Importance of Analyte-matrix Separation in U-Pb and Pb-Pb Dating Systems

Jan DOBEŠ¹, Vladimír KISTAN¹, Markéta HOLÁ¹, Vojtěch WERTICH², Viktor KANICKÝ¹

¹Masaryk University, Department of Chemistry, Faculty of Science, Kotlářská 2, 611 37 Brno, Czech Republic

²Masaryk University, Department of Geological Sciences, Faculty of Science, Kotlářská 2, 611 37 Brno, Czech Republic

Dating is nowadays widely used analytical approach combing geology and analytical chemistry. It provides useful geochronological information in specific areas. Inductively coupled plasma mass spectrometry (ICP-MS) is still one of the common methods used for elemental and isotope ratios analysis [1]. The dating principle assumes that the age of a sample can be calculated from the ratio of two or more isotopes present in the sample. To determine the isotope ratios, sample preparation is crucial in order to eliminate interferences and improve accuracy and precision [2].

Besides ore mining, manual separation, sample digestion and dissolution, the critical step in the preparation procedure is analyte-matrix separation. Uranium and lead are usually separated from sample matrix by the principles of ion exchange chromatography using previously developed procedures [3, 4] or new approaches can be introduced. In this study, series of U and Pb standard solutions were used to optimize the separation procedure using commercially available resins. Different eluting agents were tested. After the method optimization, uraninite samples were analysed. For both elemental and isotopic analysis, solution method ICP-MS was used.

Key words: ICP-MS, uranium-lead dating system, ion exchange chromatography, dating

References:

[1] F. Vanhaecke, L. Balcaen, D. Malinovsky, Use of single-collector and multi-collector ICPmass spectrometry for isotopic analysis, Journal of Analytical Atomic Spectrometry 24(7) (2009) 863-886.

[2] Y.C. Yip, J.C.W. Lam, W.F. Tong, Applications of lead isotope ratio measurements, Trac-Trends in Analytical Chemistry 27(5) (2008) 460-480.

[3] S.M. Khopkar, A.K. De, CATION-EXCHANGE BEHAVIOUR OF URANIUM(VI) ON AMBERLITE IR-120 SEPARATION FROM MIXTURES, Analytica Chimica Acta 22(2) (1960) 153-158.

[4] F.W.E. Strelow, APPLICATION OF ION EXCHANGE CHROMATOGRAPHY TO ACCURATE DETERMINATION OF LEAD URANIUM AND THORIUM IN TANTALO-NIOBATES, Analytical Chemistry 39(12) (1967) 1454-&.