

Izoelektrická fokusace

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Izoelektrická fokusace - IEF

- elektromigrační separační analytická metoda využívající existence izoelektrického stavu amfolytů, kdy efektivní náboj je nulový.
- $pH = pI$
- Analyty - proteiny
- Separace - $\Delta pI < 0,01$
- Fokusace – zakoncentrování
- Charakterizace - pI



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Protein jako amfolyt

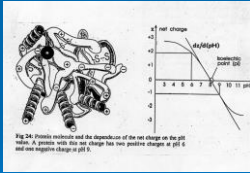


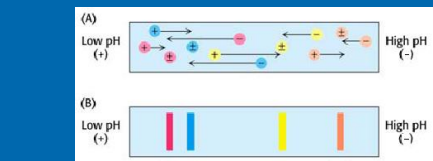
Table 5-2. Isoelectric Points of Several Common Proteins

Protein	pI
Pepsin	<1.0
Ovalbumin (hen)	4.6
Serum albumin (human)	4.9
Tropomyosin	5.1
Insulin (bovine)	5.4
Fibrinogen (human)	5.8
γ -Globulin (human)	6.6
Collagen	6.6
Myoglobin (horse)	7.0
Hemoglobin (human)	7.1
Ribonuclease A (bovine)	9.4
Cytochrome c (horse)	10.6
Histone (bovine)	10.8
Lysozyme (hen)	11.0
Salmine (salmon)	12.1

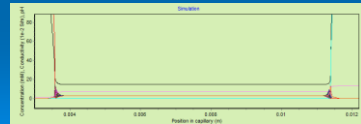
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Simul 5



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Druhy IEF

- Gelová IEF
 - S nosnými amfolyty
 - S imobilizovaným gradientem (IPG)
 - Dvourozměrná elektroforéza 2D = IEF + SDS PAGE
- Kapilární IEF
- Preparativní IEF
 - Free flow
 - Komorová (např. Rotofor)

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Typický výsledek gelové IEF

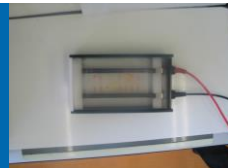


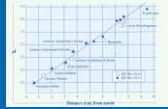
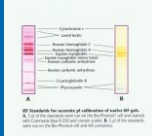
Fig. 2. Typical separation in the Model 111 Mini IEF Cell (5% polyacrylamide gel with 2% Bio-Lyte 3/10 ampholytes). Focusing was carried out in a stepped fashion (100 V for 15 minutes, 200 V for 15 minutes, 450 V for 1 hour). Samples are: lanes 1 & 10: Bio-Rad's IEF Standards; lanes 2-5: Dilutions of horseradish peroxidase; lanes 6-9: Dilutions of Japanese water moccasin snake venom.

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Standardy pI - proteiny



nestabilní,
nečisté,
drahé,
málo barevné,
málo rozpustné při pI



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Nízkomolekulární barevné pI markery



- Požadavky na pI markery
 - Škála pI od ~2 do 11, po ~ 0.5 pI
 - dobře emfolytí, $-dz/dpH > 0.05$, $\Delta pK < 2 < 4$
 - rozpustnost ve vodě při $pH = pI$, $> 1 \text{ mg/ml}$
 - různé barvy, $\lambda_{max} > 400 \text{ nm}$, $A_{1\%} > 100$
 - Čistota, $> 99\%$
 - Dostupnost, cena markeru
 - Stabilita - hydrolyza, oxidace, fotodegradace, mikroorganismy

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Aminomethylované nitrofenoly

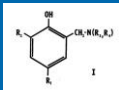


TABLE I
ISOTHERMAL DISSOCIATION CONSTANTS OF SEVERAL FORMULAS I

No.	R_1	R_2	R_3	R_4	R_5	K_{12}	K_{13}	K_{14}	K_{15}
1	H	H	H	H	H	0.000	0.000	0.000	0.000
2	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
3	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
4	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
5	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
6	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
7	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
8	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
9	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
10	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
11	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
12	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
13	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
14	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
15	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
16	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
17	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
18	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
19	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
20	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
21	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
22	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
23	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
24	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
25	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
26	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
27	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
28	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
29	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
30	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
31	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
32	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
33	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
34	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
35	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
36	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
37	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
38	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
39	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
40	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
41	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
42	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
43	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
44	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
45	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
46	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
47	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
48	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
49	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
50	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000

TABLE II
ISOTHERMAL DISSOCIATION CONSTANTS OF SEVERAL FORMULAS II

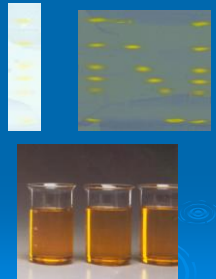
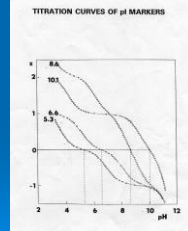
No.	R_1	R_2	R_3	R_4	R_5	K_{12}	K_{13}	K_{14}	K_{15}
1	H	H	H	H	H	0.000	0.000	0.000	0.000
2	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
3	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
4	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
5	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
6	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
7	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
8	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
9	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
10	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
11	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
12	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
13	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
14	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
15	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
16	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
17	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
18	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
19	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
20	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
21	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
22	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
23	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
24	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
25	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
26	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
27	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
28	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
29	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
30	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
31	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
32	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
33	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
34	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
35	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
36	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
37	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
38	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
39	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
40	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
41	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
42	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
43	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
44	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
45	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
46	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
47	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
48	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
49	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000
50	H	H	H	H	CH ₃	0.000	0.000	0.000	0.000

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Žluté pI markery

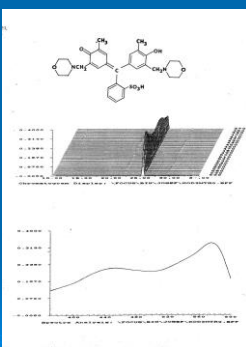


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Příklad barevného pI markeru

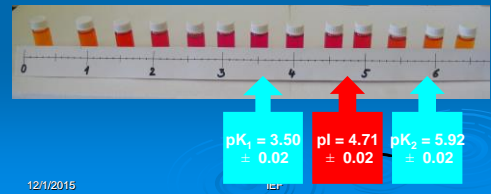
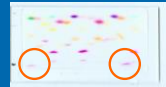
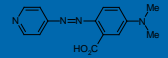


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Spektrofotometrické určení pI



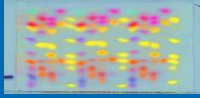
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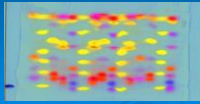
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Určení pI interpolací v gelové IEF

Gradient pH



Směs 30 jednoduchých pufrů



Biolyt 3 - 10

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Dynamika fokusace v gelové IEF

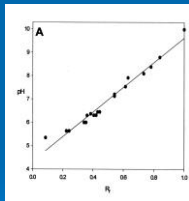


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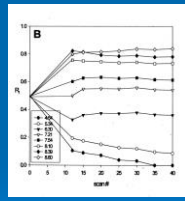
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Dynamika pH gradientu Biolyt 3-10



Lineární gradient pH 4 - 10



Po 1/4 hod malé změny pH gradientu

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Dynamika fokusace v gelové IEF



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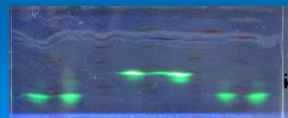
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Vývoj fluorescenčních pI markerů

Vis



fluorescence



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yellow markers

Mass spectrometric characterization of low-molecular-mass color pI markers and their use for direct determination of pI value of proteins

Mazanec, K., Slais, K., Chmelik, J.

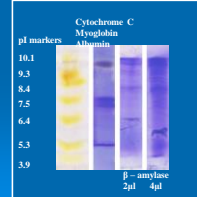
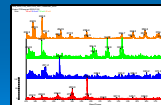
J. Mass Spectrom. 41 2006 1570-1577

Pardubice 2005



Mass spectra of nitro-substituted pI markers

pI	Marker	Structure	Chemical	Formula
10.1	1,4-DIAMINOTOLUENE	NO ₂	<chem>Nc1ccc(N)cc1</chem>	<chem>C7H9N2</chem>
9.3	1,4-DIAMINOBENZENE	NO ₂	<chem>Nc1ccc(N)cc1</chem>	<chem>C6H8N2</chem>
8.4	1,4-DIAMINOPYRIDINE	NO ₂	<chem>Nc1ccncc1N</chem>	<chem>C5H7N3</chem>
7.5	1,4-DIAMINOPYRIMIDINE	NO ₂	<chem>Nc1ccncc1N</chem>	<chem>C4H6N4</chem>
6.4	1,4-DIAMINOPYRIMIDINE	NO ₂	<chem>Nc1ccncc1N</chem>	<chem>C4H6N4</chem>
5.3	1,4-DIAMINOPYRIMIDINE	NO ₂	<chem>Nc1ccncc1N</chem>	<chem>C4H6N4</chem>
3.9	1,4-DIAMINOPYRIMIDINE	NO ₂	<chem>Nc1ccncc1N</chem>	<chem>C4H6N4</chem>

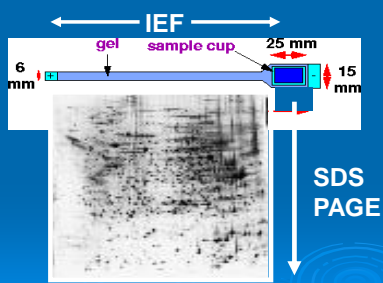


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2D Gel electrophoresis

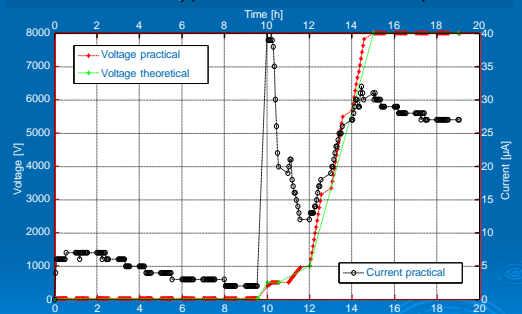


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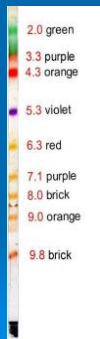
Záznam typické IEF na IPG stripu



Credit:

Deng, X., Hahne, T., Schröder, S., Redweik, S., Nebija, D., Schmidt, H., Janssen, O., Lachmann, B., Wätzig, H., (2012). The challenge to quantify proteins with charge trains due to isoforms or conformers. *Electrophoresis*, 33(2), 263–9. doi:10.1002/eips.201100321

Test of color pI markers - LM ladder



Hanspeter Schickel,
ETC Elektrophorese-Technik GmbH, Kirchentellinsfurt, Germany

Strips rehydrated 2 hours under Kerosene
run native 7 hours with Nitrogen
Amersham Multiphor.

Courtesy of Dr. Hanspeter Schickel

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IEF of mixture of chosen pI markers in the first dimension strip of 2D gel electrophoresis

in *Clinical Proteomics. From Diagnosis to Therapy*. J. Van Eyk and M.J. Dunn (Eds.), Chapter 2. Protein Separation by Two-Dimensional Electrophoresis
Pamela M. Donoghue, Miroslava Stastna, Michael J. Dunn, p 13,
2007 WILEY-VCH Verlag GmbH & Co. KGaA, Weinheim



Immobiline Dry Strip (Amersham Biosciences) pH 3–10, 18 cm.
Apparatus: Protean IEF Cell (BioRad).
Sample: 10 µl of pI markers mixture diluted with 340µl of IEF buffer (8M urea, 2M thiourea, 4% CHAPS, 1% DTT, 0.01% bromophenol blue, 1.5% (v/v) hydroxyethyl disulfide, 0.2% (v/v) IPG buffer pH 3–10).

The acidic end is on the left and the basic end on the right side of the strip.
The pI values of individual pI markers are marked in the picture

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2D - typical result – silver staining

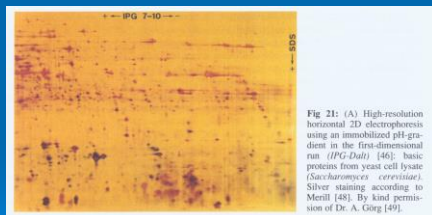


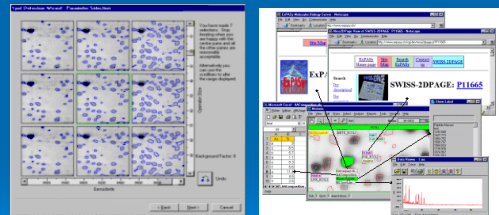
Fig 21: (A) High-resolution horizontal 2D electrophoresis using an immobilized pI-gradient in the first-dimensional run (IPG-Dial) [46]; basic proteins from yeast cell lysate (*Saccharomyces cerevisiae*). Silver staining according to Merrill [48]. By kind permission of Dr. A. Görg [49].

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2D Gel electrophoresis - Software

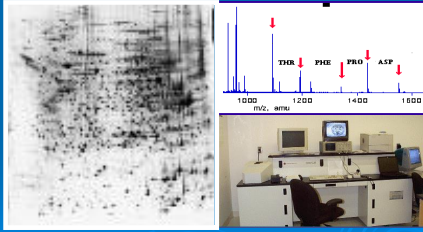


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Protein identification by 2D gel electrophoresis -MS



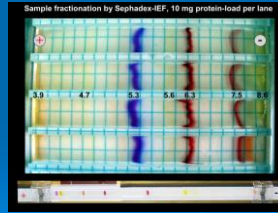
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IEF in Granulated Sephadex Gels

Methods in Molecular Biology, vol. 424: Volume 1: Sample Preparation and Pre-Fractionation, Edited by: A. Posch, Chapter 22: Sample Prefractionation in Granulated Sephadex IEF Gels, Angelika Görg, Carsten Lück, and Walter Weiss, p 277, Humana Press Inc., 2007, Totowa, NJ



Use of coloured pI - markers to determine the slope of the pH gradient and the position where to cut and remove the individual Sephadex fractions in order to fit to the corresponding narrow pH range IPGs

Courtesy of Carsten Lück

IEF in Sephadex gels and IPG strips

Hodný Z., Přidalová J.,

Institute of Experimental Medicine AV ČR, v.v.i., Prague



Courtesy of Z. Hodný



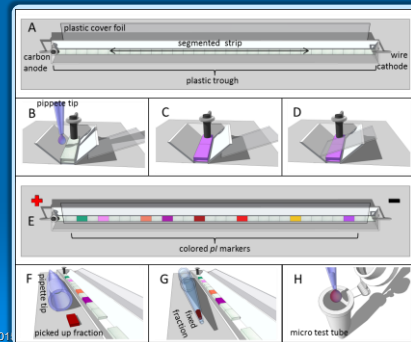
pI markers - LM ladder
Home made strip,
linear gradient pH 4-10,
11 cm,
1 min 30V,
50 min 30V => 3500V,
2 hours 3500V,
Courtesy of J. Přidalová

12/1/2015

IEF

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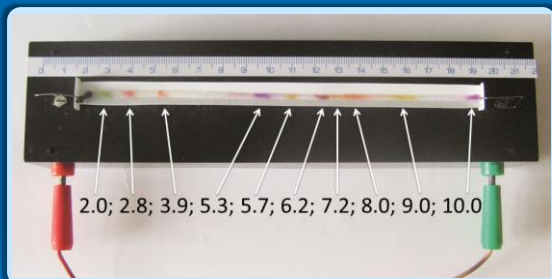
Mikropreparativní sIEF v proužku netkané textilie



28/1/201

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sIEF nosných pufrů s barevnými pI markery



fokusovaná směs:

0,15 ml zásobního roztoku 12ti nosných pufrů v hydroxidu sodném
0,05 ml zásobního roztoku barevných pI markerů
0,15 ml ethylen glykolu, 0,05 ml butanolu, 1,1 ml vody

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Animace průběhu separace

- separace probíhá většinou cca 12 hodin
 - večer nadávkování vzorku a zapnutí zdroje
 - ráno možno sbírat frakcionovaný vzorek
 - pokud není možnost extrahovat frakce, zdroj udržuje nejvyšší napětí do příchodu obsluhy



30/1/2015

IEF

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Odběr frakcí

- dvě základní konfigurace pro odběr frakcí:
 - proužek při separaci **v celku** – frakce vybrány po IEF na základě polohy markerů a vystříhány
 - proužek dopředu **nařezaný na kousky** definované délky a kousky jsou před separací vyrovnány do fokusačního korýtka a přilepeny roztokem sacharózy příp. jiné fixační látky
 - při lepení možnost přidat i směs nosných pufrů a barevných markerů → „instantní proužek“ → stačí přidat vodu a vzorek a zapnout zdroj

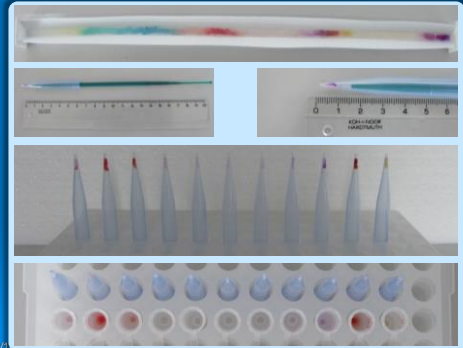


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IEF

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extrakce frakcí - promývání

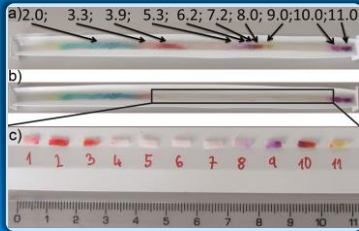


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Frakcionace syrovátky

- optimalizace pro dosažení co nejvyšší čistoty caseinomacropéptidu
 - separovaný vzorek – 1% (m/v) roztok surové syrovátky
 - přidání spacerů – IMAC (imidazol-1-yl-octové kyseliny) a Tris (tris(hydroxymethyl)aminometanu)



program ext. zdroje:

100 V – 200 V 4 hod.
200 V – 1000 V 4 hod.
1000 V – 3000 V 4 hod.
3000 V → do sběru frakcí

vzorek:
0,375 ml 1% (w/v) syrovátky;
0,05 ml zásobního roztoku;
barevných pI markerů;
0,1 ml ethylen glykolu;
0,05 ml butanolu;
0,025 ml 0,1 mol⁻¹ IMAC;
0,1 ml of 0,1 mol⁻¹ Tris;

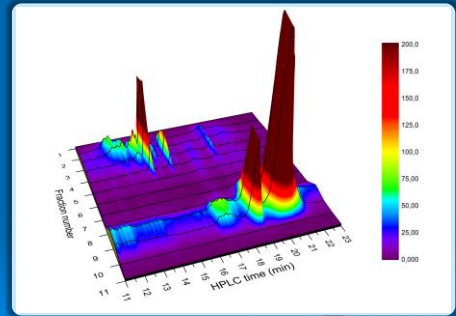
extrakce: 8mm proužek do
100 µl vody

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IEF

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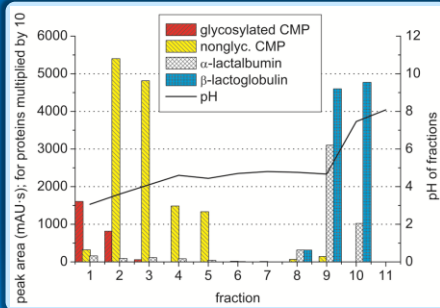
HPLC analýza frakcí získaných z IEF



Kroma Microbore Poroshel 30080-C18 (5 µm částice, 1-75 mm) + C18 předtisková, pH 7,0 °C
Proužek 20 µl min⁻¹
33/1/2015 Gradient: 5-20 min 0-20% TFA lineární gradient od 5 do 80% TFA ACN (30 min)
stávající obsah: 4 µl

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Obsah jednotlivých proteinů ve frakcích

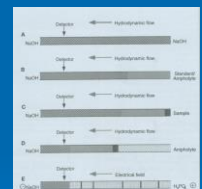
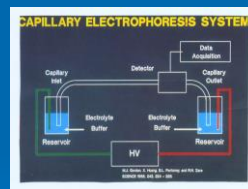


Výtěžek:
glykosylovaný CMP 44%
n glykosylovaný CMP 80%
α-laktalbumin 77%
β-laktoglobulin 101%

35/1/2015

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Kapilární IEF



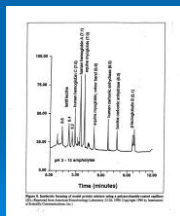
12/1/2015

IEF

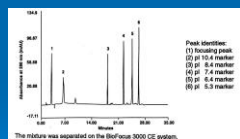
36/79

Kapilární IEF standardů

proteiny



pI markery

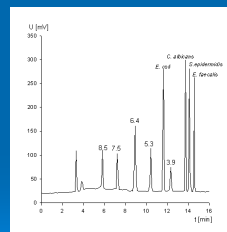


12/1/2015

IEF

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CIEF mikroorganismů



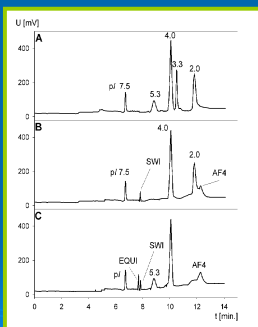
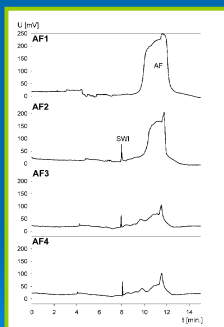
Sample: *E. coli*, *C. albicans*, *S. epidermidis*, *E. faecalis* in physiological saline solution, 4×10^9 cell ml^{-1} .

12/1/2015

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CIEF virů s UV detekcí



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Patogeny z různých zdrojů

Table 1. The strains of the plant pathogens included in this study, comparison of their isoelectric points, pI, and RSDs from three measurements of the migration times, t, for each from the strains.

Abbreviation in Figs.	Strain	pI
C. michiganensis	<i>Clavibacter michiganensis</i> subsp. <i>michiganensis</i> CCM 1835	4.6
	<i>Clavibacter michiganensis</i> subsp. <i>michiganensis</i> VURV C204	4.6
	<i>Clavibacter michiganensis</i> subsp. <i>michiganensis</i> VURV 2450	4.7
	<i>Clavibacter michiganensis</i> subsp. <i>michiganensis</i> VURV 5290	4.6
	<i>Clavibacter michiganensis</i> subsp. <i>michiganensis</i> VURV 5295	4.7
	<i>Clavibacter michiganensis</i> subsp. <i>michiganensis</i> VURV 7008	4.7
	<i>Clavibacter michiganensis</i> subsp. <i>michiganensis</i> VURV 7018	4.6
C. michiganensis	<i>Clavibacter michiganensis</i> subsp. <i>michiganensis</i> VURV 7030	4.7
		pI = 4.7, RSD = 1.9 %
X vesicatoria	<i>Xanthomonas vesicatoria</i> CCM 2101	4.0
	<i>Xanthomonas vesicatoria</i> CCM 2102	4.1
	<i>Xanthomonas vesicatoria</i> VURV P-1-1	4.0
	<i>Xanthomonas vesicatoria</i> VURV P-6-1	4.1
	<i>Xanthomonas vesicatoria</i> LMG 2804	4.1
X. vesicatoria	<i>Xanthomonas vesicatoria</i> LMG 667	4.1
		pI = 4.1, RSD = 0.7 %
P. syringae	<i>Pseudomonas syringae</i> pv. <i>tomato</i> CFSP 6422	4.0
	<i>Pseudomonas syringae</i> pv. <i>tomato</i> CFSP 2212	4.0
	<i>Pseudomonas syringae</i> pv. <i>tomato</i> IMA 1733.3	4.0
	<i>P. syringae</i>	pI = 4.0, RSD = 1.9 %
P. corrugata	<i>Pseudomonas corrugata</i> CFSP 4901	2.4
	<i>Pseudomonas corrugata</i> CFSP 5465	2.4
	<i>Pseudomonas corrugata</i> CFSP 6653	2.4
	<i>Pseudomonas corrugata</i> IMA 614.5.3	2.4
<i>P. corrugata</i>		pI = 2.4, RSD = 0.9 %

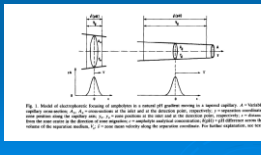
12/1/2015

IEF

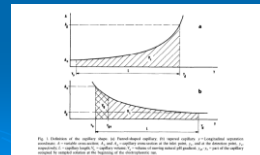
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Tapered capillary in cIEF

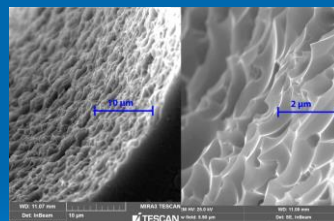
Journal of Chromatography A, 98 (2004) 399–404
Model of electrophoretic focusing in a natural pH gradient moving in a tapered capillary
 Karel Špaňálek
 Institute of Analytical Chemistry, Academy of Sciences of the Czech Republic, 602 00 Brno, Czech Republic
 Received 17 May 2004; revised manuscript received 10 June 2004



Journal of Chromatography A, 103 (2004) 241–247
Transient electrophoretic processes in capillaries of non-uniform cross-section
 K. Špaňálek
 Institute of Analytical Chemistry, Academy of Sciences of the Czech Republic, 602 00 Brno, Czech Republic



Supercritical water in preparation of tapered fused silica capillaries



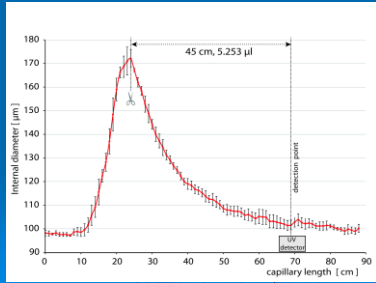
Effect of treatment of 100 μm i.d. fused silica capillary with supercritical water in semi-dynamic mode. Experimental conditions: 400 °C, 32 MPa, 20 replacements of supercritical water.

12/1/2015

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Dependence the local internal diameter of etched fused silica capillary on the capillary length.
The cutout of the segment used as the tapered capillary in cIEF and the detection window are indicated.



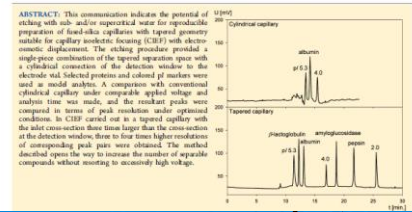
12/1/2015

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Isoelectric Focusing in Continuously Tapered Fused Silica Capillary Prepared by Etching with Supercritical Water

Karel Slais,* Marie Horká, Pavel Karišek, Josef Planeta, and Michal Roth
Institute of Analytical Chemistry of the ASCR, v. i. s. r., Vevří 97, 602 00 Brno, Czech Republic



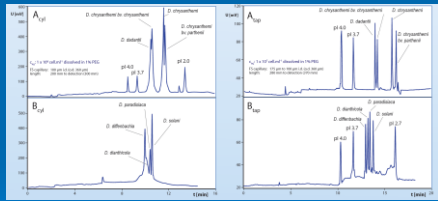
12/1/2015

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Resolution of several *Dickeya* bacterium species with similar isoelectric points by capillary isoelectric focusing employing

cylindrical (left) and tapered (right) capillary.



Combination of Capillary Isoelectric Focusing in a Tapered Capillary with MALDI-TOF MS for Rapid and Reliable Identification of *Dickeya* Species from Plant Samples
Horka, M.; Salplachta, J.; Karasek, P.; Kubesova, A.; Horky, J.; Matouskova, H.; Slais, K.; Roth, M. ANALYTICAL CHEMISTRY Volume: 85 Pages: 6806-6812 JUL 16 2013

12/1/2015

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Preparative Liquid phase IEF

Rotofor

MicroRotofor



Total 2.5 ml sample
Ten 0.25 ml fractions
2 hours run time



Preparativní autofokuse peptidů + pI markerů

pI = 3.0 pI = 4.0

Tomas, R.; Yan, L.S.; Krenkova, J.; Forest, F. ELECTROPHORESIS, 28 (13): 2283-2290 2007

18/10/2013

IEF

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ZOOM® IEF Fractionator (Invitrogen)

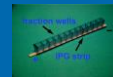
- komůrka pro fokusaci je oddělena polyakrylamidovými disky s definovaným pH
- proteiny musí procházet přes tyto disky dokud nedorazí do izoelektrického bodu
- objem každé části 650 µl
- vysoká reprodukovatelnost
- příprava vzorku: rozpustit, denaturovat, alkylovat, bez zákalu a částic



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IEF

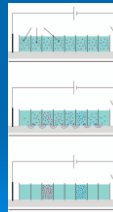
OFF - GEL electrophoresis



P.E. Michel, F. Raymond, I. L. Arnaud, J. Jossier, H. H. Girault, J. S. Rossier, Electrophoresis 2003, 24, 3-11

Agilent 3100 OFFGEL Fractionator

pI-based fractionation of proteins and peptides with liquid-phase recovery, introduced May 30, 2006, Co-developed with DiagnoSwiss S.A.



Use of pI-dye markers as on-line trackers for the focusing of peptides during electrophoresis on the OFFGEL fractionator device (OGE). 22 hours runtime

Courtesy of M. Heller, DKF, University of Bern, Switzerland

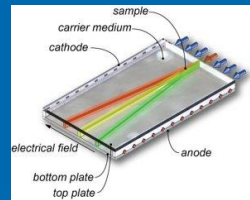
Use of pi-dye markers as on-line trackers for the focusing of peptides during electrophoresis on the OFFGEL fractionator device (OGE).



- pi-marker dyes were added at 10 ug (dark orange, pi 3.9; violet, pi 5.2; red, pi 6.2; bright orange, pi 8.0) or 30 ug (yellow, pi 10.1), respectively.
- Peptide/dye solution was distributed into the 13 wells of the OGE.
- I-PG strips pH 3-10 from BioRad re-hydrated in OGE buffer were used.
- Focusing was done by setting a maximal potential (1250 or 1500 V) and a current limit of 50 uA.

Courtesy of M. Heller, DKF, University of Bern, Switzerland

Free-Flow Electrophoresis



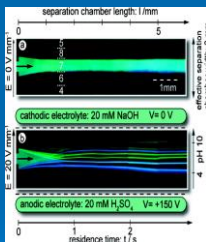
The miniaturization of FFE implies several advantages especially considering sample volume and separation speed. In contrast to the tens of milliliters of sample consumed by conventional large scale FFE devices, microfluidic FFE systems require only tens of nanoliters up to hundreds of microliters of sample. This is especially interesting in clinical analysis where often only low sample volumes are available. Furthermore, instead of residence times of up to tens of minutes, microfluidic FFE (μ -FFE) devices separate within several seconds.

12/1/2015

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Microfluidic high-resolution free-flow isoelectric focusing



FFIIEF of seven fluorescent IEF markers

Voltage = 150 V, current = 50 mA

Markers (pI 4, 5.1, 6.2, 7.2, 8.1, 9, and 10.3) are fully separated within less than 2 s.

The sample flow rate was 0.4 mL/min ($v = 2$ mm/s).

The apparent kinks in the fluorescent tracer paths are caused by merging multiple photographs.

Copyright American Chemical Society, © 2008

Kohlschütter, D., Eijkel, J. C. T., Schlaumann, S., van den Berg, A., Schasfoort, R. B. M., *Anal. Chem.* 2007, 79, 8190–8198.

12/2/2015

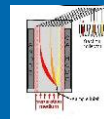
IEF

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Preparative Free Flow Electrophoresis



FFE Service GmbH
Dr. Gerhard Weber
D-85551 Kirchheim
Germany



WEBER, Gerhard; Margartenweg 23 85551 Kirchheim (DE).
WO/2002/050524, 07.12.2001,

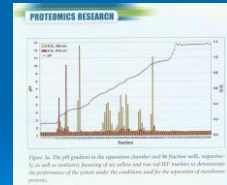


Figure 10. The pH gradient in the separation chamber used for the fractionation of the protein mixture. The pH gradient is shown as a function of the residence time in the separation chamber. The pH gradient is shown as a function of the residence time in the separation chamber.



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IEF v rozbíhavém toku (divergent flow IEF, DF IEF)

Základní idea

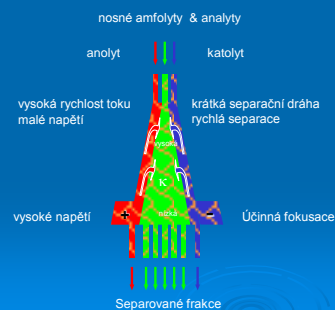
- Fluidika – kontinuální rozšiřování plochého kanálu při toku kapaliny od vstupu k výstupu při čemž je generován rozbíhavý tok
- a současně,
- IEF - malé příčné napětí na vstupu kanálu a vysoké příčné napětí na výstupu kanálu

12/1/2015

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IEF v rozbíhavém toku



Fluidika – rozbíhavý tok

IEF – řízení elektrického proudu vodivostí kapaliny

Jednoduché zařízení:

Membrány eliminovány použitím porézního lože

Separační plocha a vstupy a výstupy kapaliny tvořeny netkanou textilií

Kontakty k elektrodám tvořeny netkanou textilií

Tok generovaný hydrostaticky

12/1/2015

IEF

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Divergent flow IEF

Šlais K. Electrophoresis 29 2008 2451-2457

The polypropylene nonwoven web 0.1 mm thick lies on white polyvinylchloride flexible sheet

input strips dipped in Petri dishes containing:
above left – anolyte
above middle – solution of carriers and pI markers
above right – catholyte



middle left - carbon rod anode
middle right – carbon rod cathode

output strips - bottom - microplate

Streamlines of red pI markers from left -
pI = 3.3, 4.7, 6.2, 7.6, 11.0

Flow due to hydrostatics and capillary elevation

Constant power load 1 W

No cooling

12/1/2015

IEF

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Dynamics of divergent flow IEF

Šlais K. Electrophoresis 29 2008 2451-2457

1 W constant power load

switched off at 11 hod 30 min

switched on at 11 hod 40 min



Flow inputs:
Anolyte: 0.05 M H_2PO_4 , 5.2 mS/cm, 1 mL/h
Catholyte: 0.05 M NaOH, 11 mS/cm, 1 mL/h
Carriers and pI markers: 0.75 mS/cm, 4 mL/h,

Holdup volume: 1 ml
Separation area: 71 cm^2

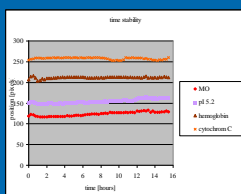
Streamlines of red pI markers from left
pI = 3.3, 4.7, 6.2, 7.6, 11.0

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Časová stabilita polohy linií IEF v rozbíhavém toku



Linie od leva
oranžová - methyl oranž, levandulová - marker pI 5.2
hnědá - hemoglobin, 0.5 mg/ml, cihlová - cytochrom
C, 0.5 mg/ml, průtok 0.18 mL.min⁻¹

Kolísání linií 3.96 %, 3.94 %, 1.26 % a 1.88 %
Štašná M., Šlais K. Electrophoresis, 2008, 29, 4503-4507

12/1/2015

IEF

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Preparativní DF IEF piva



Mazanec K., Bobalova J., Šlais K. Anal Bioanal Chem 2009, 393, 1769-1778

12/1/2015

IEF

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DF IEF extraktů ječmene, sladu a piva

Kontinuálně dávkované pI markery: 1 kapka roztoku směsi pI markerů
oranžová – marker pI 2.5
fialová – marker pI 11



Vstup - Extrakt ječmene (neodosolený) + puřy + markery pI 2.5 a 11
průtok - 0.23 ml/min,
vodivost - 1.0 mS/cm
Vstupní elektrody: 4 mA, 20 V
Výstupní elektrody: 6 mA, 800 V

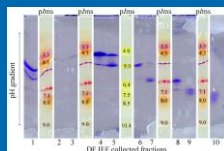
Mazanec K., Bobalova J., Šlais K.
Anal Bioanal Chem 2009, 393, 1769-1778

12/2/2015

IEF

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Kombinovaný sken IEF gelu frakcí z DF IEF piva



Barevné pI markery skenované ihned po gel IEF
Proteiny skenované po vybarvení Comassie

Mazanec K., Bobalova J., Šlais K.
Anal Bioanal Chem 2009, 393, 1769-1778

12/2/2015

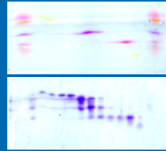
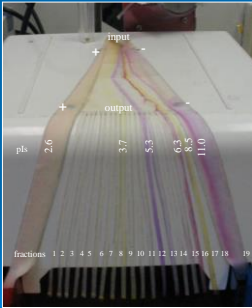
IEF

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MALDI-MS spektra 20ti DF IEF frakcí proteinů ze surového ječmenného sladu.

Preparative divergent flow IEF without carrier ampholytes for separation of complex biological samples



Separation of proteins in individual yeast lysate DF IEF fractions by polyacrylamide gel IEF.

DF IEF without carrier ampholytes with yeast lysate sample and colored pH markers.

Desalting, preconcentration, prepreparation

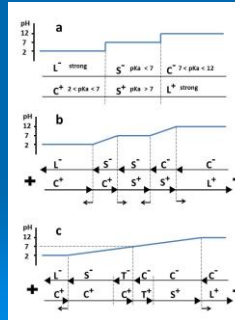
M. Stastna, K. Smit, Electrophoresis, 31, 2010, 433-439

12/2/2015

IEF

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Electrolyte system for fast preparative focusing in wide pH range based on bidirectional isotachopheresis (BITP)



L⁻ - leading anion of strong acid
 L⁺ - leading cation of strong base,
 C⁻ - anionic counter ions,
 C⁺ - cationic counter ions,
 S⁻ - anionic spacers,
 S⁺ - cationic spacers,
 T⁻ - the fastest C⁻ in LB in anionic ITP part and
 T⁺ - the fastest C⁺ in LA in cationic ITP part.

IEF

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The composition of LB, LA and spacer electrolytes used for simulation and in the experiment verification

catholyte	pH	κ (S ⁻¹ cm ⁻¹)	M ₀	standard conc (mg/L)	conc (mg/L)
lysine	10.70	-264	146130	7	10213.1
ACA	10.80	-284	13120	10	13121.0
GABA	10.76	-260	10330	10	10331.0
β-alanine	10.24	-163	8930	10	8931.0
glycine	9.20	-114	7100	7	7101.0
isopropane	9.42	-114	13000	5	6500.0
TAPS	8.30	-250	24120	5	12060.0
TAPSO	7.70	-260	20620	5	12060.0
NGRI	13.70	-1519	4000	80	24000.0

anolyte	pH	κ (S ⁻¹ cm ⁻¹)	M ₀	standard conc (mg/L)	conc (mg/L)
citric acid	2.95-4.52	1274-1216	104130	5	7007.5
β-alanine	3.40	-184	8930	10	8931.0
GABA	4.60	-284	10330	10	10331.0
ACA	4.50	-204	13120	10	13121.0
citronellol	4.85-9.20	-174-172	11330	5	5665.5
EMPS	5.20	-300	12320	5	6160.0
lysine	6.40	-260	20620	5	10310.0
threonine	6.80	-302	13120	5	6560.0
H ₂ O	8.20	-829	9600	20	24000.0

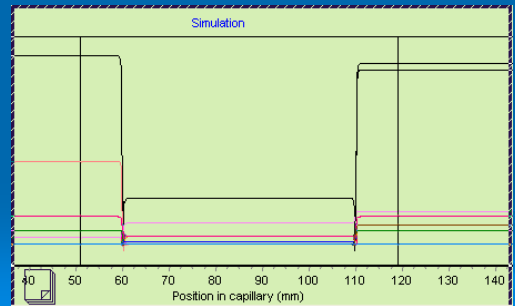
Spacer	pH	κ (S ⁻¹ cm ⁻¹)	M ₀	standard conc (mg/L)	conc (mg/L)
MEPS	7.20	-269	20930	5	10475.0
ACES	6.70	-313	14220	5	7110.0
MEB	6.00	-280	21320	5	10660.0
phosphoric acid	5.30	-204	12310	5	6155.0
acetosuccinic acid	4.70	-453-394	7700	5	3850.0
glycolic acid	3.90	-424	10400	5	5200.0
phosphoric acid	3.163-3.1240	3440-4113	30212	5	15106.0
lactonic	7.10	-1020	6830	5	3415.0
lactate	7.70	-1600	11920	5	5960.0
Trip	8.00	-2095	12130	5	6065.0
isopropylamine	8.30	-1407	8700	5	4350.0
isopropanol	8.70	-1153	10010	5	5005.0
acetamidopyrrolone	9.00	-1800	1014	5	507.0
tolueneformal	10.00	-1800	1014	5	507.0
picolinic	10.00	-1104	8510	4	3404.0

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Computer simulation of dynamics in newly suggested electrolyte system based on bidirectional ITP



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The animation of the experiment with colored indicators subjected to BITP electrofocusing in newly suggested electrolyte system and carried out on nonwoven strip in V-shape trough during 30 min.

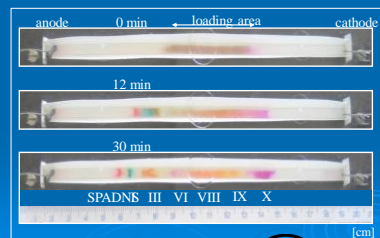


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The examples of representative images displaying bidirectional ITP electrofocusing process in nonwoven strip in V-shape trough with colored pH indicators taken at 0, 12 and 30 minutes.



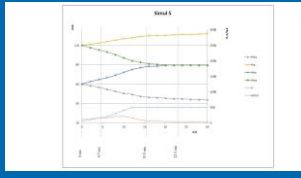
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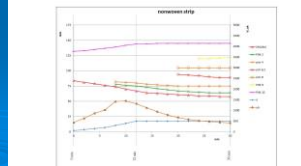
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The electrofocusing dynamics shown as dependence of zone position on analysis time

simulation

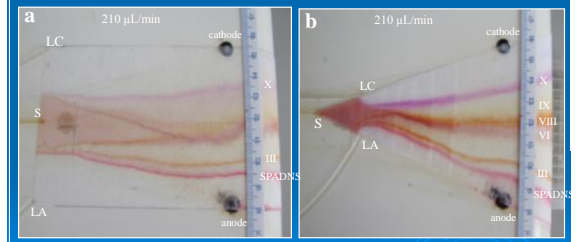


experiment carried out on linear nonwoven strip in the V-shape trough



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The images of bidirectional ITP electrofocusing with continuous flow in rectangular (a) and trapezoidal (b) separation beds under the same experimental conditions

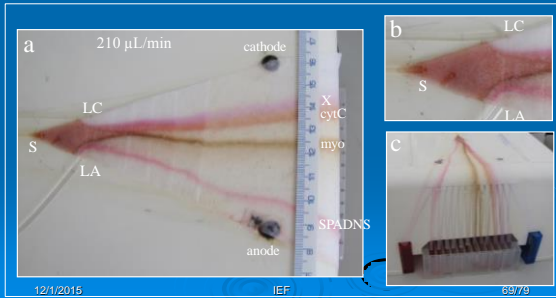


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The example of bidirectional ITP separation and electrofocusing in continuous flow of cytochrome C (cytC) and myoglobin (myo)

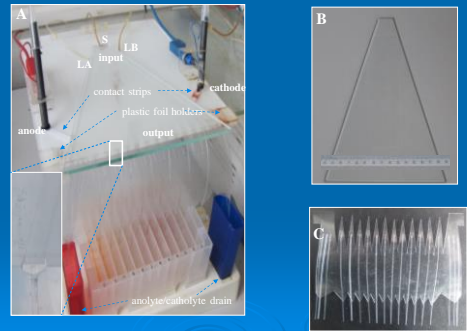


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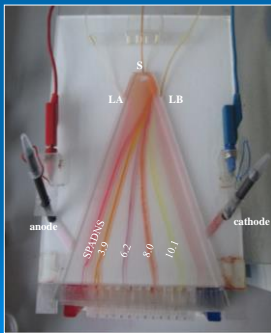
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Continuous fast focusing in trapezoidal void channel based on bidirectional isotachopheresis in wide pH range.

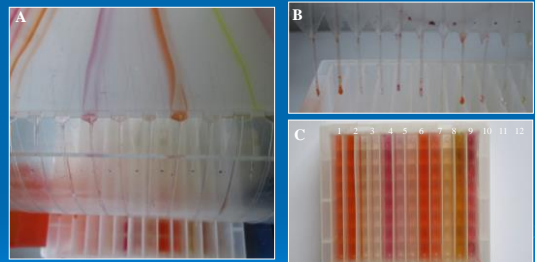


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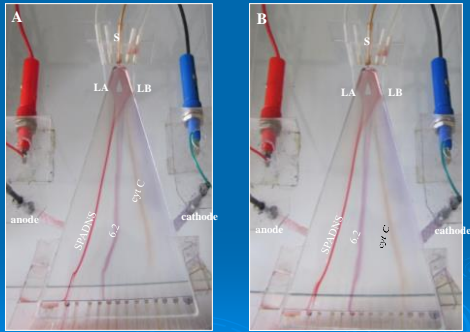
The separation of colored indicators in instrumentation with a larger void closed channel.



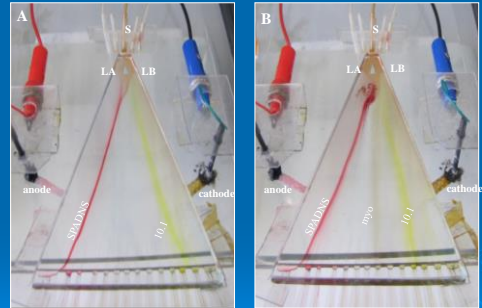
The details of the instrumentation output with collected fractions in twelve well plate.



The separation of two colored indicators and cytochrome C in smaller void channel.



The separation of two colored indicators and myoglobin in smaller void channel.



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