



Izoelektrická fokusace

K. Šlais

Izoelektrická fokusace - IEF

- elektromigrační separační analytická metoda využívající existence izoelektrického stavu amfolytů, kdy efektivní náboj je nulový.
- $pH = pI$
- Analyty - proteiny
- Separace - $\Delta pI < 0.01$
- Fokusace – zakoncentrování
- Charakterizace - pI



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Protein jako amfolyt

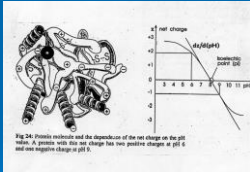


Fig 24-1 Protein structure and the dependence of the net charge on the pH value. A protein with zero net charge has two positive charges at pH 6 and one negative charge at pH 6.

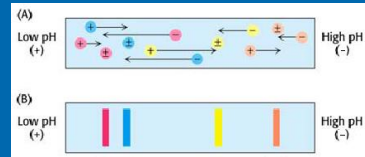
Table 5-2. Isoelectric Points of Several Common Proteins

Protein	pI
Pepsin	<1.0
Ovalbumin (hen)	4.6
Serum albumin (human)	4.9
Tropomyosin	5.1
Insulin (bovine)	5.4
Fibrinogen (human)	5.8
γ -Globulin (human)	6.6
Collagen	6.6
Myoglobin (horse)	7.0
Hemoglobin (human)	7.1
Ribonuclease A (bovine)	9.4
Cytochrome c (horse)	10.6
Histone (bovine)	10.8
Lysozyme (hen)	11.0
Salmine (salmon)	12.1

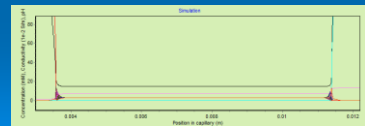
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Simul 5



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Druhy IEF

- Gelová IEF
 - S nosnými amfolyty
 - S imobilizovaným gradientem (IPG)
 - Dvourozměrná elektroforéza 2D = IEF + SDS PAGE
- Kapilární IEF
- Preparativní IEF
 - Free flow
 - Komorová (např. Rotofor)

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Typický výsledek gelové IEF



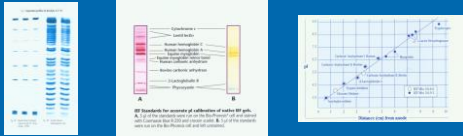
Fig. 2. Typical separation in the Model 111 Mini IEF Cell (5% polyacrylamide gel with 2% Bio-Lyte 3/10 ampholytes). Focusing was carried out in a stepped fashion (100 V for 15 minutes, 200 V for 15 minutes, 450 V for 1 hour). Samples are: lanes 1 & 10: Bio-Rad's IEF Standards; lanes 2-5: Dilutions of horseradish peroxidase; lanes 6-9: Dilutions of Japanese water moccasin snake venom.

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Standardy pl - proteiny



nestabilní,
nečisté,
drahé,
málo barevné,
málo rozpustné při pl



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Nízkomolekulární barevné pl markery



- Požadavky na pl markery
 - Škála pl od ~2 do 11, po ~ 0.5 pl
 - dobře emfolytí, $-dz/dpH - 0.5 > 0.05$, $\Delta pK - 2 < 4$
 - rozpustnost ve vodě při pH = pl, > 1 mg/ml
 - různé barvy, $\lambda_{max} > 400$ A, $\epsilon > 100$
 - Čistota, $> 99\%$
 - Dostupnost, cena markeru
 - Stabilita - hydrolyza, oxidace, fotodegradace, mikroorganismy

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Aminomethylované nitrofenoly

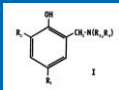


TABLE I
PHYSICO-CHEMICAL PROPERTIES OF DERIVATIVES OF GENERAL FORMULA I

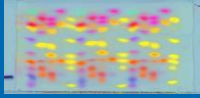
No.	R ₁	R ₂	R ₃	M.W.	pI	λ _{max}	ε
1	H	H	H	155	4.7	410	1000
2	H	H	CH ₃	169	4.7	410	1000
3	H	H	CH ₂ CH ₃	183	4.7	410	1000
4	H	H	CH ₂ CH ₂ CH ₃	197	4.7	410	1000
5	H	H	CH ₂ CH ₂ CH ₂ CH ₃	211	4.7	410	1000
6	H	H	CH ₂ CH ₂ CH ₂ CH ₂ CH ₃	225	4.7	410	1000
7	H	H	CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₃	239	4.7	410	1000
8	H	H	CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₃	253	4.7	410	1000
9	H	H	CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₃	267	4.7	410	1000
10	H	H	CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₃	281	4.7	410	1000
11	H	H	CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₃	295	4.7	410	1000
12	H	H	CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₃	309	4.7	410	1000
13	H	H	CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₃	323	4.7	410	1000
14	H	H	CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₃	337	4.7	410	1000
15	H	H	CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₃	351	4.7	410	1000
16	H	H	CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₃	365	4.7	410	1000
17	H	H	CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₃	379	4.7	410	1000
18	H	H	CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₃	393	4.7	410	1000
19	H	H	CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₃	407	4.7	410	1000
20	H	H	CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₃	421	4.7	410	1000
21	H	H	CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₃	435	4.7	410	1000
22	H	H	CH ₂ CH ₃	449	4.7	410	1000
23	H	H	CH ₂ CH ₃	463	4.7	410	1000
24	H	H	CH ₂ CH ₃	477	4.7	410	1000
25	H	H	CH ₂ CH ₃	491	4.7	410	1000
26	H	H	CH ₂ CH ₃	505	4.7	410	1000
27	H	H	CH ₂ CH ₃	519	4.7	410	1000
28	H	H	CH ₂ CH ₃	533	4.7	410	1000
29	H	H	CH ₂ CH ₃	547	4.7	410	1000
30	H	H	CH ₂ CH ₃	561	4.7	410	1000
31	H	H	CH ₂ CH ₃	575	4.7	410	1000
32	H	H	CH ₂ CH ₃	589	4.7	410	1000
33	H	H	CH ₂ CH ₃	603	4.7	410	1000
34	H	H	CH ₂ CH ₃	617	4.7	410	1000
35	H	H	CH ₂ CH ₃	631	4.7	410	1000
36	H	H	CH ₂ CH ₃	645	4.7	410	1000
37	H	H	CH ₂ CH ₃	659	4.7	410	1000
38	H	H	CH ₂ CH ₃	673	4.7	410	1000
39	H	H	CH ₂ CH ₃	687	4.7	410	1000
40	H	H	CH ₂ CH ₃	701	4.7	410	1000
41	H	H	CH ₂ CH ₃	715	4.7	410	1000
42	H	H	CH ₂ CH ₃	729	4.7	410	1000
43	H	H	CH ₂ CH ₃	743	4.7	410	1000
44	H	H	CH ₂ CH ₃	757	4.7	410	1000
45	H	H	CH ₂ CH ₃	771	4.7	410	1000
46	H	H	CH ₂ CH ₃	785	4.7	410	1000
47	H	H	CH ₂ CH ₃	799	4.7	410	1000
48	H	H	CH ₂ CH ₃	813	4.7	410	1000
49	H	H	CH ₂ CH ₃	827	4.7	410	1000
50	H	H	CH ₂ CH ₃	841	4.7	410	1000

TABLE II
PHYSICO-CHEMICAL PROPERTIES OF DERIVATIVES OF GENERAL FORMULA I

No.	R ₁	R ₂	R ₃	M.W.	pI	λ _{max}	ε
1	H	H	H	155	4.7	410	1000
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7	H	H	CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₃	239	4.7	410	1000
8	H	H	CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₃	253	4.7	410	1000
9	H	H	CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₃	267	4.7	410	1000
10	H	H					

Určení pI interpolací v gelové IEF

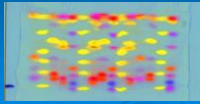
Gradient pH



Směs 30 jednoduchých pufrů



Biolyt 3 – 10



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Dynamika fokusace v gelové IEF

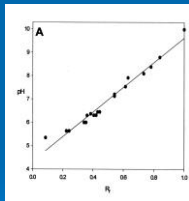


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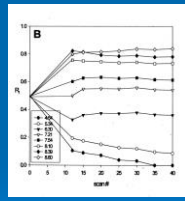
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Dynamika pH gradientu Biolyt 3-10



Lineární gradient pH 4 - 10



Po 1/4 hod malé změny pH gradientu

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Dynamika fokusace v gelové IEF



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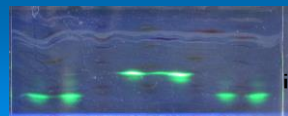
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Vývoj fluorescenčních pI markerů

Vis



fluorescence



ief446_INDEX.exe

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yellow markers

Mass spectrometric characterization of low-molecular-mass color pI markers and their use for direct determination of pI value of proteins

Mazanec, K., Slais, K., Chmelik, J.

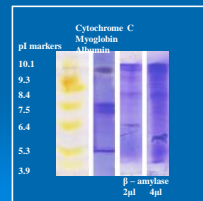
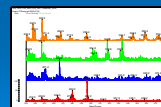
J. Mass Spectrom. 41 2006 1570-1577

Pardubice 2005



Mass spectra of nitro-substituted pI markers

pI	Marker	Structure	Chemical
10.1	cytochrome C		10.1
9.3	Myoglobin		9.3
8.4	Albumin		8.4
7.5			7.5
6.4			6.4
5.3			5.3
3.9			3.9



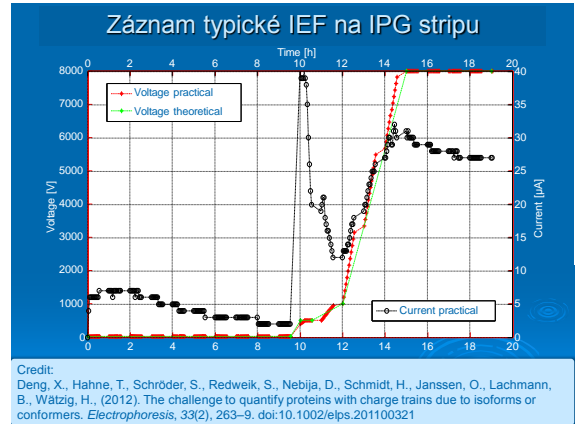
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2D Gel electrophoresis

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Test of color pI markers - LM ladder

Hanspeter Schickel,
ETC Elektrophorese-Technik GmbH, Kirchentellinsfurt, Germany

Strips rehydrated 2 hours under Kerosene
run native 7 hours with Nitrogen
Amersham Multiphor.

Courtesy of Dr. Hanspeter Schickel.

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IEF of mixture of chosen pI markers in the first dimension strip of 2D gel electrophoresis

in Clinical Proteomics. From Diagnosis to Therapy. J. Van Eyk and M.J. Dunn (Eds.),
Chapter 2. Protein Separation by Two-Dimensional Electrophoresis
Pamela M. Donoghue, Miroslava Stastna, Michael J. Dunn, p 13,
2007 WILEY-VCH Verlag GmbH & Co. KGaA, Weinheim

Immobiline Dry Strip (Amersham Biosciences) pH 3–10, 18 cm.
Apparatus: Protean IEF Cell (BioRad).
Sample: 10 µl of pI markers mixture diluted with 340µl of IEF buffer (8M urea, 2M thiourea, 4% CHAPS, 1% DTT, 0.01% bromophenol blue, 1.5% (v/v) hydroxyethyl disulfide, 0.2% (v/v) IPG buffer pH 3–10).
The acidic end is on the left and the basic end on the right side of the strip.
The pI values of individual pI markers are marked in the picture

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2D - typical result – silver staining

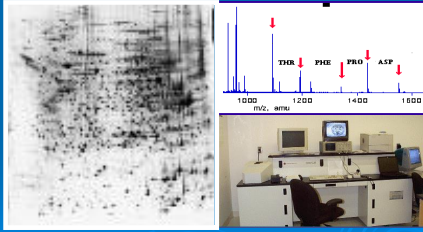
Fig 21: (A) High-resolution horizontal 2D electrophoresis using an immobilized pI-gradient in the first-dimensional run (IPG-Dial) [46]; basic proteins from yeast cell lysate (*Saccharomyces cerevisiae*). Silver staining according to Merrill [48]. By kind permission of Dr. A. Görg [49].

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2D Gel electrophoresis - Software

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Protein identification by 2D gel electrophoresis -MS



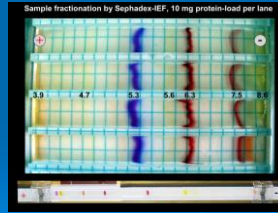
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IEF in Granulated Sephadex Gels

Methods in Molecular Biology, vol. 424: Volume 1: Sample Preparation and Pre-Fractionation, Edited by: A. Posch
 Chapter 22. Sample Prefractionation in Granulated Sephadex IEF Gels
 Angelika Görg, Carsten Lück, and Walter Weiss, p 277,
 Humana Press Inc., 2007, Totowa, NJ



Use of coloured pI - markers to determine the slope of the pH gradient and the position where to cut and remove the individual Sephadex fractions in order to fit to the corresponding narrow pH range IPGs

Courtesy of Carsten Lück

IEF in Sephadex gels and IPG strips

Hodný Z., Přidalová J.,

Institute of Experimental Medicine AV ČR, v.v.i., Prague



Courtesy of Z. Hodný



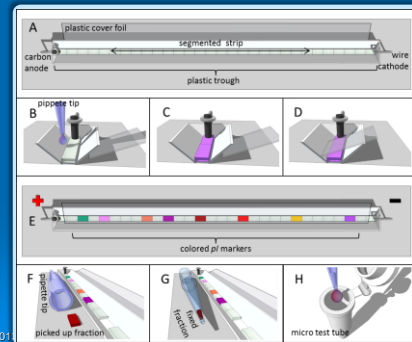
pI markers - LM ladder
 Home made strip,
 linear gradient pH 4-10,
 11 cm,
 1 min 30V,
 50 min 30V => 3500V,
 2 hours 3500V,
 Courtesy of J. Přidalová

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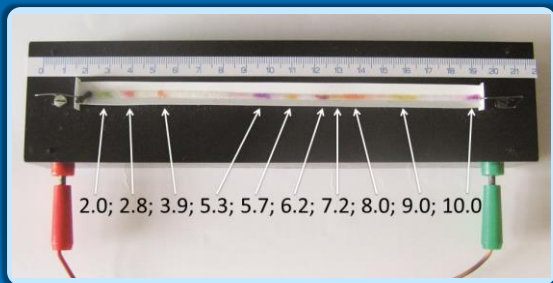
Mikropreparativní sIEF v proužku netkané textilie



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sIEF nosných pufrů s barevnými pI markery



fokusovaná směs:

0,15 ml zásobního roztoku 12ti nosných pufrů v hydroxidu sodném
 0,05 ml zásobního roztoku barevných pI markerů
 0,15 ml ethylen glykolu, 0,05 ml butanolu, 1 ml vody

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Animace průběhu separace

- separace probíhá většinou cca 12 hodin
 - večer nadávkování vzorku a zapnutí zdroje
 - ráno možno sbírat frakcionovaný vzorek
 - pokud není možnost extrahovat frakce, zdroj udržuje nejvyšší napětí do příchodu obsluhy



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Odběr frakcí

- dvě základní konfigurace pro odběr frakcí:
 - proužek při separaci **v celku** – frakce vybrány po IEF na základě polohy markerů a vystříhány
 - proužek dopředu **nařezaný na kousky** definované délky a kousky jsou před separací vyrovnány do fokusačního korýtka a přilepeny roztokem sacharózy příp. jiné fixační látky
 - při lepení možnost přidat i směs nosných pufrů a barevných markerů → „instantní proužek“ → stačí přidat vodu a vzorek a zapnout zdroj

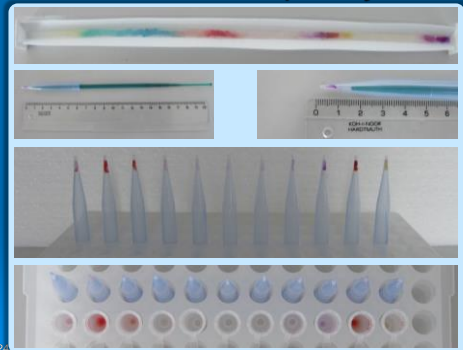


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extrakce frakcí - promývání

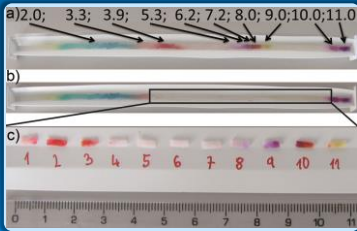


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Frakcionace syrovátky

- optimalizace pro dosažení co nejvyšší čistoty caseinomacropéptidu
 - separovaný vzorek – 1% (m/v) roztok surové syrovátky
 - přidání spacerů – IMAC (imidazol-1-yl-octové kyseliny) a Tris (tris(hydroxymethyl)aminometanu)



program ext. zdroje:

100 V – 200 V 4 hod.
200 V – 1000 V 4 hod.
1000 V – 3000 V 4 hod.
3000 V → do sběru frakcí

vzorek:
0,375 ml 1% (w/v) syrovátky;
0,05 ml zásobního roztoku;
barevných *pI* markerů;
0,1 ml ethylen glykolu;
0,05 ml butanolu;
0,025 ml 0,1 mol⁻¹ IMAC;
0,1 ml of 0,1 mol⁻¹ Tris;

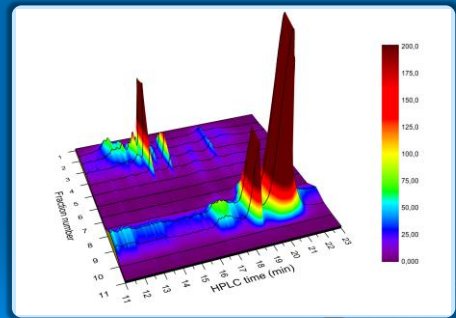
extrakce: 8mm proužek do
100 μ l vody

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HPLC analýza frakcí získaných z IEF



Kromia Microbore Poroshel 30080-C18 (5 μ m částice, 1+75 mm) + C18 předstokována při 70 °C

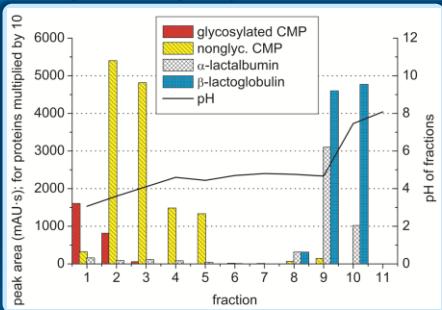
Průtok 200 μ l min⁻¹

32/24/2013 0,1% TFA lineární gradient od 5 do 80 % EBF ACN (30 min)

stokovaný vzorek 4 μ l

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Obsah jednotlivých proteinů ve frakcích



Výtěžek:

glykosylovaný CMP 44%

α -laktalbumin 77%

neglykosylovaný CMP 80%

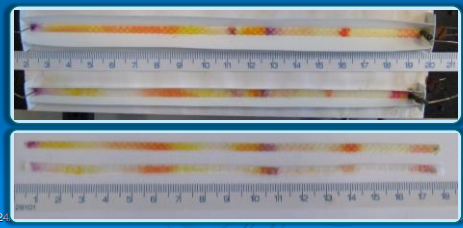
β -laktoglobulin 101%

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Testování dalších typů netkaných textilíí

- různé materiály ze zorkovníku firmy Polytex s.r.o. (Malé Svatoňovice)
- nejlepší materiál (Novolin, 100 g⁻²) podobné charakteristiky jako původní netkaná textilie ale vyšší pevnost v tahu a menší třepivost – jednodušší manipulace a omezení odpadávání krátkých vláken

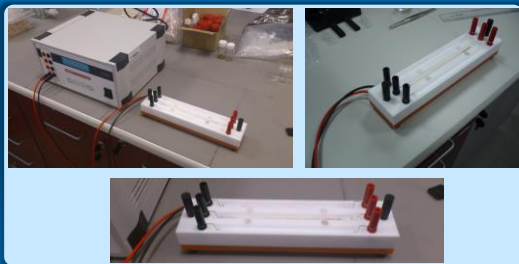


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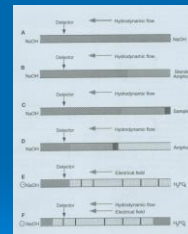
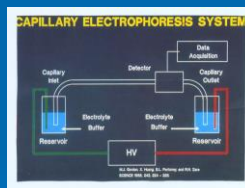
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sIEF na PŘF MU

- Diplomová práce
- Bc. Petr Švédá, školitel doc. RNDr. Přemysl Lubal, Ph.D.
- vytvoření prototypu společné se separační metodou
- následně převést metodu do výuky na ústavu analytické chemie PŘF MU



Kapilární IEF



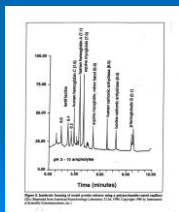
9/24/2013

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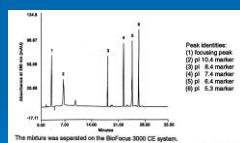
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Kapilární IEF standardů

proteiny



pI markery

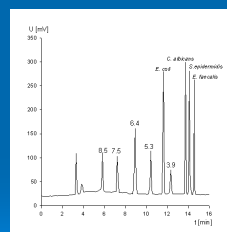


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CIEF mikroorganismů



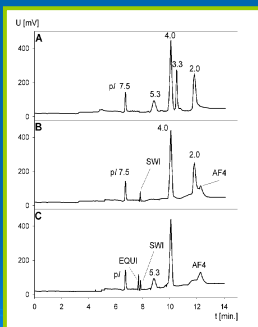
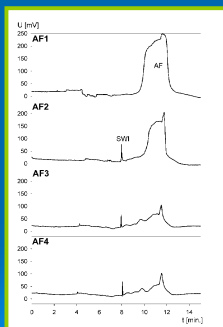
Sample: *E. coli*, *C. albicans*, *S. epidermidis*, *E. faecalis* in physiological saline solution, 4×10^8 cell ml⁻¹.

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CIEF virů s UV detekcí



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Patogeny z různých zdrojů

Table 1. The strains of the plant pathogens included in this study, comparison of their isoelectric points, pI, and RSDs from three measurement of the migration times, t, for each of the strains.

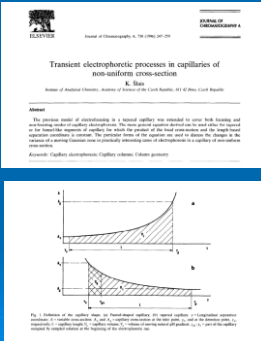
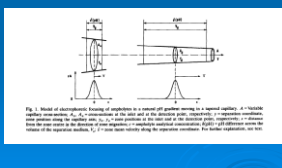
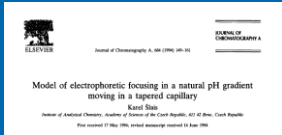
Abbreviation in Figs.	Strain	pI	RSD
C. michiganensis	<i>Clavibacter michiganensis</i> subsp. <i>michiganensis</i> CCM 1630	4.6	pI = 4.7, RSD = 1.9 %
	<i>Clavibacter michiganensis</i> subsp. <i>michiganensis</i> VURV C204	4.6	
	<i>Clavibacter michiganensis</i> subsp. <i>michiganensis</i> VURV 2459	4.7	
	<i>Clavibacter michiganensis</i> subsp. <i>michiganensis</i> VURV 5050	4.6	
	<i>Clavibacter michiganensis</i> subsp. <i>michiganensis</i> VURV 5059	4.7	
	<i>Clavibacter michiganensis</i> subsp. <i>michiganensis</i> VURV 7008	4.7	
C. michiganensis	<i>Clavibacter michiganensis</i> subsp. <i>michiganensis</i> VURV 7018	4.6	pI = 4.7, RSD = 1.9 %
	<i>Clavibacter michiganensis</i> subsp. <i>michiganensis</i> VURV 7030	4.7	
X. vesicatoria	<i>Xanthomonas vesicatoria</i> CCM 2101	4.0	pI = 4.1, RSD = 0.7 %
	<i>Xanthomonas vesicatoria</i> CCM 2102	4.1	
	<i>Xanthomonas vesicatoria</i> VURV P-1-1	4.0	
	<i>Xanthomonas vesicatoria</i> VURV P-6-1	4.1	
	<i>Xanthomonas vesicatoria</i> LMG 2804	4.1	
P. syringae	<i>Pseudomonas syringae</i> pv. <i>tomato</i> CFBP 5422	4.0	pI = 4.0, RSD = 1.9 %
	<i>Pseudomonas syringae</i> pv. <i>tomato</i> CFBP 2212	4.0	
	<i>Pseudomonas syringae</i> pv. <i>tomato</i> IVA 1733.3	4.0	
P. corrugata	<i>Pseudomonas corrugata</i> CFBP 4901	2.4	pI = 2.4, RSD = 0.9 %
	<i>Pseudomonas corrugata</i> CFBP 5465	2.4	
	<i>Pseudomonas corrugata</i> CFBP 6663	2.4	
	<i>Pseudomonas corrugata</i> IVA 614.5.3	2.4	
P. corrugata			

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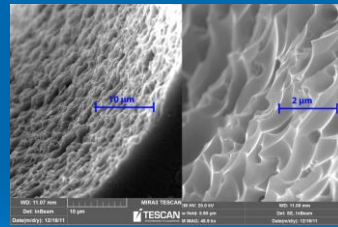
IEF

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Tapered capillary in cIEF

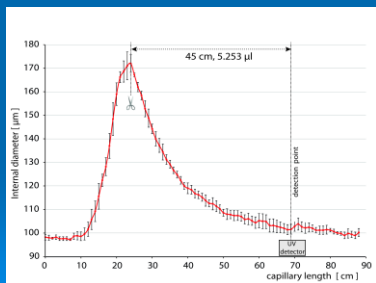


Supercritical water in preparation of tapered fused silica capillaries



Effect of treatment of 100 µm i.d. fused silica capillary with supercritical water in semi-dynamic mode. Experimental conditions: 400 °C, 32 MPa, 20 replacements of supercritical water.

Dependence the local internal diameter of etched fused silica capillary on the capillary length. The cutout of the segment used as the tapered capillary in cIEF and the detection window are indicated.



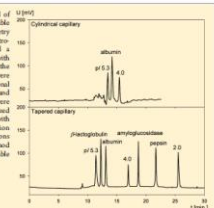
analytical chemistry

Technical Note
publications.cup.edu

Isoelectric Focusing in Continuously Tapered Fused Silica Capillary Prepared by Etching with Supercritical Water

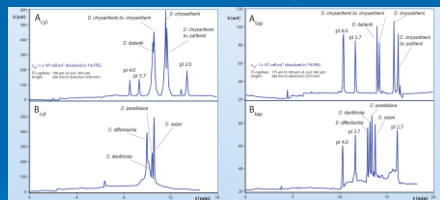
Karel Šlais,* Marie Horáková, Pavel Karáček, Josef Planeta, and Michal Roth
Institute of Analytical Chemistry of the ASCR, v. i. s. r., Veselí 97, 602 00 Brno, Czech Republic

ABSTRACT: This communication indicates the potential of etching with sub- and/or supercritical water for reproducible preparation of fused-silica capillaries with tapered geometry suitable for capillary isoelectric focusing (cIEF) with electro-osmotic displacement. The etching procedure provided a single-piece construction of the tapered separation space with a cylindrical connection of the detection window to the electrode vial. Selected proteins and colored pI markers were used as model analytes. A comparison with conventional cylindrical capillary under comparable applied voltage and analysis time was made, and the resulting peaks were compared in terms of peak resolution under optimized conditions. In cIEF applied in a tapered capillary with the inlet cross-section three times larger than the cross-section of the detection window, three to five times higher resolutions of corresponding peak pairs were obtained. The method described opens the way to increase the number of separable compounds without resorting to excessively high voltage.



Resolution of several *Dickeya* bacterium species with similar isoelectric points by capillary isoelectric focusing employing

cylindrical (left) and tapered (right) capillary.



Combination of Capillary Isoelectric Focusing in a Tapered Capillary with MALDI-TOF MS for Rapid and Reliable Identification of *Dickeya* Species from Plant Samples
Horka, M.; Salpiachta, J.; Karasek, P.; Kubesova, A.; Horky, J.; Matouskova, H.; Slais, K.; Roth, M. ANALYTICAL CHEMISTRY Volume: 85 Pages: 6806-6812 JUL 16 2013

Preparative Liquid phase IEF



Rotofor



MicroRotofor



Total 2.5 ml sample
Ten 0.25 ml fractions
2 hours run time

Preparativní autofokuse peptidů + pI markerů

pI = 3.0 pI = 5.3

Tomas, R.; Yan, L.S.; Krenkova, J.; Foret, F. ELECTROPHORESIS, 28 (13): 2283-2290 2007

ZOOM® IEF Fractionator (Invitrogen)

- komůrka pro fokusaci je oddělena polyakrylamidovými disky s definovaným pH
- proteiny musí procházet přes tyto disky dokud nedorazí do izoelektrického bodu
- objem každé části 650 µl
- vysoká reprodukovatelnost
- příprava vzorku: rozpustit, denaturovat, alkylovat, bez zákalu a částic

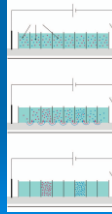
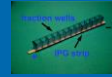


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OFF - GEL electrophoresis



P.E. Michel, F. Reymond, I. L. Atmadji, J. Jasser, H. H. Girault, J. S. Rossier, *Electrophoresis* 2003, 24, 3-11

Agilent 3100 OFFGEL Fractionator

pI-based fractionation of proteins and peptides with liquid-phase recovery, introduced May 30, 2006. Co-developed with DiagnoSwiss S.A.

Use of pI-dye markers as on-line trackers for the focusing of peptides during electrophoresis on the OFFGEL fractionator device (OGE). 22 hours runtime

Courtesy of M. Heller, DKF, University of Bern, Switzerland

Use of pI-dye markers as on-line trackers for the focusing of peptides during electrophoresis on the OFFGEL fractionator device (OGE).



- pI-marker dyes were added at 10 µg (dark orange, pI 3.9; violet, pI 5.2; red, pI 6.2; bright orange, pI 8.0) or 30 µg (yellow, pI 10.1), respectively.
- Peptide/dye solution was distributed into the 13 wells of the OGE.
- IPG strips pH 3-10 from BioRad re-hydrated in OGE buffer were used.
- Focusing was done by setting a maximal potential (1250 or 1500 V) and a current limit of 50 µA.

Courtesy of M. Heller, DKF, University of Bern, Switzerland

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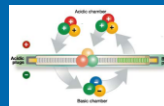
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Parallel isoelectric focusing.



Charged proteins/peptides migrate through the dPC™ chip between the acidic and basic sides until they encounter a gel plug whose pH is at or near their pI. The uncharged protein/peptide will no longer migrate and become "trapped" in these plugs



G. Zilberstein et al. Parallel isoelectric focusing chip; *Proteomics* 2004, 4, 2533-2540.

S. Buhkspan et al. Matrices, Arrays, Systems and Methods; US PT 7,166,202, 2007.

dPC™ Chip Specifications

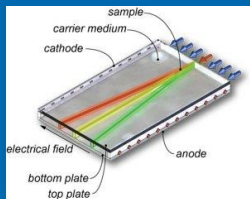
Fractionation Time	30-45 minutes
pH Ranges*	4.2-6.2; 6.0-8.0; 7.2-9.2
Max. Resolution	0.05 pH
Number of plugs per chip	41
Throughput	1-6 samples/run
	90 samples/8 hr
Loading Capacity	20-100 µg Typical Load
	2.0 mg Maximum Load

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Free-Flow Electrophoresis



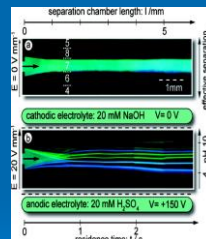
The miniaturization of FFE implies several advantages especially considering sample volume and separation speed. In contrast to the tens of milliliters of sample consumed by conventional large scale FFE devices, microfluidic FFE systems require only tens of nanoliters up to hundreds of microliters of sample. This is especially interesting in clinical analysis where often only low sample volumes are available. Furthermore, instead of residence times of up to tens of minutes, microfluidic FFE (μ -FFE) devices separate within several seconds.

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Microfluidic high-resolution free-flow isoelectric focusing



FFIEF of seven fluorescent IEF markers

Voltage = 150 V, current = 50 mA

Markers (pI 4, 5.1, 6.2, 7.2, 8.1, 9, and 10.3) are fully separated within less than 2 s.

The sample flow rate was 0.4 mL/min ($v = 2$ mm/s).

The apparent kinks in the fluorescent tracer paths are caused by merging multiple photographs.

Copyright American Chemical Society. © 2008

Kohlheyer, D., Eijkel, J. C. T., Schlautmann, S., van den Berg, A., Schasfoort, R. B. M., *Anal. Chem.* 2007, 79, 8190-8198.

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Preparative Free Flow Electrophoresis



FFE Service GmbH
Dr. Gerhard Weber
D-85551 Kirchheim
Germany

WEBER, Gerhard, Margentisweg 23 85551 Kirchheim (DE).
WO/2002/050524, 07.12.2001,

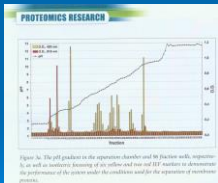
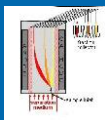


Figure 36. The pH gradient in the separation chamber used in fraction web, separation time as well as necessary flow rates of anolyte and catholyte are determined. The performance of the system under the conditions used for the separation of membrane proteins.



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IEF v rozbíhávém toku (divergent flow IEF, DF IEF)

Základní idea

- Fluidika – kontinuální rozšiřování plochého kanálu při toku kapaliny od vstupu k výstupu při čemž je generován rozbíhavý tok

a současně,

- IEF - malé příčné napětí na vstupu kanálu a vysoké příčné napětí na výstupu kanálu

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IEF v rozbíhávém toku

Fluidika – rozbíhavý tok

IEF – řízení elektrického proudu vodivostí kapaliny κ

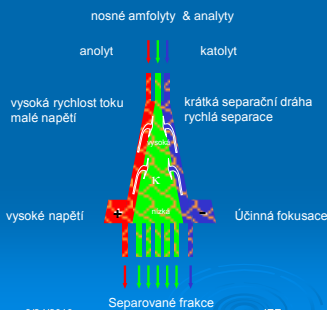
Jednoduché zařízení:

Membrány eliminovány použitím porézního lože

Separací plocha a vstupy a výstupy kapaliny tvořeny netkanou textilíí

Kontakty k elektrodám tvořeny netkanou textilíí

Tok generovaný hydrostaticky



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Divergent flow IEF

Šlais K. Electrophoresis 29 2008 2451-2457

The polypropylene nonwoven web 0.1 mm thick lies on white polyvinylchloride flexible sheet

Input strips dipped in Petri dishes containing:
above left – anolyte
above middle – solution of carriers and pI markers
above right – catholyte



middle left – carbon rod anode
middle right – carbon rod cathode

output strips – bottom – microplate

Streamlines of red pI markers from left -
pI = 3.3, 4.7, 6.2, 7.6, 11.0

Flow due to hydrostatics and capillary elevation

Constant power load 1 W

No cooling

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Dynamics of divergent flow IEF

Šlais K. Electrophoresis 29 2008 2451-2457

1 W constant power load

switched off at 11 hod 30 min

switched on at 11 hod 40 min

Flow inputs:
Anolyte: 0.05 M H_2PO_4 , 5.2 mS/cm, 1 mL/h
Catholyte: 0.05 M NaOH, 11 mS/cm, 1 mL/h
Carriers and pI markers: 0.75 mS/cm, 4 mL/h,

Holdup volume: 1 ml
Separation area: 71 cm^2

Streamlines of red pI markers from left
pI = 3.3, 4.7, 6.2, 7.6, 11.0

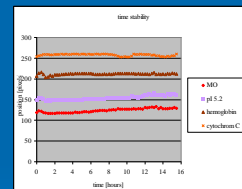


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Časová stabilita polohy linií IEF v rozbíhávém toku



Linie od leva

oranžová - methyl oranž, levandulová - marker pI 5.2
hnědá - hemoglobin, 0.5 mg/ml, chlořá - cytochrom
C, 0.5 mg/ml, průtok 0.18 mL.min⁻¹

Kolísání linií 3.96 %, 3.94 %, 1.26 % a 1.88 %

Šlais M., Šlais K. Electrophoresis, 2008, 29, 4503-4507

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Preparativní DF IEF piva



Mazanec K., Bobalova J., Slais K. Anal Bioanal Chem 2009, 393, 1769-1778

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DF IEF extraktů ječmene, sladu a piva

Kontinuálně dávkované pl markery: 1 kapka roztoku směsi pl markerů
 oranžová - marker pl 2.5
 fialová - marker pl 11



Vstup - Extrakt ječmene (neodsolený) + pufovy + markery pl 2.5 a 11
 průtok - 0.23 ml/min,
 vodivost - 1.0 mS/cm
 Vstupní elektrody: 4 mA, 20 V
 Výstupní elektrody: 6 mA, 800 V

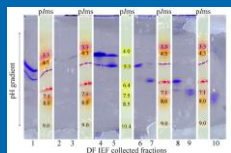
Mazanec K., Bobalova J., Slais K.
 Anal Bioanal Chem 2009, 393, 1769-1778

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Kombinovaný sken IEF gelu frakcí z DF IEF piva



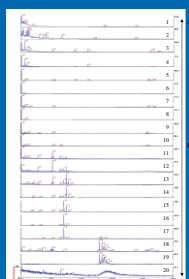
Barevné pl markery skenované ihned po gel IEF
 Proteiny skenované po vybarvení Comassie

Mazanec K., Bobalova J., Slais K.
 Anal Bioanal Chem 2009, 393, 1769-1778

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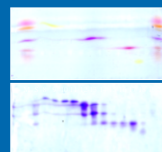
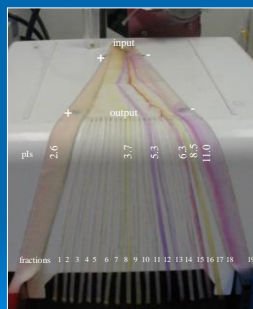
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MALDI-MS spektra 20ti DF IEF frakcí proteinů ze surového extraktu ječmenného sladu.

Preparative divergent flow IEF without carrier ampholytes for separation of complex biological samples



Separation of proteins in individual yeast lysate DF IEF fractions by polyacrylamide gel IEF.

DF IEF without carrier ampholytes with yeast lysate sample and colored pl markers.

Desalting, preconcentration, preseparator

M. Slasna, K. Slais, Electrophoresis, 31, 2010, 433-439

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Projekt FRVŠ 2012

Vypracování laboratorní úlohy na zařízení pro izoelektrickou fokusaci v rozbíhacím toku

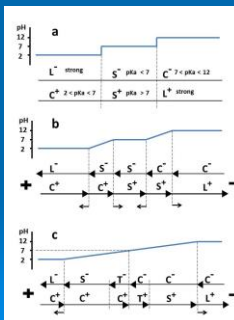
- laboratorní úloha vyučována v rámci předmětu C9320 Metody biochemického výzkumu na PIF MU
- výuky se zúčastnilo 45 studentů
- projekt byl úspěšně obhájen před komisí PIF MU 20. února 2013

- Použití chladicího boxu;
- Chlazení
- Odpařování
- Kontaminace
- Bezpečnost



8/24/2013

Electrolyte system for fast preparative focusing in wide pH range based on bidirectional isotachopheresis (BITP)



L⁻ - leading anion of strong acid
 L⁺ - leading cation of strong base,
 C⁻ - anionic counter ions,
 C⁺ - cationic counter ions,
 S⁻ - anionic spacers,
 S⁺ - cationic spacers,
 T⁻ - the fastest C⁻ in LB in anionic ITP part and
 T⁺ - the fastest C⁺ in LA in cationic ITP part.

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The composition of LB, LA and spacer electrolytes used for simulation and in the experiment verification

electrolyte	pH	$\alpha \times 10^3$ ($\mu\text{m}^2 \cdot \text{s}^{-1}$)	M_0	standard error (std)	error (imp.) (%)
LB	10.70	26.4	169.15	3	1023.3
GA	10.80	28.8	131.20	10	1312.0
GABA	10.50	20.0	103.00	10	1030.0
Phospon	10.24	10.8	89.10	10	891.0
phosfo	9.78	17.4	78.00	7	523.0
phospono	9.20	18.6	100.00	5	600.0
TAPS	8.50	25.0	243.28	5	1216.4
TAPSO	7.50	26.0	292.28	5	1268.4
NH ₄ Cl	11.70	151.9	400.00	60	2400.0

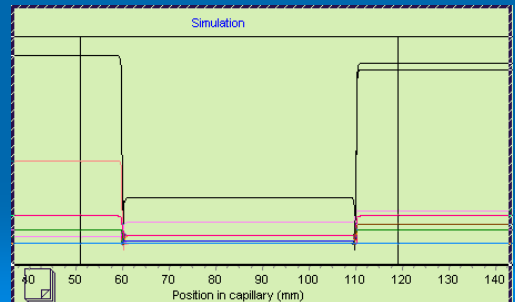
electrolyte	pH	$\alpha \times 10^3$ ($\mu\text{m}^2 \cdot \text{s}^{-1}$)	M_0	standard error (std)	error (imp.) (%)
phospono	7.20	26.9	209.50	3	627.9
ACES	6.70	11.3	162.20	3	486.6
MEK	6.00	28.0	213.20	3	639.6
phospono	5.30	20.6	123.11	3	369.3
phospono	4.76	42.3	194.77	3	584.3
glycolic acid	3.80	42.4	118.10	3	354.3
glycolic acid	2.16	12.12	348.6	421.5	1021.2
citric acid	3.10	102.0	161.00	3	483.0
citric acid	2.30	40.00	112.20	3	336.6
Tris	8.08	20.5	121.10	3	363.3
urea	8.18	141.5	87.00	3	261.0
urea	6.70	101.5	160.10	3	480.3
urea	9.00	20.0	101.14	3	303.4
phospono	10.00	20.0	101.14	3	303.4
phospono	10.10	10.0	100.8	10	1008.0

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Computer simulation of dynamics in newly suggested electrolyte system based on bidirectional ITP



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The animation of the experiment with colored indicators subjected to BIP electrofocusing in newly suggested electrolyte system and carried out on nonwoven strip in V-shape trough during 30 min.

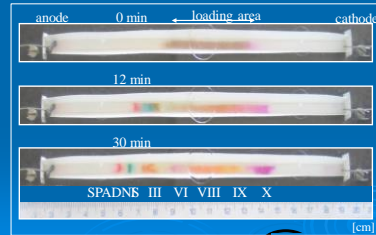


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The examples of representative images displaying bidirectional ITP electrofocusing process in nonwoven strip in V-shape trough with colored pH indicators taken at 0, 12 and 30 minutes.



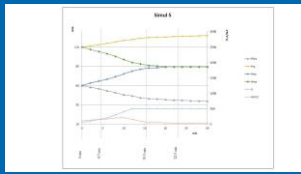
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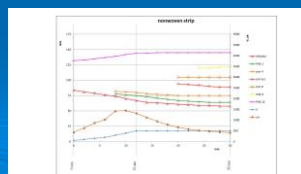
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The electrofocusing dynamics shown as dependence of zone position on analysis time

simulation



experiment carried out on linear nonwoven strip in the V-shape trough

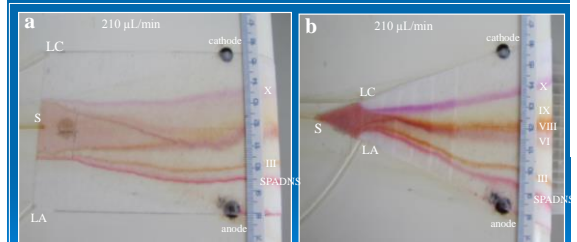


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The images of bidirectional ITP electrofocusing with continuous flow in rectangular (a) and trapezoidal (b) separation beds under the same experimental conditions

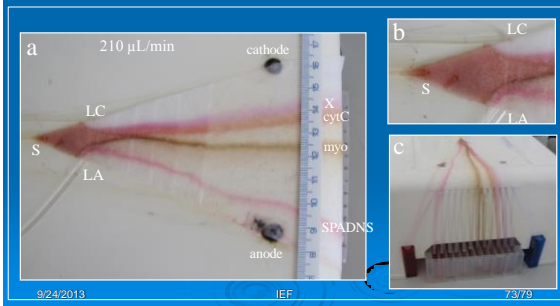


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The example of bidirectional ITP separation and electrofocusing in continuous flow of cytochrome C (cytC) and myoglobin (myo)



FF- BITP v rozbíhavém kanále

možnost analýzy větších částic (bakterie, buňky)



Conclusions

The examined electrolyte system is suggested mainly for preparative sample treatments with relatively low (reasonable) number of eluted collected separated fractions, typically between 10 to 20 fractions. Thus, the total sum of spacers and counter ions does not need to substantially exceed number of 20.

compounds needed for preparation of the running solutions can be chosen from available cheap components,

The buffers are typically low molecular organic molecules which can contribute to compatibility with downstream fraction processing.

Running solutions can be optimized in advance and, *vice versa*, observed experimental results can be directly used for simulation and further solution modification.

By nature of ITP, the described method of focusing by bidirectional ITP with use of multiple counter ions is directly applicable for the separation and enrichment of ampholytes as well as both weak and strong electrolytes.

When the suggested electrolyte system is used in continuous flow devices the properties of leading electrolytes (mainly their pH and conductivity) make them simultaneously the electrode electrolytes. This strongly simplifies both the device construction and the operation.

The main advantage of the use of the suggested electrolyte system is seen in the possibility to use high current densities at the initial stages of focusing without danger of the local overheating. This fact strongly reduces the time needed for analysis completion and it applies to both modes of suggested bidirectional ITP including linear mode and planar arrangement with continuous flow.

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Electrolyte system for fast preparative focusing in wide pH range based on bidirectional isotachopheresis.

Slais, Karel; Stastna, Miroslava

