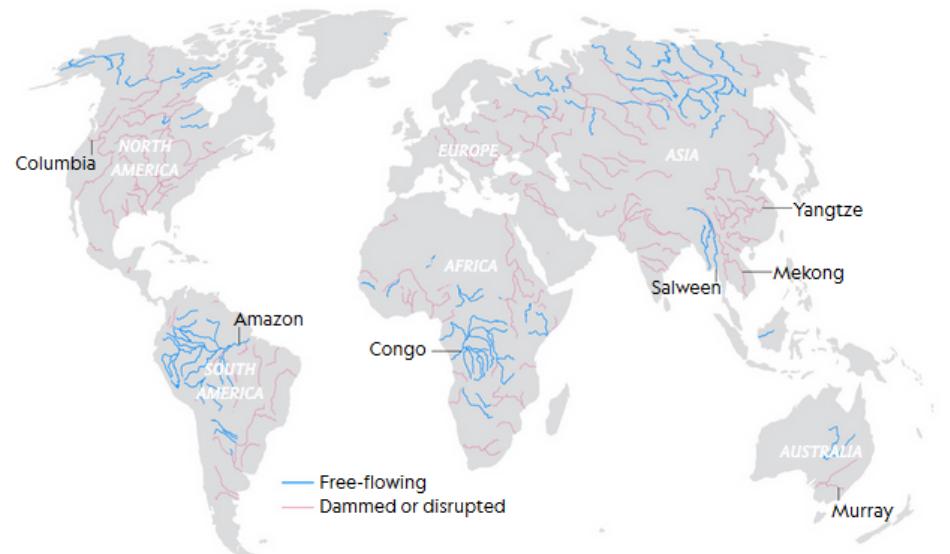
The world's remaining free-flowing rivers

Only 37 percent of world's largest rivers are free of dams or other disruptions. Free-flowing rivers are found primarily in the Amazon and Congo Basins, and in the Arctic.

Percentage of very large rivers (longer than 1,000 km) that remain free-flowing, by continent



Distribution of very large rivers

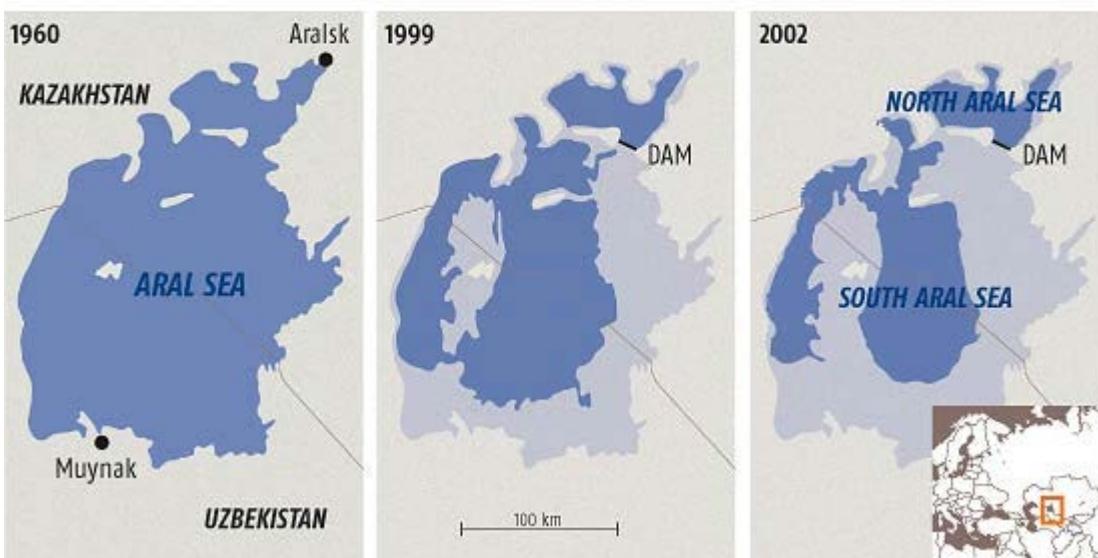


Aral See - Kazakhstan, Uzbekistan



THE SHRINKING SEA

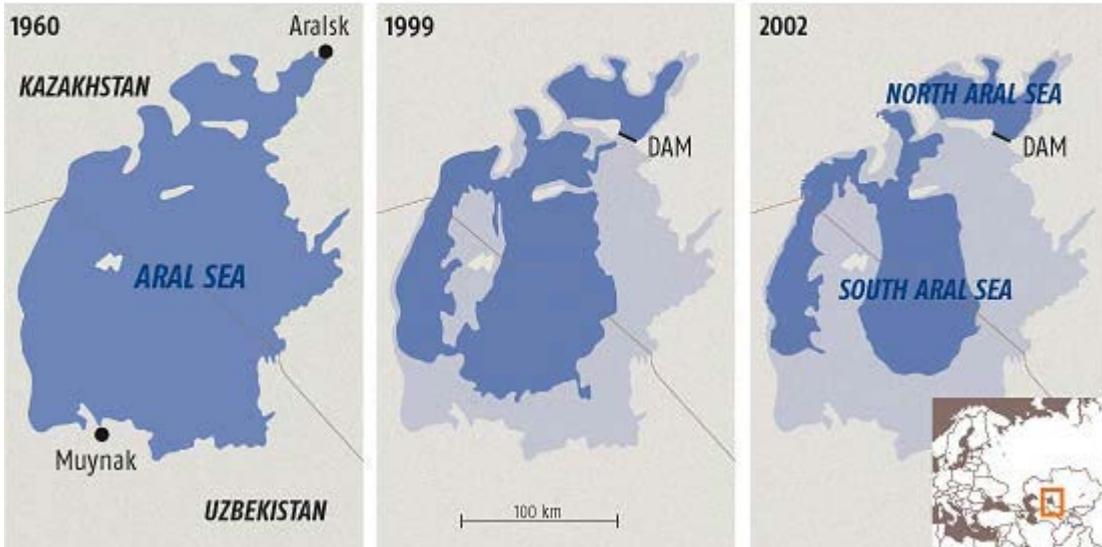
The changed shape of the Aral Sea since 1960



- 2005 – a dam between N and South part was constructed
- what has happened?

THE SHRINKING SEA

The changed shape of the Aral Sea since 1960



- 2005 – a dam between N and South part was constructed
- what has happened?



An original area of the Aral sea (1960) was:

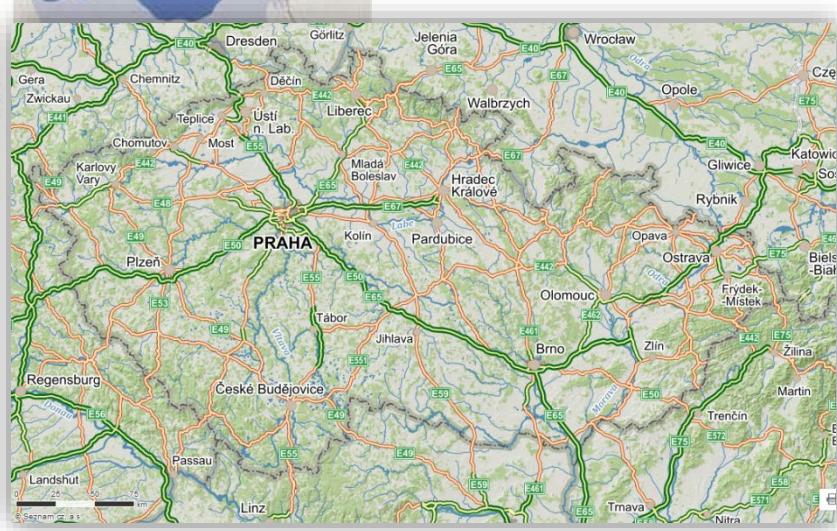
Bigger than
Czechia

Of similar
size to
Czechia

Smaller than
Czechia

THE SHRINKING SEA

The changed shape of the Aral Sea since 1960



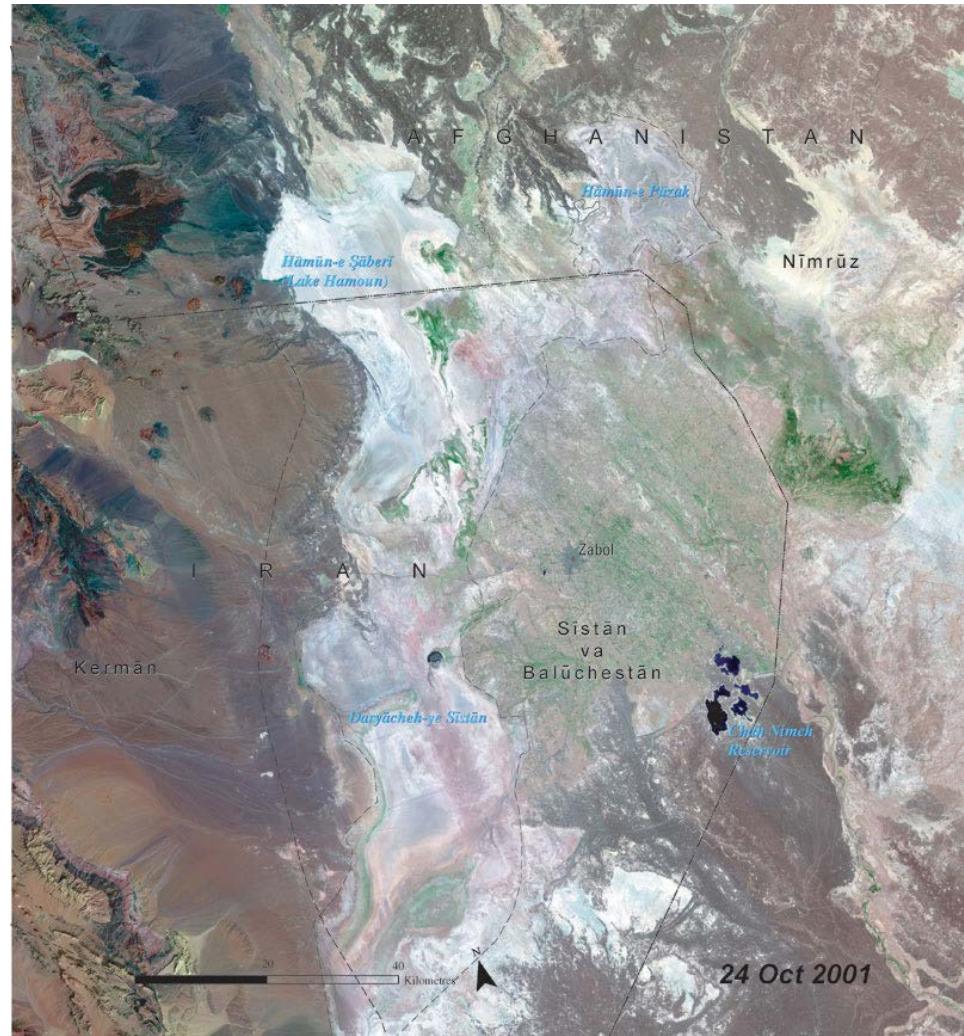
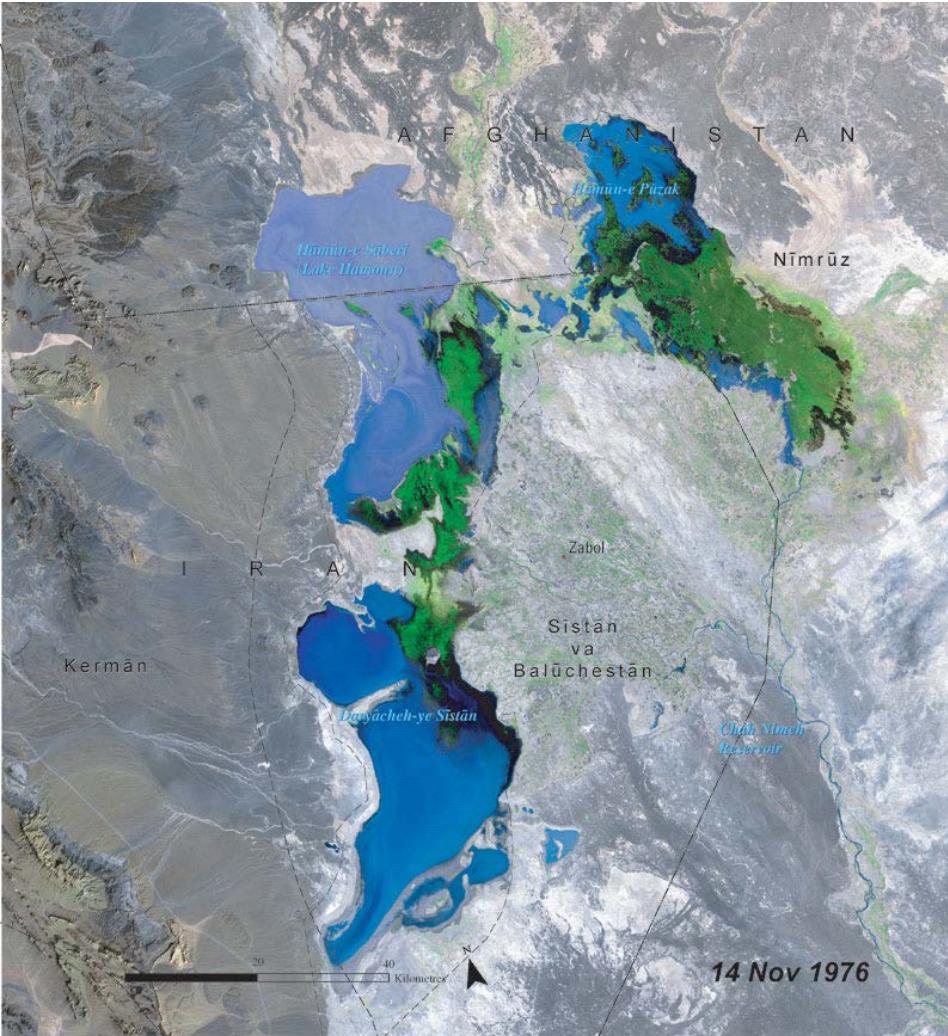
THE SHRINKING SEA

The changed shape of the Aral Sea since 1960



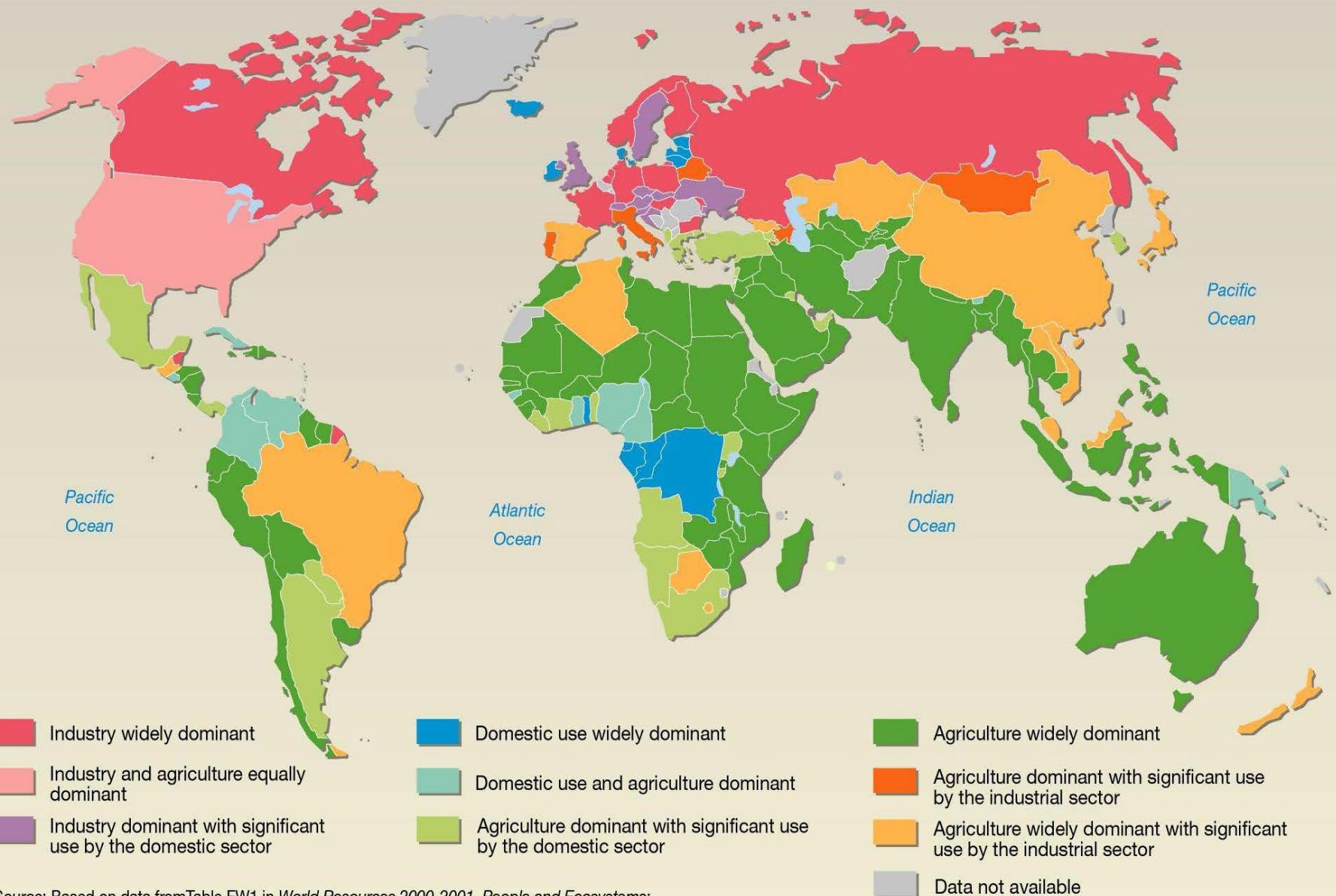


Lake Hamoun – Iran, Afghanistan





Areas and types of water consumption





Water consumption - what does it mean?

Start the presentation to see live content. For screen share software, share the entire screen. Get help at pollev.com/app





Areas and types of water scarcity

Areas around the globe suffering from depleted water resources

Physical water scarcity

Water resource development is approaching or has exceeded sustainable limits. More than 75% of river flow is extracted for agriculture

Approaching physical water scarcity

More than 60% of river flow is extracted. These areas will experience physical water scarcity in the near future

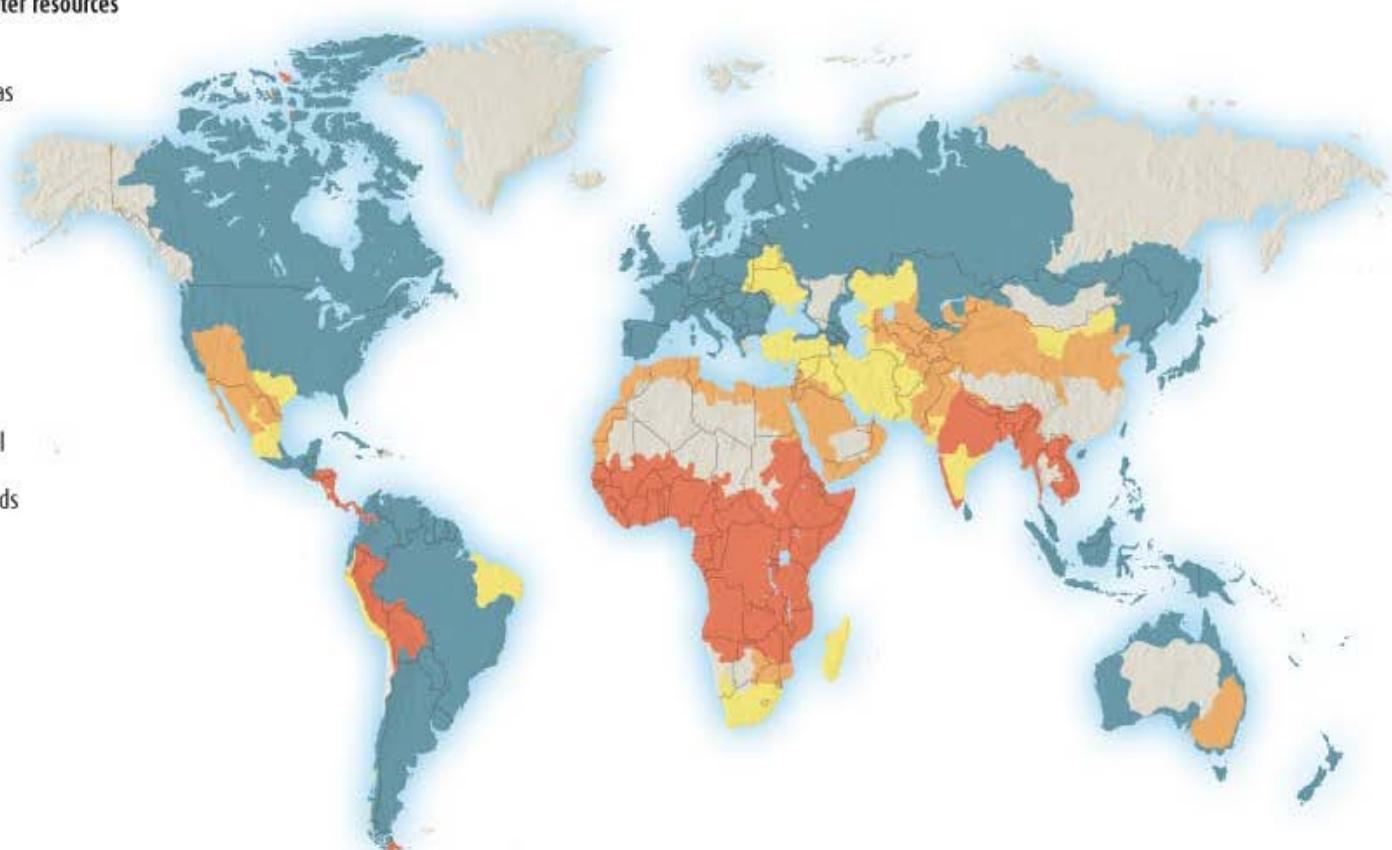
Economic water scarcity

Limited access to water even though natural local supplies are available to meet human demands. Less than 25% of water extracted for human needs

Little or no water scarcity

Abundant water resources relative to use, with less than 25% of water extracted for human purposes

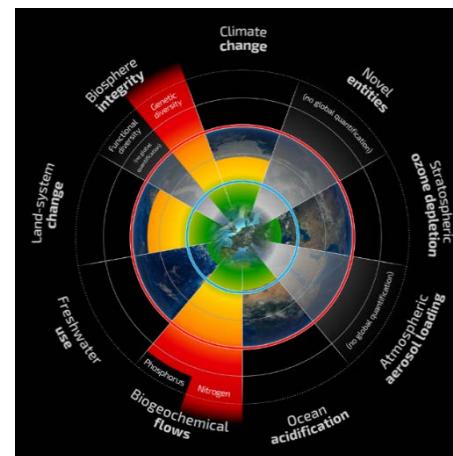
Not estimated





VIII. Land use

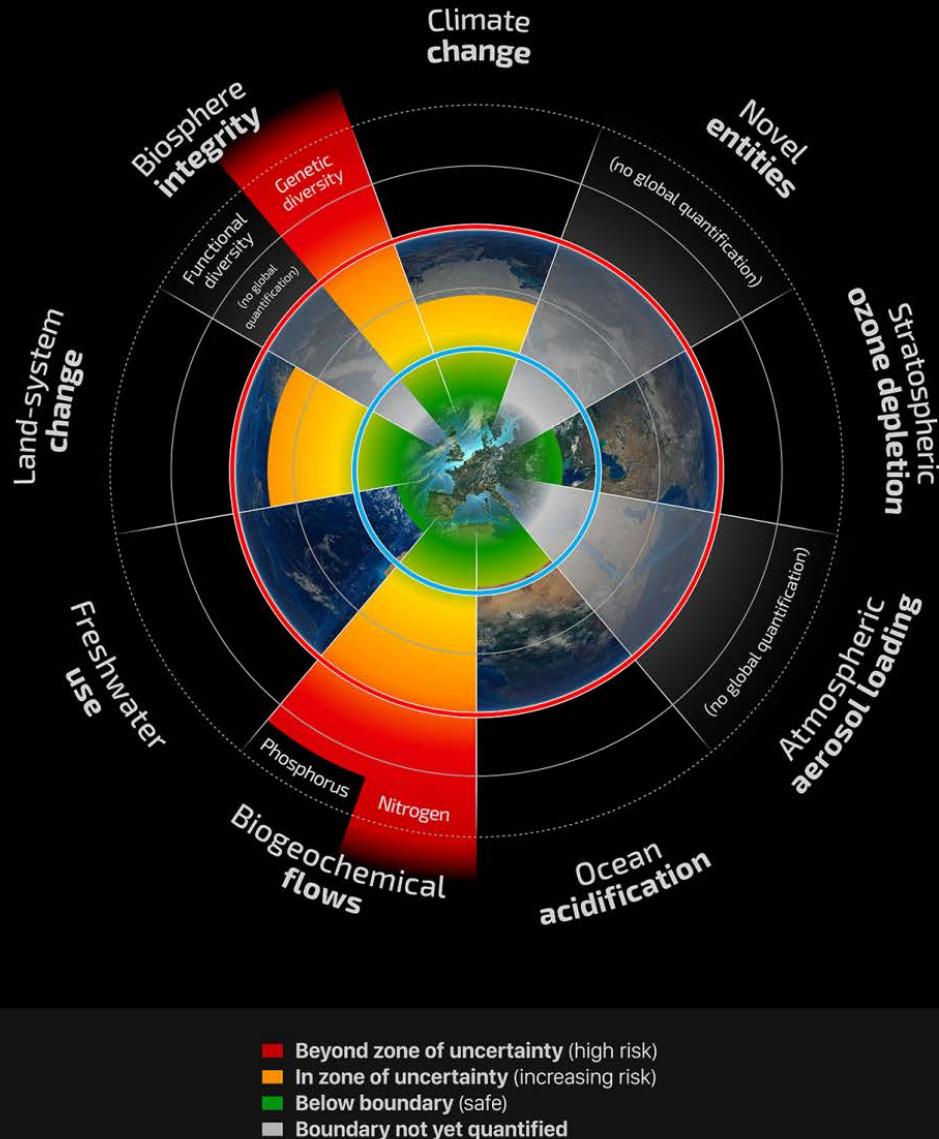
Earth-system process	Control variable(s)	Planetary boundary (zone of uncertainty)	Current value of control variable
Land-system change (R2009: same)	<i>Global</i> : Area of forested land as % of original forest cover	<i>Global</i> : 75% (75–54%) Values are a weighted average of the three individual biome boundaries and their uncertainty zones	62%
	<i>Biome</i> : Area of forested land as % of potential forest	<i>Biome</i> : Tropical: 85% (85–60%) Temperate: 50% (50–30%) Boreal: 85% (85–60%)	





Planetary Boundaries

A safe operating space for humanity





Santa Cruz, Bolívia

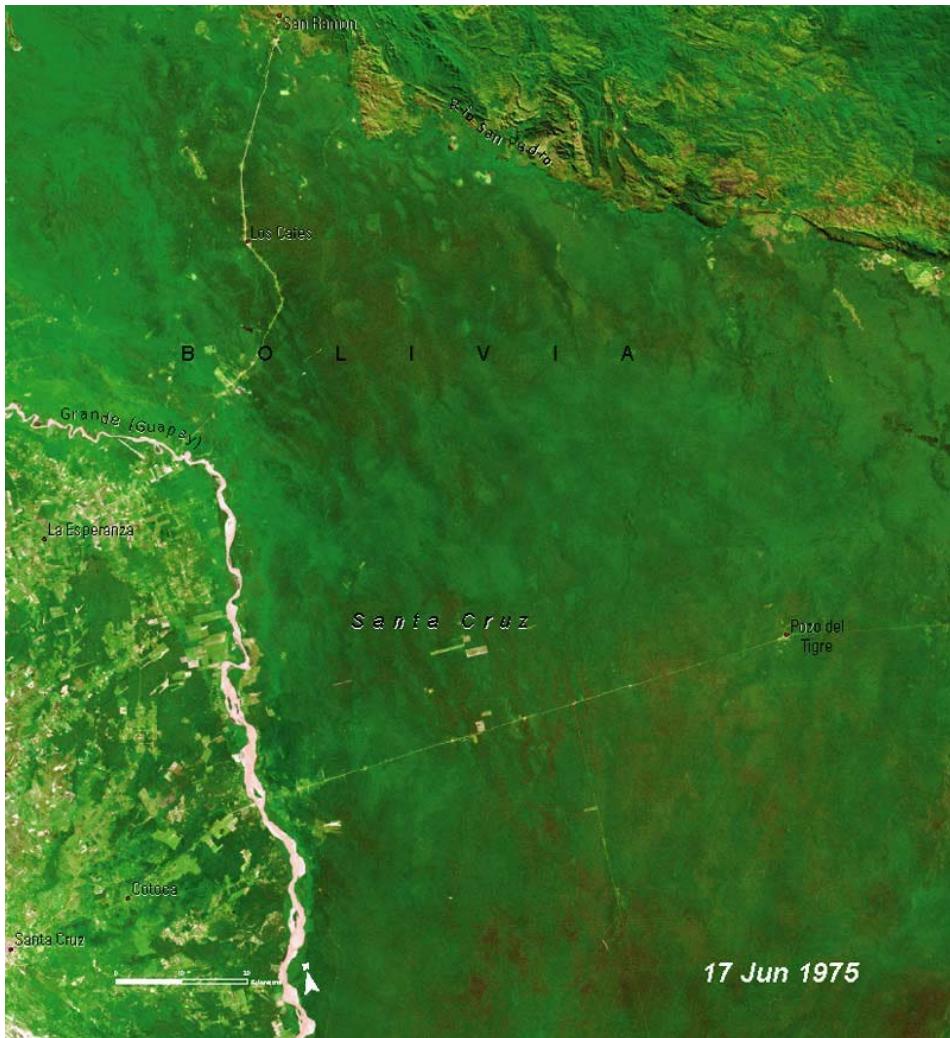
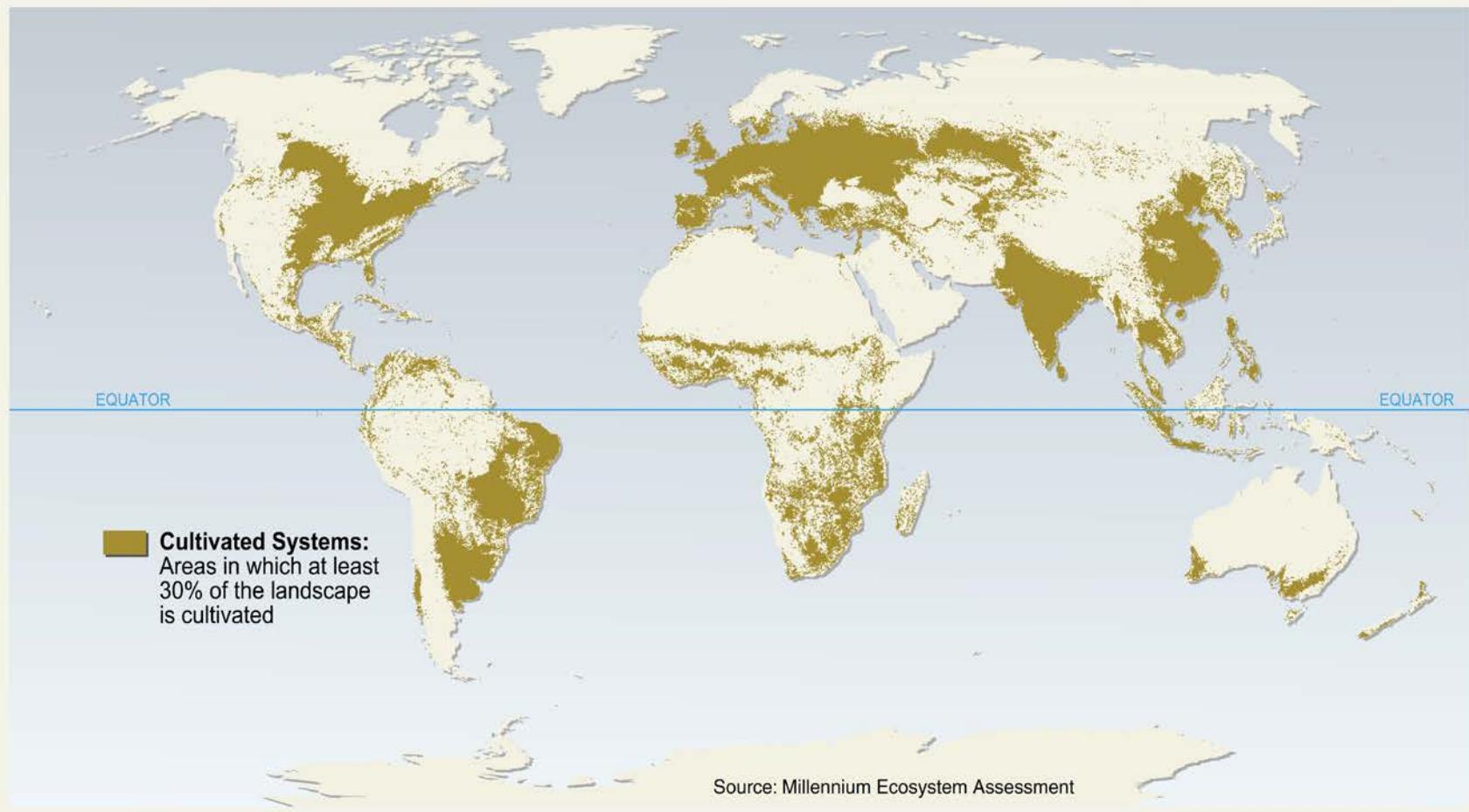




Figure 1. EXTENT OF CULTIVATED SYSTEMS, 2000. Cultivated systems cover 24% of the terrestrial surface.





Change in land system

- agriculture (main driver)
- In the last 50 years, change of the ice-free land to the agriculture land circa 0.8% per year
- main force of the changes of ecosystem services and functions (e.g. food production, water cleaning, etc.),
- habitat loss is a main force of biodiversity loss
- it undermines human well-being and sustainability
- exceeding the safe level of a land system change in a certain region can lead to a sudden change in the character of the landscape



Change in land system

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Parts of Amazon close to tipping point

- › 13:52 05 March 2009 by Catherine Brahic
- › For similar stories, visit the [Endangered Species Topic Guide](#)

The Mato Grosso, the most scarred region of the Amazon rainforest, is teetering on a deforestation "tipping point", and may soon be on a one-way route to becoming a dry and relatively barren savannah.

Mônica Carneiro Alves Senna and colleagues at the Federal University of Viçosa, Brazil, used computer models to simulate how the Amazon would recover from various amounts of deforestation. Their simulations ranged from a complete wipe-out of the entire forest to a situation where just one fifth of the forest would be removed.



Optimistic future?

Shrinking farmland

For the first time, more land is being left to return to nature than is being cleared for agriculture

