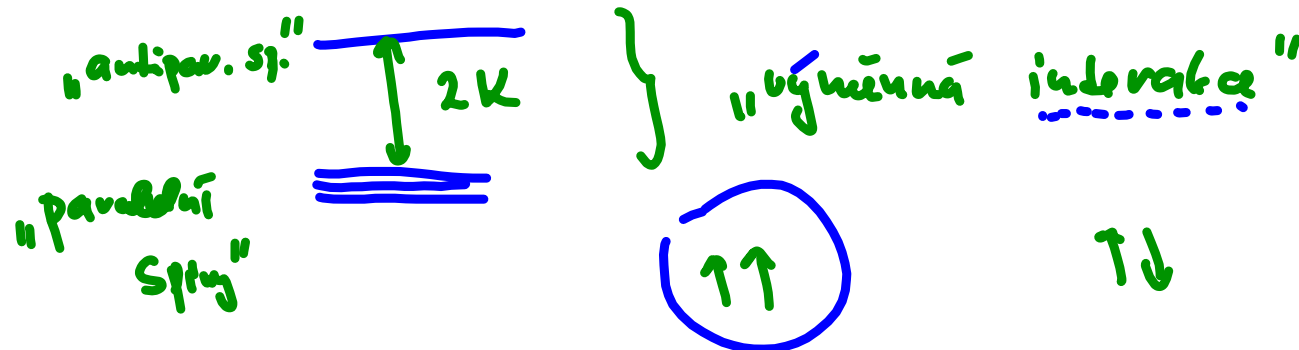


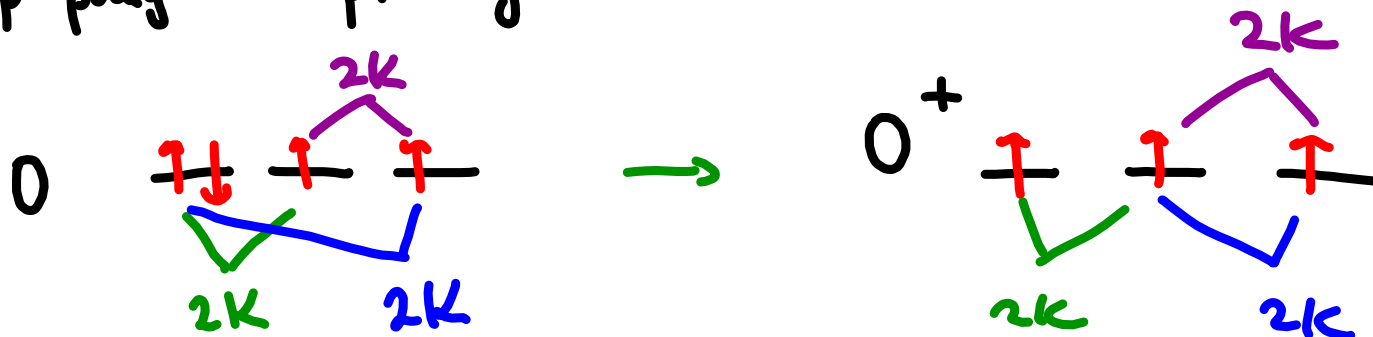
5-5 k je kladný, 2 rovnice pro

E_1 leží \equiv pod —
₃

a pro vzdálenost \equiv a — platí:



p-puky 2. periody



Σ vým. en. 6k

—||—

		vým. energie			
IP		neutrální	→	kation	Δ
EA		(anion)	←	(kation)	
B		0k		0k	0
C	(B ⁻)	2k		0k	2k
N	(C ⁻)	6k		2k	4k
O	(N ⁻)	6k		6k	0
F	(O ⁻)	8k		6k	2k
Ne	(F ⁻)	12k		8k	4k

11-5 Růvnice pro metodu Hartreeho vs. Hartree-Fock

Pi. Atom Li: A. TVAR VF
Hartree vs. Hartree-Fock

$$\Psi = 1s(1) \cdot \bar{1}s(2) \cdot 2s(3)$$

$$\Psi = \frac{1}{\sqrt{3!}} \begin{vmatrix} 1s(1) & 1s(2) & 1s(3) \\ \bar{1}s(1) & \bar{1}s(2) & \bar{1}s(3) \\ 2s(1) & 2s(2) & 2s(3) \end{vmatrix}$$

B. OPERATOR

Energie

—

—

$$\hat{H} \Psi = E \Psi$$

Douglas R. Hartree (1897-1958)

$$\hat{H} \Psi = E \Psi$$

Vladimir A. Fock (1898-1974)

$\Psi \dots AO$ (solitní)
 rovnice pro Ψ
 ↓
 Hartreeho rovnice pro AO
 ↓
 vystupuje u nich tzv. Hartreeho op.

obs. E_{kin} , $E_{pot.}$,
 Coulombův operátor
 ↳ J

Hartree
 Nabové: Energie AO (MO)

$\Psi \dots AO$ (Slater det.)
 rovnice pro Ψ
 ↓ Hartree-Fock
 rovnice pro AO
 ↓
 — || —
 tzv. Fockův operátor
 F
 obs. E_{kin} , $E_{pot.}$,
 Coulombův operátor → J
 a výměnný operátor → K
 Hartree-Fock (HF)
 a celková energie

Break 10:19 - 10:40

↓
led

Dokaz

Pokud.

Hartree-Fock

$$\hat{F} \phi_i = \epsilon_i \phi_i$$

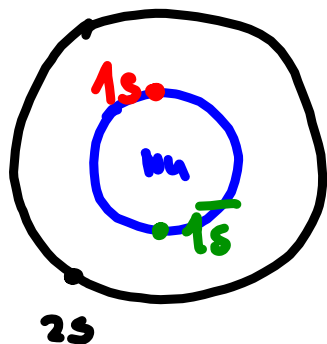
$$(11) \quad \hat{F}(1) = -\frac{1}{2} \nabla^2 - \sum_{\mu} \frac{Z_{\mu}}{r_{\mu 1}} + \sum_{j=1}^N (2\hat{J}_j - \hat{K}_j)$$

↓
elektronu

Coulombův operátor: $\hat{J}_j = \int \phi_j^*(2) \frac{1}{r_{12}} \phi_j(2) d\tau(2)$

Výměňný $\hat{K}_j \phi_i(1) = \int \phi_j^*(2) \frac{1}{r_{12}} \phi_i(2) d\tau(2) \phi_j(1)$

Pi. Atom Li (2s) $1s^2 2s^1$



$(1s)$ $\bar{1s}, 2s \dots$ approx. užit. e^-
(AO pro ion Li^{2+})

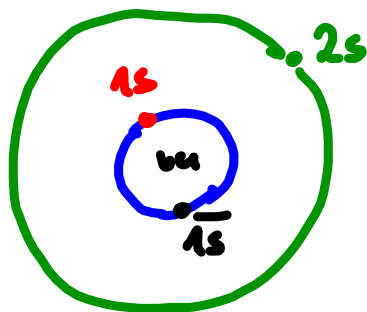
Chceme opravit na vepulze

$$\hat{J}_{\bar{1s}}(2) 1s(1) = \left[\int \bar{1s}(2) \frac{1}{r_{12}} \bar{1s}(2) d\tilde{\tau}_2 \right] 1s(1)$$

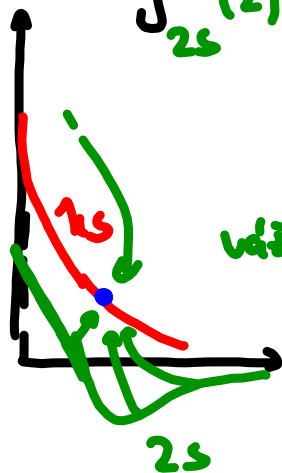
Hartree ✓
HF ✓

vděiče $\frac{1}{r_{12}}$ ke zrušení $\hat{J}_{\bar{1s}}$ poklebi
zčim $\bar{1s}$

Podobně započku uliv obsažen AO $\underline{2s}$ na AO $1s$



$$\hat{J}_{2s}(2) 1s(1) = \left[\int 2s^*(2) \frac{1}{r_{12}} 2s(2) d\tau_2 \right] 1s(1)$$

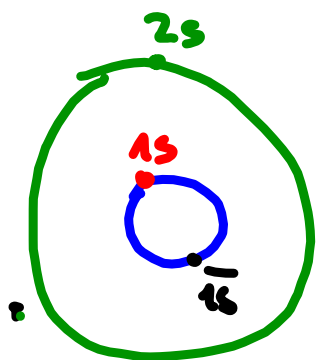


vděním $\frac{1}{r_{12}}$

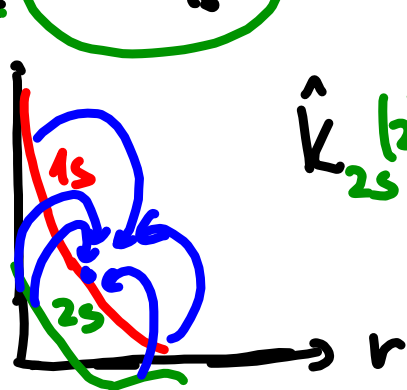
Harmon \cup
HF \cup

Pokračuji zúžit $2s_1$

chci-li „opracit“ $1s$



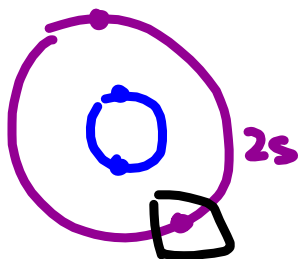
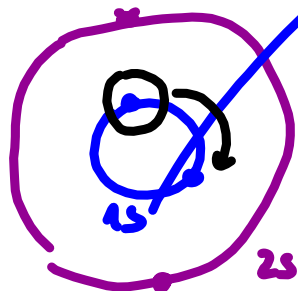
výměňový operátor „přisobí“ (středí své)
pouze v případě el. se stejným
spínem



$$\hat{K}_{2s}(2) 1s(1) = \left[\int 2s^*(2) \frac{1}{r_{12}} 1s(2) d\tau_2 \right] 2s(1)$$

chci-li opr. $1s$

11.7 Celková energie u HF - SCF

 ${}_2\text{Be}$ 

$$\epsilon_{1s} = H_{11} + J_{11} + 2J_{12} - K_{12}$$

\downarrow \downarrow
 E_{1s} E_{eff}

energetický příspěvek

do ϵ_{1s} díky umístění
2 el. v 1s

$$\epsilon_{2s} = H_{22} + J_{22} + 2J_{12} - K_{12}$$

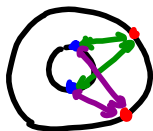
2s $\uparrow\downarrow$ 1s $\uparrow\downarrow$

$$2\epsilon_{1s} + 2\epsilon_{2s} = 2H_{11} + 2J_{11} + 4J_{12} - 2K_{12} \\ + 2H_{22} + 2J_{22} + 4J_{12} - 2K_{12}$$

Energie: Suma $2\varepsilon_1 + 2\varepsilon_2$ vs. $E_{\text{celk. pytl.}}$

$$2\varepsilon_1 + 2\varepsilon_2 = 2H_{11} + 2H_{22} + 2J_{11} + 2J_{22} + 8J_{12} - 4K_{12}$$

$$E_{\text{celk. F}} = 2H_{11} + 2H_{22} + J_{11} + J_{22} + 4J_{12} - 2K_{12}$$



$$\begin{aligned} E_{\text{celk. F}} &= 2\varepsilon_1 - (J_{11} + 2J_{12} - K_{12}) + 2\varepsilon_2 - (J_{22} + 2J_{12} - K_{12}) = \\ &= 2\varepsilon_1 - (2J_{11} - K_{11} + 2J_{12} - K_{12}) + 2\varepsilon_2 - (2J_{22} - K_{22} + 2J_{12} - K_{12}) = \end{aligned}$$

$J_{11} = K_{11}$
 $J_{22} = K_{22}$

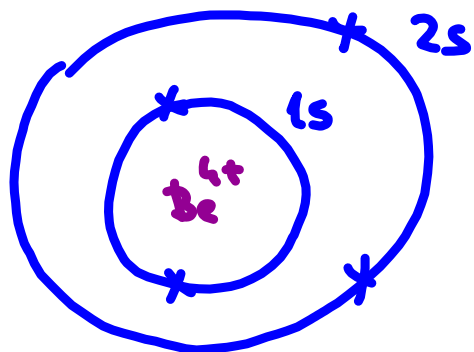
$$= 2\varepsilon_1 - \sum_{j=1}^2 (2J_{1j} - K_{1j}) + 2\varepsilon_2 - \sum_{j=1}^2 (2J_{2j} - K_{2j}) =$$

$$= \sum_{i=1}^2 \left[2\varepsilon_i - \sum_{j=1}^2 (2J_{ij} - K_{ij}) \right] \quad (11-17)$$

Suma
ob. energi

celkové energie v HF

Orbitály a Elektron



Orbital = VF pro 1 e⁻

Orbital vs. Spin-Orbital

