

<b>HW 3</b>	<b>Inorganic Materials Chemistry</b>	<b>Name:</b>	
<b>Points:</b>	<b>C7780</b>	<b>Date due:</b>	
Max. 100 points	<b>Fall 2018</b>		

**1. (30 pts.)** Use the ligand field theory to explain why  $\text{Mn}_3\text{O}_4$  is a normal spinel while  $\text{Fe}_3\text{O}_4$  is an inverse spinel. Hint: draw diagrams of energy levels of d-electrons for ions in tetrahedral and octahedral sites, use approximation  $\Delta_T = 4/9 \Delta_O$  for ligand field splitting energy, consider all  $\text{MO}_4$  and  $\text{MO}_6$  moieties as high spin complexes, calculate ligand field stabilization energy in terms of  $\Delta_O$  for both normal and inverse arrangement of ions, compare them and find which is more stable.

**2. (30 pts)** Mixed metal oxides could be prepared by sol-gel reactions from aqueous solutions of metal salts.

**a)** Order these ions  $\text{Al}^{3+}$ ,  $\text{Ba}^{2+}$ ,  $\text{Cs}^+$ ,  $\text{H}^+$ ,  $\text{Li}^+$ ,  $\text{Mg}^{2+}$  according to the increasing value of hydration enthalpy:  $\text{M}^{z+} + n \text{H}_2\text{O} \rightarrow [\text{M}(\text{H}_2\text{O})_n]^{z+} \quad \Delta H_{\text{hydration}}$

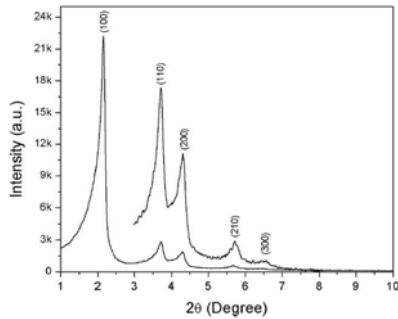
**b)** For a hydrolytic reaction  $[\text{M}(\text{H}_2\text{O})_N]^{z+} + h \text{H}_2\text{O} \rightarrow [\text{M}(\text{OH})_h(\text{H}_2\text{O})_{N-h}]^{(z-h)+} + h \text{H}_3\text{O}^+$

$$\Delta H^\circ = (75.2 - 9.6 z) \text{ kJ mol}^{-1} \quad \text{and} \quad \Delta S^\circ = (-148.4 + 73.1 z) \text{ J K}^{-1} \text{ mol}^{-1}$$

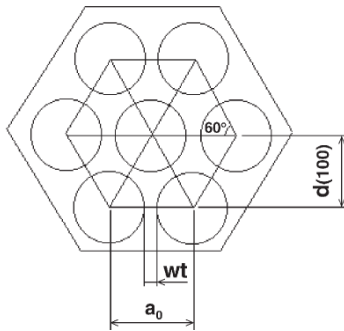
Write equation that gives a measure of spontaneity of reaction (= write a formula relating this state function to  $\Delta H^\circ$  and  $\Delta S^\circ$ ). Calculate, for which of the above listed ions is this reaction spontaneous?

**3. (40 pts.)** Calculate the wall thickness of a hexagonal MCM-41 mesoporous material, assume that it possesses cylindrical pores.

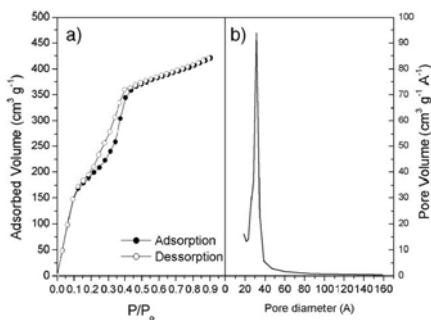
**a)** First, calculate the  $d(100)$  = interplanar distance in the (100) plane from the XRD diffractogram. CuK $\alpha$  radiation was used with  $\lambda = 1.542 \text{ \AA}$ . Diffraction maximum was found at  $2.14^\circ 2\theta$ .



**b)** Now, derive the formula relating the interplanar distance  $d(100)$  to the hexagonal mesoporous parameter  $a_0$  and calculate its value.



**c)** Derive the formula relating the diameter  $D_p$  of a pore to specific surface area  $SA$  ( $870 \text{ m}^2/\text{g}$ ) and total pore volume  $V_p$  ( $0.683 \text{ cm}^3/\text{g}$ ). Assume cylindrical pores.



**d)** Finally, calculate the wall thickness ( $wt$ ) of MCM41 material.