Advanced biochemistry and its methods Lecture 5

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Žídek: Strukturní biochemie (skripta k prednášce C9530), kapitoly 11–14

COMPUTATIONAL:

De novo structure calculation not reliable

Homology modeling

using a similar known structure as a starting model

Prediction based on Machine Learning

using sequence alignment and database including known structures (AlphaFold) using a similar known structure as a starting model

Structure determination based on experiment

 Structure determination based on experiment usually MD simulation with experimental restraints

EXPERIMENTAL:

Mass Spectrometry

mostly analysis of cross-linked fragments

Scanning surface

e.g. Atomic Force Microscopy (various modes)

Interactions of electromagnetic (other) waves with molecules:

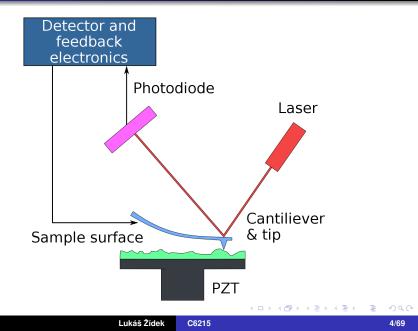
Spectroscopy

how molecules change characteristics of the wave (intensity, phase, polarization, frequency)

• diffraction methods (Microscopy)

how molecules change direction of wave in space

Atomic force microscopy

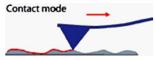


Atomic force microscopy

Scanning probe microscopy

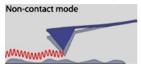


Imaging mode in atomic force microscopy



Feedback : lever deflection

the feedback system adjusts the height of the cantilever base to keep this deflection constant as the tip moves over the surface (friction force microscopy, conductive probe AFM)



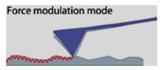
Feedback : oscillation amplitude the cantilever oscillates close to the sample surface, but without making contact with the surface.

Electrostatic / magnetic force microscopy

Intermittent contact

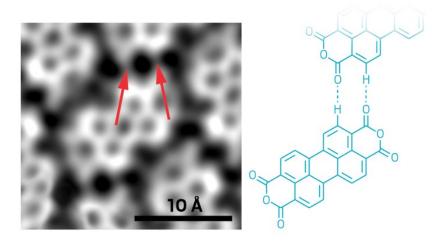


Feedback : oscillation amplitude The cantilever oscillates and the tip makes repulsive contact with the surface of the sample at the lowest point of the oscillation (Tapping mode AFM)



Feedback : lever deflection the tip does not leave the surface at all during the oscillation cycle. (interfacial force microscopy)

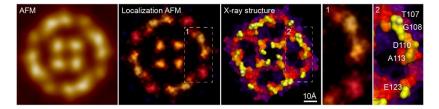
Atomic force microscopy



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image reconstruction "localization AFM" cf. super-resolution microscopy



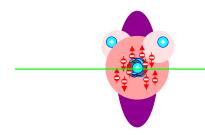
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UV/VIS spectrophotometry

absorption, transition of electrons to higher orbitals concentration, content of aromatic amino acids, heme, prosthetic groups

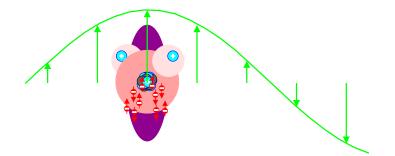
IR spectroscopy

absorption, transition of nuclei to higher vibration states



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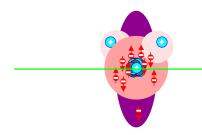
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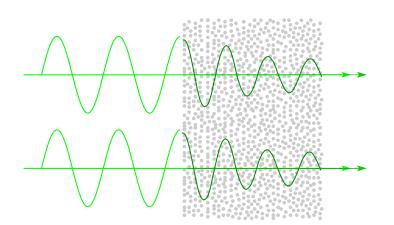
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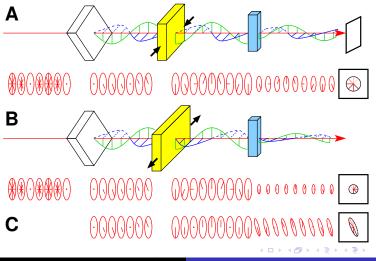
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CD Spectroscopy

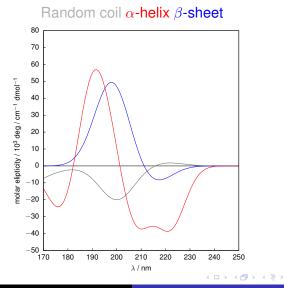
• CD spectroscopy

absorption differences of polarized light by chiral molecules



CD spectroscopy and secondary structure

overall content of secondary structures



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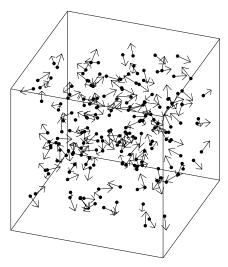
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• NMR spectroscopy

precession of magnetic moments of nuclei in magnetic field magnetic moments are slightly aligned in a static magnet axis of alignment is tilted by electromagnetic (radio) waves aligned magnetic moments precess about the static field resulting oscillating magnetic field is measured we do not observe the applied electromagnetic waves interactions of magnetic moments (mutual, with electrons) \Rightarrow resolution, structural information atomic resolution structure determination, dynamics, interactions

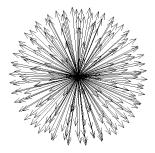
Nuclear magnetic resonance



	S	$rac{10^{-9}\gamma}{ m rads^{-1}T^{-1}}$	%
e-	1/2	-182.000	100
¹ H	1/2	0.277	99.98
¹³ C	1/2	0.067	1.1
¹⁴ N	1	0.019	99.6
¹⁵ N	1/2	-0.027	0.4
¹⁷ O	5/2	-0.036	0.04
¹⁹ F	1/2	0.252	100
³¹ P	1/2	0.108	100
¹²⁹ Xe	1/2	-0.075	24.4

quadrupolar (relax fast) rare isotopes (enrichment)

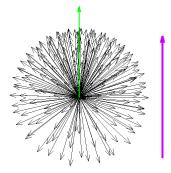
NMR sample outside magnet



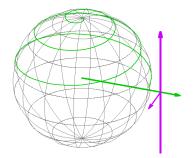
in equilibrium (spherical symmetry)

Polarization

Boltzmann distribution: $P(\theta) \propto e^{-\frac{E}{k_{\rm B}T}} = e^{\frac{\vec{\mu} \cdot \vec{B}}{k_{\rm B}T}} \Rightarrow M_{Z} = \frac{N}{V} \frac{\mu^{2}B}{3k_{\rm B}T}$ Precession (angular momentum in a field): $\vec{\omega} = -\gamma B$



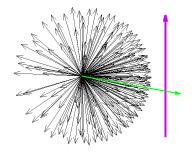
Excitation



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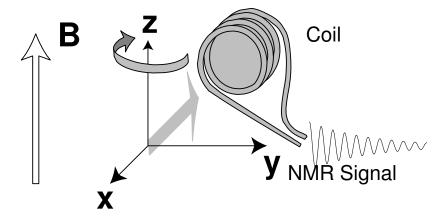
Coherent evolution



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NMR signal detection



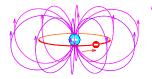
reproduced from M. H. Levitt: Spin Dynamics

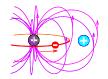
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Interactions with other nuclei and electrons





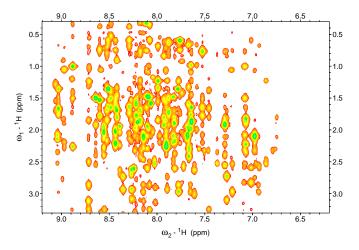


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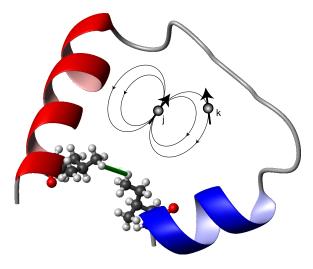
Nuclear Overhauser effect from NOESY spectra

Nuclear Overhauser effect: result of dipole-dipole interactions Peak intensities are proportional to nuclear Overhauser effect

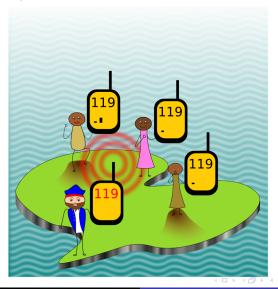


Distances from nuclear Overhauser effect

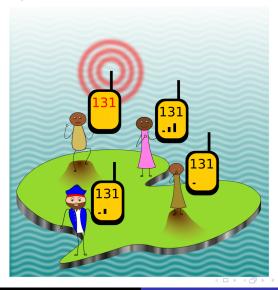
Nuclear Overhauser effect proportional to $1/d^6$ (*d* = distance)



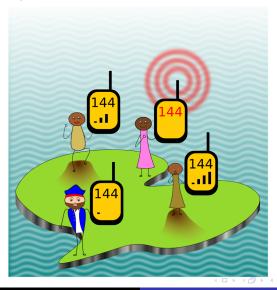
Model of the protein built from known distances



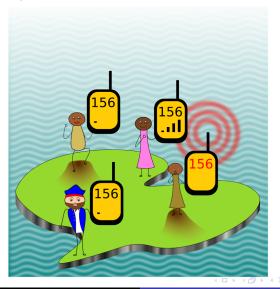
Model of the protein built from known distances



Model of the protein built from known distances



Model of the protein built from known distances



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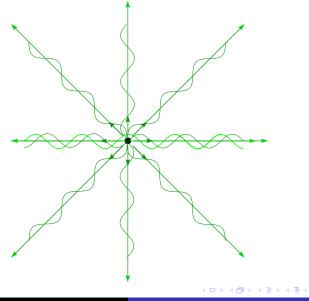
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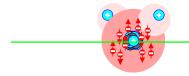
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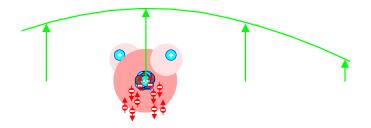
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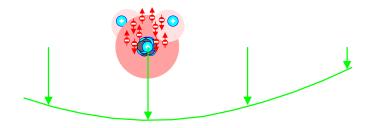




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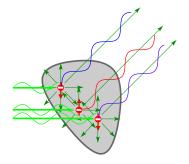
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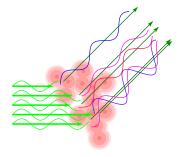
Scattering by large molecules



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Scattering by large molecules



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Scattering by large molecules

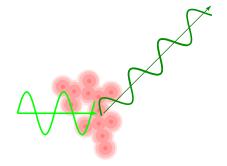
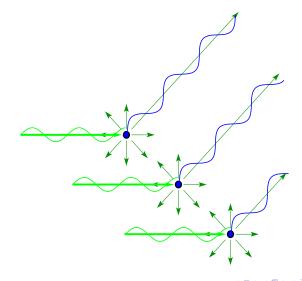


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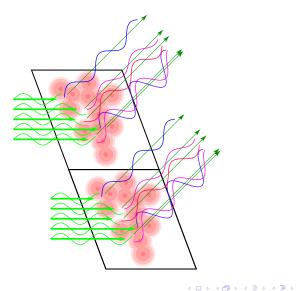
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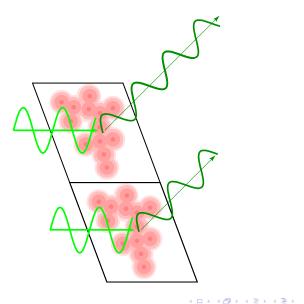
Constructive interference in crystals



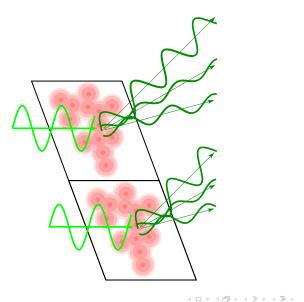
Electron density



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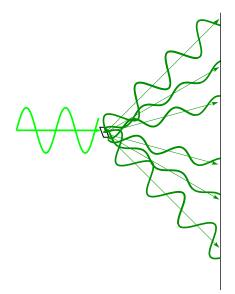


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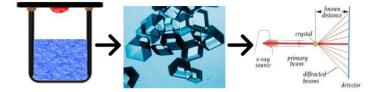


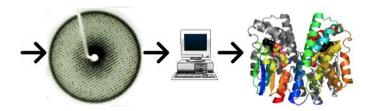
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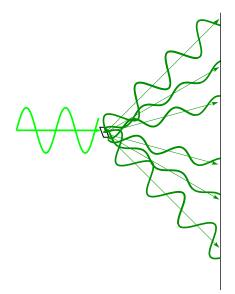
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X-ray crystallography





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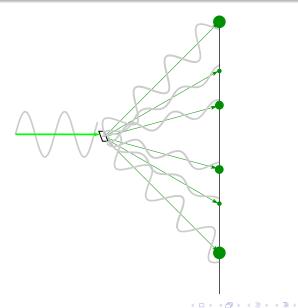


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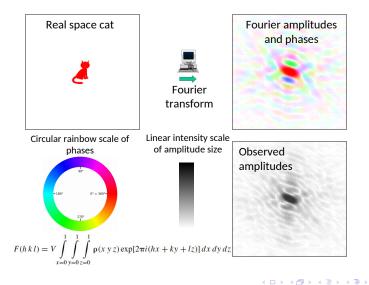
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Structure calculation



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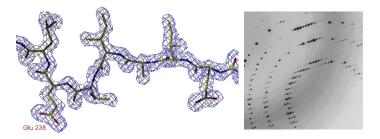
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Electron density equation & PHASE PROBLEM

$$\rho(x \ y \ z) = \frac{1}{V} \sum_{h} \sum_{k} \sum_{l} \left| F(h \ k \ l) \right| \exp\left[-2\pi i (hx + ky + lz) + i \alpha(h \ k \ l)\right]$$
$$F(h \ k \ l) = |F(h \ k \ l)| e^{i\alpha(h \ k \ l)}$$



• Direct interpretation of amplitudes

mutual positions of atoms calculated from amplitudes for simple molecules (Patterson function/map)

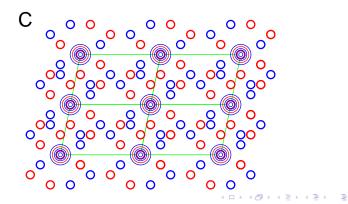
• Using heavy atoms

Molecular replacement

Diffraction back-calculated from a known structure similar to the studied proteins Orientation and position of the molecule in the crystal obtained by searching for the match of diffraction patterns (measured vs. back-calculated) Calculated phases used for the unknown molecule

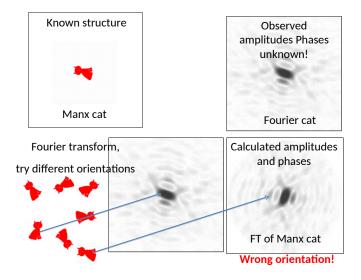
Patterson maps



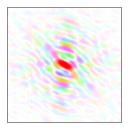


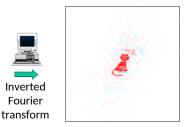
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Molecular replacement



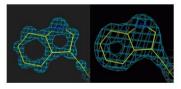
Observed **amplitudes** (tailed cat), calculated **phases** (Manx cat)





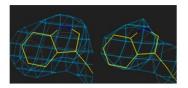
Even the tail becomes visible!

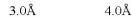
Model building & resolution









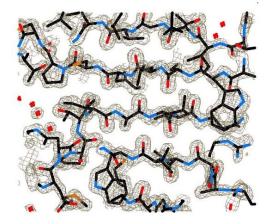




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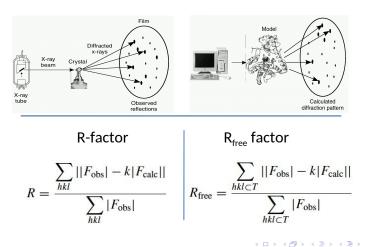
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Model building & refinement



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R-factor, R_{free}factor



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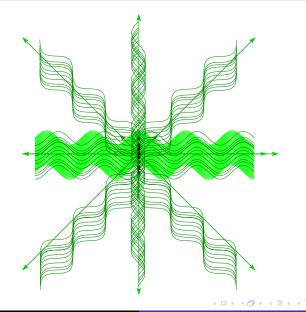




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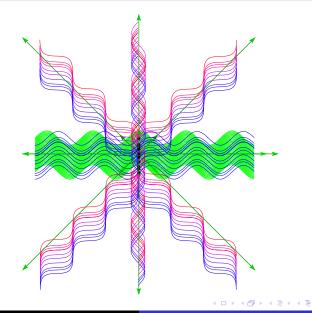


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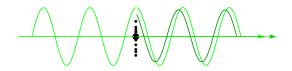
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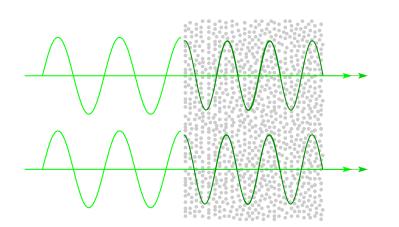


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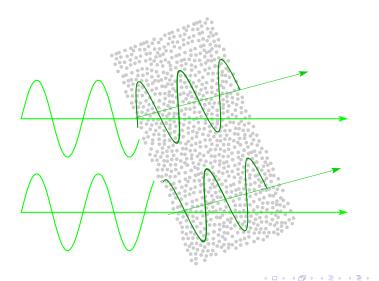
Refractive index



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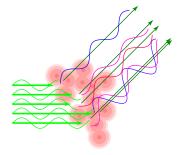
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Refractive index



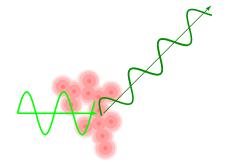
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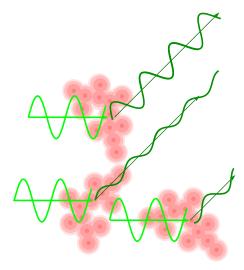
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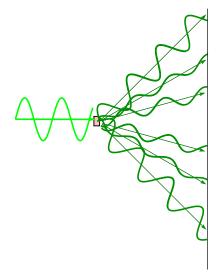
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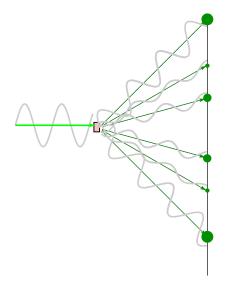
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crystal diffraction:

$$I(\vec{q}) = \int_{V} \int_{V} \Delta \rho(\vec{r}) \Delta \rho(\vec{r} + \Delta \vec{r}) e^{i\vec{q}\cdot\Delta \vec{r}} d\vec{r} d(\vec{r} + \Delta \vec{r})$$
SAXS:
$$I(q) = \left\langle \int_{V} \int_{V} \Delta \rho(\vec{r}) \Delta \rho(\vec{r} + \Delta \vec{r}) e^{i\vec{q}\cdot\Delta \vec{r}} d\vec{r} d(\vec{r} + \Delta \vec{r}) \right\rangle_{\theta_{q},\phi_{q}}$$

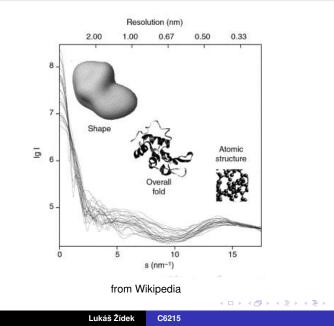
$$= 4\pi \int_{0}^{D_{\text{max}}} \underbrace{r^{2} \left\langle \int \Delta \rho(\vec{r}) \Delta \rho(\vec{r} + \Delta \vec{r}) d\Delta \vec{r} \right\rangle_{\theta_{r},\phi_{r}}}_{P(r) = r^{2} V \rho^{2} \gamma_{0}(r)} \underbrace{\frac{\sin(qr)}{qr} dr}_{P(q) = r^{2} V \rho^{2} \gamma_{0}(r)}$$
Guinier law:
$$I(q) \approx I(0) e^{-R_{g}^{2}q^{2}/3} \text{ for } q \rightarrow 0$$

 θ : scattering angle \vec{q} : scattering vector $q = |\vec{q}| = 4\pi \sin(\theta/2)/\lambda$: momentum transfer I(q): scattering intensity in direction given by q γ_0 : probability of finding a point at r from a given point R_9 : radius of gyration

Svergun and Koch, Rep. Prog. Phys. 66 (2003) 1735, stacks.iop.org/RoPP/66/1735

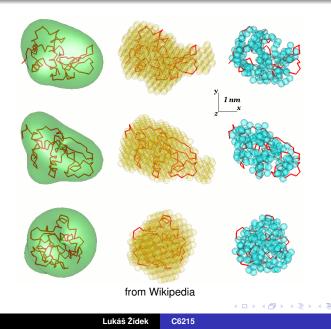
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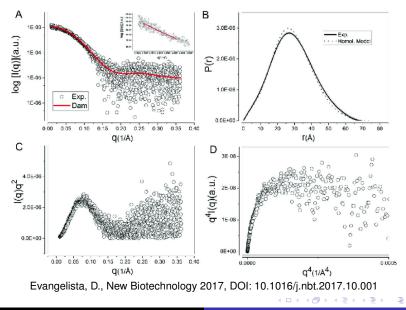


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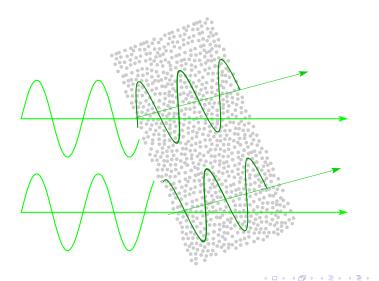


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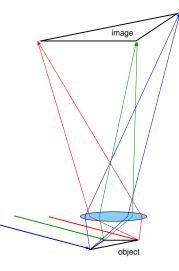
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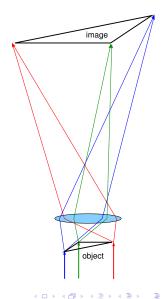
Refractive index



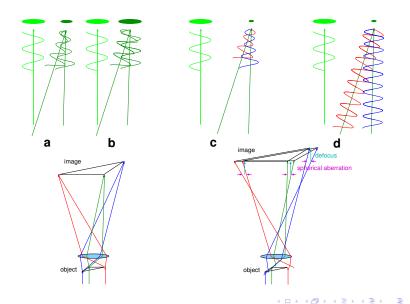
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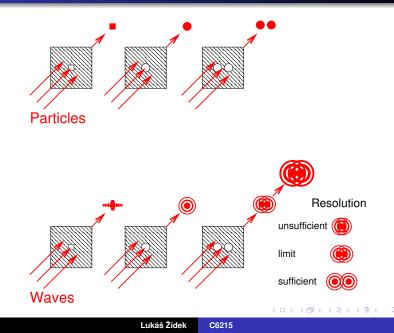
Phase vs amplitude objects



Phase vs amplitude objects

- a Perfect lens in focus provide contrast in intensity of scattered vs. original wave if the scattered wave is absorbed by the sample
- b Perfect lens in focus provide no contrast in intensity of scattered vs. original wave if the scattered wave is not absorbed by the sample, but only phase-shifted by 90°
- c Imperfect and defocused lens provide contrast in intensity of scattered vs. original wave if the scattered wave is absorbed by the sample
- d Defocus and spherical aberration of lens provide contrast in intensity of scattered vs. original wave if the scattered wave is not absorbed by the sample, but only phase-shifted by 90°. Defocus and spherical aberration introduce another phase shift ⇒ scattered have oposite phase and cancel each other.

Diffraction by aperture



Limitation of optical microscopy

Photons scatter as waves \Rightarrow limited resolution resolution < 0.61 λ/n (λ = wavelegth, n = refractive index $\Rightarrow \lambda \approx 0.1$ nm (distances of atoms in molecules), X-rays Lens are not available for X-rays ($\lambda \approx 0.1$ nm) no material has sufficient refractive index

SOLUTIONS:

• Analysis of diffraction patterns

intensity enhanced if molecules are aligned in crystals

\Rightarrow X-ray crystallography

applicable also to electron and neutron waves

• Microscopy with electrically charged waves

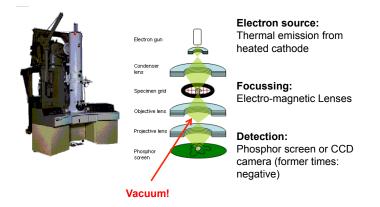
electrons scatter as waves electron beams are bent in electromagnetic field

 \Rightarrow Electron microscopy

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Transmission electron microscope

Transmission Electron Microscope

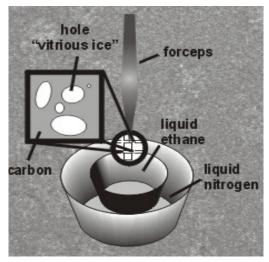




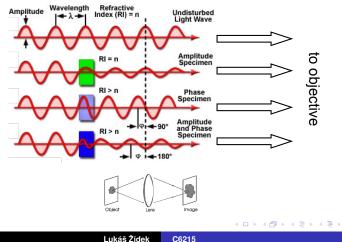
<ロト < 部ト < きと < きと < きと < 300 × 100 ×

Cryo-electron microscopy

Proteins in vitrous ice, can reach atomic resolution



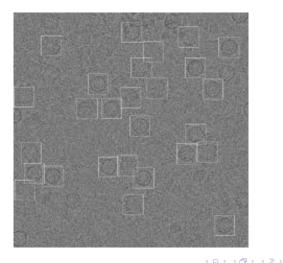
Macromolecules in water / vitreous ice are phase objects





Samples damaged by electrons

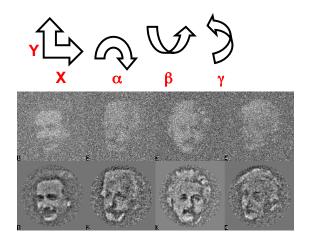
Signal to noise ratio



Lukáš Žídek

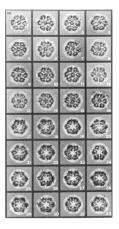
Alignment and classification

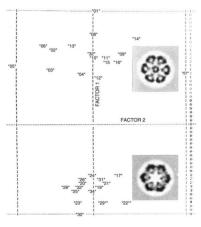
Images contain diffrent views of possibly different molecules



Alignment and classification

Classification and averaging (principal component analysis)





3D reconstruction

Iterative process:

2D projections are calculated from a 3D model alignment and classification are improved iteratively

