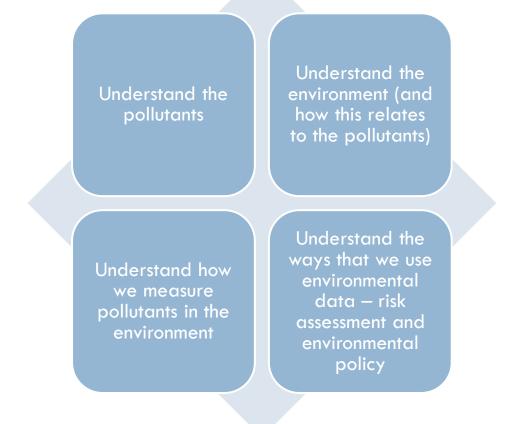
C2003 – ENVIRONMENTAL CHEMISTRY

Course goals:

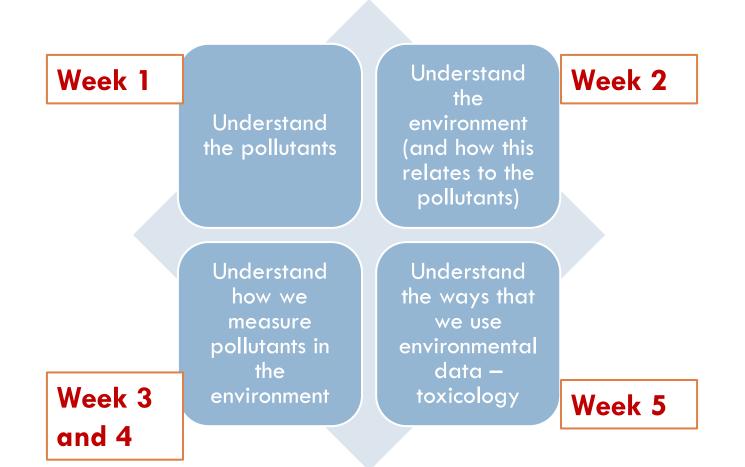
After this course, students should be able to:

understand problems related to pollution of the environment from natural and anthropogenic sources

This is pretty ambitious...so let's break it down a bit...



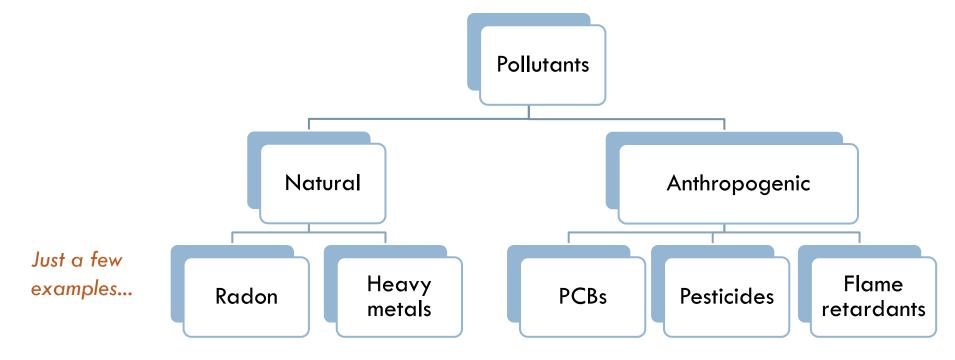
Course overview



What is pollution?

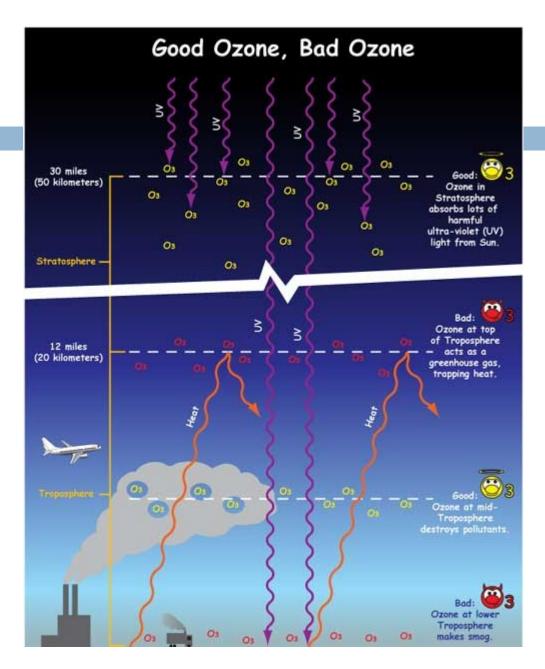
Presence of a substance in an environmental system having a harmful effect

□ The substance = pollutant or contaminant



Pollution depends on context...

- Many are have both natural and anthropogenic sources (e.g., PAHs, metals...)
- Only a pollutant when unwanted adverse effect:
 - E.g., ozone, pesticides...



http://spaceplace.nasa.gov/greenhouse/en/

Environmental chemistry

- Environmental chemistry is the study of chemical processes occurring in the environment which are impacted by human activities.
- Can be local scale, e.g., urban air pollutants or toxic substances from a chemical waste site

-or-

Can be global scale, e.g., long-range pollution transport, global warming

Why is chemistry important to understand pollution?

- A chemical's structure dictates that compound's "personality,"
 - provides a systematic basis to understand and predict chemical behavior in the environment
- With and understanding of the properties and behaviour of chemicals, we can better understand what the impact of humans is on the global environment

Schwarzenbach et al. Environmental Organic Chemistry

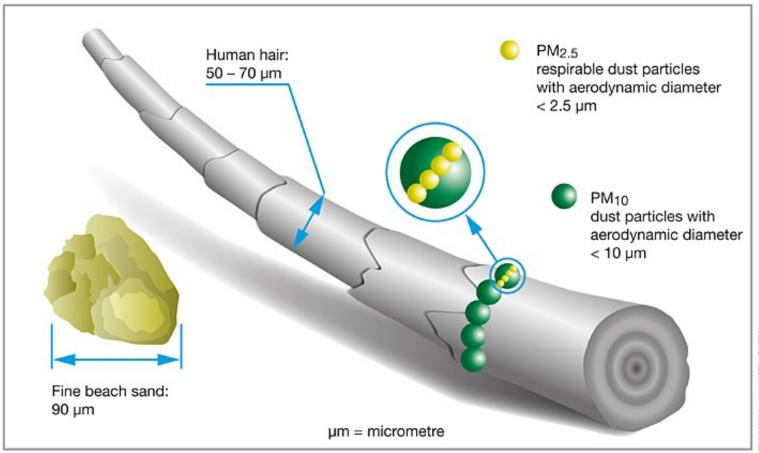
Types of pollutants

- Many classes and methods for classification exist we will consider a few of the major types of pollutants:
 - Volatile organic compounds
 - Airborne particulate matter
 - Persistent organic pollutants
 - Polycyclic aromatic hydrocarbons
 - Heavy metals
 - etc.

Particulate matter (PM)

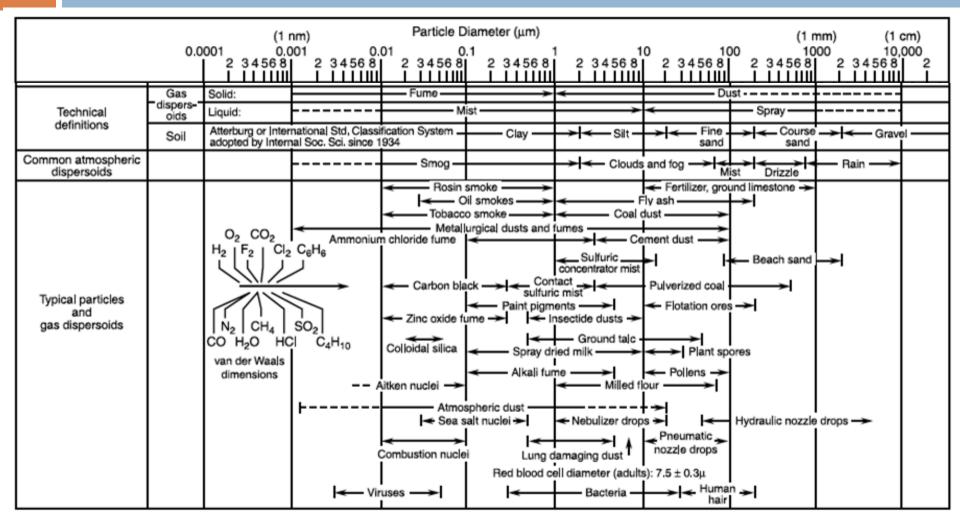
- Solid and liquid particles suspended in air
- Naturally occurring and anthropogenic
- Natural sources:
 - Salt particles from sea spray, pollen, moulds, bacteria, debris from plants and animals, soil particles entrained by wind, etc.
- □ Anthropogenic sources:
 - Industrial processes, open burning, vehicles, agriculture, mining, etc.
- PM is not a specific chemical, but a mixture of particles with different origin, composition, size, shape, etc.
- Important itself (e.g., has negative health effects) and as a carrier for other atmospheric pollutants

Particulate matter



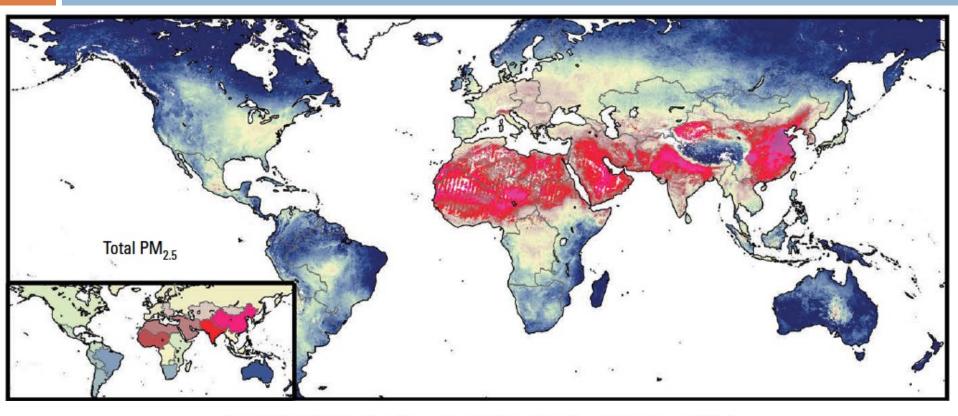
© MANN+HUMMEL GMBH

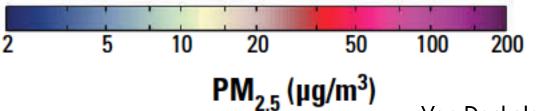
PM sizes and examples



From Finlayson-Pitts and Pitts, 2000, Chemistry of the Upper and Lower Atmosphere

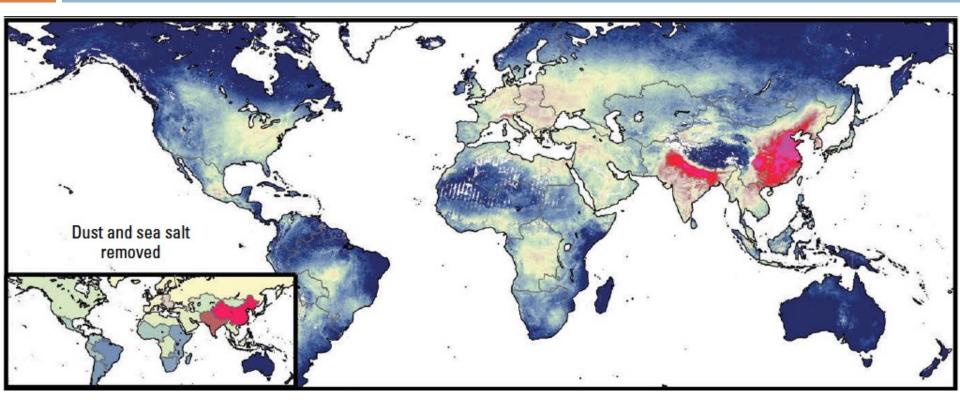
Particulate matter

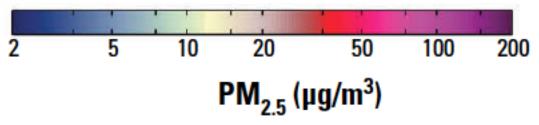




Van Donkelaar et al. EHP 2015

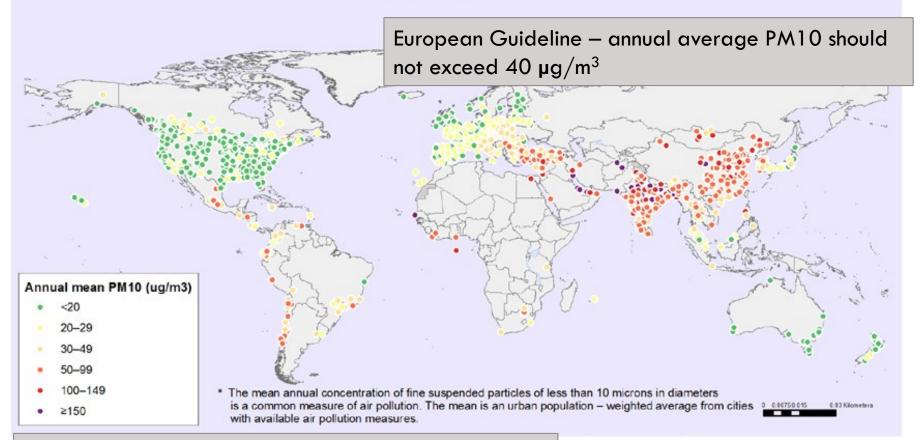
Particulate matter – excluding dust and sea spray





Particulate matter - exposure

Exposure to particulate matter with an aerodynamic diameter of 10 µm or less (PM10) in 1600 urban areas*, 2008–2013



European Environment Agency: "Particulate matter is the air pollutant that poses the greatest health risk to people in Europe." Data Source: World Health Organization Map Production: Health Statistics and Information Systems (HSI) World Health Organization



© WHO 2014. All rights reserved.

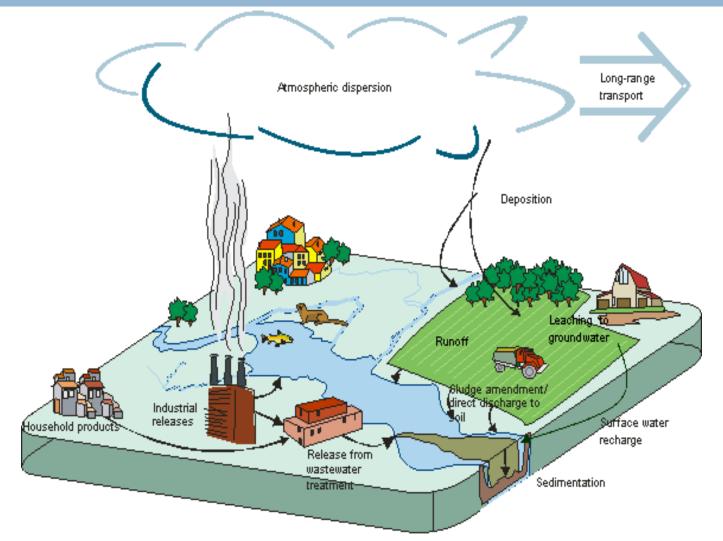
Semivolatile organic compounds (SVOCs)

- □ Not a firm grouping
- Generally determined by vapour pressure
 - **•** typically between ~ 1 and 10^{-10} Pa

Why are they important?

- Can distribute in multiple media (gas-phase air, particle-phase air, soil, water, plants, lipids, floor dust, window films...)
- Many are persistent, lipophilic, bioaccumulative
- Many chemicals of concern are in this group.

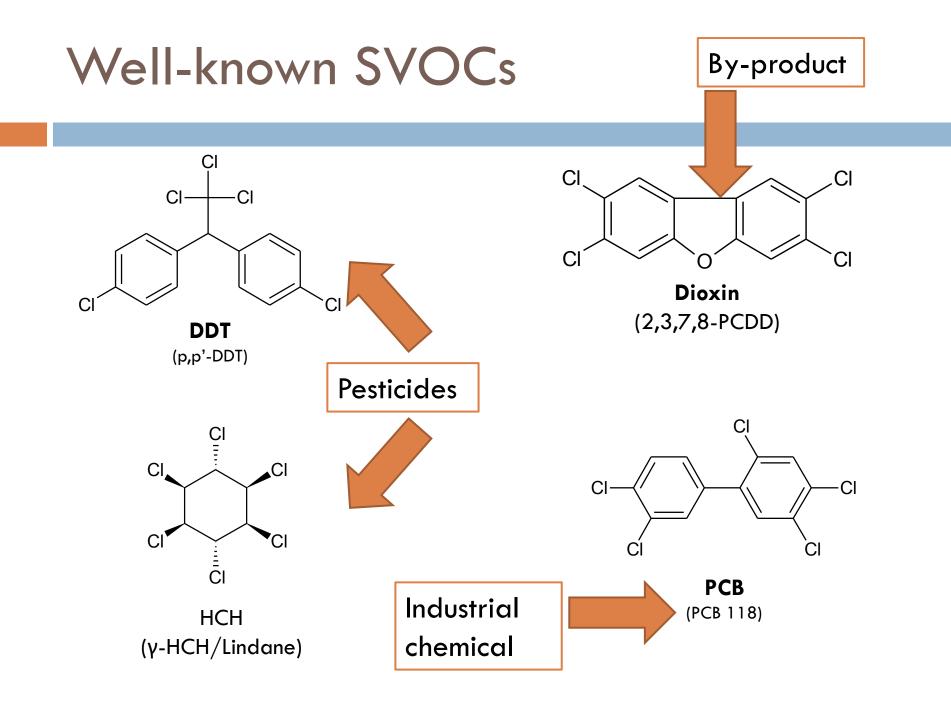
SVOCs in the environment



Canadian Environmental Protection Act (ec.gc.ca)

Examples of SVOCs

- Pesticides
- Industrial chemicals
- By-products
- Additives in consumer products



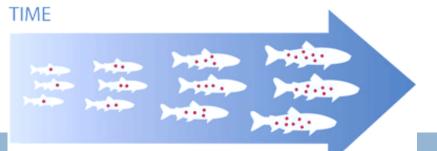


- Many SVOCs are classified as "persistent organic pollutants" (POPs) or "persistent, bioaccumulative and toxic" (PBT)
- □ 3 key terms to understand:
 - Persistence
 - Bioaccumulation
 - Toxicity

Environmental Persistence

- The length of time a chemical remains in environmental system or media
- Governed by the rates at which the compound is removed from the system by biological and chemical processes, such as environmental transport, biodegradation, hydrolysis, atmospheric reactions
- Measured as the half-life of the substance in the medium
- A chemical is considered persistent if it has a half-life of:
 - □ >2 days in air
 - \sim 2-6 months or more in water, sediment or soil

Bioaccumulation



Contaminants

- The accumulation of a chemical in tissues of an organism through any route, including respiration, ingestion, or direct contact with the contaminated environment i.e. Rate of chemical uptake >> rate of chemical loss
- If a chemical is "bioaccumulative" this means that the concentration of the chemical in the tissues of an organism can be significantly higher (e.g., several orders of magnitude) than the concentration of the chemical in the surrounding environment
- Measured by bioaccumulation factor (BAF)
 - BAF > ~1000 means a chemical is considered "bioaccumulative"

Conc. on contaminant in organism

Conc. on contaminant in ambient environment



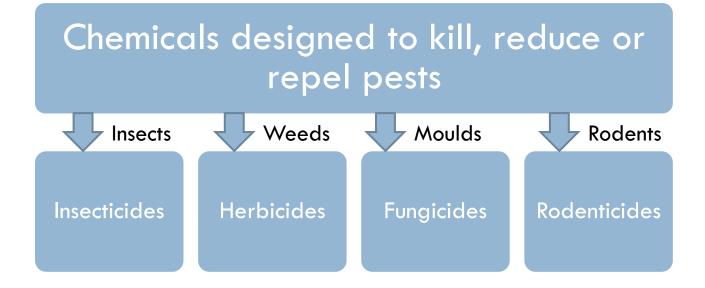
- A measure of the amount which a substance can cause harm to an organism
- Related to the dose of a chemical received by and organism
 - moderately toxic substance can cause harm if an organism receives a higher dose
 - Highly toxic substance can cause harm at low doses

Examples of SVOCs

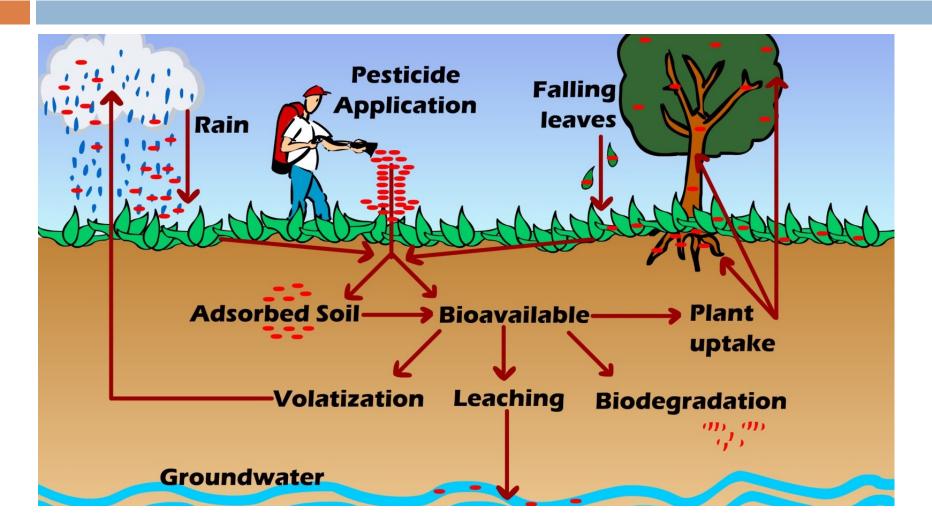
Pesticides

- Industrial chemicals
- By-products
- Additives in consumer products

Pesticides – intentionally toxic!

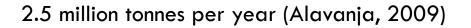


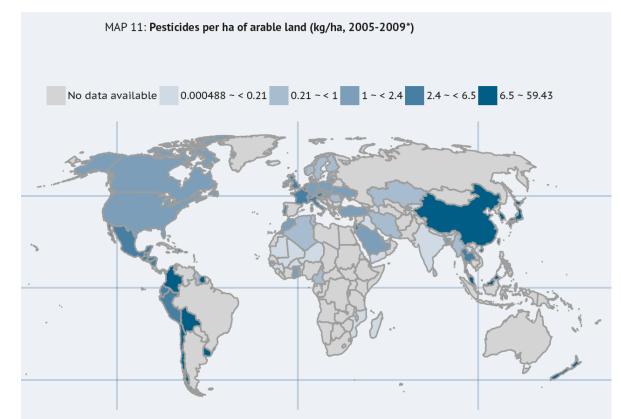
How pesticides enter the environment



Langenbach. "Persistence and bioaccumulation of Persistent Organic Pollutants" 2013

Global pesticide use





From FAO Statistical Yearbook, UN 2013

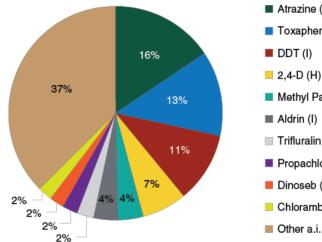


Most common pesticides

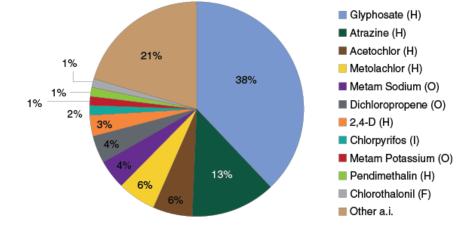
□ In 1960s...

The 10 most heavily used pesticide active ingredients in 1968 included 5 insecticides and 5 herbicides (percent total pounds active ingredient applied on 21 selected crops) 🗆 Today...

The four most heavily used pesticide active ingredients in 2008 were herbicides (percent total pounds active ingredient applied on 21 selected crops)







Note: H = herbicide, I = insecticide.

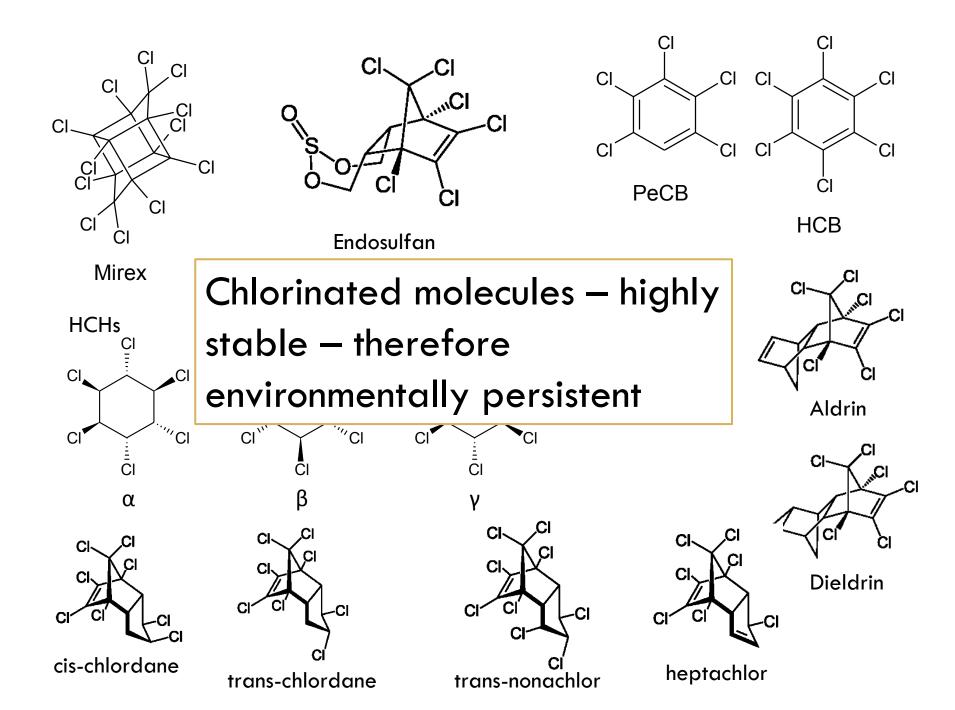
Source: USDA, Economic Research Service using USDA, National Agricultural Statistics Service and proprietary data.

Note: H = herbicide, I = insecticide, F = fungicide, and O = other.

Source: USDA, Economic Research Service using USDA, National Agricultural Statistics Service and proprietary data.

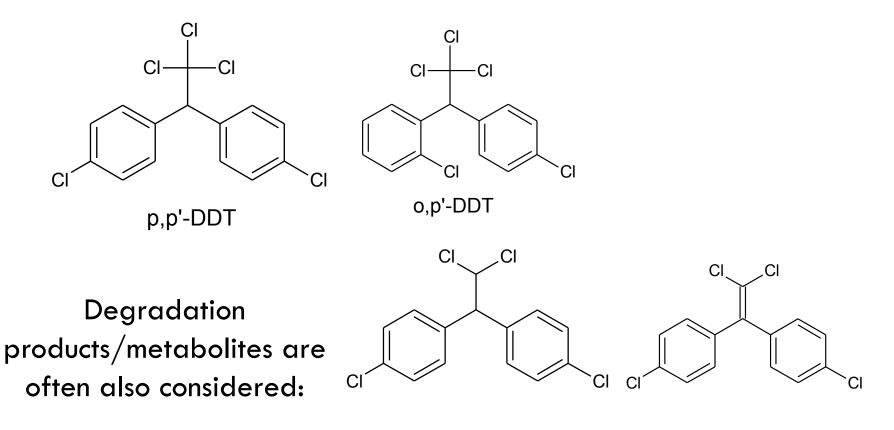
Organochlorine pesticides

- OCPs = organochlorine pesticides
- What are the OCPs?
 - DDT
 - Hexachlorobenzene (HCB)
 - Pentachlorobenzene (PeCB)
 - Hexachlorocyclohexanes (multiple isomers)
 - Heptachlor/heptachlor epoxide
 - Aldrin/dieldrin/endrin
 - Chlordane (multiple isomers)
 - Endosulfan
 - Mirex
 - ••••



Case study 2: DDT

DDT – dichlorodiphenyl trichloroethane



p,p'-DDD

p,p'-DDE

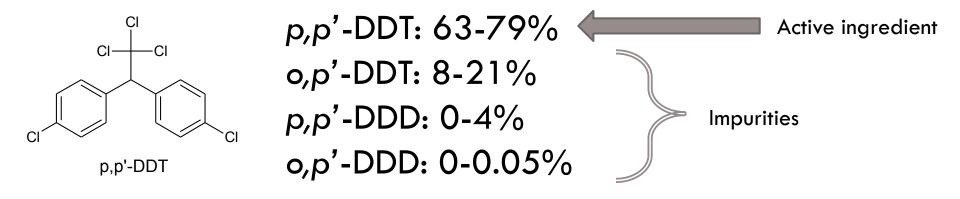
ortho

meta

para

Chemical DDT vs. Technical DDT

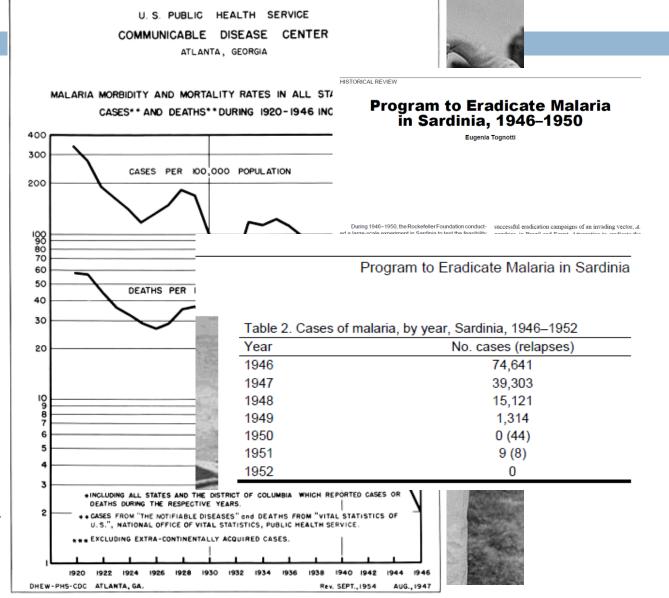
- Chemical DDT dichlorodiphenyltrichloroethane, generally p,p'-DDT – the isomer with insecticidal properties
- Technical DDT mixture of p,p'-DDT, o,p'-DDT, DDE and DDD
- DDE and DDD are impurities in technical mixture and breakdown products of DDT



DDT – a bri<u>ef history</u>

1872 – DDT was first
synthesized by Austrian
chemistry student
1939 – insecticidal
properties discovered
WW2– global use of DDT
against typhus, malaria
1945 – DDT available to
public

1940s, 1950s – WHO and country-specific programs targeting elimination of malaria – successful in Europe and North America, and large reduction in cases in India, southeast Asia



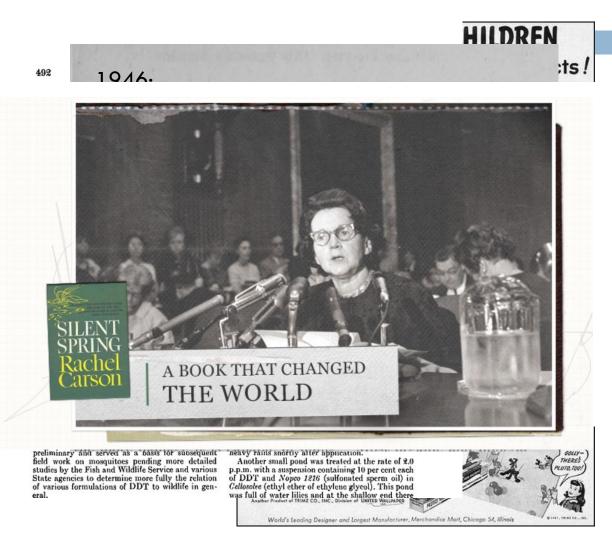
DDT – a brief history

v

1959 - More than 36
million kg of DDT was
sprayed over the US
1961 - DDT use reaches
its peak.

1940s, 1950s – Gradual increase in number of scientific studies identifying negative effects of DDT on wildlife

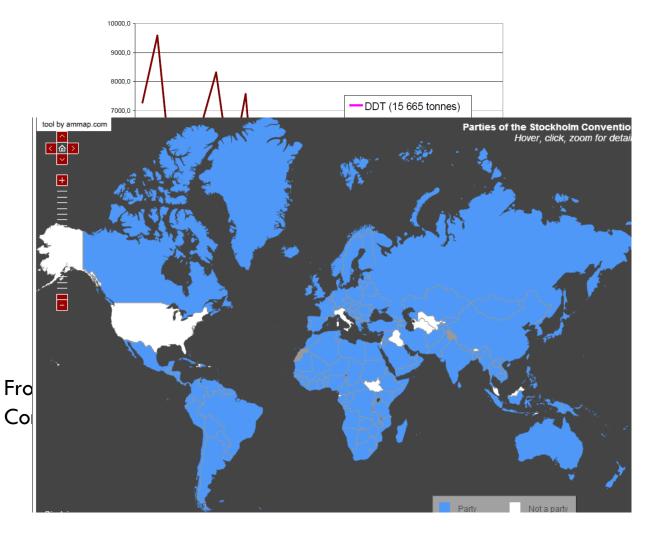
1962 - Rachel Carson's book Silent Spring blamed environmental destruction on DDT.



DDT – a brief history

Figure 2: The use of selected POPs pesticides in the former Czechoslovakia (values after the name indicate the production figures during the production period)

1972 – DDT ban in USA and Canada 1974 – DDT ban in Czechoslovakia 1970s, 1980s –ban on DDT in many countries 2001 – Stockholm Convention on POPs – DDT is banned with limited exceptions for malaria control Currently



Where is DDT still used?

- Legally for malaria control:
- Botswana, Eritrea, Ethiopia, India, Madagascar, Marshall Islands, Mauritius, Morocco, Mozambique, Namibia, Senegal, South Africa, Swaziland, Uganda, Venezuela, Yemen, Zambia
- Illegal use continues in limited locations?

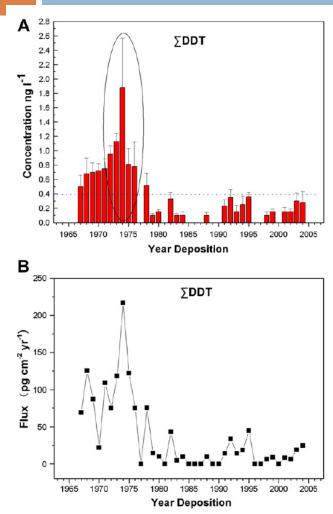
DDT – What are the concerns?

Persistence, toxicity, long-range transport and bioaccumulation/biomagnification!

What are typical trends in DDTs?

SumDDT compounds in ice core from

Mt. Everest glacier



DDT compounds in precipitation

from North America, 1995-2005

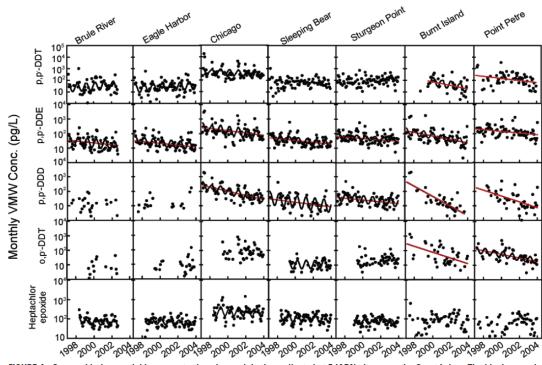


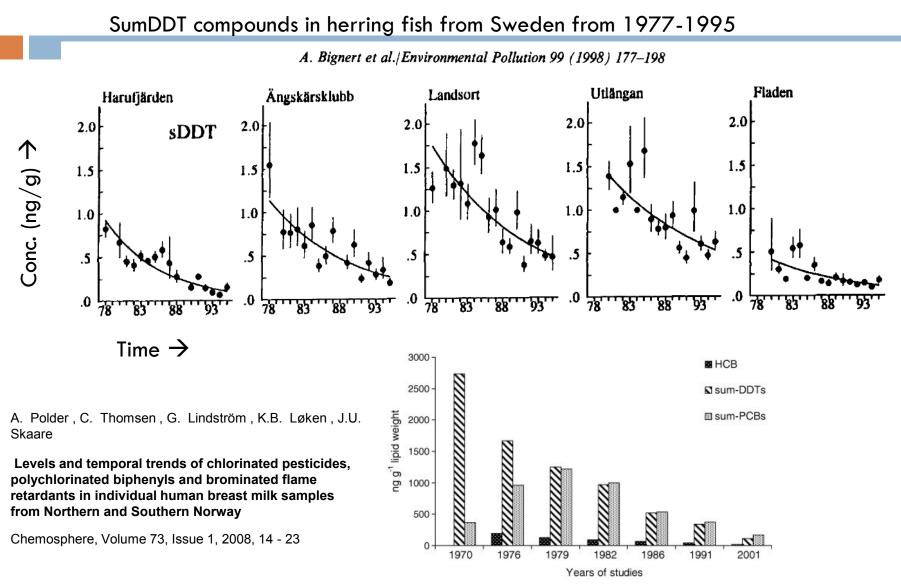
FIGURE 4. Organochlorine pesticide concentrations in precipitation collected at 7 IADN sites near the Great Lakes. The black curve is the fitted line of the sinusoidal model with the period length (a₃) set to one year. The red lines indicate long-term significant decreasing or increasing trends. Detailed information on the fitted parameters is in the Supporting Information.

(Sun et al., Environmental Science and Technology, 2006)

Fig. 4. Concentration (A) and deposition flux (B) of DDT in the ice core from East Rongbuk glacier (Mt.Everest, The Himalayas).

(Wang et al., Atmospheric Environment, 2008)

What are typical trends in DDTs?



Time trend of levels of HCB, sum-DDTs and sum-PCBs in breast milk

DDT – remaining questions?



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Should DDT Be Used to C Malaria?

DDT should be used "with caution" in combating malaria, a pan May 4, 2009 | By Marla Cone and Environmental Health News

A panel of scientists recommended today that the spraying of DDT in malaria-plagued Africa and Asia should be greatly reduced because people are exposed in their homes to high levels that may cause serious health effects.

The scientists from the United States and South Africa said the insecticide, banned decades ago in most of the world, should only be used as a last resort in combating malaria.



If Malaria's the Problem, DDT's Not the Only Answer

By May Berenbaum Sunday, June 5, 2005

In the pantheon of poisons, DDT occupies a special place. It's the only pesticide celebrated with a Nobel Prize: Swiss chemist Paul Mueller won in 1948 for having discovered its insecticidal properties. But it's also the only pesticide condemned in pop song lyrics -- Joni Mitchell's famous "Hey, farmer, farmer put away your DDT now" -- for damaging the environment. Banned in the United States more than 30 years ago, it remains America's best known toxic substance. Like some sort of rap star, it's known just by its initials; it's the Notorious B.I.G. of pesticides.

Now DDT is making headlines again. Many African governments are calling for access to the pesticide, believing that it's their best hope against malaria, a disease that infects more than 300 million people worldwide a year and kills at least 3 million, a large proportion of them children. And this has raised a controversy of Solomonic dimensions, pitting environmentalists against advocates of DDT use.



To spray or not to spray: Many African nations believe DDT is their only hope against malaria, but the powerful pesticide is not a magic bullet, the author argues. Many mosquito species have become resistant to the poison. Above, in 2001, an Ethopian girl afflicted by the disease. (By Peranders Pettersson -- Getty Images)

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Replacements for OCPs

- Current pesticide use is 2.5 million tonnes per year (Alavanja, 2009)
- OCPs are generally no longer used:
 - ~5000 tonnes DDT (produced in China, India and North Korea)¹ – 0.2% of global use
 - Only 6 countries reporting use of other OCPs (Ecuador, Honduras, Iran, Lesotho, Madagascar, Tajikistan, Ukraine) - ~2300 tonnes total in 2011²
- Replacement pesticides should have lower persistence and bioaccumulative potential

Currently used pesticides

□ Glyphosate ("Round-up")

Herbicide

- In use since 1970s
- Most widely use chemical pesticide in world
 - ~650000 tonnes per year (>30% of world pesticide market)
- Atrazine
 - Herbicide
 - banned in EU but high use in many other countries
 - 70000 tonnes per year
- Chlopyrifos
 - Most widely used insecticide
 - 170000 tonnes per year (~7% of world pesticide market)

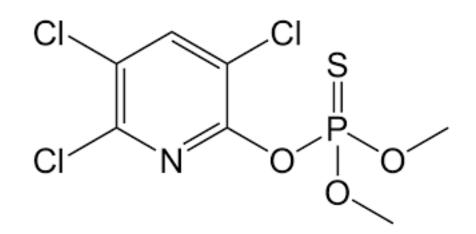
Comparing 2 insecticides: Chlorpyrifos vs. DDT

DDT

- Vapour pressure:0.0003 Pa
- □ Solubility: 0.025 mg/L
- □ Half-life in soil: 2-15 yrs
- Overall environmental half-life: 1-5 yrs
- Characteristic travel distance: 255 km



- Vapour pressure:0.001 Pa
- □ Solubility: 2 mg/L
- Half-life in soil: 60-120 days
- Overall environmental half-life: 30 days
- Characteristic travel distance: 62 km



Data from Pesticide Information Profiles, Extoxnet, Cornell University; and Mackay et al. 2014

Comparing 2 insecticides: Chlorpyrifos vs. DDT

DDT

ASTDR oral reference dose: 0.0005 mg/kg/day

□ LD50:

 \sim 100-1000 mg/kg

- Not acutely toxic to birds, but acutely toxic to aquatic organisms
- Evidence for reproductive, mutagenic and teratogenic effects in humans

Chlorpyrifos

US EPA oral reference dose: 0.003 mg/kg/day

□ LD50:

 \sim 80-300 mg/kg

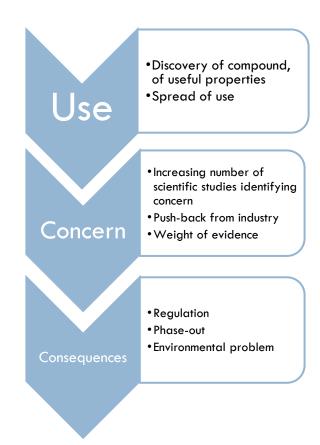
- Toxic to birds (LD50 8-100 mg/kg)
- Not identified as reproductive toxic, or teratogenic, mutagenic

Any questions about pesticides?

Examples of SVOCs

- Pesticides
- Industrial chemicals
- By-products
- Additives in consumer products

Polychlorinated biphenyls - PCBs



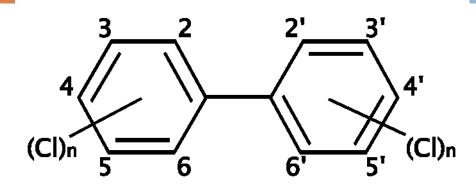
-High chemical and physical stability, even at high temperatures

 \rightarrow Desirable property!

- Industrially produced in 10 countries for a range of uses
- Can also occur as a by-product of some industrial processes, esp. cement production and pulp and paper industries
- First detected in environment in Swedish fish in 1966, many more reports followed
- Concerns about environmental persistence and bioaccumulation
- Production and new use banned by many countries in 1970s, 1980s
- Banned under Stockholm Convention

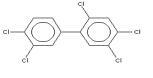
But...PCBs remain in use in old building equipment, electrical equipment, etc.

PCBs – chemical structure



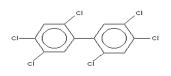
- 209 possible congeners
- 1 to 10 chlorines
- only 130 were used commercially
- Classified based on degree of chlorination

Indicator PCBs – 7 congeners:



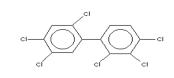
PCB

118



PCB

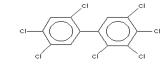
153



PCB

138

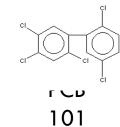
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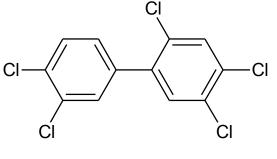


PCB

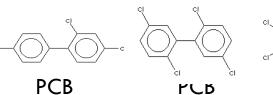
180

52





2,3',4,4',5-Pentachlorobiphenyl **PCB 118**



PCBs – health effects

- □ Acute vs. chronic effects
- Associated with cancer, liver function, skin effects at occupational exposure levels
- Prenatal exposure slows development in children
- Some evidence of link with breast cancer
- Dioxin-like PCBs

What were PCBs used for?

Transformers and capacitors

- Other electrical equipment including voltage regulators, switches, reclosers, bushings, and electromagnets
- Oil used in motors and hydraulic systems
- Old electrical devices or appliances containing PCB capacitors
- Fluorescent light ballasts
- Cable insulation
- **Thermal insulation material including fiberglass, felt, foam, and cork**
- Adhesives and tapes
- Oil-based paint

Caulking

- Plastics
- Carbonless copy paper
- **Floor finish**





PCB production

K. Breivik et al. / The Science of the Total Environment 290 (2002) 181-198

183

Producer	Country	Start	Stop	Amount	Reference				
Monsanto	USA	1930	1977	641 246	de Voogt and Brinkman (1989)				
Geneva Ind.	USA	1971	1973	454	de Voogt and Brinkman (1989				
Kanegafuchi	Japan	1954	1972	56 326	Tatsukawa (1976)				
Mitsubishi	Japan	1969	1972	2461	Tatsukawa (1976)				
Bayer AG	West Germany	1930	1983	159 062	de Voogt and Brinkman (1989)				
Prodelec	France	1930	1984	134 654	de Voogt and Brinkman (1989)				
S.A. Cros	Spain	1955	1984	29 012	de Voogt and Brinkman (1989)				
Monsanto U.K.			1977	66 542	de Voogt and Brinkman (1989)				
Caffaro	Caffaro Italy		1983	31 092	de Voogt and Brinkman (1989)				
Chemko	Czechoslovakia	1959	1984	21 482	Schlosserová (1994)				
		139	1990	141 800	AMAP (2000)				
80 J	(a) Total PCB	072	1993	32 000	AMAP (2000)				
60 -		960	1979	8000	Jiang et al. (1997)				
00	/ h)30	1993	1 324 131					
40 -									
20	\int								
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0				illion tonr	nes globally				

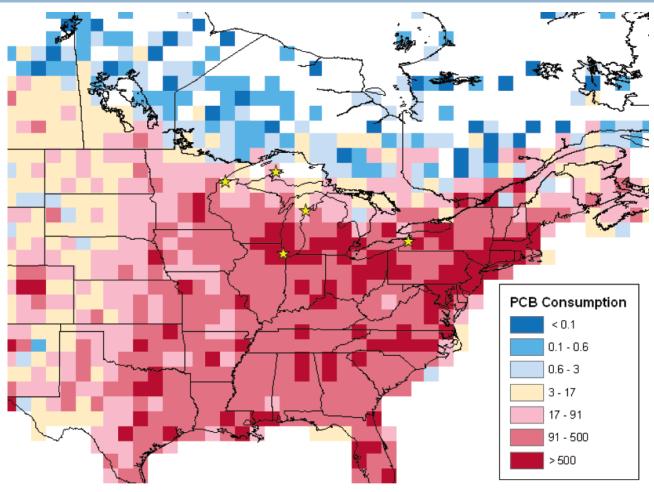
Table 1Total PCB production in t as reported in the literature

1950 1960 1970 1980 1990 2000

1930

1940

PCB use in North America – in tonnes



Hafner and Hites, ES&T, 2003

Why are PCBs still in use?

- □ Because they are so useful for their purpose!
- □ Where they were used was not well-documented
- Challenges with removing all PCBs from use current legislation only requires PCBs to be removed at >50 ppm

CN Tower, Toronto, Canada





Transformer is located in viewing area, 342 m high







Had to be cut apart by hand



Packed piece-by-piece into steel drums, removed by elevator

Any questions about PCBs?

Examples of SVOCs

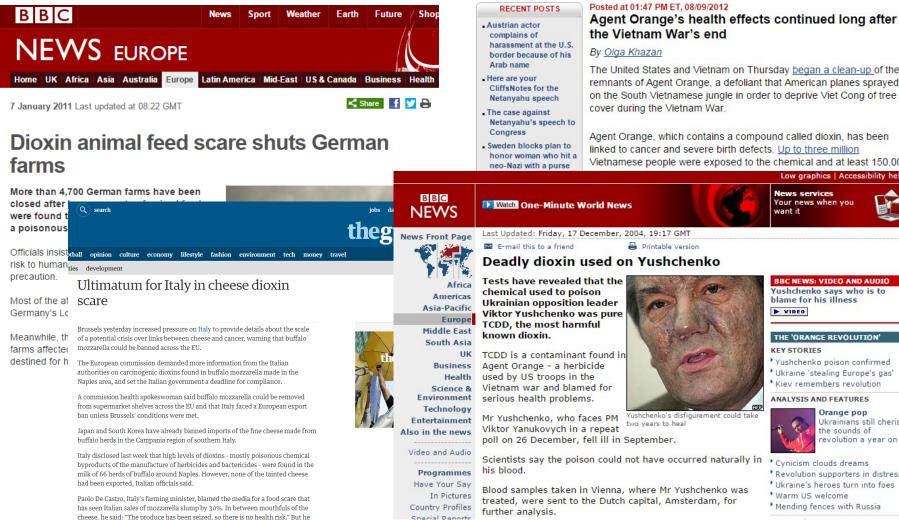
- Pesticides
- Industrial chemicals
- **By-products**
- Additives in consumer products

By-products

- By-products of industrial processes, combustion
- Unintentionally produced during industrial processes, fossil fuel combustion for heating, transportation, etc.
- Examples:
 - Polycyclic aromatic hydrocarbons
 - Polychlorinated dibenzo-p-dioxins and polychlorinated dibenzofurans

Dioxins and furans

Dioxins in the news...



The Washington Dost PostTV Politics Opinions Local Sports National World Busine

The United States and Vietnam on Thursday began a clean-up of the remnants of Agent Orange, a defoliant that American planes sprayed on the South Vietnamese jungle in order to deprive Viet Cong of tree

Agent Orange, which contains a compound called dioxin, has been linked to cancer and severe birth defects. Up to three million Vietnamese people were exposed to the chemical and at least 150,000



want it



THE 'ORANGE REVOLUTION'

KEY STORIES

Yushchenko poison confirmed * Ukraine 'stealing Europe's gas' Kiev remembers revolution

ANALYSIS AND FEATURES



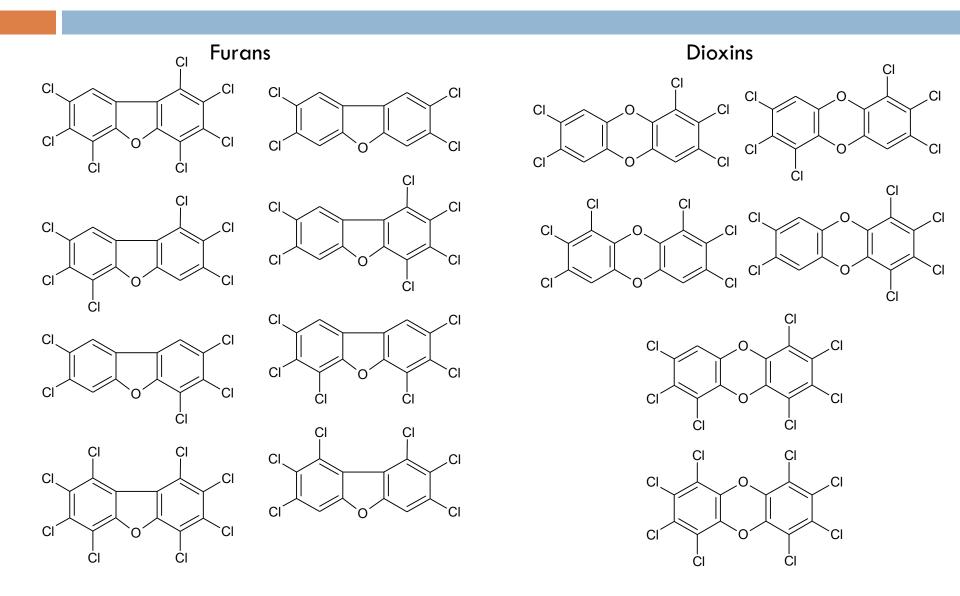
Orange pop Ukrainians still cherish the sounds of revolution a year on

Cvnicism clouds dreams

- Revolution supporters in distress
- * Ukraine's heroes turn into foes
- Warm US welcome
- Mending fences with Russia

Special Deporte

Dioxins and furans – chemical structures



Sources of PCDD/Fs

- Unintentionally produced
- During inefficient/incomplete combustion, especially waste burning
- By-product from chemicals manufacturing
- Major sources are: waste incineration, automobile emissions, metal industries, burning of peat, coal, wood

PCDD/F Source Inventory

Table 6 PCDD/PCDF release inventories for Asian countries (1) (DEH, 2004; UNEP, 2004). Releases in g TEQ/a

Cat.	Source categories	Australia – 2002				Cambodia - 2004				Sri Lanka – 2002						
		Air	Water	Land	Product	Residue	Air	Water	Land	Product	Residue	Air	Water	Land	Product	Residue
1	Waste incineration	6.5	0.36	21.9	ND	ND	40.7	0	0	0	0.78	20.3	0.055	NA	NA	0.133
2	Ferrous and non-ferrous metal production	112	0.0	44.4	ND	ND	0.41	0	0	0	1	5.52	ND	NA	NA	49.8
3	Heat and power generation	35.0	0.0	31.8	ND	ND	10.3	0	0	0	1.69	19.3	ND	ND	NA	0.096
4	Production of mineral products	1.9	0.0	0.0	ND	ND	0.099	0	0	0	0	1.37	NA	ND	ND	0.002
5	Transportation	9.1	0.0	0.0	ND	ND	0.005	0	0	0	0	0.54	NA	NA	NA	ND
6	Open burning processes	330	0.0	1030	ND	ND	218	0	14.6	0	316	121	ND	ND	NA	29.4
7	Production and use of chemicals and consumer goods	0.43	0.43	110	ND	ND						ND	ND	ND	0.446	ND
8	Miscellaneous	0.31	0.0	0.15	ND	ND	3.64	0	0	0	0	3.46	ND	ND	ND	0.074
9	Landfills and waste dumps	0.0	2.61	40.3	ND	ND						ND	0.024	ND	6	0.022
1–9	Total	495	3.42	1300	ND	ND	273	0	14.6	0	319	171.5	0.08	0.0	6.45	79.5
	Grand total	1800			607				258							

Australia, Cambodia, Sri Lanka – Main source is open burning

Secondary sources are:

Australia – metal production

Cambodia and Sri Lanka – waste incineration and heat and power generation

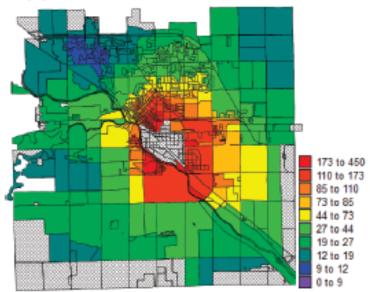
Spatial patterns of PCDD/Fs

- Higher concentrations closer to sources, in highly developed, industrialized areas
- Concentrations patterns in air, soil, sediment and biota mirror each other
- Trends on a large scale globally and small scale
 - locally

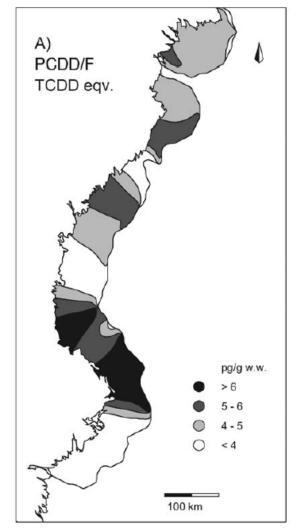
Local scale – PCDD/Fs in soil around an incinerator

Regional scale – PCDD/Fs in fish from the coast of Sweden

(c) Mean concentration

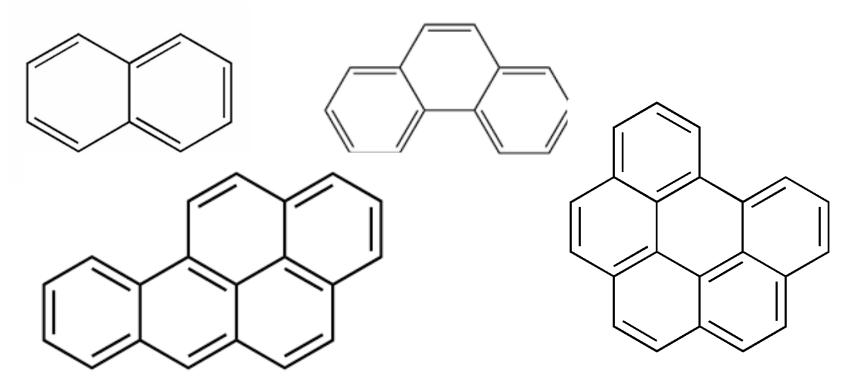


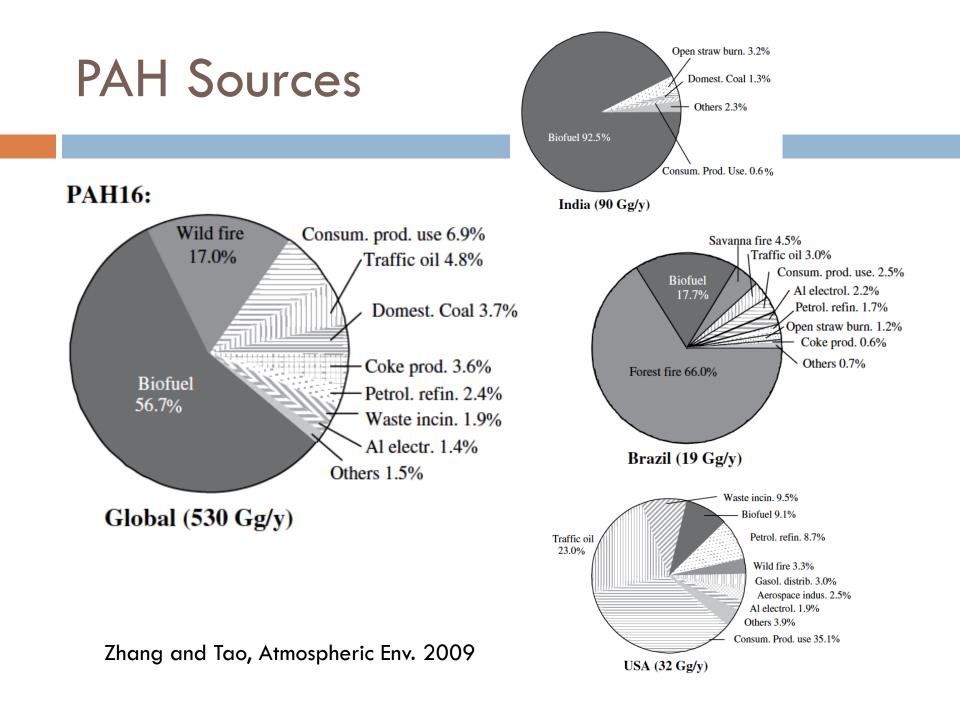
Goovaerts et al. 2008



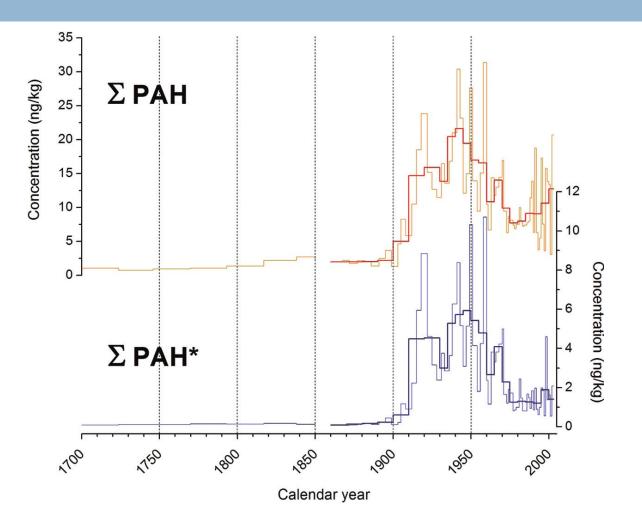
Polycyclic Aromatic Hydrocarbons (PAHs)

- By-products of combustion or fossil fuel processing
- Composed of two or more aromatic rings
- Many possible structures, but typically 3 to 6 rings





PAHs over the past 300 years



Gabrieli et al. 2010

Any questions about PCDD/Fs or PAHs?

Examples of SVOCs

- Pesticides
- Industrial chemicals
- By-products
- Additives in consumer products

Additives to consumer products

- Flame retardants
- Plasticizers

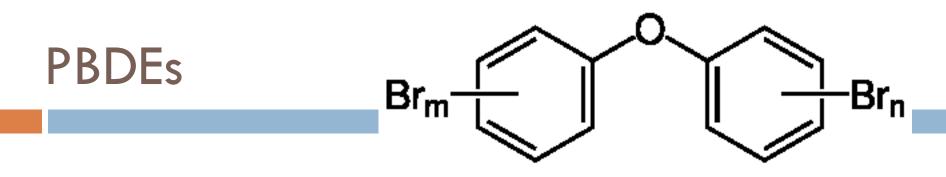
Flame retardants – organic or inorganic chemicals added to consumer products (furniture, electrical appliances, electronics) to suppress/delay/prevent the spread of fire

Plasticizers – additive chemicals that increase the flexibility, softness, fluidity of a material. Largely used in plastics.

Flame retardants

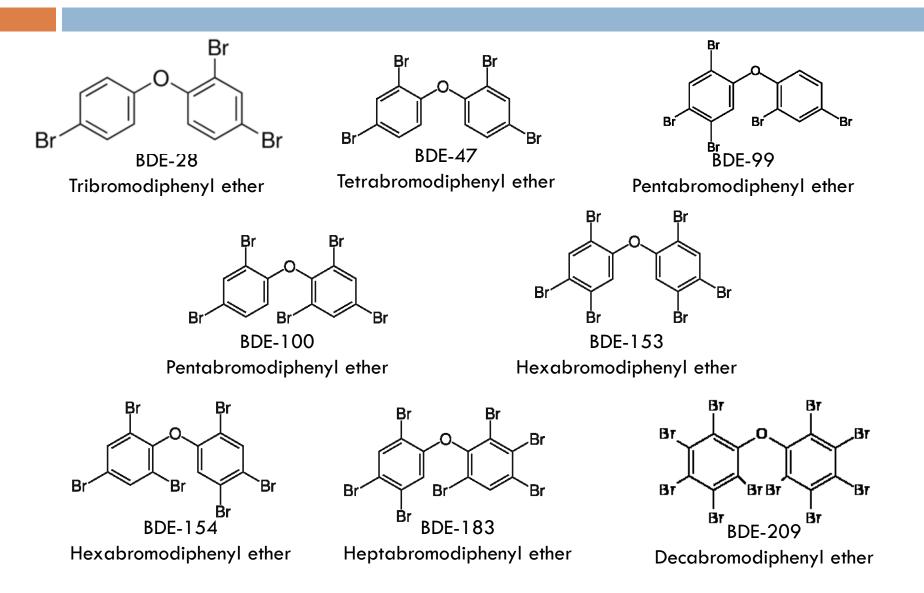


- To slow the spread of flames
- Organic or inorganic
- Wide range of applications (furniture, electronics, industrial/workplace textiles and protective equipment, vehicles)
- Required by fire safety regulations



- Polybrominated diphenyl ethers
- Flame retardants
- Classified by either technical mixture or congener group
 - Confusing!! E.g., penta-BDE can refer to either the technical mixture called "Penta" or could refer to a PBDE with 5 bromines
 - Commercial mixtures sometimes distinguished as "cpenta"

PBDE naming - congeners



Polybrominated Diphenyl Ethers: Uses

Penta

Textiles, PUF, paint, household
 plastic products, automotive parts
 banned under Stockholm Convention

Octa ABS plastic for computers, casings, circuit boards, small appliances

banned under Stockholm Convention

Deca

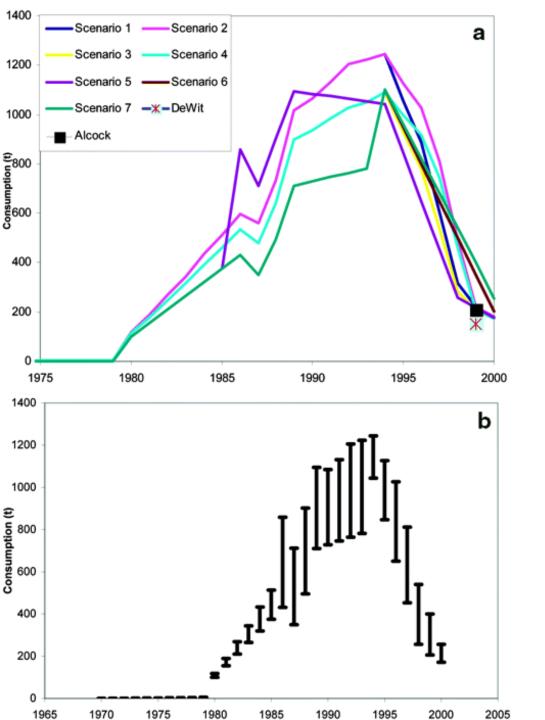
 Electrical & electronic equipment, casings for TVs, computers, textile backings (e.g., carpets)

Still in use in some areas, phased out in Europe, North America

Human health concerns for PBDEs

- Thyroid active agents
 - Neurological impairments
- Maturation
 - Delay in puberty
- Developmental neurotoxicity
 - Impaired spontaneous motor behaviour, nonhabituation behaviour
- Learning & memory
 - Worsen with age

Review: Birnbaum & Staskal 2004 EHP 112:9-17.



Estimated Historical Consumption Of Penta BDE in Europe

> Prevedouros et al. 2004 Environ Sci Technol 38:3224-3231

Estimated Consumption Of BDEs in North America

Abbasi et al. 2015 Environ Sci Technol

Figure 2. Stock of each PBDE commercial mixture in in-use products in the U.S. and Canada from 1970 to 2020, (a) pentaBDE in EEE, automotive vehicles, and PUF slabstock used in furniture, (b) octaBDE in automotive vehicles and EEE, and (c) decaBDE in plastic pallets, textiles, EEE, and automotive vehicles.

How to PBDEs get from furniture into the environment?

- Volatilization
- Abrasion, physical breakdown of the furniture
- Direct partitioning to dust

Global distributions of PCBs and PBDEs

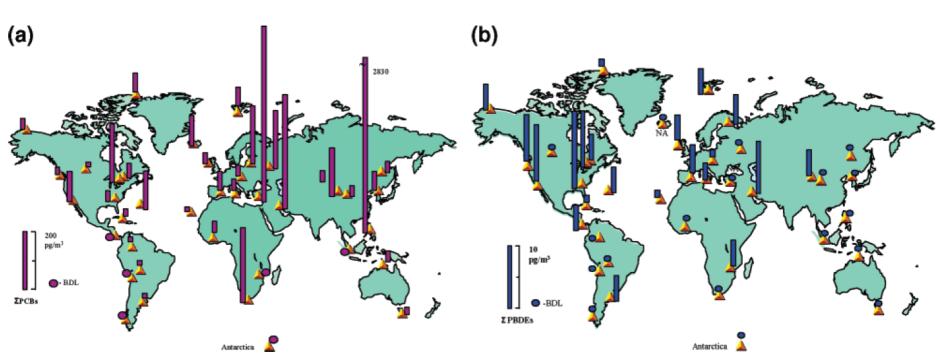


FIGURE 4. Air concentrations (pg/m³) of (a) PCBs and (b) PBDEs between December 2004 and March 2005 at GAPS sites. See Table S1 for BDL values.

From Pozo et al. 2006

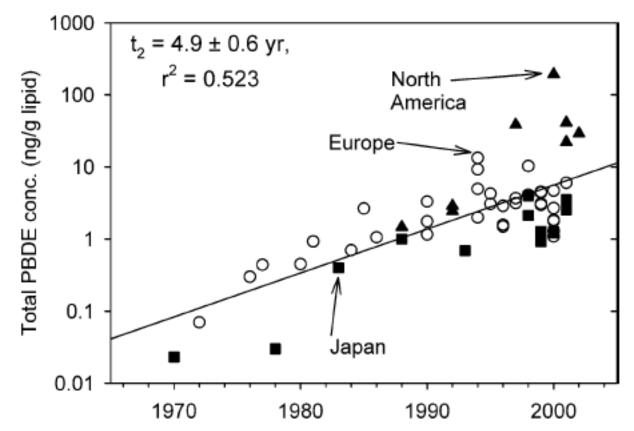
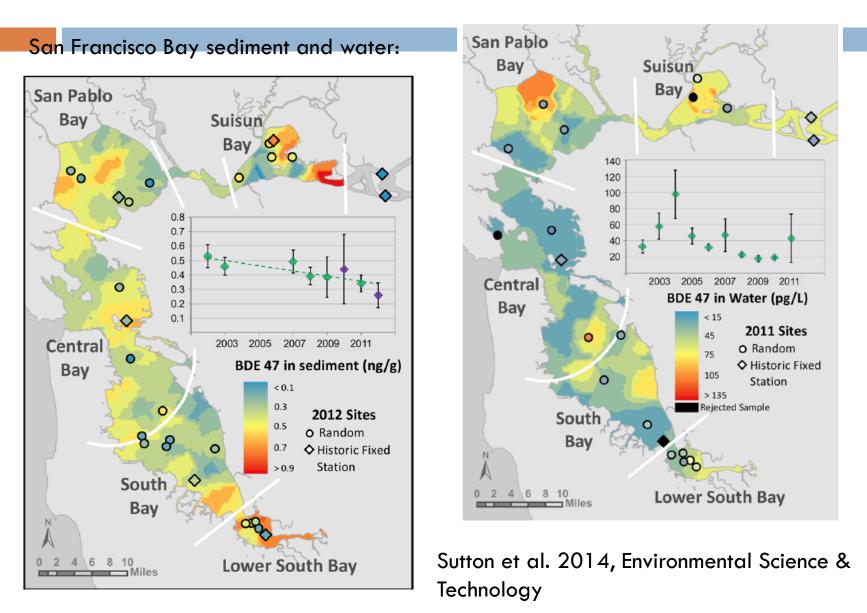
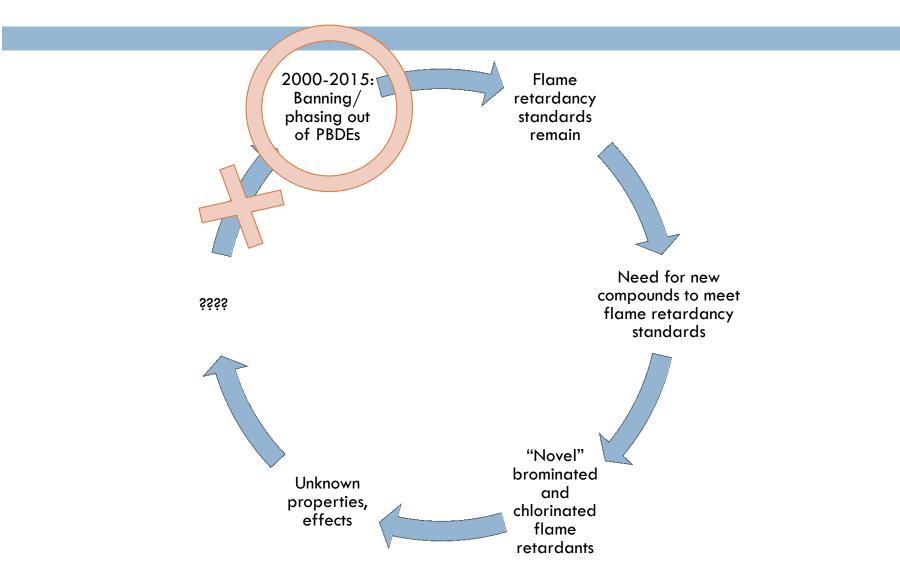


FIGURE 1. Total PBDE concentrations (\sum PBDE) in human blood, milk, and tissue (in ng/g lipid) shown as a function of the year in which the samples were taken; see Table 2. The three symbol types indicate the location from which the samples were collected. The overall regression is shown.

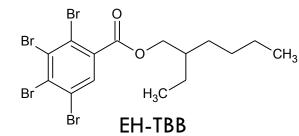
Temporal trends of PBDEs

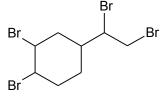


Replacement of banned compounds

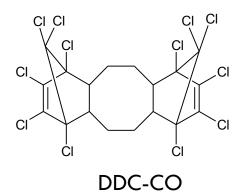


"Novel" flame retardants - NFRs – replacements for PBDEs

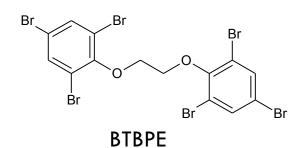








Br Br Br Br Br Br Br Br Br





PBEB



HBB



PBT

Phosphate- based flame retardants in consumer products

TABLE 1. Characteristics of the Polyurethane Foam Samples Analyzed in This Study ^a				
sample ID	source	year purchased	flame retardant detected	
1	chair	2004	unidentified	
2	mattress pad	2009	N/D	
3	leather couch	2005	unidentified	
4	sofa bed	2008	TDCPP	1.3
5	chair	2008	N/D	
6	foam from footstool	2006	TCPP	2.2
7	headrest of chair	2008	TCPP	0.5
8	chair	2006	TDCPP	3.2
9	chair	2004	TDCPP	3.0
10	chair	2007	TCPP	1.5
11	futon	N/A	pentaBDE	0.5
12	ottoman	2007	TCPP	0.7
13	chair	2003	TDCPP	1.0
14	chair	2006	TDCPP	2.9
15	pillow	2006	TDCPP	2.8
16	chair	2007	TDCPP	3.8
17	chair	2005	TDCPP	3.2
18	mattress pad	2006	TDCPP	1.2
19	couch	2007	TDCPP	5.0
20	chair	2005	TDCPP	2.5
21	office chair	2005	N/D	
22	futon	2008	TDCPP	2.8
23	nursery glider/rocker	2009	TDCPP	2.9
24	foam insulation from sieve/shaker	2008	TDCPP	2.2
25	baby stroller	2009	TDCPP	NM
26	couch	2007	TBB, TBPH	4.2

^a N/A - Not available. N/D - Not detected. NM - not measured due to low mass of foam available. TDCPP - Tris-(1,3-dichloro-2-propyl)phosphate. TCPP - Tris(1-chloro-2-propyl)phosphate. PentaBDE - Pentabromodiphenyl ether commercial mixture. TBB - ethylhexyl 2,3,4,5-tetrabromobenzoate. TBPH - bis(2-ethylhexyl) tetrabromophthalate.

Stapleton et al. Environmental Science and Technology, 2009

Chemicals to know

- PCDD/Fs
- D PAHs

What to know...

ABOUT EACH COMPOUND:

- Source/use of the compound
 - Industrial? Emission by-product?

Status

- Is the chemical still in use? Where is it legal/illegal?
- Where do we find the chemical?
 - In the environment? In humans? How are humans exposed?