

LOSCHMIDT  
LABORATORIES



# Introduction to the structure of macromolecules



# Course information



- 10 lectures  $\approx$  20 h

1. Introduction to the structure of macromolecules
2. Structure of biomolecules
3. Bioinformatics databases and structure prediction
4. Models of structures
5. Stability and dynamics of macromolecules
6. Analysis of protein structures
7. Protein-ligand complexes
8. Macromolecular complexes and interactions
9. Engineering of protein structures
10. Applications of structural biology and bioinformatics

# Course information

- 3 lecturers

- Sérgio Marques, PhD

- Lectures 1-2, 5-10



- Joan Planas, PhD

- Lectures 3, 4



- Mgr. David Bednář, PhD

- Lecture 5



# Course information



## ❑ Examination

- ❑ Written exam, multiple choices, 25 questions, 25 points
  - A: 25-22
  - B: 21-19
  - C: 18-16
  - D: 15-13
  - E: 12-10
  - F (fail): < 10

## ❑ 3 exam dates

- 12 Jan. 2021; 13:00 - B11/305
- 25 Jan 2021; 13:00 - B11/305
- 10 Feb. 2021; 09:00 - B11/305

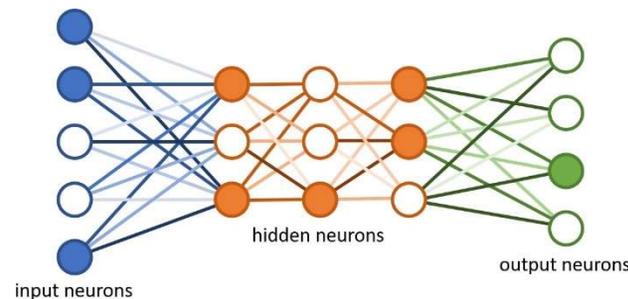
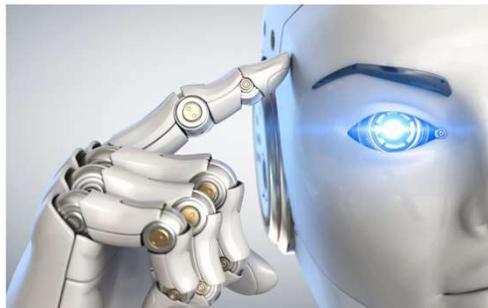
- ❑ Slides with essential information have the sign:



# AI in Biology, Chemistry, and Bioengineering - Bi9680En

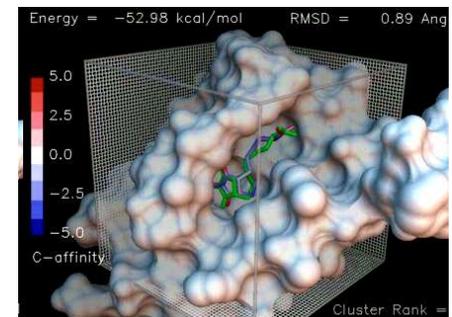
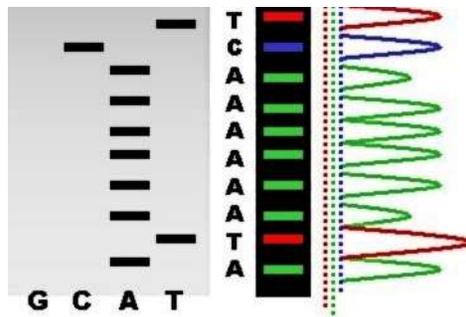


- Semester: autumn
- Lectures: 2 hours/week
- Lecturer: Dr. Stanislav Mazurenko
- Outline:
  - Modern bio-challenges: drug design, DNA interpretation, protein engineering
  - Types of AI algorithms and workflow for designing predictors
  - Clustering algorithms, random forests, artificial neural networks
  - Features, databases, and predictors used in applications



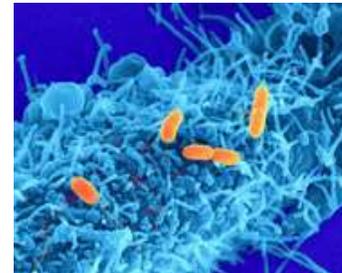
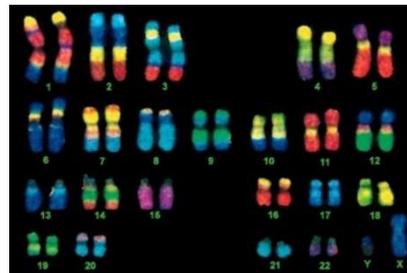
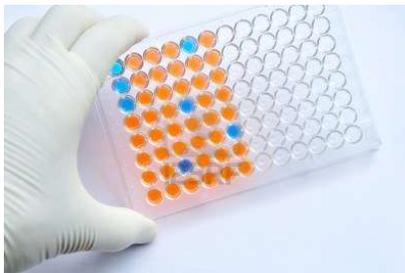
# Bioinformatics - Bi5000

- Semester: autumn
- Lectures: 2 hours/week
- Lecturer: Prof. Jiří Damborský and Dr. Roman Pantůček
- Outline:
  - Searching in bioinformatics databases
  - Analysis of nucleotide and protein sequences
  - Identification of genes
  - Analysis and prediction of protein structure



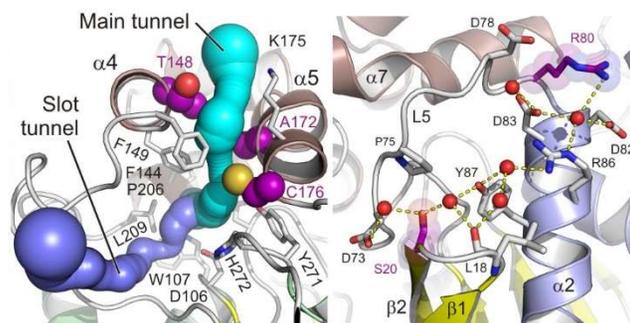
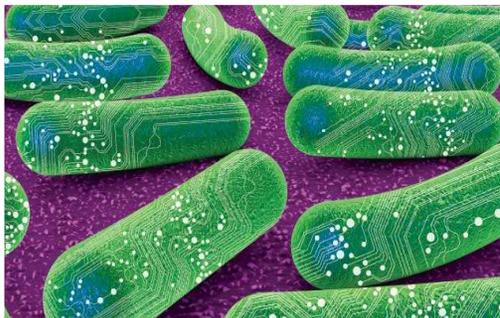
# Molecular biotechnology - Bi7430

- Semester: autumn
- Lectures: 2 hours/week; exercises: 2 hours/week
- Lecturer: Doc. Prokop, Dr. Dvořák, Dr. Bidmanová
- Outline:
  - Protein and metabolic engineering
  - Molecular diagnostics and modern vaccines
  - Cell and gene therapy and regenerative medicine
  - Molecular biotechnology in industry and agriculture



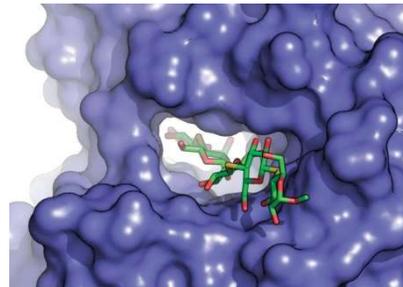
# Synthetic biology - S2015

- Semester: autumn
- Lectures: 2 hours/week
- Lecturer: Dr. Martin Marek and Dr. Karel Říha
- Outline:
  - Engineering concepts in synthetic biology
  - From genetic engineering to synthetic genomes
  - Protein engineering and design, from proteins to nanomachines
  - Metabolic engineering, artificial organelles



# Protein engineering - Bi7410

- Semester: spring
- Lectures: 1 hour/week
- Lecturer: Dr. Radka Chaloupková
- Outline:
  - Structure-function relationships in proteins
  - Expression and purification of recombinant proteins
  - Methods of structure and function analysis
  - Rational design, semi-rational design, directed evolution
  - Application of protein engineering

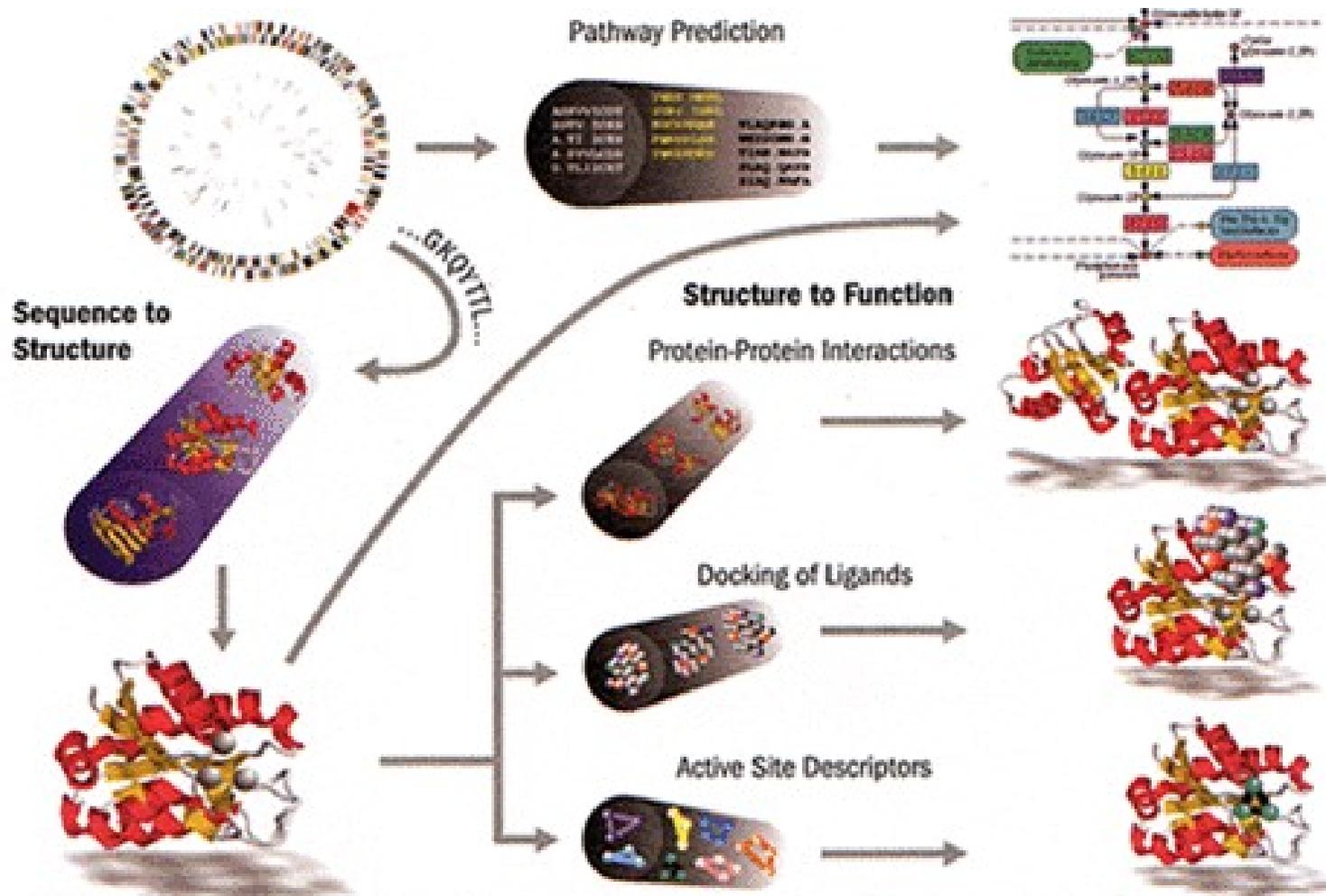


# Outline

- ❑ Motivation
- ❑ Structural biology
- ❑ Bioinformatics
- ❑ Visualization of structure
- ❑ Energetics of structures
- ❑ Molecular interactions
- ❑ Determination of structure

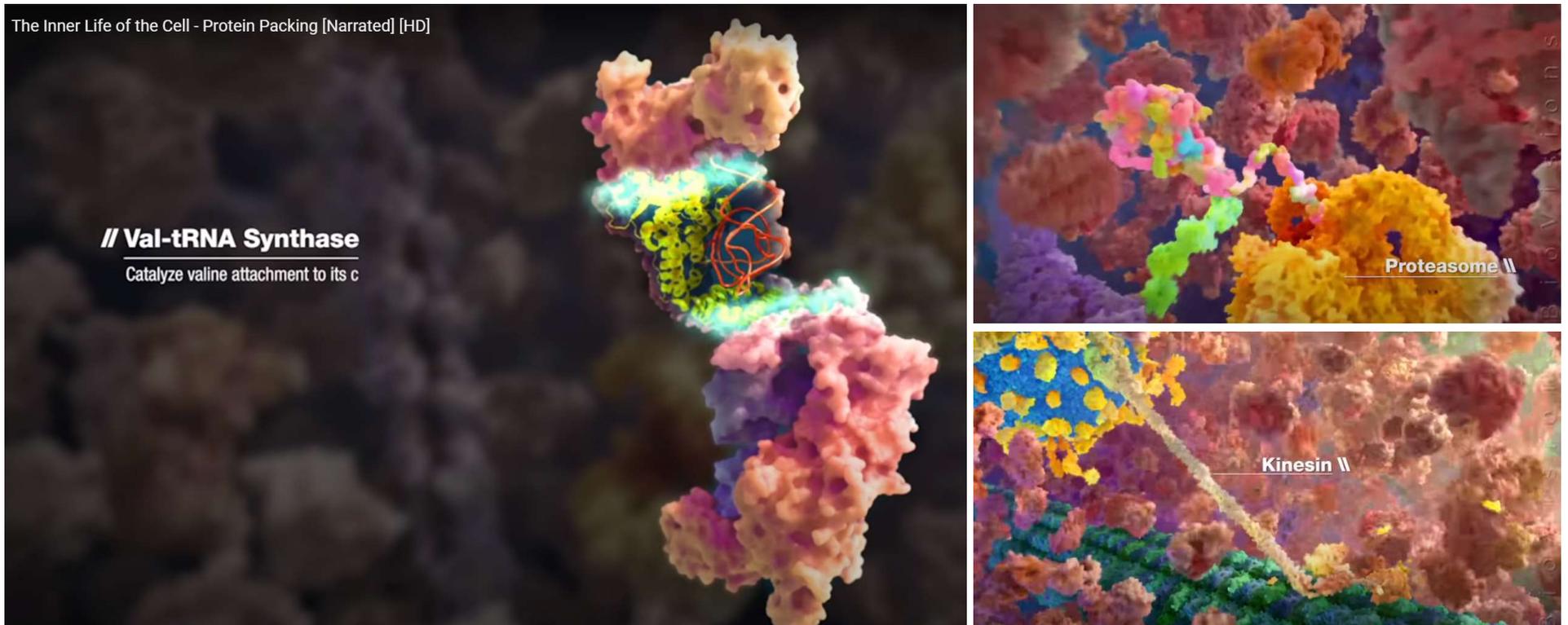
# Motivation

## □ Sequence-to-structure-to-function paradigm



# Motivation

- 3D structure ↔ biological function

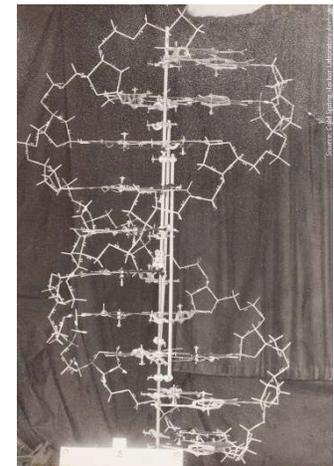
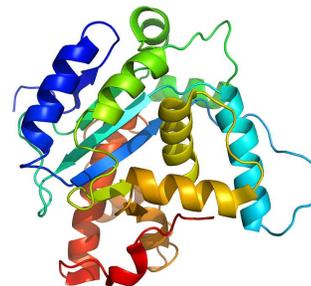
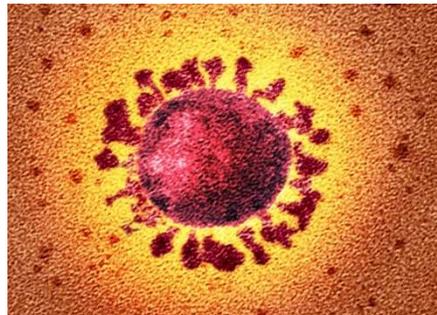
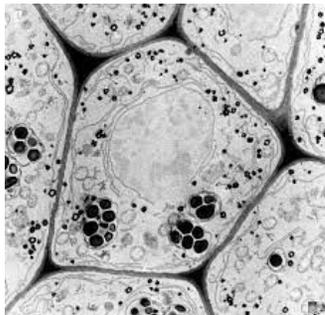


XVIVO & Harvard University – The inner life of the cell

<https://www.youtube.com/watch?v=VdmbpAo9JR4>

# Structural biology

- ❑ Focused on the three dimensional structures of biomolecules and their mutual interactions to understand their functions in the cell.
- ❑ Tries to make biological objects visible and understood
  - “Seeing is believing.”
  - “In order to understand, we need to see.”



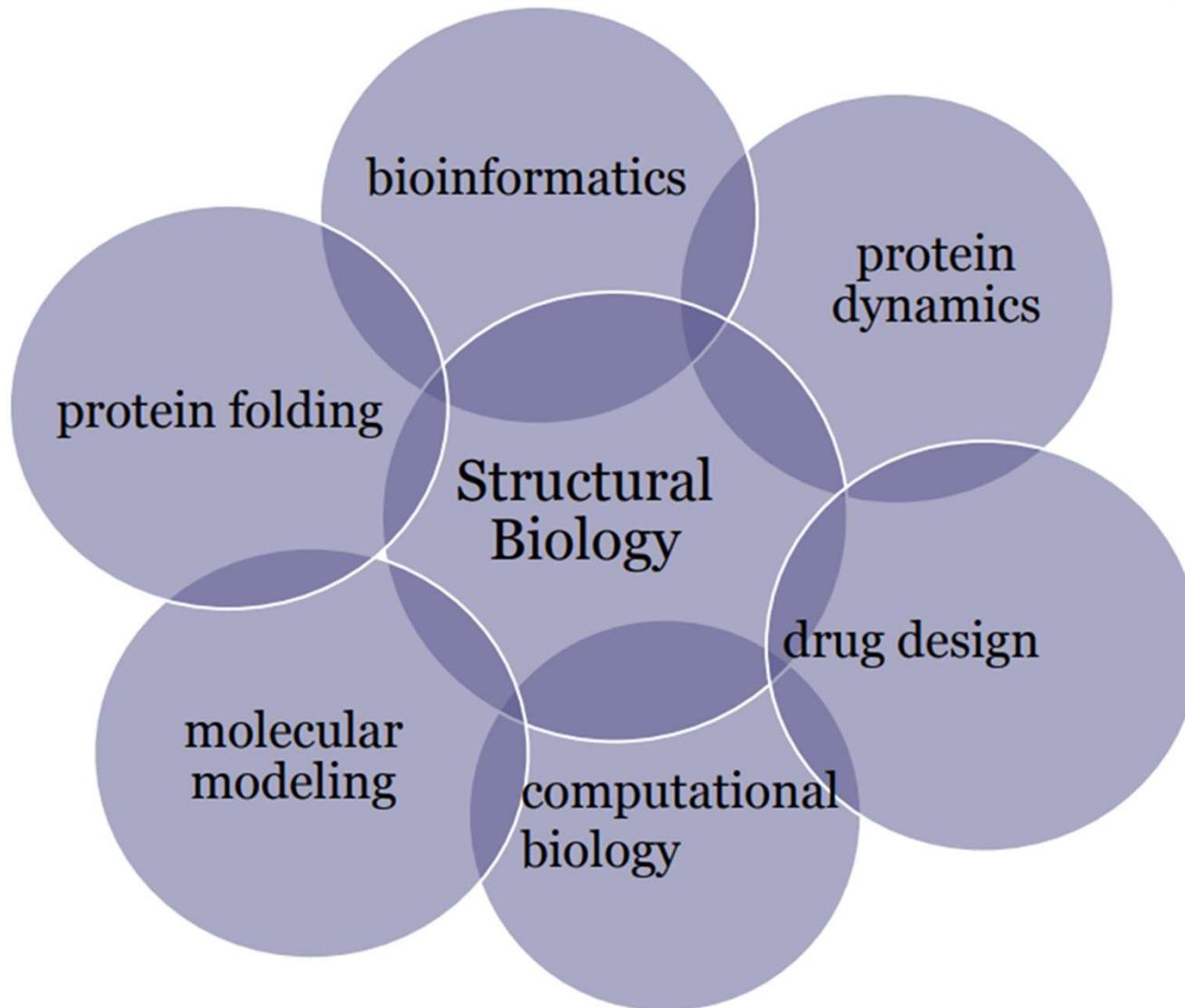
# Structural biology



- ❑ Important milestones
  - 1869 – DNA discovery
  - 1953 – DNA structure
  - 1959 – Myoglobin structure
  - 1968 – Hemoglobin structure
  
- ❑ “Unfortunately, we cannot accurately describe at the chemical level how a molecule functions unless we first know its structure”

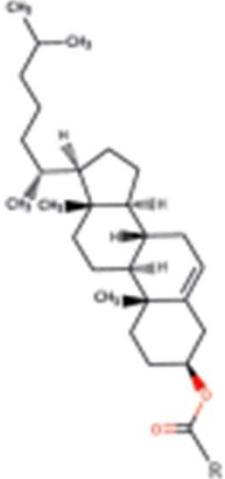
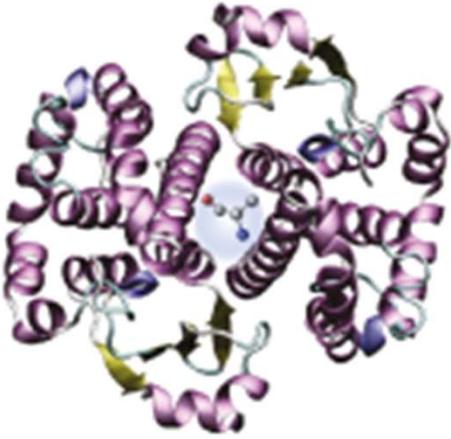
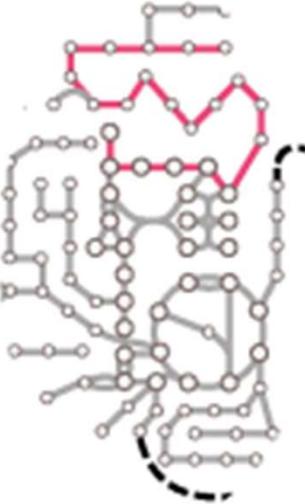
James Watson, 1964

# Structural biology



# Structural biology



	antibiotic drug	substrate-enzyme	cellular systems	organism phenotype
<b>scale</b>				
<b>objectives</b>	ligand structure similarity identifies promiscuous activity on antibiotics	prediction of ligand off-target binding to protein active or allosteric binding sites	metabolic pathway perturbations as identified by constraint-based modeling	Development of resistance; expression profiles of over expressed genes in presense of drug

# Bioinformatics

- “Bioinformatics is an interdisciplinary field that develops methods and software tools for understanding biological data, in particular when the data sets are large and complex.”

*In Wikipedia*

- Sequence analysis, genomics, proteomics, systems biology, structural bioinformatics

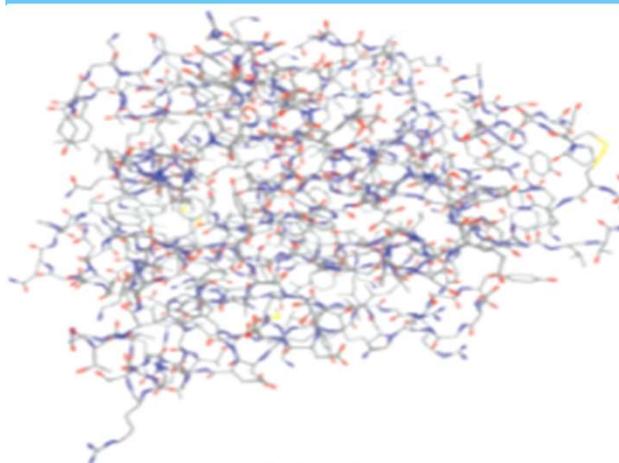
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B4F917.1 13 SIKLWPPSESTRIMLVDRMTNNLST..ESIFSRK..YRLLGKQEAHENAKTIEELCFALADE.....HFREEPDGDGSSAVQLYAKETSKMMLEVLK 100
A9S1V2.1 23 VFKLWPPSQGTREAVRQKMKALKLSS..ACFESQS..FARIELADAQE HARAIIEEVAFGAQE.....ADSGGDKTGSAVVMVYAKHASKLMLETLR 109
B9GSN7.1 13 SVKLWPPGQSTRMLMLVERMTKNFIT..PSFISRK..YGLLSKEEAEEEDAKKIEEVAFAAANQ.....HYEKQPDGDDGSSAVQIYAKESRRLMLEVLK 100
Q8H056.1 30 SFSIWPPPTQRTRDAVVRRLVDTLGG..DTILCKR..YGAVPAADAEPAAARGIEAEAFDAAAA..SGEAAATASVEEGIKALQLYSKEVSRRLDFVK 120
Q0D4Z3.2 44 SLSIWPPSQRTDRAVVRRLVQTLVA..PSILSKR..YGAVPEAEAGRAAAVEAEAYAAVTES..SSAAAAPASVEDGIEVLQAYSKEVSRRLLELAK 135
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A9NW46.1 13 SIKLWPPSESTRMLMLVERMTDNLSS..VSFFSRK..YGLLSKEEAENAKRIEETAFLAND.....HEAKEPNLDDSSVVQFYAREASKLMLEALK 100
Q9C500.1 57 SLRIWPPPTQKTRDAVLNRLIETLST..ESILSKR..YGTLSKDDATTVAKLIEEEAYGVASN.....AVSSDDDGKILELYSKEISKRMLESVK 142
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Q9M651.2 13 SIKLWPPSLPTRKALIERITNNFSS..KTIFTEK..YGLTKDQATENAKRIEDIAFSTANQ.....QFEREPDGDGSSAVQLYAKECSKLILEVLK 100
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# Structure visualization

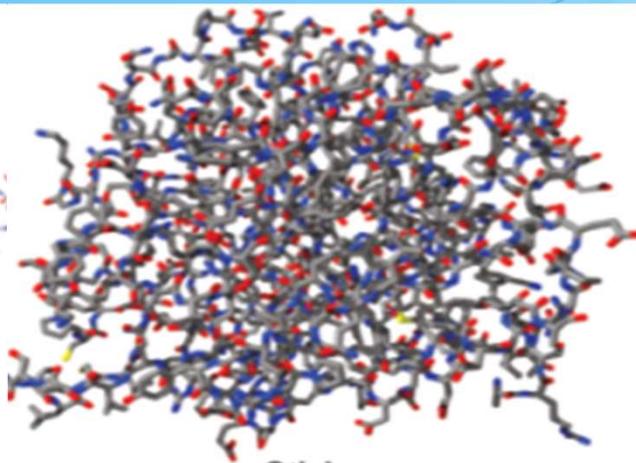


- ❑ Some widespread-used programs
  - PyMOL – <http://www.pymol.org/>
  - Chimera – <http://www.cgl.ucsf.edu/chimera/>
  - VMD – <http://www.ks.uiuc.edu/Research/vmd/>
  - Caver Analyst – <https://www.caver.cz>
- ❑ Various representation
  - bond-based
  - backbone-based
  - surface-based
- ❑ Seeing is believing but ...
  - beware of misinterpretations and over-interpretations!

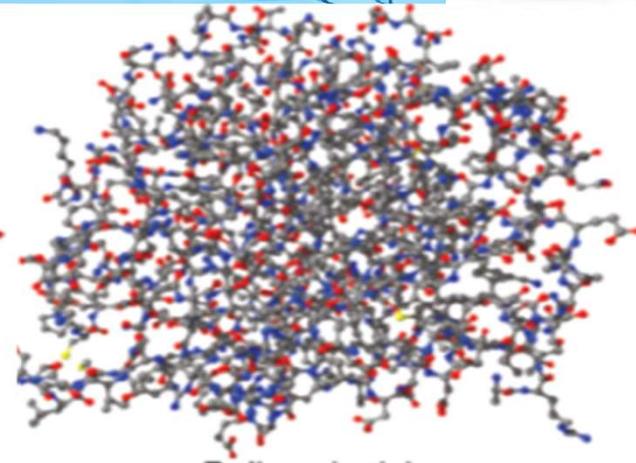
# Structure visualization



Wireframe



Stick

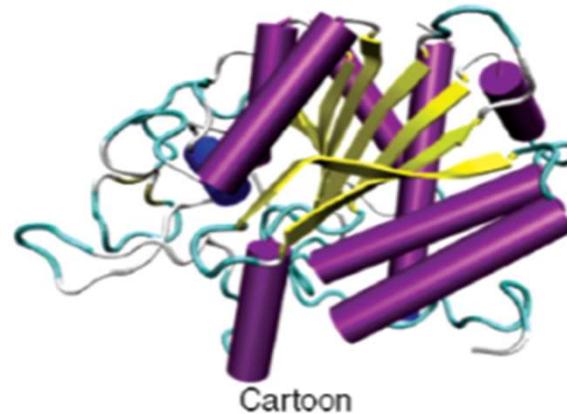


Ball and stick

## ❑ Bonds-based representation

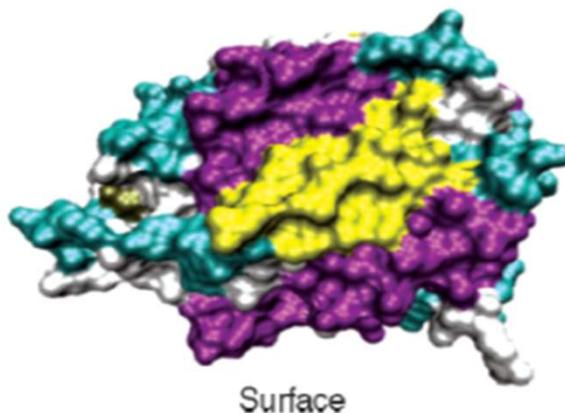
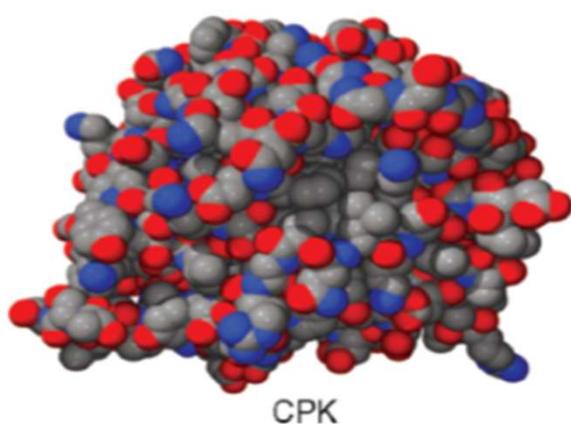
- fast, small demands on resources
- suitable to detailed analysis
- incorrect impression about atom packing (empty space) and interatomic distances

# Structure visualization



- Backbone-based representation
  - moderately fast & resource-demanding
  - suitable to investigation of folds and secondary structure
  - good for overall orientation in the structure – shows landmarks

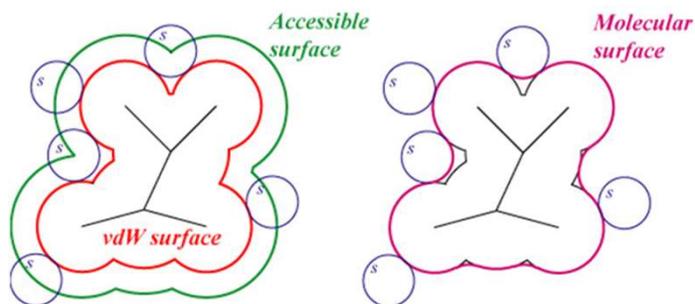
# Structure visualization



□	hydrogen (H)	white
■	carbon (C)	black
■	nitrogen (N)	blue
■	oxygen (O)	red
■	fluorine (F), chlorine (Cl)	green
■	bromine (Br)	dark red
■	iodine (I)	dark violet
■	noble gases (He, Ne, Ar, Kr, Xe)	cyan
■	phosphorus (P)	orange
■	sulfur (S)	yellow

## □ Surface-based representation

- very slow & very resource-demanding
- applicable to study of cavities & molecular contacts and shapes



# Energetics of structures



- ❑ Energy
- ❑ Entropy
- ❑ Free energy
- ❑ Energy landscape

# Energetics of structures



## □ Energy

- internal energy **U** (const. V), enthalpy **H** (constant P), ...
- total energy often inaccessible -> differences in energy
- negative energy is favorable
  
- potential **V/E<sub>p</sub>** energy – interactions of atoms in a system
- kinetic **T/E<sub>k</sub>** energy – movement of atoms

$$U = E_p + E_k$$

# Energetics of structures



## □ Entropy

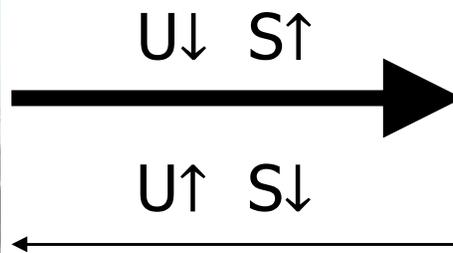
- relates to the thermal disorder of the system
- total entropy  $S > 0$
- higher entropy is more favorable

# Energetics of structures



## Free energy

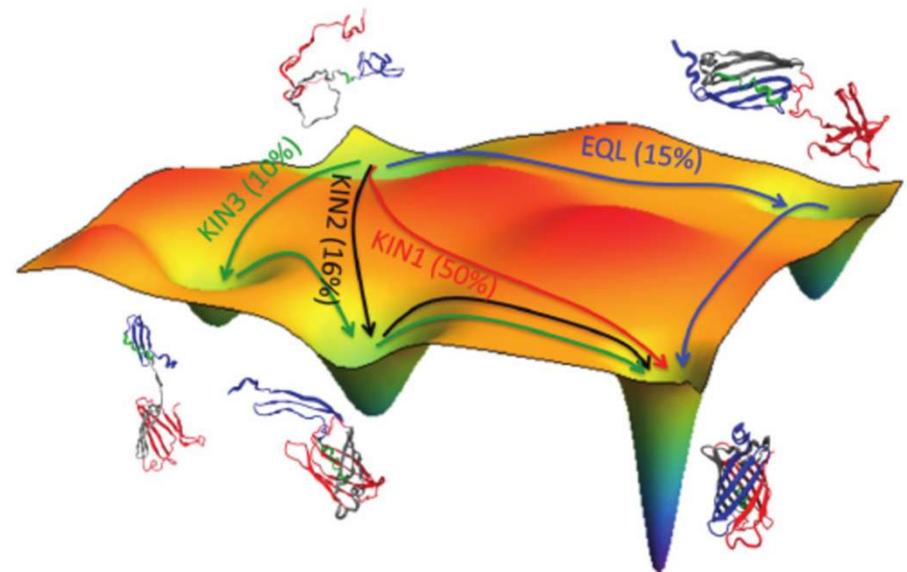
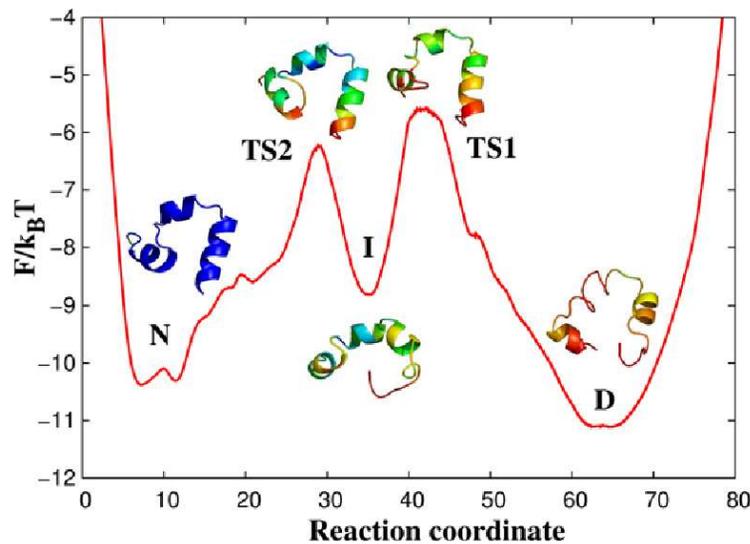
- Helmholtz **A/F** (const. V), **Gibbs G** (const. P)
- both internal energy **H** (enthalpy) and entropy **S**
- **$A = U - TS$**  and  **$G = H - TS$** ;  **$\Delta G = \Delta H - T\Delta S$**
- **negative free energy change is favorable** ( $\Delta G < 0$ )



# Energy landscape



- Relationship between structure and its potential energy
  - structure dictates potential energy – how strong are the individual interactions
  - potential energy reflects probability of finding the different structures – **lower energy structures occur more frequently**



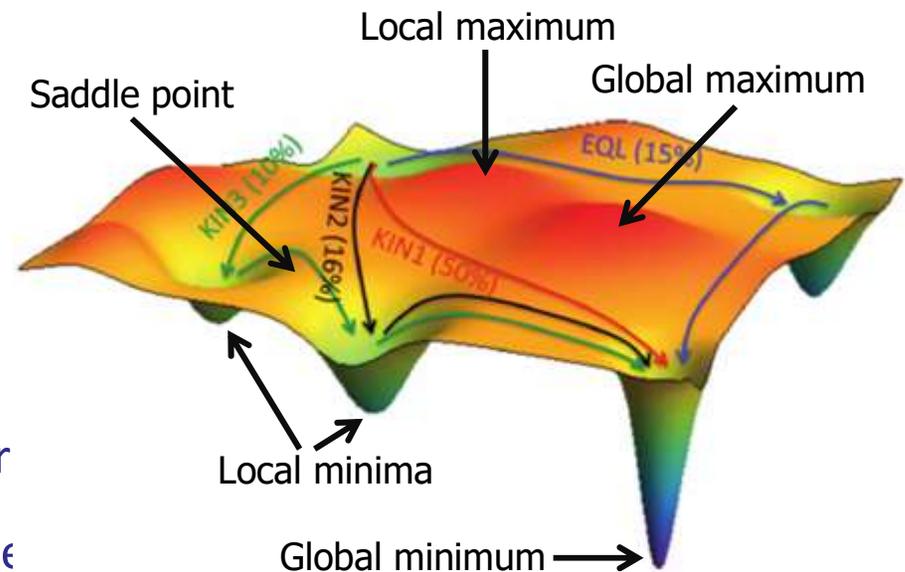
# Energy landscape



- Relationship between structure and its potential energy
  - structure dictates potential energy – how strong are the individual interactions
  - potential energy reflects probability of finding the different structures – lower energy structures occur more frequently

- Potential energy surface

- multidimensional surface
- minima – stable structures
- saddle points – transient str
- maxima – unstable structure



# Molecular interactions



- ❑ Covalent interactions (chemical bonds)
  - sharing of electrons
  - very stable under standard condition
  
- ❑ Non-covalent interactions
  - much weaker than covalent
  - electrostatic interactions
  - polar interactions
  - non-polar interactions

# Electrostatic interactions



## □ Charge-charge interactions

- Coulomb's law
- **long-range interaction** – decrease with  $r^2$
- **environment dependent**
  - permittivity:  $\epsilon = \epsilon_0 \cdot \epsilon_r$
  - relative permittivity ( $\epsilon_r$ ) = dielectric constant

$$F = \frac{q_1 \cdot q_2}{4\pi \cdot \epsilon \cdot r^2}$$

Material	$\epsilon_r$
Vacuum	1.0
Air	1.0006
Teflon	2.1
Interior of proteins, membranes	2-20
Water (20 deg C)	80.1
Water (0 deg C)	88

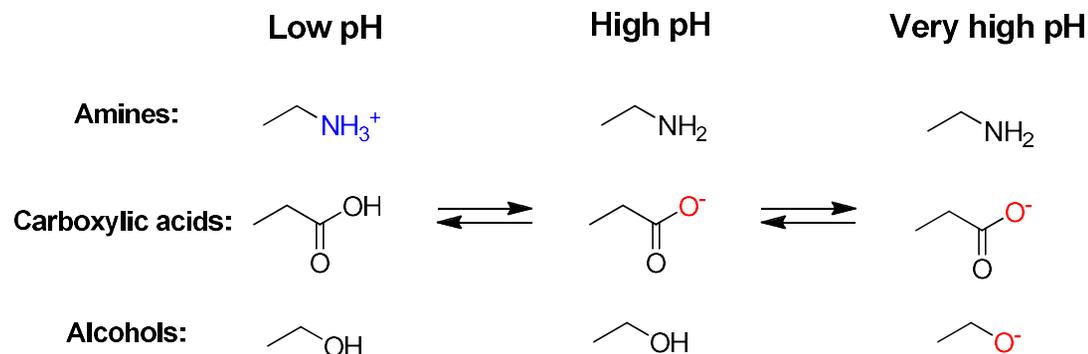
# Electrostatic interactions



## □ Charge-charge interactions

- Coulomb's law
- long-range interaction – decrease with  $r^2$
- environment dependent
  - permittivity:  $\epsilon = \epsilon_0 \cdot \epsilon_r$
  - salt concentration – presence of counter ions
  - pH – change in charge of molecule

$$F = \frac{q_1 \cdot q_2}{4\pi \cdot \epsilon \cdot r^2}$$

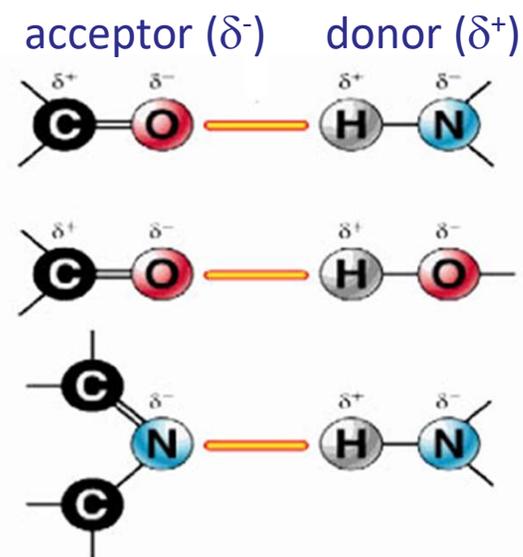


# Polar interactions



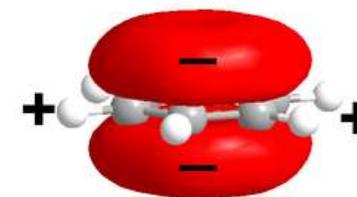
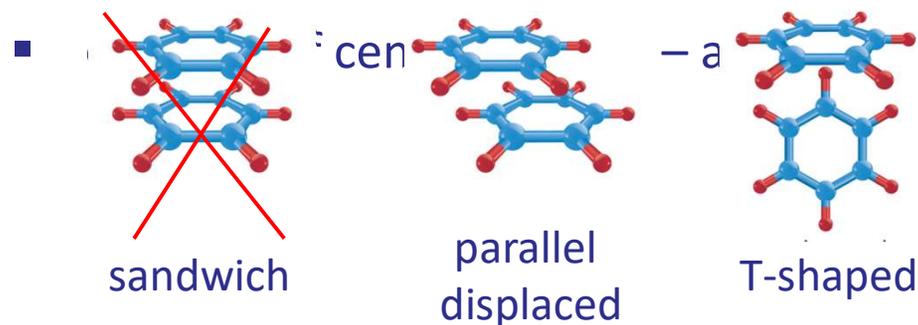
## □ Hydrogen bonds (H-bonds)

- donor and acceptor atoms sharing hydrogen
- H-bond distance: 2.8-3.4 Å
- with highly electronegative atoms: F, O, N



## □ Aromatic ( $\pi$ - $\pi$ ) interactions

- attractive interaction between aromatic rings



# Non-polar interactions

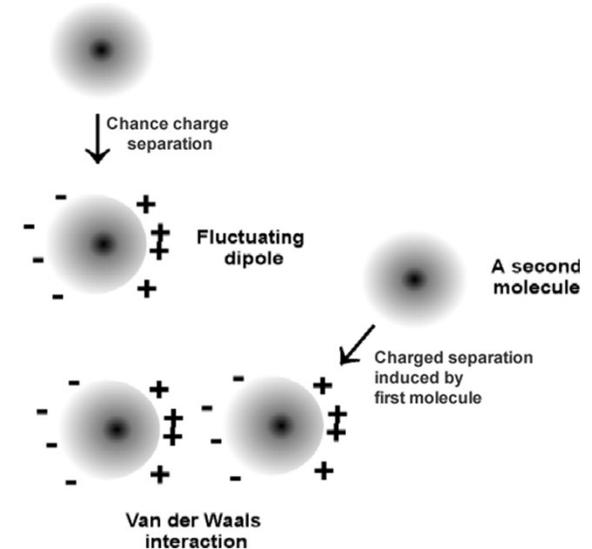


## □ van der Waals (vdW) interactions

- between any two atoms
- important in non-polar molecules
- short-range interactions – up to 5 Å

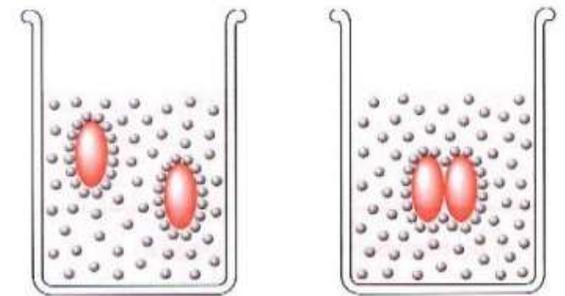
$$F_{\text{vdW}}(r) = -\frac{AR_1R_2}{(R_1 + R_2)6r^2}$$

$R_1, R_2$  – van der Waals radii  
 $r$  – distance



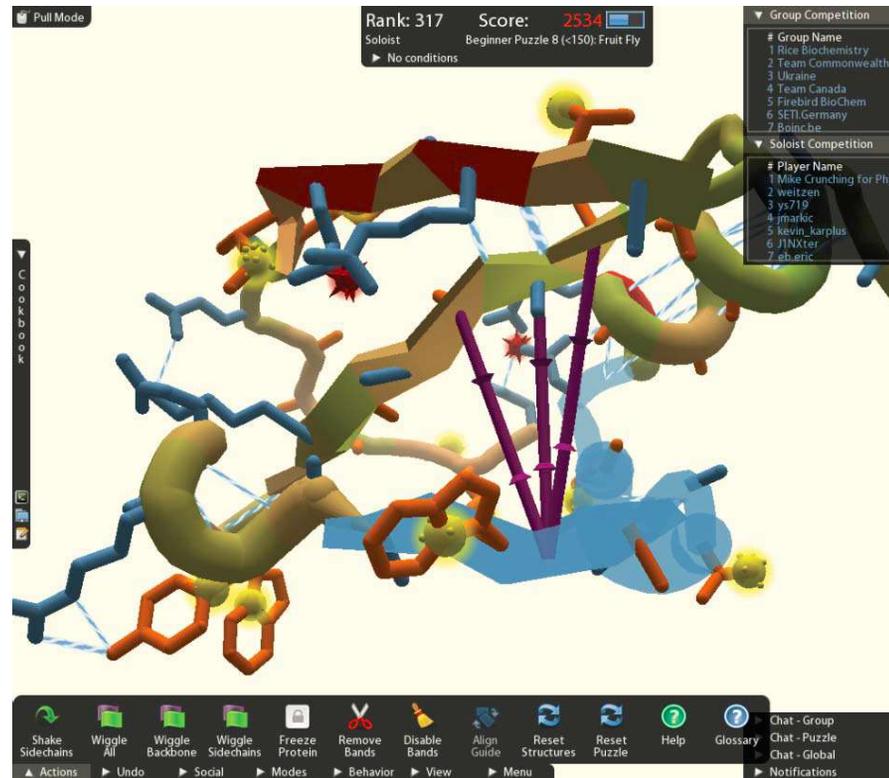
## □ Hydrophobic interactions

- entropic origin – water molecules ordered around hydrophobic moiety -> unfavorable
- hydrophobic packing -> favorable release of some ordered water molecules



# Protein folding game

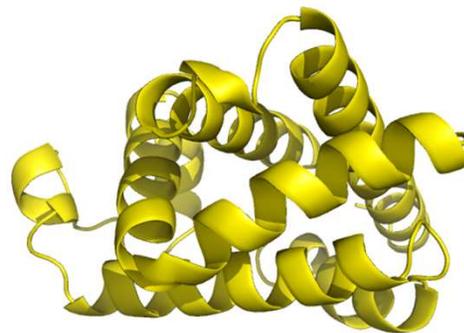
- ❑ FOLD.IT – <http://fold.it/portal/>
  - crowdsourcing online game
  - prediction of protein structures



# Structure determination

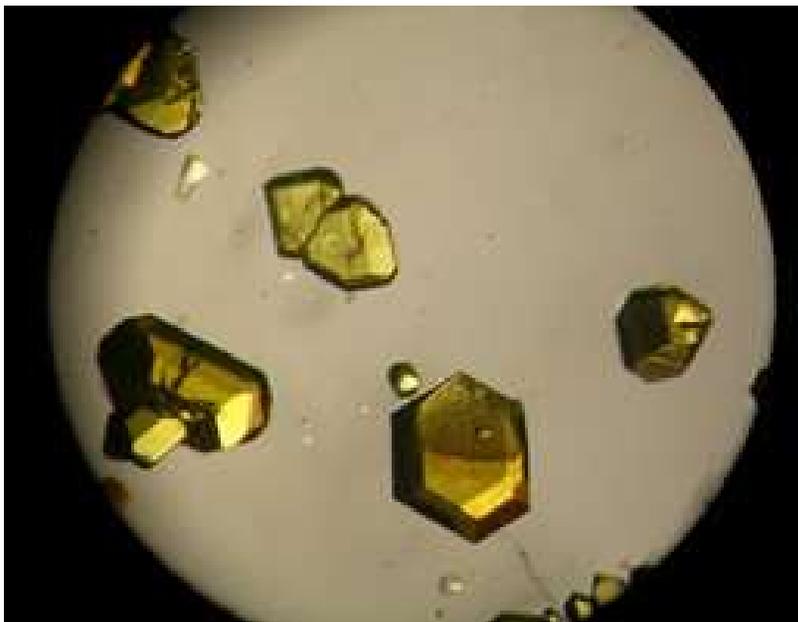


- ❑ Established methods
  - X-ray crystallography
  - NMR spectroscopy
  - electron microscopy
  - bioinformatics predictions – theoretical



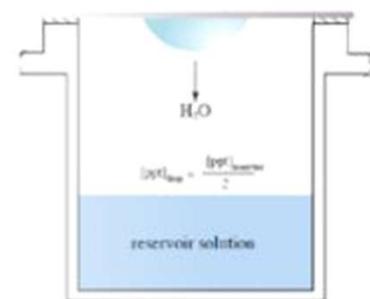
# X-ray crystallography

- Crystallization procedures
  - slow (days-weeks)
  - high risk of failure

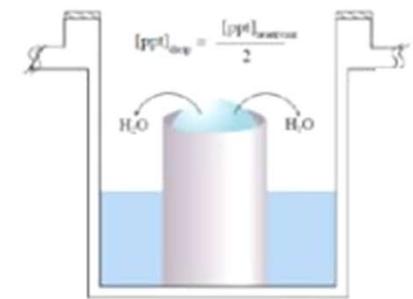


## Some Crystallization Methods:

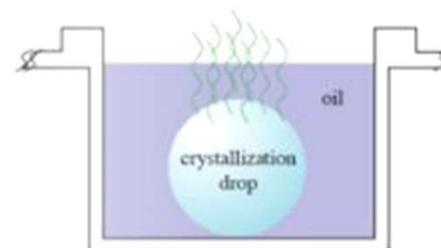
Vapor diffusion  
Hanging-drop



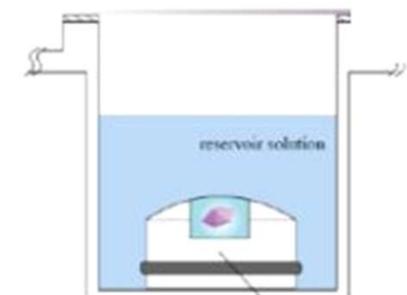
Sitting-drop



Batch:  
micro batch under oil

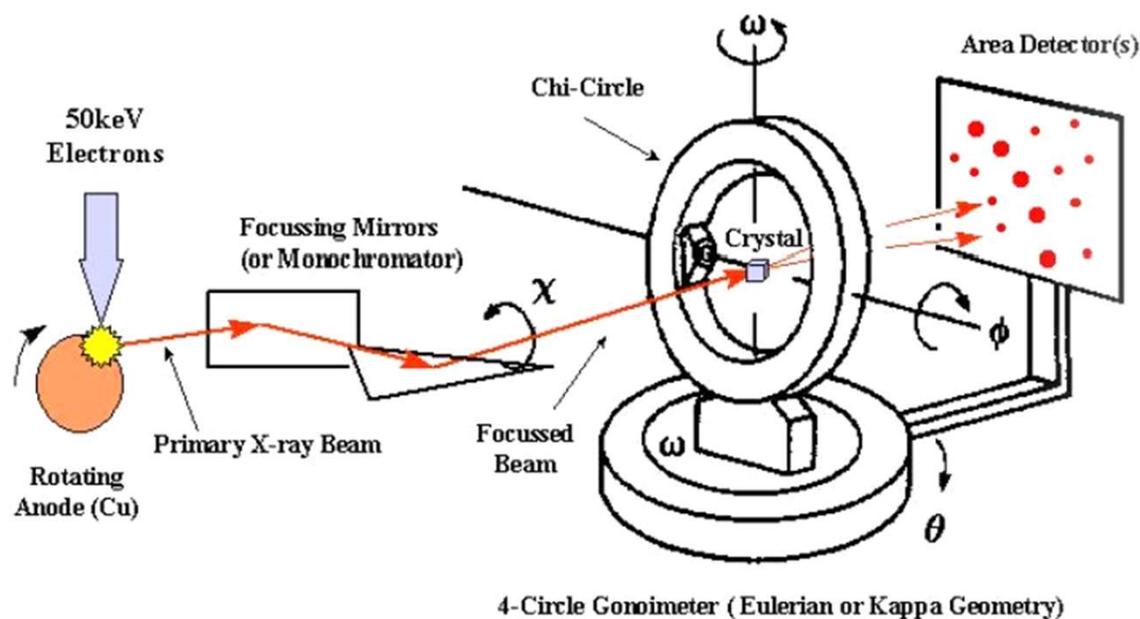


Dialysis



# X-ray crystallography

## Data Collection



**X-ray sources:** X-ray tubes, rotating anodes and synchrotrons.

Synchrotrons produce the brightest X-rays



APS Chicago

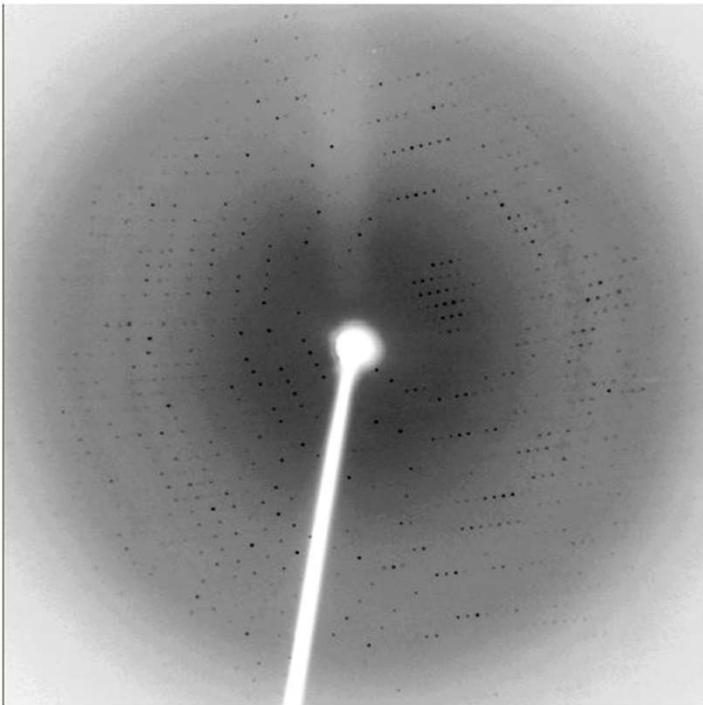


European Synchrotron Radiation Facility, Grenoble

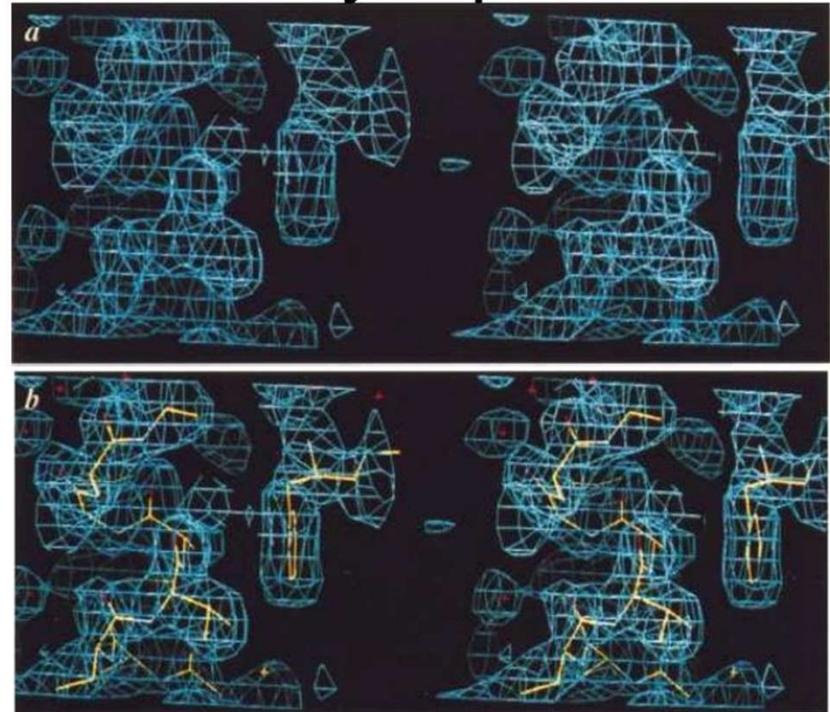
# X-ray crystallography



Image of diffraction



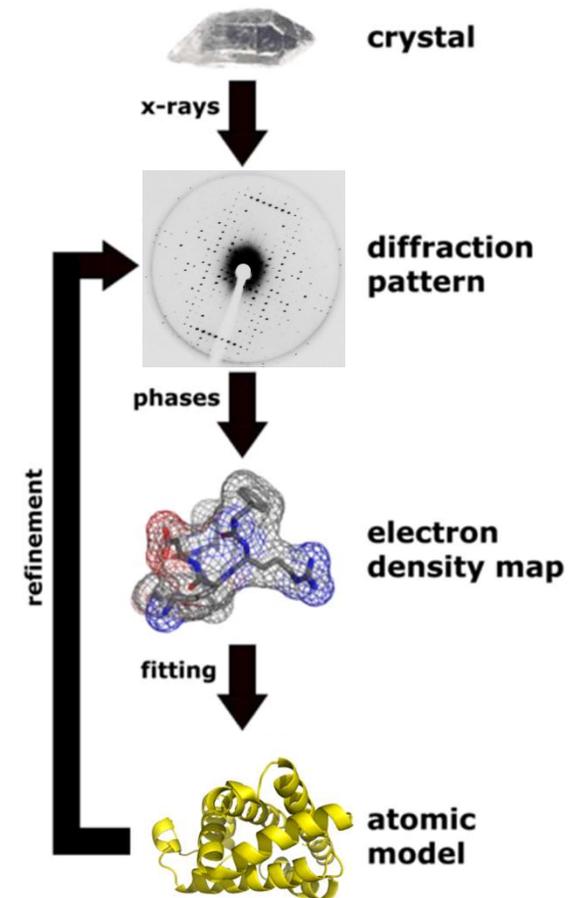
Electron density map



Building a structure model

# X-ray crystallography

- ❑ Crystallization
  - hanging drop, sitting drop, microbatch
- ❑ Data collection
  - diffractometers, synchrotrons
- ❑ Analysis of diffraction data
  - solving phase problem
    - isomorphous replacement
    - molecular replacement
    - anomalous scattering
- ❑ Iterative model building

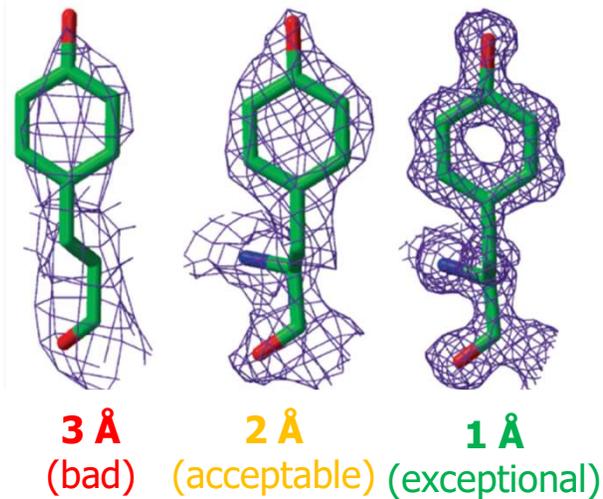


# Parameters of an X-ray structure



## □ Resolution

- measure of the level of detail present in the diffraction pattern



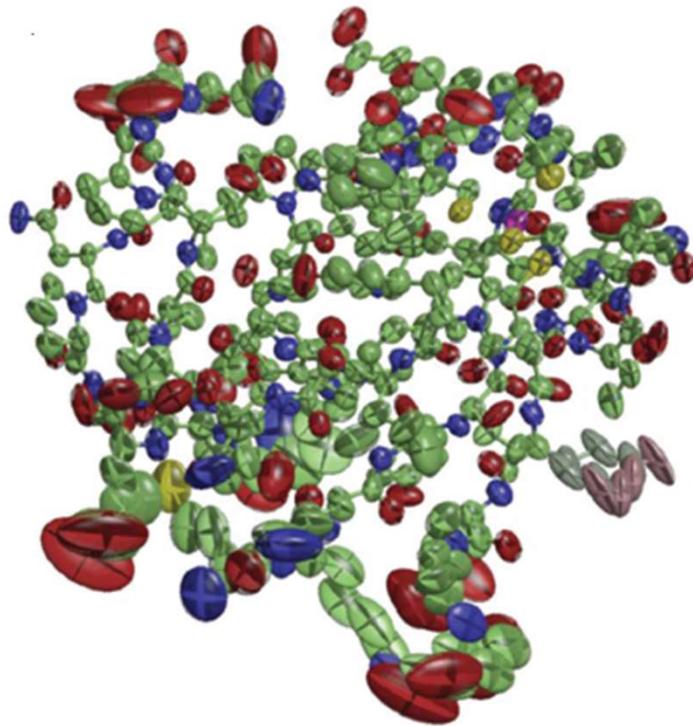
## □ R-factor (R-value)

- measure of a model quality - i.e. the agreement between the crystallographic model and the diffraction data
- values from 0 (ideal) to 0.63 (random structure), typically about 0.2

# Parameters of an X-ray structure



- ❑ Thermal factors (B-factors)
  - measure of how much an atom oscillates or vibrates around the position specified in the model



# X-ray crystallography



## ❑ Advantages

- no limitations in size
- possibility to obtain an atomic resolution

## ❑ Disadvantages

- requirement of a crystal
- structure in a crystalline (non-native) state
- static picture of macromolecule
- position of hydrogen atoms (usually) not detected

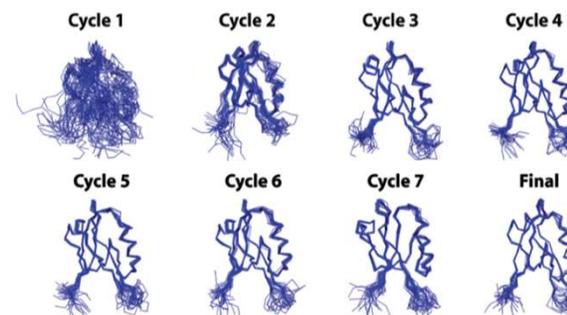
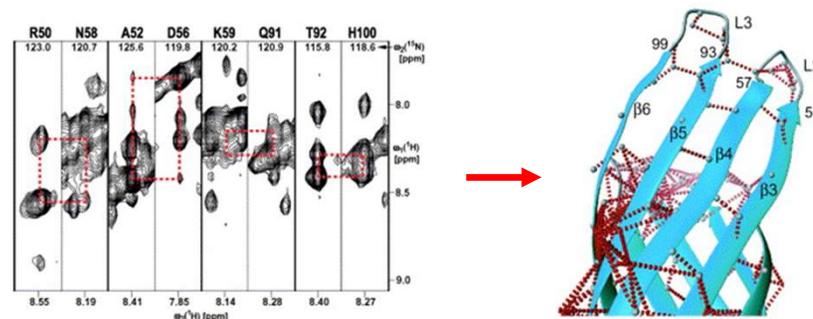
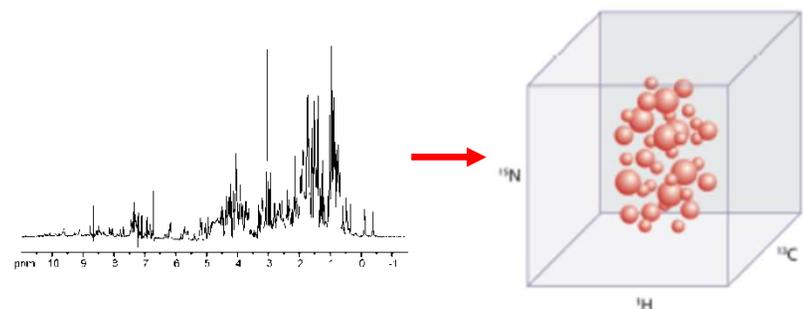
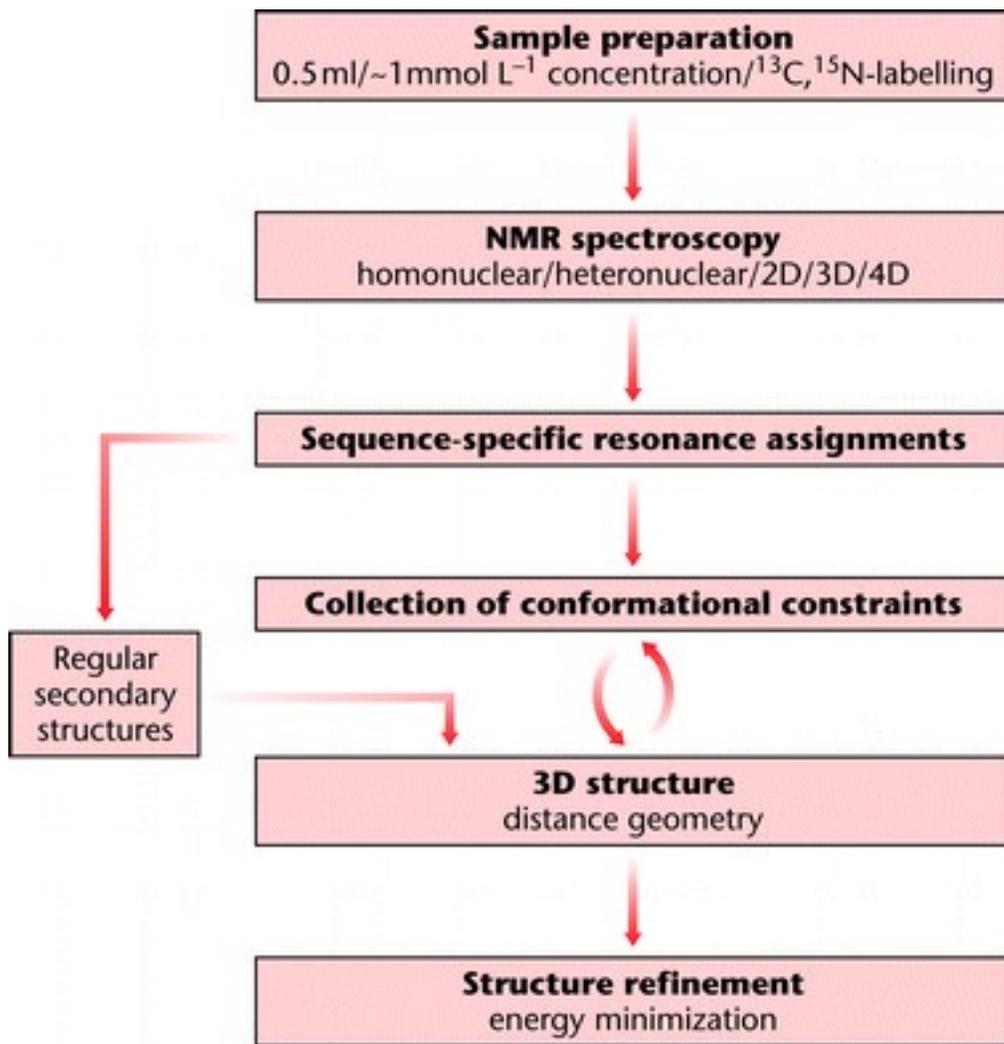
# NMR spectroscopy

- Nuclear magnetic resonance (NMR)
  - Detects energy transitions in the magnetic moments of nuclei with non-zero nuclear spins
  - Common isotopes:  $^1\text{H}$ ,  $^{13}\text{C}$ ,  $^{15}\text{N}$ ,  $^{31}\text{P}$ ,  $^{35}\text{Cl}$



900 MHz NMR spectrometer

# NMR spectroscopy

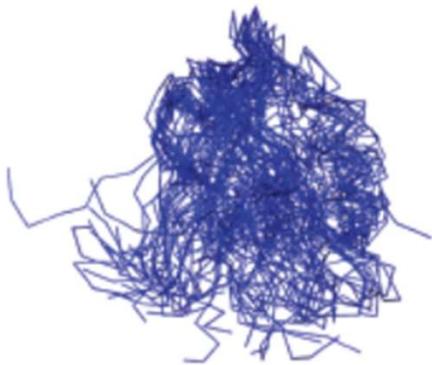


# Parameters of an NMR structure



## □ RMSD

- root-mean-squared deviation of atomic positions across the ensemble of solutions
- reveals differences between individual conformations
- **general parameter in many fields of structural biology**



**RMSD = 3.59 Å**



**RMSD = 1.06 Å**



**RMSD = 0.42 Å**

$$RMSD = \sqrt{\frac{1}{N} \sum_{i=1}^N \delta_i^2}$$

# NMR spectroscopy



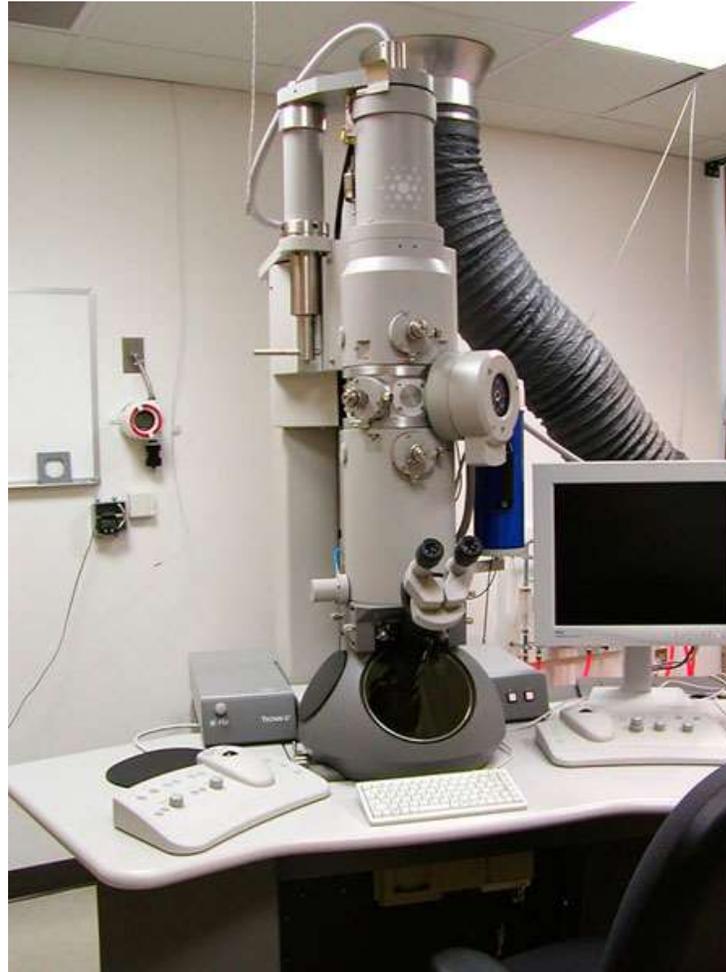
## □ Advantages

- structure in solution (native) state
- possibility to investigate dynamics of macromolecule
- position of hydrogen atoms detected

## □ Disadvantages

- size limited to approximately 40 kDa
- requirement of isotopically labeled sample

# Electron microscopy

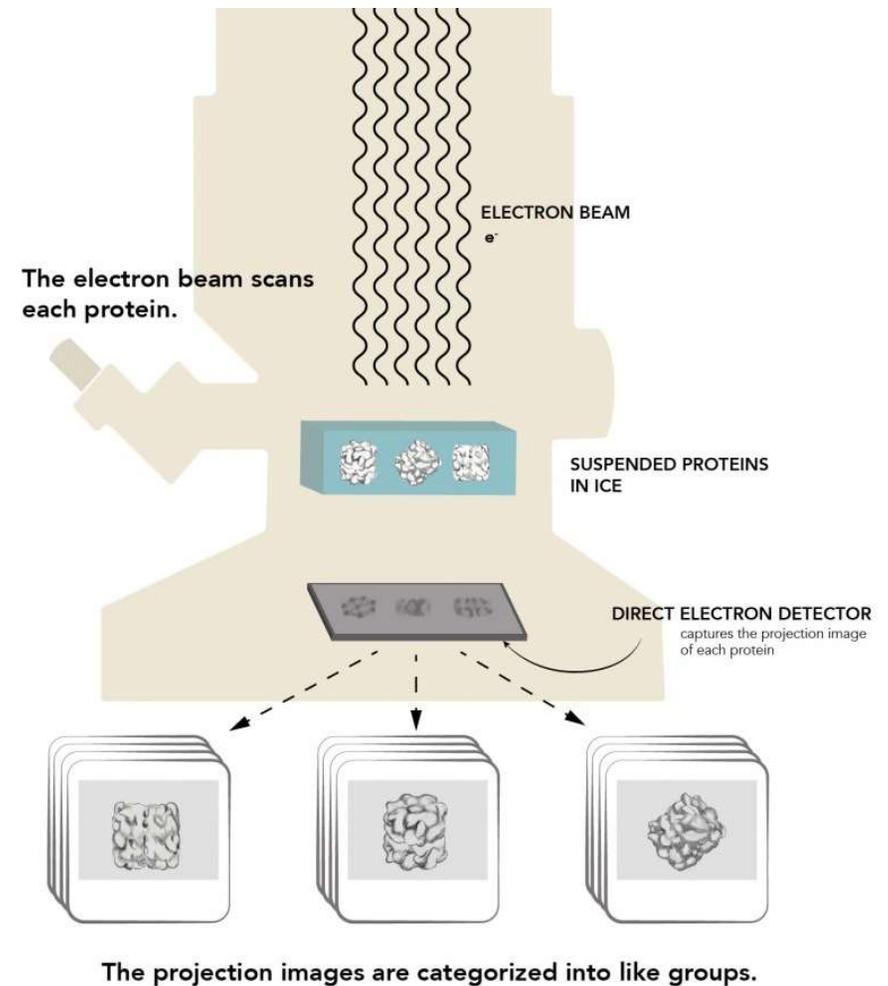
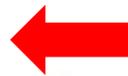
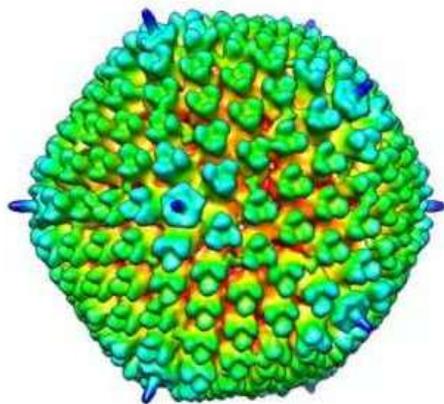


FEI Tecnai T12 Cryotransmission Electron Microscope

# Electron microscopy

- ❑ wavelength of an electron is much shorter than the wavelength of light
- ❑ so it can reveal much smaller things
- ❑ samples are flash-frozen in their natural environments (cryo-EM)
- ❑ can generate 3D images of large molecules at nearly atomic resolution

Reconstruction of Adenovirus by CryoTEM

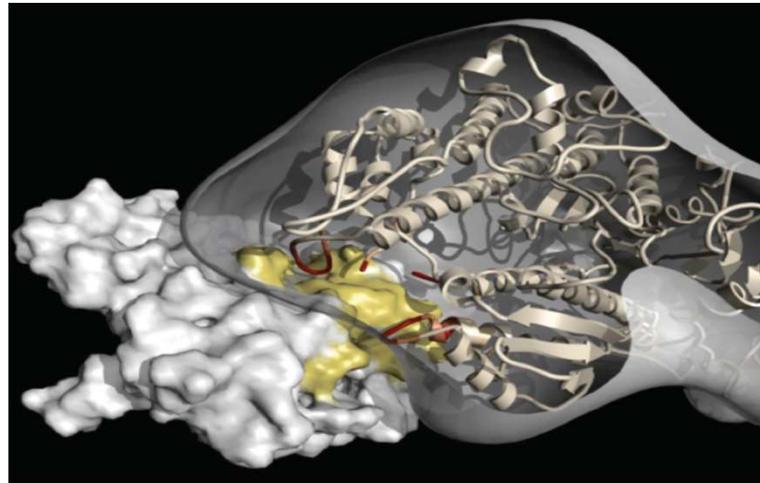


# Electron microscopy



## □ Advantages

- applicable to extremely large systems
- complements other methods e. g. X-ray, NMR



## □ Disadvantages

- lower resolution (2-3 Å at best)

# Bioinformatics predictions



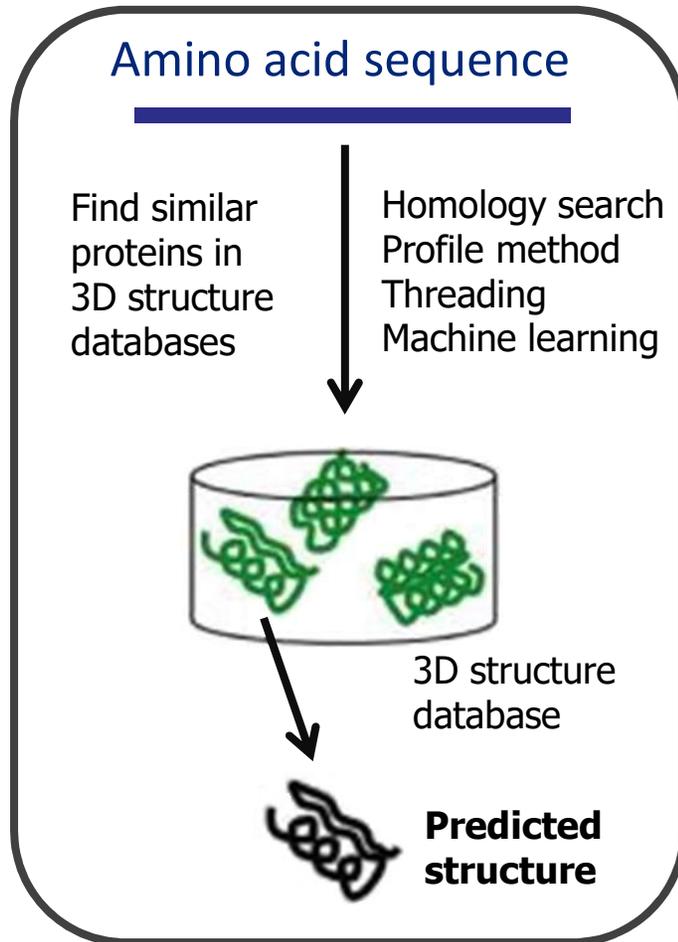
- ❑ Homology modeling
- ❑ Machine learning
- ❑ *Ab initio* prediction



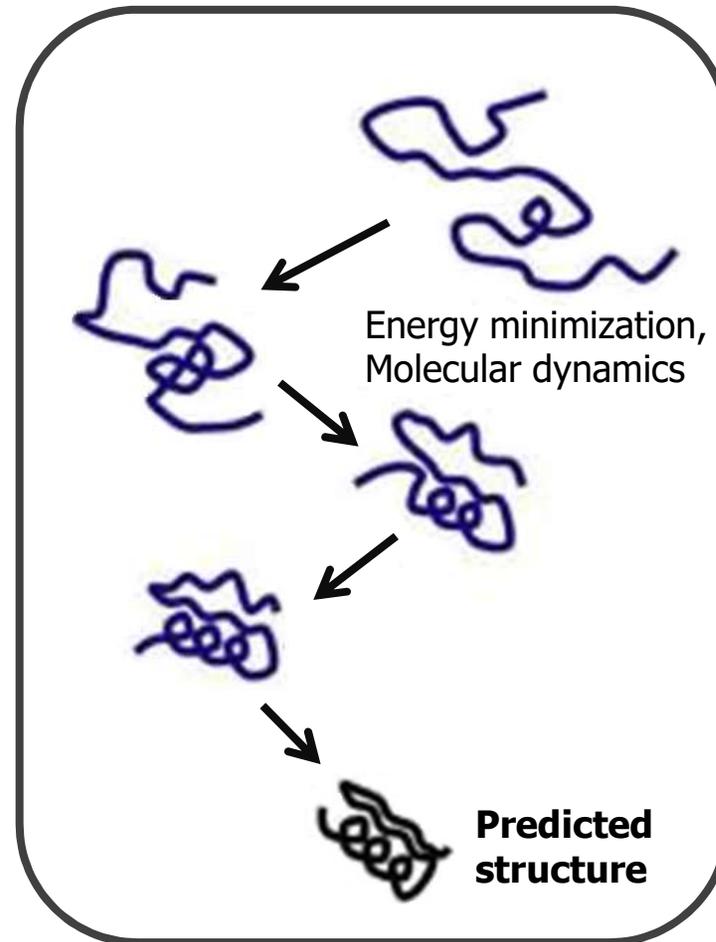
# Bioinformatics predictions



## Comparative modelling



## Ab initio predictions



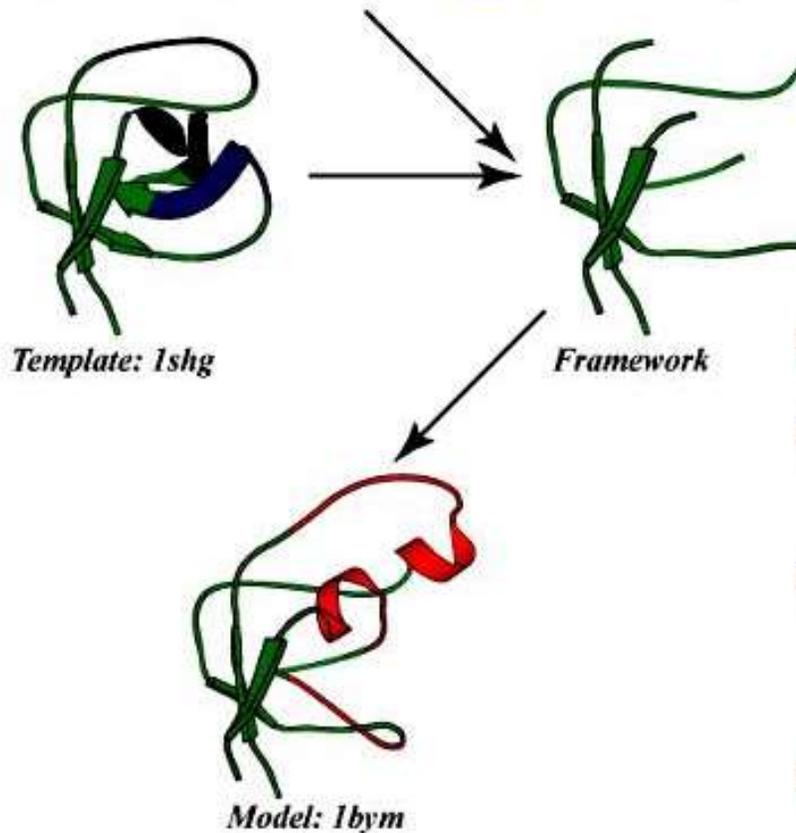
# Bioinformatics predictions



## □ Homology modeling

```
Ishg  KELVLALYDYE-----KSPREVTMKKGDILTLNNTKDNWKEVNDRCGFV---PAAKVKKLD  
Ibym  RKVRIVQIIEIFQVETDQPTQLLDADIRVGSSEVEIVDRDCHI--TISHNGKIVELLDLAEIIRIEE
```

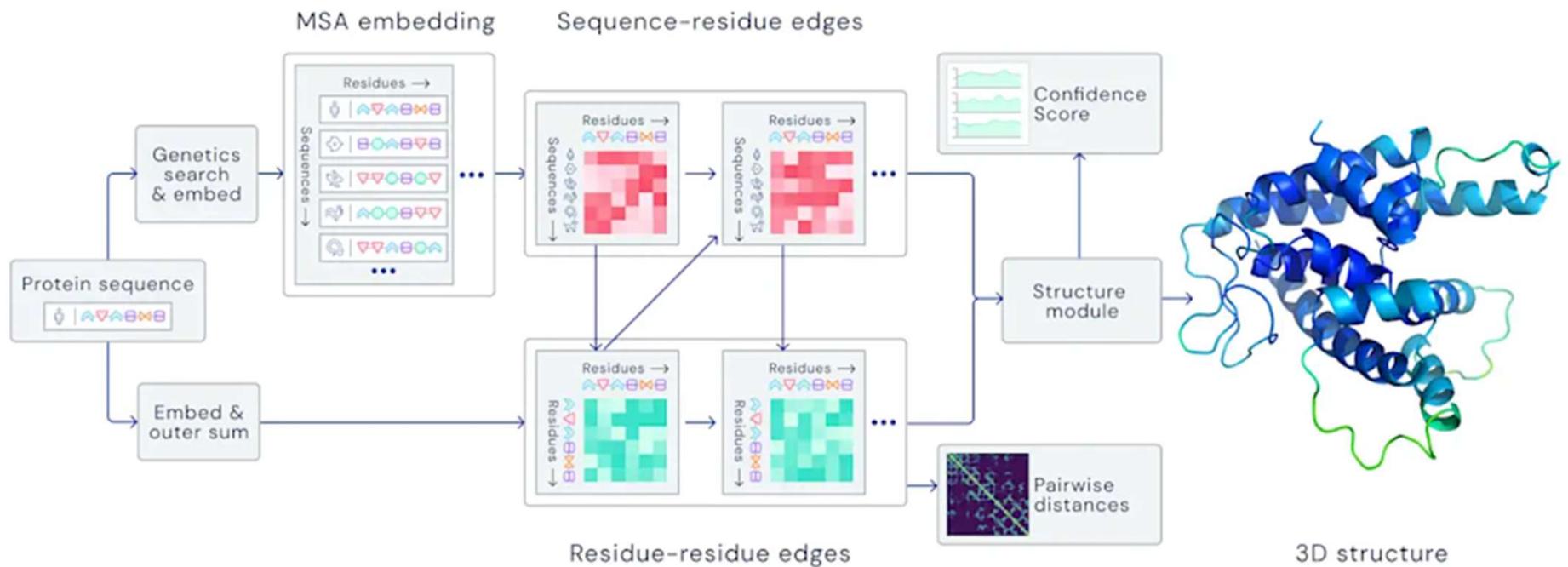
Comparison of sequences  
in databases



- Find template
- Align target sequence with template
- Generate model:
  - add loops
  - add sidechains
- Refine model

# Bioinformatics predictions

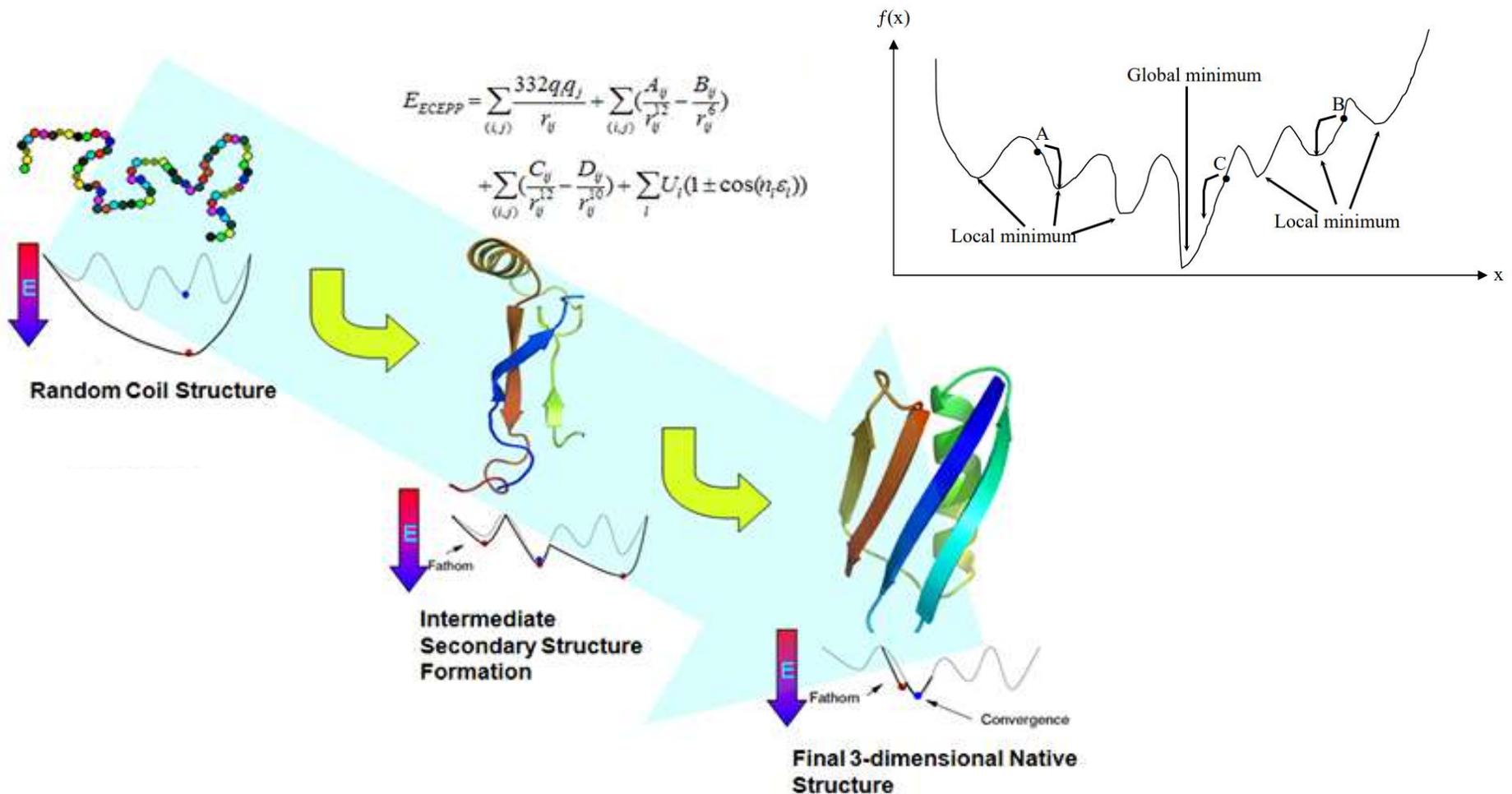
- Machine learning
  - Training on sequence and 3D databases
  - Ex.: AlphaFold 2



# Bioinformatics predictions



## Ab initio prediction



# Bioinformatics predictions



## □ Advantages

- very fast
- low cost

## □ Disadvantages

- theoretical model – experimental validation is needed
- *Ab initio* is very demanding

# References

- ❑ Petsko, G. A. & Ringe, D. (2004). **Protein Structure and Function**, New Science Press, London.
- ❑ Gu, J. & Bourne, P. E. (2009). **Structural Bioinformatics, 2<sup>nd</sup> Edition**, Wiley-Blackwell, Hoboken.
- ❑ Liljas, A. *et al.* (2009). **Textbook Of Structural Biology**, World Scientific Publishing Company, Singapore.
- ❑ Schwede, T. & Peitsch, M. C. (2008). **Computational Structural Biology: Methods and Applications**, World Scientific Publishing Company, Singapore.
- ❑ O'Donoghue, S. *et al.* (2010) Visualization of macromolecular structures. *Nature Methods* **7**: S42–S55.
- ❑ Electrostatic Interactions in Protein Structure, Folding, Binding, and
- ❑ Zhou, H-X. & Pang, X. (2018) Electrostatic interactions in protein structure, folding, binding, and condensation. *Chemical Reviews*. **118**: 1691–1741