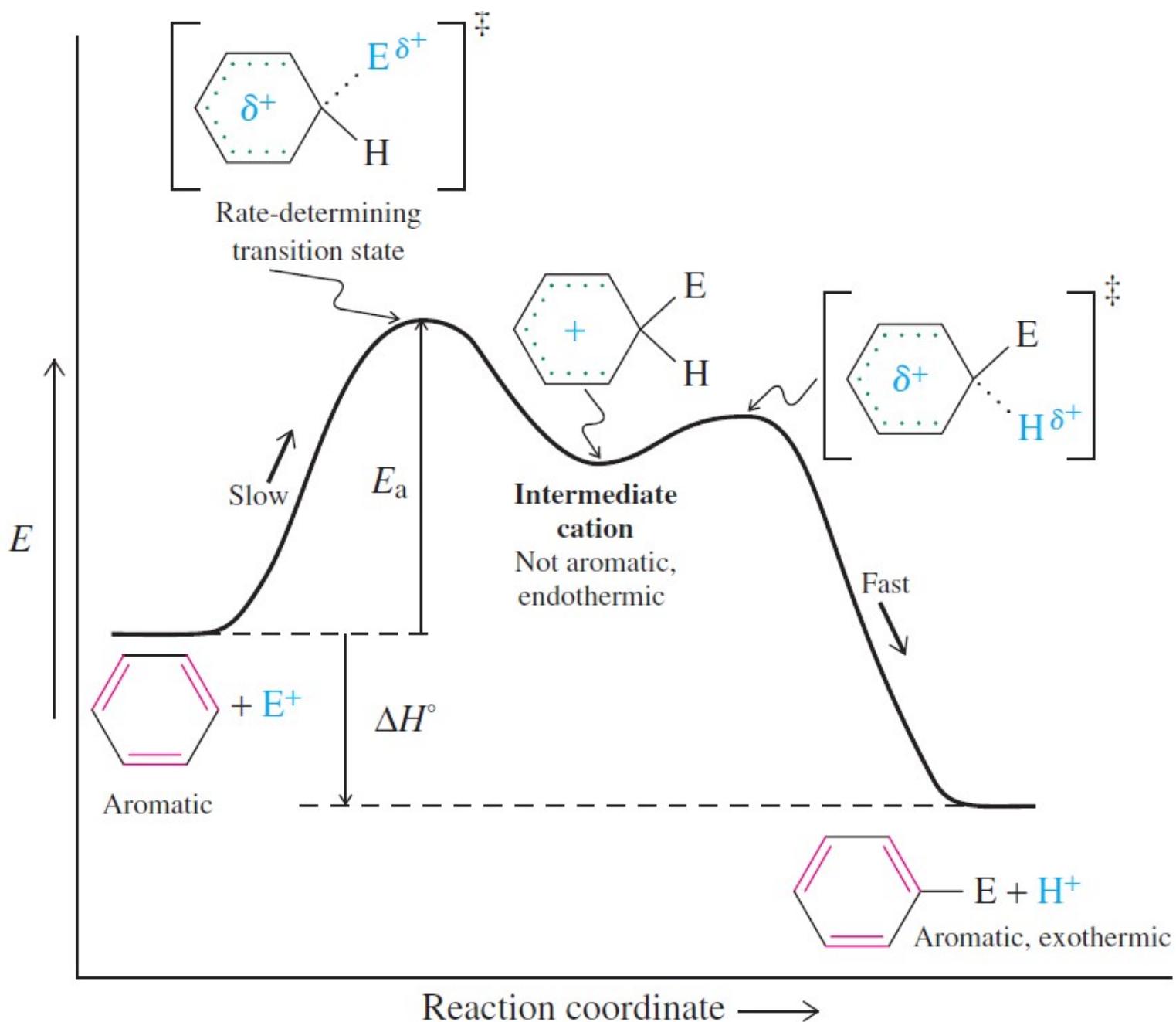
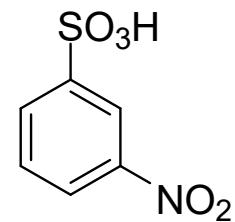
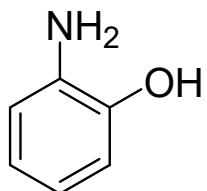
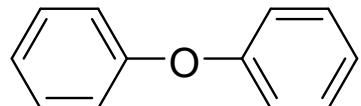
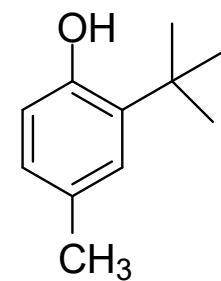
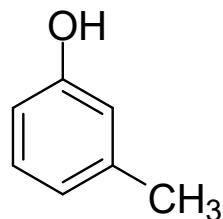
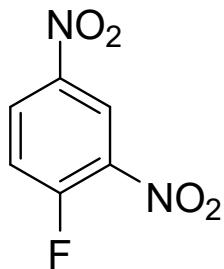
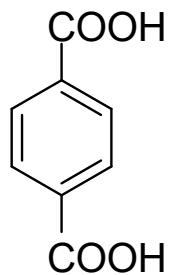


AROMATICKÉ SLOUČENINY

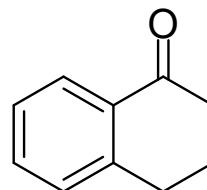
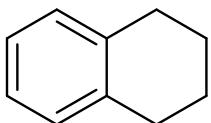
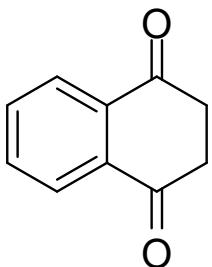
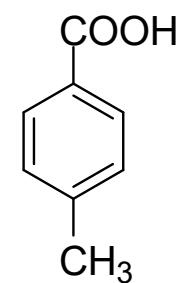
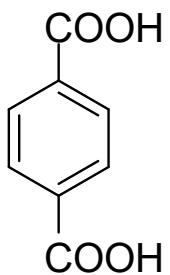
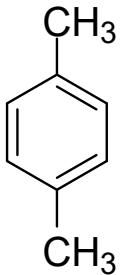
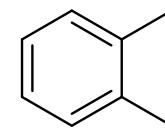
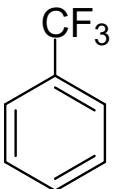
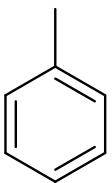
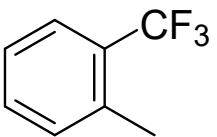




Určete, zda je benzenové jádro aktivované nebo deaktivované pro elektrofilní aromatickou substituci



Sloučeniny seřaďte podle jejich vzrůstající reaktivity při elektrofilní aromatické substituci

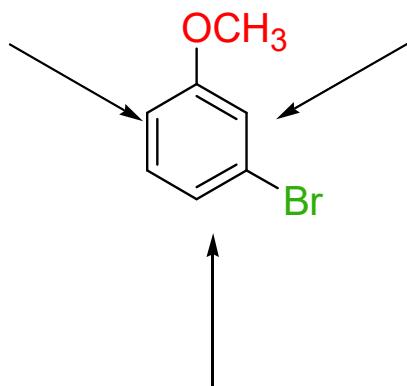
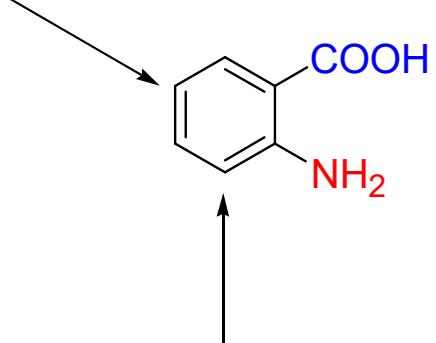
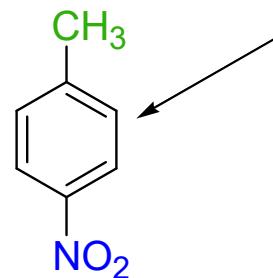
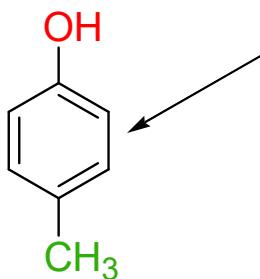


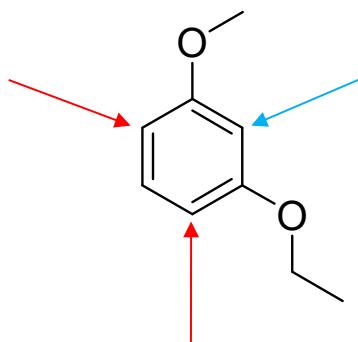
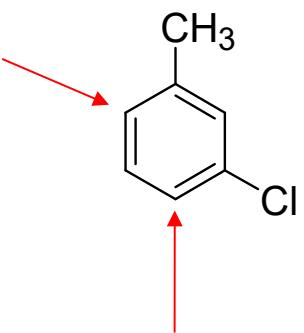
aktivující skupiny

alkyly aktivující slabě

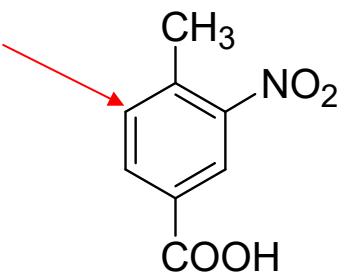
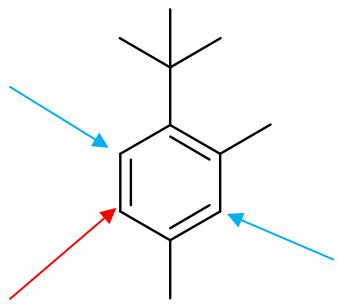
deaktivující skupiny

**halogeny o-, p- dirigující
ale deaktivující**

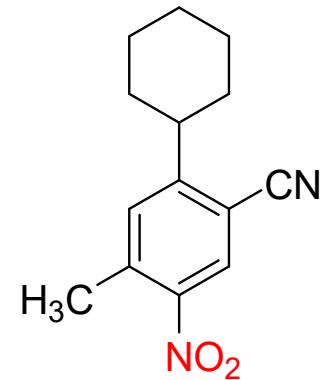
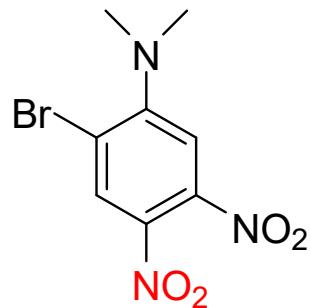
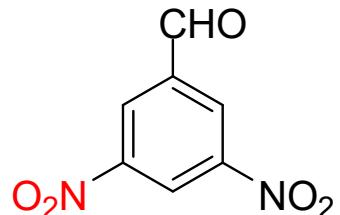
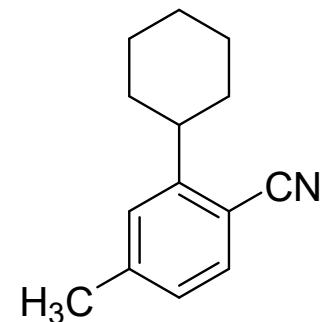
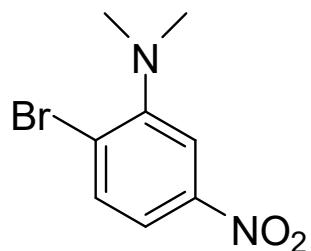
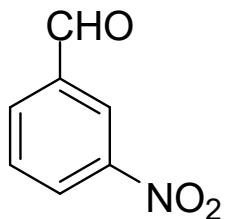




Stérická zábrana



Předpovězte produkt mononitrační reakce



a) Nitrace

b) Sulfonace

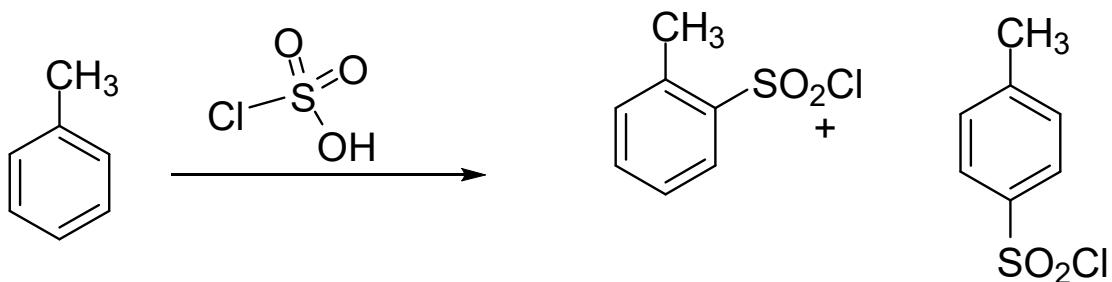
c) Halogenace

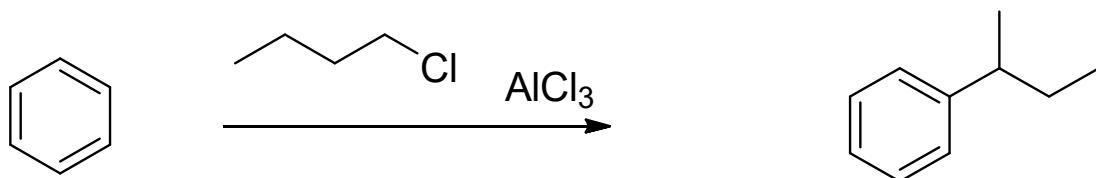
d) Friedl-Craftsova alkylace a acylace

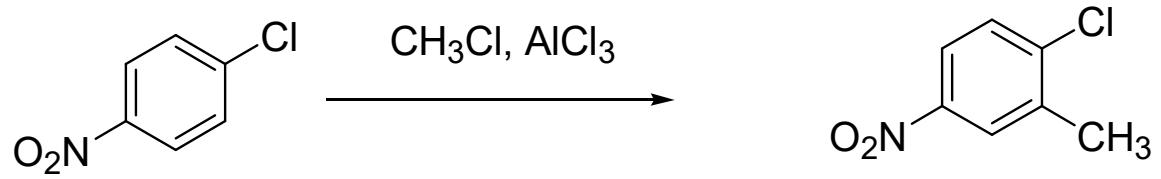
- 1) ne na aromátech se silnými akceptory (CN , NO_2)
- 2) nereagují aromatické halogenderiváty a vinylhalogenidy
- 3) problém vícenásobné substituce, protože zavedením alkylu je produkt pro další elektrofilní substituci reaktivnější
- 4) přesmyky karbokationtů
- 5) nelze provádět na substrátech s aminoskupinou, komplexace Lewisovy kyseliny na volný elektronový pár dusíku

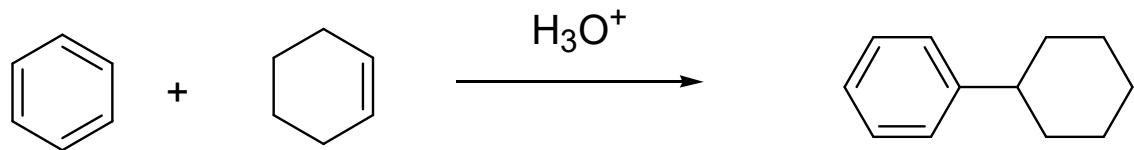
e) Kopulace

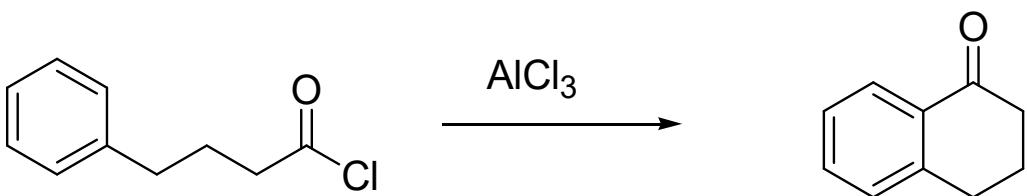


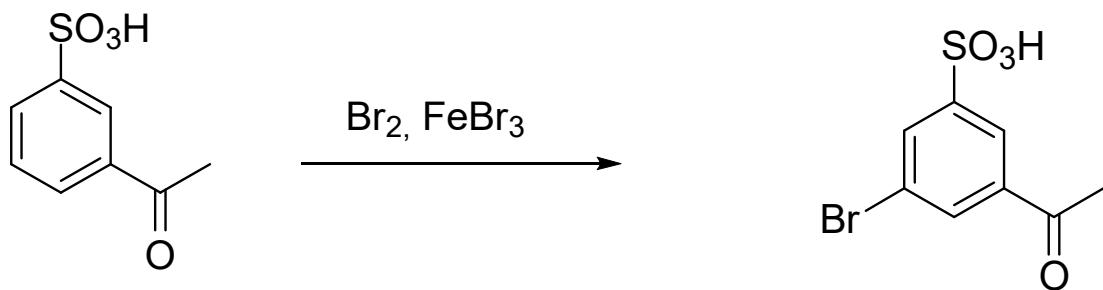


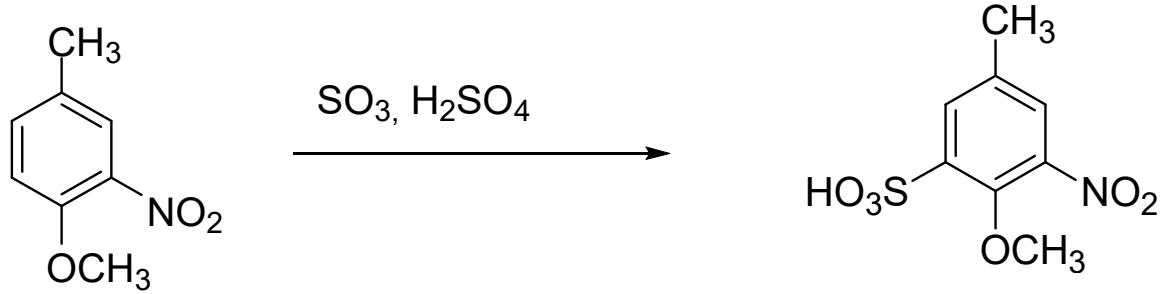


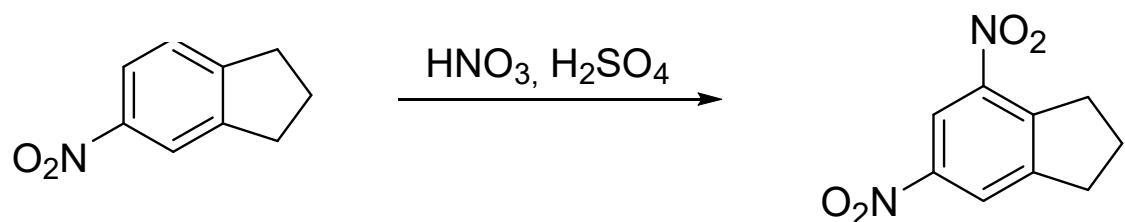




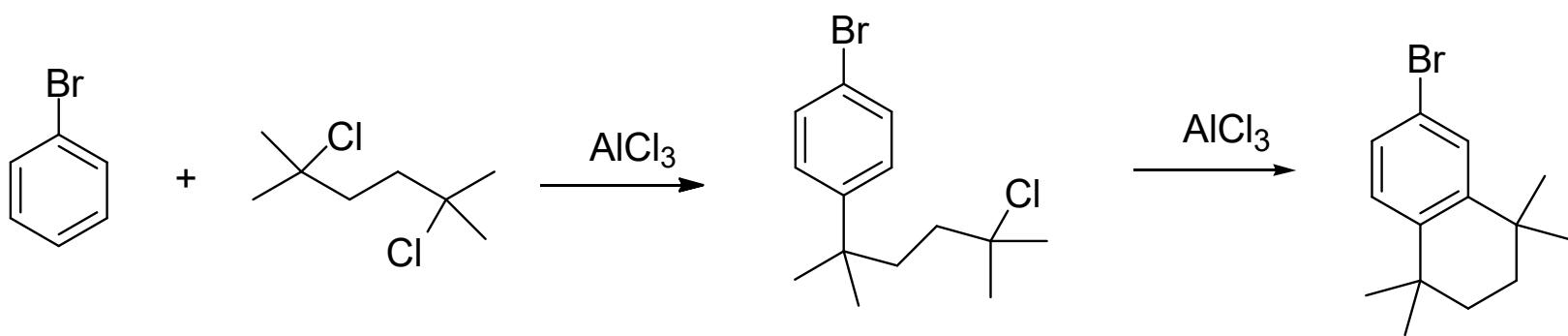


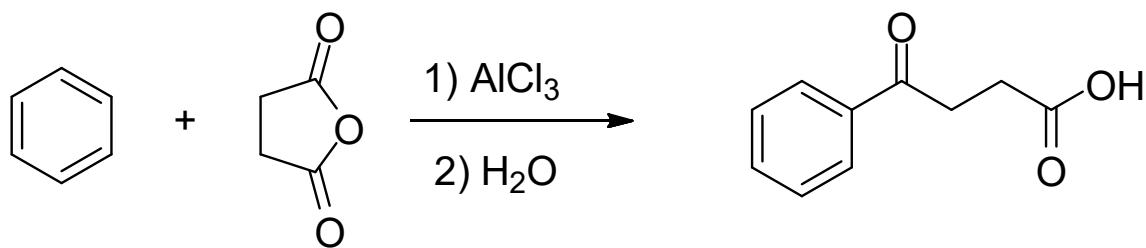


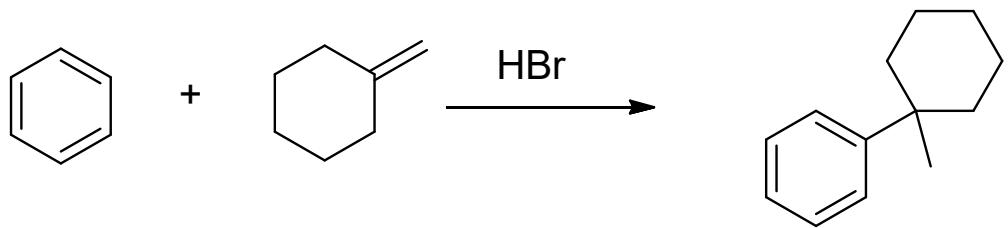


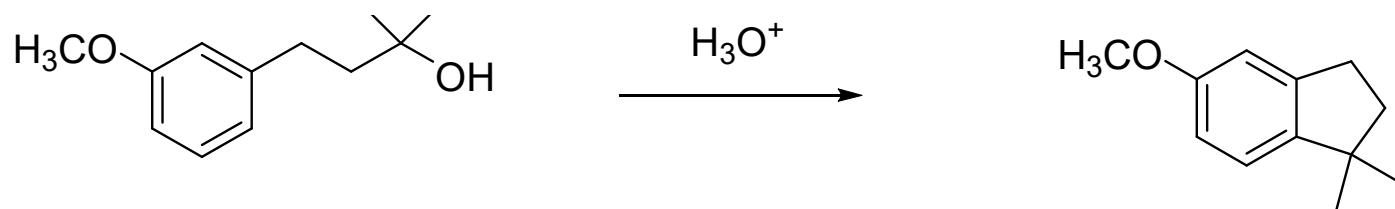


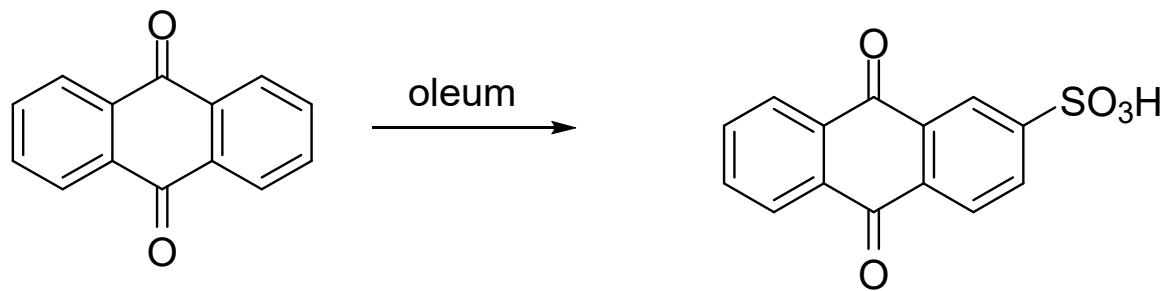


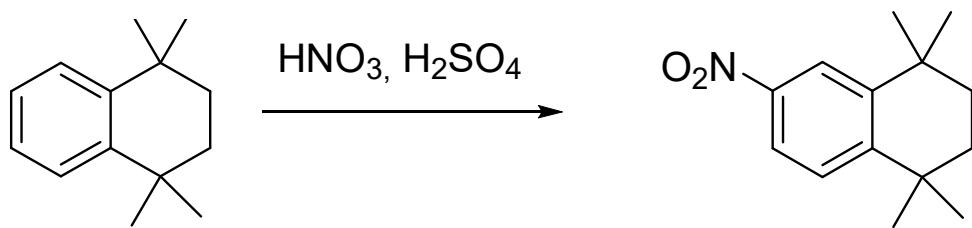


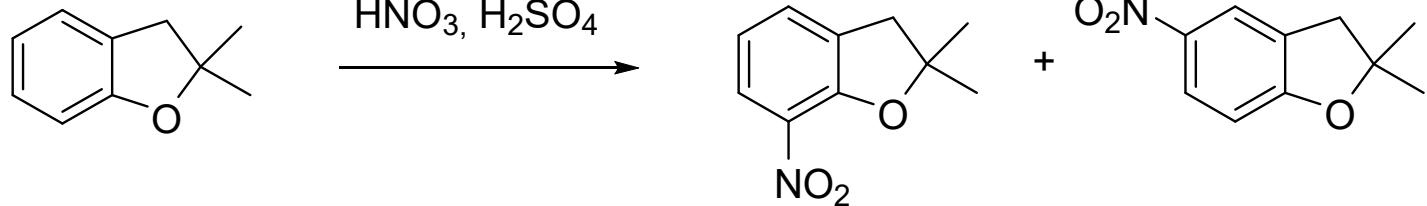












Navrhněte syntézu uvedené sloučeniny

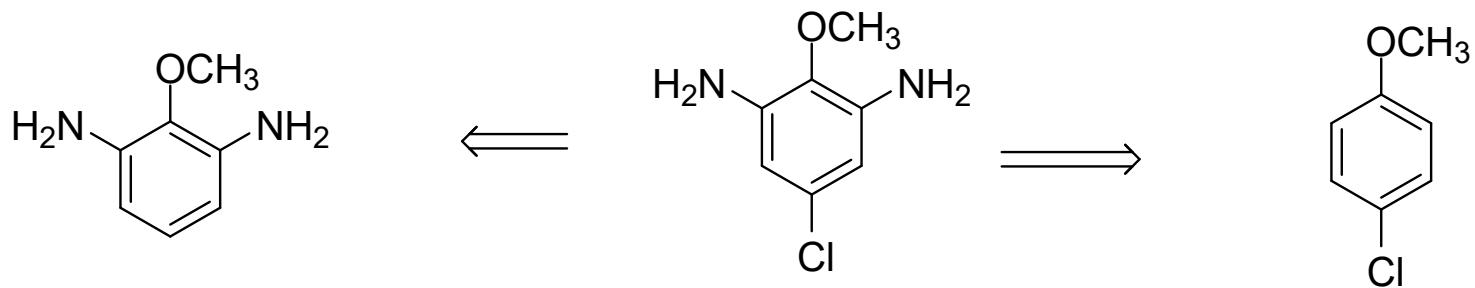
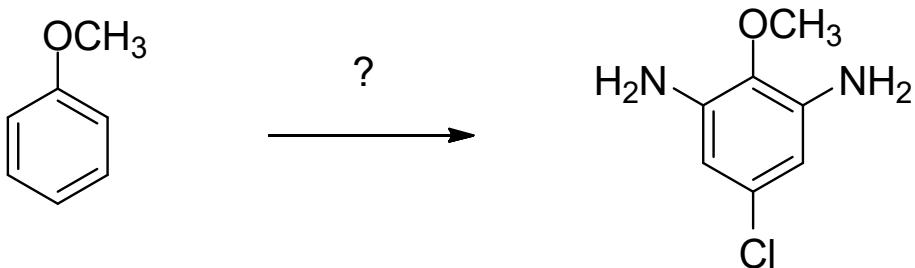


Schéma 1

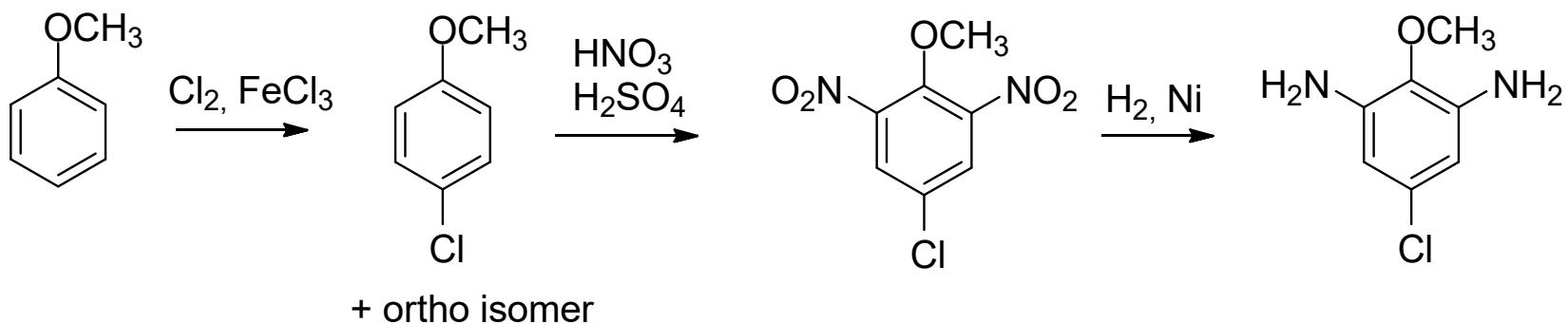
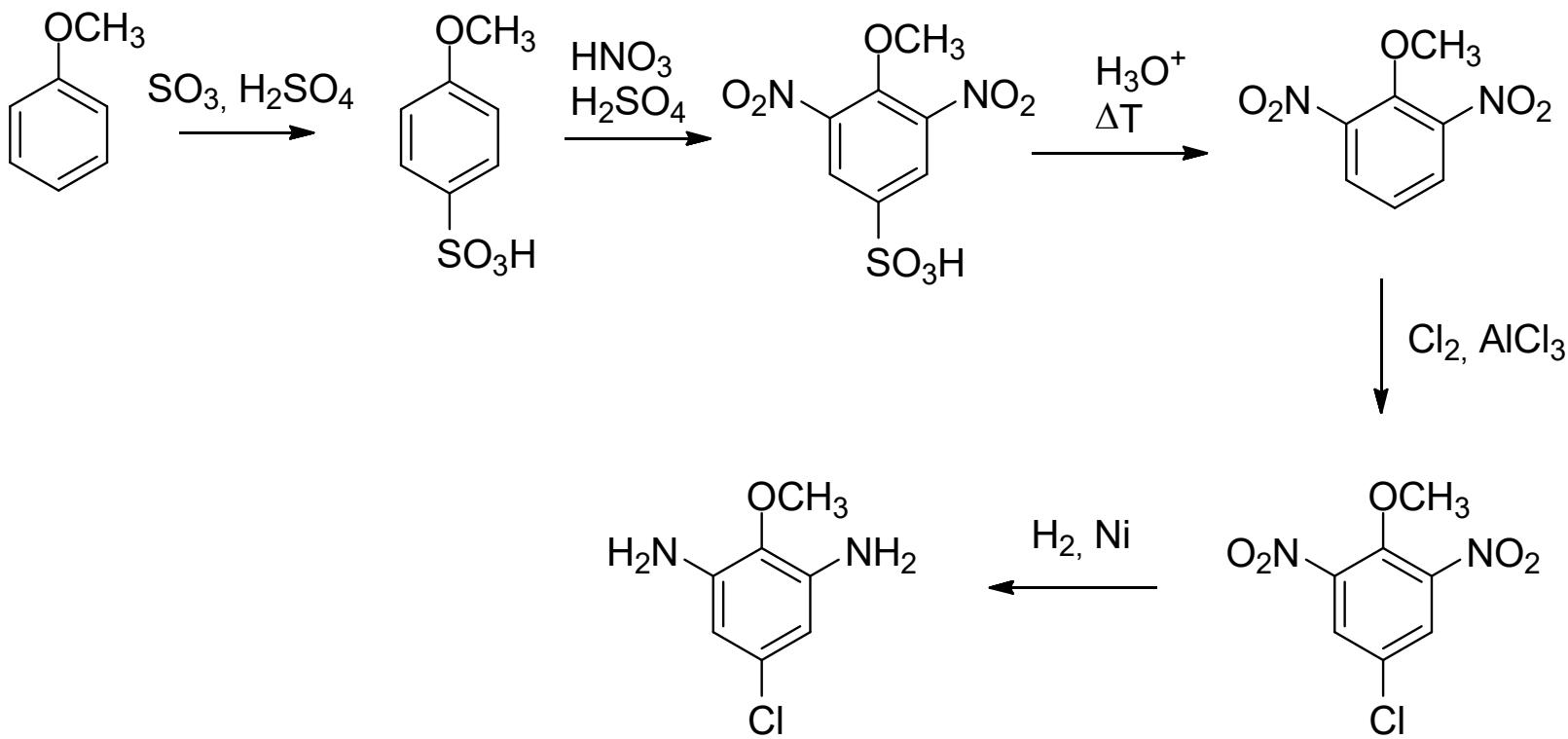
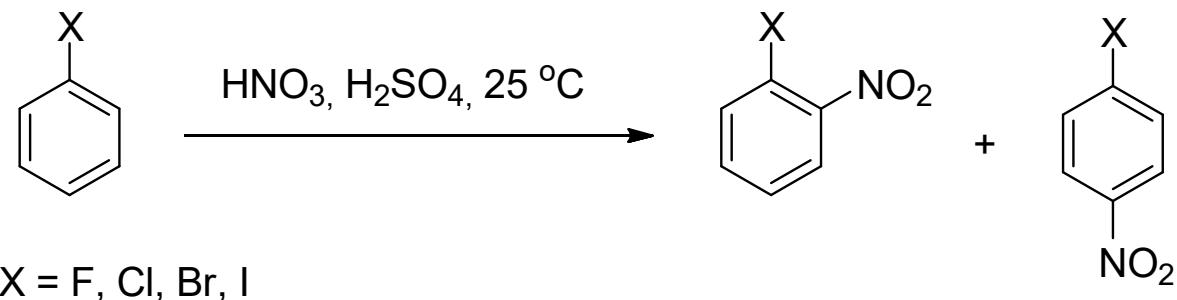


Schéma 2



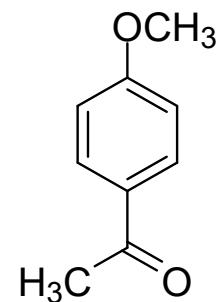
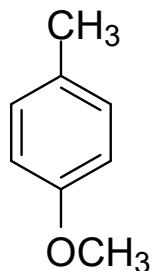
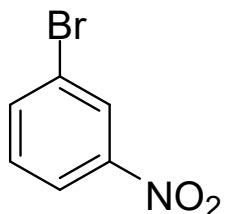
Vysvětlete uvedené pozorování



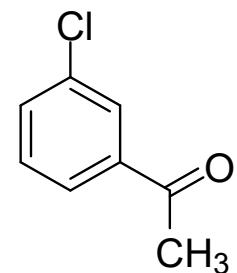
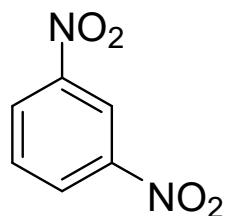
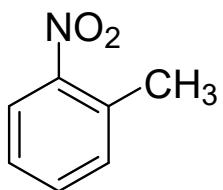
	ortho (%)	para (%)
F	13	86
Cl	35	64
Br	43	56
I	45	54



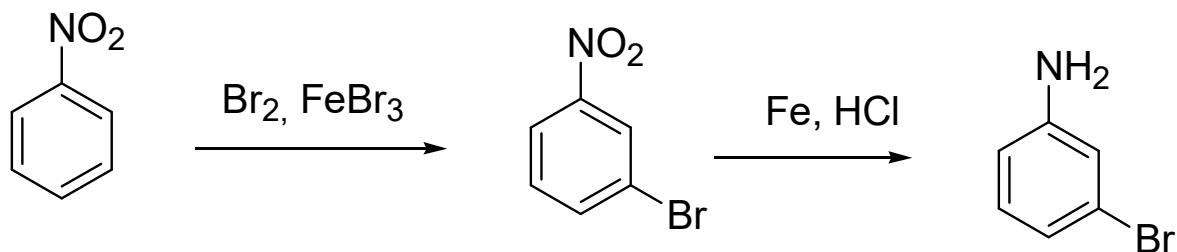
**Navrhněte jednokrokovou syntézu uvedených sloučenin
z vhodného prekurzoru**



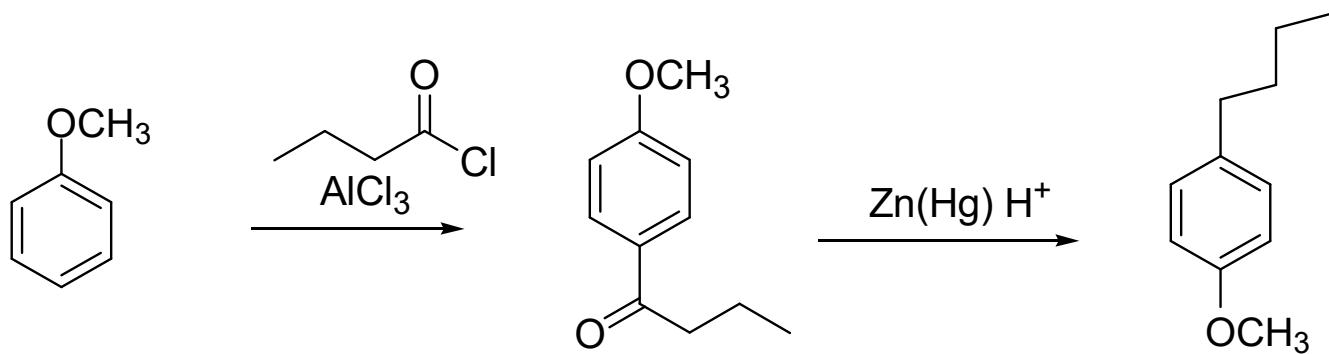
**Navrhněte jednokrokovou syntézu uvedených sloučenin
z vhodného prekurzoru**



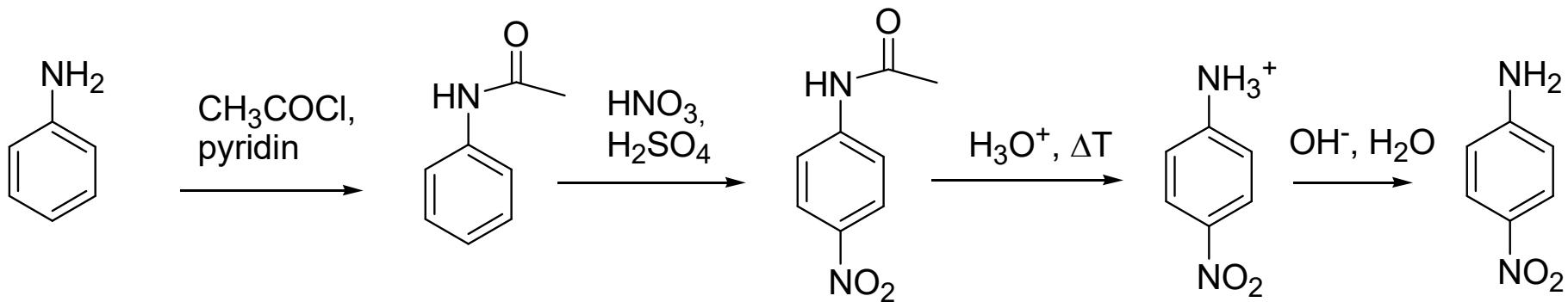
Navrhňete syntézu uvedené sloučeniny

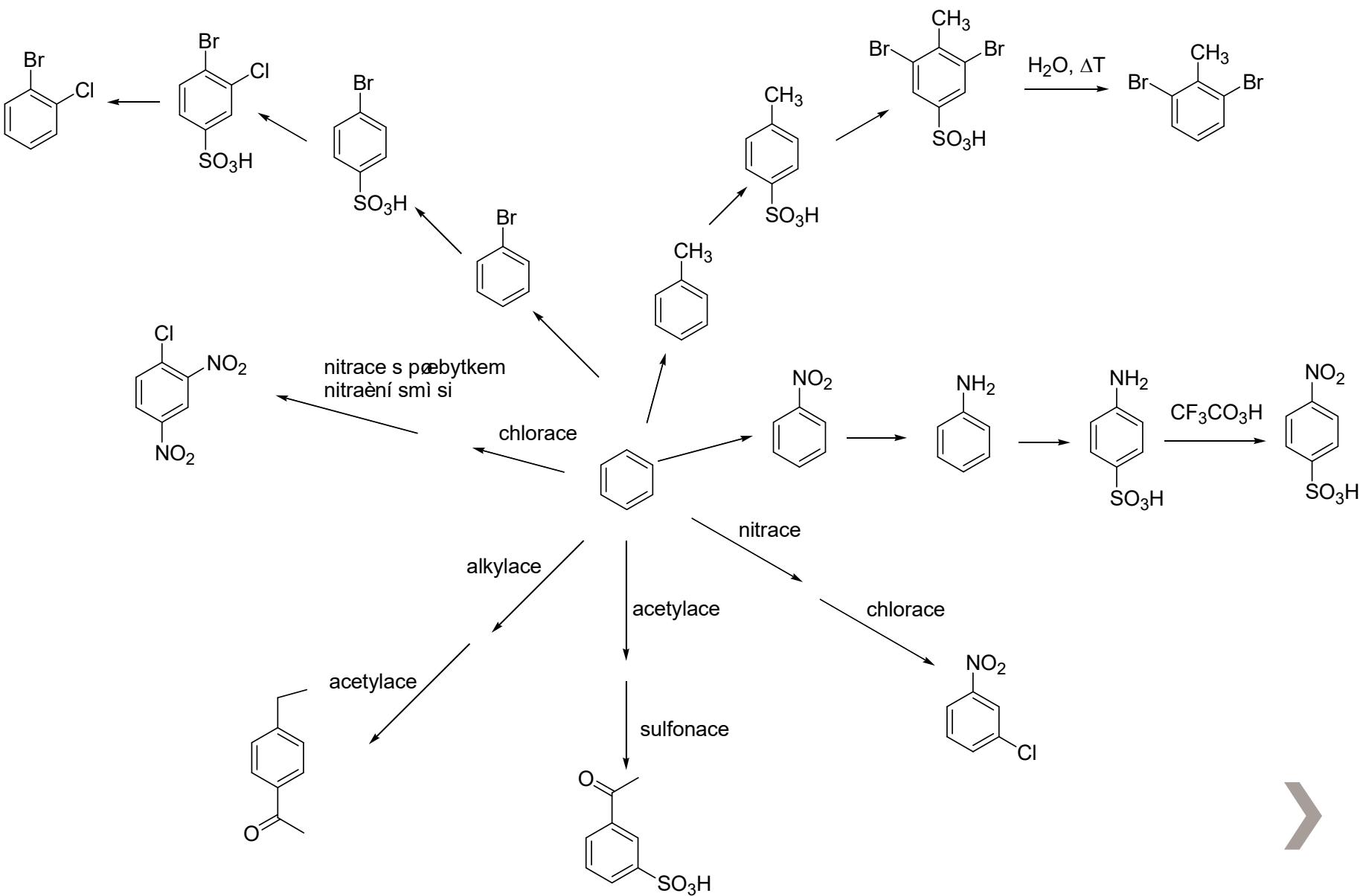


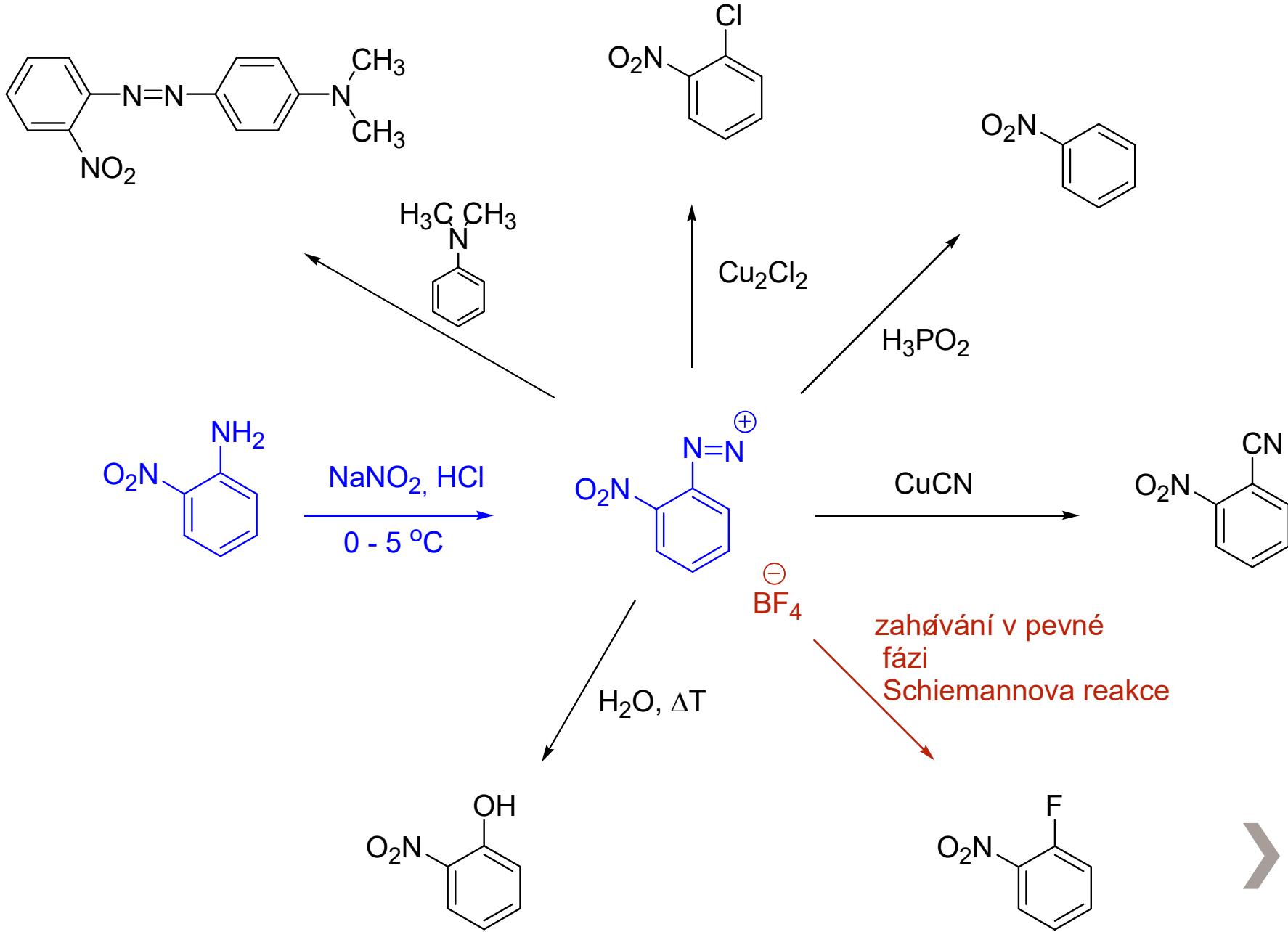
Navrhňete syntézu uvedené sloučeniny



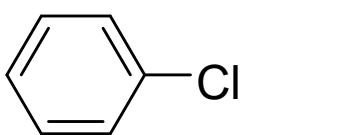
Navrhňte syntézu uvedené sloučeniny





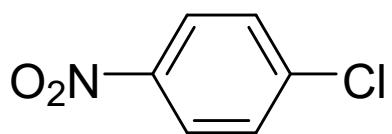


Nukleofilní aromatická substituce mechanismus Ad - E

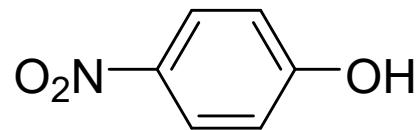


KOH, var

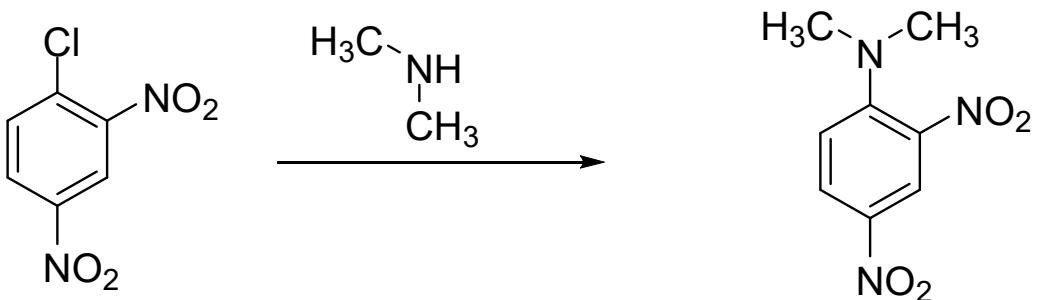
nereaguje



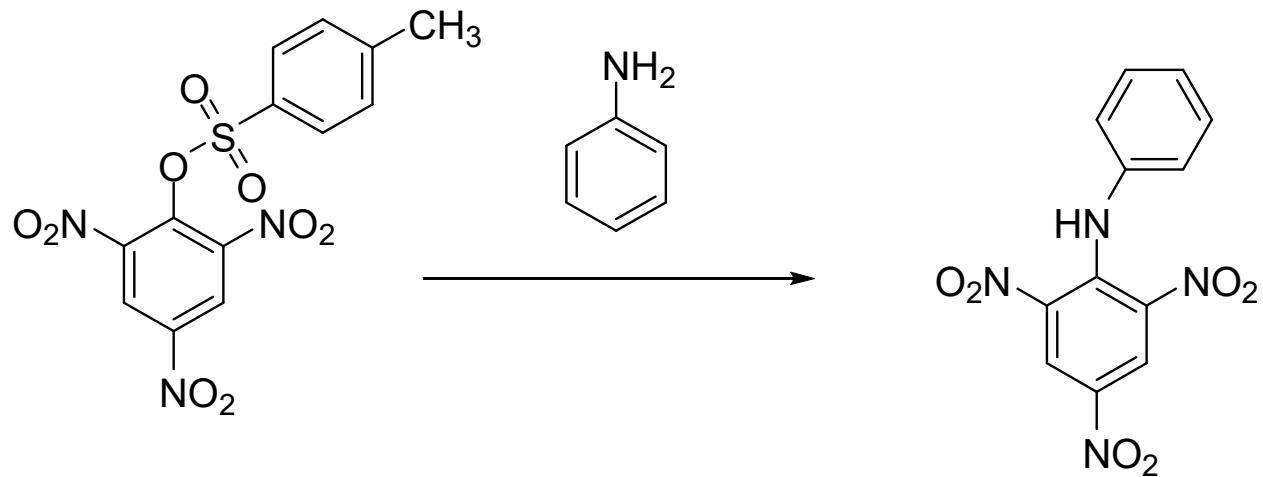
OH^- $130\text{ }^\circ\text{C}$



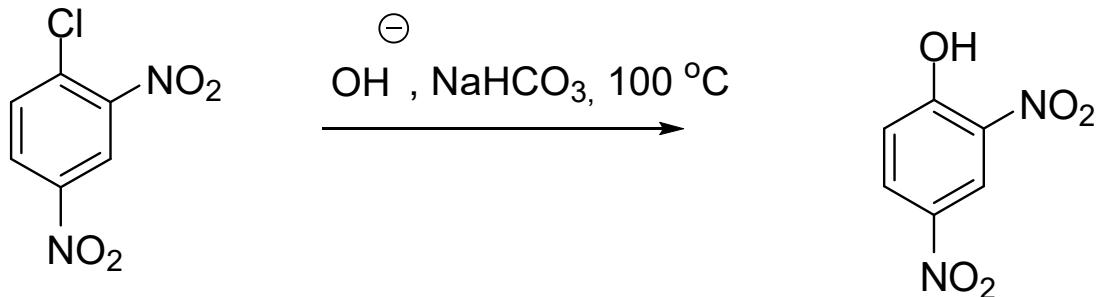
Nukleofilní aromatická substituce mechanismus Ad - E



Nukleofilní aromatická substituce mechanismus Ad - E



Nukleofilní aromatická substituce mechanismus Ad - E



Nukleofilní aromatická substituce mechanismus E - Ad

