SHORT COURSE ICP – MASS SPECTROMETRY INSTRUMENTATION

<u>Frank Vanhaecke</u> Ghent University, Belgium











WHERE EXACTLY IS **B**ELGIUM?

Facts

- Area: 30,528 km²
- Population ~ 11,000,000
- Inhabitation density: 360 /km²
- Capital: Brussels
 - European Parliament
 - NATO headquarters
- Northern part: Flanders (Dutch)
- Southern part: Wallonia (French)

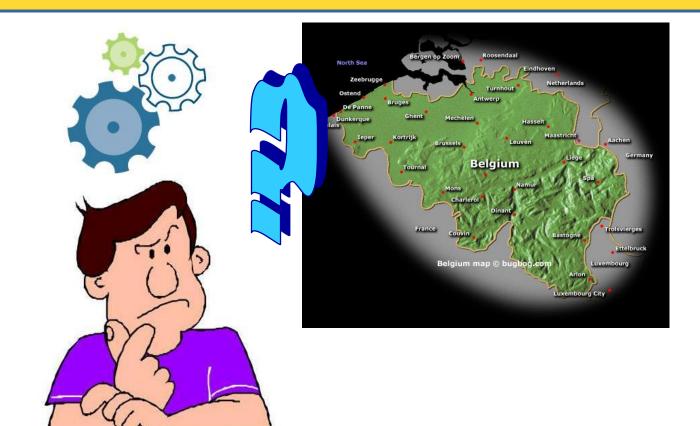








WHERE SHOULD I KNOW BELGIUM FROM?



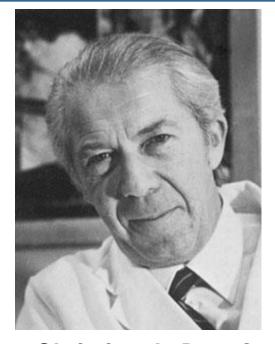






PERHAPS ... WHO KNOWS?





<u>Leo Baekeland</u>? Chemist, inventor "Father of plastics" Bakelite phenol-formaldehyde resin

<u>Christian de Duve</u>? Biochemist Nobel Prize for Medicine in 1974 Discovered lysosomes and peroxisomes as cell organelles







PERHAPS ... WHO KNOWS?



<u>Eddy Merckx</u>? 1960-70s, 5 times winner of Tour de France World Champion World hour record holder "The cannibal"



<u>Kim Clijsters</u>? Recently "retired" 3 times US open, 1 time Australian open Achieved nr. 1 world ranking







PERHAPS ... WHO KNOWS?





<u>Tintin</u>









WHERE SHOULD YOU KNOW BELGIUM FROM?

MICHAEL JACKSON'S I GREA **BEERS OF** BELGIUM THIRD EDITION Trappist beers – alcohol content: 6 – 12%

WHERE SHOULD YOU KNOW BELGIUM FROM?









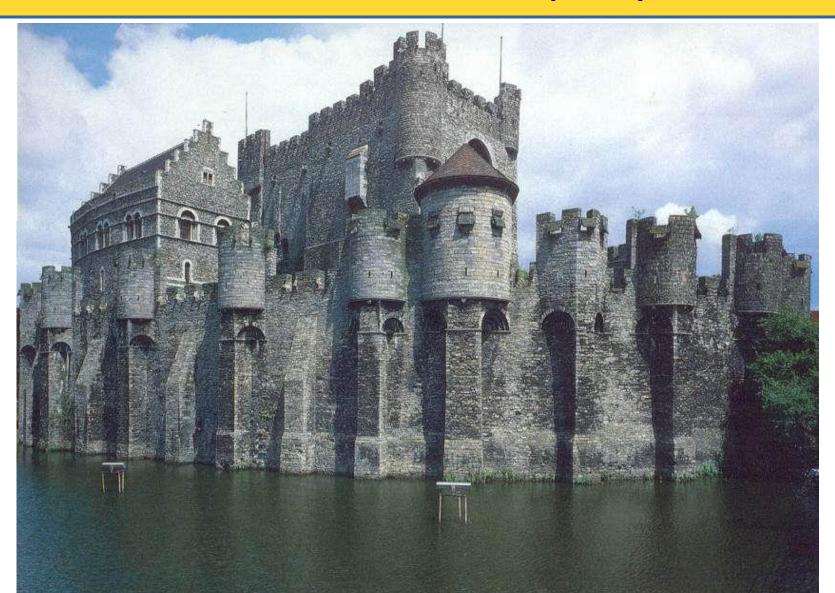
WHAT ABOUT GHENT?

• Wikipedia?

Ghent started as a settlement at the confluence of the Rivers <u>Scheldt</u> and <u>Lys</u> and became in the <u>Middle Ages</u> one of the largest and richest cities of northern <u>Europe</u>. Today it is a busy city with a <u>port</u> and a <u>university</u>.



THE MIDDLE AGES IN GHENT GRAVENSTEEN CASTLE (1180)



THE MIDDLE AGES IN GHENT FRIDAY'S MARKET SINCE 1199 !







THE MIDDLE AGES IN GHENT GRASLEI – MEDIAEVAL PORT



THE MIDDLE AGES IN GHENT THE BELFRY & GOTHIC CHURCHES



GHENT UNIVERSITY °1817 - ~38,000 STUDENTS & ~7,000 STAFF MEMBERS



DEPARTMENT OF ANALYTICAL CHEMISTRY



ATOMIC & MASS SPECTROMETRY RESEARCH GROUP A&MS









INDUCTIVELY COUPLED PLASMA – MASS SPECTROMETRY ICP-MS

What is ICP-MS?

Wikipedia?

- Inductively coupled plasma mass spectrometry (ICP-MS) is a type of mass spectrometry which is capable of detecting metals and several non-metals at concentrations as low as one part in 10¹² (part per trillion). This is achieved by ionizing the sample with inductively coupled plasma and then using a mass spectrometer to separate and quantify those ions.
- Compared to <u>atomic absorption techniques</u>, ICP-MS has greater speed, precision, and sensitivity. However, analysis by ICP-MS is also more susceptible to trace contaminants from <u>glassware</u> and reagents. In addition, the presence of some ions can interfere with the detection of other ions.
- A powerful technique for the determination of (ultra)trace elements



ICP – MASS SPECTROMETRY

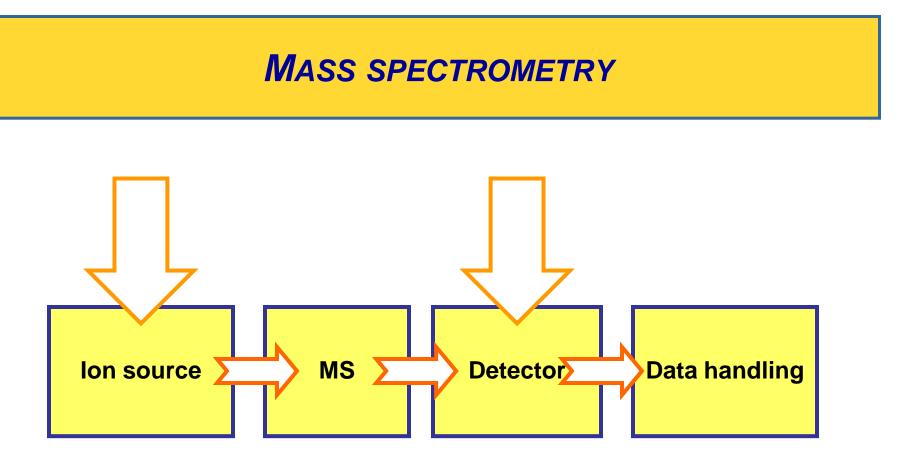
- ° 1983, at present: thousands of instruments in use
- Research tool \rightarrow more robust & well-established technique
- <u>Advantages</u>
 - Low limits of detection
 - Multi-element capabilities
 - Wide linear dynamic range
 - High sample throughput
 - Relatively simple spectra
 - Ability to obtain isotopic information
 - Ease of combination with
 - Alternative sample introduction systems
 - Chromatographic separation techniques













High vacuum region







THE INDUCTIVELY COUPLED PLASMA ICP



INDUCTIVELY COUPLED PLASMA – ICP

PLASMA = GAS MIXTURE AT HIGH TEMPERATURE, CONTAINING MOLECULES, ATOMS, IONS AND ELECTRONS

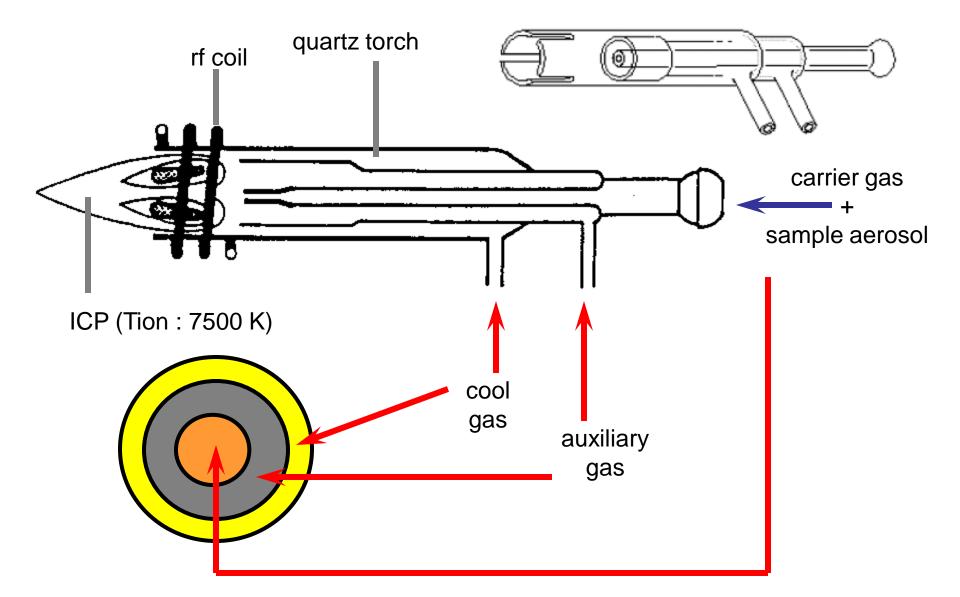
Presence of charged particles
 Energy supply via induction



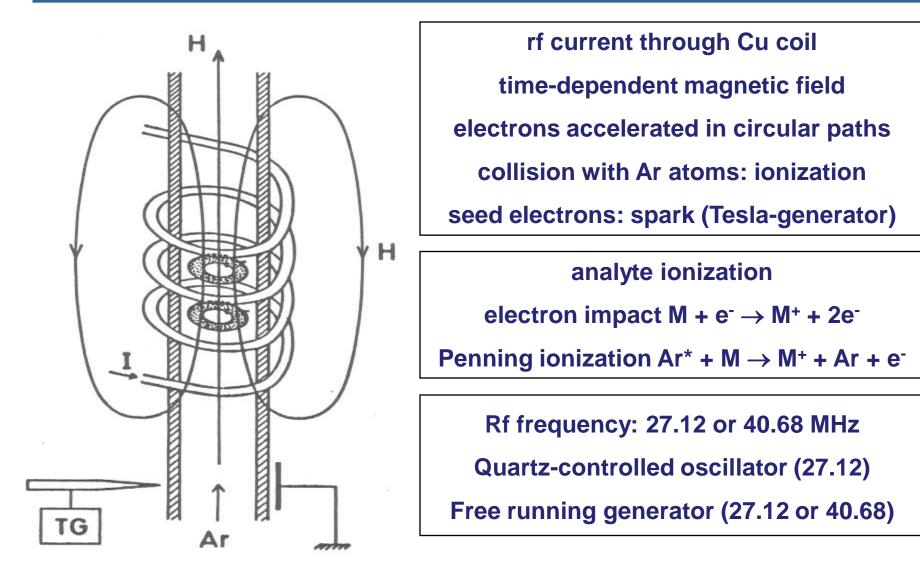




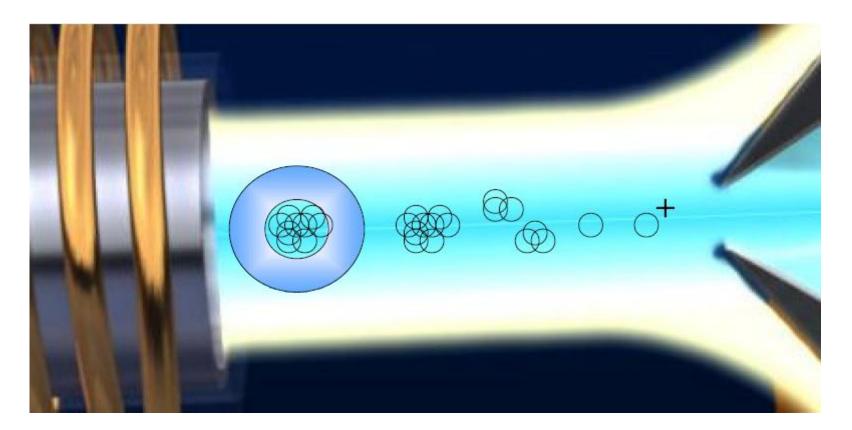
PLASMA TORCH & ICP



FORMATION OF ICP & ANALYTE IONIZATION



PROCESSES IN THE ICP SAMPLE INTRODUCTION VIA NEBULIZATION



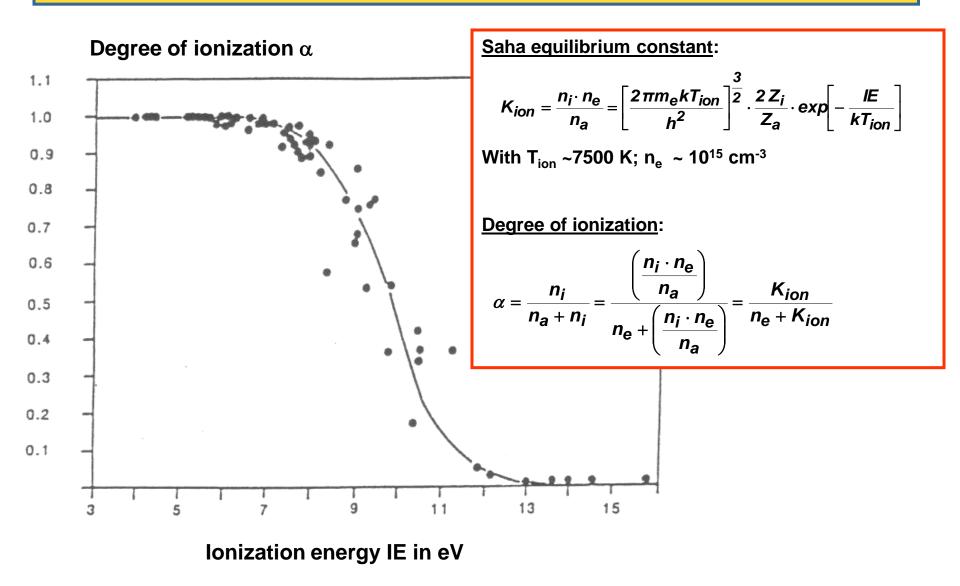
Residence time of analyte in ICP: ~ ms







IONIZATION EFFICIENCY OF THE ICP



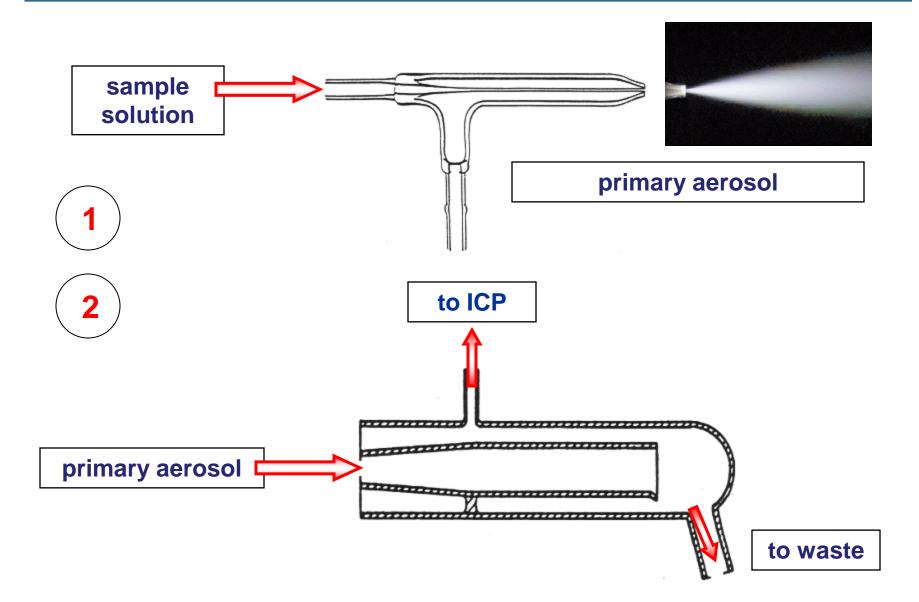
SAMPLE INTRODUCTION INTO THE ICP

- General
 - Sample or representative part \rightarrow ICP
 - Convert sample into form transportable by Ar carrier gas
- Standard sample introduction system
 - Pneumatic nebulizer + spray chamber
- Pneumatic nebulizer
 - Conversion of sample solution into aerosol
- Spray chamber
 - Removal of larger droplets
 - Avoid plasma overloading
 - Warrant efficient atomization & ionization in ICP





STANDARD SAMPLE INTRODUCTION SYSTEM PNEUMATIC NEBULIZER & SPRAY CHAMBER

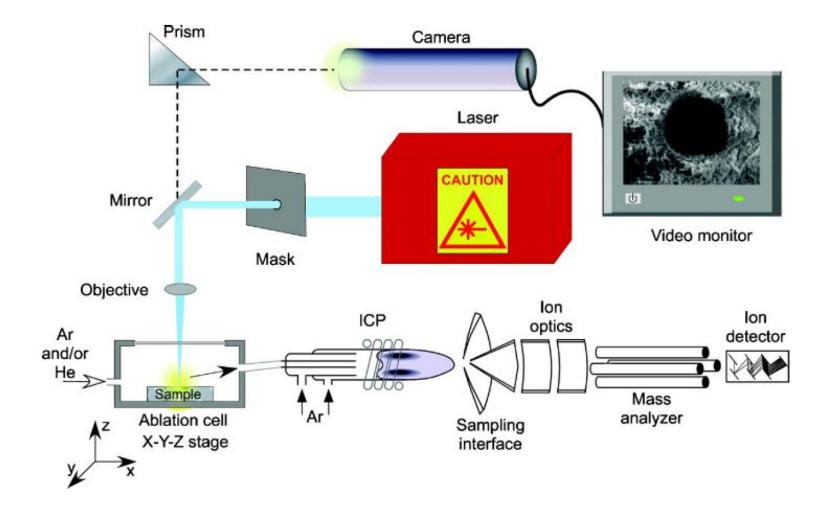


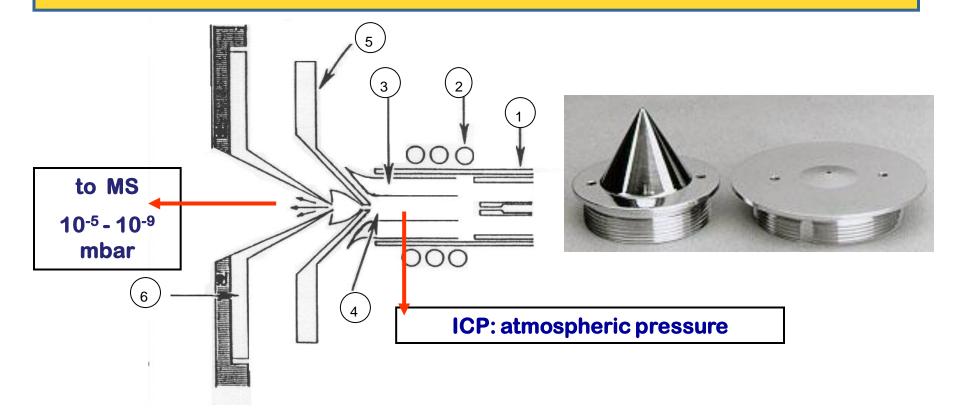
LASER ABLATION

AS A MEANS OF SAMPLE INTRODUCTION

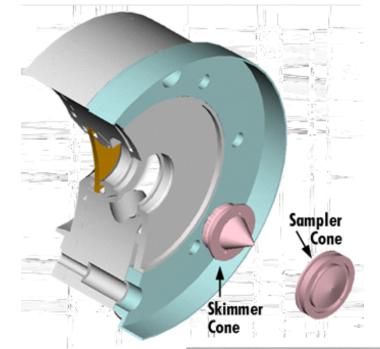
• Direct analysis of solid materials

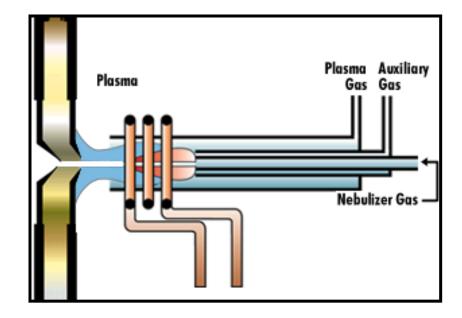
Conducting & non-conducting / Opaque & transparent





- Sampling cone & skimmer, central aperture ~1 mm
- Expansion chamber (1 mbar) \Rightarrow supersonic expansion of extracted gas
 - Composition of plasma gas is 'frozen'
- Central beam via skimmer aperture \Rightarrow lens system & MS





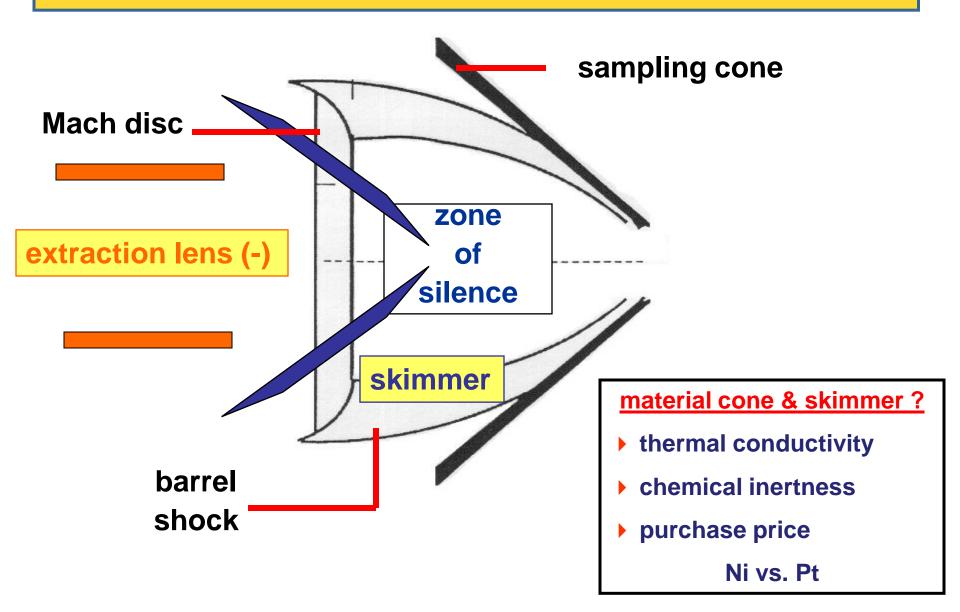








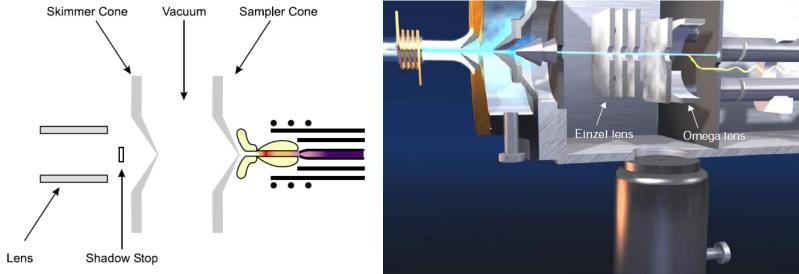
T_{ion} ~7500 K, n_e- ~10¹⁵ cm⁻³ Sampled via 2-cone interface



LENS SYSTEM

Set-up & complexity

from a single lens to complicated set-ups ...

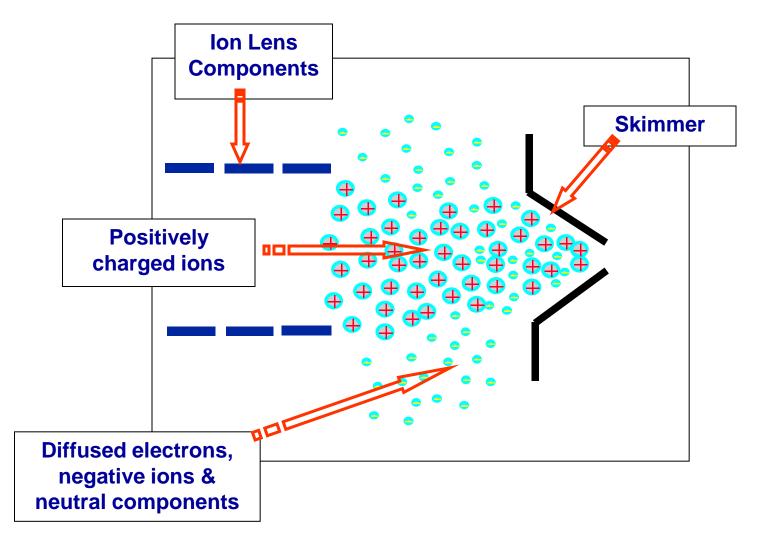


• Goals?

- Selection of positive ions
- Efficient transport to & introduction into mass analyser



SELECTION OF POSITIVE IONS IN THE LENS SYSTEM



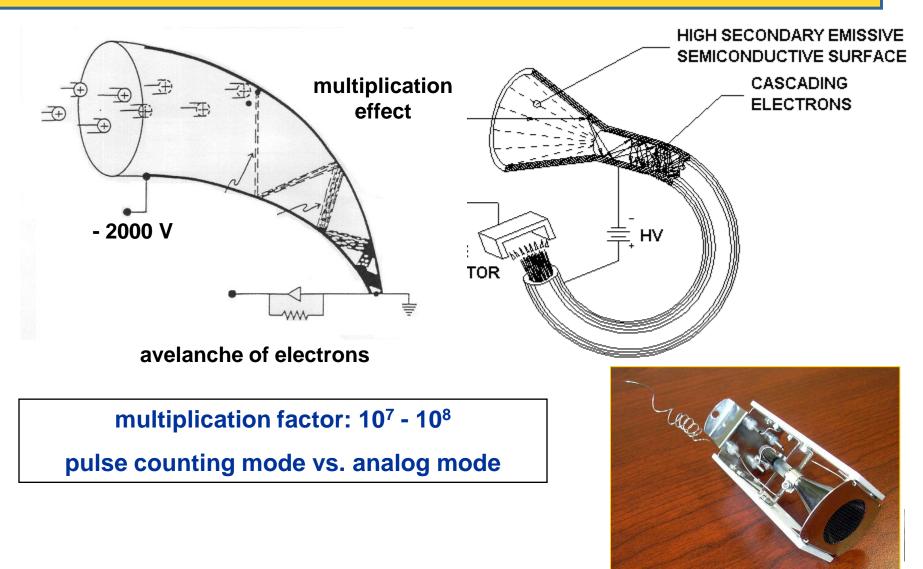






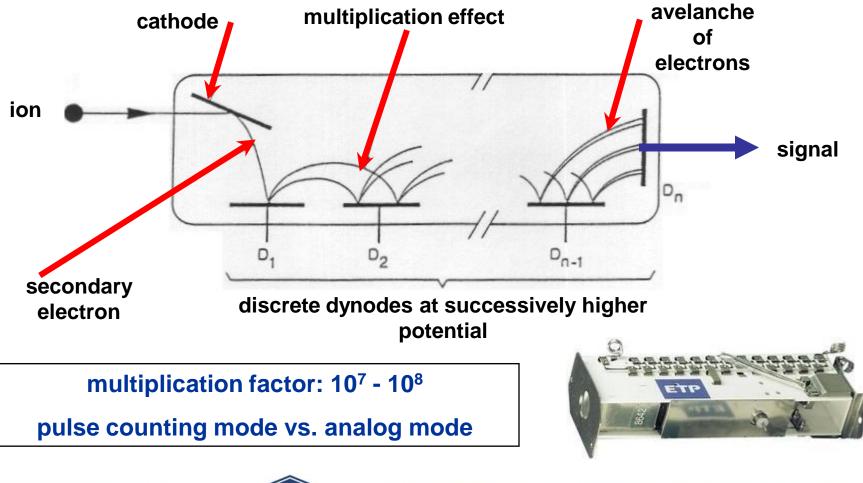


ION DETECTION CONTINUOUS DYNODE ELECTRON MULTIPLIER



ION DETECTION

DISCRETE DYNODE ELECTRON MULTIPLIER







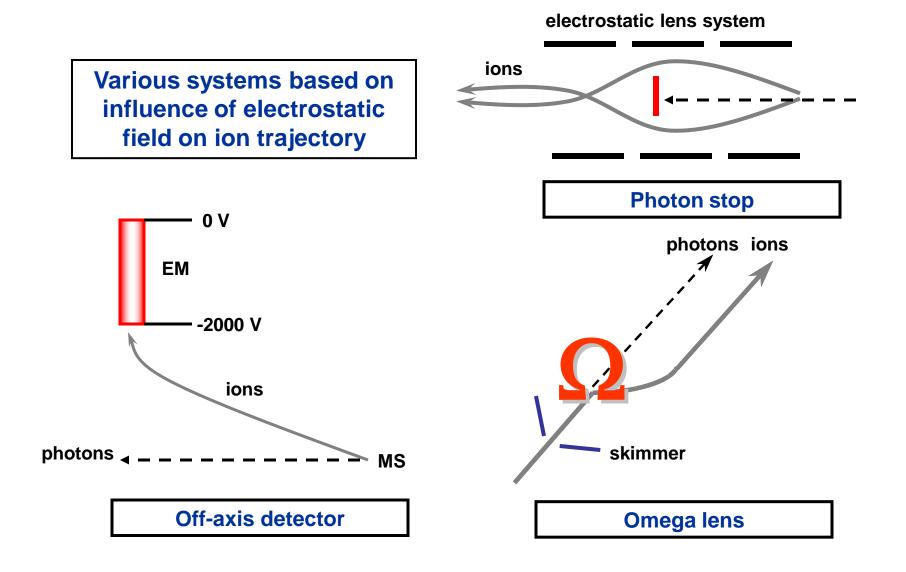
ION DETECTION / ELECTRON MULTIPLIER

- Comparison of detector signal with threshold value
 - Background < 0.1 count / s</p>
- Avoid photons reaching detector
 - Photon stop in ion beam
 - Detector mounted off-axis
 - Ion lens system, quadrupole filter & detector off axis
 - Omega-lens (Agilent)
- Limited life-time (= consumable)
 - 1 2 years
- Detector dead time
 - Handling of one ion, no possibility to detect another one
 - More pronounced effects at higher count rates
 - Accurate isotope ratio determination requires correction

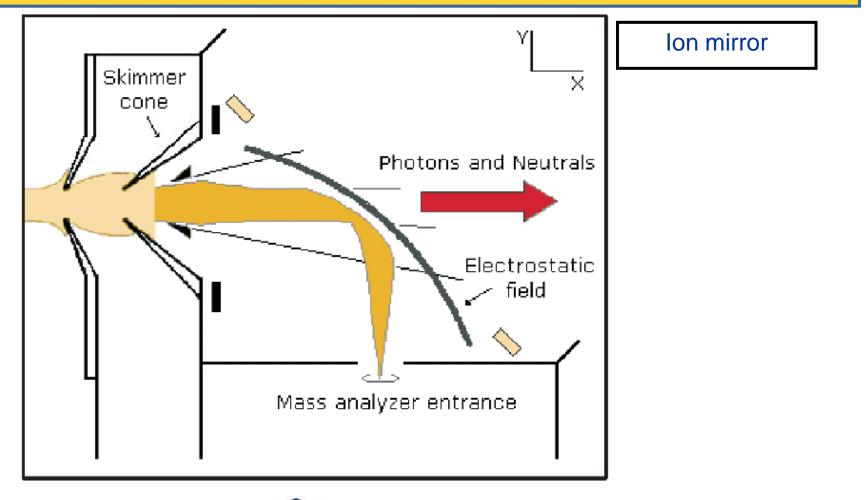




PREVENTING PHOTONS FROM REACHING THE DETECTOR



PREVENTING PHOTONS FROM REACHING THE DETECTOR





UNIVERSITEIT GENT

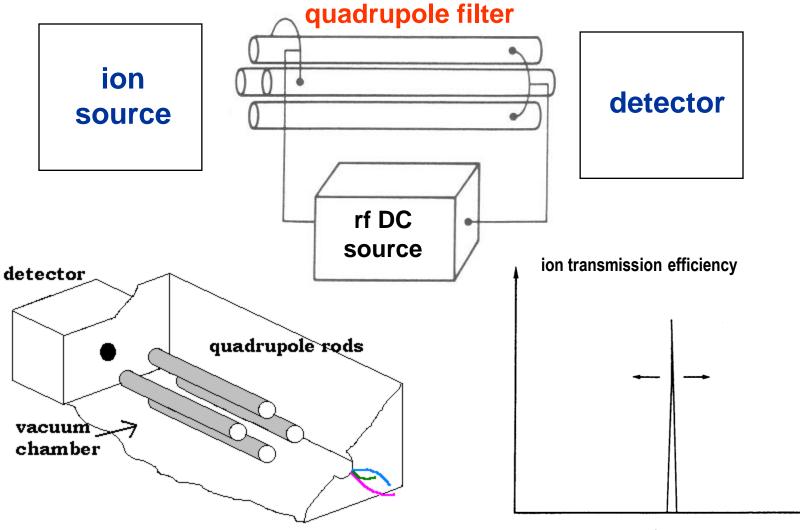
THE MASS SPECTROMETER





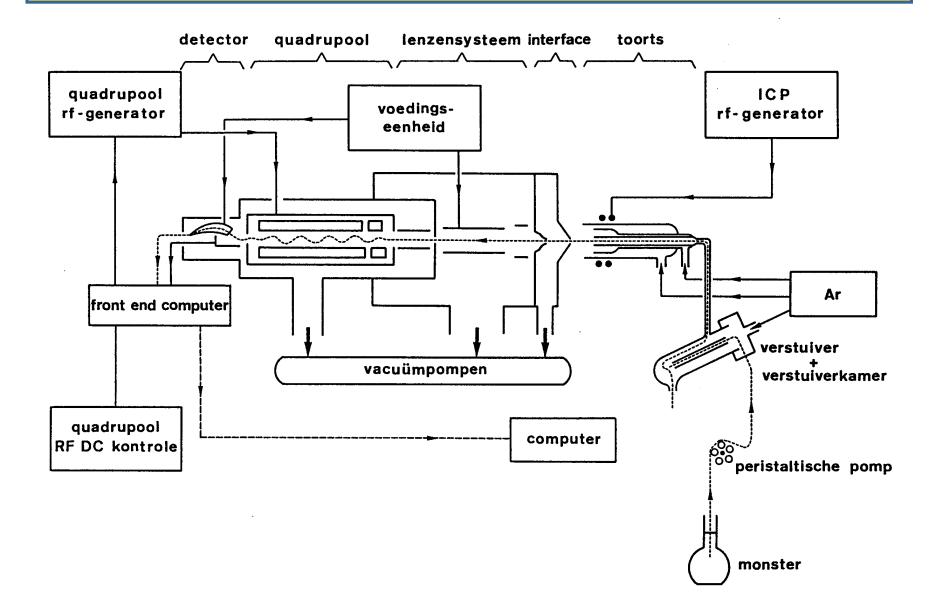


MASS ANALYSIS – THE QUADRUPOLE FILTER



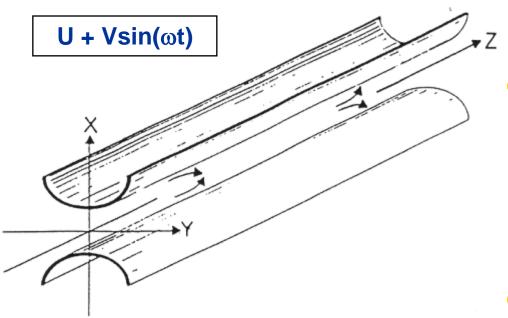
mass/charge ratio

QUADRUPOLE-BASED ICP – MASS SPECTROMETRY





each pair of quadrupole rods on ion paths

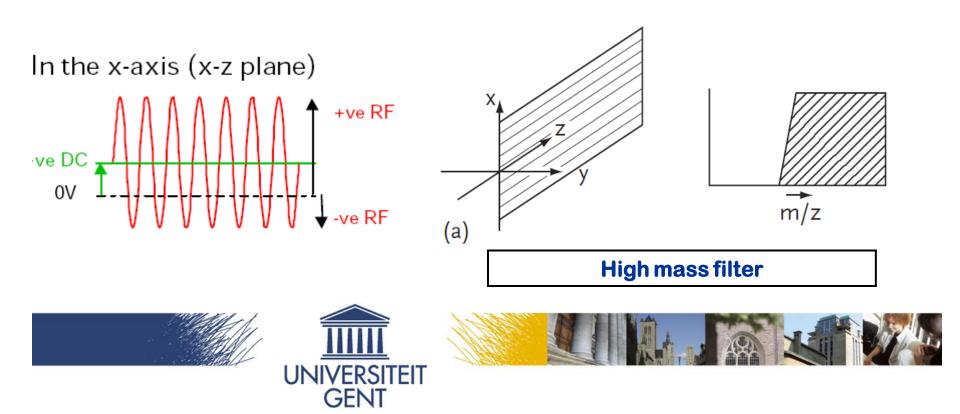


Electrode potential XZ vs. YZ

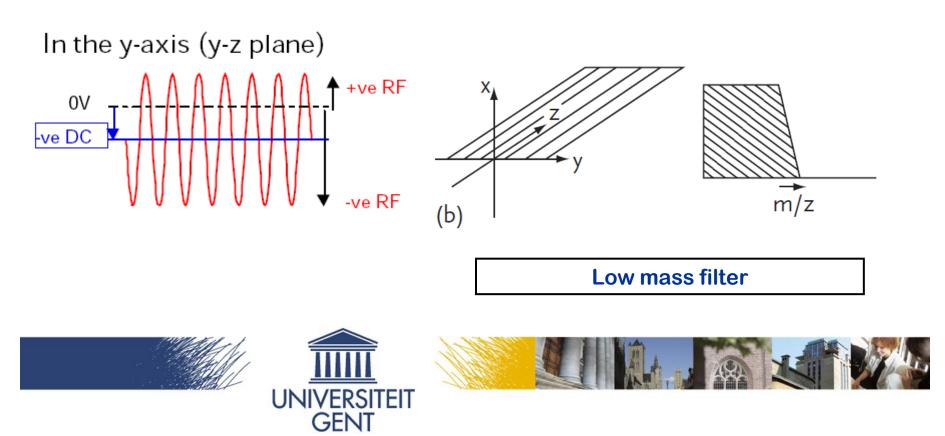
equal magnitude, different polarity

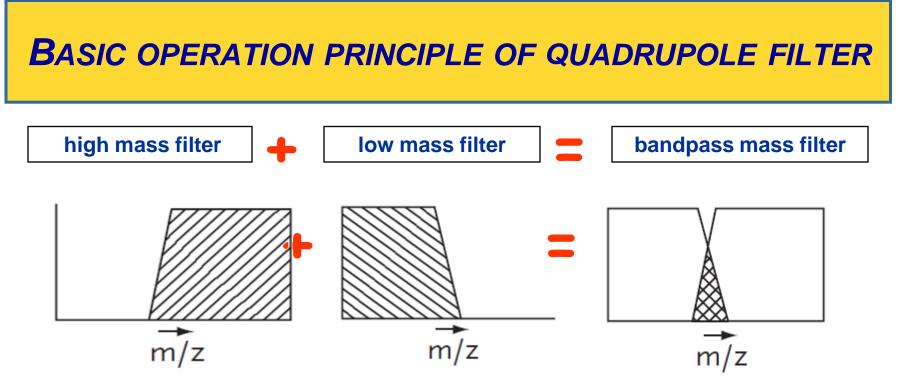
- Quadrupole rods
 - DC component
 - AC (rf) component
- Instable path ?
 - Magnitude actual negative potential
 - frequency of AC component
 - Position of ion
 - Velocity of ion
 - Mass-to-charge ratio of ion
- Heavy ions
 - Average (DC) potential
- Lighter ions
 - Motion corrected by AC field

- XZ-plane
 - DC component : +
 - Sufficiently heavy ions only undergo focusing effect (DC)
 - Path of lighter ions affected by AC-field

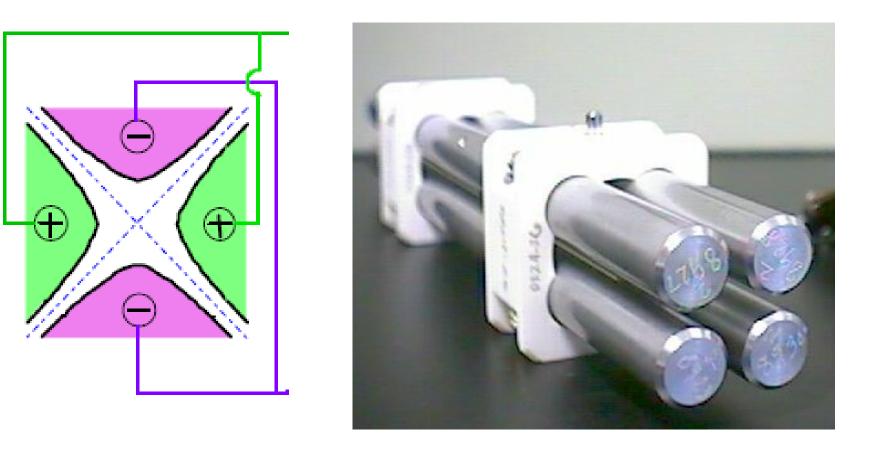


- YZ-plane
 - DC component : -
 - Sufficiently heavy ions only undergo defocusing effect (DC)
 - Path of lighter ions affected by AC-field





- Advantages
 - High scanning speed
 - Can be used at relatively high pressures
 - Instrumental simplicity & low purchase price
- Disadvantages
 - Low mass resolution (R)
 - lons only separated if ∆m ≥ 1/2 u
- Data acquisition modes
 - Spectral scanning vs. peak hopping / peak jumping









SPECTRAL INTERFERENCES IN QUADRUPOLE-BASED ICP-MS

- Disadvantage of quadrupole filter
 - Low mass resolution (R)
 - lons separated if ∆m ≥ 1/2 u
 - Otherwise: overlap of ion signals (spectral interferences)
- Overlap of signals of isobaric nuclides
 - Not problematic
 - For every element (except In): one isotope interference-free
- Polyatomic ions & doubly charged ions
 - Spectral overlap of ions showing the same <u>nominal</u> m/z ratio

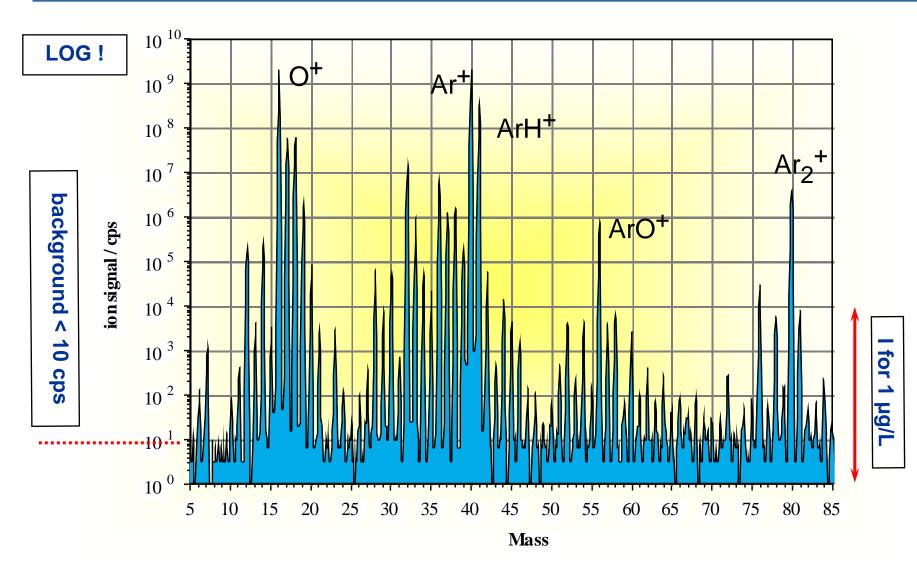


TYPES OF POLYATOMIC IONS IN ICP-MS

- Ar-containing ions
 - Ar introduced in ICP at ~ 20 L/min
 - Ar⁺ and Ar₂⁺
 - Ar + elements from solvent, surrounding air and/or matrix
 - ArO+, ArOH+, ArN+, ArC+, ArCl+, ArNa+, ...



ICP-MS BACKGROUND SPECTRUM FOR HIGH-PURITY WATER



ICP-MS BACKGROUND SPECTRUM FOR HIGH-PURITY WATER

Mass	Molecular ion	Interferes with
28	N ₂ +	Si
29	N,H⁺	Si
30	NO⁺	Si
31	NOH⁺	Р
32	0 ₂ +	S
33	O₂H⁺	S
39	³⁸ ArH⁺	К
40	40Ar+	Ca
41	⁴⁰ArH⁺	Ca
44	CO ₂ +	Ca
54	⁴⁰ ArN⁺	Fe, Cr
55	⁴⁰ArNH⁺	Mn
56	⁴⁰ ArO⁺	Fe
57	₄⁰ArOH⁺	Fe
76	⁴⁰ Ar ³⁶ Ar⁺	Se
78	40Ar38Ar+	Se
80	40 Ar ₂ +	Se



TYPES OF POLYATOMIC IONS IN ICP-MS

- Ar-containing ions
 - Ar introduced in ICP at ~ 20 L/min
 - Ar⁺ and Ar₂⁺
 - Ar + elements from solvent, surrounding air and/or matrix
 - ArO+, ArOH+, ArN+, ArC+, ArCl+, ArNa+, ...
- Oxide & hydroxide ions
 - ▶ *MO*⁺ (*m*/*z* + 16) and *MOH*⁺ (*m*/*z* + 17)
 - MO+/M+ determined by M-O bond strength
 - Usually MO+/M+ > MOH+/M+
 - ► Optimization of instrument settings MO+/M+ ≤ 0.05 (5%)
 - Still problematic if $m/z(M_1O^+) = m/z(M_2^+)$ and $c(M_1) >> c(M_2)$
 - Formed in ICP : low t° in neighbourhood of vaporizing droplets
- Other molecular ions
 - SO₂⁺, SO₂H⁺, SiCl⁺, ...



DOUBLY-CHARGED IONS IN ICP-MS

- Doubly charged ions
 - ▶ *M*²⁺ (*m*/z ÷ 2)
 - \blacktriangleright M^{2+}/M^{+} determined by $(IP_2 IP_1)$
 - ► Optimization of instrument settings M²⁺/M⁺ ≤ 0.05 (5%)
 - Still problematic if $m/z(M_1^{2+}) = m/z(M_2^{+})$ and $c(M_1) >> c(M_2)$
 - ▶ if m/z of $M_1 = uneven \Rightarrow M_1^{2+}$ is no problem
 - *M*₁²⁺ signal at non-integral m/z
 - Quadrupole filter shows sufficient resolution
 - Formed in ICP





Some IMPORTANT SPECTRAL INTERFERENCES DUE TO THE PRESENCE OF CHLORINE

Matrix Element	Mass	Molecular ion	Interferes with
Chlorine (Cl)	51	³⁵ CIO⁺	V
e.g. HCI, HCIO ₄	52	³⁵ CIOH+	Cr
Cŀ, ClO₄-	53	³⁷ CIO+	Cr
35	54	³⁷ CIOH+	Cr, Fe
CI	75	⁴⁰ Ar ³⁵ Cl⁺	As
CHLORINE			



17





Some IMPORTANT SPECTRAL INTERFERENCES DUE TO THE PRESENCE OF CARBON

Matrix Element	Mass	Molecular In ion	terferes with
Carbon (C)	24	C_2^+	Mg
e.g. organics, (CO ₂ , 25	C ₂ H⁺, ¹² C ¹³ C⁺	Mg
CO₃=	26	CN⁺	Mg
12	28	CO+	Si
	44	CO,+	Ca
	45	CO₂H⁺, ¹ 3CO₂⁺	Sc
CARBON 6	52	ArC⁺	Cr





2.2.2

Some IMPORTANT SPECTRAL INTERFERENCES DUE TO THE PRESENCE OF SULPHUR

Matrix Element	Mass	Molecular In ion	terferes with
Sulfur (S)	48	³² SO+	Ti
e.g. H₂SO₄, SO₄=	49	³² SOH⁺	Ti
	50	³⁴ SO+	Ti, V
32	51	³₄SOH⁺	V
C	64	³² S ₂ ⁺ , ³² SO ₂ ⁺	Zn
	70	³⁸ Ar ³² S+	Ge
SULPHUR 16	72	⁴⁰ Ar ³² S⁺, ³⁸ Ar ³⁴ S⁺	Ge
19	74	40Ar34S+	Ge







Some IMPORTANT SPECTRAL INTERFERENCES DUE TO THE PRESENCE OF PHOSPHORUS

Matrix Element	Mass	Molecular ion	Interferes with
Phosphorus (P)	47	PO⁺	Ti
e.g. H ₃ PO ₄ , PO ₄ ³⁻	48	POH⁺	Ti
31	63	PO₂⁺ ArP⁺	Cu
D	71	ArP+	Ge
PHOSPHOROUS 15			



SPECTRAL INTERFERENCES IN ICP-MS











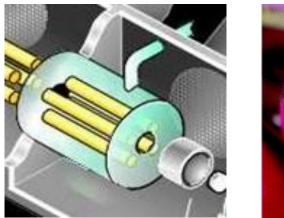
COLLISION/REACTION CELLS

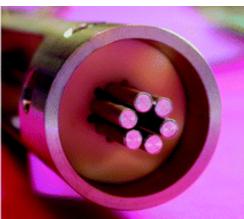
Multipole assembly with (2n + 2) rods:

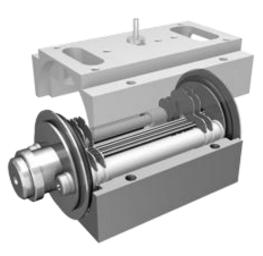
- Quadrupole
- Hexapole
- Octopole

Quadrupole cell Perkin Elmer Hexapole cell Thermo Scientific Xseries II

Octopole cell Agilent Technologies







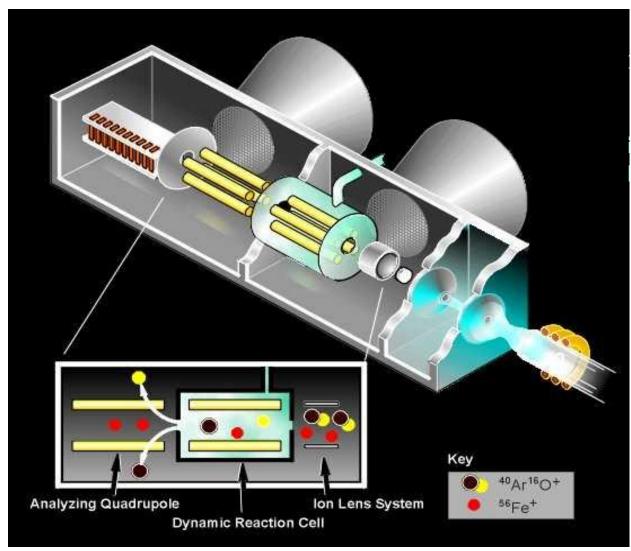






Use of a collision/reaction cell IN QUADRUPOLE-BASED ICP-MS: GENERAL CONCEPT

• Analyte & interfering ions in / analyte ions out



DIFFERENCES BETWEEN

DIFFERENT TYPES OF COLLISION/REACTION CELLS?

• Hexapole & octopole cell

Guide ions from point a to point b

• Quadrupole cell

Guides ions from point a to point b

Can be used as a mass filter



Slightly different ways of application







PERKIN ELMER DYNAMIC REACTION CELL – DRC

Typical use: highly reactive gases

- Determination of Fe
 - ${}^{40}Ar^{16}O^+ + NH_3 \rightarrow {}^{40}Ar^{16}O + NH_3^+$
 - ⁵⁶Fe⁺ + $NH_3 \rightarrow NO$ reaction
- Determination of Fe
 - ${}^{40}Ar^{16}O^+ + CO \rightarrow {}^{40}Ar^+ + CO_2$
 - ${}^{56}Fe^+ + CO \rightarrow NO$ reaction
- Determination of Pd
 - ${}^{90}ZrO^+ + O_2 \rightarrow {}^{90}ZrO_2^+$
 - ${}^{106}Pd^+ + O_2 \rightarrow NO$ reaction
- Determination of S
 - ${}^{32}S^+ + O_2 \rightarrow {}^{32}SO^+ + O$
 - ${}^{16}O_2^+ + O_2 \rightarrow NO$ reaction

charge transfer to reaction gas

atom transfer to reaction gas

atom transfer from reaction gas

atom transfer from reaction gas

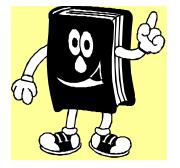






PERKIN ELMER DYNAMIC REACTION CELL – DRC THEORETICAL CONSIDERATIONS

- Reaction is thermodynamically allowed ($\Delta G < 0$)
 - ► ~ exothermic reaction (△H < 0)</p>
 - ion-molecule reaction MAY proceed
 - is usually fast
- Reaction is thermodynamically not allowed (∆G > 0)
 - ► ~ endothermic reaction (△H > 0)
 - ion-molecule reaction will not proceed
 - no energy is being supplied
- Consultation of thermodynamic / kinetic data

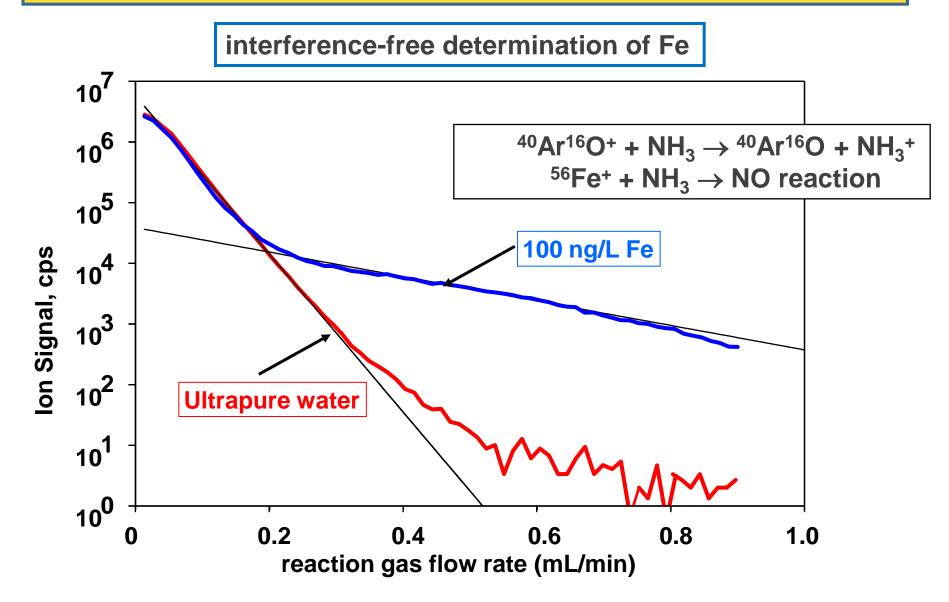




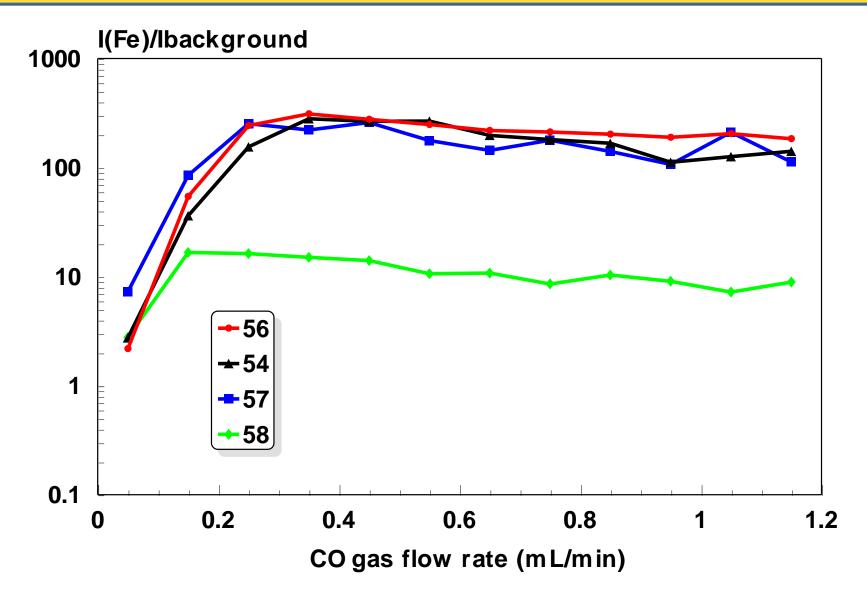




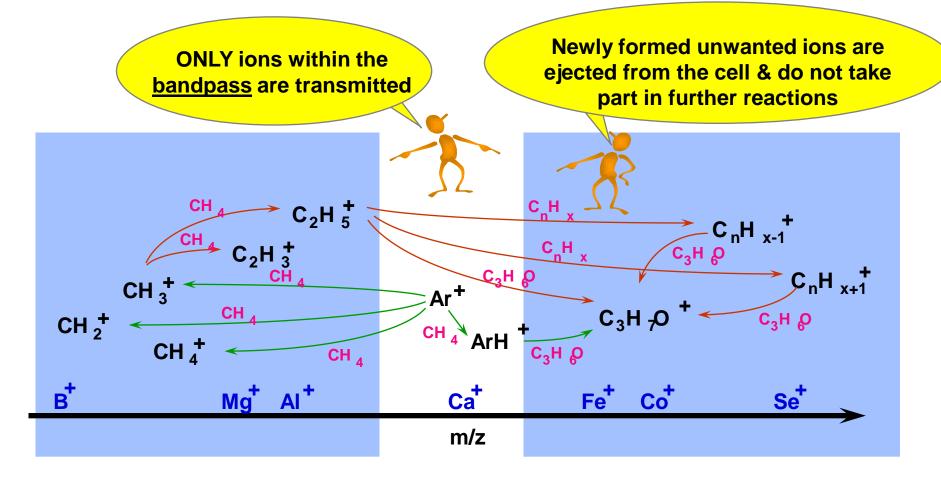
PERKIN ELMER DYNAMIC REACTION CELL – DRC OPTIMIZATION OF REACTION GAS FLOW RATE



PERKIN ELMER DYNAMIC REACTION CELL – DRC OPTIMIZATION OF REACTION GAS FLOW RATE



SIDE REACTIONS WHEN USING REACTIVE GASES ? TACKLED BY USING QUADRUPOLE AS MASS FILTER



Bandpass can be shifted in synchronicity with the mass analyzer

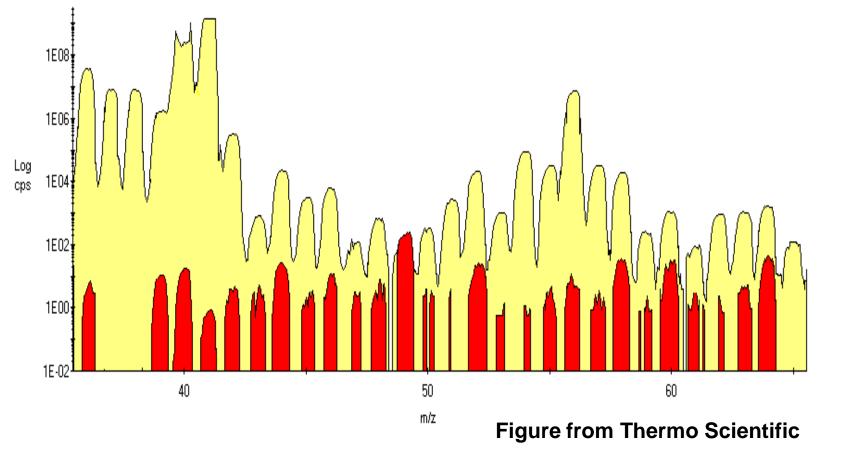
Figure from Scott Tanner / Sciex

SIDE REACTIONS WHEN USING REACTIVE GASES ? TACKLED BY KINETIC ENERGY DISCRIMINATION

- Ions produced inside the hexapole / octopole cell
 Velocity ~ 0
- Ions extracted from ICP
 Higher velocity
 - selection of ions according to their velocity or E_{kin}
 Accomplished via decelerating potential = energy filter
 'Kinetic energy discrimination'



BACKGROUND SPECTRUM WITH / WITHOUT H_2/He pressurized hexapole collision cell



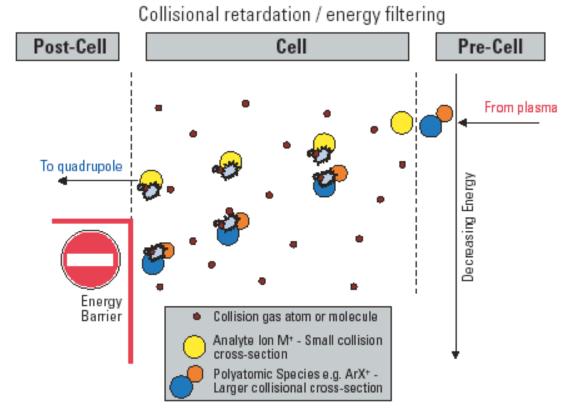






ALTERNATIVE USE OF KINETIC ENERGY DISCRIMNATION USE OF INERT GAS (HELIUM) ONLY

CCT^{ED} - Kinetic Energy Discrimination

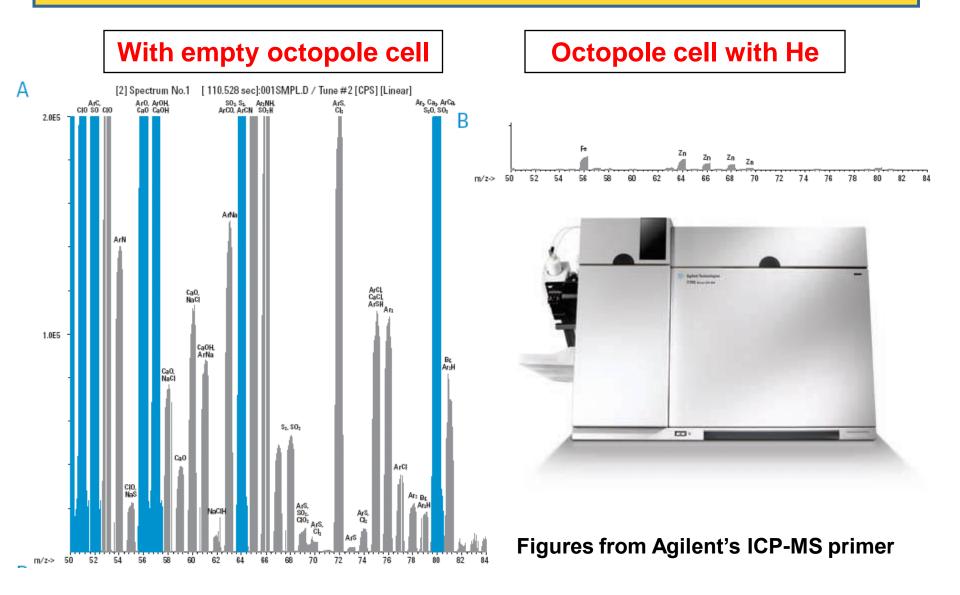


• Molecular ions are larger and thus, undergo more collisions

Loose more E_{kin} than atomic ions

Energy barrier selectively discriminates against molecular ions

BACKGROUND SPECTRUM WITH / WITHOUT HE-ONLY PRESSURIZED OCTOPOLE COLLISION CELL



PRESENT-DAY QUADRUPOLE-BASED ICP-MS INSTRUMENTS

Agilent 7700

- Octopole Collision reaction cell
- Preferred modus of operation
 - Use of He as inert collision gas & kinetic energy discrimination



- Perkin Elmer Nexion
 - Quadrupole reaction cell
 - Preferred modus of operation
 - Use of He as inert collision gas & kinetic energy discrimination
 - Use of reaction gases for selective ion-molecule reaction



PERKIN-ELMER NEXION UNIVERSAL CELL TECHNOLOGY

• Use as Q-based reaction cell with mass filtering capabilities

and

• Use as collision cell with kinetic energy discrimination

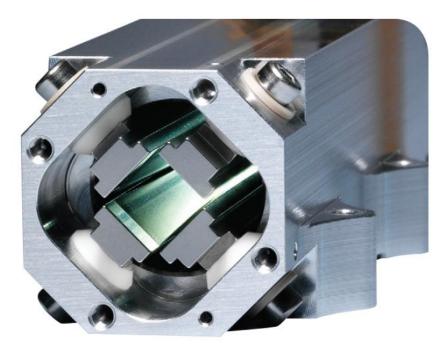


Figure from Perkin Elmer



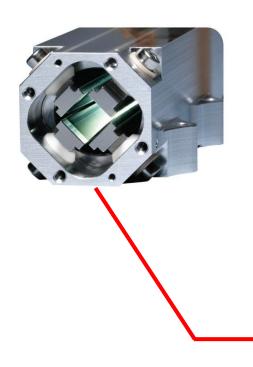


- ThermoScientific ICAP-Q
 - Flatapole reaction cell
 - Preferred modus of operation
 - Use of He as inert collision gas & kinetic energy discrimination
 - Use of reaction gases for selective ion-molecule reaction





ThermoScientific ICAP-Q



Vertical design for a small footprint and easy access to all components Completely new electronics for optimum scan speeds and class-leading reliability

Thermo

switching; flatapole technology for improv ion transmission and reliable interference reduction with He.

RAPID lens for optimum ion focusing and transmission

Efficient heat management for reduced installation costs and reliable operation

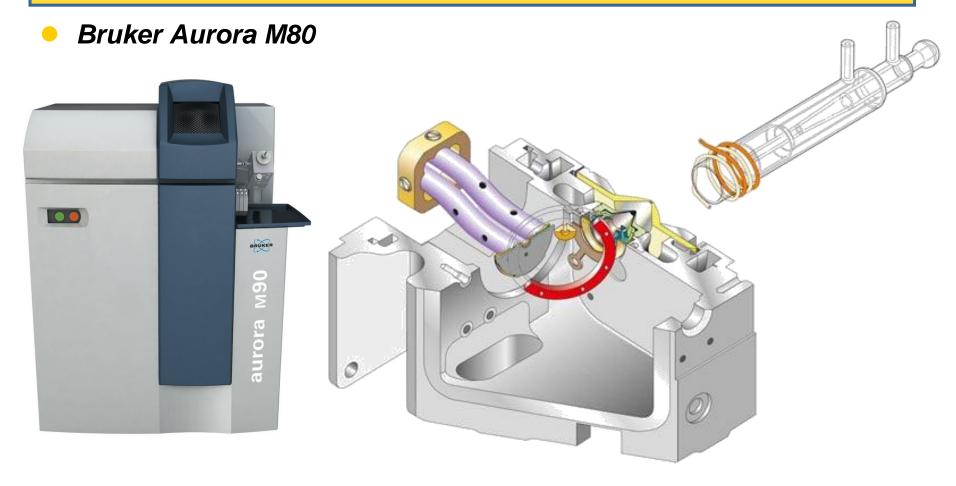
iCAP Q

New SEM with >9 orders of linear dynamic range and long lifetime

New interface for high matrix robustness, performance and long term reliability

Lightning fast switching RF generator without requiring a grounded shield New modular, bench-height sample introduction system with no complicated connections for easy handling independent of analyst skill level

Simple, bench-height access to cones and lenses









Collision/reaction interface – CRI Bruker Aurora M90

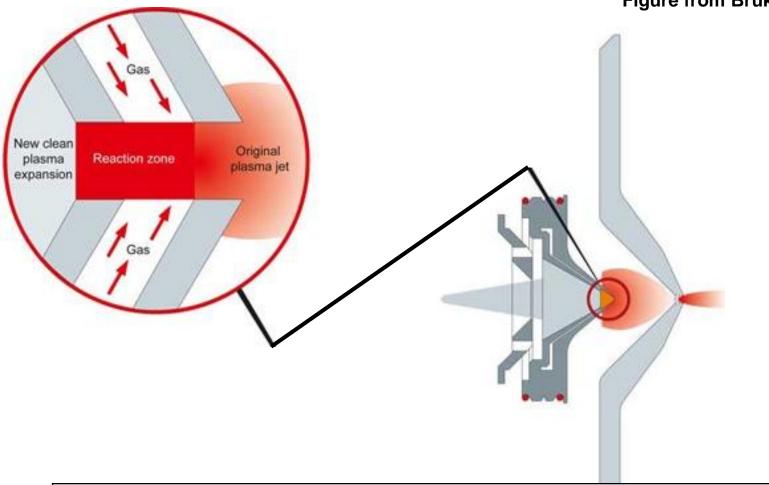
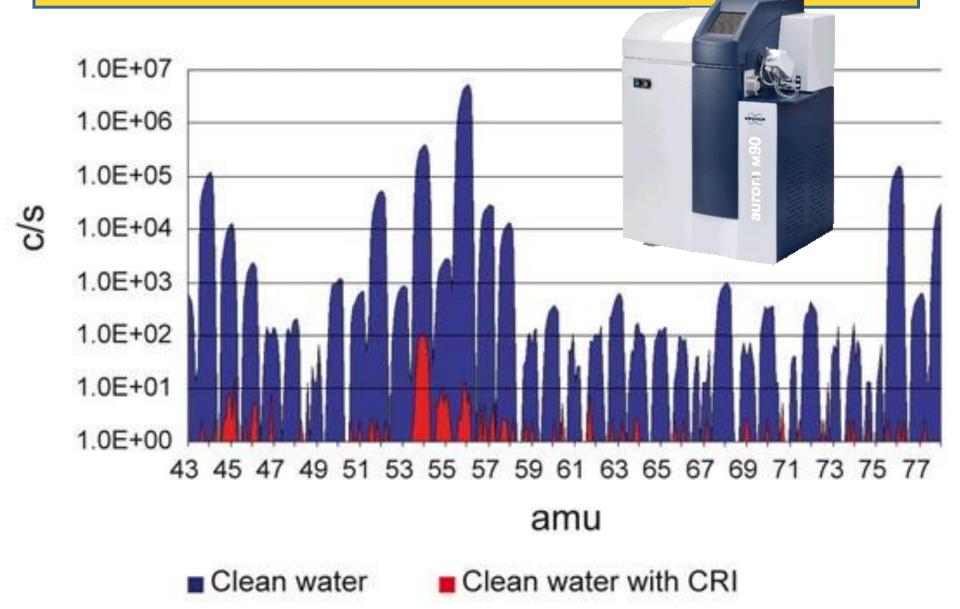


Figure from Bruker website

Introduction of reaction gas (H₂ + He) via skimmer in plasmajet

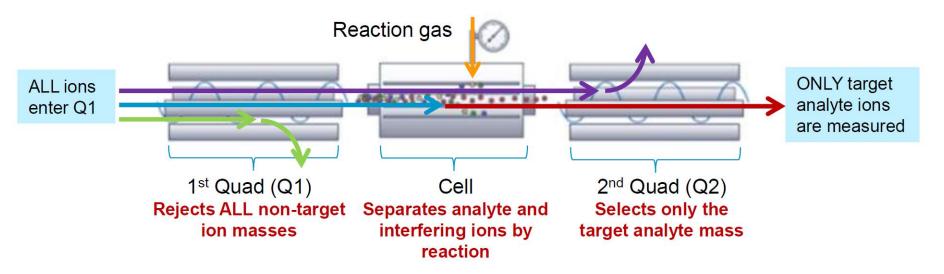
COLLISION/REACTION INTERFACE – CRI BRUKER AURORA M90



ICP-MS/MS OR ICP-QQQ

Agilent 8800

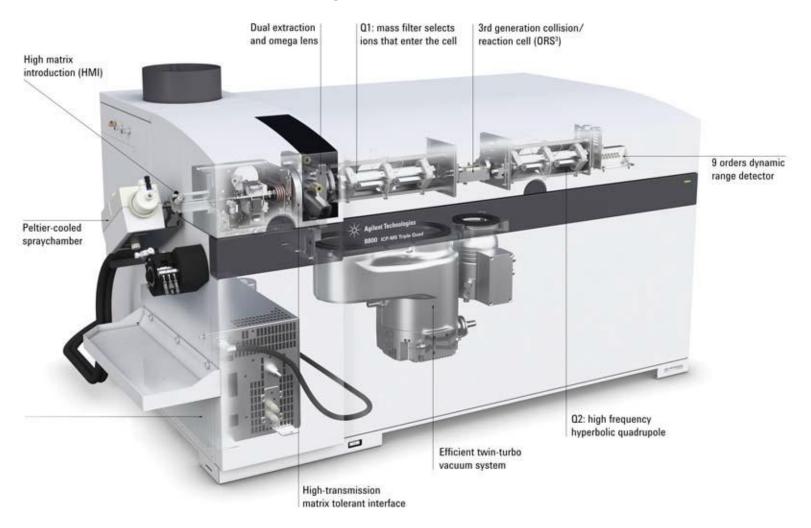
- Introduced in 2012
- Quadrupole filter 1 / Octopole reaction cell / Quadrupole filter 2
 - Double mass selection





Agilent 8800 ICP-QQQ

Superior tool for avoiding spectral overlap



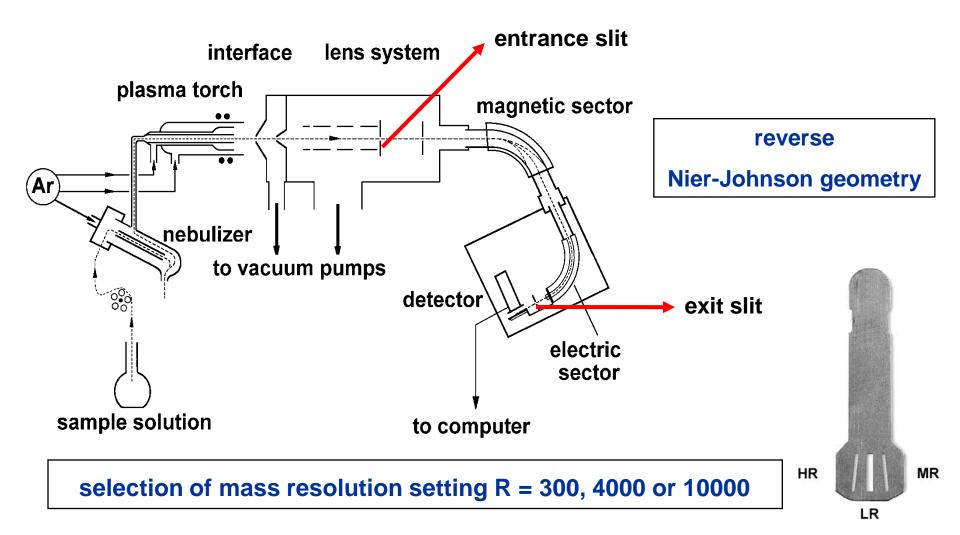
SECTOR FIELD ICP-MS (HIGH RESOLUTION ICP-MS)



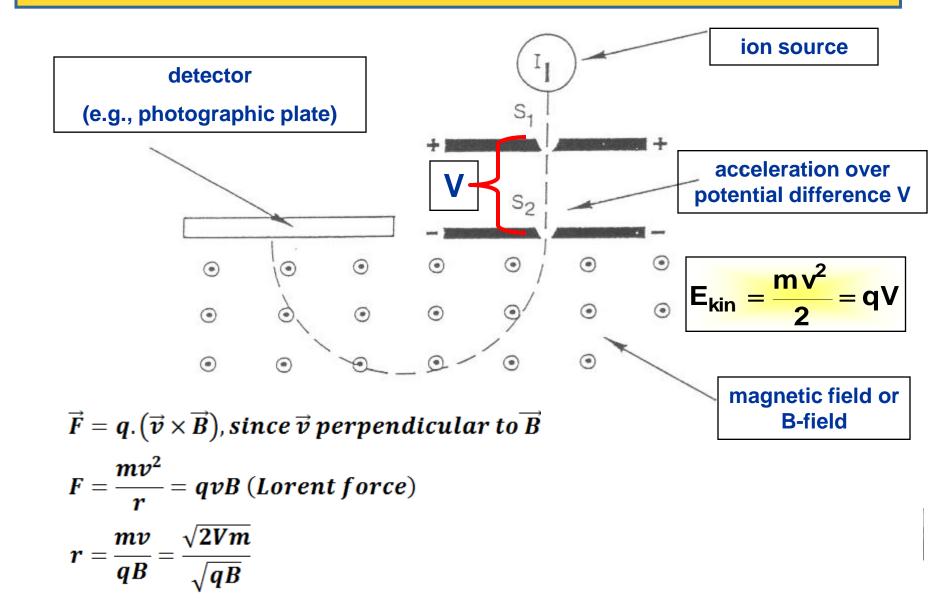




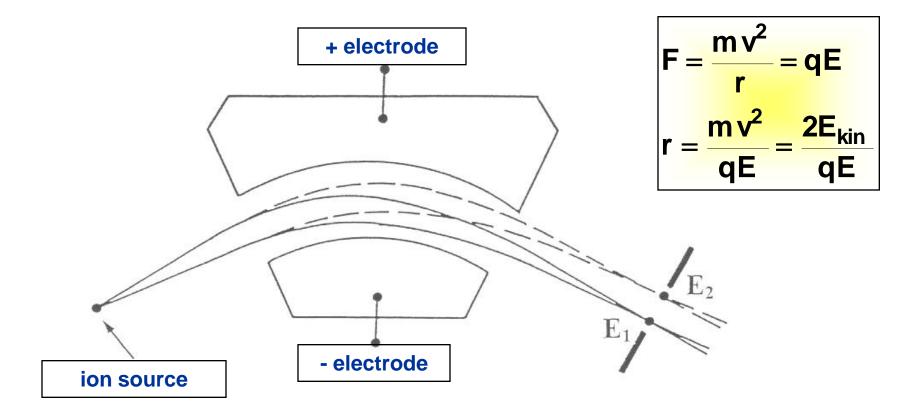
SECTOR FIELD ICP-MS



SEPARATION OF IONS IN A MAGNETIC SECTOR



USE OF AN ELECTROSTATIC SECTOR AS ENERGY FILTER



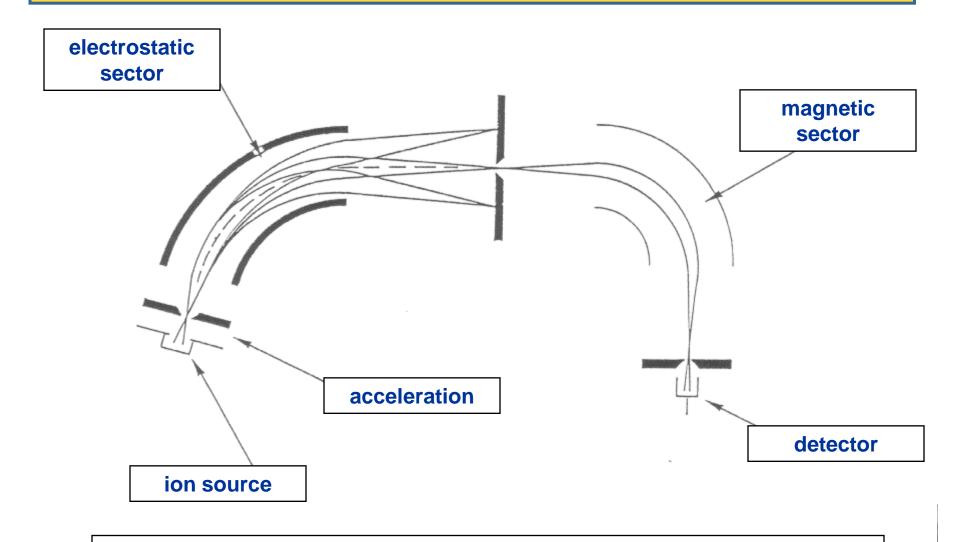
energy filtering leads to improved mass resolution





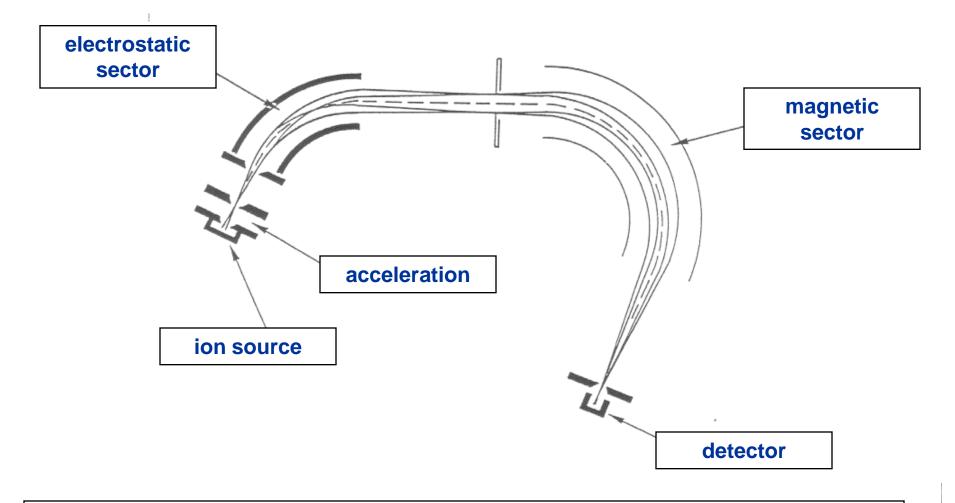


COMBINATION OF ELECTROSTATIC AND MAGNETIC SECTOR



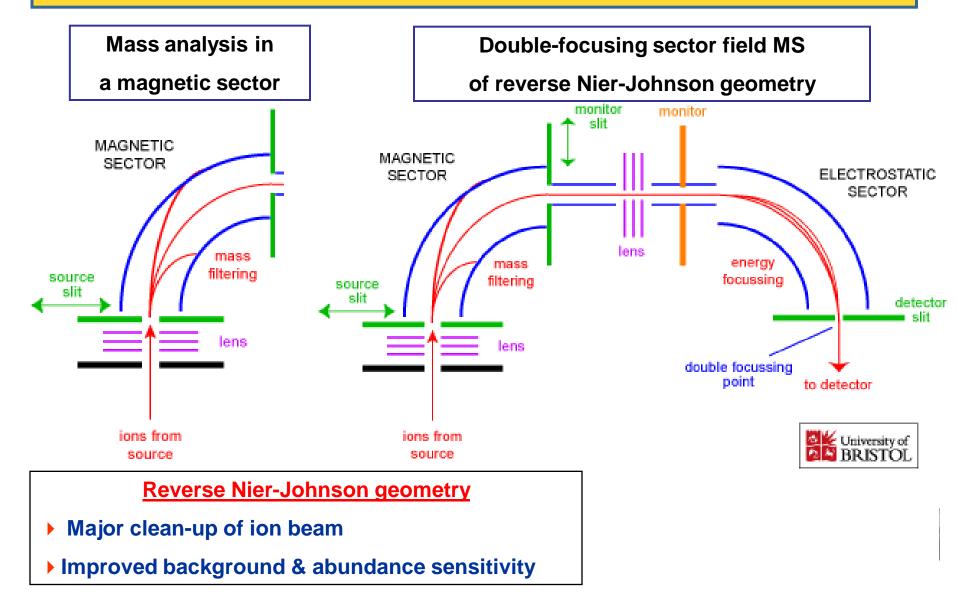
high mass resolution / huge loss in transmission efficiency

DOUBLE-FOCUSING SET-UP NIER-JOHNSON GEOMETRY

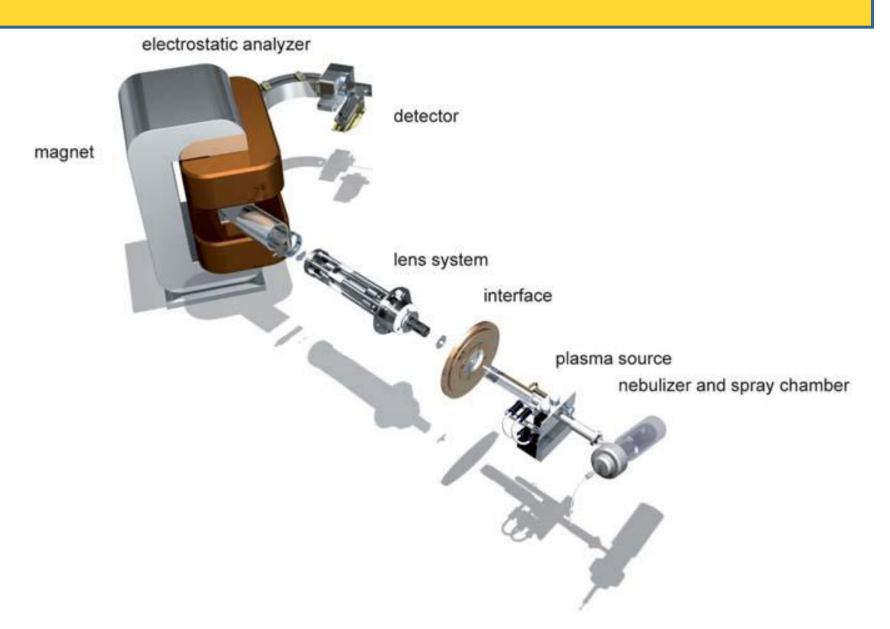


high mass resolution / limited loss in transmission efficiency

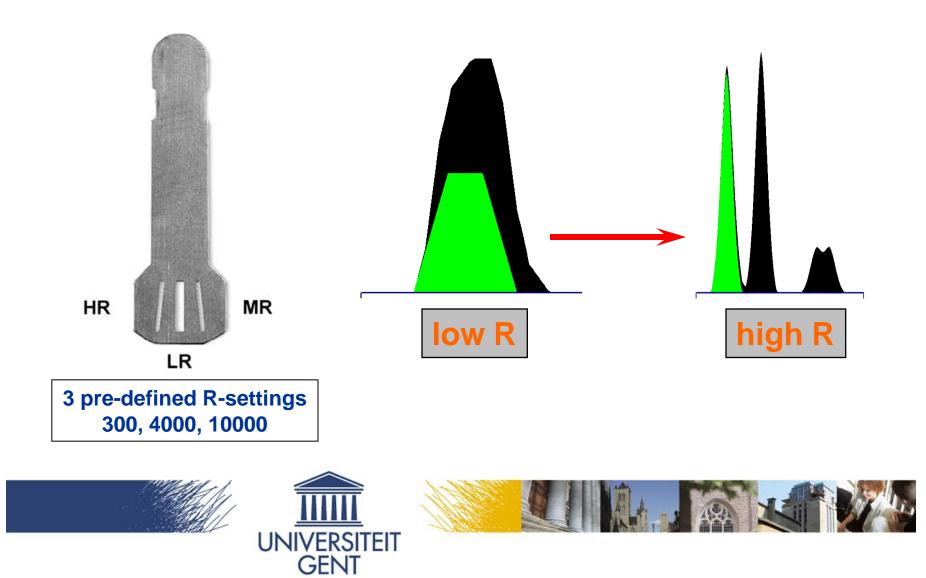
DOUBLE-FOCUSING SECTOR FIELD MS OF REVERSE NIER-JOHNSON GEOMETRY

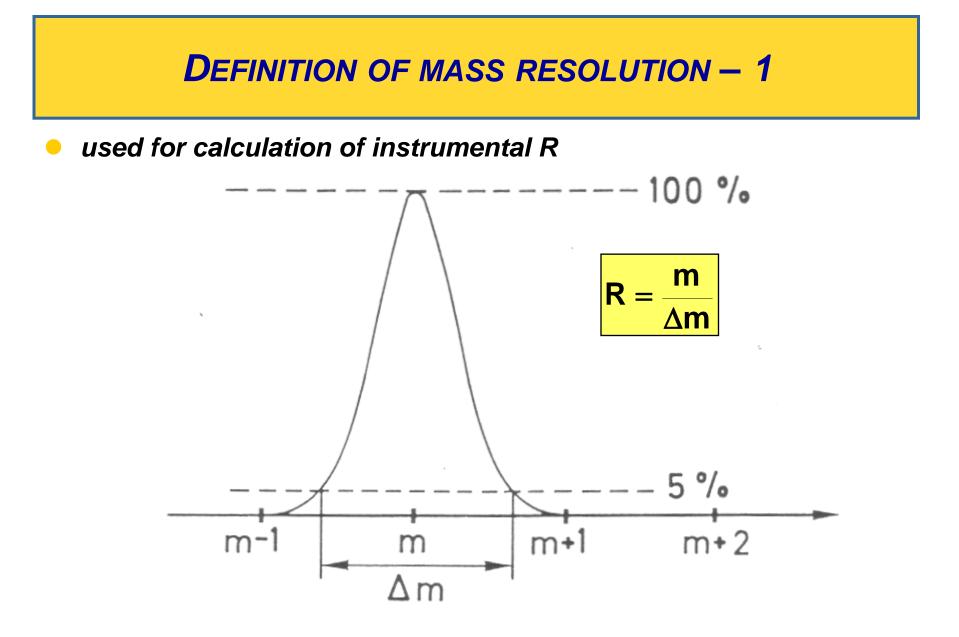


SECTOR FIELD ICP-MS



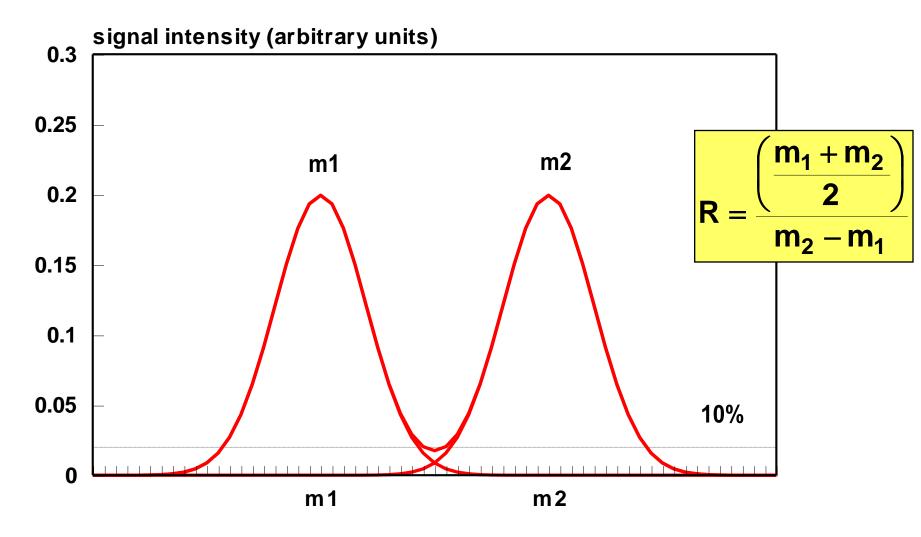
SECTOR FIELD ICP-MS 'HIGH RESOLUTION' ICP-MS





DEFINITION OF MASS RESOLUTION – 2 10% VALLEY DEFINITION

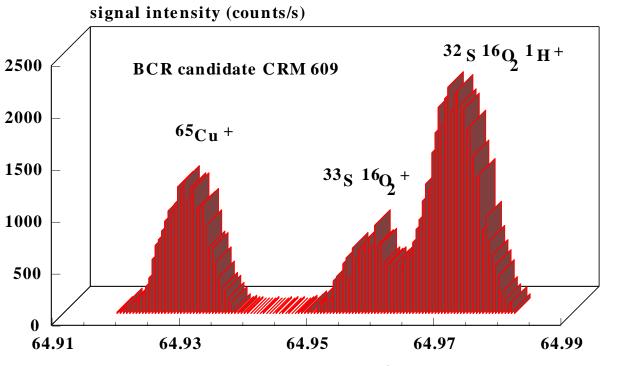
Calculation of mass resolution that is required



SECTOR FIELD ICP-MS 'HIGH RESOLUTION' ICP-MS

Determination of Cu in groundwater

At low R: overlap of ⁶³Cu⁺ and ⁴⁰Ar²³Na⁺, is ⁶⁵Cu⁺ interference-free?

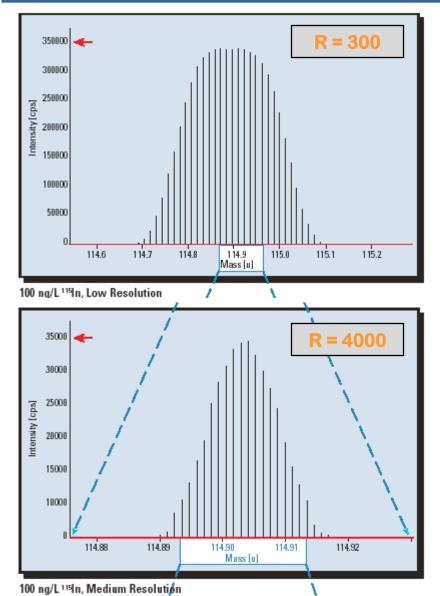


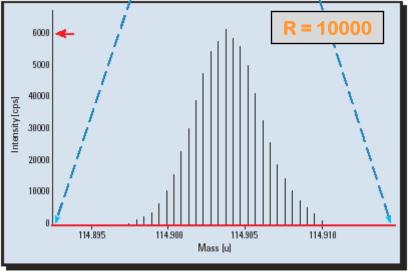
mass-to-charge ratio

GENT

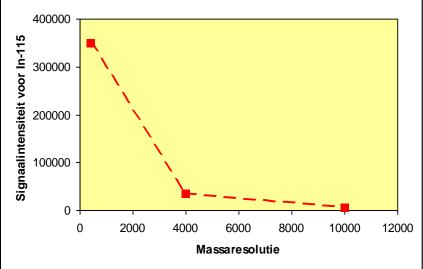


EFFECT OF MASS RESOLUTION SETTING ON SIGNAL INTENSITY









PRESENT-DAY SECTOR FIELD ICP-MS INSTRUMENTS OFFERING HIGH MASS RESOLUTION

Thermo Scientific Element 2 / Element XR







PRESENT-DAY SECTOR FIELD ICP-MS INSTRUMENTS OFFERING HIGH MASS RESOLUTION

Nu Instruments AttoM







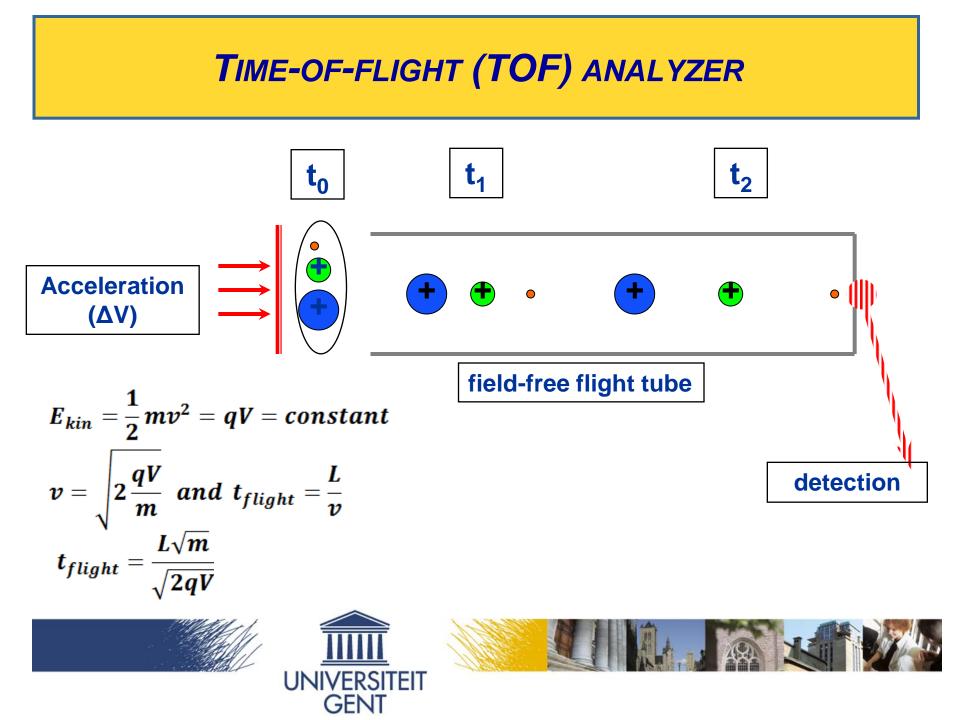


OTHER TYPES OF ICP-MS INSTRUMENTATION





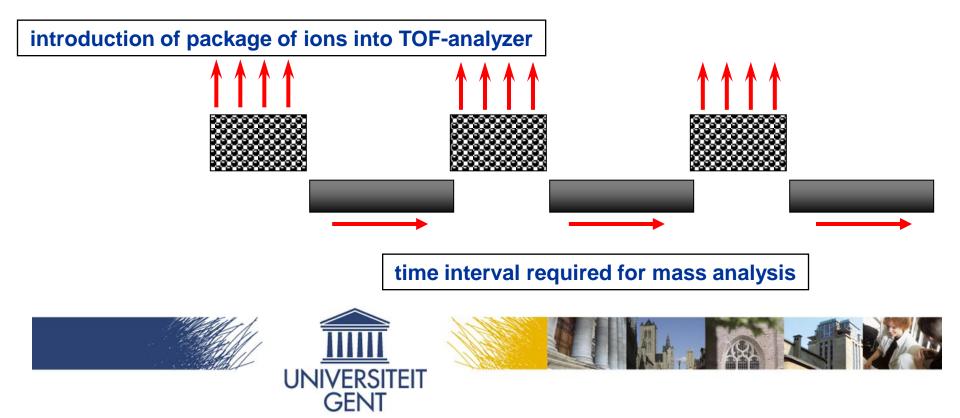




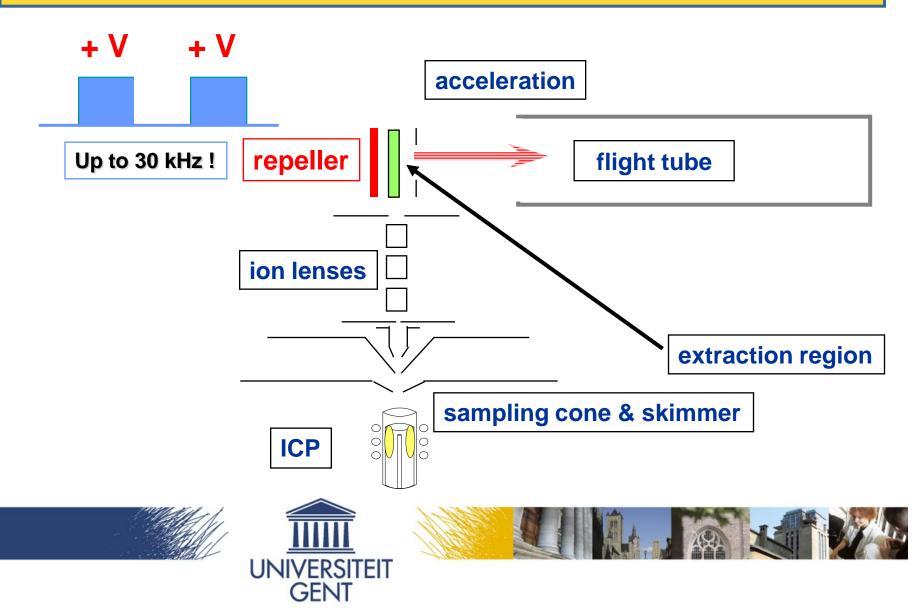
Use of TOF-ANALYZER IN ICP-MS

Ions have to be introduced pulse-wise (no continuous introduction)

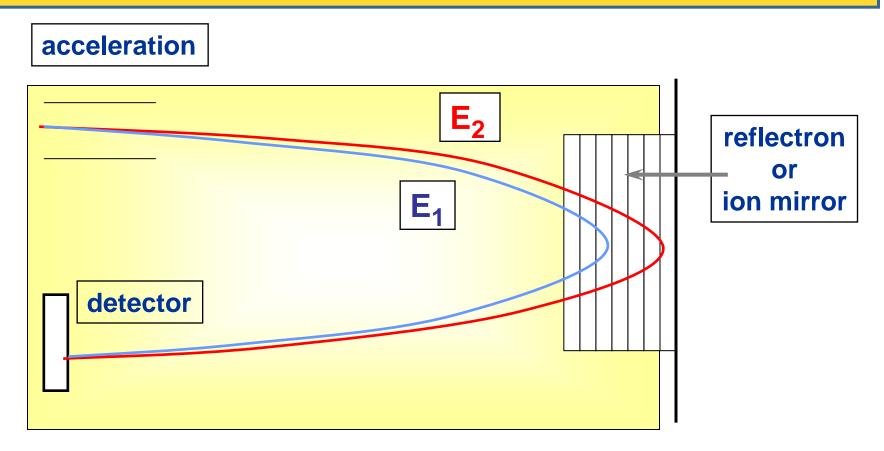
- Otherwise: simultaneous arrival at detector of
 - heavy ion introduced at t & lighter ion introduced at $t + \Delta t$
 - ICP = continuous ion source
- beam modulation required



ORTHOGONAL ACCELERATION AS A MEANS OF BEAM MODULATION



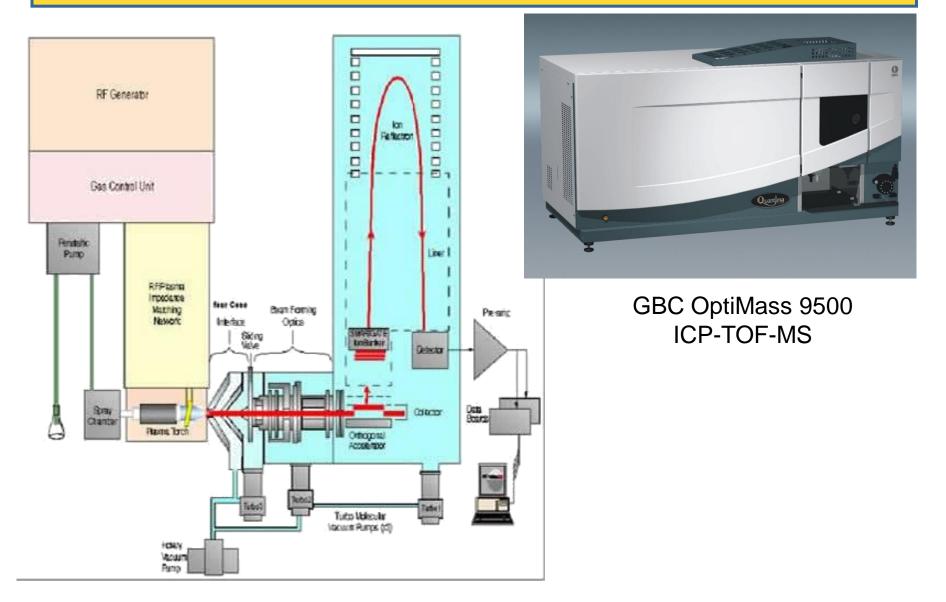
USE OF ION MIRROR (REFLECTRON) IN TOF-ICP-MS TO IMPROVE MASS RESOLUTION



ion, mass m, E_2 ion, mass m, E_1

$$E_2 > E_1$$

ORTHOGONAL ACCELERATION AS A MEANS OF BEAM MODULATION



FIGURES OF MERIT OF TOF-ICP-MS

- Simultaneous handling of ions extracted from ICP
- Fast + pronounced multi-element capabilities
 - up to 30,000 full mass spectra per second
 - well-suited for transient signals with short duration
 - ETV-ICP-MS
 - LA-ICP-MS
 - GC-ICP-MS
- Sensitivity & LODs
 - inferior to ICP-QMS
- Other figures of merit
 - Unit mass resolution or better
 - similar to ICP-QMS in many ways
- No commercial success (so far) ...

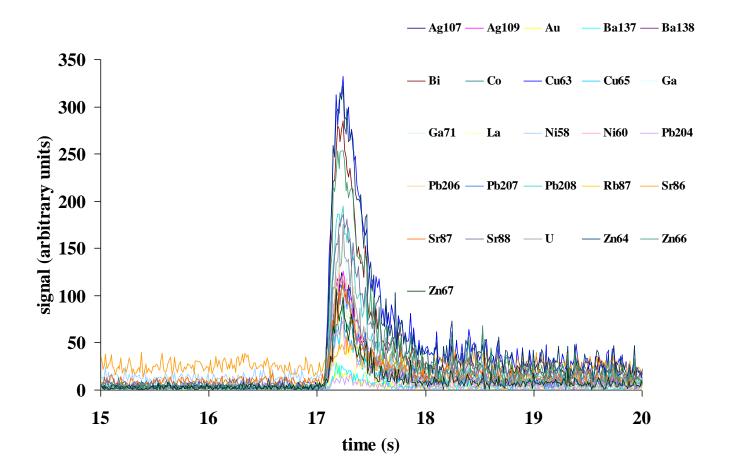


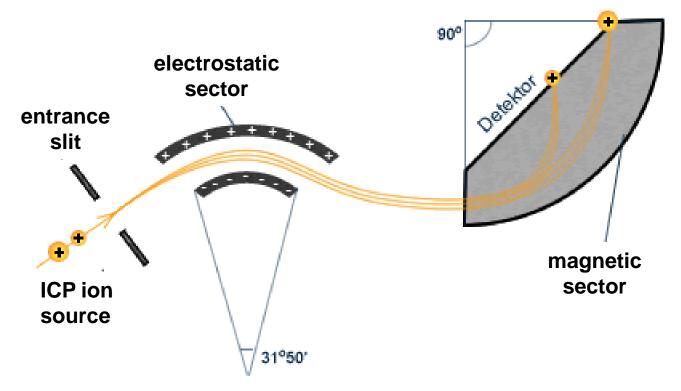


FIGURES OF MERIT OF TOF-ICP-MS

Most promising for LA-ICP-MS applications

e.g., single-shot analysis of NIST SRM 612 glass (FWHM = 0.30 s)

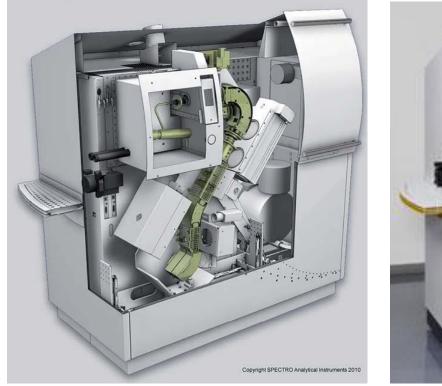




All ion beams focused in one focal plane

- Detector with 4800 miniaturized semi-conductor based detectors
- Simultaneous monitoring of entire elemental mass spectrum (Li to U)

Introduced commercially @ Pittcon-2010 by Spectro



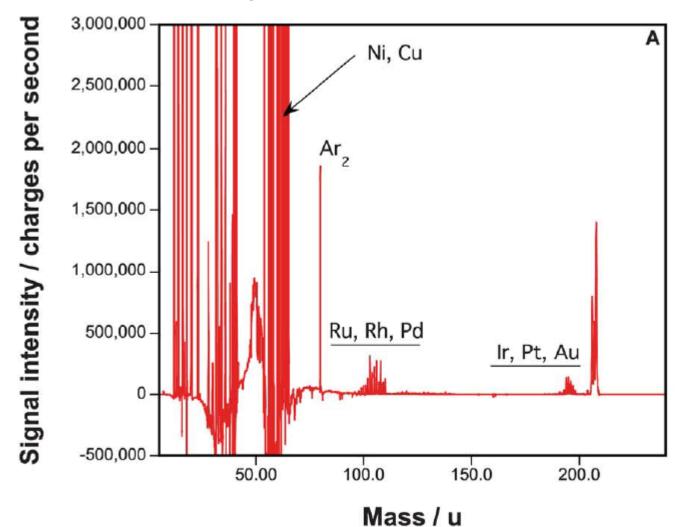




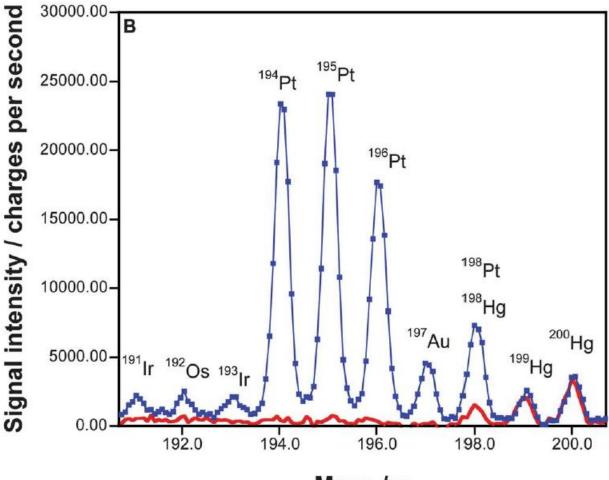




• Simultaneous monitoring of entire elemental mass spectrum

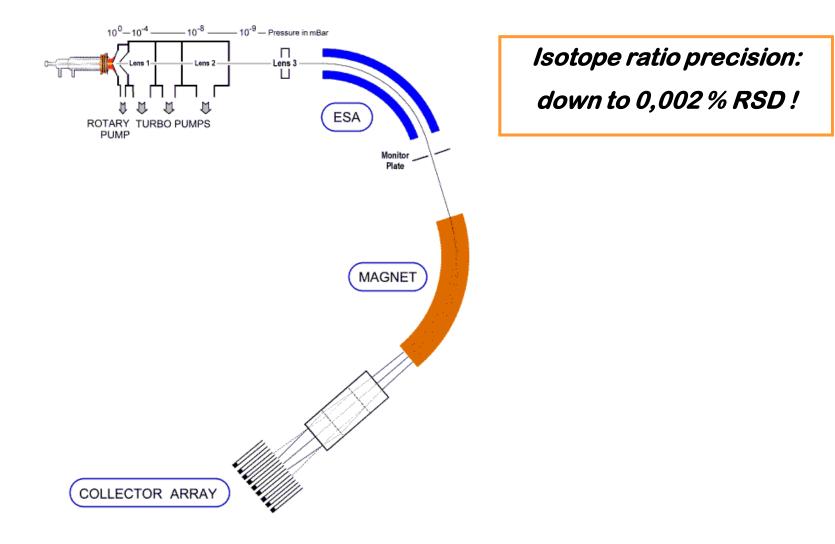


Simultaneous monitoring of entire elemental mass spectrum

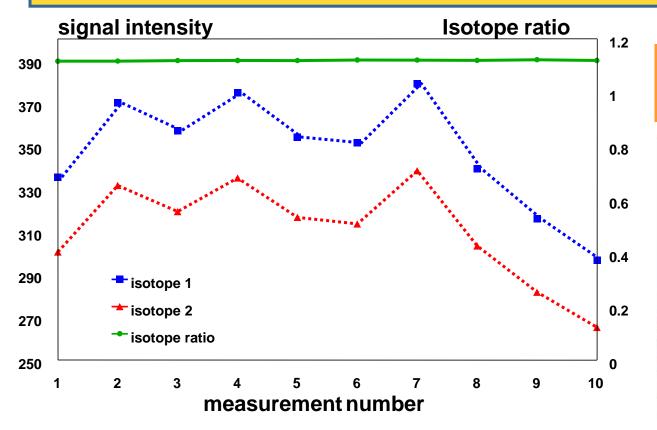


Mass / u

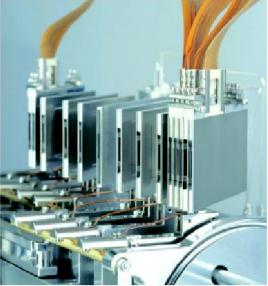
MULTI-COLLECTOR ICP-MS A DEDICATED TOOL FOR ISOTOPIC ANALYSIS



ARRAY OF FARADAY COLLECTORS: SIMULTANEOUS MONITORING OF ION SIGNAL INTENSITIES



Isotope ratio precision: down to 0,002 % RSD !

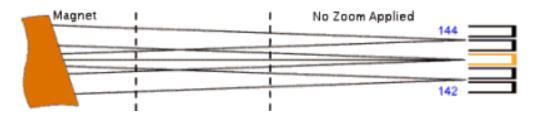


Simultaneous monitoring:

- Automatic correction for signal instability & signal drift
- Higher isotope ratio precision
- With ICP-MS instrument equipped with only one detector:
 - Mimicked by fast 'hopping'

ION BEAMS → FARADAY COLLECTORS ?

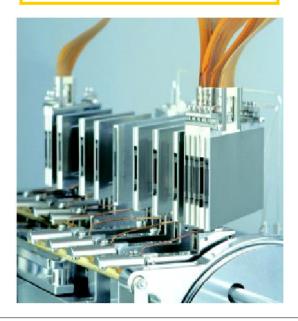
Zoom optics



The ion beams are steered into the appropriate collectors by applying suitable voltages on the zoom "optics" (= electrostatic lenses).

Moveable detectors

(motorized)



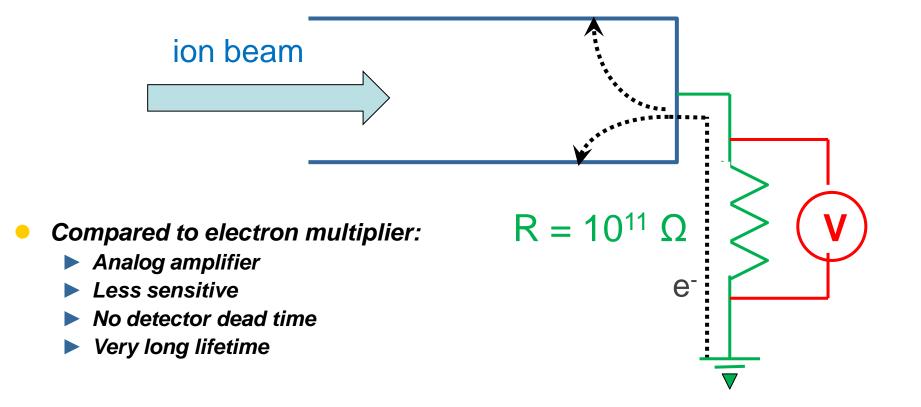
The position of the Faraday collectors can be optimised with respect to the respective ion beams

Or a combination of both ...





FARADAY COLLECTOR – OPERATING PRINCIPLE



GENT





ICP-MS FIGURES OF MERIT

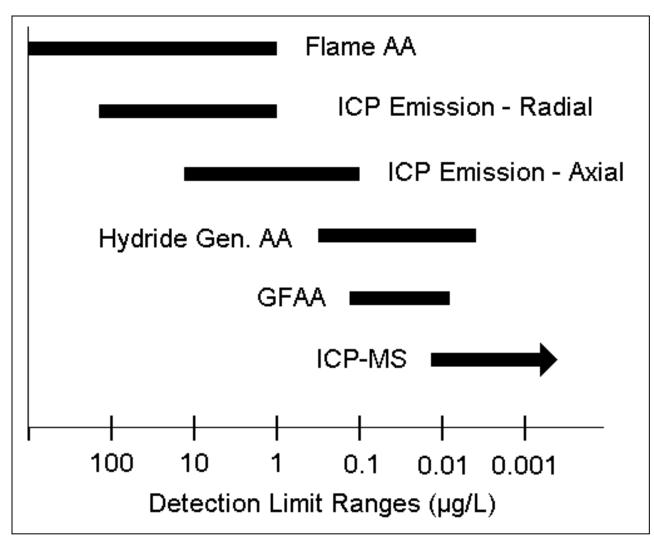




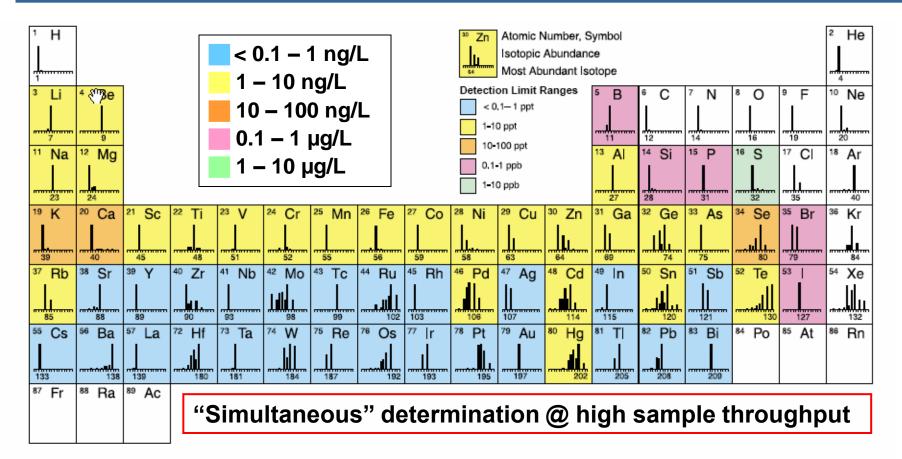


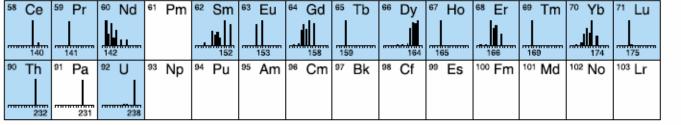
FIGURES OF MERIT OF ICP-MS LIMITS OF DETECTION - LODS

Lowest detection limits attainable



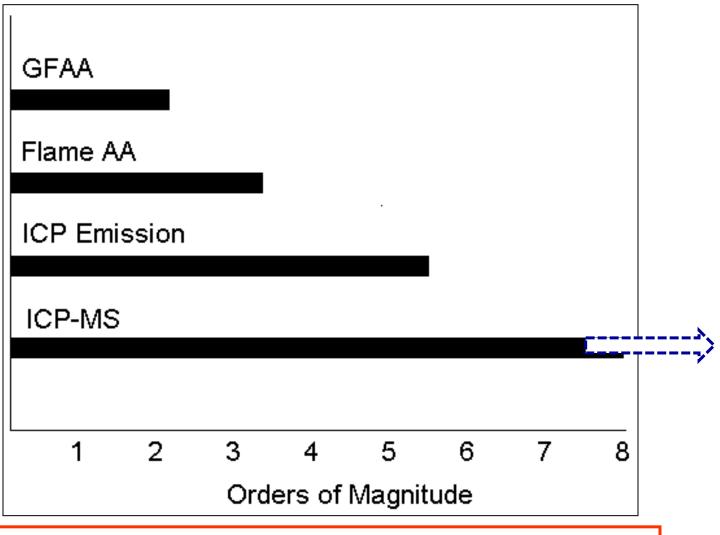
QUADRUPOLE-BASED ICP-MS LIMITS OF DETECTION – LODS





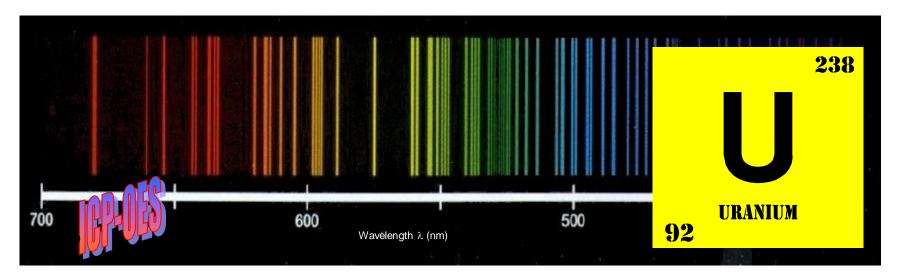


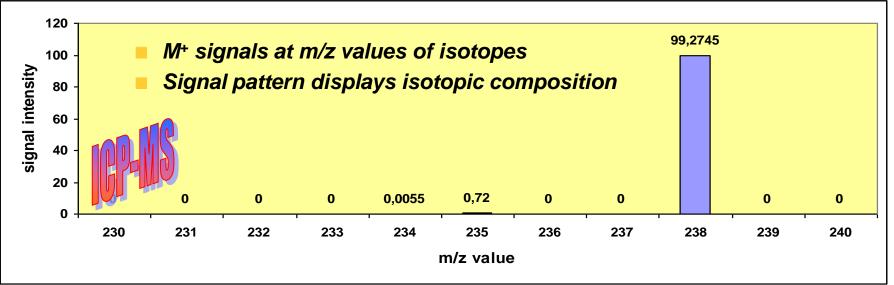
FIGURES OF MERIT OF ICP-MS LINEAR DYNAMIC RANGE



Dual mode electron multiplier (pulse counting vs. analog mode)

ADVANTAGES OF ICP-MS RELATIVELY SIMPLE MASS SPECTRA





ADVANTAGES OF ICP-MS

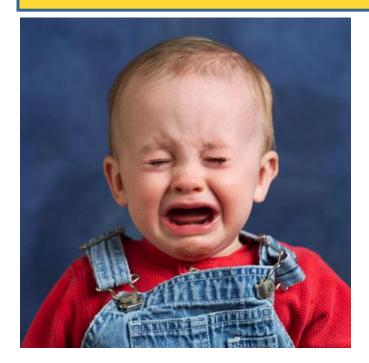
- Low limits of detection
- Multi-element capabilities
- Wide linear dynamic range
- High sample throughput
- Relatively simple spectra
- Ease of combination with
 - Alternative sample introduction systems
 - Chromatographic separation techniques
- Ability to obtain isotopic information







DISDVANTAGES OF ICP-MS



Purchase price

- ► Quadrupole-based ICP-MS ~ 150,000 €
- Sector field ICP-MS ~ 350,000 €
- Multi-collector ICP-MS ~ 600.000 €
- Costs of operation
 - High purity Ar gas (20 L/min)
 - Consumables
- Spectral interferences
 - Collision/reaction cell in ICP-QMS
 - High mass resolution in ICP-SFMS
 - No solution in
 - ► TOF-ICP-MS
 - Mattauch-Herzog ICP-MS







DISDVANTAGES OF ICP-MS OPERATING COSTS (PER YEAR IN US \$)

Technique	Gases	Power	Supplies	Total
Flame AA	2500	200	1300	4000
Furnace AA	200	400	5000	5600
ICP-OES	3500	1000	2300	6800
ICP-MS	3500	1000	11000	15500

Calculated taking into account an average of 4 hrs of operation / day = 1000 hrs / year