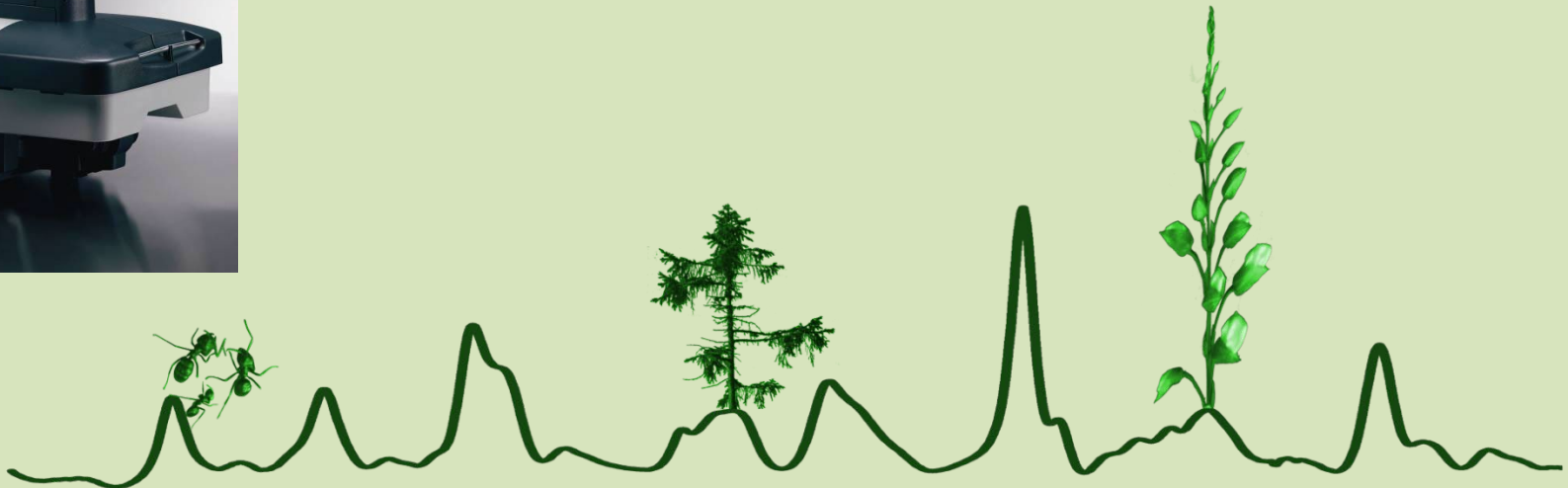


Raman spectroscopy / microscopy of bone fragments



Samples

- Sample 1: 1400
- Sample 2: CCB01

Powdered and sieved bone fragments
(> 25 microns) / **Untreated samples.**

Measurement technique

DXR3 Raman microscope, 2023 (3th gen.)





**Sir Chandrasekhara
Venkata Raman
1888 – 1970**

**Nobel Price in Physics
1930**

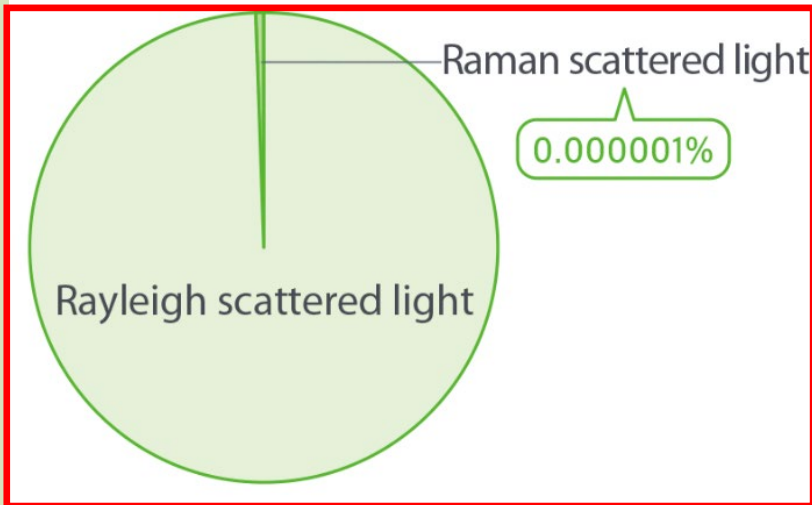
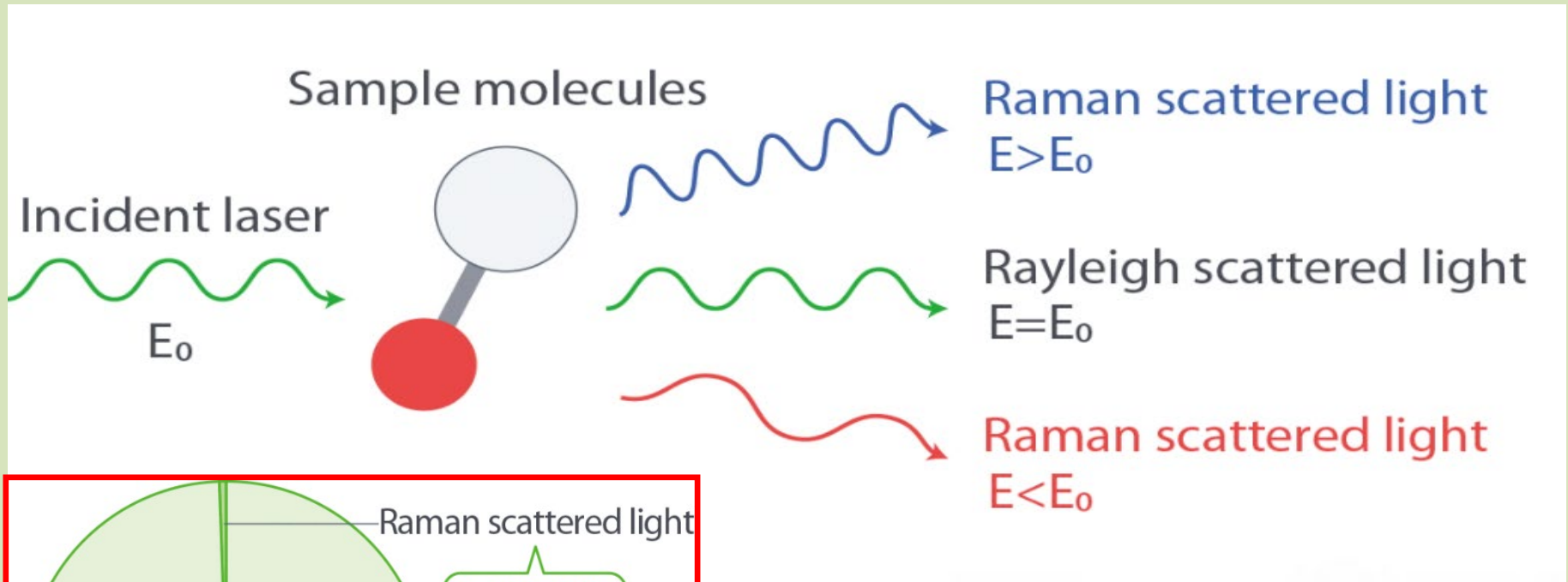
A New Type of Secondary Radiation

C. V. Raman and K. S. Krishnan, *Nature*, 121(3048), 501, March 31, 1928

The experiments we have made have confirmed this anticipation, and shown that in every case in which light is scattered by the molecules in dust-free liquids or gases, the diffuse radiation of the ordinary kind, having the same wavelength as the incident beam, is accompanied **by a modified scattered radiation of degraded frequency.**



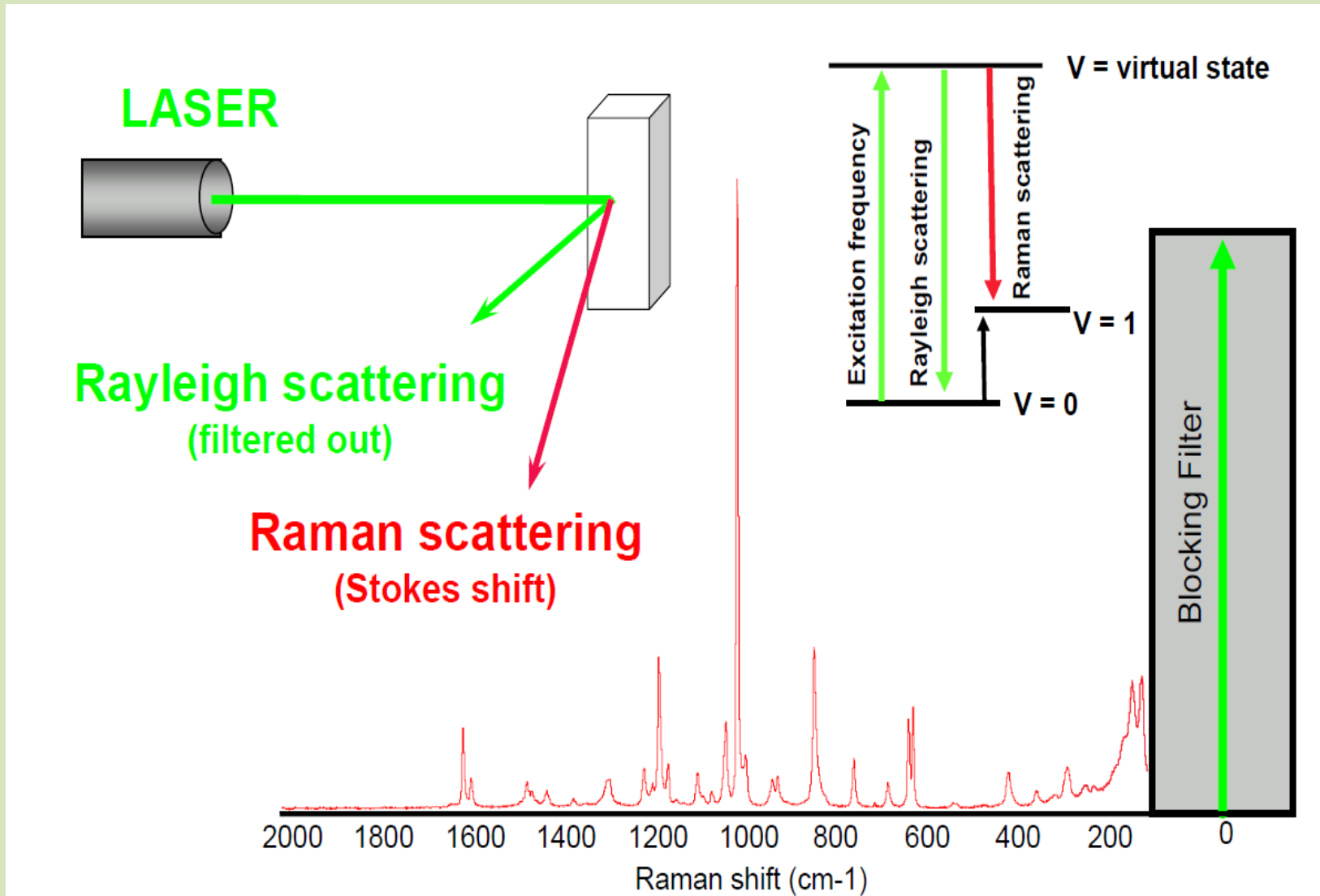
Raman microscopy: theory



Optical inelastic scattering

Dominant process is Rayleigh scattered light, Raman is extremely weak
 One photon from $10^6 - 10^8$ photons is Raman scattering!

Raman microscopy: theory



Raman versus infrared spectroscopy

Raman

It is due to the **scattering** of light by the vibrating molecules.

The vibration is Raman active if it causes a change in **polarisability**.

The molecule need not possess a permanent dipole moment.

Water can be used as a solvent

Sample preparation is not very elaborate sample can be almost in any state

Gives an indication of covalent character in the molecule.

More suitable for anorganic features
Cutoff usually 50 cm^{-1}

Microscopy: 0,4 micron resolution

IR

It is the result of **absorption** of light by vibrating molecules.

The vibration is IR active if there is a change in **dipole moment** during the vibration.

The vibration concerned should have a change in dipole moment

Water cannot usually be used due to its intense absorption IR spectrum

Sample preparation is elaborate for some techniques

Gives an indication of ionic character in the molecule.

Cutoff usually 400 cm^{-1}

Microscopy: 10 x 10 micron resolution

DXR3 Raman microscope

Several possibilities of Raman spectra measurement modes:

- „Point and Shoot“: manual selection of measurement spot on the samples (software, or hardware joystick)
- **Point map**: selection of measurements points on visual picture or overall sample image mosaic – number of points is not limited
- **Linear map**: line of measurement points – microlayers, cuts, linear section of sample
- **Area map 2D**: area grid of measurement points – 2D imaging of samples
- **Depth profiling**: microscope confocality allows depth profiling without any sample preparations. Depth profile resolution usually lower than 2 micrometers
- **Combination of depth profiling and area/linear mapping: 3D imaging**

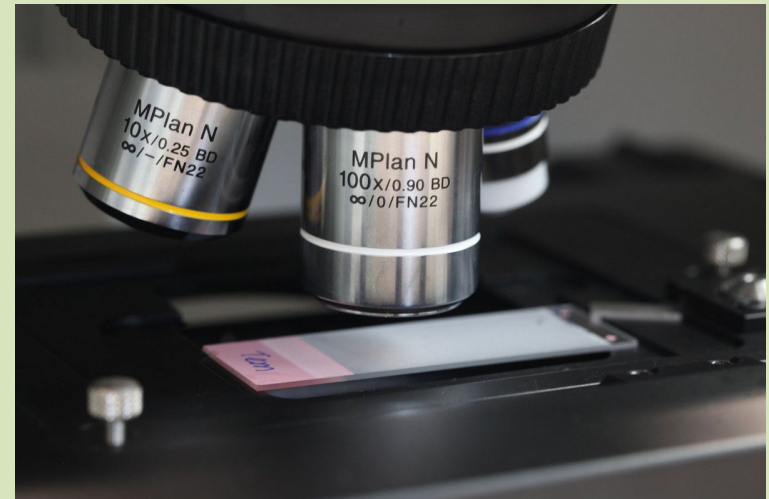
Possibility of great selection of excitation lasers wavelengths, gratings with different range and spectral resolution. Software tracks all parts and optimizing measurement parameters.

Choice of excitation in nm, maximum laser power on sample in mW. Polarizer/analyser experiments option etc.



Raman spectroscopy of bones

- Usually in literature FT-IR spectroscopy/microscopy option preferred
- Raman spectroscopy / microscopy as complementary method
- Raman has better spatial resolution = better understanding to material homogeneity + small particle detection
- „Inorganic“ peaks in Raman spectra (vibrational modes) of bone materials mostly present with minimum fluorescence
- Measurement range 6 700 – 50 cm^{-1} (FT-IR cut-off usually around 400 cm^{-1})
- Raman easier for fiber optics coupling, remote/on-site analysis etc.



Vibration description	Frequency (cm ⁻¹)
CaCO ₃ / hydroxyapatite lattice vibrations	50 – 280 cm ⁻¹
$\nu_2(\text{PO}_4^{3-})$	423 - 433 cm ⁻¹ (P-O stretching/ O-P-O bending)
$\nu_2(\text{PO}_4^{3-})$	446 - 449 cm ⁻¹ (O-P-O bending)
$\nu_4(\text{PO}_4^{3-})$	561 – 617 cm ⁻¹ (O-P-O bending)
Proline	851 – 855 cm ⁻¹ (benzene ring)
Hydroxyproline	873 - 875 cm ⁻¹ (benzene ring)
$\nu_2(\text{CO}_3^{2-})$	880 cm ⁻¹ (C-O stretching)
Proline	917 cm ⁻¹
$\nu_1(\text{PO}_4^{3-})$	955 - 962 cm ⁻¹ (dominant P-O symmetric stretch)
$\nu_3(\text{PO}_4^{3-})$	1023 – 1054 cm ⁻¹ (P-O asymmetric stretch)
$\nu_1(\text{CO}_3^{2-})$	1065 - 1077 cm ⁻¹ (B-type substitution / C-O stretching in plane)
$\nu_1(\text{CO}_3^{2-})$	1079 – 1080 cm ⁻¹ (amorphous K ₂ CO ₃)
$\nu_1(\text{CO}_3^{2-})$	1095 - 1105 cm ⁻¹ (A-type substitution / C-O stretching in plane)
Amide III	1230 - 1289 cm ⁻¹ (C-N-H stretch, bending N-H)
	Continued on next slide

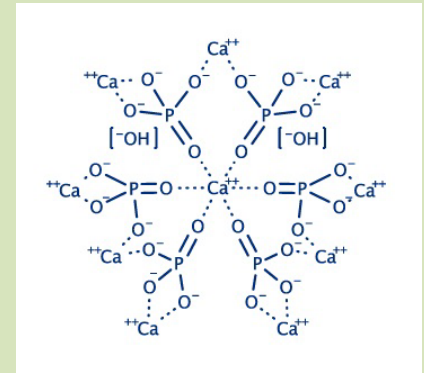
Bone is a material comprising usually a protein component (mostly collagen I) with inorganic matrix of hydroxyapatite (Ca₁₀(PO₄)₆OH_x), HA, where hydroxyl (OH) and phosphate (PO₄) groups being partly substituted by carbonate – CO₃

Raman vibrations in table are prepared mostly according to literature, not all these vibrations are present in analyzed samples

(main factor will be sample preparation and origin)

Vibration description	Frequency (cm ⁻¹)
CH ₂ wag	1455 – 1465 cm ⁻¹
Amide II	1455 – 1465 cm ⁻¹ (C-N-H stretch) (bending N-H)
Tyrosine Y8a	1610 – 1620 cm ⁻¹ (resonance-enhanced side-chain vibrations)
Amide I	1655 – 1675 cm ⁻¹ (C=O stretch)
v(CH)	2820 – 3035 cm ⁻¹ (hydrocarbons?)
v(OH)	3570 – 3576 cm ⁻¹
unknown	4367, 4397, 4500 cm ⁻¹

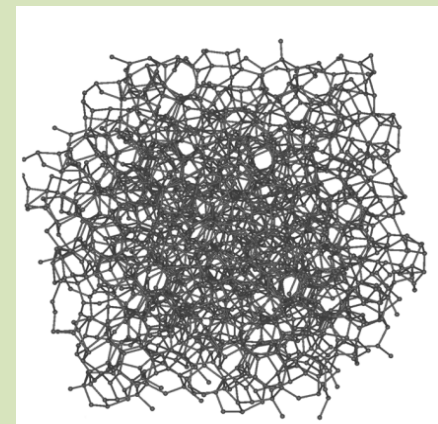
Hydroxyapatite HA:
(Ca₁₀(PO₄)₆OH_x)



Both samples are not perfectly pure, there are black particles visible with microscope, ID is carbon (form of „ash“), with amorphous carbon Raman peaks:

Vibration description	Frequency (cm ⁻¹)
D-band	1320 – 1360 cm ⁻¹
G-band	1500 – 1600 cm ⁻¹

Carbon

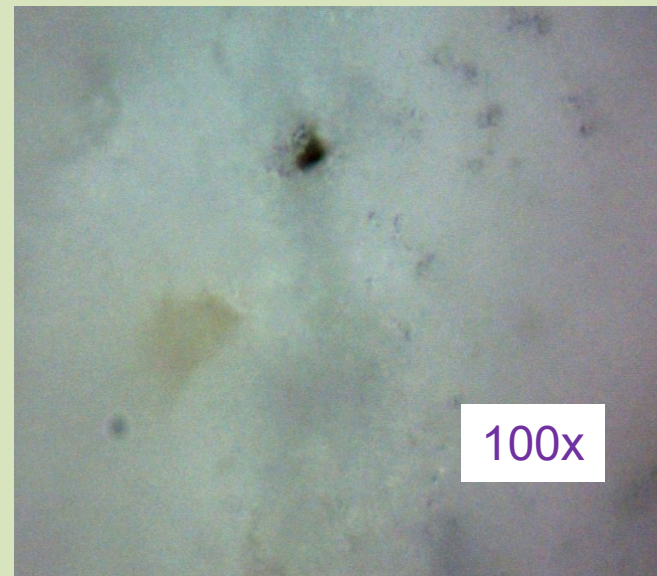
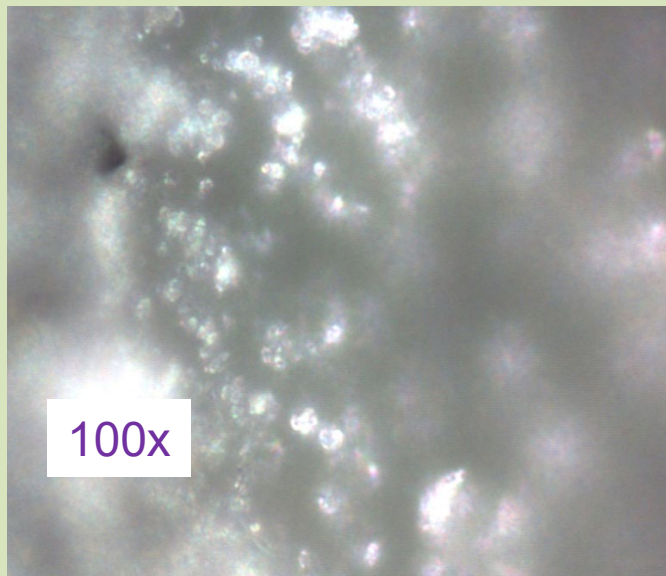
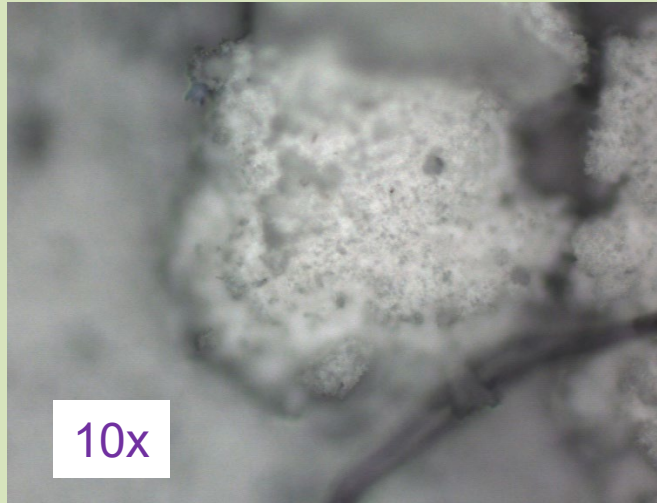
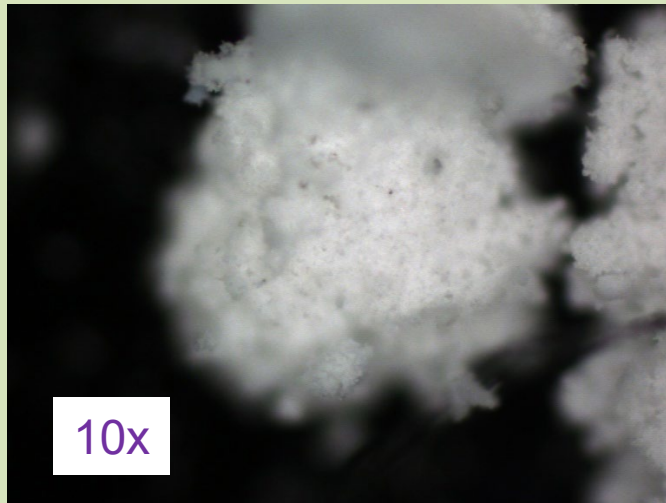


Raman spectroscopy of bones: theory

Notes from literature

- Two main types of carbonate (CO_3) substitution that occur in hydroxyapatite:
 - A-type: when CO_3 replaces OH groups
 - B-type. when CO_3 replaces PO_4 in the apatite structureIf these substitutions take place concurrently, an AB-type substitution occurs.
- Broadening of the phosphate symmetric stretching ($\nu_1 \text{PO}_4^{3-}$ 955 - 962 cm^{-1}) for decreasing temperatures reflects a higher disorder of the bioapatite lattice, associated with a larger amount of CO_3^{2-} ions (substituting either OH^- or PO_4^{3-}) in hydroxyapatite
- Symmetric stretching vibration $\nu_1(\text{PO}_4^{3-})$ has a high sensitivity to the mineral encirclement: the frequency and form of this line depends on the local surroundings and change as a result of substitution of anionic groups and changes in the degree of crystallinity.
- Substituted carbonate-apatite B-type phosphate ν_1 line appears in the range of 955–959 cm^{-1} . In a crystalline unsubstituted hydroxyapatite line ν_1 shifted to 962–965 cm^{-1}
Frequency line 945–950 cm^{-1} indicates the presence of a disordered lattice phosphate apatite.
- Phosphate band ν_1 is a superposition of three components; it usually has an asymmetric shape due to the disordered vibrations contribution of phosphate and unsubstituted hydroxyapatite.
- Variation in the bone's carbonate content is closely related to alterations in the growth morphology and crystallite size, that are known to take place upon temperature changes – a higher atomic disorder corresponding to smaller crystal dimensions, lattice becoming progressively more ordered with increasing temperatures (upon CO_3^{2-} loss)
- Etc....

Samples - description

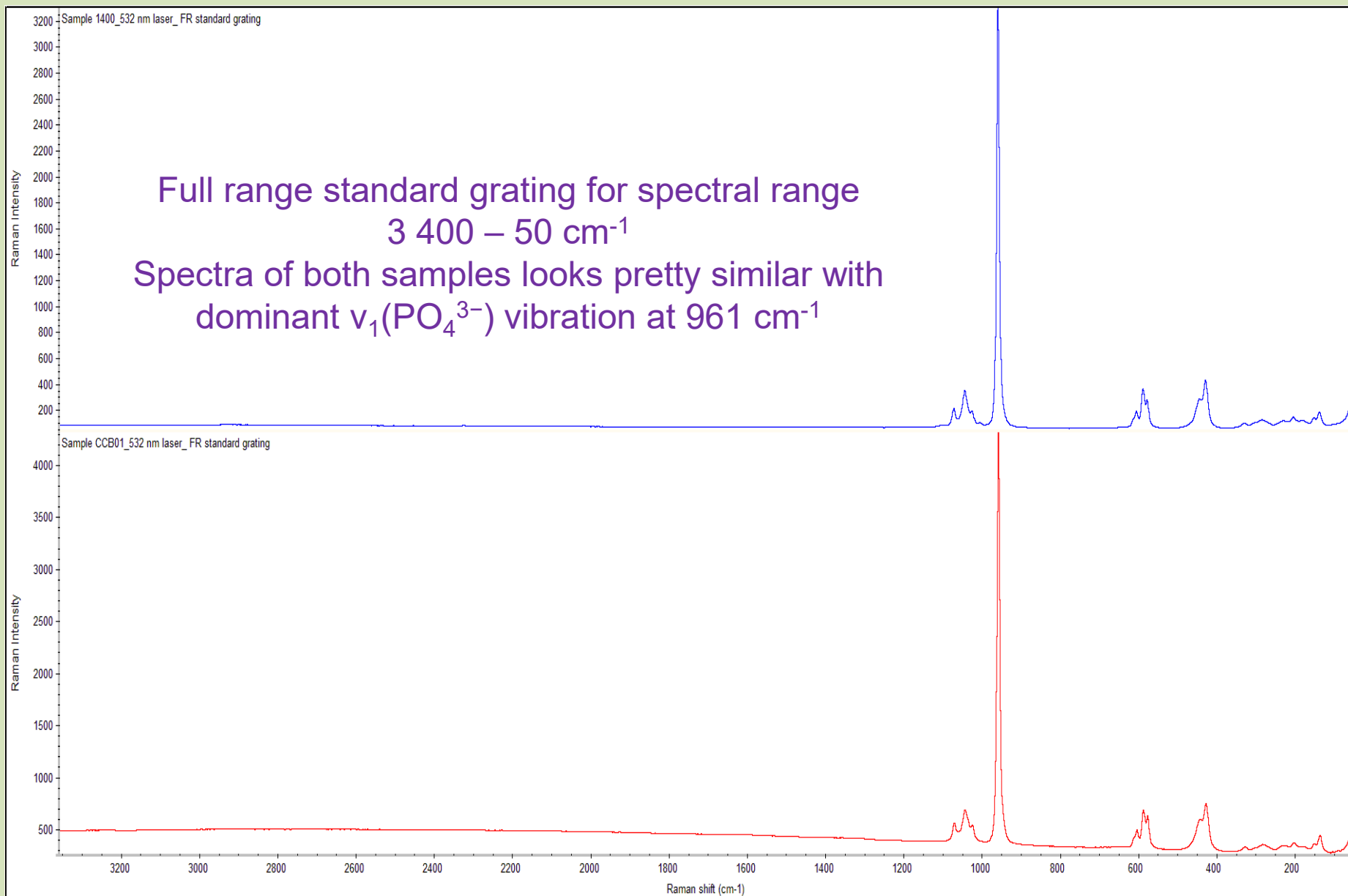


Both samples are homogenic white powders with randomly scattered black dots

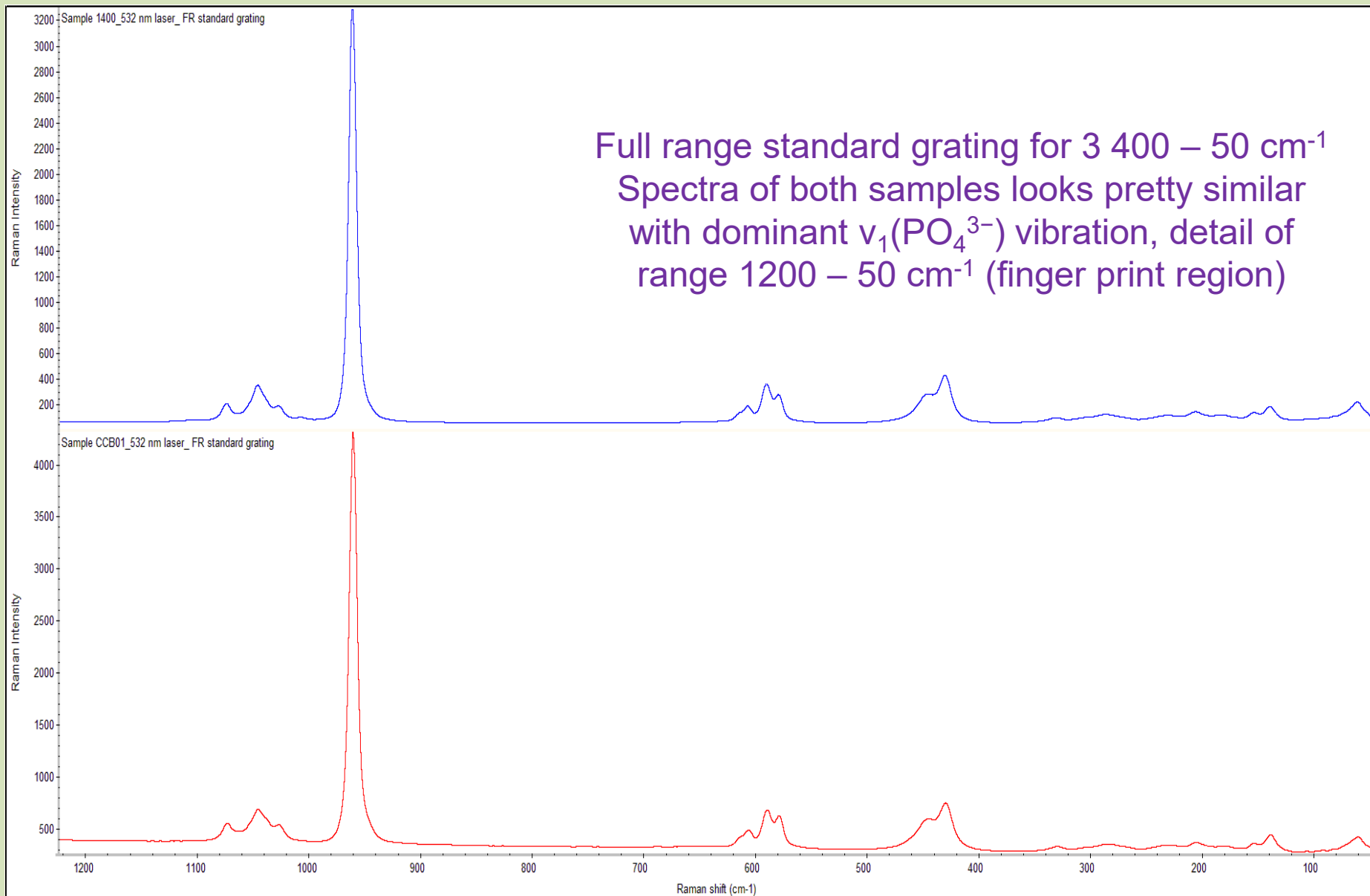
Raman microscope allows classic microscopy with glass objectives, polarization and mapping options

Sample volume needed for Raman analysis is very low, just couple of 25 micron particles (even one particle is enough)

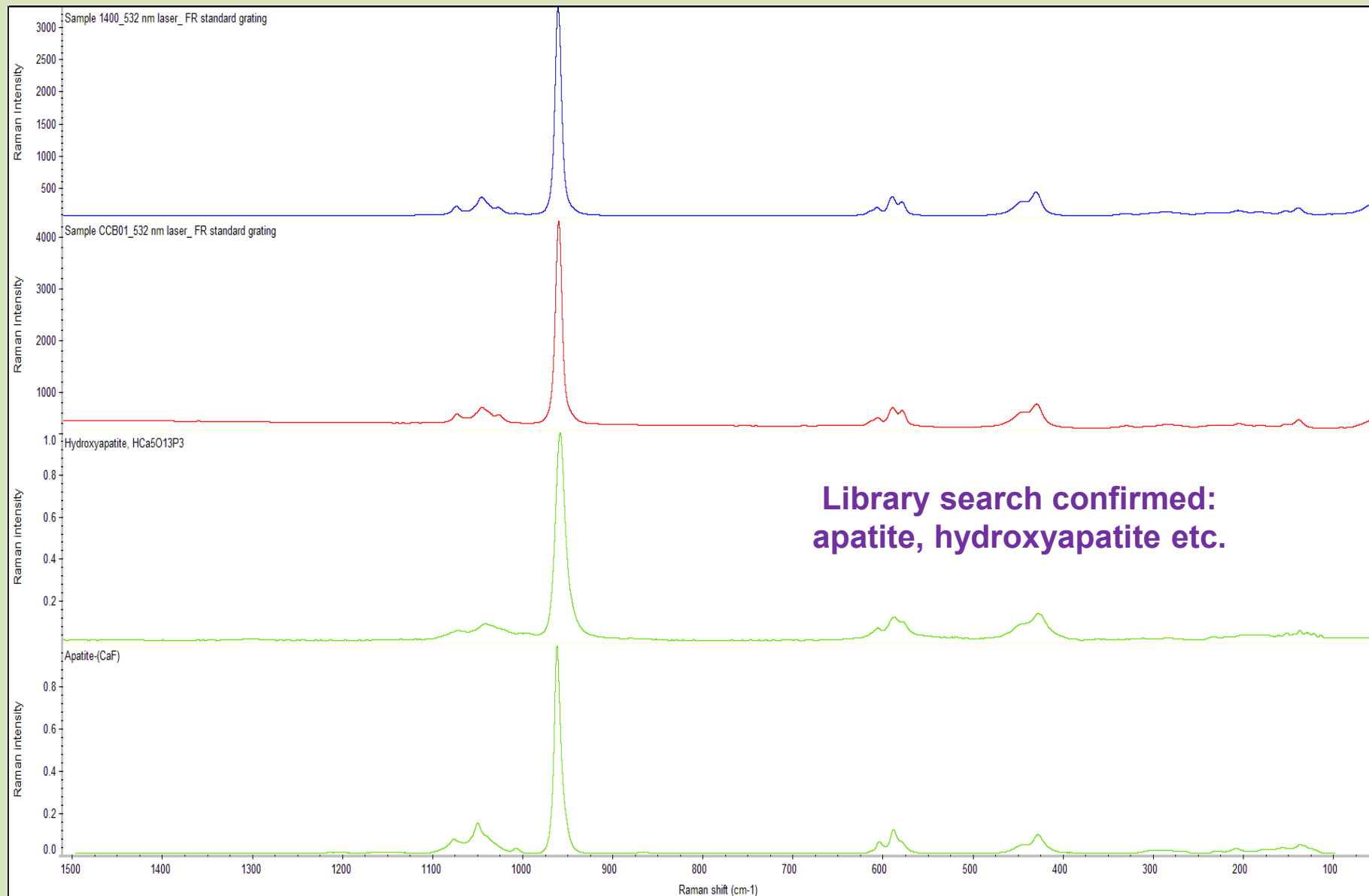
Raman spectra of samples



Raman spectra of samples

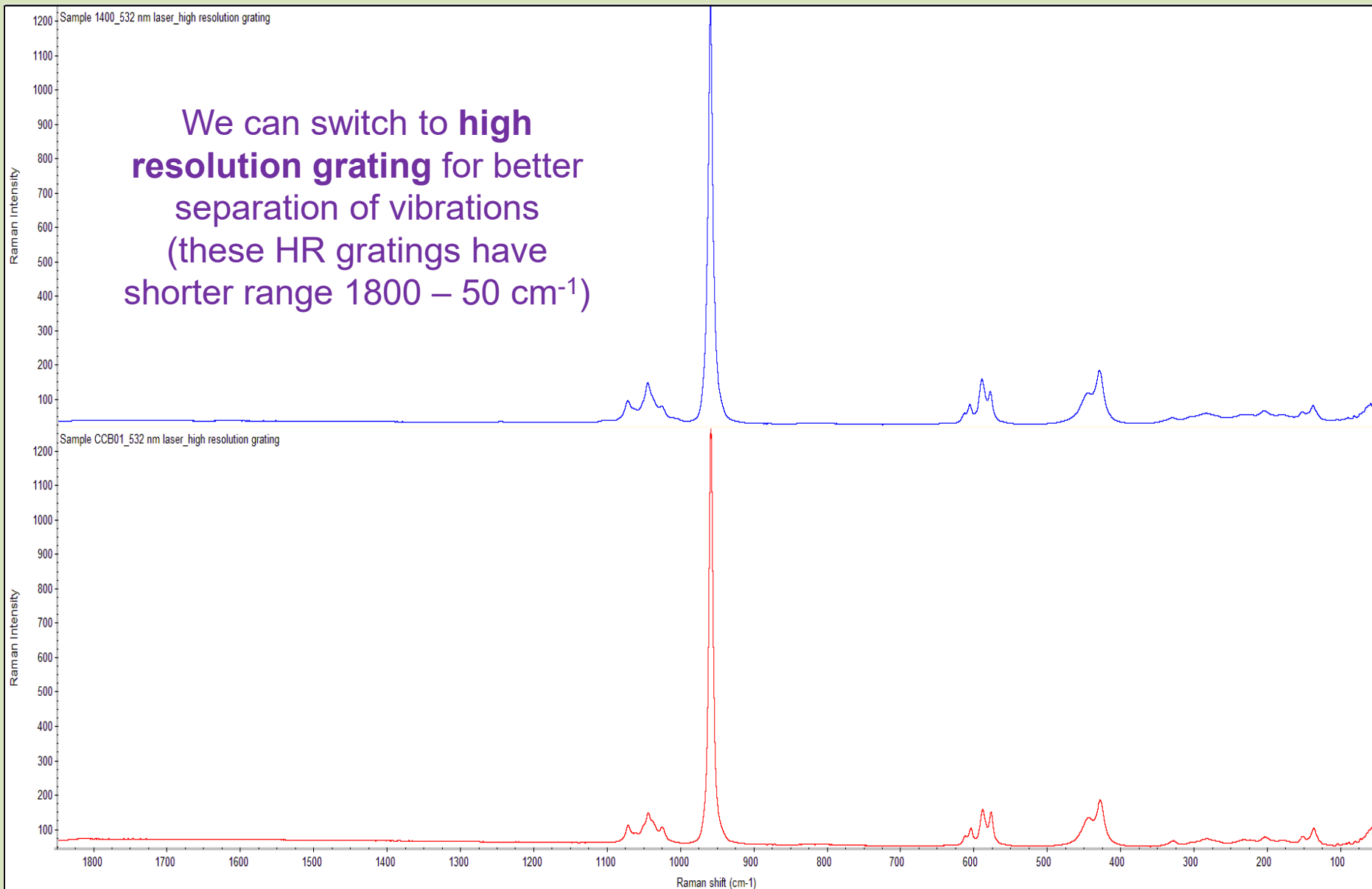


Raman spectra of samples

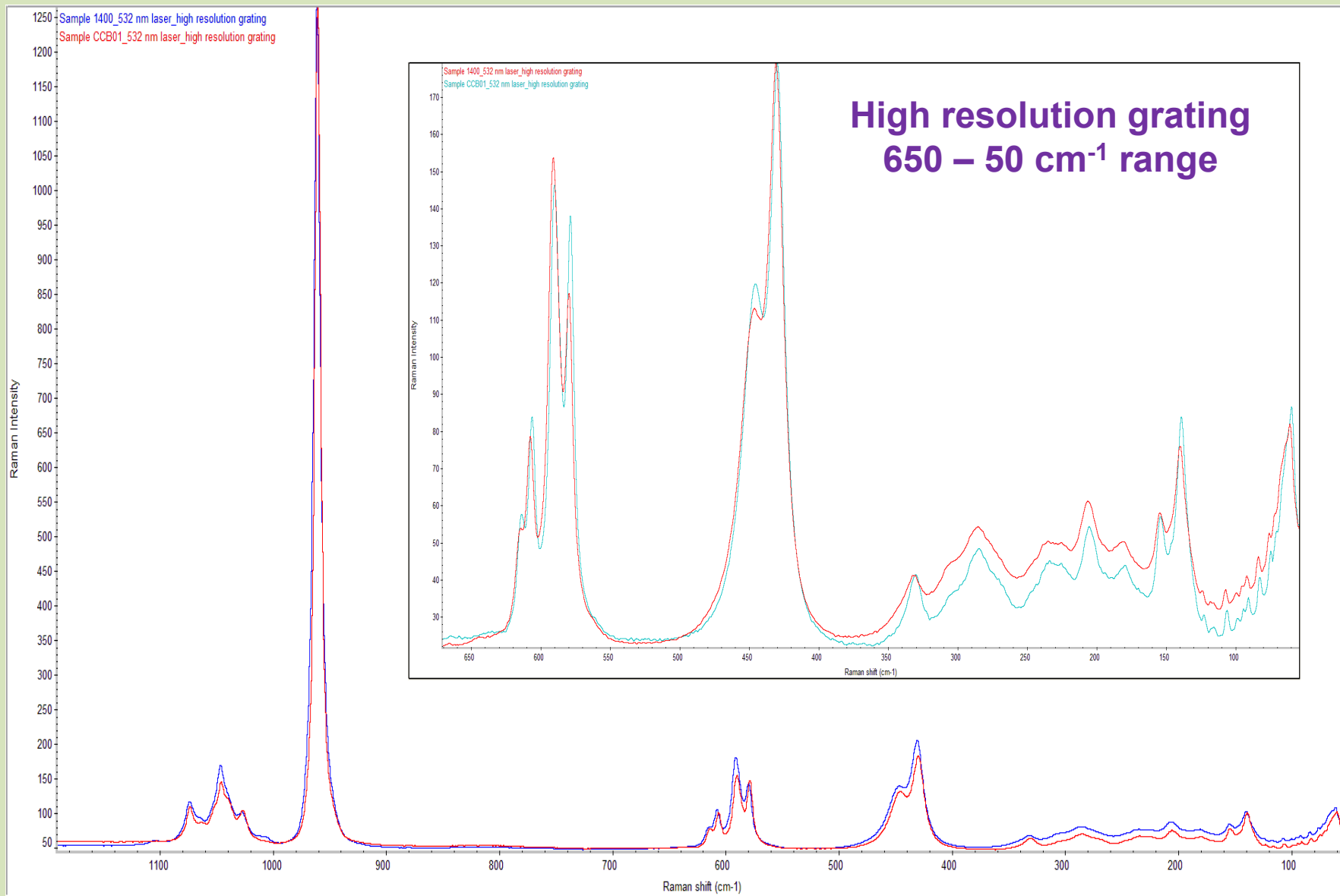


Raman spectra of samples

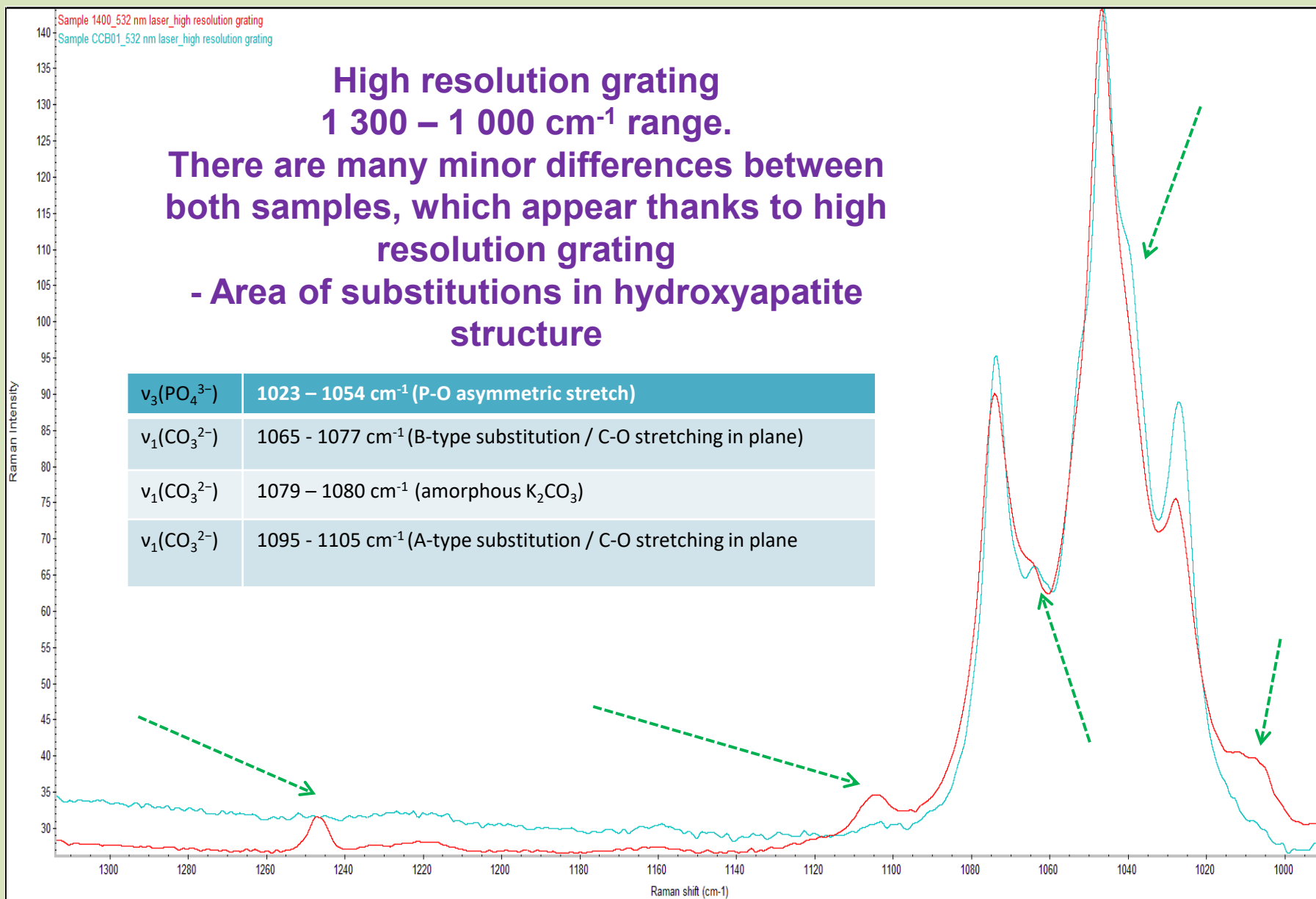
We can switch to **high resolution grating** for better separation of vibrations (these HR gratings have shorter range 1800 – 50 cm^{-1})



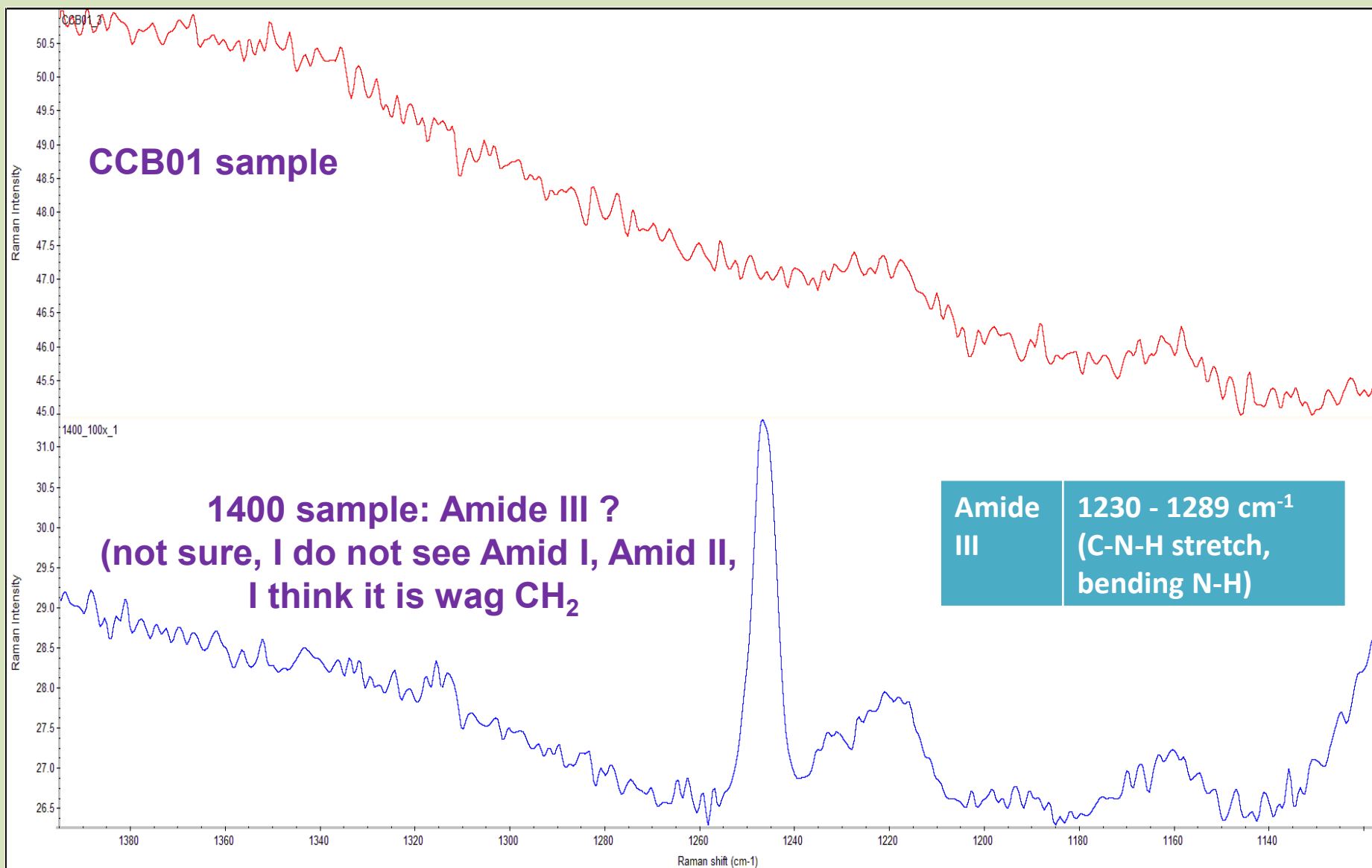
Raman spectra of samples



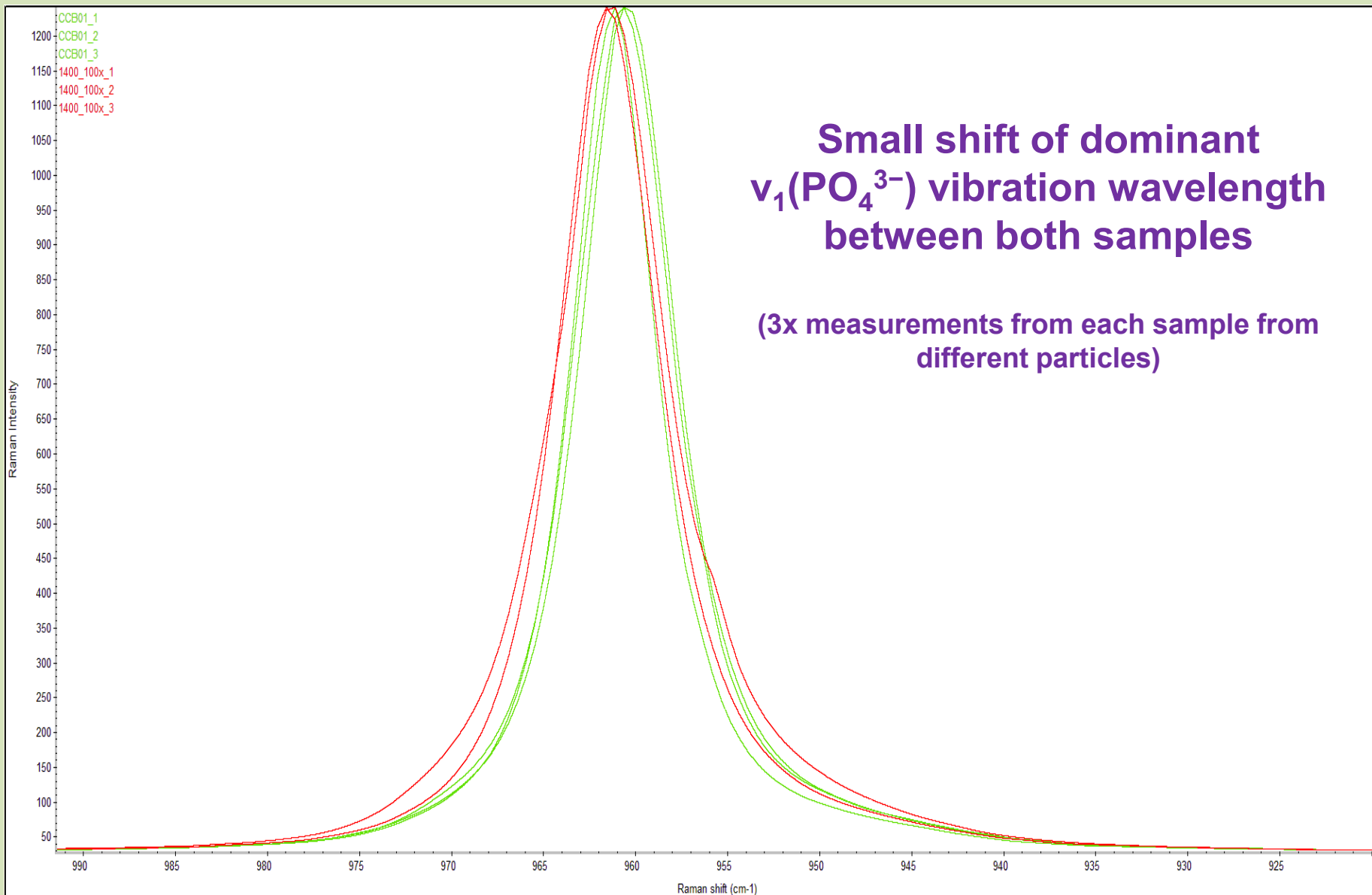
Raman spectra of samples



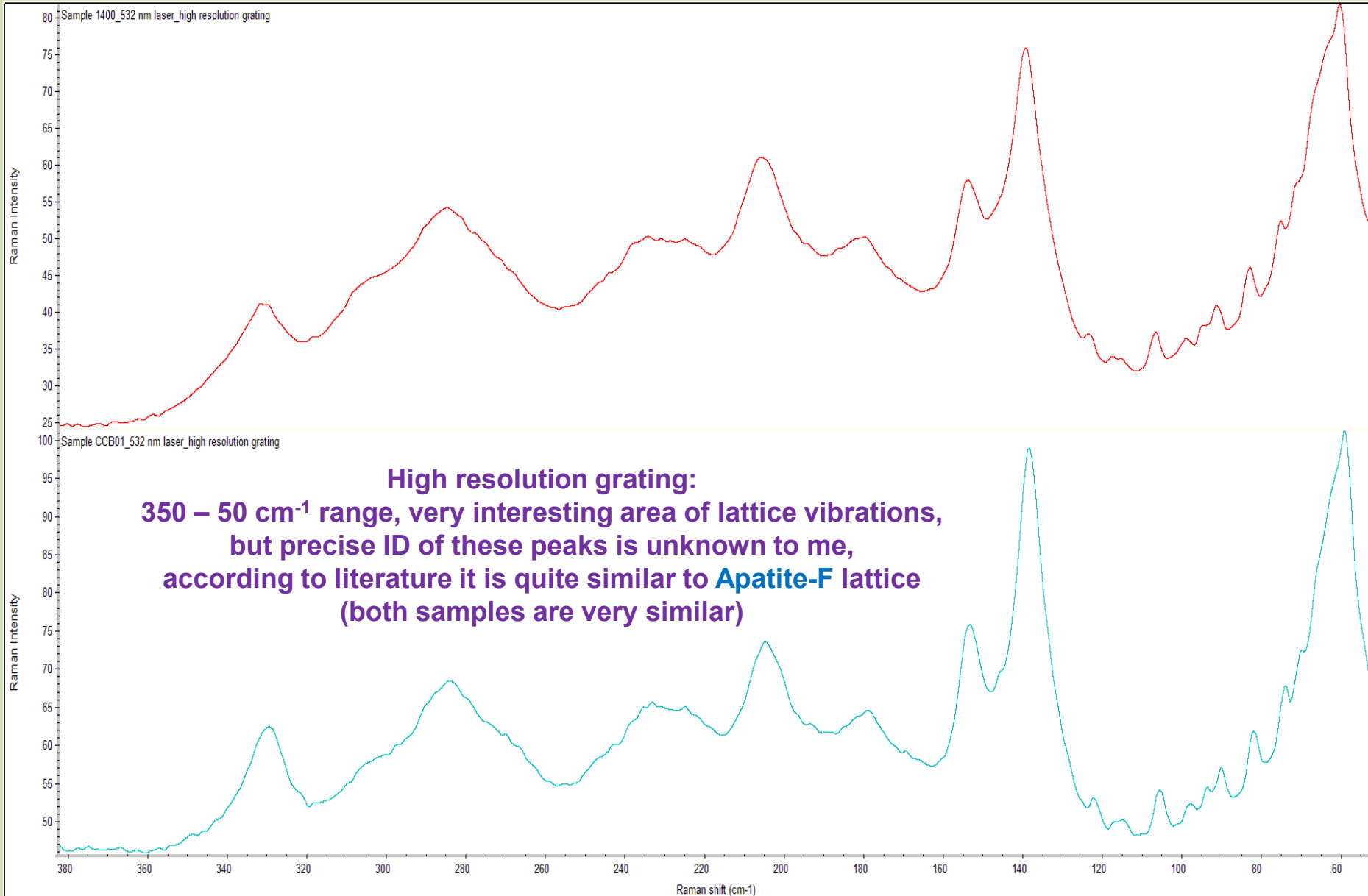
Raman spectra of samples



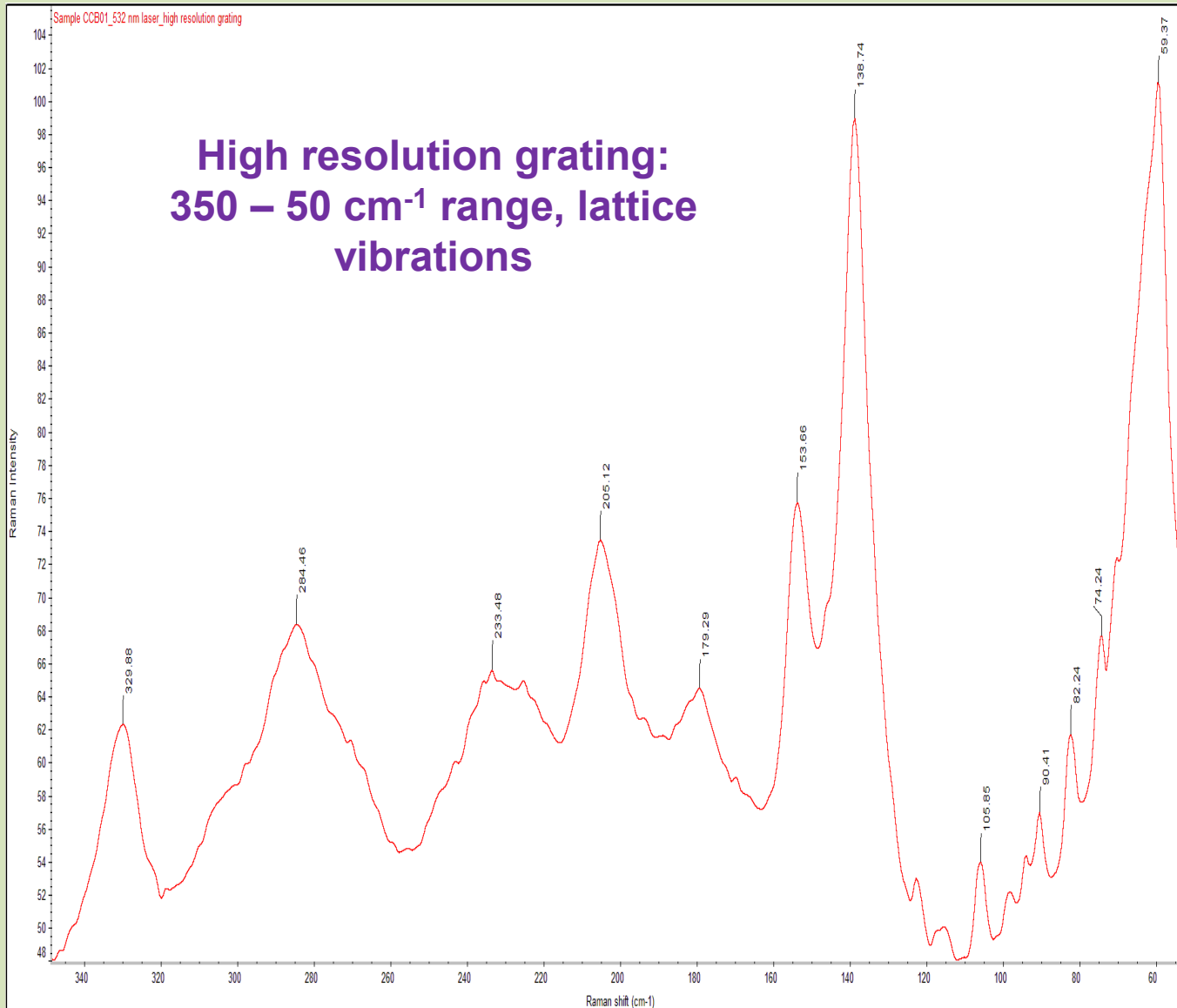
Raman spectra of samples



Raman spectra of samples



Raman spectra of samples



X Position (lattice)

59.4

74.2

82.2

90.4

105.9

138.7

153.7

179.3

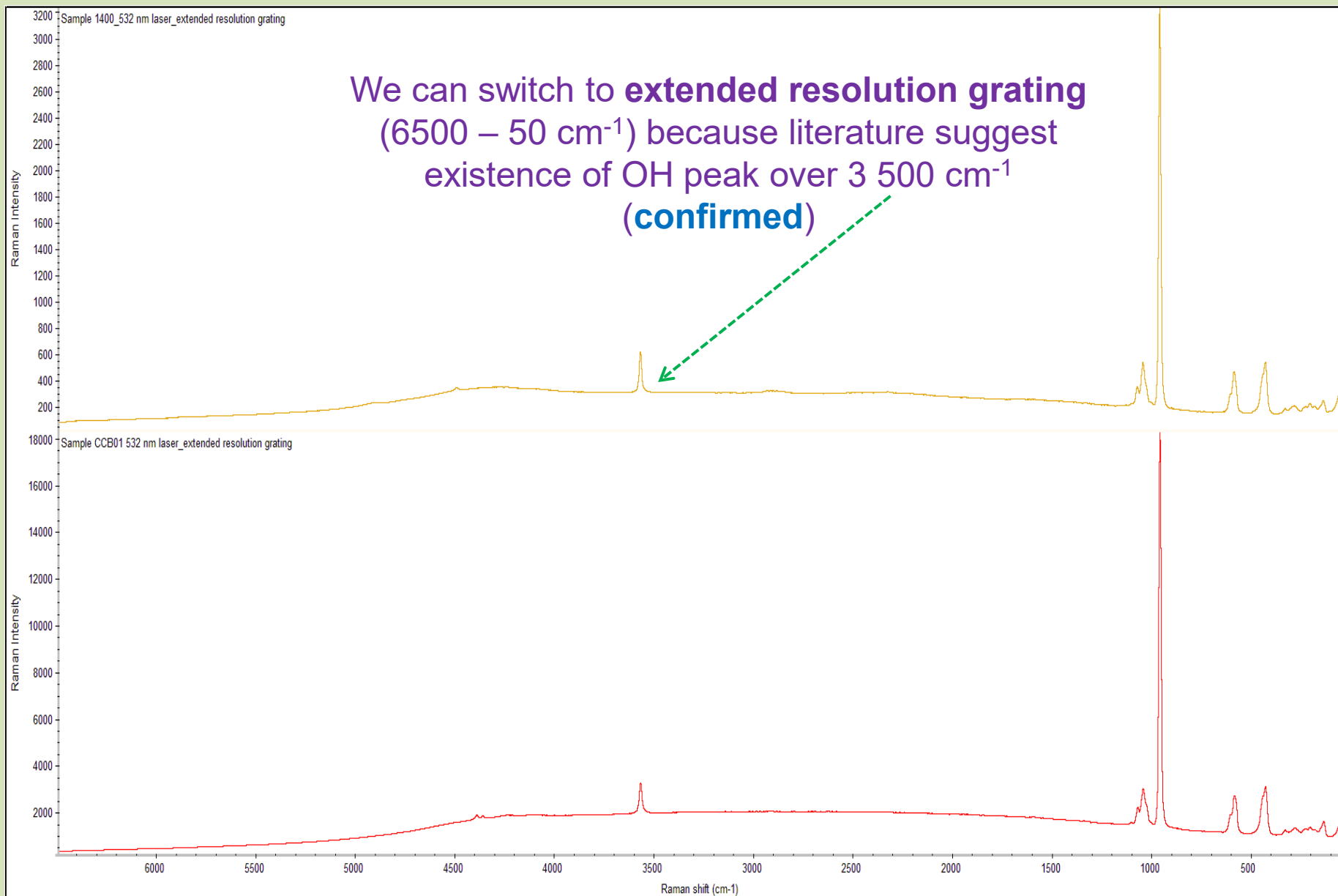
205.1

233.5

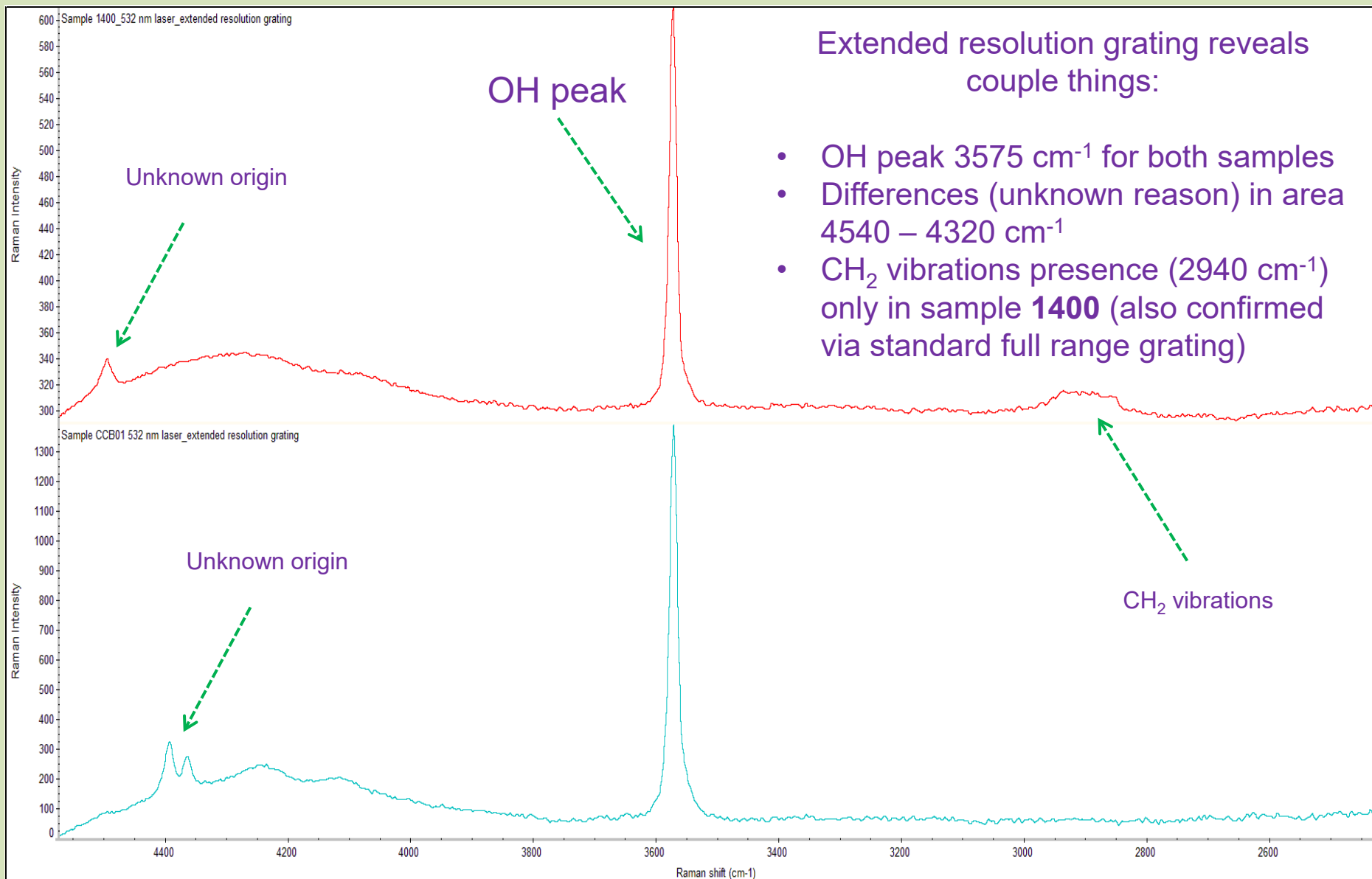
284.5

329.9

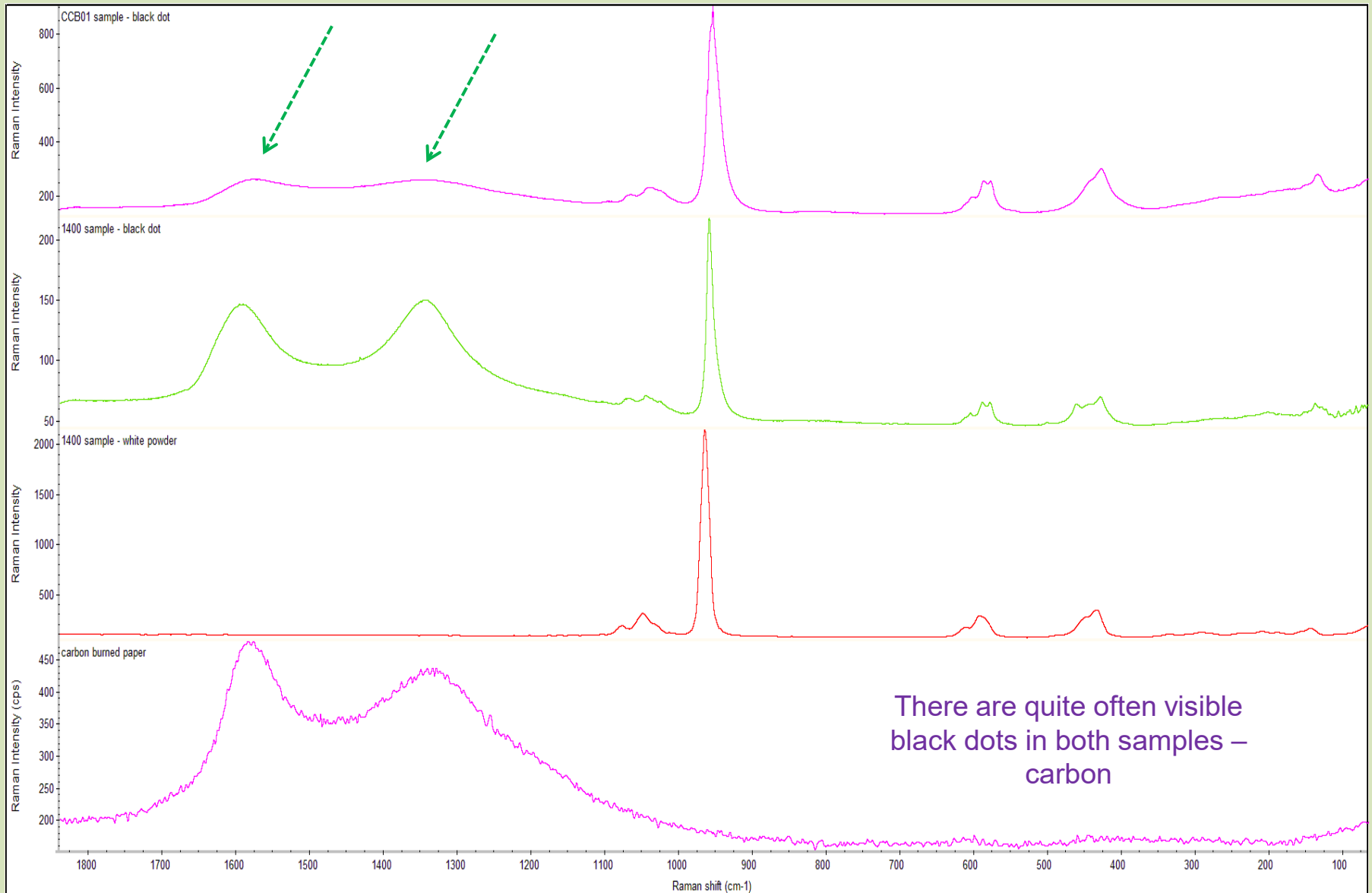
Raman spectra of samples



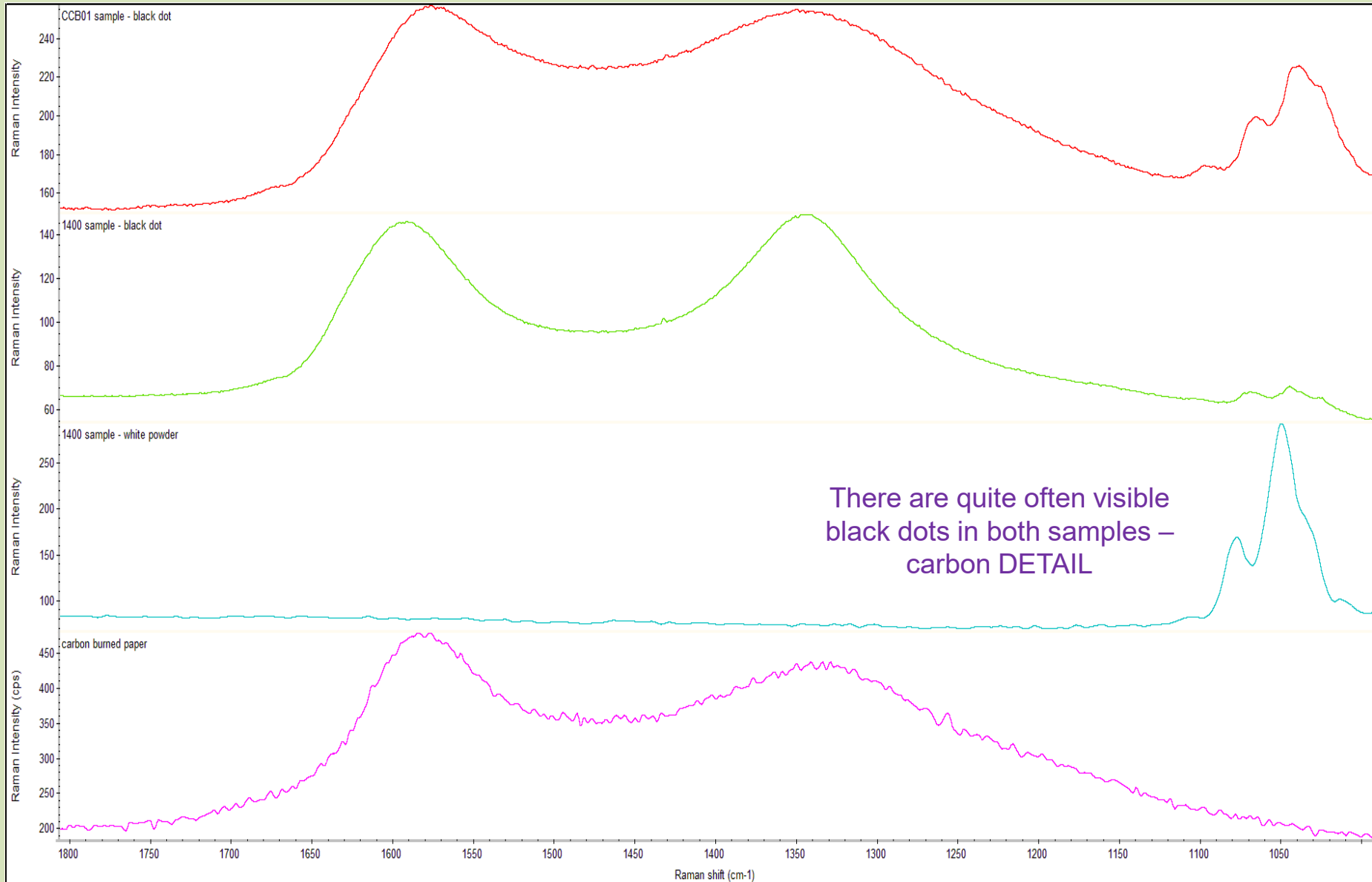
Raman spectra of samples



Raman spectra of samples



Raman spectra of samples



Raman spectra of samples: Summary and future research ideas

- Raman features of both samples are very similar, nevertheless there are differences, f.e. sample 1400 includes hydrocarbons CH_2 vibrations and unknown features over 4320 cm^{-1} range
- More samples Raman measurements required for better understanding of spectral features, maybe collect Raman spectra during whole sample preparation process / processes.
- Measurement of hydroxyapatite and apatite standards with high resolution gratings, again for better understanding of spectra and their differences
- More research according to notes on slide 11 (from literature)
- Temperature dependency experiments with high resolution grating: Linkam environmental chamber(s):

	TS1000
Temperature Range	Ambient to 1000°C
Heating/Cooling Rates	0.1°C to $200^\circ\text{C}/\text{min}$
Temperature Stability	$< 1^\circ\text{C}$
Sample Cup Sizes	17mm \varnothing x 3mm depth
Objective Lens Working Distance	7.0mm
Light Aperture	1.7mm

